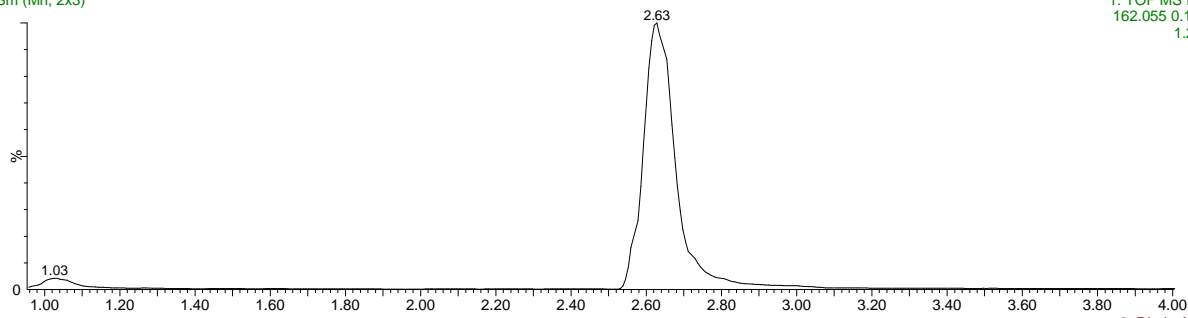


0.15 E 10

DHQ37b-5 Sm (Mn, 2x3)

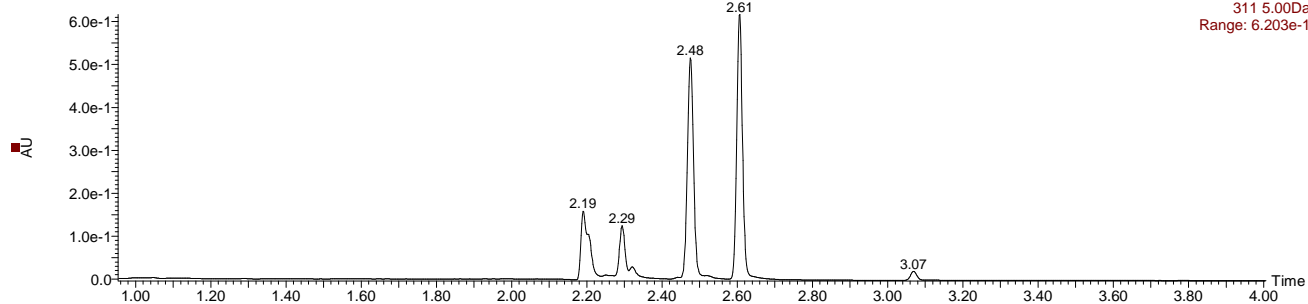


11-May-2009

3::0::0

1: TOF MS ES+
162.055 0.10Da
1.20e3

DHQ37b-5



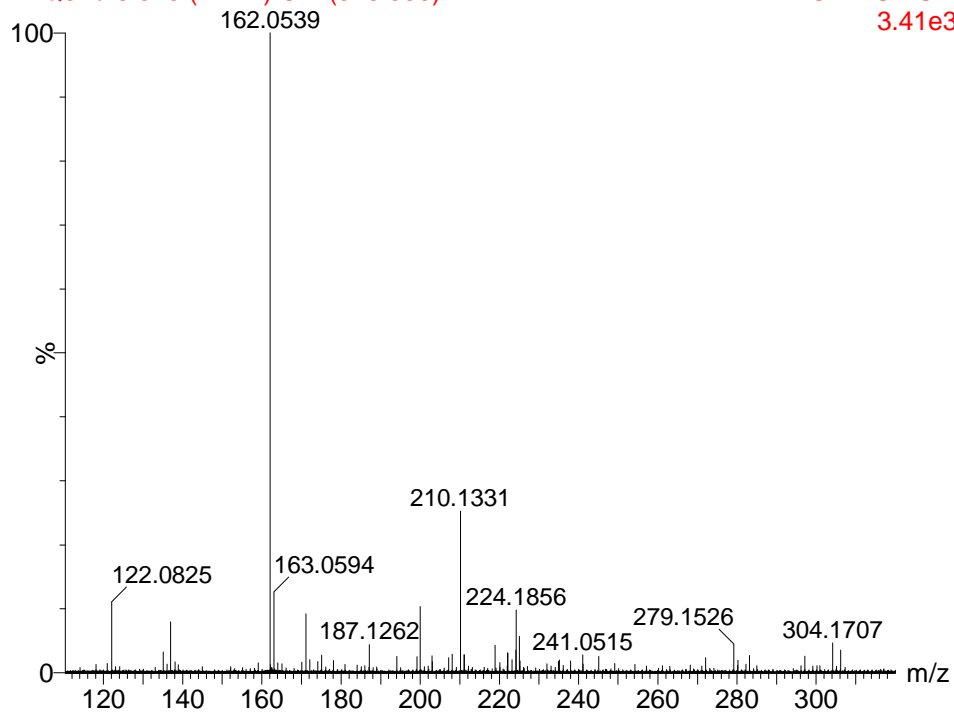
3: Diode Array
311 5.00Da
Range: 6.203e-1

Supplementary Figure 1A. HPLC analysis of PqsD activity. Upper panel, extracted ion chromatogram of DHQ ($m/z = 162.055$; retention time, 2.63 min) identifies the peak at 2.62 minutes in the lower panel as DHQ.

0.15 E 10

DHQ37b-5 325 (2.744) Cm (325:330)

1: TOF MS ES+
3.41e3

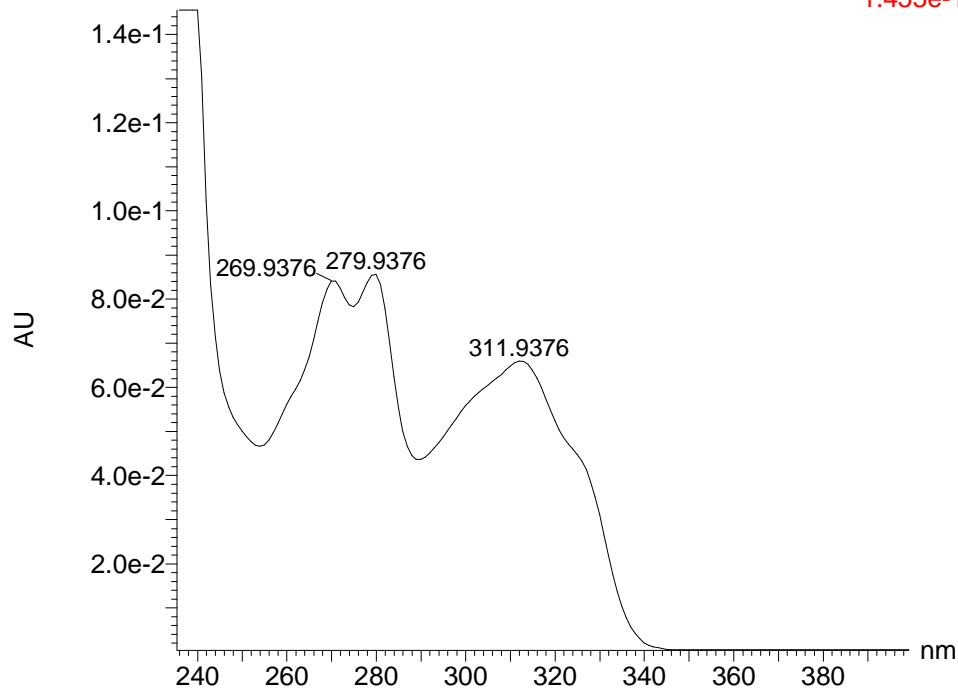


Supplementary Figure 1B. Electrospray ionization mass spectrum of the material eluting at 2.62 min in *Supplementary Figure 1B* above confirms that it is DHQ ($m/z_{obsd} = 162.0539$; $m/z_{calcd} = 162.0555$).

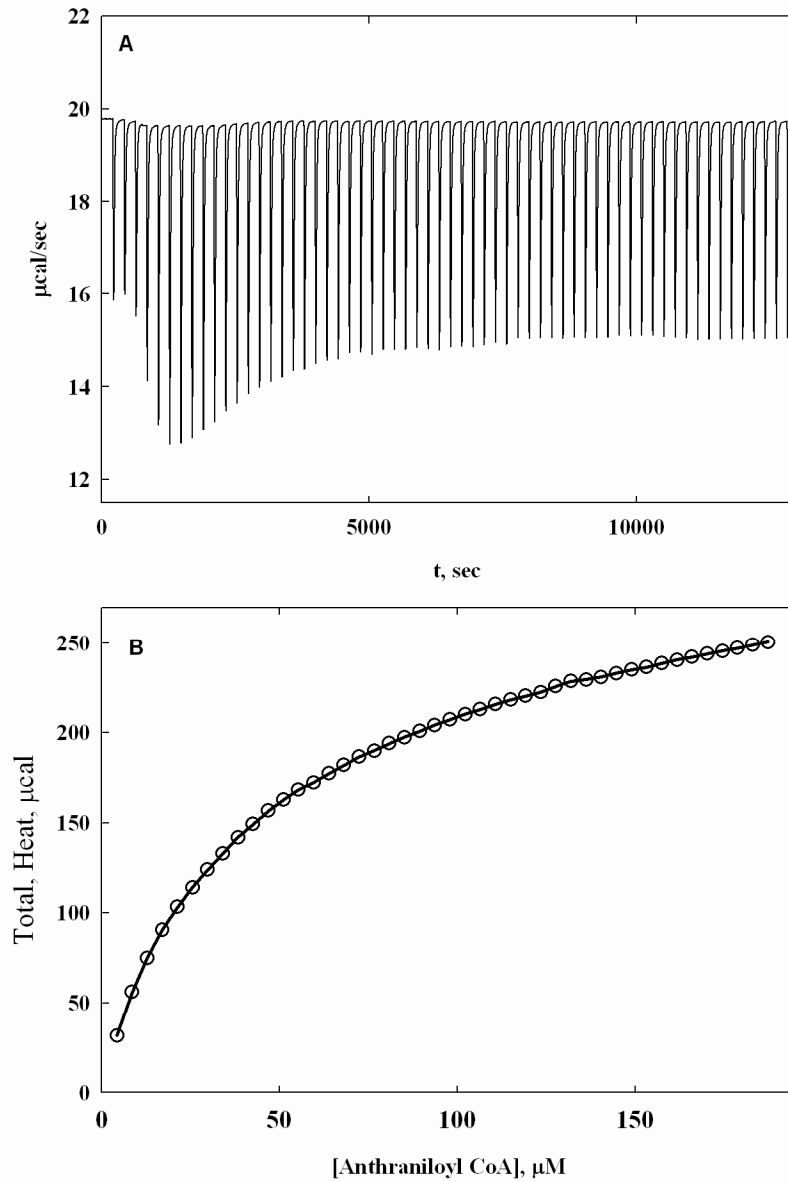
0.15 E 10

DHQ37b-5 1565 (2.607) Cm (1561:1577)

3: Diode Array
1.455e-1



Supplementary Figure 1C. UV spectrum of the material eluting at 2.62 min in *Supplementary Figure 1B* above is consistent with the UV spectrum of authentic DHQ.



Supplementary Figure 2. Analysis of the interaction between Cys112Ala PqsD and ACoA by isothermal titration calorimetry. (A) Raw ITC data. A considerable heat of dilution is apparent and was noted in similar injections into buffer only. (B) Data were analyzed by plotting total heat evolved vs [ACoA] and fitting to the following hyperbolic function where y is cumulative heat evolved, y_0 is an offset term for a nonzero baseline, a

is the amplitude of the curve, and b is the apparent k_d . The first four injections were excluded from the fit due to noise.

$$y = y_0 + \frac{ax}{b+x} \quad (\text{Supplementary Equation 1})$$