Supporting Information

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SI Methods.

Expression, Purification, and Characterization of GPIb α -N. The fulllength human TPST-2 cDNA coexpressed with the cDNA coding for the human GPIb α fragment -2 to 290 with the Cys⁶⁵ \rightarrow Ala substitution was cloned into the Drosophila melanogaster expression vector pMT/BiP/V5-HisA (Invitrogen, Carlsbad, CA). Sitedirected mutagenesis for expressing mutant GPIba-N, including all Tyr \rightarrow Phe substitutions, was performed by polymerase chain reaction using oligonucleotide primers containing the desired mutation(s) (QuickChange Site-Directed Mutagenesis Kit, Stratagene, La Jolla, CA). All GPIbα-N species were purified by ion exchange and gel permeation chromatography. To demonstrate differences in sulfate content, duplicate 40 to 100 µg aliquots of each species were equilibrated with 0.05 M sodium acetate, pH 5.5; one was mixed with 60 µg of abalone sulfatase (type VIII; Sigma-Aldrich Co., St. Louis, MO) and both were placed in a heating block at 37 °C. After various incubation times, the reaction was stopped by placing the tubes on ice for 30 min followed by dilution in phosphate-buffered saline. Sulfate removal was analyzed by comparative ion exchange chromatography and isoelectric focusing; the latter was performed with 5 µg of protein in 5% polyacrylamide-2% ampholytes (pH 3-10) applying 100 V for 1 h, 200 V for 1 h, and 500 V for 30 min. Gels were then fixed and proteins stained with Coomassie Blue.

Crystallization of the Complex of Fully Sulfated GPIb α -N with FIIa. Human FIIa was treated with a 25-fold molar excess of PPACK (both from Haematologic Technologies) for eight hours at 22-25 °C and three days at 4 °C. Catalytically blocked FIIa was separated from free inhibitor by gel permeation chromatography (GPC) on a PD10 column (GE Healthcare, Buckinghamshire, United Kingdom) equilibrated with a buffer composed of 20 mM Hepes, 135 mM NaCl, pH 7.4 [Hepes-buffered saline, (HBS)]. Residual proteolytic activity was assessed from the clotting time of a 4 mg/mL fibrinogen solution mixed with 0.1 mg/mL P-FIIa in the presence of 2 mM CaCl₂; it was less than 1:1,000,000 of that of untreated FIIa. The complex of GPIba-N with three Tys residues and P-FIIa was obtained and crystallized as described (1). In brief, the two proteins were mixed at 1:1 molar ratio and the resulting complex was purified by GPC, concentrated to 20 mg/mL in HBS, and crystallized by the hanging drop technique mixing 1 µL of protein solution and 1 µL of reservoir solution containing 16% PEG 600 and 200 mM ammonium phosphate, pH 7.0. The crystals were first characterized using an R-axis 1V++ image plate detector (Rigaku Corp, Tokyo, Japan); diffraction data to 3.2 Å resolution were collected at the Stanford Synchrotron Radiation Lightsource on beam lines 9-2.

Preparation of Chimeric GPIb α **-N-Long.** To check if FIIa binding to surface-anchored GPIb α -N is affected by the spatial orientation of the molecule relative to the surface, we used a recombinant chimeric protein formed by residues -2 to 288 of GPIb α followed by 132 residues of the SV40 large T antigen; this molecule was designated GPIb α -N-Long. We generated a monoclonal antibody (LJ-3A2) that reacted specifically with the extended SV40 sequence and not with GPIb α -N; this was used in experiments based on receptor immobilization as an alternative to using LJ-P3 specific for an epitope located in the NH2-terminal region of GPIb α -N.

Evaluation of FIIa Binding to GPIb α -N by Complex Formation in Solution and by SPR. To measure the amount of soluble complex

formed, different GPIba-N species and P-FIIa were mixed in a 3:1 mass ratio and incubated at 37 °C for 30 min. The components of the mixtures were then separated by GPC using two high performance liquid chromatography (HPLC) Superdex columns in series, the first packed with S75 and the second with S200 (GE Healthcare). As a reference, corresponding amounts of the individual complex components were analyzed under the same conditions. Fractions were collected and analyzed for GPIba-N and P-FIIa content by reversed-phase HPLC, using a 0.2×11 cm column packed with 10 µm Poros RH-1 beads (Applied Biosystems, Foster, CA) and, as the eluant, a gradient starting with 0.1% trifluoroacetic acid in Milli-Q (Millford, MA) water and ending with 0.1% trifluoroacetic acid in acetonitrile (Fisher Optima, NJ); light absorbance of the effluent was measured at 215 nm and plotted against elution time. Known amounts of purified GPIbα-N and P-FIIa were analyzed individually using the same procedure, and the relation between known protein mass and area under the corresponding absorption peak provided the calibration to measure unknown quantities of the same protein. The fractions containing P-FIIa bound to GPIba-N or control P-FIIa not incubated with GPIba-N did not overlap; thus, the amount of complex formed was expressed as percent P-FIIa present in the former relative to the amount present in the latter. With the excess GPIb α -N used, typically >95% P-FIIa was in the complex fractions.

To generate SPR sensorgrams, GPIba-N was diluted in HBS containing 3 mM EDTA and 0.005% P20 surfactant (HBS-EP), and injected over one of the chip flow cells to which the antibody LJ-P3 was covalently bound. Injection at a flow rate of 40 μ L/ min continued until the desired ligand concentration at the chip surface (measured in Biacore Units, RU; 1 RU = 1×10^{-6} Refractive Index Units) was reached. P-FIIa diluted in HBS-EP (concentration range 3.125-10,000 nM) was then injected over the flow cell with bound GPIba and a control cell where only LJ-P3 was on the surface (nonspecific binding). Association was followed for 3 min, after which HBS-EP was injected for 5 min to assess dissociation. Two 1-min pulses of 3.2 mM NaOH, pH 11.5, were injected at 60 µL/min to detach LJ-P3-bound GPIb α -N before reusing the chip. Binding at saturation (B_{max}) and equilibrium dissociation constant (K_D , binding affinity) were obtained by nonlinear fitting of specific binding (ΔRU , the difference in P-FIIa RU on the surface with and without immobilized GPIba-N at equilibrium) as a function of P-FIIa concentration; a one-site model of interaction yielded consistently the best fitting (GraphPad Prism version 5.0; GraphPad Software, San Diego, CA). B_{max} was reported as molar ratio of bound P-FIIa and GPIba-N immobilized onto the chip, the latter calculated with the assumption that 1 RU measured by SPR corresponds to 1 pg of protein/mm² of surface. To derive the kinetics rate constants, we assumed that the dissociation phase followed an exponential model and derived the dissociation rate constant (off-rate, k_d) by nonlinear regression fitting of the experimental data using the equation:

$$B_t = \mathbf{NS} + (B_0 - \mathbf{NS}) \cdot e^{-k_d \cdot t}$$
 [S1]

Where t is time, B_t is observed binding at time = t, NS is nonspecific binding, and B_0 is total binding at time = 0 (before dissociation begins). Then, the association rate constant (on-rate, k_a) was obtained by nonlinear regression fitting of the association data using the equation:

$$B_t = B_{\mathrm{Eq}} \cdot (1 - e^{-k_{\mathrm{ob}} \cdot t})$$
 [S2]

Where B_t is binding at time = t, B_{Eq} is maximum (equilibrium) binding measured at any given analyte concentration, and k_{ob} (expressed in units of inverse time) is a constant expressing how quickly binding reaches equilibrium; then, k_a was derived from k_{ob} using the equation:

$$k_a = (k_{\rm ob} - k_d) / [\text{Analyte}]$$
 [S3]

Where k_d is the calculated dissociation rate constant.

Flla Binding to Platelets. Following an institutionally approved protocol, human blood was obtained from healthy donors who gave their informed consent according to the Declaration of Helsinki. Blood (five parts) collected in acid-citrate-dextrose (1 part; 71 mM citric acid, 85 mM sodium citrate, 111 mM dextrose, pH 4.5) was centrifuged at 600 g for 12 min to prepare platelet-rich plasma. A platelet pellet was then obtained by centrifuging platelet-rich plasma at 800 g for 15 min in the presence of 10 mM PG E_1 (Enzo Life sciences, Farmingdale, NY) and 0.6 U/mL Apyrase Grade VII (Sigma). The pellet was washed once by resuspension and centrifugation in modified Tyrode buffer, pH 6.5 (135 mM sodium chloride, 2.9 mM magnesium chloride, 1 mM sodium phosphate monobasic, 10 mM Hepes, 2.5 mM dextrose), and finally resuspended in the same buffer but at pH 7.4; the platelet count was adjusted to $2 \cdot 10^7$ /mL. For these assays, FIIa was blocked in the active site with biotin-PPACK (B-P-FIIa; Haematologic Technologies) and characterized as described above for P-FIIa. In initial assays, B-P-FIIa was added at the desired concentration (maximum 800 nM) into aliquots of the platelet suspension and incubated for the indicated time at room temperature (22–25 °C), followed by phycoerythrin-conjugated

- 1. Celikel R, et al. (2003) Modulation of α-thrombin function by distinct interactions with platelet glycoprotein lbα. *Science* 301:218–221.
- Handa M, Titani K, Holland LZ, Roberts JR, Ruggeri ZM (1986) The von Willebrand factor-binding domain of platelet membrane glycoprotein lb. Characterization by monoclonal antibodies and partial amino acid sequence analysis of proteolytic fragments. J Biol Chem 261:12579–12585.

streptavidin (S-PE; Invitrogen; 1.25 µg/mL) for 5 min; in subsequent experiments (see Fig. S5*C*), B-P-FIIa and S-PE were mixed at a fixed molar ratio (4:1) before addition to platelets. The two methods gave similar results, but the latter was more reproducible and was used for evaluating the time-course of binding. When indicated, the anti-GPIb α monoclonal antibody, LJ-Ib10 (2), which specifically inhibits FIIa binding (3), was added to the platelet suspension five minutes before the addition of S-PE/B-P-FIIa. Binding was measured by flow-cytometry, without sample dilution, using a FACS Calibur II equipped with a 488 nm argon laser and the software CellQuest for data evaluation (Becton Dickinson and Company, Franklin Lakes, NJ). Results were expressed as geometric mean of the fluorescence intensity of 10,000 events and analyzed with GraphPad Prism version 5.0.

Additional Materials Used in Supporting Experiments. In addition to DNA aptamers, we used other FIIa exosite inhibitors to evaluate their effect on binding to GPIb α . Lepirudin (recombinant [Leu¹-Thr²]-63-desulfohirudin; Refludan; Bayer Corp., Pharmaceutical Div., West Haven, CT) and hirugen (synthetic 3-O-sulfo-Tyr⁶³-hirudin [54–65]; Anaspec, Fremont, CA) were used to block exosite I; heparin (sodium salt, Grade II, from porcine intestinal mucosa; Sigma-Aldrich Co., St. Louis, MO) to block exosite II. Alexa Fluor 488-conjugated streptavidin was from Invitrogen, Carlsbad, CA.

Evaluation of Results Obtained with Inhibitors of FIIa Binding to GPIba. The inhibition constants (K_i) of FIIa exosite inhibitors were calculated according to the Cheng-Prusoff equation (4) from the corresponding IC₅₀ values derived from inhibition dose-response curves fitted with GraphPad Prism (version 5.0; GraphPad Software, San Diego, CA). Results were expressed as mean \pm SEM of at least two independent determinations.

- De Marco L, Mazzucato M, Masotti A, Fenton JW, II, Ruggeri ZM (1991) Function of glycoprotein Ibα in platelet activation induced by α-thrombin. J Biol Chem 266:23776–23783.
- Cheng Y, Prusoff WH (1973) Relationship between the inhibition constant (K_i) and the concentration of inhibitor which causes 50 per cent inhibition (i50) of an enzymatic reaction. *Biochem Pharmacol* 22:3099–3108.



Fig. S1. Exosite I-specific ligands fail to inhibit P-FIIa binding to soluble GPIb α -N. Top. Elution time from a GPC column of P-FIIa (peak 2); wild type GPIb α -N (peak 3); P-FIIa plus hirugen (8-fold molar excess; peak 4); P-FIIa plus GPIb α -N (peak 5); and hirugen plus P-FIIa plus GPIb α -N (peak 6). Mixtures yielding new peaks eluting earlier than individual components indicates complex formation in solution. Note that hirugen alone (peak 1) has no significant absorption under these experimental conditions. Bottom. The same as above, but using lepirudin (peak 1) instead of hirugen and FIIa instead of P-FIIa; lepirudin has a larger molecular mass than hirugen, thus the positional shift of peak 6 as compared to peak 5 is more evident.



Fig. 52. Inhibition of P-FIIa binding to immobilized GPIb α -N by hirugen and heparin. (*A*). SPR analysis of the inhibition of P-FIIa (used at K_D concentration, 80 nM) binding to immobilized wild type GPIb α -N peak 3 by hirugen; the calculated K_i was 178.6 ± 0.6 nM. (*B*). The same as in *A*, but using the exosite II inhibitor, heparin; the calculated K_i was 464.5 ± 3.9 nM. Results are expressed as percent binding relative to a control without inhibitor.



Fig. S3. Competitive inhibition of P-FIIa binding to immobilized GPIb α -N by HD1. SPR analysis of dose-dependent P-FIIa binding to immobilized wild type GPIb α -N peak 3 with or without HD1 added at approximately IC₅₀ concentration (see Fig. 5A). Results are expressed as percent binding relative to that measured at the highest concentration of P-FIIa added. The apparent K_D (250 nM) calculated from the fitting of FIIa binding in the presence of HD1 is in agreement with a model of competitive antagonism according to the equation: $K_{Dapp} = K_D (1 + [Inhibitor]/K_i)$, where K_{Dapp} and K_D are the equilibrium dissociation constants in the presence of inhibitor, respectively. Indeed, substituting in the equation the values for $K_D = 76$ nM (Table 2); $K_i = 77.3$ nM (Fig. 5A), then $K_{Dapp} = 233.31$ nM.



Fig. S4. Inhibition of P-FIIa binding to immobilized GPIb α N-Long by HD1. (*A*). Dose-response of P-FIIa binding to fully sulfated GPIb α N-Long measured by SPR. Binding is reported as mole of P-FIIa bound per mole of GPIb α N-Long immobilized onto the chip. (*B*). Inhibition of P-FIIa binding (used at K_D concentration) to immobilized GPIb α N-Long by exosite I-specific aptamer HD1. Results are expressed as binding in percent of a control with no inhibitor. The corresponding IC₅₀ and K_i values (nM; mean ± standard error of the mean; n = 2), respectively, were: 156.5 ± 1.02 and 67.07 ± 0.58.



Fig. S5. B-P-FIIa binding to human platelets and inhibition by hirugen and heparin. (*A*). Time course of B-P-FIIa (used at K_D concentration) binding to washed human platelets (2 · 10⁷ platelets/mL) with or without addition of the anti-human GPIb α monoclonal antibody, LJ-Ib10 (50 µg/mL). Alexa Fluor 488-conjugated streptavidin (0.25 molar ratio) was mixed with B-P-FIIa before adding to the platelet suspension. After incubation at room temperature (22–25 °C) for the indicated time periods, samples were analyzed by flow cytometry without dilution. Results are expressed as geometric mean of the fluorescence of 10,000 measured events. (*B*). Experiment performed as in *A* except that, after incubation of B-P-FIIa/Alexa Fluor 488-conjugated streptavidin with platelets for 20 min, increasing concentrations of P-FIIa, as indicated, were added. After additional 20 min, the displacement of B-P-FIIa was evaluated by flow cytometry. Results are reported as percent residual binding relative to that measured before the addition of P-FIIa. (C). Dose-response curves of B-P-FIIa binding to platelets and of the inhibition by increasing concentrations of LJ-Ib10; in the latter experiment, B-P-FIIa was used at saturating concentration and binding in the presence of 10-fold molar excess of exosite I-specific hirugen or exosite I-specific heparin (average molecular weight 15 kDa) was evaluated after 20 min at room temperature. In (*C* and *D*) detection was by PE-conjugated streptavidin.



Fig. S6. Stable association of Alexa Fluor 488-streptavidin with B-P-FIIa. Top. Elution time from a GPC column of HD1 (peak 1), B-P-FIIa (peak 2), B-P-FIIa/Alexa Fluor 488-streptavidin (peak 3), or B-P-FIIa/Alexa Fluor 488-streptavidin (1:1 molar ratio) in the presence of a 9-fold molar excess HD1 (peak 4). Light transmittance measured at 279 nm to detect protein. Bottom. Light transmittance measured at 495 nm, corresponding to the maximum absorption peak of the fluorochrome. Note that the volume analyzed for different mixtures was not constant.



Fig. 57. HD1 and hirudin prevent the interaction of GPIb α -N Tys²⁷⁹ with FIIa Trp¹⁴⁸. Superposition of the crystal structures of the complex of FIIa with fully sulfated GPIb α -N (PDB ID: 3PMH) and FIIa in complex with HD1 (left, PDB ID: 1HUT) or hirudin (right, PDB ID: 2HTC). In the FIIa/ GPIb α -N complex, Tys²⁷⁹ anchors a string of negatively charged GPIb α residues (279–284) to FIIa by interacting with the backbone of Trp¹⁴⁸ flanking exosite I. This interaction is unattainable when Tyr²⁷⁹ is substituted by Phe or the position of Trp¹⁴⁸ is changed upon binding of the two exosite I ligands. GPIb α -N is colored green; FIIa is red in complex with GPIb α -N and blue in complex with HD1 or hirudin; the latter two are light blue and dark orange, respectively. Tys²⁷⁹ in GPIb α -N and Thr¹⁴⁷/Trp¹⁴⁸ in FIIa are shown with side chains in stick and ball representation.



Fig. S8. Schematic representation of the mechanism of FIIa binding to platelets based on crystal structure information. (*A*). Structure of the complex showing the contacts between one GPIbα-N (receptor) and two FIIa (ligand) molecules. (*B*). Structure of the complex showing the contacts between one FIIa and two GPIbα-N molecules. This mode of interaction is required for stable association of the ligand with the immobilized receptor. (*C*). Schematic representation of a spatial arrangement, compatible with molecular dimensions in the crystal and 1:1 stoichiometry, in which each of four ligand molecules is bound concurrently to two receptors through distinct sites, and each receptor binds two distinct ligands. The indicated distances between the centroids of receptor molecules are estimates based on the proposed clustering and compatible with the proposed binding mechanism.

Sulfate oxygen atoms of GP Ib α	Site I contacts (exosite I)		Site II contacts (exosite II)	
Tys276-O1 Tys276-O2			Arg126-NH1 Arg126-CG Arg126-CD <u>Arg126-NE</u> Phe232-CE1	3.69 3.29 3.76 <u>3.83</u> 3.23
			Arg126-CZ <u>Arg126-NH1</u>	3.03 3.41 <u>2.65</u>
Tys276-O3	Asp60e-CG <u>Asp60e-O</u> Asp60e-CB	3.89 <u>3.15</u> 3.68	Arg126-CG Arg126-CD <u>Arg126-NE</u> Arg126-NH1	3.13 3.92 <u>3.71</u> 3.71
Tvs278-O1	none		AIG120-INFL 3.82	
Tys278-O2	Pro37-CD Pro37-CB PRO37-CG	2.85 3.35 2.65		
Tys278-O3	Arg35-CZ <u>Arg35-NH1</u> Arg35-NH2	3.84 <u>3.31</u> 3.80	<u>Lys236-NZ</u>	<u>3.46</u>
Tys279-O1	Thr147-OG1 Thr147-CG2 Thr147-CB	3.79 3.00 3.99		
Tys279-O2	Thr147-CG2 Thr147-C <u>Thr147-O</u> <u>Trp148-N</u> Trp148-CA Trp148-CB Trp148-C	3.74 3.23 <u>3.09</u> <u>3.10</u> 2.75 3.36 3.92	Trp237-CD1	3.63
Tys279-O3	110-0-0	5.52	Lys240-CG <u>Lys240-NZ</u>	3.74 <u>3.49</u>

Table S1. Contact distances (in Å) between sulfate oxygen atoms of three sulfated Tyr residues of GPIb α and atoms of residues in FIIa site I and site II

Hydrophilic contacts are underlined.

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