Supporting Information

A bulky DNA lesion derived from a highly potent polycyclic aromatic tumorigen stabilizes nucleosome core particle structure

Yuqin Cai^a, Lihua Wang^a, Shuang Ding^a, Adam Schwaid^b, Nicholas E. Geacintov^b, and Suse Broyde*^a*, *.

^aDepartment of Biology and ^{*b*}Chemistry, New York University, New York, N.Y., 10003, ***Corresponding Author:** Suse Broyde, broyde@nyu.edu Tel. (212) 998-8231 Fax. (212) 995-4015

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Methods

Molecular modeling of the initial lesion-containing and unmodified NCP structure for MD simulations

Unmodified NCP Since our 14*S* (−)-*trans-anti*-DB[*a*,*l*]P-*N*⁶-dA adduct adopts an intercalative conformation (see below), we used a nucleosome structure with naturally basestretched segment of nucleosomal DNA (PDB (*1*) ID: 2NZD) (*2*) that can serve as an intercalation pocket upon alkylation by an aromatic chemical (*3*). The base sequence contexts around the intercalation pocket (at the 1.5 SHL, Figure 1d) was modified to the sequence context CAC:GTG (C202-A203-C204:G87-T88-G89) (Figure 1d), which has been extensively investigated (*4-9*).

Lesion-containing NCP Structural characterizations of *R* and *SN*⁶-dA adducts derived from the planar bay region PAH B[*a*]P (Figure S1) (*10-16*), and the twisted fjord region PAH B[*c*]Ph (Figure S1) (*5, 6, 17*) and B[*g*]C (*8*) have been carried out by NMR methods and computational approaches. These studies have shown that the *N* 6 -dA adducts adopt classical intercalation-type conformations: the *S* stereoisomers are intercalated on the 3'-side of the modified adenine, while the *R* stereoisomers are positioned on the 5'-side, and Watson-Crick base pairing, although distorted, remains at the damaged A:T pair. Accordingly, the initial

model for the 14*S* (−)-*trans-anti-*DB[*a,l*]P-*N 6 -*dA adduct was built from the duplex 11-mer high resolution NMR solution structure of the fjord region B[*c*]Ph analog with the same stereochemistry at the linkage site, 1*S* (−)-*trans-anti*-B[*c*]Ph-*N*⁶-dA (6) (Figure S1). Two rings were added to the B[*c*]Ph to create the DB[*a,l*]P initial model. This model on the nucleotide level was subjected to geometry optimization utilizing the Gaussian 03 package from Gaussian, Inc (*18*), and the optimized DB[*a,l*]P was relinked, with identical torsion angles, to the damaged adenine of the duplex 11-mer, and subjected to 30-ns of MD simulations (CaiY., Broyde S. et al, manuscript in preparation). Spectroscopic observations reveal that the UV absorption maximum for this adduct is red-shifted (Figure S2) which is fully consistent with an intercalative conformation.

Using the best representative structure (*19*) (shown in Figure 1b) from this ensemble, the lesion was cut (at the C14- N^6 bond (Figure S1)) and covalently linked to the adenine (A203) at the nucleosome stretch site (Figure 1d). The lesion-DNA linkage site torsion angles α' and β' assumed the same values as in the initial B[*c*]Ph adduct NMR structure (*6*) (α' = 130.6°, β' = 110.7°, Figure S1). Thus, we obtained a classical intercalation structure of the 14*S* (−) *trans-anti-*DB[*a,l*]P-*N 6 -*dA adduct covalently linked in the nucleosome stretch site, with the lesion occupying the intercalation pocket (Figures 1d and S3). Modeling software InsightII and Discovery Studio (Accelrys Inc.) were used.

Force field

The Cornell et al. force field (*20*) with modifications (*21, 22*) and the parm99 parameter set (*23*) modified by the parmbsc0 DNA parameters (*24*) were employed for all simulations.

Partial charges for the 14*S* (−)-*trans-anti*-DB[*a,l*]P-*N* 6 -dA adduct were calculated on the nucleotide level for the initial models, using Hartree Fock quantum mechanical calculations with 6-31G^{*} basis set without geometry optimization, employing the Gaussian 03 package from Gaussian, Inc (*18*). The charges were then fitted to each atomic center with the RESP algorithm (*25, 26*). The fitted charges are shown in Table S5. Missing bond and angle parameters were added to the force field by analogy to chemically similar atom types already exist in the parm99 parameter set and are shown in Table S6.

Protonation

The online protonation server PDB2PQR (http://pdb2pqr-1.wustl.edu/pdb2pqr/) (*27, 28*) was employed to determine the protonation state of the amino acid side chains. Based on the results of the PDB2PQR calculation, together with close inspection of the surrounding environment for possible hydrogen bonds, the ionization state of acids and bases in HIS, ARG, LYS, GLU and ASP were manually assigned. In our case, only HIS amino acids need to be assigned since the protonation state of a histidine is not clear from the crystal structure. The HIS can be protonated either on the N^{δ} or N^{ϵ} ² or both to produce HID, HIE, and HIP, respectively. A HIS residue must be the HID type to accept a hydrogen bond, whereas it must be either HIE or HIP to donate a hydrogen bond. The decision to choose between an HIE and an HIP was made based on the availability of an acceptor in the proximity. The results are: HIE: 487, 538, 746, 923; HIP: 594, 654, 979; HID: 627, 672, 770, 827, 872, 1012, 1039.

MD protocols

The AMBER 9.0 simulation package (*29*) was utilized to carry out all minimization and MD simulations. The LEaP module was used to add hydrogen atoms and neutralize the system with $K⁺$ counterions. Hydrogen atoms of the nucleosome core particle were minimized for 600 steps of steepest-descent followed by 600 steps of conjugate gradient. In this minimization, an implicit solvent model was used with a distance-dependent dielectric function of $\varepsilon = 4.0$ r, with r being the distance between two atoms. The systems were then reoriented using SIMULAID (*30*) to minimize the number of water molecules needed to solvate the system. A periodic rectangular box of TIP3P water (*31*) with 10.0 Å buffer was created around the NCPs with the LEaP module. For the unmodified NCP, the box dimensions were approximately 136 Å x 142 Å x 94 Å, with a total of \sim 41571 water molecules. For the 14*S* (−)-*trans-anti*-DB[*a,l*]P-*N* 6 -dA adduct-containing NCP, the box dimensions were approximately 136 Å x 142 Å x 94 Å, with a total of \sim 41563 water molecules.

The following minimization, heating, MD equilibration and production protocols were utilized. To begin with, the counterions and water molecules were minimized for 2500 steps of steepest descent and 2500 steps of conjugate gradient, with 50 kcal/mol restraints on the solute atoms. Then, 30 ps initial MD at 10K with 25 kcal/mol restraints on solute were performed to allow the solvent to relax. Next, the system was heated up from 10K to 300K at constant volume over 30 ps with 10 kcal/mol restraint on the NCP. The restraints on the solute were relaxed from 10 kcal/mol (for 30 ps) to 1 kcal/mol (for 40 ps) to 0.1 kcal/mol (50 ps). Subsequently, production MD was conducted at 1 atm, 300K for 65 ns, with 1 ps coupling constant for both pressure and temperature. These simulation protocols are similar to those given in earlier publication from our group (*32*).

In all MD simulations, the Particle-Mesh Ewald (*33, 34*) method with 9.0 Å cutoff for the non-bonded interactions was used. A 2.0 fs time step and the SHAKE algorithm (*35*) were applied in the MD simulations. All other parameters were default values in the AMBER 9 simulation package.

The stability of the MD simulation was evaluated. For each system, the root mean square deviation (RMSD) of each snapshot in the trajectory relative to its respective initial structure was plotted as a function of time and is shown in Figure S5. RMSDs for the local region (within 15 Å to the damaged base $A203^*$) of the damaged residue and its undamaged counterpart (within 15 Å to the base A203) of each snapshot in the trajectory relative to its respective initial structure were plotted as a function of time and are also shown (Figure 3d). For both cases, the MD achieved stability, fluctuating around the mean after 20.0 ns of simulation, and we employed the structural ensembles from the $20 - 65.0$ ns time frame at 10 ps intervals for further analyses.

Structural analyses

The PTRAJ module of the AMBER 9 package (*29*) was employed for structural analyses. The CARNAL module of the AMBER 7 package (*36*) was utilized to compute hydrogen bond occupancies. Frames were selected at 10 ps intervals from the last 45.0 ns of simulation. The DNA duplex helicoidal parameters and groove dimensions were analyzed using MD Toolchest (*37, 38*); for the groove dimensions, 5.8 Å was subtracted from the pairwise phosphorous-phosphorous distances to account for the van der Waals radius of the P atoms (*39*). Van der Waals interaction energies for the damaged NCP at the intercalation

pocket were computed between the DB[*a,l*]P aromatic rings and the adjacent bases, namely, T88, A203, C87, and G204. For the unmodified NCP, the interaction energies between the same base pairs that form the intercalation pocket were computed for comparison. The ANAL module of the AMBER 9 package (*29*) was used for these computations. The best representative structures (*19*) for each system were obtained using the methods described in (*32*).

Hydrogen bond quality index analyses

We employed our hydrogen bond quality index (HBI) (*13, 40*), to quantitatively assess the quality of Watson-Crick hydrogen bonding, in terms of the deviation from ideal Watson-Crick hydrogen bond distances and angles:

$$
I_{H} = \sum_{\text{Snapshot D-H...A}} [(d_{DA} - d_{DA}^{0})^{2} + (1 + \cos \gamma)^{2}]
$$

where d_{DA} is the instantaneous donor-acceptor distance, d_{DA}^0 is an ideal donor-acceptor distance (41) [for G:C pair: O6 (G) to N4 (C) is 2.91 Å, N1 (G) to N3 (C) is 2.95 Å, and N2 (G) to O2 (C) is 2.86 Å; for A:T pair: N (A) to O4 (T) is 2.95 Å, and N1 (A) to N (T) is 2.82 Å] and γ is the instantaneous donor-hydrogen \cdots acceptor (D-H \cdots A) hydrogen bond angle with an ideal value of 180 °. The summation is over all the Watson-Crick hydrogen bonds in a base pair over the trajectory. Snapshots were selected at 10 ps intervals over the 20-65 ns of the trajectory and 4500 snapshots were used. The lower the value of I_H , the better the quality of the Watson-Crick hydrogen bonding. *IH* is zero when the Watson-Crick hydrogen bonding is ideal during the dynamics. However, in real, even unmodified DNA, sequence-dependent deviations from ideal Watson-Crick hydrogen bonding are normal (*42*). Comparing the *I^H* value of a modified step to its analogue step in the unmodified control duplex provides an estimate of the integrity of its Watson-Crick hydrogen bond upon modification.

Modeling and computational resources

The INSIGHT II 2005 program (Accelrys Software, Inc.) was employed to visualize and build models. PyMOL (Delano Scientific, LLC.) (*43*) was employed to make molecular images and movies. Computations were carried out on our own cluster of Silicon Graphic Origin and Altix high-performance computers and through TeraGrid resources TG-MCB060028N at the Texas Advanced Computing Center provided by the National Science Foundation.

Figure S1. Chemical structures of B[*c*]Ph, B[*a*]P, DB[*a,l*]P, the 1*S* (−)-*trans-anti*-B[*c*]Ph-*N* 6 dA adduct and the 14*S* (−)-*trans-anti*-DB[*a,l*]P-*N* 6 -dA adduct with designation of fjord and bay regions. The DB[*a,l*]P adduct was built from the B[*c*]Ph by adding two rings. The lesion-DNA linkage site torsion angles α´ and β´ are defined as follows: α´, N1-C6-*N*-C1 (B[*c*]Ph); β', C6-N-C1 (B[*c*]Ph)-C2 (B[*c*]Ph) for the 1*S* (−)-*trans-anti*-B[*c*]Ph-N⁶-dA adduct; and α´, N1-C6-*N*-C14(DB[*a,l*]P); β´, C6-*N*-C14(DB[*a,l*]P)-C13(DB[*a,l*]P) for the 14*S* (−) *trans-anti*-DB[*a,l*]P-*N* 6 -dA adduct.

 Figure S2. Titration of the DB[*a,l*]P diol epoxide-modified strand containing a single 14*S* (−)-*trans-anti*-DB[*a,l*]P-*N* 6 -dA adduct with the complementary unmodified strand. The redshifted absorption spectra in the double-stranded DNA molecules are characteristic of intercalative adduct conformations. The broadened red-shifted absorption spectra reveal that there is a significant heterogeneity in the base-stacking interactions in these double-stranded duplexes (Cai Y. , Broyde S., manuscript in preparation).

Background. As shown earlier (*44, 45*), intercalated benzo[*a*]pyrene-derived DNA lesions are characterized by red-shifted UV absorption spectra relative to the absorption spectra of the same PAH residue in single-stranded DNA. External minor groove adduct conformations, on the other hand, are characterized by blue shifted spectra. These differences are revealed by titrating the PAH-modified single-stranded oligonucleotides with the complementary strand and observing the shifts in the absorption maxima as the double-stranded molecules are formed.

Figure S3. (a) Initial structure of the nucleosome core particle containing the 14*S* (−)-*transanti*-DB[a, l]P- N^6 -dA lesion (red) and (b) Following 65-ns.

Figure S4. Views down the helix axis showing the more distorted hydrogen bond angle in the unmodified NCP compared to the modified one (See Table S2). Stacking of the DB[*a,l*]P ring system with the T88 partner base stabilizes this hydrogen bond angle.

Figure S5. Time dependence of RMSDs for the entire NCP containing the 14*S* (−)-*transanti*-DB[a, l]P- N^6 -dA adduct (red) and the entire unmodified NCP (green). The RMSDs were calculated relative to their respective initial structures. Only the solute (DNA and protein) was considered. The ensemble average RMSDs with standard deviations are given.

Table S1. Ensemble average van der Waals stacking interaction energies (kcal/mol) between adjacent base pairs and between the DB[*a,l***]P aromatic rings and surrounding bases in the interaction pocket for the damaged NCP, and between the analogous base pairs in the unmodified NCP.***^a*

^a Standard deviations are given in parentheses. The van der Waals stacking interaction energies between DB[*a,l*]P and partner T88 to the damaged A203 is highlighted in yellow.

Table S2. Ensemble average hydrogen bond distances and angles at the damaged base pair and the analogous base pair in the unmodified NCP *^a*

^a Standard deviations are given in parentheses. Hydrogen bond angles in ideal B-DNA are 180 °. Highlighted in yellow is the hydrogen bond angle that is more ideal in the lesioncontaining NCP than in the unmodified NCP. Also note that the standard deviations for all angles are greater in the unmodified.

^{*a*} Criteria for hydrogen bonding: heavy atom-heavy atom distance $<$ 3.3 Å, and donorhydrogen-acceptor angle > 140 °. Note the hydrogen bonds (highlighted in yellow) in the 14*S*-DB[*a,l*]P-NCP involving Arg 702 which are not present in the unmodified NCP. All other hydrogen bond differences provide no net gain in total hydrogen bonds in the 14*S*-DB[*a,l*]P-NCP compared to the unmodified NCP. This can be seen by adding the occupancies for the two cases: 650 %, and 550 %, (or 6.5 and 5.5 hydrogen bonds) respectively, for the 14*S*-DB[*a,l*]P-NCP and unmodified NCP.

| Major groove width | $14S\text{-}DB[a,l]P\text{-}NCP$ | Unmodified |
|--------------------|----------------------------------|-------------------|
| P83-P205 | 10.5 ± 0.8 | 12.0 ± 0.8 |
| P84-P204 | 10.3 ± 1.1 | 11.1 ± 1.4 |
| P85-P203 | 15.8 ± 1.3 | 14.9 ± 1.8 |
| P86-P202 | 18.2 ± 1.7 | 16.9 ± 2.0 |
| P87-P201 | 18.9 ± 1.5 | 16.1 ± 2.2 |
| P88-P200 | 15.9 ± 1.6 | 14.5 ± 1.5 |

Table S4. Ensemble average groove widths (Å).*^a* **Standard deviations are given in parentheses.**

^aThe lesion site and its analogous site in the unmodified NCP are highlighted in yellow. Note the sharp minor groove narrowing in the damaged NCP (see also Figure 3c). The major groove opening in the damaged compared to the unmodified NCP stems from the intercalation on the major groove side.

| | Atom Name | Atom Type | Topology | Partial Charge |
|----------------|------------------|------------------|------------------|-----------------------|
| $\mathbf{1}$ | \mathbf{P} | \mathbf{P} | M | 1.2687 |
| $\mathbf{2}$ | O1P | O ₂ | E | -0.795 |
| 3 | O ₂ P | O ₂ | E | -0.795 |
| $\overline{4}$ | O5' | OS | M | -0.476 |
| 5 | C5' | CT | $\mathbf M$ | -0.169 |
| 6 | H5'1 | H1 | E | 0.1079 |
| 7 | H5'2 | H1 | E | 0.1079 |
| 8 | C4' | CT | $\mathbf M$ | 0.3142 |
| 9 | H4' | H1 | E | 0.0639 |
| 10 | O4' | OS | S | -0.451 |
| 11 | C1 | CT | \boldsymbol{B} | 0.2203 |
| 12 | H1' | H2 | E | 0.1191 |
| 13 | N ₉ | ${\bf N}^*$ | S | -0.081 |
| 14 | C8 | CK | \boldsymbol{B} | 0.1243 |
| 15 | H8 | H ₅ | E | 0.1856 |
| 16 | N7 | NB | S | -0.661 |
| 17 | C ₅ | CB | S | 0.0524 |
| 18 | C6 | CA | \boldsymbol{B} | 0.4648 |
| 19 | $\mathbf N$ | N2 | B | -0.535 |
| 20 | HN | H | E | 0.3617 |
| 21 | C14 | CT | \boldsymbol{B} | -0.025 |
| 22 | HC14 | H1 | E | 0.1766 |
| 23 | C13 | CT | 3 | 0.0299 |
| 24 | HC13 | H1 | E | 0.1574 |
| 25 | O13 | OH | S | -0.625 |
| 26 | HO13 | HO | E | 0.3875 |
| 27 | C12 | CT | 3 | 0.2376 |
| 28 | HC12 | H1 | E | 0.0834 |
| 29 | O12 | OH | S | -0.659 |
| 30 | HO12 | HO | E | 0.4245 |
| 31 | C11 | CT | 3 | 0.1874 |
| 32 | HC11 | H1 | E | 0.0222 |
| 33 | O11 | OH | S | -0.637 |
| 34 | HO11 | HO | E | 0.4095 |
| 35 | C16 | CA | S | -0.019 |

Table S5. Partial charges, atom types and topologies for the 14*S* **(**−**)-***trans-anti***-DB[***a,l***]P-***N* **6 -dA adduct.**

| Angle | K_{θ} (kcal·mol ⁻¹ ·rad ⁻²) | θ_{eq} (deg) | |
|--|---|---------------------|--|
| $N2$ -CT-CA | 80.0 | 111.20 | |
| H ₁ -C _T -C _A | 50.0 | 109.50 | |
| OH-CT-CA | 50.0 | 109.50 | |

Table S6. Added force field parameters *^a*

^a Missing bond and angle parameters were added to the force field by analogy to chemically similar atom types already exist in the parm99 parameter set.

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