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Rapid Total Syntheses Utilizing "Supersilyl" Chemistry**

Brian J. Albert, Yousuke Yamaoka, and Hisashi Yamamoto*

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General Techniques.

All non-aqueous reactions were carried out in flame-dried glassware under an atmosphere of dry nitrogen or argon and stirred via magnetic stir-plates. All reactions were carried out with anhydrous solvents unless otherwise noted. Anhydrous THF, CH_2Cl_2 , Et_2O , and hexane were dried with an M BRAUN solvent purification system (A2 Alumina). Anhydrous DMF was purchased from Aldrich and used without further purification. Yields refer to chromatographically and spectroscopically (¹H NMR) homogeneous materials, unless otherwise stated. Silyl enol ethers **1**,¹ **8**,² and **14**³ were prepared according to the literature procedures. Dimethylaluminum triflimide (Tf₂NAIMe₂) was prepared according to the literature method from Me₃Al and Tf₂NH.⁴ The ruthenium metathesis catalyst, Zhan catalyst 1B, was purchased from Strem.

All reactions were monitored by thin–layer chromatography (TLC) carried out on 0.25 mm EMD silica gel plates (60F–254) using UV light (254 nm) with 2.4% phosphomolybdic acid/1.4% phosphoric acid/5% sulfuric acid in water, cerium sulfate in aqueous sulfuric acid, or anisaldehyde in ethanol and heat as developing agents. TSI silica gel (230–400 mesh) was used for flash chromatography.

Infrared spectra were recorded as thin films on sodium chloride plates using a Nicolet 20 SXB FTIR. ¹H NMR and ¹³C spectra were recorded on a Bruker Avance 400 or a Bruker Avance 500. Chemical shift values (δ) are reported in ppm and calibrated to the residual solvent peak (CDCl₃ δ 7.26 ppm for ¹H, 77.0 ppm for ¹³C; C₆D₆ δ 7.16 ppm for ¹H, 128.0 ppm for ¹³C; CD₃OD δ 3.31 ppm for ¹H, 49.0 ppm for ¹³C). The following abbreviations are used to indicate the multiplicities; s, singlet; d, doublet; t, triplet; q, quartet; quint, quintet; m, multiplet; br, broad; app, apparent.

Abbreviations: http://pubs.acs.org/paragonplus/submission/joceah_joceah_abbreviations.pdf

¹ Boxer, M. B.; Yamamoto, H. *Nature Protocols* **2006**, *1*, 2434.

² Boxer, M. B.; Akakura, M.; Yamamoto, H. *J. Am. Chem. Soc.* **2008**, *130*, 1580.

³ Yamaoka, Y.; Yamamoto, H *J. Am. Chem. Soc.* **2010**, *13*2, 5354.

⁴ Marx, A.; Yamamoto, H. Angew. Chem. Int. Ed. **2000**, 39, 178.



Preparation of 7 from 1-tetradecanol: The preparation of tetradecanal was accomplished by the IBX induced oxidation of commercially available 1-tetradecanol (our yield = 7.22 g, 85%), according to the literature.⁵

A stirred solution of **1** (4.92 g, 16.9 mmol) and tetradecanal (1.48 mg, 6.97 mmol) in CH₂Cl₂ (40 mL) was cooled to -40 °C, and then to the mixture was added Tf₂NAIMe₂ (0.10 M in CH₂Cl₂, 140 μ L, 14 μ mol) over a period of 2 min. The reaction mixture was stirred for 30 min at the same temperature then was quenched by the addition of pH 7 buffer (50 mL). The layers were separated, and the aqueous phase was extracted with CH₂Cl₂ then hexanes (10 mL each). The combined organic layers were dried over Na₂SO₄, filtered through cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (250 mL) eluting with CH₂Cl₂/hexanes (1:9 \rightarrow 1:4) to give **7** (4.68 g, 85%) as a colorless viscous oil.

Data for 7: $R_f = 0.31$ (1:4 CH_2Cl_2 /hexanes); IR (neat): 2926, 2855, 1727 (C=O), 1467, 1395, 1245, 1084, 836 cm⁻¹; ¹H NMR (500 MHz, CDCl₃, 293K) δ 9.77 (dd, 1H, J = 4.0, 1.1 Hz), 4.14–4.08 (m, 1H), 3.39 (app tt, 1H, J = 8.5, 4.3 Hz), 2.57 (ddd, 1H, J = 15.2, 4.7, 1.1 Hz), 2.36 (ddd, 1H, J = 15.2, 6.8, 4.0 Hz), 1.74 (ddd, 1H, J = 14.0, 9.0, 5.5 Hz), 1.64–1.56 (m, 1H), 1.47 (ddd, 1H, J = 13.2, 9.3, 4.0 Hz), 1.33–1.23 (m, 23H), 0.88 (t, 3H, J = 6.8 Hz), 0.19 (s, 27H), 0.18 (s, 27H); ¹³C NMR (126 MHz, CDCl₃, 293K) δ 202.4, 73.4, 69.9, 49.5, 43.6, 37.9, 31.9, 29.9, 29.7 (3 carbons), 29.6 (2 carbons), 29.5, 29.4, 25.3, 22.7, 14.1, 0.7, 0.5; LRMS (APCI+) C₂₇H₆₁O₂Si₄ [M – OSi(TMS)₃]⁺ 409.3 (26%).



Preparation of 9 from 7: A stirred solution of aldehyde **7** (1.57 g, 1.98 mmol), silvl enol ether **8** (1.16 g, 3.81 mmol), and 1-iodo-2-phenylacetylene (43.1 mg, 0.189 mmol) in CH₂Cl₂ (10 mL) was cooled to -40 °C, and then to the mixture was added Tf₂NH (0.010 M in CH₂Cl₂, 600 µL, 6.0 µmol). The reaction mixture was stirred for 1 h at the same temperature then was quenched by the addition of pH 7 buffer (10 mL). The layers were separated, and the aqueous phase was extracted with CH₂Cl₂ (5 mL). The combined organic layers were dried over Na₂SO₄, filtered through cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography⁶ on silica gel (150 mL)

⁵ Wiseman, J. M.; McDonald, F. E.; Liotta, D. C. Org. Lett. 2005, 7, 3155.

⁶ This diasteromer was separated from its minor diastereomers (major R_f = 0.39), with 1 or 2 chromatographies.

eluting with CH_2Cl_2 /hexanes (1:9 \rightarrow 1:4) to give **9** (1.71 g, 78% along with 19% of its diastereomers) as a colorless viscous oil.

Data for 9: $R_f = 0.29$ (1:4 CH₂Cl₂/hexanes); IR (neat): 2947, 1724 (C=O), 1438, 1379, 1244, 1070, 1005, 836, 689 cm⁻¹; ¹H NMR (500 MHz, CDCl₃, 293K) δ 4.12 (br app tt, 1H, J = 8.7, 4.8 Hz), 3.67 (app tt, 1H, J = 9.3, 4.7 Hz), 3.42 (br app tt, 1H, J = 8.6, 4.3 Hz), 2.51 (dd, 1H, J = 14.8, 8.6 Hz), 2.41 (dd, 1H, J = 14.8, 4.4 Hz), 2.12 (s, 3H), 1.78 (ddd, 1H, J = 12.9, 9.2, 5.7 Hz), 1.68 (ddd, 1H, J = 12.9, 9.4, 5.9 Hz), 1.64–1.56 (m, 1H), 1.37 (ddd 1H, J = 13.0, 8.4, 4.7 Hz), 1.35–1.19 (m, 24H), 0.88 (t, 3H, J = 6.9 Hz), 0.193 (s, 27H), 0.181 (s, 27H), 0.177 (s, 27H); ¹³C NMR (126 MHz, CDCl₃, 303K) δ 206.3, 73.7, 70.3, 69.7, 50.3, 45.5, 44.9, 38.3, 32.0, 31.5, 29.9, 29.6 (5 carbons), 29.5, 29.4, 26.0, 22.7, 14.1, 0.85, 0.79, 0.71; LRMS (APCI+) C₃₉H₉₃O₃Si₈ [M - OSi(TMS)₃]⁺ 833.4 (9%), C₃₀H₆₅O₂Si₄ [M - OSi(TMS)₃ - HOSi(TMS)₃]⁺ 569.4 (38%).



Preparation of tetraol S1-*syn/syn* from 9 for the determination of the relative stereochemistry: A stirred solution of 9 (220 mg, 0.200 mmol) in THF (1.0 mL) was cooled to -78 °C, then to this mixture was added MeMgBr (3.0 M in Et₂O, 350 µL, 0.405 mmol) dropwise. After 30 min the reaction was warmed to 0 °C, and after another 60 min at the same temperature, the reaction was warmed to 23 °C. After 75 additional min the reaction mixture was quenched by the addition of sat. aq. NH₄Cl (4 mL). The layers were separated, and the aqueous phase was extracted with hexanes (2 × 3 mL). The combined organic layers were dried over Na₂SO₄, filtered through cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (10 mL) eluting with CH₂Cl₂/hexanes (1:9 \rightarrow 1:2) to give the alcohol (155 mg, 70%) as a colorless oil.

A stirred solution of this alcohol (151 mg, 0.135 mmol) in THF (0.80 mL) and MeOH (80 μ L) was cooled to 0 °C, then to this mixture was added TBAF (1.0 M in THF, 200 μ L, 0.20 mmol) dropwise over 3 min. After an additional 2 min at the same temperature, the reaction mixture was warmed to 23 °C. After an additional 40 min at the same temperature, AcOH (50 μ L) was added and the resulting solution was concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (20 mL) eluting with *i*-PrOH/hexanes (1:19 \rightarrow 1:7) to give **S1-syn/syn** (51 mg, quant.) as a colorless viscous oil.

Data for S1-*syn/syn*:[†] R_f = 0.38 (1:4 *i*-PrOH/hexanes); IR (neat): 3370 (br, O-H), 2919, 2850, 1467, 1325, 1115, 1089, 847 cm⁻¹; ¹H NMR (500 MHz, 5% CDCl₃ in CD₃OD, 293K) δ 4.15 (app tt, 1H, J = 7.5, 5.1 Hz), 3.97 (app tt, 1H, J = 8.3, 4.4 Hz), 3.77–3.71 (m, 1H), 1.64–1.50 (m, 5H), 1.48–1.38 (m, 3H), 1.34–1.24 (m, 25H), 1.23 (s, 3H), 0.88 (t, 3H, J = 6.9 Hz); ¹³C NMR (126 MHz, 5% CDCl₃ in CD₃OD, 293K) δ 71.9, 71.5, 70.4, 69.0, 46.1, 44.8, 38.6, 32.9, 30.9, 30.6 (8 carbons), 30.3, 28.7, 26.3, 23.5, 14.4; LRMS (APCI–) C₂₂H₄₆³⁵ClO₄ [M + ³⁵Cl]⁻ 409.3 (100%), C₂₂H₄₆³⁷ClO₄ [M + ³⁵Cl]⁻ 411.2 (37%).

[†] This compound was confirmed to be *syn/syn* by the Kishi method for 1,3,5,7-tetraols (the 3- and 5- carbon chemical shifts are ~69 and ~70 ppm, then a *syn/syn* relationship exists).⁷

Preparation of 10. See *Angew. Chem. Int. Ed.* **2008**, 47, 7520 for the epoxidation reaction. Then see *J. Am. Chem. Soc.* **1990**, *112*, 5583 for the *para*-methoxybenzyl protection reaction (our yield = 90%).



Preparation of S2. A solution of epoxide **10** (1.11 g, 5.02 mmol) in DMSO (10 mL) and H₂O (10 mL) was stirred at 23 °C under an air atmosphere. To this solution was added KOH (1.12 g, 20.0 mmol), and after 5 min at the same temperature, the reaction mixture was warmed to 75 °C. After an additional 3 h at the same temperature, the reaction mixture was cooled to 23 °C, and it was subsequently poured onto aqueous HCI solution (0.1 N, 200 mL). The resulting mixture was extracted with EtOAc (6 × 25 mL). The combined extracts were dried over Na₂SO₄, filtered though cotton, and concentrated under reduced pressure. The resulting residue was purified by flash chromatography on silica gel (25 mL) eluting with EtOAc/hexanes (1:4 \rightarrow 3:2) to afford **S2** as a colorless oil (1.11 g, 93%).

Data for S2: $R_f = 0.13$ (1:1 EtOAc/hexanes); $[\alpha]_D^{23} = +36.1$ (*c* 1.44, MeOH); IR (neat): 3407 (br, O-H), 3075, 2934, 2871, 1613, 1514, 1248, 1035, 821 cm⁻¹; ¹H NMR (500 MHz, 293K, CD₃OD) δ 7.25 (br d, 1H, *J* = 8.8 Hz), 6.88 (br d, 1H, *J* = 8.8 Hz), 5.85 (ddd, 1H, *J* = 17.6, 10.4, 8.0 Hz), 5.36 (d, 1H, *J* = 10.4 Hz), 5.32 (d, 1H, *J* = 17.6 Hz), 4.52 (d, 1H, *J* = 11.3 Hz), 4.32 (d, 1H, *J* = 11.3 Hz), 3.82–3.76 (m, 4H), 3.66–3.61 (m, 2H), 3.51 (dd, 1H, 12.2, 7.8 Hz); ¹³C NMR (126 MHz, 293K, CD₃OD) δ 160.7, 136.8, 131.7, 130.6, 119.9, 114.7, 82.5, 75.2, 71.2, 64.3, 55.7; LRMS (APCI+) C₁₃H₁₇O₄ [M - H]⁺ 237.1 (100%), C₈H₉O [PMB]⁺ 121.1 (68%).

⁷ Kobayashi, Y.; Tan, C.-H.; Kishi, Y. Helv. Chim. Acta 2000, 83, 2562.



Preparation of 11. A solution of diol **S2** (360 mg, 1.50 mmol) in THF (7.5 mL) and H₂O (10 mL) was stirred at 0 °C under an air atmosphere. To this solution was added NalO₄ (390 mg, 1.82 mmol), and after 5 min at the same temperature, the reaction mixture was warmed to 23 °C. After an additional 1 h at the same temperature, the reaction mixture was diluted with H₂O (25 mL). The resulting mixture was extracted with Et₂O/heanes (9:1 v/v, 4 × 10 mL). The combined extracts were dried over Na₂SO₄, filtered though cotton, and concentrated under reduced pressure. The resulting crude **11**, unstable to flash chromatography, was obtained colorless oil (294 mg, 95%) and used without further purification.

Data for 11: $R_f = 0.34 \rightarrow 0.50$ (1:9 EtOAc/hexanes); $[α]_D^{23} = +13.2$ (*c* 1.26, CH₂Cl₂); IR (neat): 3001, 2936, 2837, 1736 (C=O), 1613, 1514, 1249, 1076, 1034, 934, 822 cm⁻¹; ¹H NMR (500 MHz, 293K, CDCl₃) δ *J* = 9.53 (d, 1H, *J* = 1.0 Hz), 7.29 (ddd, 1H, *J* = 9.5, 3.0, 2.0 Hz), 6.90 (ddd, 1H, *J* = 9.5, 3.0, 2.0 Hz), 5.76 (ddd, 1H, *J* = 17.3, 10.5, 6.5 Hz), 5.49 (ddd, 1H, *J* = 17.3, 1.5, 1.5 Hz), 5.45 (ddd, 1H, *J* = 10.5, 1.5, 1.3 Hz), 4.63 (d, 1H, *J* = 11.5 Hz), 4.52 (d, 1H, *J* = 11.5 Hz), 4.27 (app dq, 1H, *J* = 6.5, 1.5 Hz), 3.81 (s, 3H); ¹³C NMR (126 MHz, 293K, CDCl₃) δ 200.0, 159.5, 130.8, 129.7, 129.0, 120.7, 113.9, 84.4, 71.1, 55.3; LRMS (APCI+) C₁₂H₁₃O₃ [M − H]⁺ 205.1 (5%), C₈H₉O [PMB]⁺ 121.1 (100%).

($TMS_{3}SiQ \qquad \bigcirc \qquad $			РМВ
Entry	Solvent(s)	Temp. (°C)	Yield (%) of 12	dr
1	DMF	-65	6	48:44:6:2
2	DMF/THF (5:1)	-65	10	48:44:6:2
3	THF	-78	56	47:38:12:3
4	Et ₂ O/DMF (19:1)	-78	43	47:40:10:3
5	<i>t</i> -BuOMe/DMF (19:1)	-78	36	47:40:10:3
6	CH ₂ Cl ₂ /DMF (19:1)	-78	29	48:43:7:2
7	toluene/DMF (19:1)	-78	63	48:43:7:2
8	CyMe/DMF (19:1)	-78	50	48:44:6:2

Table S1. Optimization of the aldol reaction of 9 and 11.



Preparation of 12 (Table S1, entry 7): A solution of ketone **9** (221 mg, 0.201 mmol) in toluene (1.6 mL) was stirred at -40 °C, and to this solution was added LiHMDS (1.0M in hexanes, 240 μ L, 0.240 mmol). After 10 min at the same temperature, the reaction mixture was warmed to 0 °C. After an additional 2 h at the same temperature, the reaction was cooled to -78 °C, then to it was added DMF (100 μ L, 1.29 mmol) in toluene (150 μ L, then 50 μ L rinse) down the flask side. After an additional 10 min, a solution of **11** in toluene (150 μ L, then 50 μ L rinse) was added down the flask side. After an additional 1 h at -78 °C, the mixture was quenched with sat. aq. NH₄Cl (5 mL). The resulting mixture was extracted with CH₂Cl₂ (3 mL). The combined organic layers were dried over Na₂SO₄, filtered though cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (150 mL) eluting with CH₂Cl₂/hexanes/EtOAc (10:90:0 \rightarrow 20:78:2) to give **12** (166 mg, 63% as a 48:43:7:2 mixture of diastereomers) as a colorless viscous oil and returned **9** (70.0 mg, 32%).



Preparation of EBC-23 from ketoalcohol 12: A solution of ketoalcohol **12** (261 mg, 0.200 mmol) in THF (0.80 mL) was stirred at -78 °C, and to this solution was added LiHMDS (1.0M in hexanes, 220 μ L, 0.220 mmol). After 30 min at the same temperature, acryloyl chloride (24 μ L, 0.30 mmol) was added dropwise. After an additional 30 min at the same temperature, the reaction was warmed to ambient temperature and concentrated under high vacuum (4 mm Hg).

The residue was dissolved in toluene (5.0 mL), and to the solution was added Zhan-1B (2.9 mg, 4.0 μ mol) in one portion. The mixture was then heated to 95 °C, and after 15 and 26 h more Zhan-1B was added (2.9 and 3.0 mg = 8.8 mg, 12 μ mol total). After a total of 48 h at 95 °C, the mixture was cooled to ambient temperature and concentrated under high vacuum (4 mm Hg).

The residue was dissolved in THF (2.0 mL), and the stirred solution was cooled to 0 °C. To the mixture was added pyridine (50 μ L, 0.62 mmol) followed by HF•py (70% v/v HF, 200 μ L). After 2 h at the same temperature, the reaction was warmed to 23 °C. After an additional 2 h at the same temperature, the reaction was quenched by the addition of sat. aq. NaHCO₃ (10 mL). The resulting

mixture was extracted with Et₂O (4 × 5 mL). The combined extracts were dried over Na₂SO₄, filtered though cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (10 mL) eluting with hexanes/EtOAc (2:3 \rightarrow 1:9) to give a mixture of hemiketal anomers **S3** (39.5 mg, 33%).

Hemiketals **S3** (27.9 mg, 0.0472 mmol) in CH₂Cl₂ (380 µL) and H₂O (95 µL) was rapidly stirred at 0 °C under an atmosphere of air, and to this mixture was added 2,3-dichloro-5,6-dicyanobenzoquinone (24.5 mg, 0.108 mmol). After 5 min at the same temperature, the reaction was warmed to 23 °C. After an addition 2.5 h at this temperature, the mixture was diluted with H₂O (5 mL) and sat. aq. NaHCO₃ (5 mL). The resulting mixture was extracted with Et₂O (4 × 5 mL). The combined extracts were dried over Na₂SO₄, filtered though cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (6 mL) eluting with hexanes/EtOAc (2:3→1:9) to give **EBC-23** (15.4 mg, 72%) as a white powder.

Data for EBC-23: $R_f = 0.39$ (3:2 CH₂Cl₂/EtOAc), 0.31 (1:4 hexanes/EtOAc); $[\alpha]_D^{24}$ +15.4 (*c* 0.20, CHCl₃); IR (neat): 3524, 3479, 2918, 2852, 1727 (C=O), 1470, 1280, 1243, 1103, 1067 cm⁻¹; LRMS (APCI-) $C_{26}H_{44}^{35}CIO_6$ [M + ${}^{35}CI$]⁺ 487.1 (100%), $C_{26}H_{44}^{37}CIO_6$ [M + ${}^{37}CI$]⁺ 489.1 (40%).

Natural EBC-23 ¹ H NMR Data: 750 MHz, CDCl ₃	Williams' ¹ H NMR Data: 500 MHz, CDCl ₃	Our ¹ H NMR Data: 500 MHz, CDCl ₃
6.89 (dd, 1H, <i>J</i> = 10.0, 5.2 Hz)	6.87–6.90 (dd, 1H, <i>J</i> = 9.9, 5.1 Hz)	6.91 (dd, 1H, <i>J</i> = 9.9, 5.2 Hz)
6.21 (d, 1H, <i>J</i> = 10.0 Hz)	6.22–6.20 (dd, 1H, <i>J</i> = 9.9 Hz)	6.23 (d, 1H, <i>J</i> = 9.9 Hz)
5.04 (ddd, 1H, <i>J</i> = 6.9, 4.5, 2.5 Hz)	5.04–5.02 (ddd, 1H, <i>J</i> = 6.9, 4.5, 2.6 Hz)	5.06 (ddd, 1H, <i>J</i> = 6.9, 4.6, 2.5 Hz)
4.51 (dd, 1H, <i>J</i> = 5.2, 4.5 Hz)	4.51 –4.49 (t, 1H, <i>J</i> = 4.8 Hz)	4.52 (app t, 1H, $J = 4.9$ Hz)
4.40-4.35 (m, 1H)	4.39–4.34 (m, 1H)	4.43-4.37 (m, 1H)
4.13-4.09 (m, 1H)	4.11 (m, 1H)	4.16–4.10 (m, 1H)
3.82–3.76 (m, 1H)	3.81-3.76 (m, 1H)	3.84-3.78 (m, 1H)
3.05 (d, 1H, J = 9.2 Hz)	3.05 (br s, 1H)	3.08 (br s, 1H)
2.54 (dd, 1H, <i>J</i> = 14.9, 6.9 Hz)	2.56–2.51 (dd, 1H, <i>J</i> = 14.9, 6.8 Hz)	2.56 (dd, 1H, <i>J</i> = 14.9, 6.9 Hz)
2.30 (dd, 1H, <i>J</i> = 14.9, 2.5 Hz)	2.31–2.27 (dd, 1H, <i>J</i> = 15.0, 2.6 Hz)	2.32 (dd 1H, <i>J</i> = 14.9, 2.5 Hz)
2.05-1.98 (m, 2H)	2.05-1.97 (m, 2H)	2.08-1.99 (m, 2H)
1.80-1.75 (m, 1H)	1.79–1.75 (m, 1H)	1.82–1.76 (m, 1H)
1.65-1.56 (m, 2H)	1.65–1.56 (m, 3H)	1.68-1.58 (m, 2H)
1.52-1.47 (m, 1H)	1.52–1.47 (m, 2H)	1.58-1.35 (m, 3H)
1.47-1.42 (m, 1H)		
1.42-1.35 (m, 2H)		
1.33-1.20 (m, 21H)	1.23 (s, 23H)	1.32–1.20 (m, 23H)
0.86 (t, 3H, J = 7.1 Hz)	0.84 (t, 3H, $J = 14.0^8$ Hz)	0.88 (t, 3H, J = 7.0 Hz)

Table S2. Comparison of the isolated, Williams' and our ¹H NMR data for EBC-23.

⁸ This J value is assumed to be a typographic error. Their data for the isolated material indicated a J value of 7.1 Hz.

Natural EBC-23 ¹³ C NMR Data:	Williams' ¹³ C NMR Data:	Our ¹³ C NMR Data:
125 MHZ, CDCl ₃ 161.0	125 MHZ, CDCI ₃ 161.0	126 MHZ, CDCl ₃ 161.0
138.6	138.6	138.6
124.6	124.6	124.6
106.6	106.6	106.6
78.8	78.8	78.8
71.8	71.8	71.9
68.9	68.9	68.9
67.6	67.8	67.8
64.2	64.2	64.2
47.7	47.7	47.7
42.2	42.2	42.2
38.8	38.7	38.7
37.7	37.7	37.7
31.9	31.9	31.9
29.67	29.67	29.68
29.65	29.66	29.67
29.63	29.64	29.65
29.61	29.62	29.62
29.58	29.59	29.60
29.3	29.3	29.3
25.4	25.4	25.4
22.7	22.7	22.7
14.1	14.1	14.1

Table S3. Comparison of the isolated, Williams' and our ¹³C NMR data for EBC-23.



Preparation of 15 from 3-butenal: The preparation of 3-butenal was prepared from glyoxal according to the literature procedure.⁹

A stirred solution of 3-butenal (40.0 g, 15.0 mmol, 1/37 w/w in CH_2Cl_2) and silyl enol ether **14** (5.27 g, 12.2 mmol) in CH_2Cl_2 (120 mL) was cooled to -78 °C, and then to the mixture was added Tf_2NAIMe_2 (0.10 M in CH_2Cl_2 , 610 µL, 0.061 µmol). The reaction mixture was stirred for 12 h at 0 °C then was quenched by the addition of pH 7 buffer (10 mL). The layers were separated, and the aqueous phase was extracted with CH_2Cl_2 (20 mL). The combined organic layers were dried over Na_2SO_4 , filtered through cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (100 mL) eluting with CH_2Cl_2 /hexanes (1:7 \rightarrow 1:5) to give **15** (4.57 g, 75%) as a colorless semisolid.

Data for 15: $R_f = 0.42$ (1:4 CH_2Cl_2 /hexanes); IR (neat): 2952, 2909, 2875. 1717, 1416 cm⁻¹; ¹H NMR (500 MHz, CDCl₃, 293K) δ 5.79–5.66 (m, 1H), 5.08–4.94 (m, 2H), 3.92–3.85 (m, 1H), 2.61 (dd, J = 16.4, 8.0 Hz), 2.47 (dd, J = 16.4, 4.4 Hz), 2.32–2.18 (m, 2H), 1.04 (t, 27H, J = 7.9 Hz), 0.77 (q, 18H, J = 7.9 Hz); ¹³C NMR (126 MHz, CDCl₃, 293K) δ 207.2, 134.1, 117.7, 72.2, 49.5, 41.0, 31.3, 8.8, 5.4; LRMS (ES+) $C_{25}H_{56}O_2Si_4$ [M + Na]⁺ 523.5 (100%).



Preparation of 16: A stirred solution of silyl enol ether **1** (2.90 g, 10.0 mmol), *n*-hexanal (240 μ L, 2.00 mmol) and 1-iodo-2-phenylacetylene (45.3 mg, 0.199 mmol) in CH₂Cl₂ (10.0 mL) was cooled to -40 °C, and then to the mixture was added Tf₂NH (0.010 M in CH₂Cl₂, 400 μ L, 4.0 μ mol). The reaction mixture was stirred for 30 min at the same temperature then was quenched by the addition of pH 7 buffer (5 mL). The layers were separated, and the aqueous phase was extracted with CH₂Cl₂ (5 mL). The combined organic layers were dried over Na₂SO₄, filtered through cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (100 mL) eluting with CH₂Cl₂/hexanes (1:20 \rightarrow 1:7) to give **16** (1.66 g, 85%, dr = 80:12:5:3) as a white foam.

Data for 16: $R_f = 0.52$ (1:4 CH_2Cl_2 /hexanes); IR (neat): 2949, 2894, 1733 (C=O), 1438, 1375 cm⁻¹; ¹H NMR (500 MHz, CDCl₃, 293K) δ 9.94 (dd, 1H, J = 2.9, 1.3 Hz), 4.49 (m, 1H), 3.95 (m, 1H), 3.67 (m, 1H), 2.71 (ddd, 1H, J = 15.4, 3.9, 1.2 Hz), 2.63 (ddd, 1H, J = 15.3, 8.7, 3.1 Hz), 2.06 (ddd, 1H, J = 12.8,

⁹ Crimmins, M. T.; Kirincich, S. J.; Wells A. J.; Choy, A. L. Synth. Commun. **1998**, 28, 3675.

9.7, 5.4 Hz), 1.98 (ddd, 1H, J = 12.9, 9.9, 5.6 Hz), 1.93–1.84 (m, 1H), 1.74 (ddd, 1H, J = 13.2, 9.0, 4.5 Hz), 1.60 (ddd, 1H, J = 12.5, 8.9, 3.4 Hz), 1.50–1.36 (m, 7H), 0.94 (t, 3H, J = 6.9 Hz), 0.33 (s, 27H), 0.324 (s, 27H), 0.318 (s, 27H); ¹³C NMR (126 MHz, CDCl₃, 293K) δ 199.8, 73.9, 70.17, 70.15, 50.5, 46.1, 45.0, 38.9, 32.4, 26.2, 23.0, 14.1, 1.0, 0.9, 0.8; LRMS (ES+) C₃₉H₁₀₂NaO₄Si₁₂ [M + Na]⁺ 993.7 (7%), C₃₉H₁₀₃NaO₄Si₁₂ [M + H]⁺ 971.7 (10%).



Preparation of 17. To a stirred solution of ketone **15** (446 mg, 0.890 mmol) and LiBF₄ (418 mg, 4.45 mmol) in DMF (9.0 mL) was added LiHMDS (1.0 M in hexane, 1.1 mL, 1.1 mmol) at -40 °C. After being stirred at same temperature for 30 min, the solution was cooled to -60 °C. After an additional 5 min at the same temperature, the toluene solution (1 mL) of the aldehyde **16** (1.73 g, 1.78 mmol, dr = 87:13) was slowly added. The reaction mixture was stirred for 1 h at the same temperature then was quenched by the addition of water (1 mL). The reaction mixture was diluted with hexanes (20 mL) and the organic phase was washed with brine (2 × 10 mL). The combined organic layers were dried over Na₂SO₄, filtered through cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (100 mL) eluting with CH₂Cl₂/hexanes (1:7 \rightarrow 1:5) to give **17** (506 mg, 39%) as a white foam and other diastereomers (340 mg, 25%, R_f ~ 0.45, dr ~ 73:20:7).¹⁰

Data for 17: $R_f = 0.27$ (1:4 $CH_2Cl_2/hexanes$); IR (neat): 2950, 2875, 1716, 1461, 1244, 836 cm⁻¹; ¹H NMR (500 MHz, C_6D_6 , 293K) δ 6.00–5.87 (m, 1H), 5.15–5.05 (m, 2H), 4.53–4.43 (m, 1H), 4.26-4.14 (m, 2H), 4.02–3.94 (m, 1H), 3.74–3.64 (m, 1H), 3.63 (br s, 1H), 2.90–2.77 (m, 1H), 2.75–2.62 (m, 2H), 2.55–2.62 (m, 3H), 2.00 (ddd, 1H, J = 12.6, 6.5 Hz), 1.92 (ddd, 1H, J = 12.7, 6.4 Hz), 1.87–1.77 (m, 3H), 1.73–1.65 (m, 1H), 1.64–1.57 (m, 1H), 1.49–1.36 (m, 7H), 1.16 (t, 27H, J = 7.8 Hz), 0.97–0.91 (m, 21H), 0.35 (s, 27H), 0.334 (s, 27H), 0.325 (s, 27H); ¹³C NMR (126 MHz, C_6D_6 , 338K) δ 207.1, 134.7, 117.8, 74.2 (2 carbons), 72.6, 70.4, 67.5, 52.1, 50.2, 46.1, 45.7, 43.9, 41.7, 38.8, 32.6, 26.1, 14.0, 9.1, 6.0, 1.2, 1.1 (2 carbons); LRMS (ES+) $C_{64}H_{158}NaO_6Si_{16}$ [M + Na]⁺ 1496.3 (5%), $C_{25}H_{56}O_2Si_4$ [15 + Na]⁺ 523.5 (100%).

¹⁰ Fortunately, the all-syn diastereomer **17** was easily separated from the mixture. Unfortunately, a more accurate diastereomeric ratio cannot be given due to the poor separation of the minor diastereomers. Since an excess of aldehyde was employed and 39% yield of **17** was obtained, a matched 1,3- and 1,5-selective reaction probably occurred.



Preparation of 18. To a stirred solution of NaBH₄ (36.0 mg, 0.952 mmol) in MeOH (1 mL) was added the THF (1 mL) solution of **17** (140 mg, 0.095 mmol) at -20 °C. After being stirred at same temperature for 12h, the reaction mixture was diluted with hexanes (5 mL). The combined organic layers were dried over Na₂SO₄, filtered through cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (10 mL) eluting with CH₂Cl₂/hexanes (1:7 \rightarrow 1:4) to give **18** (128 mg, 89%) as a white foam.

Data for 18: $R_f = 0.22$ (1:4 CH_2Cl_2 /hexanes); IR (neat): 2940, 2875, 1432, 1244, 835 cm⁻¹; ¹H NMR (500 MHz, C_6D_6 , 293K) δ 6.21–6.09 (m, 1H), 5.22–5.12 (m, 2H), 4.61–4.50 (m, 1H), 4.41–4.33 (m, 1H), 4.32–4.16 (m, 3H), 4.15–4.06 (m, 1H), 4.01–3.91 (m, 1H), 3.67–3.58 (m, 1H), 2.72–2.60 (m, 1H), 2.59–2.48 (m, 1H), 2.07-1.98 (m, 2H), 1.97–1.87 (m, 4H), 1.84–1.75 (m, 2H), 1.73–1.64 (m, 2H), 1.60–1.53 (m, 1H), 1.47–1.34 (m, 1H), 1.19 (t, 27H, J = 7.8 Hz), 1.02–0.94 (m, 21H), 0.32 (s, 81H); ¹³C NMR (126 MHz, C_6D_6 , 338K) δ 135.5, 116.9, 75.7, 74.4, 74.1, 72.9, 70.4, 69.5, 46.21, 46.19, 45.5, 44.4, 44.3, 40.7, 39.2, 32.5, 26.2, 23.0, 13.9, 9.1, 6.1, 1.13 (2 carbons), 1.06; LRMS (ES+) $C_{64}H_{160}NaO_6Si_{16}$ [M + Na]⁺ 1498.3 (10%), 345.5 (100%) .



Preparation of 13. To a stirred solution of **18** (100 mg, 0.0678 mmol) in MeOH/CH₂Cl₂ (4:1) (5.0 mL) was irradiated in a quartz flask at room temperature. After being stirred for 12h, the solvent was removed under reduced pressure. The crude mixture was used without purification in the next step.

To a stirred solution of NaH (56.0 mg, 1.34 mmol) in THF (1 mL) was a solution of the crude mixture (THF, 1 mL) at 0 °C. After being stirred at same temperature for 10 min, and then to the mixture was added MeI (89.3 μ L, 2.68 mmol). The reaction mixture was stirred for 12 h at room temperature then was quenched by the addition of water (1 mL). The layers were separated, and the aqueous phase was extracted with AcOEt (5 mL). The combined organic layers were dried over Na₂SO₄, filtered through cotton, and concentrated under reduced pressure. The resulting residue was purified by flask chromatography on silica gel (5 mL) eluting with AcOEt/hexanes (1:5 \rightarrow 1:3) to give **13** (18.3 mg, 61%) as a colorless oil.

Data for 13^{11,12}: $R_f = 0.31$ (1:3 AcOEt/hexanes); IR (neat): 2929, 2820, 1457, 1379, 1188, 1092 cm⁻¹; ¹H NMR (500 MHz, CDCl₃, 293K) δ 5.88–5.76 (m, 1H), 5.18–5.04 (m, 2H), 3.34–3.26 (m, 6H), 3.34 (s, 3H), 3.32 (s, 3H), 3.312 (s, 6H), 3.308 (s, 6H), 2.41–2.23 (m, 2H), 1.88–1.76 (m, 4H), 1.64–1.45 (m, 8H), 1.39–1.22 (m, 6H), 0.90 (t, 3H, J = 6.9 Hz); ¹³C NMR (126 MHz, CDCl₃, 293K) δ 134.4, 117.3, 77.9, 77.3, 75.4, 75.29, 75.28, 75.27, 56.4, 56.3, 56.24, 56.23, 56.21, 56.18, 38.16, 38.15, 38.1, 37.9, 37.64, 37.55, 33.4, 32.1, 24.6, 22.7, 14.1; LRMS (ES+) C₂₅H₅₀O₆ [M + Na]⁺ 469.5 (100%).

¹¹ Liu, K.; Arico, J. W.; Taylor, R. E. *J. Org. Chem.* **2010**, *75*, 3953; ¹H, ¹³C data reported in CDCl₃, which matches our data. ¹² Mori, Y.; Kohchi, Y.; Suzuki, M.; Carmeli, S.; Moore, R. E.; Paterson, G. M. *J. Org. Chem.* **1991**, *56*, 631; ¹H, ¹³C data reported in C₆D₆, which also matches our data (not shown).

¹H NMR for 7: CDCl₃, 500 MHz, 293K





¹H NMR for S1-*syn/syn*: 5% CDCl₃ in CD₃OD, 500 MHz, 293K











¹H NMR for 16: CDCI₃, 500 MHz, 293K





¹H NMR for 17: C₆D₆, 500 MHz, 338K



¹H NMR for 19: CDCl₃, 500 MHz, 293K

