

Supplementary Material for
Structural features promoting dioxygen production by *Dechloromonas*
***aromatica* chlorite dismutase**

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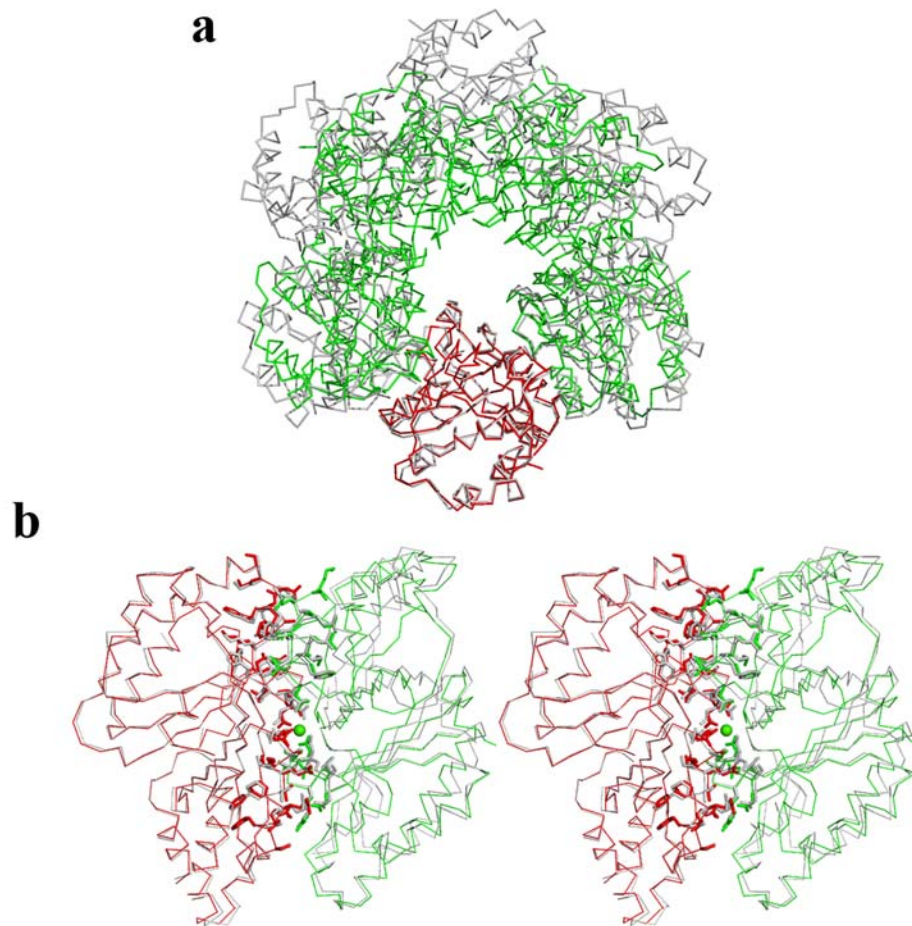


Fig. S1. Overlay of *D. aromatica* chlorite dismutase and *A. oryzae* chlorite dismutase (PDB entry: 2vdx). The *D. aromatica* pentamer is drawn as a green and red colored $\text{Ca}\alpha$ trace and the *A. oryzae* hexamer as a grey $\text{Ca}\alpha$ trace. The Figure was generated by overlaying the red *D. aromatica* monomer onto an *A. oryzae* monomer. **(a)** Superimposition of *D. aromatica* chlorite dismutase pentamer on *A. oryzae* chlorite dismutase hexamer. **(b)** Stereo representation of the chlorite dismutase monomer/monomer interface. The calcium ion found in *D. aromatica* chlorite dismutase is shown as a green sphere. This figure was generated using PyMOL (<http://www.pymol.org/>).

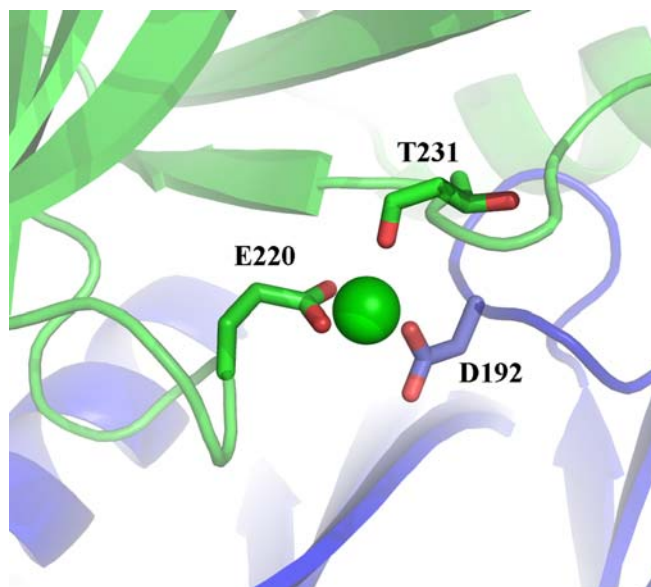


Fig. S2. Calcium binding site at the monomer/monomer interface of the chlorite dismutase pentamer. The two monomers are shown as green and blue cartoon respectively, and the calcium ion is shown as a green sphere. This figure was generated using PyMOL (<http://www.pymol.org/>).

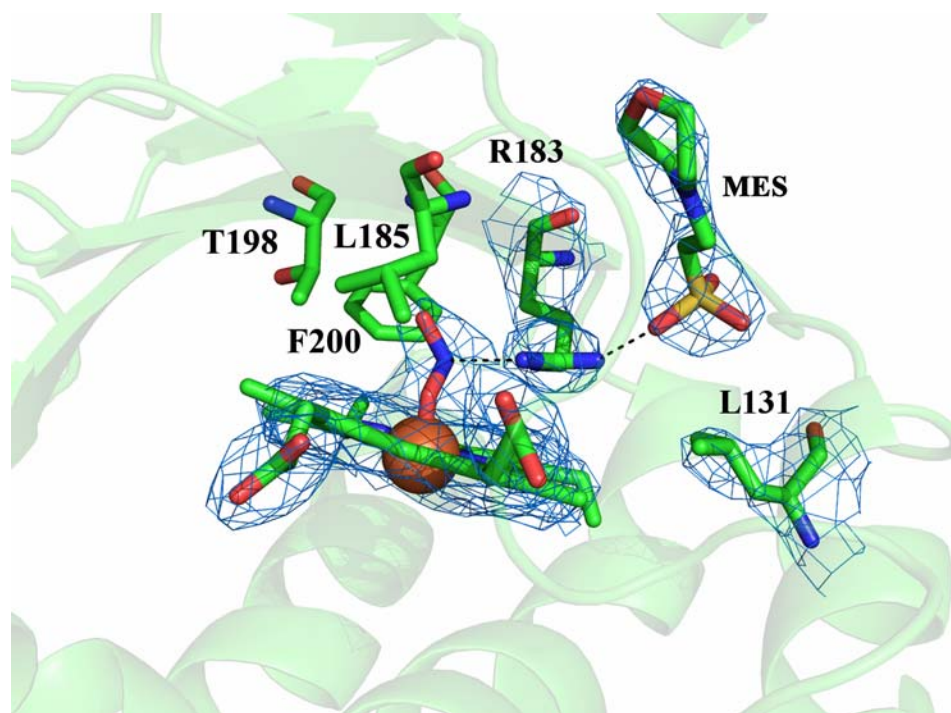


Fig. S3. Molecule of MES buffer bound close to the chlorite dismutase active site at pH=6.5. Electron density shown in blue is 20-fold NCS averaged 2Fo-Fc maps contoured at 3 σ . The heme, MES, nitrite and residues are drawn as stick colored by atom (carbon, green), with the Fe as an orange sphere. This figure was generated using PyMOL (<http://www.pymol.org/>).

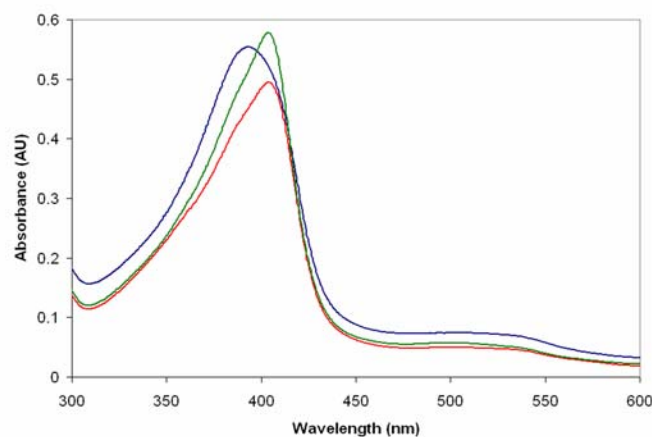


Fig. S4. UV-visible absorbance spectra of nitrite and acetate binding to chlorite dismutase. Ligand concentrations are equivalent to those in the crystallization; nitrite 47-fold excess per active site, and acetate 373-fold excess per active site. Spectra are colored as follows; chlorite dismutase alone, blue ($\lambda_{\text{max}} = 392$ nm); chlorite dismutase + nitrite, red ($\lambda_{\text{max}} = 405$ nm); chlorite dismutase + acetate, green ($\lambda_{\text{max}} = 404$ nm).

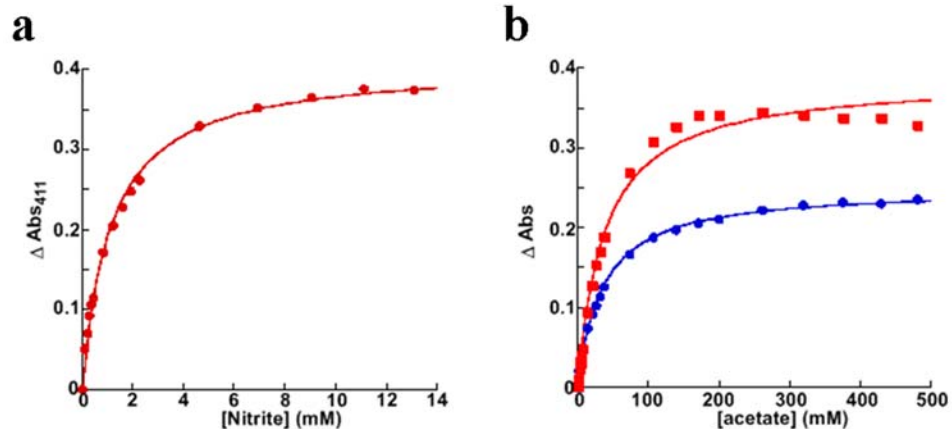


Fig. S5. Nitrite and acetate binding curves to Cld at pH 6.5. **(a)** Change of the absorbance at 411 nm as a function of nitrite added. The data were fit to the equation described in Materials and Methods, and yield a apparent K_d value of $1120 \pm 50 \mu\text{M}$. **(b)** Change in absorbance at 407 nm (red) or 381 nm (blue) as a function of acetate added. A fit of the data to the equation described in Materials and Methods yields an apparent K_d value of $38 \pm 3 \text{ mM}$ (407 nm) or $35 \pm 1 \text{ mM}$ (381 nm).