

Synthesis of Ruthenium Boryl Analogues of the Shvo Metal–Ligand Bifunctional  
Hydrogenation Catalyst

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*and*

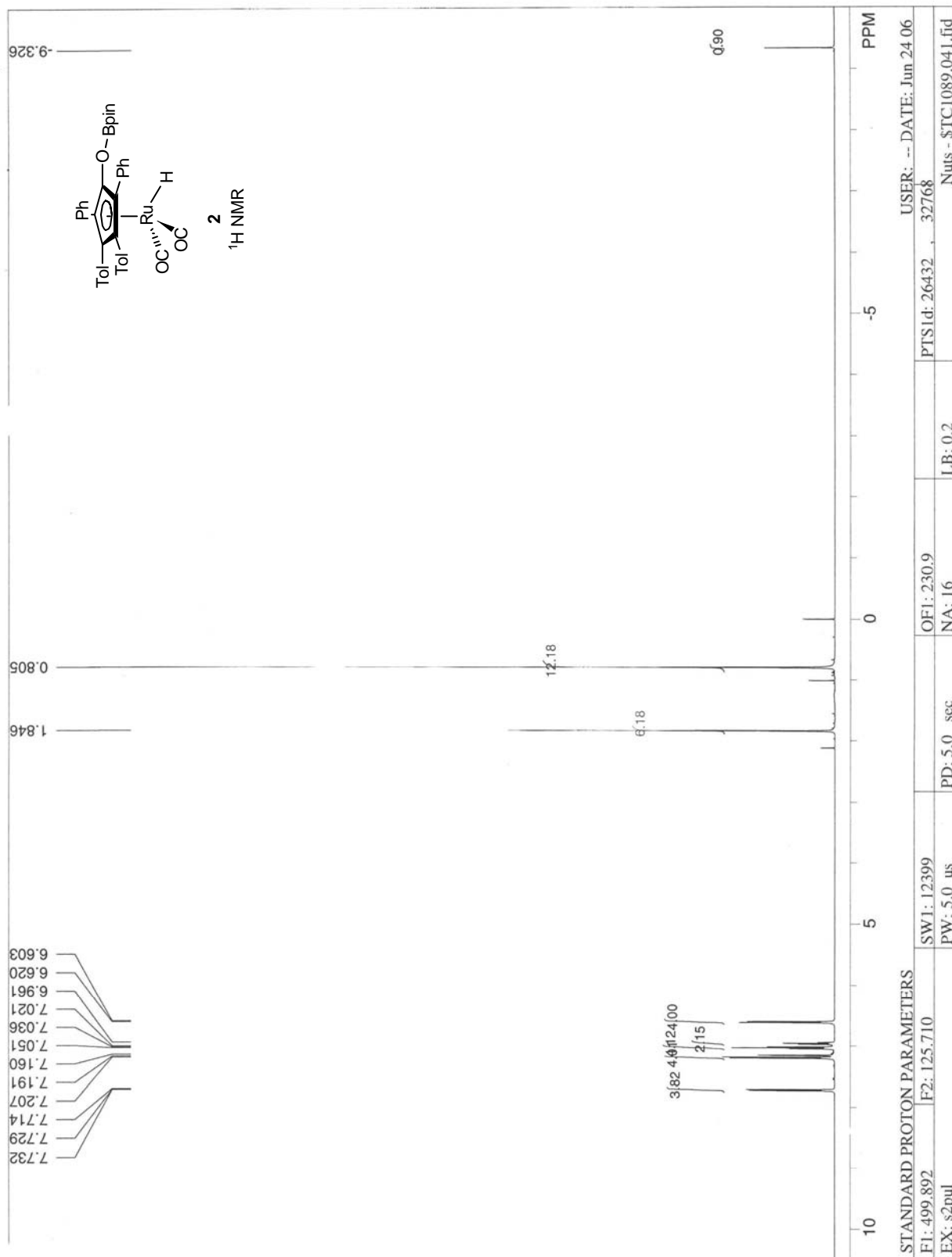
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**Supporting Information**

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I. NMR Spectra of Complexes 2, 3, 5, 7, 8

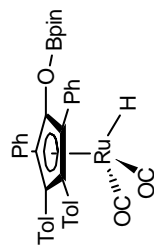


## STANDARD PROTON PARAMETERS

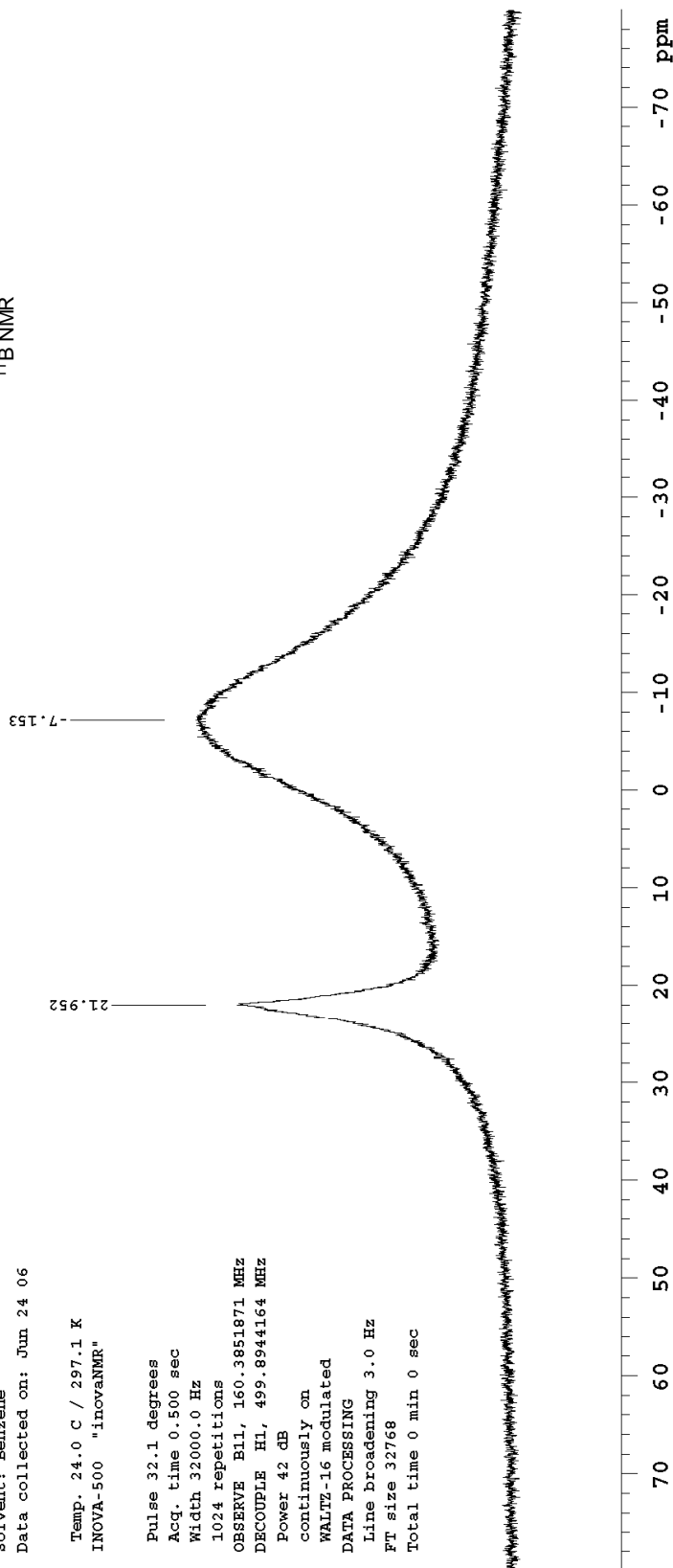
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 Data collected on: Jun 24 06

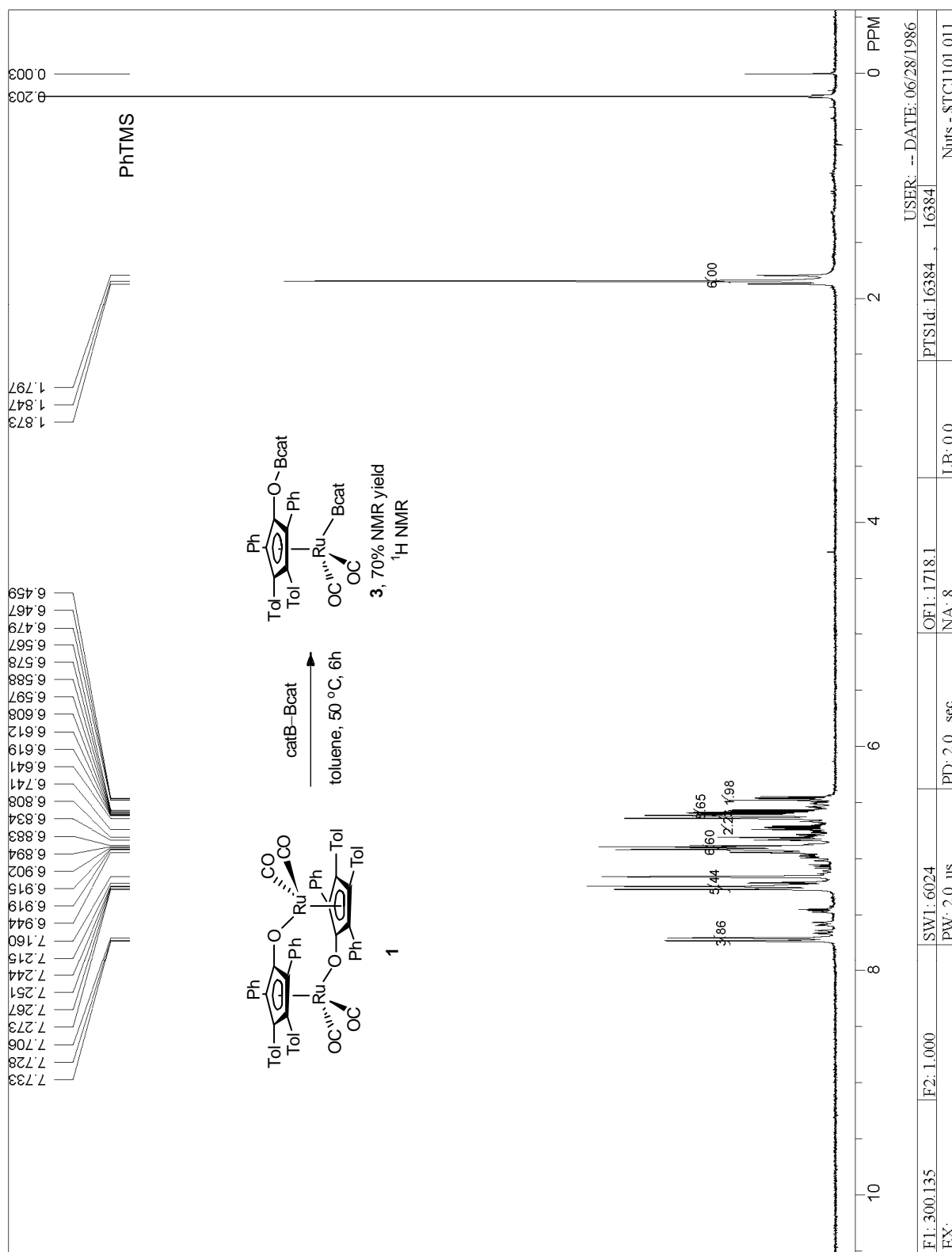
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 continuously on  
 WALTZ-16 modulated  
 DATA PROCESSING  
 Line broadening 3.0 Hz  
 FT size 32768  
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2  
<sup>11</sup>B NMR





## STANDARD PROTON PARAMETERS

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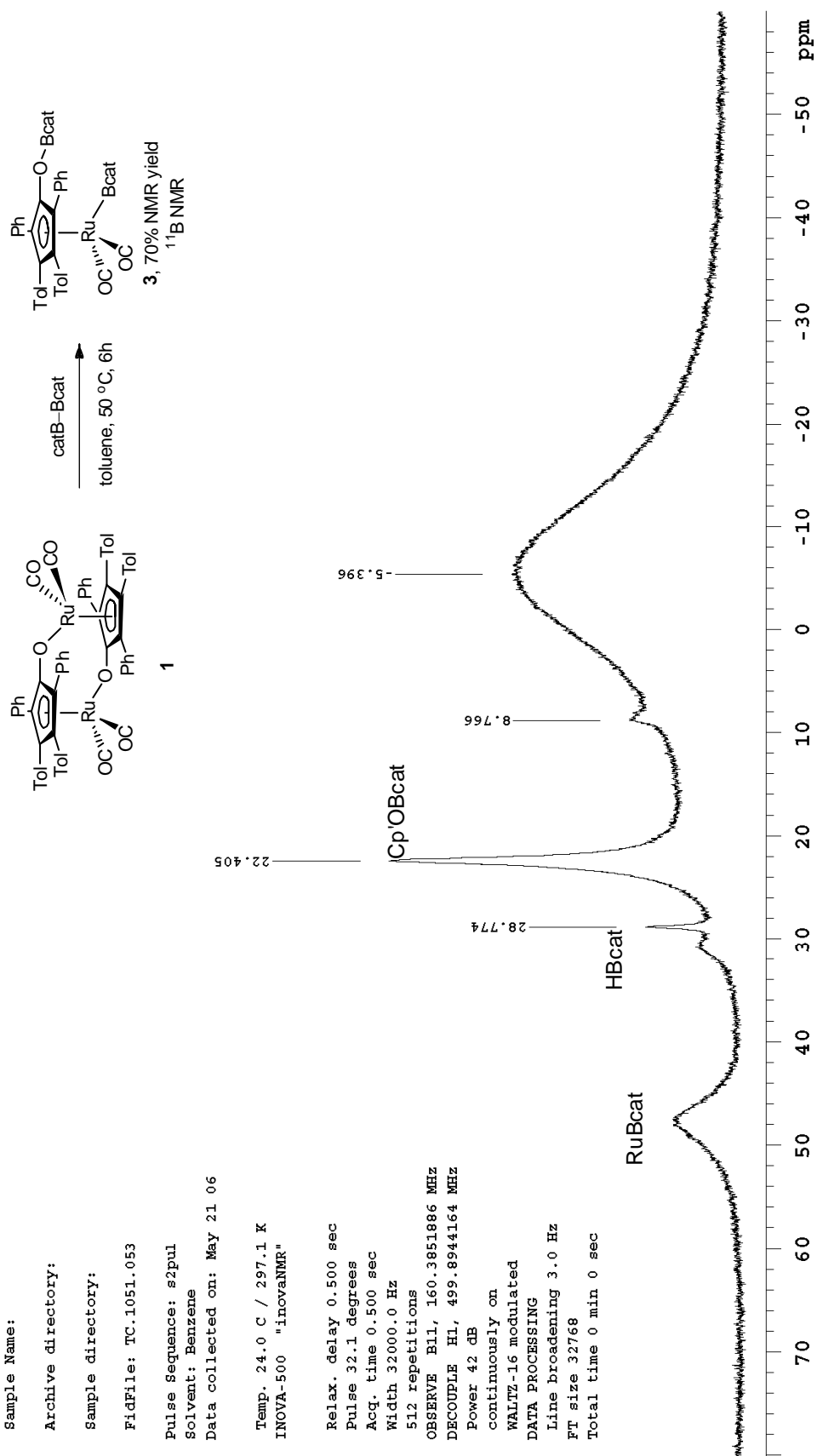
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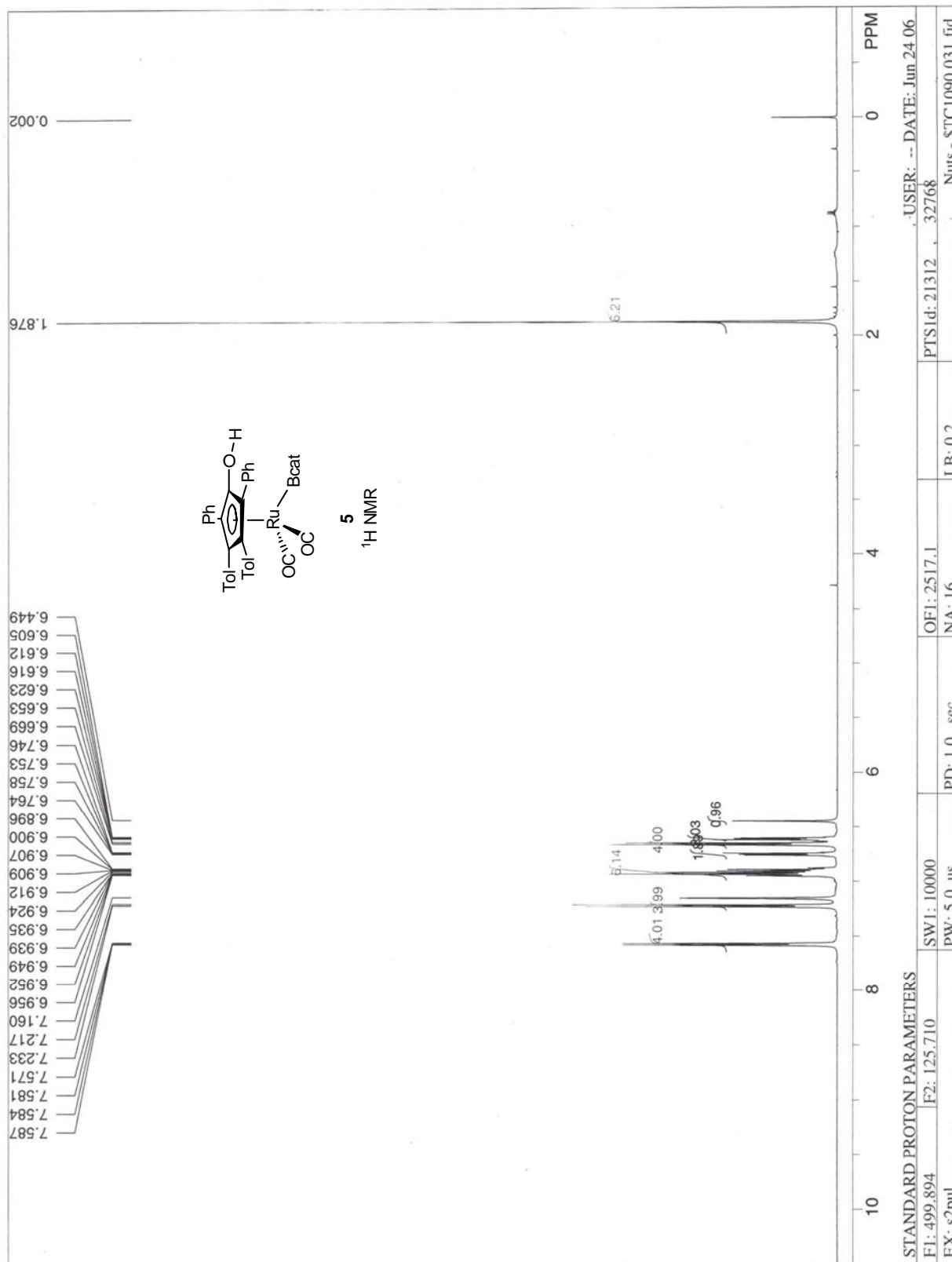
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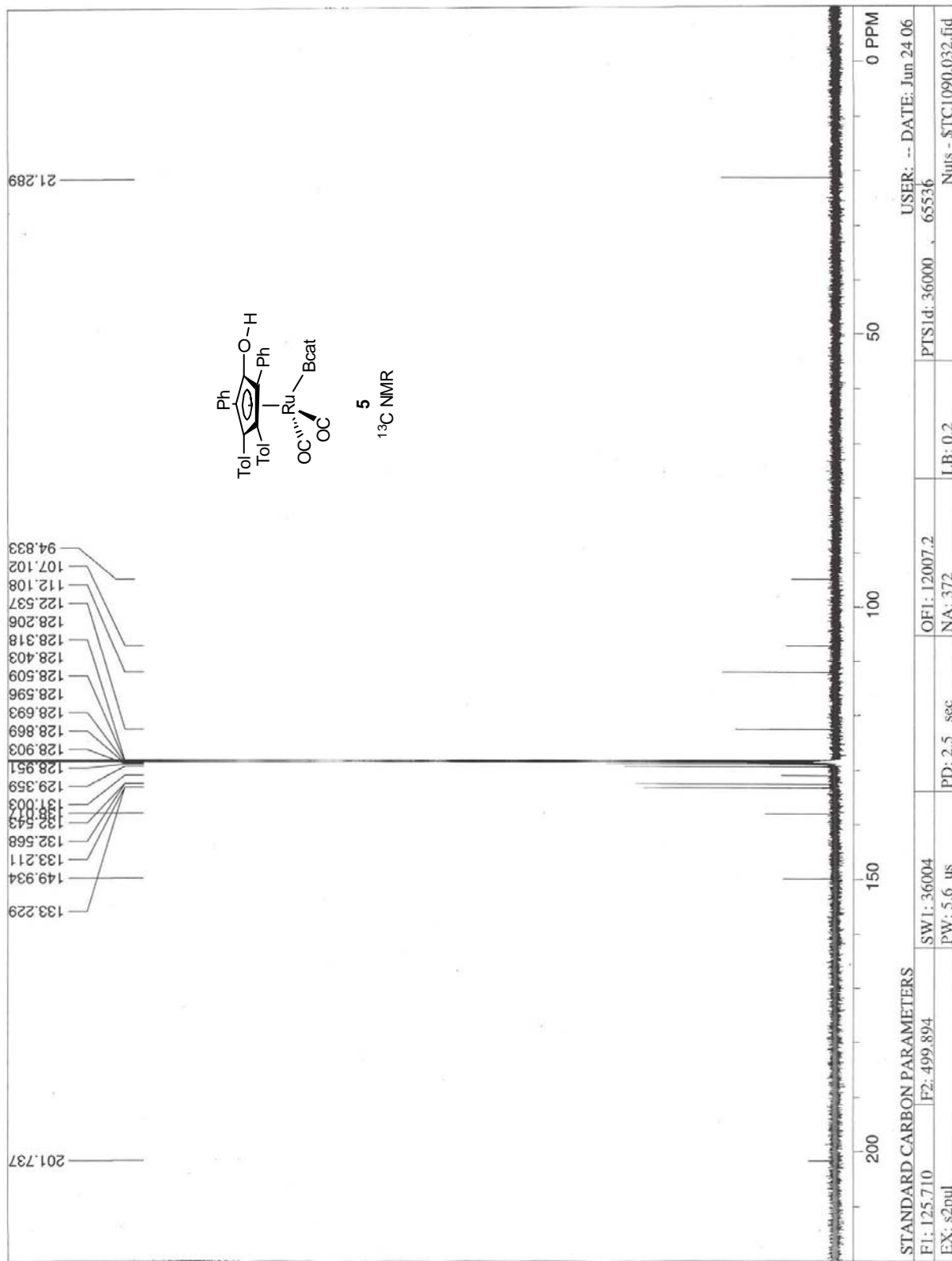
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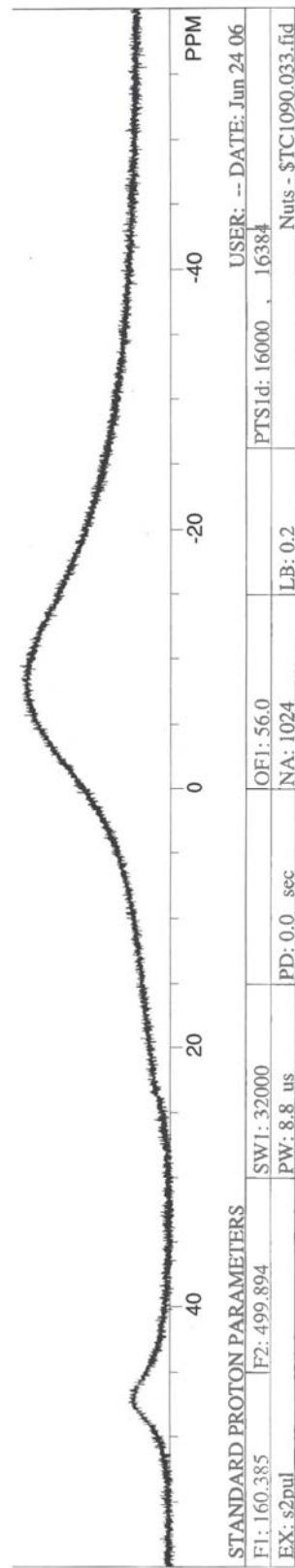
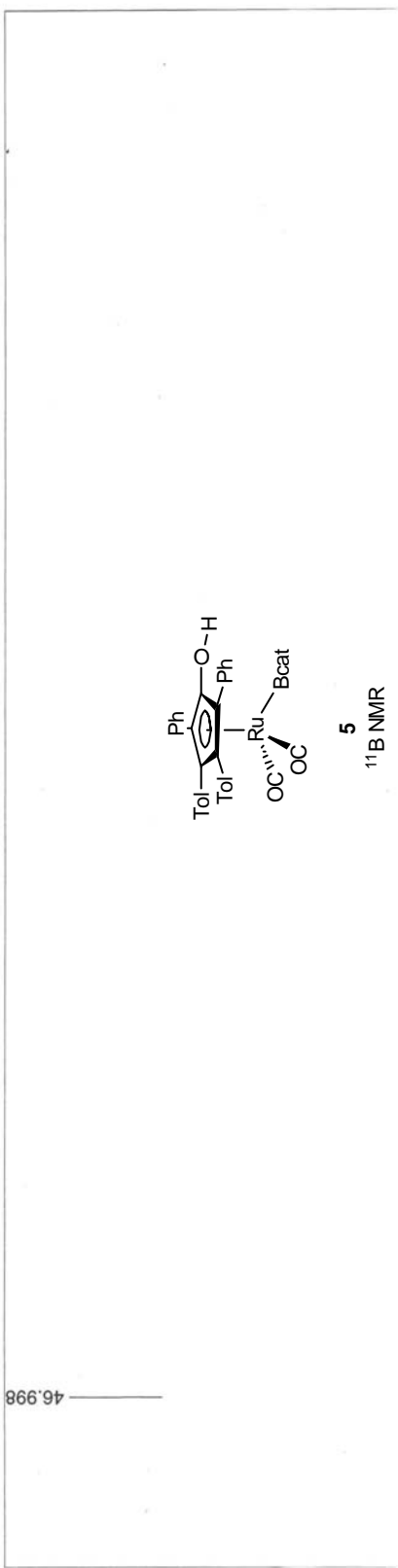
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 Total time 0 min 0 sec

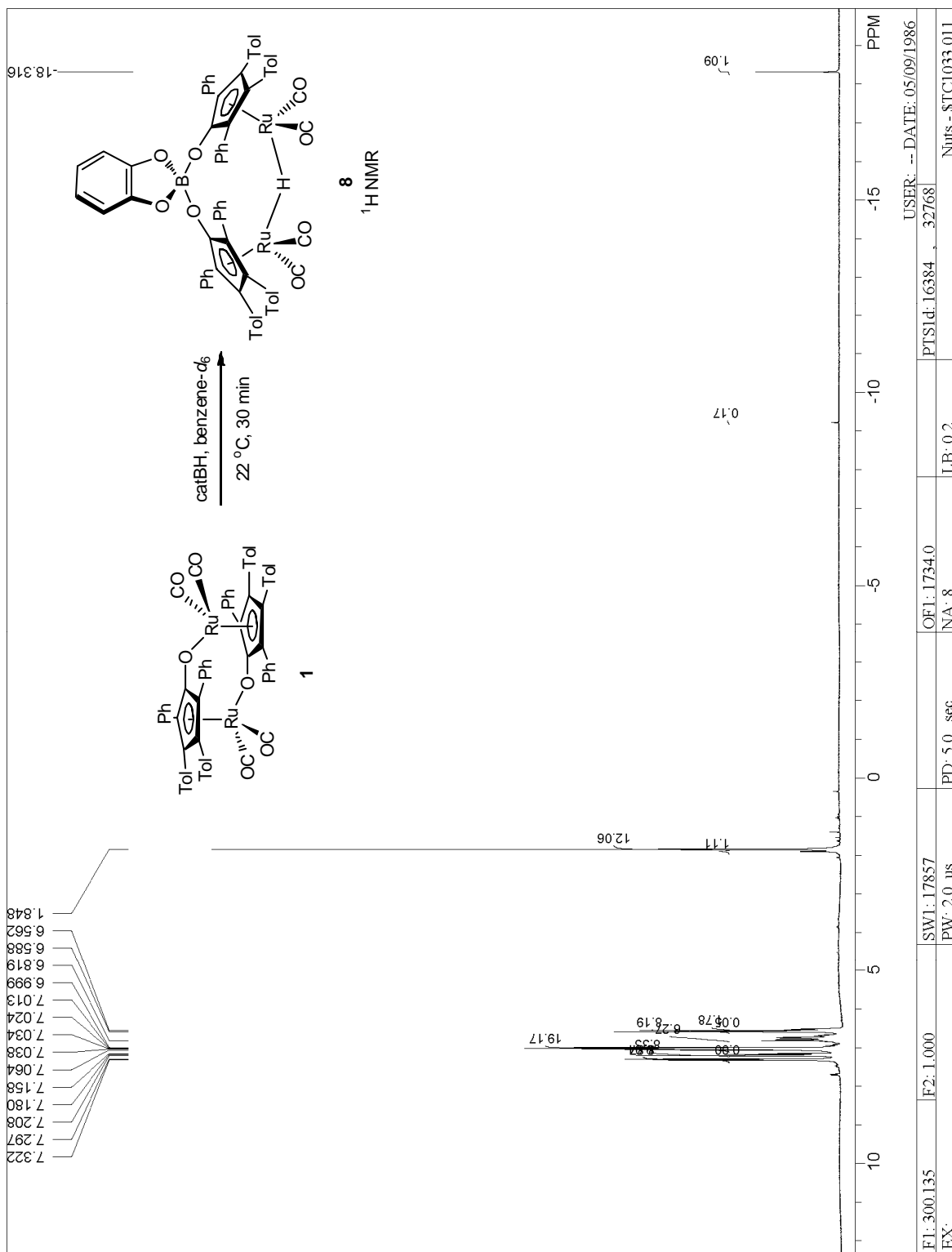


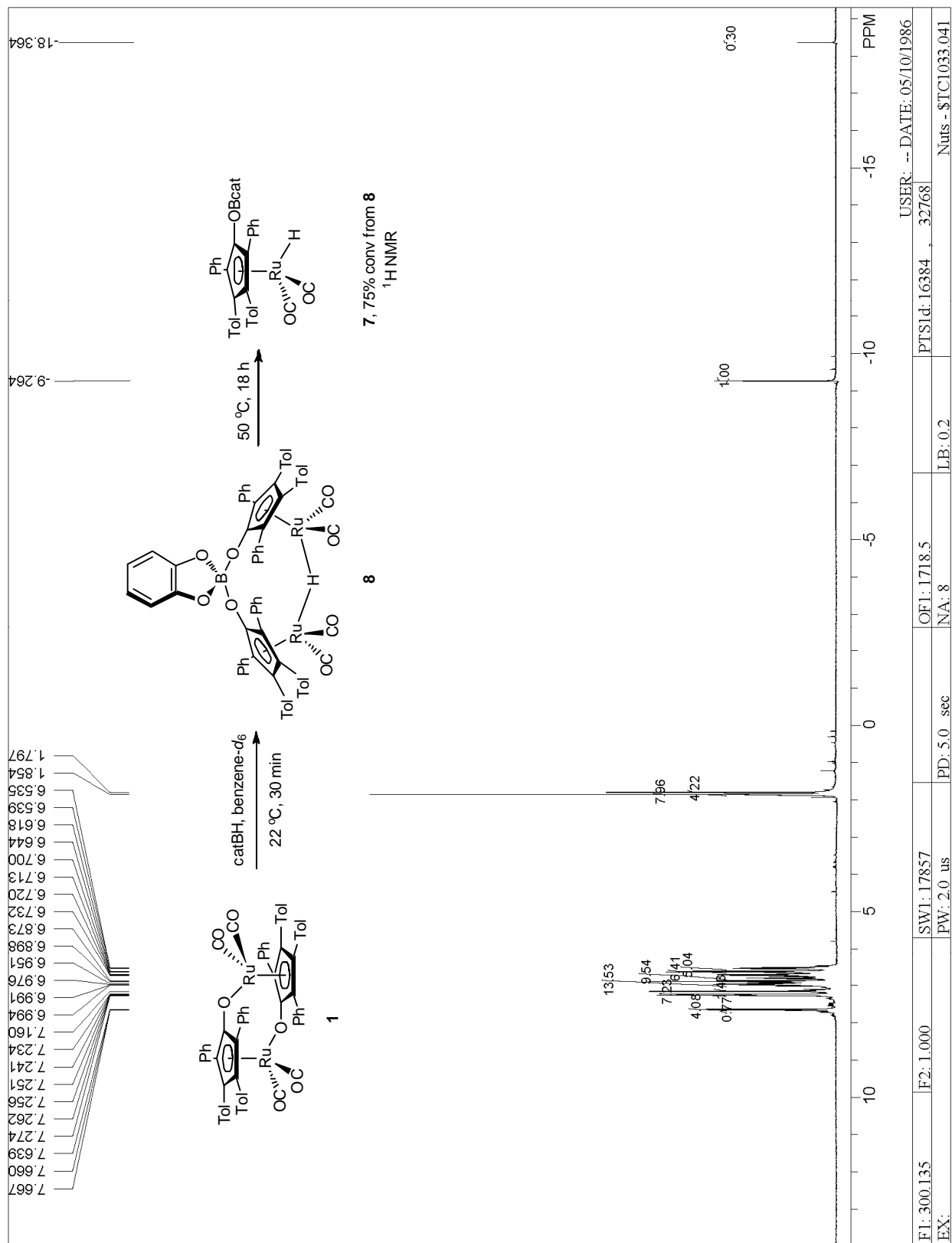








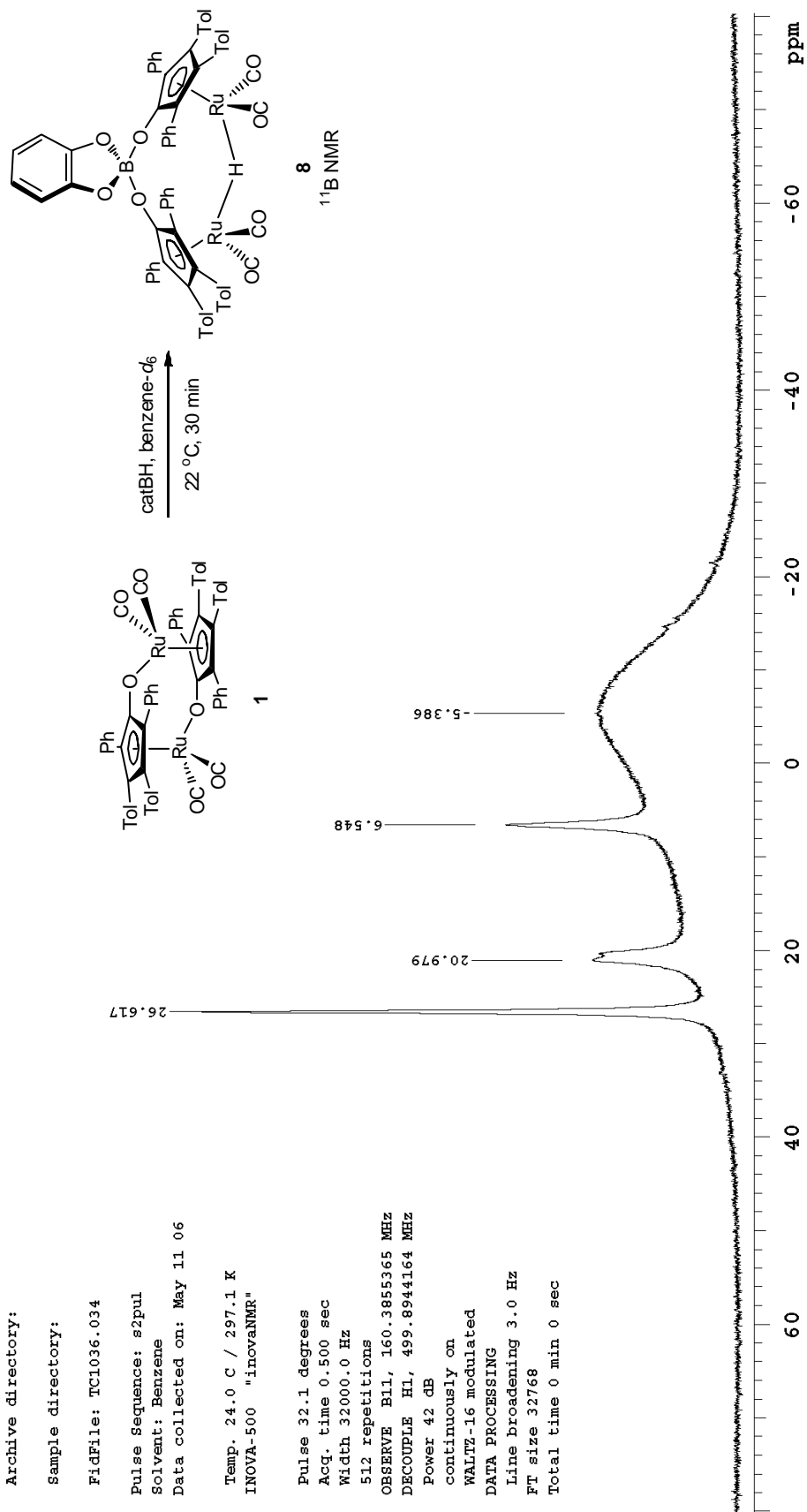




## STANDARD PROTON PARAMETERS

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 continuously on  
 WALTZ-16 modulated  
 DATA PROCESSING  
 Line broadening 3.0 Hz  
 FT size 32768  
 Total time 0 min 0 sec



## II. X-ray Crystal Data of Complex 5

### Data Collection.

A yellow crystal with approximate dimensions 0.35 x 0.20 x 0.20 mm<sup>3</sup> was selected under oil under ambient conditions and attached to the tip of a nylon loop. The crystal was mounted in a stream of cold nitrogen at 100(2) K and centered in the X-ray beam by using a video camera.

The crystal evaluation and data collection were performed on a Bruker CCD-1000 diffractometer with Mo K<sub>α</sub> ( $\lambda = 0.71073 \text{ \AA}$ ) radiation and the diffractometer to crystal distance of 4.9 cm.

The initial cell constants were obtained from three series of  $\omega$  scans at different starting angles. Each series consisted of 20 frames collected at intervals of 0.3° in a 6° range about  $\omega$  with the exposure time of 10 seconds per frame. A total of 85 reflections were obtained. The reflections were successfully indexed by an automated indexing routine built in the SMART program. The final cell constants were calculated from a set of 9582 strong reflections from the actual data collection.

The data were collected by using the hemisphere data collection routine. The reciprocal space was surveyed to the extent of a full sphere to a resolution of 0.80 Å. A total of 41042 data were harvested by collecting three sets of frames with 0.25° scans in  $\omega$  with an exposure time 30 sec per frame. These highly redundant datasets were corrected for Lorentz and polarization effects. The absorption correction was based on fitting a function to the empirical transmission surface as sampled by multiple equivalent measurements. [1]

### Structure Solution and Refinement

The systematic absences in the diffraction data were uniquely consistent for the space group *C2/c* that yielded chemically reasonable and computationally stable results of refinement [1].

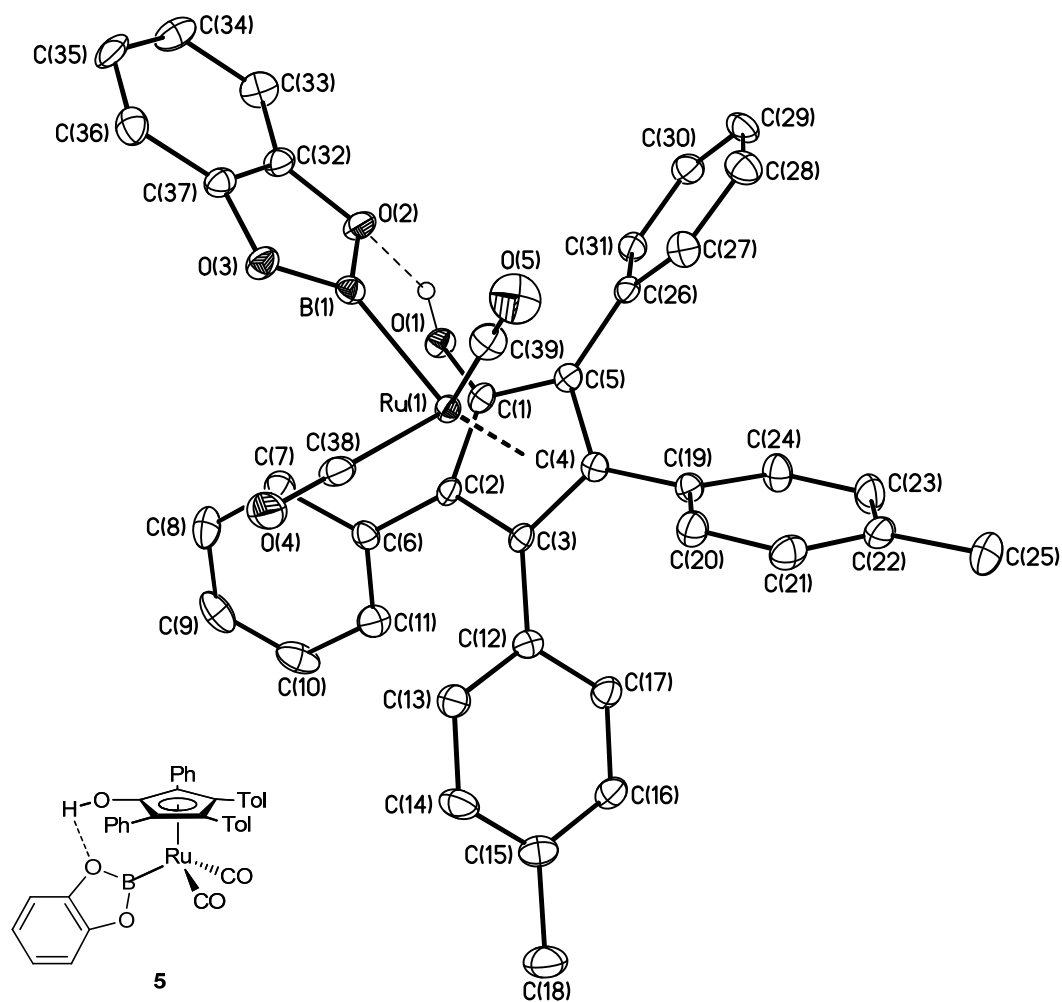
A successful solution by the direct methods provided most non-hydrogen atoms from the *E*-map. The remaining non-hydrogen atoms were located in an alternating series of least-squares cycles and difference Fourier maps. All non-hydrogen atoms were refined with anisotropic displacement coefficients. All hydrogen atoms were included in the structure factor calculation

at idealized positions and were allowed to ride on the neighboring atoms with relative isotropic displacement coefficients.

The final least-squares refinement of 418 parameters against 7745 data resulted in residuals  $R$  (based on  $F^2$  for  $I \geq 2\sigma$ ) and  $wR$  (based on  $F^2$  for all data) of 0.0476 and 0.1167, respectively. The final difference Fourier map was featureless.

#### **References**

- [1] **Bruker-AXS. (2000-2003) SADABS V.2.05, SAINT V.6.22, SHELXTL V.6.10 & SMART 5.622 Software Reference Manuals. Bruker-AXS, Madison, Wisconsin, USA.**



**Figure 2:** Molecular drawing of **5** shown with 50% probability ellipsoids. All hydrogen atoms except for the O-H are omitted for clarity. The one hydrogen bond is shown as the thin dashed line. The thick dashed line links Ru(1) to the center of the centroid.

Table 1. Crystal data and structure refinement for **5**.

Empirical formula	C <sub>39</sub> H <sub>29</sub> BO <sub>5</sub> Ru	
Formula weight	689.50	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	C2/c	
Unit cell dimensions	a = 10.2918(11) Å	α = 90°.
	b = 17.2418(19) Å	β = 98.069(2)°.
	c = 35.581(4) Å	γ = 90°.
Volume	6251.3(12) Å <sup>3</sup>	
Z	8	
Density (calculated)	1.465 Mg/m <sup>3</sup>	
Absorption coefficient	0.547 mm <sup>-1</sup>	
F(000)	2816	
Crystal size	0.35 x 0.20 x 0.20 mm <sup>3</sup>	
Theta range for data collection	1.16 to 28.32°.	
Index ranges	-13 ≤ h ≤ 13, -22 ≤ k ≤ 22, -47 ≤ l ≤ 47	
Reflections collected	41042	
Independent reflections	7745 [R(int) = 0.0719]	
Completeness to theta = 28.32°	99.3 %	
Absorption correction	Multi-scan with SADABS	
Max. and min. transmission	0.8984 and 0.8316	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	7745 / 0 / 418	
Goodness-of-fit on F <sup>2</sup>	1.030	
Final R indices [I > 2σ(I)]	R1 = 0.0476, wR2 = 0.1061	
R indices (all data)	R1 = 0.0657, wR2 = 0.1167	
Largest diff. peak and hole	0.951 and -1.102 e.Å <sup>-3</sup>	

Table 2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **5**.  $U(\text{eq})$  is defined as one third of the trace of the orthogonalized  $U^{\text{ij}}$  tensor.

	x	y	z	U(eq)
Ru(1)	3622(1)	520(1)	1174(1)	13(1)
O(1)	4195(2)	1038(1)	2077(1)	18(1)
O(2)	3517(2)	-433(1)	1903(1)	20(1)
O(3)	1993(2)	-884(1)	1424(1)	20(1)
O(4)	892(2)	383(2)	744(1)	33(1)
O(5)	4577(3)	-799(2)	727(1)	34(1)
B(1)	2996(3)	-340(2)	1515(1)	15(1)
C(1)	4354(3)	1205(2)	1713(1)	16(1)
C(2)	3449(3)	1665(2)	1467(1)	14(1)
C(3)	4053(3)	1821(2)	1128(1)	14(1)
C(4)	5270(3)	1415(2)	1164(1)	14(1)
C(5)	5457(3)	1008(2)	1525(1)	14(1)
C(6)	2217(3)	1993(2)	1571(1)	16(1)
C(7)	1334(3)	1525(2)	1734(1)	23(1)
C(8)	202(4)	1838(2)	1837(1)	29(1)
C(9)	-67(4)	2615(2)	1779(1)	32(1)
C(10)	773(4)	3080(2)	1623(1)	29(1)
C(11)	1934(3)	2780(2)	1518(1)	22(1)
C(12)	3601(3)	2416(2)	840(1)	16(1)
C(13)	2362(3)	2416(2)	627(1)	21(1)
C(14)	1941(3)	3031(2)	391(1)	25(1)
C(15)	2737(3)	3676(2)	365(1)	22(1)
C(16)	3985(3)	3675(2)	574(1)	21(1)
C(17)	4416(3)	3052(2)	806(1)	21(1)
C(18)	2298(4)	4350(2)	113(1)	32(1)
C(19)	6219(3)	1468(2)	889(1)	16(1)
C(20)	5844(3)	1290(2)	506(1)	20(1)
C(21)	6734(3)	1361(2)	249(1)	21(1)
C(22)	8015(3)	1600(2)	365(1)	21(1)
C(23)	8371(3)	1794(2)	743(1)	25(1)
C(24)	7485(3)	1732(2)	1003(1)	21(1)
C(25)	8975(4)	1629(2)	83(1)	30(1)
C(26)	6642(3)	574(2)	1692(1)	15(1)
C(27)	7325(3)	91(2)	1469(1)	20(1)
C(28)	8468(3)	-286(2)	1624(1)	23(1)
C(29)	8930(3)	-203(2)	2008(1)	23(1)
C(30)	8269(3)	278(2)	2234(1)	21(1)
C(31)	7140(3)	664(2)	2076(1)	18(1)
C(32)	2798(3)	-1025(2)	2046(1)	19(1)
C(33)	2957(4)	-1335(2)	2401(1)	24(1)
C(34)	2125(4)	-1952(2)	2458(1)	26(1)



C(35)	1195(4)	-2224(2)	2164(1)	25(1)
C(36)	1062(3)	-1906(2)	1810(1)	22(1)
C(37)	1889(3)	-1299(2)	1754(1)	18(1)
C(38)	1926(3)	430(2)	901(1)	21(1)
C(39)	4212(3)	-307(2)	895(1)	21(1)

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Table 3. Bond lengths [Å] and angles [°] for **5**.

Ru(1)-C(38)	1.881(3)	C(15)-C(16)	1.391(5)
Ru(1)-C(39)	1.886(3)	C(15)-C(18)	1.499(5)
Ru(1)-B(1)	2.075(4)	C(16)-C(17)	1.388(4)
Ru(1)-C(2)	2.250(3)	C(16)-H(16)	0.9500
Ru(1)-C(5)	2.275(3)	C(17)-H(17)	0.9500
Ru(1)-C(1)	2.286(3)	C(18)-H(18A)	0.9800
Ru(1)-C(3)	2.296(3)	C(18)-H(18B)	0.9800
Ru(1)-C(4)	2.297(3)	C(18)-H(18C)	0.9800
Ru(1)-centroid	1.925(3)	C(19)-C(24)	1.386(4)
O(1)-C(1)	1.361(3)	C(19)-C(20)	1.397(4)
O(1)-H(1)	0.8400	C(20)-C(21)	1.388(4)
O(2)-C(32)	1.397(4)	C(20)-H(20)	0.9500
O(2)-B(1)	1.419(4)	C(21)-C(22)	1.388(5)
O(3)-C(37)	1.392(4)	C(21)-H(21)	0.9500
O(3)-B(1)	1.399(4)	C(22)-C(23)	1.386(5)
O(4)-C(38)	1.134(4)	C(22)-C(25)	1.505(4)
O(5)-C(39)	1.132(4)	C(23)-C(24)	1.390(4)
C(1)-C(2)	1.425(4)	C(23)-H(23)	0.9500
C(1)-C(5)	1.435(4)	C(24)-H(24)	0.9500
C(2)-C(3)	1.457(4)	C(25)-H(25A)	0.9800
C(2)-C(6)	1.482(4)	C(25)-H(25B)	0.9800
C(3)-C(4)	1.425(4)	C(25)-H(25C)	0.9800
C(3)-C(12)	1.479(4)	C(26)-C(31)	1.400(4)
C(4)-C(5)	1.454(4)	C(26)-C(27)	1.403(4)
C(4)-C(19)	1.479(4)	C(27)-C(28)	1.389(5)
C(5)-C(26)	1.482(4)	C(27)-H(27)	0.9500
C(6)-C(11)	1.394(5)	C(28)-C(29)	1.389(5)
C(6)-C(7)	1.401(5)	C(28)-H(28)	0.9500
C(7)-C(8)	1.380(5)	C(29)-C(30)	1.396(5)
C(7)-H(7)	0.9500	C(29)-H(29)	0.9500
C(8)-C(9)	1.377(6)	C(30)-C(31)	1.387(4)
C(8)-H(8)	0.9500	C(30)-H(30)	0.9500
C(9)-C(10)	1.355(6)	C(31)-H(31)	0.9500
C(9)-H(9)	0.9500	C(32)-C(33)	1.363(4)
C(10)-C(11)	1.400(5)	C(32)-C(37)	1.380(4)
C(10)-H(10)	0.9500	C(33)-C(34)	1.398(5)
C(11)-H(11)	0.9500	C(33)-H(33)	0.9500
C(12)-C(13)	1.389(4)	C(34)-C(35)	1.396(5)
C(12)-C(17)	1.396(4)	C(34)-H(34)	0.9500
C(13)-C(14)	1.383(5)	C(35)-C(36)	1.362(5)
C(13)-H(13)	0.9500	C(35)-H(35)	0.9500
C(14)-C(15)	1.392(5)	C(36)-C(37)	1.382(5)
C(14)-H(14)	0.9500	C(36)-H(36)	0.9500

C(38)-Ru(1)-C(39)	90.55(14)	C(4)-C(3)-C(2)	107.9(3)
C(38)-Ru(1)-B(1)	84.27(14)	C(4)-C(3)-C(12)	126.1(3)
C(39)-Ru(1)-B(1)	85.21(14)	C(2)-C(3)-C(12)	125.0(3)
C(38)-Ru(1)-C(2)	100.60(13)	C(4)-C(3)-Ru(1)	71.94(17)
C(39)-Ru(1)-C(2)	163.57(13)	C(2)-C(3)-Ru(1)	69.60(17)
B(1)-Ru(1)-C(2)	107.71(12)	C(12)-C(3)-Ru(1)	132.9(2)
C(38)-Ru(1)-C(5)	162.38(13)	C(3)-C(4)-C(5)	108.7(3)
C(39)-Ru(1)-C(5)	105.38(13)	C(3)-C(4)-C(19)	124.4(3)
B(1)-Ru(1)-C(5)	104.18(12)	C(5)-C(4)-C(19)	126.7(3)
C(2)-Ru(1)-C(5)	62.26(11)	C(3)-C(4)-Ru(1)	71.90(17)
C(38)-Ru(1)-C(1)	130.39(13)	C(5)-C(4)-Ru(1)	70.64(16)
C(39)-Ru(1)-C(1)	137.64(13)	C(19)-C(4)-Ru(1)	127.5(2)
B(1)-Ru(1)-C(1)	88.36(12)	C(1)-C(5)-C(4)	106.4(3)
C(2)-Ru(1)-C(1)	36.60(10)	C(1)-C(5)-C(26)	126.4(3)
C(5)-Ru(1)-C(1)	36.68(11)	C(4)-C(5)-C(26)	126.6(3)
C(38)-Ru(1)-C(3)	102.67(12)	C(1)-C(5)-Ru(1)	72.09(17)
C(39)-Ru(1)-C(3)	128.44(13)	C(4)-C(5)-Ru(1)	72.29(16)
B(1)-Ru(1)-C(3)	144.91(12)	C(26)-C(5)-Ru(1)	127.3(2)
C(2)-Ru(1)-C(3)	37.37(10)	C(11)-C(6)-C(7)	118.7(3)
C(5)-Ru(1)-C(3)	61.58(10)	C(11)-C(6)-C(2)	120.6(3)
C(1)-Ru(1)-C(3)	60.74(10)	C(7)-C(6)-C(2)	120.7(3)
C(38)-Ru(1)-C(4)	132.99(12)	C(8)-C(7)-C(6)	120.4(3)
C(39)-Ru(1)-C(4)	101.93(13)	C(8)-C(7)-H(7)	119.8
B(1)-Ru(1)-C(4)	141.23(12)	C(6)-C(7)-H(7)	119.8
C(2)-Ru(1)-C(4)	61.64(11)	C(9)-C(8)-C(7)	120.1(3)
C(5)-Ru(1)-C(4)	37.08(10)	C(9)-C(8)-H(8)	120.0
C(1)-Ru(1)-C(4)	60.62(10)	C(7)-C(8)-H(8)	120.0
C(3)-Ru(1)-C(4)	36.15(11)	C(10)-C(9)-C(8)	120.7(3)
C(1)-O(1)-H(1)	109.5	C(10)-C(9)-H(9)	119.7
C(32)-O(2)-B(1)	106.8(2)	C(8)-C(9)-H(9)	119.7
C(37)-O(3)-B(1)	106.9(2)	C(9)-C(10)-C(11)	120.6(3)
O(3)-B(1)-O(2)	108.4(3)	C(9)-C(10)-H(10)	119.7
O(3)-B(1)-Ru(1)	128.6(2)	C(11)-C(10)-H(10)	119.7
O(2)-B(1)-Ru(1)	123.0(2)	C(6)-C(11)-C(10)	119.6(3)
O(1)-C(1)-C(2)	123.1(3)	C(6)-C(11)-H(11)	120.2
O(1)-C(1)-C(5)	127.0(3)	C(10)-C(11)-H(11)	120.2
C(2)-C(1)-C(5)	109.8(3)	C(13)-C(12)-C(17)	118.0(3)
O(1)-C(1)-Ru(1)	128.4(2)	C(13)-C(12)-C(3)	123.6(3)
C(2)-C(1)-Ru(1)	70.32(17)	C(17)-C(12)-C(3)	118.0(3)
C(5)-C(1)-Ru(1)	71.22(16)	C(14)-C(13)-C(12)	121.0(3)
C(1)-C(2)-C(3)	107.0(3)	C(14)-C(13)-H(13)	119.5
C(1)-C(2)-C(6)	124.7(3)	C(12)-C(13)-H(13)	119.5
C(3)-C(2)-C(6)	127.9(3)	C(13)-C(14)-C(15)	121.1(3)
C(1)-C(2)-Ru(1)	73.08(17)	C(13)-C(14)-H(14)	119.5
C(3)-C(2)-Ru(1)	73.03(17)	C(15)-C(14)-H(14)	119.5
C(6)-C(2)-Ru(1)	125.2(2)	C(16)-C(15)-C(14)	118.1(3)

C(16)-C(15)-C(18)	120.0(3)	C(29)-C(28)-H(28)	120.2
C(14)-C(15)-C(18)	121.9(3)	C(28)-C(29)-C(30)	120.0(3)
C(17)-C(16)-C(15)	120.9(3)	C(28)-C(29)-H(29)	120.0
C(17)-C(16)-H(16)	119.6	C(30)-C(29)-H(29)	120.0
C(15)-C(16)-H(16)	119.6	C(31)-C(30)-C(29)	120.0(3)
C(16)-C(17)-C(12)	120.9(3)	C(31)-C(30)-H(30)	120.0
C(16)-C(17)-H(17)	119.6	C(29)-C(30)-H(30)	120.0
C(12)-C(17)-H(17)	119.6	C(30)-C(31)-C(26)	120.8(3)
C(15)-C(18)-H(18A)	109.5	C(30)-C(31)-H(31)	119.6
C(15)-C(18)-H(18B)	109.5	C(26)-C(31)-H(31)	119.6
H(18A)-C(18)-H(18B)	109.5	C(33)-C(32)-C(37)	123.1(3)
C(15)-C(18)-H(18C)	109.5	C(33)-C(32)-O(2)	128.5(3)
H(18A)-C(18)-H(18C)	109.5	C(37)-C(32)-O(2)	108.4(3)
H(18B)-C(18)-H(18C)	109.5	C(32)-C(33)-C(34)	115.8(3)
C(24)-C(19)-C(20)	118.5(3)	C(32)-C(33)-H(33)	122.1
C(24)-C(19)-C(4)	120.5(3)	C(34)-C(33)-H(33)	122.1
C(20)-C(19)-C(4)	120.9(3)	C(35)-C(34)-C(33)	121.2(3)
C(21)-C(20)-C(19)	120.4(3)	C(35)-C(34)-H(34)	119.4
C(21)-C(20)-H(20)	119.8	C(33)-C(34)-H(34)	119.4
C(19)-C(20)-H(20)	119.8	C(36)-C(35)-C(34)	121.8(3)
C(22)-C(21)-C(20)	121.2(3)	C(36)-C(35)-H(35)	119.1
C(22)-C(21)-H(21)	119.4	C(34)-C(35)-H(35)	119.1
C(20)-C(21)-H(21)	119.4	C(35)-C(36)-C(37)	117.1(3)
C(23)-C(22)-C(21)	118.0(3)	C(35)-C(36)-H(36)	121.4
C(23)-C(22)-C(25)	122.1(3)	C(37)-C(36)-H(36)	121.4
C(21)-C(22)-C(25)	119.8(3)	C(32)-C(37)-C(36)	121.0(3)
C(22)-C(23)-C(24)	121.3(3)	C(32)-C(37)-O(3)	109.5(3)
C(22)-C(23)-H(23)	119.4	C(36)-C(37)-O(3)	129.4(3)
C(24)-C(23)-H(23)	119.4	O(4)-C(38)-Ru(1)	178.4(3)
C(19)-C(24)-C(23)	120.6(3)	O(5)-C(39)-Ru(1)	179.3(3)
C(19)-C(24)-H(24)	119.7		
C(23)-C(24)-H(24)	119.7		
C(22)-C(25)-H(25A)	109.5		
C(22)-C(25)-H(25B)	109.5		
H(25A)-C(25)-H(25B)	109.5		
C(22)-C(25)-H(25C)	109.5		
H(25A)-C(25)-H(25C)	109.5		
H(25B)-C(25)-H(25C)	109.5		
C(31)-C(26)-C(27)	118.2(3)		
C(31)-C(26)-C(5)	120.2(3)		
C(27)-C(26)-C(5)	121.6(3)		
C(28)-C(27)-C(26)	121.2(3)		
C(28)-C(27)-H(27)	119.4		
C(26)-C(27)-H(27)	119.4		
C(27)-C(28)-C(29)	119.7(3)		
C(27)-C(28)-H(28)	120.2		

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Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **5**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [ h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12} ]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Ru(1)	12(1)	15(1)	12(1)	1(1)	1(1)	-2(1)
O(1)	21(1)	20(1)	12(1)	1(1)	2(1)	-1(1)
O(2)	23(1)	16(1)	19(1)	4(1)	2(1)	-3(1)
O(3)	22(1)	21(1)	16(1)	4(1)	1(1)	-4(1)
O(4)	24(1)	29(2)	41(2)	9(1)	-13(1)	-8(1)
O(5)	43(2)	33(2)	27(1)	-12(1)	9(1)	4(1)
B(1)	15(2)	15(2)	16(2)	0(1)	3(1)	2(1)
C(1)	17(2)	15(2)	14(1)	-2(1)	2(1)	-4(1)
C(2)	14(1)	16(2)	13(1)	2(1)	1(1)	-5(1)
C(3)	14(1)	13(2)	14(1)	-1(1)	1(1)	-4(1)
C(4)	14(1)	15(2)	13(1)	-2(1)	1(1)	-2(1)
C(5)	15(1)	15(2)	12(1)	-1(1)	-1(1)	-3(1)
C(6)	14(2)	20(2)	14(1)	0(1)	0(1)	1(1)
C(7)	21(2)	20(2)	30(2)	0(1)	8(1)	1(1)
C(8)	21(2)	35(2)	35(2)	-1(2)	13(2)	-5(2)
C(9)	19(2)	41(2)	35(2)	-7(2)	4(2)	12(2)
C(10)	33(2)	27(2)	25(2)	-3(1)	3(1)	16(2)
C(11)	25(2)	22(2)	22(2)	0(1)	5(1)	1(1)
C(12)	19(2)	15(2)	15(1)	0(1)	3(1)	1(1)
C(13)	19(2)	17(2)	26(2)	0(1)	-2(1)	-4(1)
C(14)	24(2)	23(2)	26(2)	4(1)	-8(1)	1(1)
C(15)	29(2)	20(2)	17(2)	1(1)	2(1)	2(1)
C(16)	23(2)	19(2)	21(2)	2(1)	4(1)	-4(1)
C(17)	19(2)	25(2)	17(2)	4(1)	0(1)	-5(1)
C(18)	35(2)	23(2)	35(2)	9(2)	-4(2)	-1(2)
C(19)	15(2)	16(2)	16(1)	0(1)	4(1)	1(1)
C(20)	19(2)	24(2)	18(2)	0(1)	3(1)	-2(1)
C(21)	26(2)	24(2)	15(1)	2(1)	5(1)	0(1)
C(22)	23(2)	21(2)	23(2)	6(1)	10(1)	4(1)
C(23)	16(2)	34(2)	25(2)	3(1)	2(1)	-3(1)
C(24)	16(2)	29(2)	17(1)	-3(1)	1(1)	-4(1)
C(25)	26(2)	38(2)	29(2)	8(2)	13(2)	0(2)
C(26)	14(1)	14(2)	17(1)	2(1)	1(1)	-3(1)
C(27)	19(2)	22(2)	17(1)	-4(1)	0(1)	-2(1)
C(28)	23(2)	18(2)	27(2)	-4(1)	1(1)	2(1)
C(29)	14(2)	19(2)	34(2)	1(1)	-5(1)	2(1)
C(30)	20(2)	22(2)	20(2)	4(1)	-3(1)	-2(1)
C(31)	17(2)	20(2)	18(1)	2(1)	2(1)	-1(1)
C(32)	21(2)	13(2)	22(2)	1(1)	6(1)	1(1)
C(33)	29(2)	26(2)	17(2)	-2(1)	2(1)	0(2)
C(34)	40(2)	20(2)	21(2)	4(1)	11(1)	-2(2)

C(35)	29(2)	12(2)	37(2)	2(1)	14(2)	-6(1)
C(36)	20(2)	21(2)	27(2)	-6(1)	3(1)	-2(1)
C(37)	20(2)	16(2)	17(1)	2(1)	6(1)	1(1)
C(38)	24(2)	16(2)	21(2)	4(1)	-2(1)	-4(1)
C(39)	24(2)	24(2)	15(1)	-1(1)	1(1)	-1(1)

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Table 5. Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **5**.

	x	y	z	U(eq)
H(1)	4227	556	2110	26
H(7)	1515	988	1773	28
H(8)	-392	1518	1948	35
H(9)	-849	2827	1850	38
H(10)	572	3614	1584	34
H(11)	2526	3110	1412	27
H(13)	1794	1986	643	25
H(14)	1095	3014	245	30
H(16)	4551	4105	558	25
H(17)	5277	3059	943	25
H(18A)	2245	4192	-153	48
H(18B)	2928	4776	164	48
H(18C)	1432	4524	163	48
H(20)	4974	1120	421	24
H(21)	6461	1245	-11	26
H(23)	9237	1972	827	30
H(24)	7750	1872	1260	25
H(25A)	9514	2098	126	45
H(25B)	8494	1638	-175	45
H(25C)	9541	1170	114	45
H(27)	6998	21	1208	24
H(28)	8933	-599	1468	28
H(29)	9695	-474	2117	28
H(30)	8594	342	2495	25
H(31)	6700	994	2231	22
H(33)	3593	-1143	2599	29
H(34)	2193	-2190	2701	31
H(35)	641	-2643	2212	30
H(36)	427	-2094	1611	27

Table 6. Torsion angles [°] for **5**.

C(37)-O(3)-B(1)-O(2)	0.7(3)
C(37)-O(3)-B(1)-Ru(1)	-176.0(2)
C(32)-O(2)-B(1)-O(3)	-1.3(3)
C(32)-O(2)-B(1)-Ru(1)	175.7(2)
C(38)-Ru(1)-B(1)-O(3)	23.4(3)
C(39)-Ru(1)-B(1)-O(3)	-67.7(3)
C(2)-Ru(1)-B(1)-O(3)	122.7(3)
C(5)-Ru(1)-B(1)-O(3)	-172.4(3)
C(1)-Ru(1)-B(1)-O(3)	154.2(3)
C(3)-Ru(1)-B(1)-O(3)	127.3(3)
C(4)-Ru(1)-B(1)-O(3)	-170.8(2)
C(38)-Ru(1)-B(1)-O(2)	-152.9(3)
C(39)-Ru(1)-B(1)-O(2)	116.0(3)
C(2)-Ru(1)-B(1)-O(2)	-53.6(3)
C(5)-Ru(1)-B(1)-O(2)	11.3(3)
C(1)-Ru(1)-B(1)-O(2)	-22.1(3)
C(3)-Ru(1)-B(1)-O(2)	-49.0(4)
C(4)-Ru(1)-B(1)-O(2)	12.9(4)
C(38)-Ru(1)-C(1)-O(1)	75.0(3)
C(39)-Ru(1)-C(1)-O(1)	-87.2(3)
B(1)-Ru(1)-C(1)-O(1)	-6.1(3)
C(2)-Ru(1)-C(1)-O(1)	117.2(3)
C(5)-Ru(1)-C(1)-O(1)	-122.8(4)
C(3)-Ru(1)-C(1)-O(1)	156.5(3)
C(4)-Ru(1)-C(1)-O(1)	-161.8(3)
C(38)-Ru(1)-C(1)-C(2)	-42.2(2)
C(39)-Ru(1)-C(1)-C(2)	155.6(2)
B(1)-Ru(1)-C(1)-C(2)	-123.29(19)
C(5)-Ru(1)-C(1)-C(2)	120.1(3)
C(3)-Ru(1)-C(1)-C(2)	39.33(17)
C(4)-Ru(1)-C(1)-C(2)	81.03(19)
C(38)-Ru(1)-C(1)-C(5)	-162.29(18)
C(39)-Ru(1)-C(1)-C(5)	35.6(3)
B(1)-Ru(1)-C(1)-C(5)	116.65(19)
C(2)-Ru(1)-C(1)-C(5)	-120.1(3)
C(3)-Ru(1)-C(1)-C(5)	-80.73(19)
C(4)-Ru(1)-C(1)-C(5)	-39.03(17)
O(1)-C(1)-C(2)-C(3)	170.7(3)
C(5)-C(1)-C(2)-C(3)	-5.1(3)
Ru(1)-C(1)-C(2)-C(3)	-65.7(2)
O(1)-C(1)-C(2)-C(6)	-2.2(5)
C(5)-C(1)-C(2)-C(6)	-178.0(3)
Ru(1)-C(1)-C(2)-C(6)	121.5(3)
O(1)-C(1)-C(2)-Ru(1)	-123.7(3)



C(5)-C(1)-C(2)-Ru(1)	60.5(2)
C(38)-Ru(1)-C(2)-C(1)	148.61(19)
C(39)-Ru(1)-C(2)-C(1)	-79.4(5)
B(1)-Ru(1)-C(2)-C(1)	61.3(2)
C(5)-Ru(1)-C(2)-C(1)	-35.75(17)
C(3)-Ru(1)-C(2)-C(1)	-114.4(2)
C(4)-Ru(1)-C(2)-C(1)	-77.97(18)
C(38)-Ru(1)-C(2)-C(3)	-97.03(18)
C(39)-Ru(1)-C(2)-C(3)	35.0(5)
B(1)-Ru(1)-C(2)-C(3)	175.65(17)
C(5)-Ru(1)-C(2)-C(3)	78.61(18)
C(1)-Ru(1)-C(2)-C(3)	114.4(2)
C(4)-Ru(1)-C(2)-C(3)	36.39(16)
C(38)-Ru(1)-C(2)-C(6)	27.8(3)
C(39)-Ru(1)-C(2)-C(6)	159.8(4)
B(1)-Ru(1)-C(2)-C(6)	-59.6(3)
C(5)-Ru(1)-C(2)-C(6)	-156.6(3)
C(1)-Ru(1)-C(2)-C(6)	-120.8(3)
C(3)-Ru(1)-C(2)-C(6)	124.8(3)
C(4)-Ru(1)-C(2)-C(6)	161.2(3)
C(1)-C(2)-C(3)-C(4)	3.6(3)
C(6)-C(2)-C(3)-C(4)	176.1(3)
Ru(1)-C(2)-C(3)-C(4)	-62.1(2)
C(1)-C(2)-C(3)-C(12)	-165.5(3)
C(6)-C(2)-C(3)-C(12)	7.0(5)
Ru(1)-C(2)-C(3)-C(12)	128.8(3)
C(1)-C(2)-C(3)-Ru(1)	65.7(2)
C(6)-C(2)-C(3)-Ru(1)	-121.8(3)
C(38)-Ru(1)-C(3)-C(4)	-151.30(18)
C(39)-Ru(1)-C(3)-C(4)	-50.3(2)
B(1)-Ru(1)-C(3)-C(4)	110.5(2)
C(2)-Ru(1)-C(3)-C(4)	117.8(2)
C(5)-Ru(1)-C(3)-C(4)	37.15(16)
C(1)-Ru(1)-C(3)-C(4)	79.25(18)
C(38)-Ru(1)-C(3)-C(2)	90.94(19)
C(39)-Ru(1)-C(3)-C(2)	-168.05(18)
B(1)-Ru(1)-C(3)-C(2)	-7.2(3)
C(5)-Ru(1)-C(3)-C(2)	-80.61(18)
C(1)-Ru(1)-C(3)-C(2)	-38.51(16)
C(4)-Ru(1)-C(3)-C(2)	-117.8(2)
C(38)-Ru(1)-C(3)-C(12)	-28.4(3)
C(39)-Ru(1)-C(3)-C(12)	72.6(3)
B(1)-Ru(1)-C(3)-C(12)	-126.6(3)
C(2)-Ru(1)-C(3)-C(12)	-119.3(4)
C(5)-Ru(1)-C(3)-C(12)	160.1(3)
C(1)-Ru(1)-C(3)-C(12)	-157.8(3)

C(4)-Ru(1)-C(3)-C(12)	122.9(4)
C(2)-C(3)-C(4)-C(5)	-0.7(3)
C(12)-C(3)-C(4)-C(5)	168.2(3)
Ru(1)-C(3)-C(4)-C(5)	-61.4(2)
C(2)-C(3)-C(4)-C(19)	-175.8(3)
C(12)-C(3)-C(4)-C(19)	-6.9(5)
Ru(1)-C(3)-C(4)-C(19)	123.6(3)
C(2)-C(3)-C(4)-Ru(1)	60.6(2)
C(12)-C(3)-C(4)-Ru(1)	-130.5(3)
C(38)-Ru(1)-C(4)-C(3)	39.8(2)
C(39)-Ru(1)-C(4)-C(3)	141.99(18)
B(1)-Ru(1)-C(4)-C(3)	-120.7(2)
C(2)-Ru(1)-C(4)-C(3)	-37.61(16)
C(5)-Ru(1)-C(4)-C(3)	-118.2(2)
C(1)-Ru(1)-C(4)-C(3)	-79.62(18)
C(38)-Ru(1)-C(4)-C(5)	158.1(2)
C(39)-Ru(1)-C(4)-C(5)	-99.78(19)
B(1)-Ru(1)-C(4)-C(5)	-2.5(3)
C(2)-Ru(1)-C(4)-C(5)	80.62(18)
C(1)-Ru(1)-C(4)-C(5)	38.61(17)
C(3)-Ru(1)-C(4)-C(5)	118.2(2)
C(38)-Ru(1)-C(4)-C(19)	-80.0(3)
C(39)-Ru(1)-C(4)-C(19)	22.1(3)
B(1)-Ru(1)-C(4)-C(19)	119.4(3)
C(2)-Ru(1)-C(4)-C(19)	-157.5(3)
C(5)-Ru(1)-C(4)-C(19)	121.9(3)
C(1)-Ru(1)-C(4)-C(19)	160.5(3)
C(3)-Ru(1)-C(4)-C(19)	-119.9(3)
O(1)-C(1)-C(5)-C(4)	-170.9(3)
C(2)-C(1)-C(5)-C(4)	4.7(3)
Ru(1)-C(1)-C(5)-C(4)	64.6(2)
O(1)-C(1)-C(5)-C(26)	0.8(5)
C(2)-C(1)-C(5)-C(26)	176.4(3)
Ru(1)-C(1)-C(5)-C(26)	-123.6(3)
O(1)-C(1)-C(5)-Ru(1)	124.4(3)
C(2)-C(1)-C(5)-Ru(1)	-60.0(2)
C(3)-C(4)-C(5)-C(1)	-2.3(3)
C(19)-C(4)-C(5)-C(1)	172.5(3)
Ru(1)-C(4)-C(5)-C(1)	-64.5(2)
C(3)-C(4)-C(5)-C(26)	-174.1(3)
C(19)-C(4)-C(5)-C(26)	0.8(5)
Ru(1)-C(4)-C(5)-C(26)	123.7(3)
C(3)-C(4)-C(5)-Ru(1)	62.2(2)
C(19)-C(4)-C(5)-Ru(1)	-122.9(3)
C(38)-Ru(1)-C(5)-C(1)	50.0(5)
C(39)-Ru(1)-C(5)-C(1)	-156.01(19)

B(1)-Ru(1)-C(5)-C(1)	-67.1(2)
C(2)-Ru(1)-C(5)-C(1)	35.66(17)
C(3)-Ru(1)-C(5)-C(1)	78.24(19)
C(4)-Ru(1)-C(5)-C(1)	114.5(3)
C(38)-Ru(1)-C(5)-C(4)	-64.5(5)
C(39)-Ru(1)-C(5)-C(4)	89.53(19)
B(1)-Ru(1)-C(5)-C(4)	178.39(18)
C(2)-Ru(1)-C(5)-C(4)	-78.80(18)
C(1)-Ru(1)-C(5)-C(4)	-114.5(3)
C(3)-Ru(1)-C(5)-C(4)	-36.23(17)
C(38)-Ru(1)-C(5)-C(26)	172.5(4)
C(39)-Ru(1)-C(5)-C(26)	-33.5(3)
B(1)-Ru(1)-C(5)-C(26)	55.4(3)
C(2)-Ru(1)-C(5)-C(26)	158.2(3)
C(1)-Ru(1)-C(5)-C(26)	122.6(3)
C(3)-Ru(1)-C(5)-C(26)	-159.2(3)
C(4)-Ru(1)-C(5)-C(26)	-123.0(3)
C(1)-C(2)-C(6)-C(11)	128.5(3)
C(3)-C(2)-C(6)-C(11)	-42.8(5)
Ru(1)-C(2)-C(6)-C(11)	-138.4(3)
C(1)-C(2)-C(6)-C(7)	-49.6(4)
C(3)-C(2)-C(6)-C(7)	139.0(3)
Ru(1)-C(2)-C(6)-C(7)	43.5(4)
C(11)-C(6)-C(7)-C(8)	0.5(5)
C(2)-C(6)-C(7)-C(8)	178.6(3)
C(6)-C(7)-C(8)-C(9)	0.2(5)
C(7)-C(8)-C(9)-C(10)	-0.3(6)
C(8)-C(9)-C(10)-C(11)	-0.3(6)
C(7)-C(6)-C(11)-C(10)	-1.0(5)
C(2)-C(6)-C(11)-C(10)	-179.2(3)
C(9)-C(10)-C(11)-C(6)	1.0(5)
C(4)-C(3)-C(12)-C(13)	132.0(3)
C(2)-C(3)-C(12)-C(13)	-60.9(4)
Ru(1)-C(3)-C(12)-C(13)	33.1(5)
C(4)-C(3)-C(12)-C(17)	-54.4(4)
C(2)-C(3)-C(12)-C(17)	112.7(3)
Ru(1)-C(3)-C(12)-C(17)	-153.3(2)
C(17)-C(12)-C(13)-C(14)	-0.8(5)
C(3)-C(12)-C(13)-C(14)	172.8(3)
C(12)-C(13)-C(14)-C(15)	-1.1(5)
C(13)-C(14)-C(15)-C(16)	1.9(5)
C(13)-C(14)-C(15)-C(18)	-179.8(3)
C(14)-C(15)-C(16)-C(17)	-1.0(5)
C(18)-C(15)-C(16)-C(17)	-179.2(3)
C(15)-C(16)-C(17)-C(12)	-0.9(5)
C(13)-C(12)-C(17)-C(16)	1.7(5)

C(3)-C(12)-C(17)-C(16)	-172.2(3)
C(3)-C(4)-C(19)-C(24)	120.1(3)
C(5)-C(4)-C(19)-C(24)	-54.0(5)
Ru(1)-C(4)-C(19)-C(24)	-147.0(3)
C(3)-C(4)-C(19)-C(20)	-56.7(4)
C(5)-C(4)-C(19)-C(20)	129.1(3)
Ru(1)-C(4)-C(19)-C(20)	36.2(4)
C(24)-C(19)-C(20)-C(21)	1.4(5)
C(4)-C(19)-C(20)-C(21)	178.4(3)
C(19)-C(20)-C(21)-C(22)	0.8(5)
C(20)-C(21)-C(22)-C(23)	-2.3(5)
C(20)-C(21)-C(22)-C(25)	176.7(3)
C(21)-C(22)-C(23)-C(24)	1.7(5)
C(25)-C(22)-C(23)-C(24)	-177.3(3)
C(20)-C(19)-C(24)-C(23)	-2.0(5)
C(4)-C(19)-C(24)-C(23)	-179.0(3)
C(22)-C(23)-C(24)-C(19)	0.5(6)
C(1)-C(5)-C(26)-C(31)	-34.6(5)
C(4)-C(5)-C(26)-C(31)	135.5(3)
Ru(1)-C(5)-C(26)-C(31)	-129.3(3)
C(1)-C(5)-C(26)-C(27)	148.1(3)
C(4)-C(5)-C(26)-C(27)	-41.7(5)
Ru(1)-C(5)-C(26)-C(27)	53.5(4)
C(31)-C(26)-C(27)-C(28)	-0.3(5)
C(5)-C(26)-C(27)-C(28)	177.0(3)
C(26)-C(27)-C(28)-C(29)	1.8(5)
C(27)-C(28)-C(29)-C(30)	-2.2(5)
C(28)-C(29)-C(30)-C(31)	1.0(5)
C(29)-C(30)-C(31)-C(26)	0.5(5)
C(27)-C(26)-C(31)-C(30)	-0.8(5)
C(5)-C(26)-C(31)-C(30)	-178.2(3)
B(1)-O(2)-C(32)-C(33)	178.3(3)
B(1)-O(2)-C(32)-C(37)	1.3(3)
C(37)-C(32)-C(33)-C(34)	-0.6(5)
O(2)-C(32)-C(33)-C(34)	-177.1(3)
C(32)-C(33)-C(34)-C(35)	-0.1(5)
C(33)-C(34)-C(35)-C(36)	0.4(6)
C(34)-C(35)-C(36)-C(37)	0.0(5)
C(33)-C(32)-C(37)-C(36)	1.1(5)
O(2)-C(32)-C(37)-C(36)	178.2(3)
C(33)-C(32)-C(37)-O(3)	-178.0(3)
O(2)-C(32)-C(37)-O(3)	-0.9(4)
C(35)-C(36)-C(37)-C(32)	-0.7(5)
C(35)-C(36)-C(37)-O(3)	178.2(3)
B(1)-O(3)-C(37)-C(32)	0.1(3)
B(1)-O(3)-C(37)-C(36)	-178.9(3)

C(39)-Ru(1)-C(38)-O(4)	156(12)
B(1)-Ru(1)-C(38)-O(4)	71(12)
C(2)-Ru(1)-C(38)-O(4)	-36(12)
C(5)-Ru(1)-C(38)-O(4)	-49(12)
C(1)-Ru(1)-C(38)-O(4)	-12(12)
C(3)-Ru(1)-C(38)-O(4)	-74(12)
C(4)-Ru(1)-C(38)-O(4)	-97(12)
C(38)-Ru(1)-C(39)-O(5)	136(25)
B(1)-Ru(1)-C(39)-O(5)	-140(25)
C(2)-Ru(1)-C(39)-O(5)	3(25)
C(5)-Ru(1)-C(39)-O(5)	-36(25)
C(1)-Ru(1)-C(39)-O(5)	-57(25)
C(3)-Ru(1)-C(39)-O(5)	30(25)
C(4)-Ru(1)-C(39)-O(5)	2(25)

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Symmetry transformations used to generate equivalent atoms:

Table 7. Hydrogen bonds for **5** [ $\text{\AA}$  and  $^\circ$ ].

D-H...A	d(D-H)	d(H...A)	d(D...A)	$\angle(\text{DHA})$
O(1)-H(1)...O(2)	0.84	1.96	2.680(3)	143.6

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Symmetry transformations used to generate equivalent atoms: