## *N*-Isopropylidene-*N*'-2-Nitrobenzenesulfonyl Hydrazine. A Reagent for Conversion of Alcohols to the Corresponding Monoalkyl Diazenes.

Massachusetts Institute of Technology, Department of Chemistry, Massachusetts 02139

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**General Procedures.** All reactions were performed in oven-dried or flame-dried round bottomed flasks or modified Schlenk (Kjeldahl shape) flasks. The flasks were fitted with rubber septa and reactions were conducted under a positive pressure of argon. Stainless steel syringes or cannulae were used to transfer air- and moisture-sensitive liquids. Flash column chromatography was performed as described by Still et al. using silica gel (60-Å pore size,  $32-63 \mu m$ , standard grade).<sup>1</sup> Analytical thin–layer chromatography was performed using glass plates pre-coated with 0.25 mm 230–400 mesh silica gel or neutral alumina gel impregnated with a fluorescent indicator (254 nm). Thin layer chromatography plates were visualized by exposure to ultraviolet light and/or by exposure to an ethanolic phosphomolybdic acid (PMA), an acidic solution of *p*-anisaldehyde (anis), an aqueous solution of ceric ammonium molybdate (CAM), an aqueous solution of potassium permanganate (KMnO<sub>4</sub>) or an ethanolic solution of ninhydrin followed by heating (<1 min) on a hot plate (~250 °C). Organic solutions were concentrated on rotary evaporators at ~20 Torr (house vacuum) at 25–35 °C, then at ~1 Torr (vacuum pump) unless otherwise indicated.

**Materials.** Commercial reagents and solvents were used as received with the following exceptions: Dichloromethane, diethyl ether, tetrahydrofuran, acetonitrile, toluene, and triethylamine were purified by the method of Grubbs et al. under positive argon pressure.<sup>2</sup> The molarity of *n*-butyllithium solutions was determined by titration using diphenylacetic acid as an indicator (average of three determinations).<sup>3</sup> Chlorobenzene was distilled over CaCl<sub>2</sub> under an argon atmosphere, 2,6-lutidene over CaH<sub>2</sub> under an argon atmosphere, and 2,2,2-trifluoroethanol over calcium sulfate and sodium

<sup>1.</sup> Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923.

<sup>2.</sup> Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. Organometallics 1996, 15, 1518.

<sup>3.</sup> Kofron, W. G.; Baclawski, L. M. J. Org. Chem. 1976, 41, 1879.

bicarbonate under an argon atmosphere. Sodium hydride was purchased as a 60% dispersion in mineral oil, washed four times with hexanes and stored in a glove-box.

**Instrumentation.** Proton nuclear magnetic resonance (<sup>1</sup>H NMR) spectra were recorded with 400 and 500 MHz spectrometers. Chemical shifts are recorded in parts per million from internal tetramethylsilane on the  $\delta$  scale and are referenced from the residual protium in the NMR solvent (CHCl<sub>3</sub>:  $\delta$  7.27, CD<sub>3</sub>CN:  $\delta$  1.96). Data are reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constant(s) in Hertz, integration, assignment]. Carbon-13 nuclear magnetic resonance (<sup>13</sup>C NMR ) spectra were recorded with 400 and 500 MHz spectrometers and are recorded in parts per million from internal tetramethylsilane on the  $\delta$  scale and are referenced from the carbon resonances of the solvent (CDCl<sub>3</sub>:  $\delta$  77.2). Data are reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constant(s) in Hertz, assignment]. Infrared data were obtained with an FTIR and are reported as follows: [frequency of absorption (cm<sup>-1</sup>), intensity of absorption (s = strong, m = medium, w = weak, br = broad), assignment]. Gas chromatography was performed on a HP-5 5% Phenyl Methyl Siloxane column.



#### <u>N-Isopropylidene-N'-2-nitrobenzenesulfonyl hydrazine (IPNBSH, eq 1):</u><sup>4</sup>

A sample of NBSH<sup>5</sup> (1.14 g, 5.24 mmol, 1 equiv) was dissolved in acetone (8.0 mL) at 0 °C. After 30 min, the solvent was removed in vacuo and the residue was dissolved in acetone (4 mL). Slow addition of the acetone solution to a volume of hexanes (150 mL), collection of the fine powder by filtration, followed by sequential hexanes rinses (2 × 5 mL), and removal of volatiles by high vacuum provided triturated IPNBSH<sup>4</sup> as a white solid (1.20 g, 89%). mp: 139–140 °C (decomposition). TLC (66% ethyl acetate in hexanes), *Rf*: 0.7 (UV, CAM, anis). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, 20 °C): 8.30–8.28 (m, 1H), 7.87–7.85 (m, 2H) 7.79–7.77 (m, 2H), 1.96 (s, 3H), 1.92 (s, 3H). <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>, 20 °C): 158.9, 134.2, 133.5, 132.9, 132.0, 125.4, 25.5, 17.2. FTIR (thin film): 3264 (s), 1551 (s), 1375 (s), 1347 (m), 1177 (s). HRMS–ESI: calc'd for C<sub>9</sub>H<sub>12</sub>N<sub>3</sub>O<sub>4</sub>S [M+H]<sup>+</sup>: 258.0543, found: 258.0548.

<sup>4.</sup> Movassaghi, M.; Piizzi, G.; Siegel, D. S.; Piersanti, G. Angew. Chem. Int. Ed. 2006, 45, 5859.

<sup>5. (</sup>a) Dann, A. T.; Davies, W. J. Chem. Soc. **1929**, 1050 (b) Myers, A. G.; Zheng, B.; Movassaghi, M. J. Org. Chem. **1997**, 62, 7507. (c) Myers, A. G.; Movassaghi M. In *e-Encyclopedia of Reagents for Organic Synthesis*; Paquette, L. A., Ed; John Wiley & Sons, 2003.

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## <u>N-Isopropylidene-N'-((2E,6E)-3,7,11-trimethyl-dodeca-2,6,10-trienyl)-N'-2-nitrobenzene-</u> sulfonyl hydrazide (4a, Scheme 2):

DEAD (0.30 mL, 1.9 mmol, 1.2 equiv) was added drop-wise to a solution of IPNBSH (487 mg, 1.89 mmol, 1.20 equiv), *trans,trans*-farnesol (**1a**, 400  $\mu$ L, 1.57 mmol, 1 equiv), and triphenylphosphine (496 mg, 1.89 mmol, 1.20 equiv) in anhydrous THF (30 mL) at 0 °C under an argon atmosphere. After 20 min, the volatiles were removed and the residue was purified by flash column chromatography on silica gel (30% ethyl acetate in hexanes) to afford the sulfonyl hydrazide **4a** (624 mg, 86%). TLC (40% ethyl acetate in hexanes), *Rf*: 0.4 (UV, KMnO<sub>4</sub>). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, 20 °C): 7.98–7.96 (m, 1H), 7.73–7.68 (m, 2H), 7.55–7.54 (m, 1H), 5.08–5.04 (m, 3H), 3.84 (d, 2H, *J* = 7.0), 2.12 (m, 3H), 2.11 (m, 3H), 2.07–2.01 (m, 4H), 1.98–1.94 (m, 4H), 1.68 (s, 3H), 1.62 (s, 3H), 1.60 (s, 3H), 1.58 (s, 3H). <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>, 20 °C): 182.8, 149.7, 142.5, 135.6, 134.1, 131.9, 131.5, 130.9, 128.3, 124.3, 123.7, 123.6, 117.0, 49.5, 39.9, 39.7, 26.6, 26.5, 25.9, 25.2, 21.2, 17.8, 16.5, 16.1. FTIR (thin film): 2918 (br s), 1644 (m), 1589 (m), 1547 (s), 1439 (s), 1372 (s). HRMS-ESI: calc'd for C<sub>24</sub>H<sub>36</sub>N<sub>3</sub>O<sub>4</sub>S [M+H]<sup>+</sup>: 462.2421, found: 462.2412.



#### (E)-3,7,11-Trimethyldodeca-1,6,10-triene (5a, eq 2):<sup>6</sup>

A solution of IPNBSH (96 mg, 0.37 mmol, 1.3 equiv) in anhydrous DMF (1.5 mL) was added slowly to a suspension of sodium hydride (8.6 mg, 0.36 mmol, 1.2 equiv) in anhydrous DMF (1.5 mL) at 0 °C under an argon atmosphere to give an orange solution. After 1.5 h, trans, trans-farnesyl bromide (6, 81 µL, 0.30 mmol, 1 equiv) was added to the sodium amide solution and the resulting mixture was allowed to warm to 23 °C. After 3 h, excess base was guenched with glacial acetic acid (86  $\mu$ L, 1.5 mmol, 5.0 equiv), and the reaction mixture was diluted by the addition of a mixture of trifluoroethanol and water (1:1, 1.5 mL). After 11 h, the reaction mixture was diluted with water (25 mL) and extracted with diethyl ether  $(3 \times 25 \text{ mL})$ . The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated. The residue was purified by flash column chromatography on silica gel (100% "pentane) to afford the triene 5a (54 mg, 88%). All spectroscopic data were in agreement with the literature.<sup>6</sup> TLC (40% ethyl acetate in hexanes), Rf: 0.8 (CAM, KMnO<sub>4</sub>). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, 20 °C): 5.25–5.15 (m, 1H), 5.13–5.10 (m, 2H), 4.99 (t, 1H, J = 2.0 Hz), 4.94–4.91 (m, 2H), 2.14–1.98 (m, 7H), 1.69 (s, 3H), 1.61 (s, 3H), 1.60 (s, 3H), 1.36–1.33 (m, 2H), 0.99 (d, 3H, J = 8 Hz). <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>, 20 °C): 145.0, 135.1, 131.5, 124.7, 124.6, 112.7, 39.9, 37.5, 36.9, 26.9, 25.9, 25.8, 20.4, 20.4, 17.9, 16.2. FTIR (thin film): 3077 (m), 2966 (s), 2915 (s), 2856 (s), 1640 (m), 1453 (s), 1376 (s). MS  $(m/z) C_{15}H_{26} [M]^+$ : 206.

<sup>6.</sup> For prior syntheses of **5a**, see Saplay, K. M.; Sahni, R.; Damodaran, N. P.; Dev, S. *Tetrahedron* **1980**, *36*, 1455 and Myers, A. G.; Zheng, B. *Tetrahedron Lett.* **1996**, *37*, 4841.

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#### (E)-N'-(2,2-dimethyl-propylidene)-2-nitrobenzenesulfono-hydrazide (7b, Scheme 3):

NBSH (1.00 g, 4.62 mmol, 1 equiv) was dissolved in a solution of excess 2,2-dimethylpropionaldehyde in 'butanol (80% v/v, 8.0 mL) at 0 °C. After 30 min, the volatiles were removed under reduced pressure, and the residue was dissolved in dichloromethane (2 mL). Slow addition of this solution to a volume of hexanes (125 mL), collection of the resulting fine brown powder by filtration, followed by successive hexane rinses (2 × 5 mL) provided the hydrazone **7b** as a brown solid (1.16g, 92%). TLC (50% ethyl acetate in hexanes), R*f*: 0.8 (UV, CAM). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, 20 °C)  $\delta$ : 8.24–8.21 (m, 1H), 7.91 (s, 1H), 7.86–7.76 (m, 3H), 7.26 (s, 1H), 0.99 (s, 9H). <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>, 20 °C)  $\delta$ : 163.1, 134.4, 133.3, 132.6, 131.6, 125.1, 35.4, 27.2. FTIR (thin film): 3238 (s), 1548 (s), 1366 (s), 1321 (m), 1173 (s). HRMS-ESI (*m*/*z*): calcd for C<sub>11</sub>H<sub>15</sub>N<sub>3</sub>O<sub>4</sub>S [M+H]<sup>+</sup>: 286.0856, found: 286.0848.



## 2,2-Dimethyl-propionic acid N'-{2-[1-(4-chloro-benzoyl)-5-methoxy-2-methyl-1H-indol-3-yl]ethyl} hydrazide (11b, Scheme 3):

DEAD (78 µL, 0.50 mmol, 1.6 equiv) was added drop-wise to a solution of (E)-N-(2,2dimethyl-propylidene)-2-nitrobenzenesulfono-hydrazide (7b, 137 mg, 0.502 mmol, 1.63 equiv), N-(4-chlorobenzoyl)-3-(2-hydroxyethyl)-5-methoxy-2-methylindole (1b, 106 mg, 0.308 mmol, 1 equiv), and triphenylphosphine (133 mg, 0.505 mmol, 1.64 equiv) in anhydrous THF (3 mL) at 0 °C under an argon atmosphere. After 5 min, the reaction mixture was allowed to warm to 23 °C. After 30 min, a mixture of trifluoroethanol and water (1:1, 1.5 mL) was added to the reaction. After 11 h, the reaction mixture was partitioned between diethyl ether (25 mL) and water (10 mL). The organic layer was washed with water  $(3 \times 10 \text{ mL})$ , was dried over anhydrous sodium sulfate, was concentrated under reduced pressure, and the residue was purified by flash column chromatography on silica gel (75% ethyl acetate in hexanes) to give the hydrazide **11b** (107 mg, 79%). TLC (75% ethyl acetate in hexanes), Rf: 0.3 (UV, CAM, anis). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, 20 °C): 7.67–7.63 (m, 2H), 7.48–7.46 (m, 2H), 7.11 (s, 1H), 6.96–6.95 (d, 1H, J = 4.0 Hz), 6.91–6.89 (d, 1H, J = 8.0Hz), 6.68–6.66 (dd, 1H, J = 8.0, 4.0 Hz), 4.77 (br s, 1H), 3.85 (s, 3H), 3.10–3.06 (t, 2H, J = 8.0 Hz), 2.87–2.83 (t, 2H, J = 8.0 Hz), 2.35 (s, 3H), 1.15 (s, 9H). <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>, 20 °C): 177.9, 168.4, 156.1, 139.2, 134.7, 134.2, 131.3, 130.7, 129.2, 117.3, 115.2, 111.4, 101.3, 99.9, 55.9, 51.7, 38.0, 27.3, 23.4, 13.5. FTIR (this film): 3305 (br m), 2960 (m), 1679 (s), 1478 (s), 1365 (s), 1325 (s). HRMS-ESI: calc'd for  $C_{24}H_{29}CIN_3O_3$  [M+H]<sup>+</sup>: 442.1897, found: 442.1900.

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### But-3-envloxymethyl-benzene (S5c, Table 1, Entry 2):7

DEAD (66  $\mu$ L, 0.42 mmol, 1.2 equiv) was added dropwise to a solution of IPNBSH (107 mg, 0.416 mmol, 1.20 equiv), (*Z*)-4-(benzyloxy)but-2-en-1-ol (**S1c**, 62 mg, 0.347 mmol, 1 equiv), and triphenylphosphine (109 mg, 0.417 mmol, 1.20 equiv) in anhydrous THF (3.5 mL) at 0 °C under an argon atmosphere. After 5 min, the reaction mixture was allowed to warm to 23 °C. After 2 h, a mixture of trifluoroethanol and water (1:1, 1.8 mL) was added to the reaction mixture. After 3 h, the reaction mixture was diluted with water (30 mL) and extracted with diethyl ether (3 × 30 mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated. The residue was purified by flash column chromatography on silica gel (4% diethyl ether in "pentane) to give **S5c** (50 mg, 89%). TLC (4% diethyl ether in pentane), *Rf*: 0.3 (anis). All spectroscopic data were in agreement with the literature.<sup>7</sup>



#### (E)-3,7,11-Trimethyldodeca-1,6,10-triene (5a, Table 1, entry 3):

DEAD (74  $\mu$ L, 0.47 mmol, 1.2 equiv) was added drop-wise to a solution of IPNBSH (122 mg, 0.474 mmol, 1.21 equiv), *trans,trans*-farnesol (**1a**, 0.100 mL, 0.393 mmol, 1 equiv), and triphenylphosphine (124 mg, 0.473 mmol, 1.21 equiv) in anhydrous THF (9.0 mL) at 0 °C under an argon atmosphere. After 5 min, the reaction mixture was allowed to warm to 23 °C. After 20 min, a mixture of trifluoroethanol and water (1:1, 4.5 mL) was added to the reaction mixture to enable formation of the allylic diazene intermediate. After 3 h, the reaction mixture was partitioned between diethyl ether (25 mL) and water (25 mL), and the aqueous layer was extracted with diethyl ether (2 × 25 mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated. The residue was purified by flash column chromatography on silica gel (100% "pentane) to give triene **5a** (71 mg, 87%). See above for characterization data.<sup>6</sup>



#### (E)-5-Phenylpent-2-ene (S5d, Table 1, Entry 4):<sup>8</sup>

DEAD (55  $\mu$ L, 0.35 mmol, 1.1 equiv)\_was added drop-wise to a solution of IPNBSH (89 mg, 0.35 mmol, 1.1 equiv), 5-phenylpent-1-en-3-ol (**S1d**, 50 mg, 0.31 mmol, 1 equiv), and triphenylphosphine (91 mg, 0.35 mmol, 1.1 equiv) in anhydrous THF (2.9 mL) at 0 °C under an argon atmosphere. After 5 min, the reaction mixture was allowed to warm to 23 °C. After 2 h, a mixture of trifluoroethanol and water (1:1, 2.9 mL) was added to the reaction mixture to enable the formation of the intermediate allyl diazene. After 2 h, the reaction mixture was partitioned between diethyl ether (25 mL) and water (25 mL) and the aqueous layer was extracted with diethyl ether (2 × 25mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated. The residue was purified by flash column chromatography on silica gel (100%)

<sup>7.</sup> For a prior synthesis of S5c, see Cleary, P. A.; Woerpel, K. A, Org. Lett., 2005, 7, 5531.

<sup>8.</sup> For a prior synthesis of **S5d**, see Buss, A. D.; Warren, S. J. Chem. Soc., Perkin Trans I, **1985**, 2307 and Myers, A. G.; Zheng, B. *Tetrahedron Lett.* **1996**, *37*, 4841.

<sup>*n*</sup>pentane) to give the alkene **S5d** (34 mg, 74%). TLC (pentane), *Rf*: 0.5 (KMnO<sub>4</sub>). <sup>1</sup>H NMR (400 MHz) analysis revealed an *E*:*Z* ratio of 93:7. All spectroscopic data were in agreement with the literature.<sup>8</sup>



## <u>5α-Cholest-3-ene (S5e, Table 1, Entry 5):<sup>9</sup></u>

DEAD (45  $\mu$ L, 0.28 mmol, 2.0 equiv) was added drop-wise to a solution of IPNBSH (72 mg, 0.28 mmol, 2.0 equiv), cholest-4-en-3 $\beta$ -ol (**S1e**, 54 mg, 0.14 mmol, 1 equiv), and triphenylphosphine (74 mg, 0.28 mmol, 2.0 equiv) in anhydrous THF (1.4 mL) at 0 °C under an argon atmosphere. After 5 min, the reaction mixture was allowed to warm to 23 °C. After 2 h, a mixture of trifluoroethanol and water (1:1, 0.7 mL) was added to the reaction mixture. After 2 h, the reaction mixture was diluted with water (25 mL) and extracted with diethyl ether (3 × 25mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated. The residue was purified by flash column chromatography on silica gel (100% hexanes) to give the alkene **S5e** (31 mg, 60%). TLC (Pentane), *Rf*: 0.8 (CAM). <sup>1</sup>H NMR (500 MHz) analysis revealed the presence of <8% of the C<sub>5</sub>-diastereomers. All spectroscopic data were in agreement with the literature.<sup>9</sup>



# 5-Phenylpenta-1,2-diene (S5f, Table 1, Entry 6):10

DEAD (67.5 µL, 0.429 mmol, 1.20 equiv) was added drop-wise to a solution of IPNBSH (110 mg, 0.429 mmol, 1.20 equiv), 5-phenylpent-1-yn-3-ol (**S1f**, 57 mg, 0.36 mmol, 1 equiv), and triphenylphosphine (112 mg, 0.428 mmol, 1.20 equiv) in anhydrous THF (3.5 mL) at 0 °C under an argon atmosphere. After 5 min, the reaction mixture was allowed to warm to 23 °C. After 2 h, a mixture of trifluroethanol and water (1:1, 1.7 mL) was added to the reaction mixture to enable the formation of the propargylic diazene intermediate. After 2 h, the reaction mixture was partitioned between "pentane (25 mL) and water (25 mL). The organic layer was washed with water (2 × 25mL), was dried over anhydrous sodium sulfate, was filtered, and was concentrated. The residue was purified by flash column chromatography on silica gel (100% "pentane) to give the allene **S5f** (36 mg, 70%). TLC (100% "pentane), *Rf*: 0.6 (anis). All spectroscopic data were in agreement with the literature.<sup>10</sup>



## Trimethyl-(5-phenylpenta-1,2-dienyl)-silane (S5g, Table 1, Entry 7):<sup>11</sup>

DEAD (43  $\mu$ L, 0.27 mmol, 1.2 equiv) was added drop-wise to a solution of IPNBSH (69 mg, 0.27 mmol, 1.2 equiv), 5-phenyl-1-(trimethylsilyl)pent-1-yn-3-ol (**S1g**, 52 mg, 0.22, 1 equiv), and triphenylphosphine (71 mg, 0.27, 1.2 equiv) in anhydrous THF (2.2 mL) at 0 °C under an argon

<sup>9.</sup> For prior syntheses of S5e, see patent-JP,05-051329, A and Myers, A. G.; Zheng, B. Tetrahedron Lett. 1996, 37, 4841.

<sup>10.</sup> For a prior synthesis of **S5f**, see Ohno, H.; Miyamura, K.; Tanaka, T. J. Org. Chem. **2002**, 67, 1359.

<sup>11.</sup> For a prior synthesis of S5g, see Danheiser, R. L.; Carini, D. J., Fink, D. M., Basak, A. Tetrahedron, 1983, 39, 935.

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atmosphere. After 5 min, the reaction mixture was allowed to warm to 23 °C. After 2 h, a mixture of trifluoroethanol and water (1:1, 1.1 mL) was added to the reaction mixture to enable the formation of the propargylic diazene intermediate. After 2.5 h, the reaction mixture was diluted with water (25 mL) and extracted with diethyl ether ( $3 \times 25$ mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated. The residue was purified by flash column chromatography on silica gel (100% "pentane) to afford the allene **S5g** (41 mg, 86%). TLC ("pentane), R*f*: 0.4 (UV, anis). All spectroscopic data were in agreement with the literature.<sup>11</sup>



## 1-(4-Chlorobenzoyl)-3-ethyl-5-methoxy-2-methyl-indole (S5b, Table 1, Entry 8):<sup>12</sup>

DEAD (44  $\mu$ L, 0.28 mmol, 1.6 equiv) was added drop-wise to a solution of IPNBSH (71 mg, 0.28 mmol, 1.6 equiv), *N*-(4-chlorobenzoyl)-3-(2-hydroxyethyl)-5-methoxy-2-methylindole (**1b**, 59 mg, 0. 17 mmol, 1 equiv), and triphenylphosphine (73 mg, 0.28 mmol, 1.6 equiv) in anhydrous THF (1.7 mL) at 0 °C under an argon atmosphere. After 5 min, the reaction mixture was allowed to warm to 23 °C. After 30 min, phenylhydrazine (85  $\mu$ L, 0.86 mmol, 5.0 equiv) was added via syringe and the mixture was kept at ambient temperature. After 4 h, the reaction mixture was diluted with diethyl ether (25 mL) and washed with water (25 mL). The aqueous layer was extracted with diethyl ether (2 × 25 mL), and the combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated. The residue was purified by flash column chromatography on silica gel (5% ethyl acetate in hexanes) to afford **S5b** (50 mg, 89%). TLC (5% ethyl acetate in hexanes), *Rf*: 0.3 (UV, CAM anis). All spectroscopic data were in agreement with the literature.<sup>12</sup>



## 1-Phenylheptane (S5h, Table 1, Entry 9):<sup>12</sup>

DEAD (290  $\mu$ L, 1.84 mmol, 2.50 equiv) was added drop-wise to a solution of IPNBSH (474 mg, 1.84 mmol, 2.50 equiv), 1-phenylheptan-3-ol (**S1h**, 142 mg, 0.737 mmol, 1 equiv), and triphenylphosphine (485 mg, 1.85 mmol, 2.51 equiv) in anhydrous THF (5.2 mL) at 0 °C under an argon atmosphere. After 5 min, the reaction mixture was allowed to warm to 23 °C. After 20 h, phenyl hydrazine (360  $\mu$ L, 3.67 mmol, 4.98 equiv) was added via syringe and the resulting mixture was kept at ambient temperature. After 13 h, the reaction mixture was partitioned between "pentane (25 mL) and water (25 mL). The organic layer was washed with water (4 × 25 mL), was dried over anhydrous sodium sulfate, was filtered, and was concentrated. The residue was purified by flash column chromatography (100% "pentane) to afford 1-heptylbenzene (**S5h**, 46 mg, 35%). TLC (30% EtOAc in hexanes), *Rf*: 0.9 (CAM). All spectroscopic data were in agreement with the literature.<sup>12</sup>

<sup>12.</sup> For prior syntheses of S5b and S5h, see Myers, A. G.; Movassaghi, M.; Zheng, B. J. Am. Chem. Soc. 1997, 119, 8572.

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#### Crystal Structure of IPNBSH.



Table S1. Crystal data and structure refinement for IPNBSH.

Empirical formula Formula weight Temperature Wavelength Crystal system Space group Unit cell dimensions Volume Ζ Density (calculated) Absorption coefficient F(000) Crystal size Theta range for data collection Index ranges Reflections collected Independent reflections Completeness to theta =  $29.57^{\circ}$ Absorption correction Max. and min. transmission Refinement method Data / restraints / parameters Goodness-of-fit on F<sup>2</sup> Final R indices [I>2sigma(I)] R indices (all data) Largest diff. peak and hole

Identification code

06182 C9 H11 N3 O4 S 257.27 100(2) K 0.71073 Å Monoclinic P2(1)/n $\alpha = 90^{\circ}$ . a = 8.1746(3) Å b = 15.1930(5) Å $\beta = 107.4690(10)^{\circ}$ . c = 9.8516(3) Å $\gamma = 90^{\circ}$ . 1167.11(7) Å<sup>3</sup> 4 1.464 Mg/m<sup>3</sup> 0.285 mm<sup>-1</sup> 536 0.50 x 0.50 x 0.40 mm<sup>3</sup> 2.55 to 29.57°. -10<=h<=11, -21<=k<=21, -13<=l<=13 22799 3274 [R(int) = 0.0201]100.0 % Semi-empirical from equivalents 0.8946 and 0.8707 Full-matrix least-squares on F<sup>2</sup> 3274 / 1 / 159 1.060 R1 = 0.0299, wR2 = 0.0848R1 = 0.0310, wR2 = 0.08580.485 and -0.387 e.Å-3

	Х	У	Z	U(eq)
O(4)	14328(1)	638(1)	10896(1)	19(1)
O(1)	10353(1)	1177(1)	10334(1)	16(1)
O(2)	8502(1)	2204(1)	8564(1)	19(1)
S(1)	9667(1)	1478(1)	8889(1)	13(1)
N(2)	8713(1)	623(1)	7983(1)	14(1)
N(1)	8126(1)	787(1)	6509(1)	20(1)
C(2)	7743(2)	110(1)	5704(1)	21(1)
C(1)	7901(2)	-824(1)	6199(1)	37(1)
C(3)	7091(2)	277(1)	4136(1)	42(1)
O(3)	12382(1)	-78(1)	9276(1)	17(1)
N(3)	13246(1)	580(1)	9726(1)	13(1)
C(4)	11421(1)	1759(1)	8265(1)	13(1)
C(5)	13012(1)	1342(1)	8763(1)	13(1)
C(6)	14415(1)	1592(1)	8345(1)	16(1)
C(8)	12641(1)	2704(1)	6852(1)	19(1)
$\dot{C(9)}$	11247(1)	2452(1)	7304(1)	16(1)
C(7)	14214(1)	2279(1)	7368(1)	18(1)

**Table S2.** Atomic coordinates (  $x \ 10^4$ ) and equivalent isotropic displacement parameters (Å<sup>2</sup>x  $10^3$ ) for IPNBSH. U(eq) is defined as one third of the trace of the orthogonalized U<sup>ij</sup> tensor.

Table S3. Bond lengths [Å] and angles [°] for IPNBSH.

O(4)-N(3)	1.2281(11)	O(1)-S(1)-C(4) 107.68(4)	
O(1)-S(1)	1.4388(7)	N(2)-S(1)-C(4)	107.74(4)
O(2)-S(1)	1.4292(7)	N(1)-N(2)-S(1)	112.40(7)
S(1)-N(2)	1.6356(9)	C(2)-N(1)-N(2)	116.15(9)
S(1)-C(4)	1.7735(10)	N(1)-C(2)-C(1)	125.50(10)
N(2)-N(1)	1.4077(11)	N(1)-C(2)-C(3)	116.64(10)
N(1)-C(2)	1.2791(14)	C(1)-C(2)-C(3)	117.86(10)
C(2)-C(1)	1.4929(16)	O(3)-N(3)-O(4)	124.90(8)
C(2)-C(3)	1.4971(15)	O(3)-N(3)-C(5)	117.30(8)
O(3)-N(3)	1.2277(11)	O(4)-N(3)-C(5)	117.75(8)
N(3)-C(5)	1.4725(12)	C(9)-C(4)-C(5)	118.36(9)
C(4)-C(9)	1.3937(13)	C(9)-C(4)-S(1)	119.30(7)
C(4)-C(5)	1.3972(13)	C(5)-C(4)-S(1)	122.25(7)
C(5)-C(6)	1.3836(13)	C(6)-C(5)-C(4)	122.59(9)
C(6)-C(7)	1.3954(14)	C(6)-C(5)-N(3)	116.71(8)
C(8)-C(7)	1.3916(15)	C(4)-C(5)-N(3)	120.65(8)
C(8)-C(9)	1.3958(14)	C(5)-C(6)-C(7)	118.33(9)
		C(7)-C(8)-C(9)	120.64(9)
O(2)-S(1)-O(1)	120.16(5)	C(4)-C(9)-C(8)	119.85(9)
O(2)-S(1)-N(2)	108.32(5)	C(8)-C(7)-C(6)	120.21(9)
O(1)-S(1)-N(2)	105.58(4)		
O(2)-S(1)-C(4)	106.84(5)	Symmetry transformation	ons used to generate ec
			•

quivalent atoms:

*N*-Isopropylidene-*N*'-2-Nitrobenzenesulfonyl Hydrazine. A Reagent for Conversion of Alcohols to the Corresponding Monoalkyl Diazenes. Mohammad Movassaghi\* and Omar K. Ahmad Page S10 / S27

	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	U <sup>12</sup>
<del>O(4)</del>	17(1)	23(1)	14(1)	2(1)	1(1)	2(1)
O(1)	17(1)	19(1)	12(1)	-1(1)	6(1)	-1(1)
O(2)	16(1)	17(1)	24(1)	-1(1)	7(1)	5(1)
S(1)	12(1)	13(1)	13(1)	-1(1)	5(1)	0(1)
N(2)	15(1)	14(1)	12(1)	1(1)	2(1)	-2(1)
N(1)	26(1)	19(1)	13(1)	2(1)	1(1)	-3(1)
C(2)	27(1)	19(1)	15(1)	1(1)	3(1)	-3(1)
C(1)	67(1)	18(1)	20(1)	-1(1)	3(1)	-3(1)
C(3)	72(1)	29(1)	14(1)	2(1)	-2(1)	-10(1)
O(3)	16(1)	14(1)	22(1)	1(1)	7(1)	-2(1)
N(3)	12(1)	15(1)	14(1)	2(1)	5(1)	2(1)
C(4)	13(1)	13(1)	13(1)	-1(1)	5(1)	-1(1)
C(5)	14(1)	12(1)	12(1)	1(1)	4(1)	-1(1)
C(6)	13(1)	17(1)	17(1)	0(1)	5(1)	-2(1)
C(8)	23(1)	16(1)	18(1)	3(1)	7(1)	-2(1)
C(9)	17(1)	14(1)	16(1)	2(1)	4(1)	1(1)
C(7)	18(1)	19(1)	18(1)	1(1)	7(1)	-5(1)

**Table S4.** Anisotropic displacement parameters (Å<sup>2</sup>x 10<sup>3</sup>)for IPNBSH. The anisotropic displacement factor exponent takes the form:  $-2\pi^2$ [  $h^2a^{*2}U^{11} + ... + 2h$  k a\* b\* U<sup>12</sup> ]

Table S5. Hydrogen coordinates (x 10<sup>4</sup>) and isotropic displacement parameters (Å<sup>2</sup>x 10<sup>3</sup>) for IPNBSH.

	Х	У	Z	U(eq)
	9188(18)	134(8)	8292(15)	17
H(1A)	9116	-978	6588	56
H(1B)	7356	-1211	5395	56
H(1C)	7333	-895	6938	56
H(3A)	7065	913	3958	63
H(3B)	5930	36	3757	63
H(3C)	7850	-8	3665	63
H(6)	15489	1304	8713	19
H(8)	12514	3171	6186	22
H(9)	10183	2751	6958	19
H(7)	15154	2457	7053	22

Table S6. Hydrogen bonds for IPNBSH [Å and °].

D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)
N(2)-H(2)O(3)	0.851(12)	2.518(13)	3.0751(11)	123.9(12)
N(2)-H(2)O(1)#1	0.851(12)	2.374(13)	3.1702(11)	156.0(13)

Symmetry transformations used to generate equivalent atoms: #1 -x+2,-y,-z+2





















S19









![](_page_23_Figure_0.jpeg)

S24

![](_page_24_Figure_0.jpeg)

![](_page_25_Figure_0.jpeg)

![](_page_26_Figure_0.jpeg)

S27