

Supporting Information

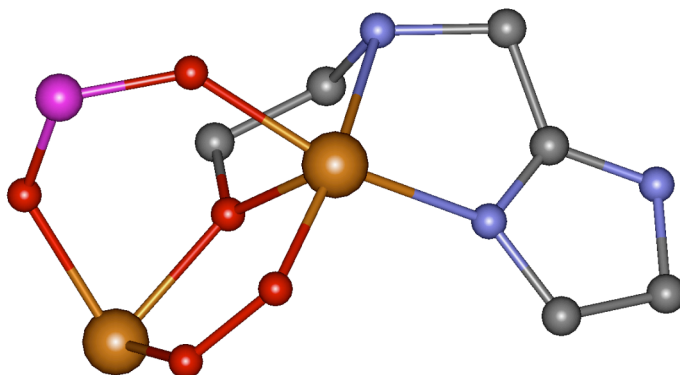


Figure S1. FEFF models for $2\bullet\text{O}_2\text{PPh}_2$ and $2\bullet\text{O}_2\text{AsMe}_2$, based on combination of crystal structures of $1\bullet\text{O}_2\text{PPh}_2$, $1\bullet\text{O}_2\text{AsMe}_2$ and $[\text{Fe}_2(\text{N-EtHPTB})(\text{O}_2)(\text{OPPh}_3)_2]^{3+}$ (Dong *et al.*)¹ using Accelrys DS Visualizer 1.7 and ChemBio 3D Ultra 11.0. Black = C, Blue = N, Red = O, Brown = Fe, Pink = P/As.

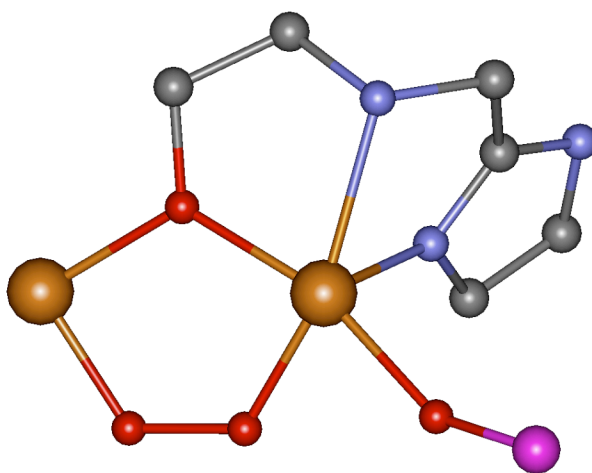


Figure S2. FEFF model for $3\bullet\text{O}_2\text{PPh}_2$, based on crystal structures of $[\text{Fe}_2(\text{N-EtHPTB})(\text{O}_2)(\text{OPPh}_3)_2]^{3+}$ (Dong *et al.*)¹ using Accelrys DS Visualizer 1.7 and ChemBio 3D Ultra 11.0. Black = C, Blue = N, Red = O, Brown = Fe, Pink = P.

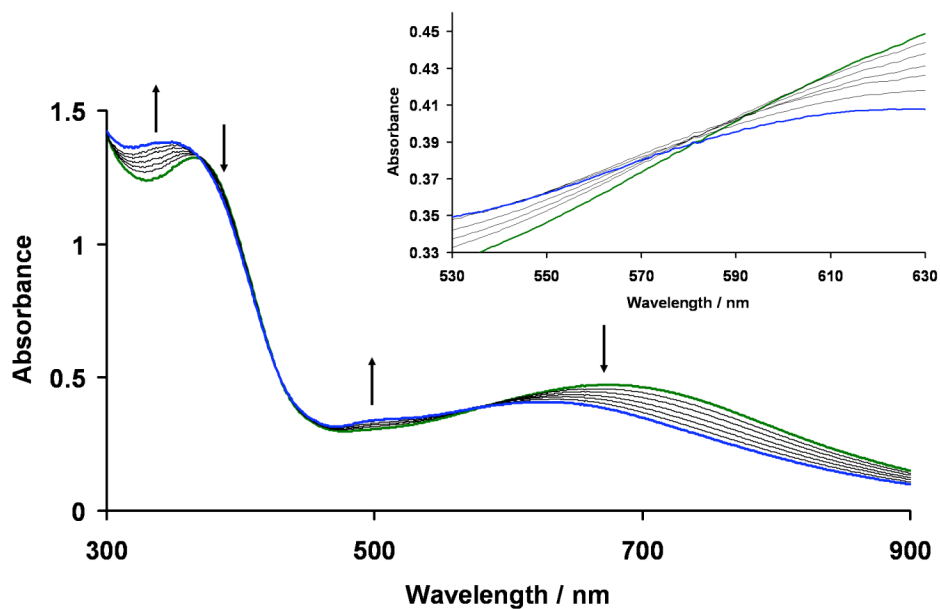


Figure S3. Partial conversion of a 0.5 mM solution of $2\bullet\text{O}_2\text{PPh}_2$ to $3\bullet\text{O}_2\text{PPh}_2$ in CH_2Cl_2 in a 0.5 cm cuvette at $-30\text{ }^\circ\text{C}$. The inset reveals the pseudo-isosbestic point at $\sim 580\text{ nm}$.

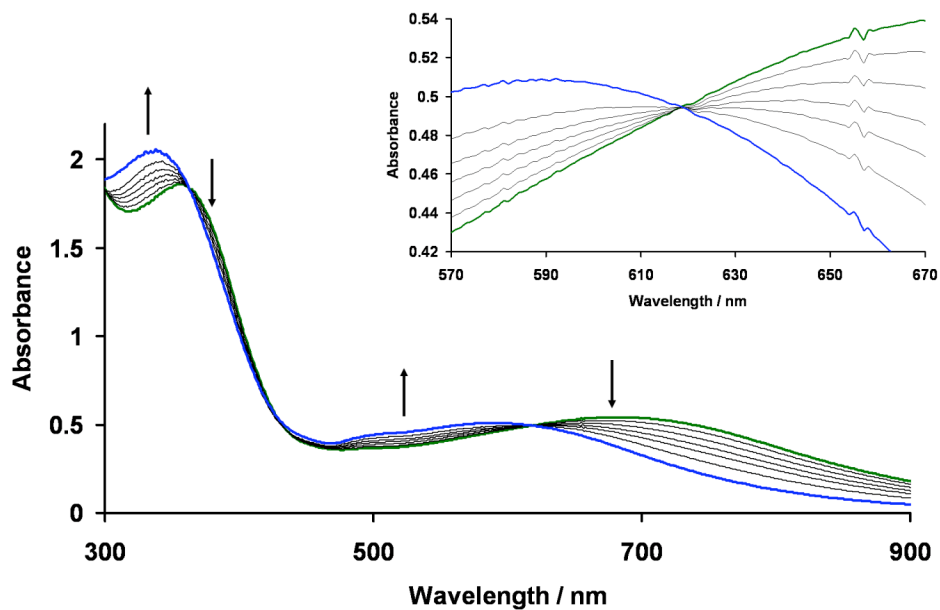


Figure S4. Conversion of a 0.5 mM solution of $2\bullet\text{O}_2\text{PPh}_2$ to $3\bullet\text{O}_2\text{PPh}_2$ in MeCN in a 0.5 cm cuvette at $-30\text{ }^\circ\text{C}$. The inset reveals the isosbestic point at $\sim 620\text{ nm}$.

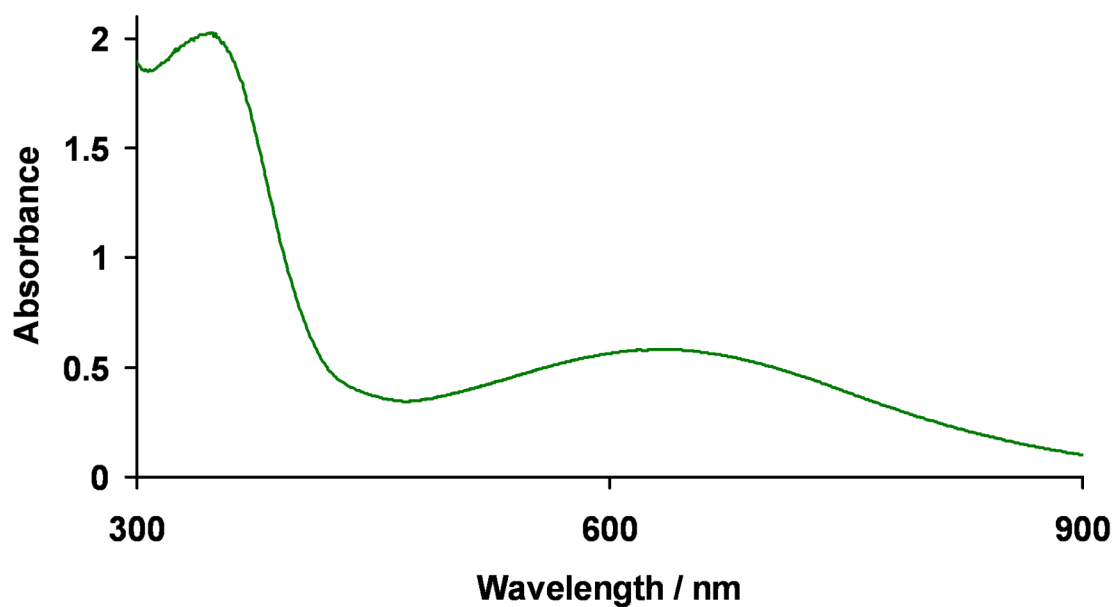


Figure S5. UV-Vis spectrum of a 0.5 mM solution of $2\bullet\text{O}_2\text{AsMe}_2$ in MeCN in a 0.5 cm cuvette at $-40\text{ }^\circ\text{C}$.

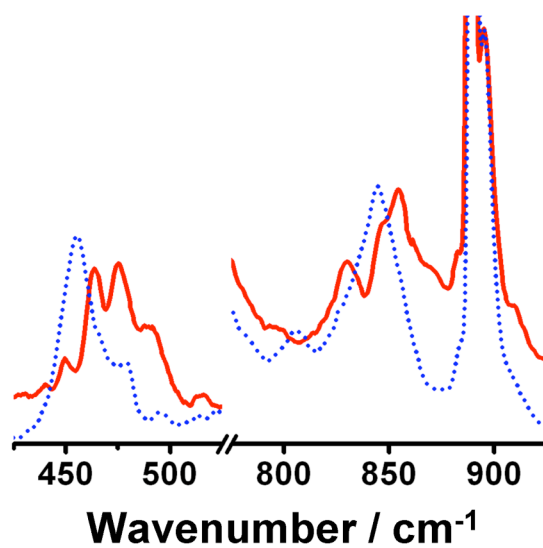


Figure S6. Resonance Raman spectra of frozen solutions of $3\bullet\text{O}_2\text{PPh}_2$ in CH_2Cl_2 . Samples are contaminated with $2\bullet\text{O}_2\text{PPh}_2$. Solid-red = $^{16}\text{O}_2$. Dotted-blue = $^{18}\text{O}_2$.

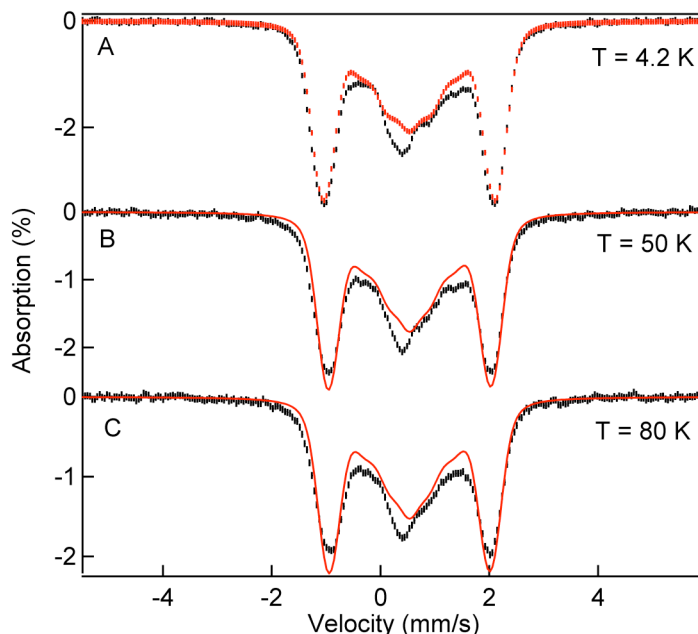


Figure S7. 8.0T Mössbauer spectra of $3\bullet\text{O}_2\text{PPh}_2$ in MeCN recorded at 4.2 K (A) and 50 K (B) and 80 K (C). The solid lines are theoretical curves using the parameters listed in Table 3. The solid line in (B) was obtained for $J = 60 \text{ cm}^{-1}$.

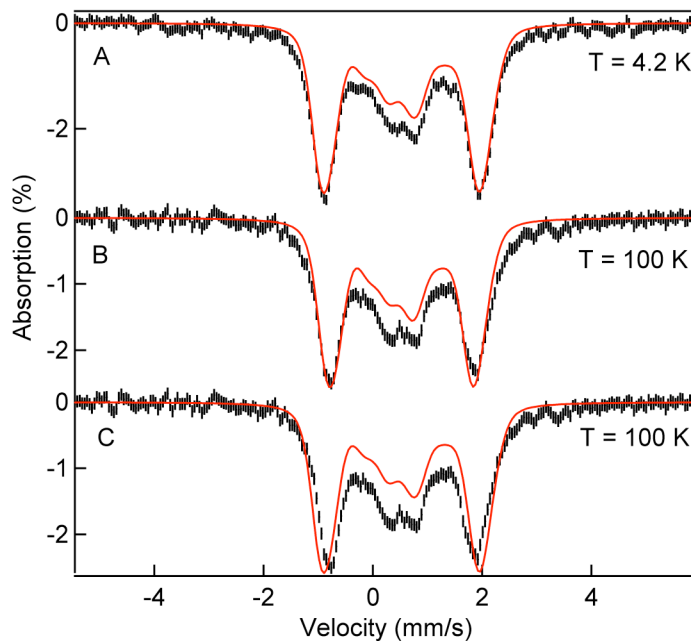


Figure S8. 8.0T Mössbauer spectra of $3\bullet\text{O}_2\text{PPh}_2$ in CH_2Cl_2 recorded at 4.2 K (A) and 100 K (B). The solid lines are theoretical curves using the parameters listed in Table 3. The data in (C) are the same as in (B); the solid line in (C) is a simulation obtained by assuming that only the $S = 0$ ground state is occupied at 100 K, i. e. using $J = 1000 \text{ cm}^{-1}$. The solid line in (B) was obtained for $J = 60 \text{ cm}^{-1}$.

Table S1. EXAFS fitting results for **2•O₂PPh₂**, **2•O₂AsMe₂**, and **3•O₂PPh₂**.^a

Complex	Fit	Fe-O/N			Fe-O/N			Fe-O/N			Fe...C			Fe...X			F ^c	F ^d
		N	R	σ ²	N	R	σ ²	N	R	σ ²	N	R	σ ²	N	R	σ ²		
2•O₂PPh₂	1	6	1.97	11.37													1.384	0.117
	2	5	2.00	8.64	1	2.30	1.13										1.225	0.104
	3	4	2.00	6.23	1	2.29	2.92				5	2.99	1.98				0.799	0.052
	4	4	2.00	6.32	1	2.29	2.00				5	2.98	3.84	1Fe	3.27	3.59	0.601	0.035
	5	4	1.99	6.14	1	2.29	3.31				5	2.94	10.66	1P	3.12	-0.30	0.488	0.023
	6	4	2.00	6.54	1	2.31	3.64				5	2.99	2.19	5C	2.21	2.26	0.662	0.042
	7	4	1.99	6.22	1	2.29	2.50				5	2.95	9.28	1Fe	3.24	9.60	0.430	0.022
														1P	3.12	0.00		
	8	4	2.00	6.69	1	2.31	2.71	1	2.54	0.58	5	2.99	3.55	1Fe	3.28	3.18	0.392	0.018
9	4	2.00	6.57	1	2.30	1.50	1	2.51	3.00^b	5	2.96	7.95	1Fe	3.25	4.06	0.304	0.013	
													1P	3.14	3.00^b			
2•O₂AsMe₂	1	6	1.94	14.63													1.517	0.149
	2	5	1.97	10.13	1	2.26	1.35										1.368	0.140
	3	5	1.99	10.86	1	2.25	0.12				3	2.97	-1.00				1.091	0.104
	4	5	1.97	10.82	1	2.25	2.19				3	2.93	3.86	1Fe	3.25	0.98	0.611	0.040
	5	5	1.97	10.95	1	2.26	2.85				3	2.90	7.46	1As	3.19	1.94	0.595	0.038
	6	5	1.97	10.66	1	2.25	1.03				3	2.95	1.63	6C	3.28	0.40	0.809	0.079
	7	3	1.97	3.94	2	2.17	4.28	1	2.35	5.35	3	2.94	2.76	1Fe	3.27	1.24	0.381	0.018
	8	3	1.97	3.87	2	2.17	3.82	1	2.36	3.80	3	2.93	5.20	1As	3.21	2.24	0.369	0.019
	9	3	1.97	3.85	2	2.17	3.89	1	2.35	4.24	3	2.92	3.34	1Fe	3.13	19.43	0.344	0.022
	10	3	1.97	3.84	2	2.17	3.90	1	2.36	4.71	3	2.94	2.22	1As	3.21	1.97		
													1Fe	3.27	1.02	0.316	0.019	
													1As	3.45	12.71			
3•O₂PPh₂	1	6	1.98	14.22													1.200	0.081
	2	5	2.00	7.84				1	1.84	1.30							1.143	0.083
	3	5	2.00	11.17	1	2.34	0.34										0.976	0.061
	4	4	2.03	5.68	1	2.34	1.01	1	1.87	1.16							0.914	0.061
	5	4	2.03	5.64	1	2.34	0.42	1	1.88	1.14	4	2.96	2.20				0.568	0.028
	6	4	2.04	5.73	1	2.33	0.48	1	1.88	1.23	4	2.95	2.39	1Fe	3.47	4.59	0.366	0.012
	7	4	2.04	5.74	1	2.33	0.52	1	1.88	1.28	4	2.95	2.30	1P	3.61	1.50	0.452	0.018
	8	4	2.03	5.73	1	2.33	0.67	1	1.88	1.24	4	2.95	2.63	5C	3.50	2.99	0.494	0.021
	9	4	2.04	5.58	1	2.33	0.80	1	1.88	1.12	4	2.95	2.43	1Fe	3.44	3.73	0.332	0.011
													0.5P	3.39	0.69			

^a Resolution ~ 0.12 Å for **2•O₂PPh₂** and **3•O₂PPh₂** and ~ 0.14 Å for **2•O₂AsMe₂**; σ² = respective Debye-Waller factor in units of 10⁻³ Å².

^b σ² value held fixed during optimization.

^c F = goodness of fit calculated as $F = \sqrt{\sum k^6 (\div_{exp} - \div_{cal})^2 / N}$, where N = the number of data points.²

^d F' = F²/ν, where ν = n_{idp} - n_{var}, n_{idp} is the number of independent points in each data set and n_{var} is the number of variables used in each optimization step. F' is used to indicate the improvement of fit upon the introduction of a shell.²

The σ^2 values of the Fe...Fe path (1.24) and the Fe...As path (2.24) in $2\bullet\text{O}_2\text{AsMe}_2$ (Fits 7 and 8, respectively) are significantly smaller than the σ^2 value of 4.06 found for the Fe...Fe paths in $2\bullet\text{O}_2\text{PPh}_2$ (Fit 9). Smaller σ^2 leads to larger calculated amplitudes of Fe...Fe or Fe...As peaks in $2\bullet\text{O}_2\text{AsMe}_2$ as compared to the Fe...Fe path in $2\bullet\text{O}_2\text{PPh}_2$ (Figure S9). In fact, two Fe...Fe/As paths can also lead a good simulation of data in the k-range of 8 to 13 \AA^{-1} (Figure S10).

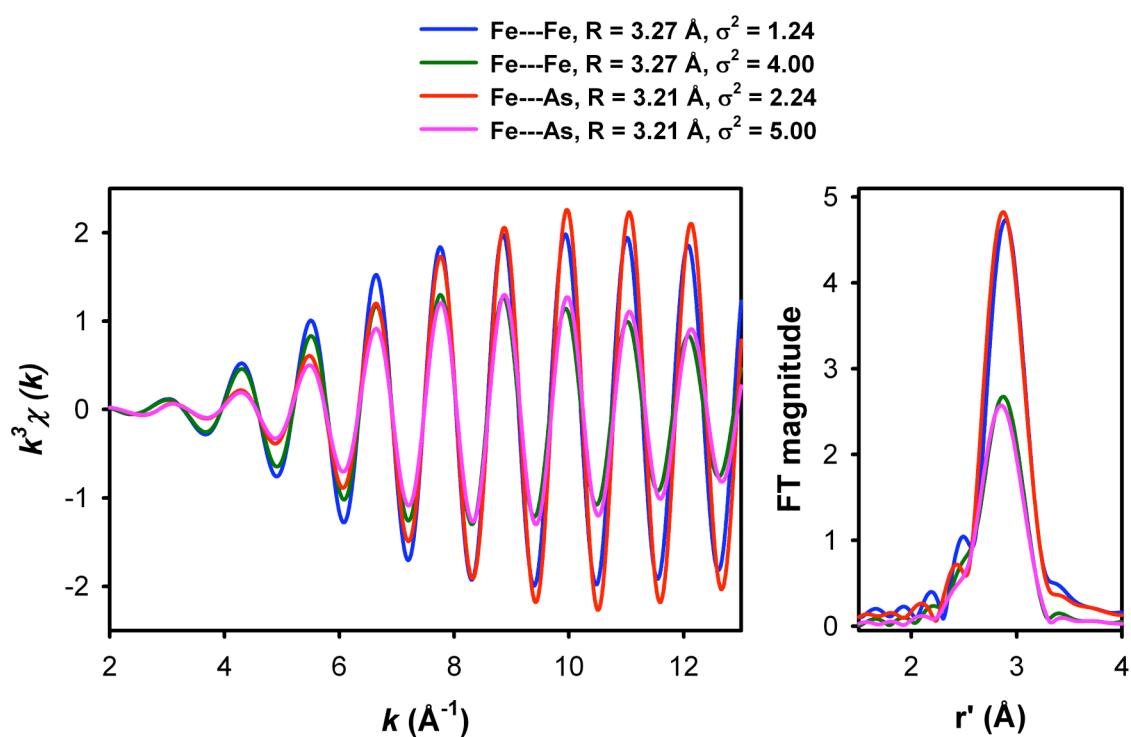


Figure S9. Theoretical EXAFS amplitudes and phases and FT amplitudes at different σ^2 values using FEFF84_87^{3,4} and Artemis.⁵

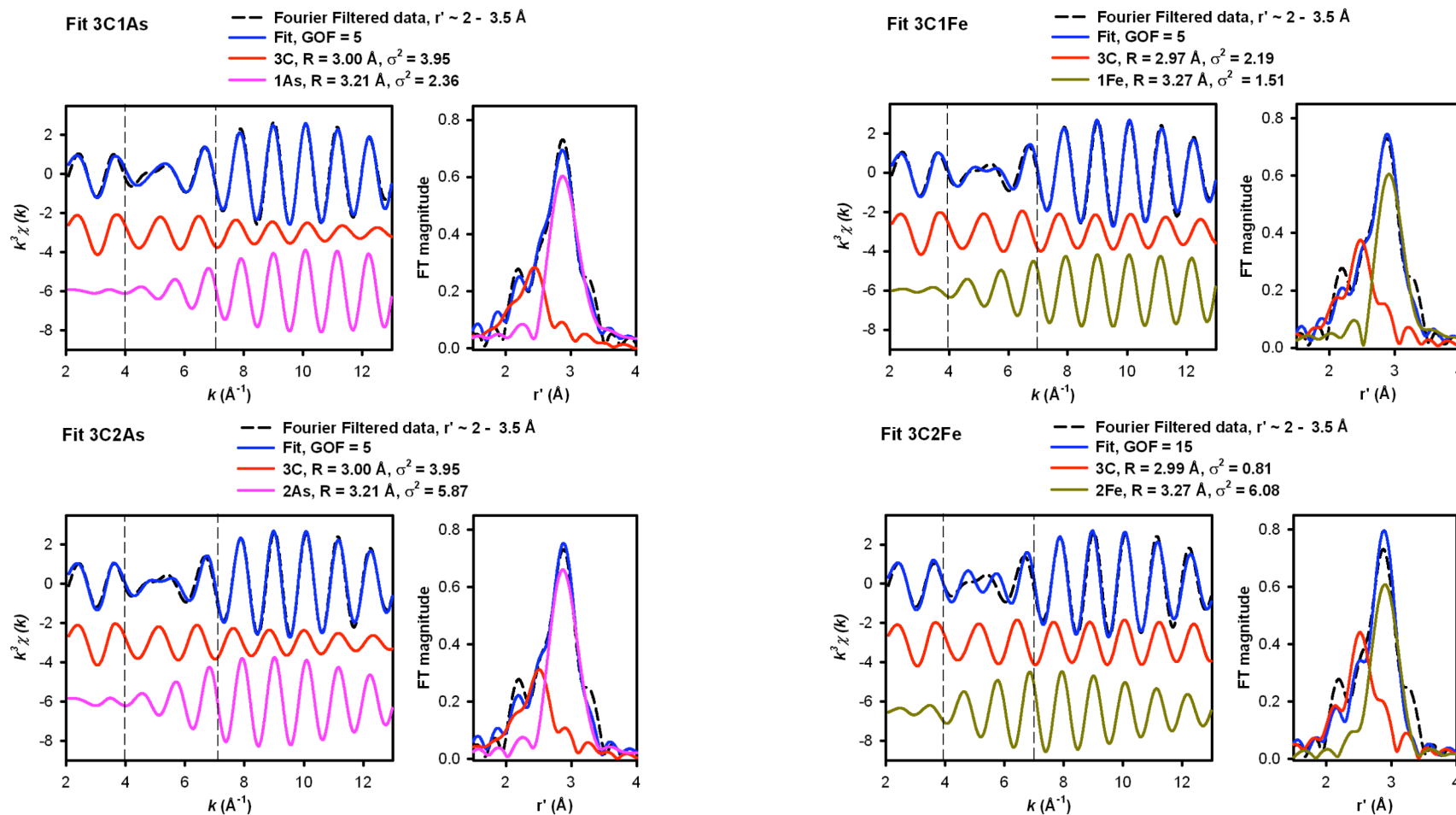


Figure S10. Fits to the outer shell with three Fe...C and one (3C1Fe, 3C1As) or two (3C2Fe, 3C2As) Fe...Fe/As paths. Fits 3C1Fe, 3C1As, and 3C2As are equally good. In fit 3C2Fe, the feature corresponding to the high Z scatterers is also well fitted. However, the destructive effect of two Fe...Fe paths on the 3 Fe...C paths in the range of 4 to 7 \AA^{-1} results in a lower quality of fit for 3C2Fe.

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- (2) Riggs-Gelasco, P. J.; Stemmler, T. L.; Penner-Hahn, J. E. *Coord. Chem. Rev.* **1995**, *144*, 245-286.
- (3) Rehr, J. J.; Mustre de Leon, J.; Zabinsky, S. I.; Albers, R. C. *J. Am. Chem. Soc.* **1991**, *113*, 5135-5140.
- (4) Leon, J. M. d.; Rehr, J. J.; Zabinsky, S. I.; Albers, R. C. *Phys. Rev. B* **1991**, *44*, 4146-4156.
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