Supporting Information

Breaking the regioselectivity rule for acrylate insertion in the Mizoroki-Heck reaction

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Materials and Experimental General Considerations

Unless noted otherwise, all manipulations of phosphorous halides or palladium complexes were carried out under an inert nitrogen or argon atmosphere using standard glovebox or Schlenk techniques. Methylene chloride, DMSO and pentane were destilled from calcium hydride, toluene and benzene from sodium, THF and diethyl ether from blue sodium/benzophenone ketyl, and methanol from activated (iodine) magnesium under argon prior to use. Acetone p.a. and methyl acrylate were degassed by repetitive freeze/thaw-cycles and used without further purification. All other solvents were commercial grade. 2,4,6-Trimethylaniline (98%) and 2,6-di(isopropyl)aniline (90%, technical grade) was purchased from Acros, and palladium(II)-chloride was obtained from Merck. Trichlorophosphine (99%) was purchased from Riedel-de Haën. Cyclooctadiene palladium methyl chloride ([(COD)PdMeCI]) (1) and 2-chloro-1,3-di(aryl)-1,3,2-diazaphospholidines (2) were synthesized according to literature procedures. All deuterated solvents were supplied by Eurisotop.

NMR spectra were recorded on a Varian Unity Inova 400, a Bruker Avance III DRX 400 or a Bruker Avance DRX 600 instruments. ¹H NMR spectra were referenced to residual protiated solvent signals. ¹³C NMR spectra were referenced to the solvent signals, and ³¹P NMR spectra to external 85% H₃PO₄. ¹H NMR spectra are reported as follows: chemical shift (δ ppm), multiplicity, coupling constant (Hz), and integration. Multiplicities are given as follows (or combinations thereof): s: singlet, d: doublet, t: triplet, sept: septet, m: multiplet, v[m]: virtual multiplet; br.: broad, v. br.: very broad. The identity of metal complexes and detailed NMR assignments were established by 2D NMR experiments (¹H-¹H gCOSY, ¹H-¹³C gHSQC and ¹H-¹³C gHMBC) in addition to 1D NMR experiments. NMR temperature calibration was performed using pure methanol (low temperature) or ethylene glycol (high temperature) samples.

Experimental Procedures

Ligand Synthesis



Lithium 2-(1,3-dimesityl-1,3,2-diazaphospholidin-2-yl)benzenesulfonate (L1a)



To a solution of benzenesulfonic acid (1582 mg, 10 mmol) in thf (35 mL) in a septum capped Schlenk tube was slowly added by syringe butyl lithium (14 mL 1.6 M pentane solution, 22.4 mmol) at 298 K under stirring. The formed suspension was stirred for 30 min at 298 K, then cannula transferred into a Schlenk frit, and filtrated. After washing with pentane (4 x 15 mL) the off-white powder was dried under high vacuum (10^{-3} mbar) to leave *o*-dilithiobenzenesulfonate•(0.75 THF) (2180 mg, 9.60 mmol, 96 %) which was used in the next step without further purification.

To o-dilithiobenzenesulfonate (0.75 THF) (493 mg, 2.2 mmol) and 2-chloro-1,3dimesityl-1,3,2-diazaphospholidine (742 mg, 2.06 mmol) in a 100 mL Schenk-tube was added THF (50 mL) and the resulting suspension was stirred for 4 hours at room temperature while a clear solution was obtained. The solvent was removed in vacuo (10⁻³ mbar), the remaining solid suspended in pentane (20 mL), and the suspension cannula transferred into a Schlenk-frit and washed with pentane $(3 \times 10 \text{ mL})$ to yield ligand L1a containing ca 0.2 equiv of THF (910 mg, 1.81 mmol, 87.8 %). ¹H NMR (400 MHz, 298K, methanol- d_4): δ 8.37 (d, ${}^{3}J_{HH}$ = 7.6 Hz, 1H, 6-H), 7.83 (m, 1H, 3-H), 7.49 (m, 1H, 5-H), 7.30 (m, 1H, 4-H), 6.70 (s br., 4H, 9-, 11-, 20-, and 22-H), 4.19 and 3.36 (m each, 2:2H, 16- and 17-H₂), 2.15 (s br., 18H, 13-, 14-, 15-, 24-, 25-, and 26-H₃). ¹³C NMR (100 MHz, 298K, methanol- d_4): δ 149.51 (C_q, ²J_{PC} = 26.8 Hz, C2), 148.70 (C_q, d, ${}^{1}J_{PC}$ = 77.4 Hz, C1), 141.86 (C_q, d, ${}^{2}J_{PC}$ = 16.5 Hz, C7 and C18), 138.94 and 138.94 (C_a each, C13, C15, C24, and C26), 135.68 and 135.66 (C_a each, C10 and C21), 134.60 (CH, C6), 129.73 (CH, d, ${}^{3}J_{PC}$ = 1.3 Hz, C4), 129.25 (CH, C5), 127.73 (CH, d, ${}^{2}J_{PC}$ = 2.0 Hz, C3), 53.37 (CH₂, d, ${}^{2}J_{PC}$ = 6.4 Hz, C16 and C17), 20.83, 20.76, and 20.72 (CH₃ each, C13, C14, C15, C24, C25, and C26). ³¹P NMR (161 MHz, 298K, methanol-d₄): δ 86.81. Anal Calc (%) for C₂₆H₃₀N₂O₃PSLi•(0.2 C₄H₈O) (497.4): C: 64.72, H: 6.40, N: 5.63. Found C: 65.16, H 6.94, N 5.08.

Lithium 2-[1,3-di-(2,6-diisiopropyl)phenyl]-1,3,2-diazaphospholidin-2-yl) benzenesulfonate (L1b)



To o-dilithiobenzenesulfonate (0.75 THF) (448 mg, 2 mmol) and 2-chloro-1.3-[di-(2.6-diisopropyl)phenyl]-1.3.2-diazaphospholidine (935 mg, 2.1 mmol) in a 20 mL Schenk-tube was added THF (10 mL) and the resulting suspension was heated under stirring to 340 K for 1 hour. The clear solution was cooled to room temperature and the solvent removed in vacuo (10⁻³ mbar). Diethyl ether (10 mL) was added to the remaining glassy solid and the mixture was sonicated for 20 min while a fine white precipitate formed. The white precipitate was cannula transferred into a Schlenk frit, washed with diethyl ether $(2 \times 5 \text{ mL})$ and pentane $(2 \times 10 \text{ mL})$, dried under high vacuum, and dissolved in hot benzene (50 mL). The resulting opaque mixture was filtrated through a pad of celite, and the solvent removed in vacuo. The remaining solid was triturated with pentane (10 mL) to yield ligand L1b containing 1 equiv of diethyl ether (1072 mg, 1.66 mmol, 82.8 %) after drying under vacuum. ¹H NMR (400 MHz, 298 K, CD₂Cl₂): δ 8.17 (d, ³J_{HH} = 7.6 Hz, 1H, 6-H), 7.57 (m, 1H, 3-H), 7.37 (m, 1H, 5-H), 7.20 (m, 1H, 4-H), 7.09 (m, 4H, 9-, 11-, 23-, and 25-H), 6.91 (m, 2H, 10- and 24-H), 4.13 and 3.36 (m each, 2:2H each, 19- and 20-H₂), 3.68 (m, 2H, 13- and 16-H), 3.21 (m, 2H, 27- and 30-H), 1.32 and 1.21 (d each, ${}^{3}J_{HH}$ = 6.8 Hz, 6:6H, 14-, 15-, 17-, and 18-H), 1.03 and 0.29 (d each, ${}^{3}J_{HH}$ = 6.8 Hz, 6:6H, 28-, 29-, 31-, and 32-H). ¹³C NMR (100 MHz, , 298 K, CD₂Cl₂): δ 152.0 (C_q, d, ²J_{PC} = 28 Hz, C1), 148.74 and 148.72 (C_q each, C8 and C12), 147.7 (C_q , C22 and C26), 144.8 (C_q , d, ${}^1J_{PC}$ = 81 Hz, C2), 140.2 (C_q , d, ${}^2J_{PC}$ = 15.5 Hz, C7 and C21), 131.5 (CH, C6), 126.9 (CH, C4), 126.1 (CH, C5), 125.4 (CH, C5, C10, and C24), 123.2 (CH, C9 and C11), 122.9 (CH, C23 and C25), 54.6 (CH₂, d, ²J_{PC} = 5.2 Hz, C19 and C20), 27.55 and 27.47 (CH each, C13 and C16), 27.2 (CH, C27 and C30), 24.9 (CH₃, C28 and C29), 23.8 and 23.6 (CH₃ each, C28, C29, C31, and C32), 23.75 and 23.72 (CH₃ each, C17 and C18). ³¹P NMR (161 MHz, 298 K, CD₂Cl₂): δ 101.3. Anal Calc for C₃₂H₄₂N₂O₃PSLi•(C₄H₁₀O) (646.8): C: 66.85, H: 8.10, N: 4.33. Found: C: 67.27, H: 8.42, N: 4.00.

{(κ^2 -*P*,*O*)-2-[1,3-dimesityl-1,3,2-diazaphospholidin-2-yl)benzenesulfonato]palladium(II)-methyl} lithium chloride adduct 2a-LiCl



To ligand L1a•(0.2 THF) (408 mg, 0.82 mmol) and [(COD)PdMeCl] (214 mg, 0.81 mmol) in a 10 mL Schlenk tube was added methylene chloride (5 mL) and the resulting mixture was stirred for 10 min at 298K. The solvent was removed in vacuo to leave a glassy solid, which was dispersed in pentane and washed diethyl ether. A white powder of 2a-LiCl containing traces of diethyl ether was isolated after dissolution of the crude reaction product in methylene chloride, precipitation with pentane, filtration, and drying under vacuum (440 mg, 0.665 mmol, 81 %). ¹H NMR (400 MHz, 298K, acetone- d_6): δ 8.32 (vt, J = 8.0 H, 1H, 6-H), 7.88 (m, 1H, 3-H), 7.67 (vt, J = 7.4 Hz, 1H 5-H), 7.55 (vt, J = 7.6 Hz, 1H, 4-H), 6.86 and 6.72 (s br. each,2:2H, 9-, 11-, 20-, and 22-H), 4.03 and 3.90 (m each, 2:2H, 16- and 17-H₂), 2.65 and 1.78 (s br. each, 6:6H, 13-, 15-, 24- and 26-H₃), 2.20 (s, 6H, 14- and 25-H₃), 0.21 (s, 3H, Pd-CH₃). ¹³C NMR (100 MHz, 298K, acetone- d_6): δ 147.94 (C_q, d, ² J_{PC} = 17.5 Hz, C1), 140.90 (C_a, d, ${}^{1}J_{PC}$ = 26.7 Hz, C2), 139.92 (C_a, d, ${}^{2}J_{PC}$ = 8.5 Hz, C7 and C18), 137.62 (C_g br., C8, C12, C19, and C23), 136.07 (C_g, C10 and C21), 132.09 (CH, C6); 130.80, 130.65, and 130.61 (CH each, C4, C5, C9, C11, C20 and C22), 128.93 (CH, d, ${}^{2}J_{PC}$ = 6.8 Hz, C3), 52.49 (CH₂, C16 and C17), 21.34 (CH₃ br., C13, C15, C24, and C26), 20.74 (CH₃, C14 and C25), -7.12 (CH₃, Pd-CH₃). ³¹P NMR (161 MHz, 298K, acetone-d₆): δ 82.56. Anal Calc for C₂₇H₃₃N₂O₄PSPdLiCl (661.4): C: 49.03, H: 5.03, N: 4.24. Found: C: 49.81, H: 5.52, N: 4.25. Recrystallization yields 2a-LiCl (acetone) within 2 d after layering an acetone solution of 2a-LiCl with pentane. Crystals of 2a-LiCl•(MeOH) suitable for X-Ray diffraction were obtained by layering a methanol solution of 2a-LiCl with pentane in an NMR tube.

{(κ²-*P*,*O*)-2-[1,3-di(2,6-diisopropyl)phenyl-1,3,2-diazaphospholidin-2-yl) benzenesulfonato]-palladium(II)-methyl} lithium chloride adduct [2b-LiCl•(0.5 Et₂O)/ 2b-LiCl•(2 THF)]

To ligand L1b•(Et₂O) (65 mg, 100 μ mol) and [(COD)PdMeCI] (26 mg, 98 μ mol) in a 10 mL Schlenk tube was added methylene chloride (5 mL) and the resulting mixture was stirred for 30 min at 298K. The solvent was removed in vacuo to leave a glassy solid, which was suspended in diethyl ether (7 mL). The mixture was transferred to a centrifugation vial and the solid separated by repeated centrifugation and extraction of the solid with diethyl ether (2 × 7 mL). After drying the solid under high vacuum (10⁻³ mbar) compound 2b-LiCl•(0.5 Et₂O) was obtained as white microcrystalline material (70.5 mg, 45.9 mmol, 93.8 %) containing traces of methylene chloride. Different batches, however contained up to 1 equiv of diethyl ether per litihium depending on the time under high vacuum.



¹H NMR (400 MHz, 298K, acetone- d_6): δ 8.21 (vt, J = 8.2 Hz, 1H, 6-H), 7.90 (m, 1H, 3-H), 7.71 (vt, J = 7.3 Hz, 1H, 5-H), 7.58 (vt, J = 7.5 Hz, 1H, 4-H), 7.22 (m, 4H, 9-, 11-, 23-, and 25-H), 7.04 (m, 2H, 10- and 24-H), 4.21 and 3.93 (m each, 2:2H, 19- and 20-H₂), 3.77 and 2.96 (m each, 2:2H, 13-, 16-, 27-, and 30-H), 1.56, 137, 0.96, and 0.76 (d each, ${}^{3}J_{HH}$ = 6.8 Hz each, 6:6:6:6H, 14-, 15-, 17-, 18-, 28-, 29-, 31-, and 32-H₃), 0.22 (s br., 3H, Pd-CH₃). ¹³C NMR (100 MHz, 298K, acetone- d_6): δ 148.23 and 147.80 (C_q each, d each, ³ J_{PC} = 1.2 and 2.7 Hz, C8, C12, C22, and C26), 147.78 (C_q, d, ${}^{2}J_{PC}$ = 18.8 Hz, C1), 139.90 (C_q, d, ${}^{1}J_{PC}$ = 27.4 Hz, C2), 139.46 (C_q, d, ${}^{2}J_{PC}$ = 7.7 Hz, C7 and C21), 129.96 (CH br., C6), 129.90 (CH, d, ³J_{PC} = 2.0 Hz, C4), 129.61 (CH, d, ${}^{4}J_{PC}$ = 5.0 Hz, C5), 129.29 (CH, d, ${}^{2}J_{PC}$ = 7.2 Hz, C3), 127.37, 127.36 and 125.04 (CH each, C9, C11, C23, and C25), 124.27 (CH, C10 and C24), 54.05 (CH₂, C19 and C20), 28.97 and 28.68 (CH each, C13, C16, C27, and C30), 26.20, 25.24, 24.25, and 23.23 (CH₃ each, C14, C15, C17, C18, C28, C29, C31, and C32), -7.54 (CH₃, d, ${}^{2}J_{PC}$ = 3.7 Hz, Pd-CH₃). ${}^{31}P$ NMR (161 MHz, 298K, acetone- d_{6}): δ 89.91. Anal Calc (%) for C₃₃H₄₅N₂O₃PSPdLiCl•(0.5C₄H₁₀O) (766.6): C: 54.83, H: 6.57, N: 3.65. Found: C: 55.01, H: 6.78, N: 4.01. Crystals of 2b-LiCl-(2 THF) suitable for Xray diffraction analysis were obtained within 1d at 298K by layering a solution of 8 mg **2b-LiCl**•(0.5 Et₂O) in 60 μL THF with pentane (2 mL) in a NMR tube.

$\{[(\kappa^2 - P, O) - 2 - (1, 3 - dimesity] - 1, 3, 2 - diazaphospholidin - 2 - yl)benzenesulfonato] palladium(II)-methyl-(<math>\kappa$ -O)-dimethylsulfoxide $\}$ (2a-DMSO)



To a solution of 2a-LiCL•(acetone) (84.3 mg, 59.9 µmol mmol) in methylene chloride (3 mL) was added DMSO (30.1 mg, 385 µmol). Addition of silvertetrafluoroborate (76.5 mg, 393 µmol) caused formation of a white precipitate which was filtered off by a syringe filter containing a celite plug after stirring for 3 h at 298K. After removal of all volatiles the resulting brown solid was suspended in pentane and sonicated. The insoluble light brown solid was filtered off and washed with pentane and diethyl ether, respectively, to yield **2a** which contains LiBF₄ by ¹⁹F-NMR spectroscopy. Elemental analysis suggests a composition of ca 2a-DMSO•(LiBF₄) (yield: 80.6 mg, 104 µmol, 87 %). ¹H NMR (400 MHz, 298K, CD₂Cl₂): δ 7.98 (vt, J = 8.6 Hz, 1H, 6-H), 7.88 – 7.80 (m, 1H, 3-H), 7.50 (vt, J = 7.2 Hz, 1H, 5-H), 7.43 (vt, J = 7.6 Hz, 1H, 4-H), 6.77 (br, 4H, 9-, 11-, 20- and 22-H), 3.99 and 3.97 (s br. each, 2:2H, 16- and 17H₂),2.54 (s, 6H, DMSO-CH₃), 2.52 and 1.63 (br each, 6:6H, 13-, 15-, 24-, and 26-H₃), 2.16 (s, 6H, 14- and 25-H₃), 0.06 (s, 3H, Pd-CH₃). ³¹P NMR (161 MHz, 298K, CD₂Cl₂): δ 89.85. ESI-MS: calc: 680.11 g mol⁻¹; Found: 603.2 (MH⁺-DMSO), 625.0 (M+Na-DMSO)⁺, 648.2 (M+HCOOH-DMSO)⁺, 704.1 (M+HNEt₃-DMSO)⁺. Anal Calc (%) for C₂₉H₃₉N₂O₄PS₂Pd•LiBF₄ (774.9): C, 44.95; H, 5.07; N, 3.62; Found: C, 46.14; H, 5.49; N, 3.68. Crystals of 2a-DMSO suitable for X-ray diffraction were grown from a solution of **2a-DMSO**•(LiBF₄) in benzene layered with pentane in an NMR-tube.

{[(κ^2 -*P*,*O*)-2-(1,3-dimesityl-1,3,2-diazaphospholidin-2-yl)benzenesulfonato] palladium(II)-[(κ^2 -*C*,*O*)-3-methoxy-2-methyl-3-oxopropyl]} (3a)



To a solution of **2a-LiCl**•(acetone) (200 mg, 284 μmol) in methylene chloride (5 mL) were added BHT (10 mg), methyl acrylate (MA; 190 mg, 2.2 mmol), and silver tetrafluoroborate (180 mg, 924 µmol). The reaction mixture was stirred for 2 h at 298 K in the dark and then filtered through a syringe filter to afford a vellow solution. Diethyl ether (20 mL) was added, the resulting precipitate separated by filtration and washed with diethyl ether $(2 \times 5 \text{ mL})$ and pentane (5 mL) to yield **3a** containing ca 1 equiv of LiBF₄ (by elemental analysis) as off-white solid after drying under high vacuum (187 mg, 239 μmol, 84.1%). ¹H NMR (400 MHz, 298K, CD₂Cl₂): δ 8.11 (vt, J = 8.6 Hz, 1H, 6-H), 7.97 (m, 1H, 3-H), 7.62 (vt, J = 7.5 Hz, 1H, 5-H), 7.54 (vt, J = 7.5 Hz, 1H, 4-H), 6.93 (br, 4H, 9-, 11-, 20- and 22-H), 3.91 (m, 4H, 16- and 17-H₂), 3.81 (s, 3H, 31-H₃), 2.65, 2.51, 1.85, and 1.72 (br. each, 3:3:3:3H, 13-, 15-, 24-, and 26-H₃), 2.35 (m, 1H, 28-H), 2.23 (s, 6H, 14- and 25-H₃), 1.39 (m, 1H, 27-H), 0.97 (m, 1H, 27-H), 0.87 (d, ${}^{3}J_{HH}$ = 7.0, 3H, 29-H₃). 13 C NMR (100 MHz, 298K, CD₂Cl₂): δ 189.77 $(C_{q}, C29), 146.48 (C_{q}, d^{2}J_{PC} = 20.0 Hz, C1), 139.38 (C_{q}, d, {}^{1}J_{PC} = 36.8 Hz, C2),$ 138.99 and 138.90 (C_{α} each, C7 and C18), 138.34, and 138.26 (C_{α} each, C10 and C21), 136.89, 136.87, 136.63, and 136.61 (C_α each, C8, C12, C19, and C23), 131.47 (d, ${}^{2}J_{PC}$ = 1.5 Hz, C4), 131.37 (CH, ${}^{2}J_{PC}$ = 2.0 Hz, C6) 130.59 (CH, d, ${}^{3}J_{PC}$ = 5.7 Hz C5), 130.21 (CH each, C9, C11, C20 and C22) 128.48 (d, 2J_{PC}= 7.8 Hz, C3), 55.55 (s, C30), 51.99 (CH₂, C16 and C17), 44.77 (CH, d, ³J_{PC} = 2.4 Hz, C28), 21.05 (CH₃ br., C13, C14, C15, C24, C25, and C26) 20.98 (CH₂, d, ²J_{PC} = 6.4 Hz, C27), 17.83 (CH₃, d, ⁴J_{PC} = 2.5 Hz, C31). ³¹P NMR (161 MHz, 298K, CD₂Cl₂): δ 85.67. Anal. Calcd. (%) for C₃₁H₃₉N₂O₅PSPd•LiBF₄ (782.9): C, 47.56; H, 5.02; N, 3.58; Found: C, 48.72; H, 5.35; N, 3.80. Crystals suitable for X-Ray diffraction were obtained by layering a methylene chloride solution of **3a** with pentane in a NMR tube. A kinetic analysis of a solution obtained from 2a-LiCl (acetone) (21.8 mg, 29.8 μ mol), methyl acrylate (115 mg, 1.34 mmol; 22.5 equiv vs palladium; ¹H and ³¹P

NMR recorded), and silver triflate (9.0 mg, 35 μ mol) in methylene-chloride- d_2 after filtration reveals $k_{obs} = 6.0 \ (\pm 0.1) \times 10^{-4} \text{ s}^{-1}$ for the disappearance of the palladium methyl signal at 298 K.

{[(κ^2 -*P*,*O*)-2-(1,3-di(2,6-diisopropyl)phenyl-1,3,2-diazaphospholidin-2-yl) benzenesulfonato]palladium(II)-[(κ^2 -*C*,*O*)-3-methoxy-2-methyl-3-oxopropyl]} (3b)



To a solution of complex 2b-LiCl•(0.85 Et₂O) (48 mg, 30.3 μ mol) in methylene chloride- d_2 (350 µL) in a NMR tube (¹H and ³¹P NMR recorded) was added methyl acrylate (115 mg, 1.34 mmol; 22 equiv vs palladium; ¹H and ³¹P NMR recorded) and silver triflate (17.1 mg, 66.6 µmol) at 298K. The NMR tube was shaken for 2 min and then centrifuged, which resulted in the deposition of silver chloride. The solution was filtered through a syringe filter into a J. Young NMR tube, the syringe filter rinsed with additional methylene chloride- d_2 (120 μ L), and the sample heated to 318 K in the NMR probe while continuously monitoring the reaction progress by ¹H NMR spectra. After complete consumption of the palladium methyl species (1 h at 318 K, k_{obs} = 4.8 $(\pm 0.1) \times 10^{-4} \text{ s}^{-1}$), the J. Young tube was connected to a vacuum manifold and all volatiles were carefully removed at 243K to leave a highly viscous, sticky oil in the tube. Benzene (0.5 mL) was added, the tube was cooled to 273K (ice bath), and the frozen solvent (and residual volatiles) were removed by sublimation under vacuum. The sample thus obtained contains traces of methyl crotonate, methyl acrylate, and one not further identified ester apart from complex 3b in ca 95 % NMR purity as was determined by 1- and 2D NMR experiments. Chemical shifts reported in the NMR assignment are based on the analysis of the crude reaction product obtained by the procedure reported here. Isolation of compound **3b** (40.6 mg, 52.5 µmol, 86.6%) was accomplished by crystallization after removing the solvent under vacuum, washing the residue with diethyl ether, redissolution of the residue in methylene chloride (85 μ L) and layering pentane (1.5 mL) on top of the methylene chloride solution. ¹H NMR (600 MHz, 298K, CD₂Cl₂): δ 8.01 (m, 2H, 6- and 3-H), 7.63 (vt, J = 7.4 Hz, 1H, 5-H), 7.53 (vt, J = 7.7 Hz, 1H, 4-H), 7.28 and 7.24 (m each, 2:2H, 9-, 11-, 23-, and 25-H), 7.06 (m, 2H, 10- and 24-H), 4.20, 4.07 and 3.93 (m each, 1:1:2H, 19- and 20-H₂), 3.74 (s, 3H, 36-H₃), 3.72, 3.55, 2.88, and 2.73 (sept each, ${}^{3}J_{HH}$ = 6.8 Hz each, 1:1:1:1H, 13-, 16-, 27-, and 30-H), 2.48 (m, 1H, 34-H), 1.54, 1.51, 1.39, 1.38, 0.99, 0.98, 0.70, and 0.63 (d each, ${}^{3}J_{HH}$ = 6.8 Hz each, 3:3:3:3:3:3:3H, 14-, 15-, 17-, 18-, 28-, 29-,31-, and 32-H₃), 1.12 (m, 1H, 34-H), 0.77 (vt, J = 9.6 Hz, 34-H), 0.76 (d, ${}^{3}J_{HH}$ = 7.1 Hz, 3H, 37-H₃). ¹³C NMR (150 MHz, 298K, CD₂Cl₂): δ 189.58 (C_q, br., C35), 149.29, 149.12, 148.10, and 147.85 (C_q each, d each, ${}^{3}J_{PC}$ = 1.2, 1.5, 1.8, and 2.2 Hz, C8, C12, C22, and C26), 147.52 (C_q br., d, ²J_{PC} = 17.7 Hz, C1), 138.86 and 137.97 (C_a each, d each, ${}^{2}J_{PC}$ = 7.8 and 7.4 Hz, C7 and C21), 138.29 (C_a, d, ${}^{1}J_{PC}$ = 37.6 Hz, C2), 131.46 (CH, d, ⁴J_{PC} = 1.5 Hz, C5), 130.17 (CH, C4), 130.13 (CH, d, ²J_{PC} = 3.6 Hz, C3), 129.6 (CH, C6), 128.45, 125.56 and 125.45 (CH, C9, C11, C23, and C25), 125.16 (CH, C10 and C24), 55.62 (CH₃, C36), 54.72 and 54.11 (CH₂ each, C19 and C20), 44.65 (CH, C34), 29.69, 29.40, 29.38, and 29.36 (CH each, C13, C16, C27, and C30), 27.10, 26.90, 25.73, 25.67, 24.84, 24.71, 23.66, and 23.61 (CH₃ each, C14, C15, C17, C18, C28, C29, C31, and C32), 17.91 (CH₃ br.; C37), 17.86 (CH₂, d, ²J_{PC} = 2.0 Hz, C33). ³¹P NMR (161 MHz, 298K, CD₂Cl₂): δ 94.31. Anal Calc (%) for C₃₇H₅₁N₂O₅PSPd (773.3): C: 57.47, H: 6.65, N 3.62. Found: C: 57.89, H: 6.97, N 3.41. Crystals suitable for X-ray analysis were obtained by layering a solution of 12 mg **3b** in methylene chloride (70 μ L) with pentane (1.5 mL) in a NMR tube.

Methyl methacrylate

Thermolysis of isolated **3b** (16.3 mg, 21 μ mol) in methylene chloride-*d*₂ solution (450 μ L) in a J. Young NMR tube yields quantitatively methyl methacrylate after 16 h at 353 K according to ¹H NMR spectroscopy (residual protio-methylene chloride is used as internal reference).

{(κ²-*P*,*O*)-[2-[di(2-anisyl)phosphin-2-yl]benzenesulfonato]palladium(II)-phenyl (methanol)} (4-MeOH)

[(DMSO)₂Pd(Ph)Cl]. In a 100 mL Schlenk tube palladium dichloride (887 mg, 5 mmol) was dissolved in DMSO (6 mL) at 383 K under stirring. After cooling to 298 K, diethyl ether was added (70 mL) under stirring, the resulting orange solid collected by filtration, washed with diethyl ether (4 \times 10 mL), and dried under vacuum to yield [(DMSO)₂PdCl₂] (1.630 g, 4.89 mmol, 97.7 %). To a suspension of [(DMSO)₂PdCl₂] (334 g, 1 mmol) in a mixture of DMSO (3 mL) and THF (15 mL) in a 50 mL septum capped Schlenk tube was added tetraphenyl tin (215 mg, 503.3 µmol). The solution was stirred for 8 h at 298 K while the orange starting material dissolved and an orange-red solution containing some palladium black formed. The mixture was concentrated to ca 5 mL, DMSO (5 mL) was added and the resulting mixture filtered through a syringe filter into a 100 mL Schlenk flask. Removal of all volatiles under high vacuum (298 K, 10⁻³ mbar) gave a orange solid, which was dispersed in THF (15 mL) and cannula transferred into a Schlenk frit. The was solid filtered off, washed with THF (4 \times 10 mL), and dried under vacuum to yield a white-grey powder of [(DMSO)₂Pd(Ph)Cl] which contains traces of palladium black (361.1 mg, 0.965 mmol, 96.5 %). **[(DMSO)₂Pd(Ph)Cl]**: ¹H NMR (400 MHz, 298K, DMSO-*d*₆): δ 7.18 (m, 2H, ortho-H, Ph), 6.96 (m, 2H, meta-H, Ph), 6.94 (m, 1H, para-H, Ph), 2.55 (s, 12H, 2 × DMSO-*h*₆). ¹³C NMR (100 MHz, 298K, DMSO-*d*₆): δ 134.59 (CH, *ortho*-C, Ph) 126.97 (CH, meta-C, Ph), 123.82 (CH, para-C, Ph), 40.43 (CH₃, 2 × DMSO-h₆), ipso-C, Ph not detected. Anal Calc (%) for C₁₀H₁₇O₂S₂CIPd (375.2): C: 32.01, H: 4.57. Found: C: 31.36, H 4.01.

4-MeOH. To sodium [di(2-anisyl)phosphinyl]benzene sulfonate (**L2**) (3) (42.4 mg, 100 μ mol) and [(DMSO)₂Pd(Ph)Cl] (41.6 mg, 111 μ mol) in a NMR tube was added CD₃OD (400 μ L). The tube was shaken for 5 min at 298 K and centrifuged. ¹H and ³¹P NMR monitoring indicated consumption of **L2** and formation of one new ³¹P containing species in > 95 % NMR yield. Complex **4-MeOH** started to separate by crystallization after ca 10 min at 298 K. The solvent was removed in vacuo, the residue dissolved in methylene chloride (3 mL), and the solution filtrated. After removal of all volatiles the residue was recrystalized from refluxing methanol to yield **4-MeOH** (50.7 mg, 78.1 μ mol, 78.1 %) containing traces of residual DMSO after 16h under high vacuum.



¹H NMR (600 MHz, 298K, CD₃OD): δ 8.03 (ddd, ³*J*_{HH} = 7.9, ⁴*J*_{HH} = 1.2 and ⁴*J*_{PH} = 4.7 Hz, 1H, 6-H), 7.78 (dd, ³*J*_{HH} = 7.5 and ³*J*_{PH} = 14.8 Hz, 2H, 12- and 18-H), 7.52 (m, 1H, 5-H), 7.49 (m, 1H, 3-H), 7.47 (m, 2H, 10- and 16-H), 7.39 (m, 1H, 4-H), 6.94 (m, 6H, 9-, 11-, 15-, 17-, 20-, and 24-H), 6.58 (m, 1H, 22-H), 6.53 (m, 2H, 21- and 23-H), 3.61 (s, 6H, 2 × OCH₃), 2.67 (s, 1.4H residual DMSO). ¹³C NMR (150 MHz, 298K, CD₃OD): δ 161.60 (C_q, d, ²*J*_{PC} = 1.1 Hz, C8 and C14), 148.05 (C_q, d, ²*J*_{PC} = 14.3 Hz,

C1), 147.04 (C_q br., C19), 139.16 (C_q, d, ${}^{2}J_{PC}$ = 13.7 Hz, C12 and C18), 136.17 (CH, d, ${}^{2}J_{PC}$ = 2.8 Hz, C3), 135.23 (CH, d, ${}^{3}J_{PC}$ = 4.4 Hz, C20 and C24), 134.72 CH, d, ${}^{4}J_{PC}$ = 2.2 Hz, C10 and C16), 131.31 (CH, d, ${}^{4}J_{PC}$ = 2.5 Hz, C5), 129.84 (CH, d, ${}^{3}J_{PC}$ = 7.8 Hz, C4), 129.22 (C_q, d, ${}^{1}J_{PC}$ = 53.1 Hz, C2), 128.32 (CH, d, ${}^{3}J_{PC}$ = 8.3 Hz, C6), 127.65 (CH, d, ${}^{4}J_{PC}$ = 2.8 Hz, C21 and C23), 124.44 (CH, C22), 121.46 (CH, d, ${}^{3}J_{PC}$ =13.2 Hz, C11 and C17), 116.14 (C_q, d, ${}^{1}J_{PC}$ = 62.9 Hz, C7 and C13), 112.33 (CH, d, ${}^{3}J_{PC}$ = 4.7 Hz, C9 and C15), 55.53 (CH₃, 2 × OCH₃), 40.57 (CH₃, residual DMSO). ³¹P NMR (161 MHz, 298K, CD₃OD): δ 17.97. Anal Calc (%) for C₂₇H₂₇O₆PSPd (617.0): C: 52.56; H: 4.41. Found: C: 52.10; H: 4.23. Crystals of **4-MeOH**•(MeOH) suitable for X-ray diffraction analysis were grown from refluxing CD₃OD.

(*E*)-Methyl cinnamate (5) and methyl acrylate insertion products into palladium hydride (9 and 10,10')

To a solution of **4-MeOH** (14 mg, 22.7 μ mol) in methanol- d_4 (400 μ L) in a J. Young NMR tube was added methyl acrylate (45.5 mg, 529 μ mol) and the sample was monitored by ¹H NMR spectroscopy periodically over a period of 2.5 h. After 61 min at 298 K consumption of **4-MeOH** was complete while formation of (*E*)-methyl cinnamate (**5**) and two products arising from insertion of methyl acrylate into (P^O)Pd(H) were observed. After removal of methanol- d_4 and excess methyl acrylate under vacuum, methyl cinnamate (**5**) was identified as solely organic product formed by insertion of the remaining material in methylene chloride- d_2 . In addition, palladium complex **9** formed by 1,2-methyl acrylate insertion into (P^O)Pd(H) as well diastereomeric palladium complexes **10,10**' formed by 1- and 2-dimensional NMR experiments.



5: ¹H NMR (400 MHz, 298K, CD₂Cl₂): δ 7.68 (d, ³J_{HH} = 16.1 Hz, 1H, PhC*H*=CH), 7.54 and 7.40 (m each, 2:3H, *ortho*-, *meta*-, and *para*-H, Ph), 6.46 (d, ³J_{HH} = 16.1 Hz, 1H, PhCH=CH), 3.77 (s, 3H, OCH₃). ¹³C NMR (100 MHz, 298K, CD₂Cl₂): δ 167.71 (Cq, CO₂Me), 145.09 (CH, PhCH=CH), 135.00 (C_q, *ipso*-C, Ph), 130.83 (CH, *para*-C, Ph), 129.43 and 128.60 (CH each, *ortho*- and *meta*-C, Ph), 118.43 (CH, PhCH=CH), 52.08 (CH₃, OCH₃).

9: ¹H NMR (400 MHz, 298K, CD₂Cl₂, key resonances only): δ 1.95 and 1.27 (m each, 1:1H, β -H₂), 0.88 (t, ³*J*_{HH} = 7.2 Hz 2H, α -H₂).

10 [**10'** (minor diastereomer) in ()]: ¹H NMR (400 MHz, 298K, CD₂Cl₂, key resonances only): δ .3.37 (3.22) (m, 1H, γ -H), 2.07 (1.92) (m, 1H, α -H), 1.27 and 1.00 (1.45 and 1.24) (m each, 1:1H, β -H₂), 1.17 (1.32) (d, ³J_{HH} = 6.4 (7.2) Hz, γ -CH₃).

{(κ²-*P*,*O*)-2-[1,3-di(2,6-diisopropyl)phenyl-1,3,2-diazaphospholidin-2-yl) benzenesulfonato]-palladium(II)-phenyl} lithium chloride adduct [6b-LiCl•(DMSO)]

To **L1b**•(**Et**₂**O**) (65 mg, 100 μ mol) and [(DMSO)₂Pd(Ph)Cl] (previous step; 39.8 mg, 106 μ mol) in a NMR tube was added methylene chloride- d_2 (400 μ L). The tube was shaken for 5 min at 298 K and then centrifuged. Monitoring of the solution by ¹H and ³¹P NMR indicated consumption of **L1b**•(**Et**₂**O**) and formation of one new ³¹P containing species in > 90 % NMR yield. The solution was filtrated, concentrated to dryness, and the resulting solid extracted with pentane (2 × 3 mL) to yield complex **6b-LiCl**•(**dmso**) after drying under vacuum (66.7 mg, 72 μ mol, 72 %).



¹H NMR (600 MHz, 298K, acetone- d_6): δ 8.25 (vt, J = 8.7 Hz, 1H, 6-H), 7.90 (m, 1H, 3-H), 7.77 (vt, J = 7.7 Hz, 1H, 5-H), 7.66 (vt, J = 7.6 Hz, 1H, 4-H), 7.02 (br., 4H, 9-,11-,23-, and 25-H), 6.90 (br., 2H, 10-, and 24-H), 6.57 (v. br., 2H, 34- and 38-H), 6.28 (m, 1H, 36-H), 6.23 (br., 2H, 35-, and 37-H); 4.07, 3.74, and 2.98 (v. br.:br.: v. br., 4:2:2H, 13-, 16-, 27-, 30-H, and 19- and 20-H₂), 2.51 (s, 6H, DMSO), 1.66, 1.29, 1.12, and 0.83 (br.:d:d:br., ${}^{3}J_{HH}$ = 6.8 Hz, 6:6:6:6H, 14-, 15-, 17-, 18-, 28-, 29-, 31-, and 32-H₃). ¹³C NMR (150 MHz, 298K, acetone- d_6): δ 148.4 (C_a, d br., ² J_{PC} =18 Hz, C1), 147.9 and 147.45 (C_q each, C8, C12, C22, and C26), 137.2 (CH, C34 and C38), 136.8 (C_q br., d, ${}^{1}J_{PC}$ = 27 Hz, C2), 136.78 (C_q, C33), 135.8 (C_q br., C7 and C12), 130.4 (CH, d, ${}^{3}J_{PC}$ = 1.4 Hz, C4), 129.8 (CH br., C3), 129.46 (CH, d, ${}^{4}J_{PC}$ =5.0 Hz, C5), 128.7 (CH br., C6), 127.47 and 124.5 (CH each, C9, C11, C23, and C25), 125.26 (CH, d, ⁴J_{PC} =3.9 Hz, C35 and C37), 123.8 (CH, C10 and C24), 121.7 (CH, C36), 54.7 (CH₂ br, C19 and C20), 40.18 (CH₃, DMSO), 29.46 and 28.79 (no correlation detected, C13, C16, C27, and C30), 27.24, 26.24, 23.32, and 21.74 (CH₃, C14, C15, C17, C18, C28, C29, C31, and C32). ³¹P NMR (161 MHz, 298K, acetone*d*₆): δ 88.95. Anal Calc (%) for C₄₃H₅₉N₂O₅PS₂PdLiCl (927.87): C: 55.66; H: 6.41; N: 3.02. Found C: 55.99; H: 6.87; N: 2.55. Crystals of 6b-LiCl (1.8 acetone, 0.2 DMSO) suitable for X-ray diffraction analysis were grown from 6b-LiCl•(DMSO) (12 mg) in acetone (60 μ L) after layering with pentane.

{[(\kappa^2-P,O)-2-(1,3-di(2,6-diisopropyl)phenyl-1,3,2-diazaphospholidin-2-yl) benzenesulfonato]palladium(II)-[(\kappa^2-C,O)-3-methoxy-2-phenyl-3-oxopropyl]} (7b) To a solution complex **6b-LiCI**•(**DMSO**) (38 mg, 41 µmol) in methylene chloride- d_2 in a NMR tube was added methyl acrylate (88 mg, 1.03 mmol, 25 equiv). The mixture was monitored by ¹H and ³¹P NMR, then silver triflate (10.8 mg, 42 µmol, 1.02 equiv) was added and the tube was shaken for 2 min and centrifuged whereby silver chloride deposited. The resulting solution was monitored by ¹H and ³¹P NMR at 318 K. Formation of complex **7b** (ca 90 % NMR yield by ¹H and ³¹P NMR) together with methyl cinnamate (**5**) (ca 7% NMR yield) was complete after ca 25 min at 318 K. Isolation of complex **7b** was accomplished by filtration, removing all volatiles under vacuum, washing the residue with pentane (4 × 2 mL) and cold diethyl ether (1 mL, 243 K), and crystallization from methylene chloride (100 µL) after layering with pentane (1.5 mL) in a NMR tube (28.3 mg, 33.9 µmol, 82.6 %). The full NMR characterization was obtained from the crude reaction mixture, after filtration, removal of all volatiles and redissolution acetone- d_6 .



¹H NMR (600 MHz, 298K, acetone- d_6): δ 8.22 (vt, J = 8.4 Hz, 1H, 6-H), 7.99 (m, 1H, 3-H), 7.69 (vt, J = 7.3 Hz, 1H, 5-H), 7.61 (vt, J = 7.5 Hz, 1H, 4-H), 7.20-6.96 (m, 11H, 9-11-, 23-25-, and 34-38-H), 4.24 and 3.98 (m each, 2:2H, 19- and 20-H₂), 3.75 (s, 3H, 36-H₃), 3.72, 3.63, 2.97, and 2.92 (sept each, ${}^{3}J_{HH}$ = 6.8 Hz, 1:1:11H, 13-, 16-, 27-, and 30-H), 3.72 (m, 1H, 34-H), 1.51, 1.36, 1.33, 0.96,0.92, 0.77, and 0.60 (d each, ${}^{3}J_{HH}$ = 6.8 Hz, 3:6:3:3:3:3:3H, 14-, 15-, 17-, 18-, 28-, 29-, 31-, and 32-H₃), 1.42 (m, 1H, 33-H), 1.16 (vt, J = 9.4 Hz, 1H, 33-H). ¹³C NMR (150 MHz, 298K, acetone d_6): δ 188.17 (C_q, d ${}^3J_{PC}$ = 3.0 Hz, C35), 149.92 (C_q, ${}^2J_{PC}$ = 28.5 Hz, C1), 149.30, 149.22, 148.08, and 148.03 (C_q each, d each, ${}^3J_{PC}$ = 2.6, 2.4, 3.6, and 3.9 Hz, C8, C12, C22, and C26), 139.97 (C_q, d, ${}^{2}J_{PC}$ = 2.4 Hz, C37), 139.26 and 139.22 (C_q each, d each, ${}^{2}J_{PC}$ = 11.4 Hz, C7 and C21), 138.71 (C_q, d, ${}^{1}J_{PC}$ = 60.6 Hz, C2), 131,4 (CH, d, ${}^{3}J_{PC}$ = 3.6 Hz, C4), 130.56 (CH, C6), 129.85 (CH, d, ${}^{4}J_{PC}$ = 7.5 Hz, C5), 129.53 (CH, d, ²*J*_{PC} = 12.6 Hz, C3), 129.79, 129.20, 129.11, 129.00, 128.71, 128.39, 127.78, 125.89, 125.67, 125.29, and 125.17 (CH each, C9-C11, C23-C25, and C38-C42), 56.26 (CH, d, ${}^{3}J_{PC}$ = 3.8 Hz, C34), 55.66 (CH₃, C36), 54.57 and 54.44 (CH₂ each, C19 and C20), 29.53 and 29.41 (CH each, C13, C16, C27, and C30), 26.93, 26.90, 25.77, 25.67, 25.06, 24.57, 24.22, and 23,71 (CH₃ each, C14, C15, C17, C18, C28, C29, C31, and C32), 18.69 (CH₂, d, ²J_{PC} = 7.7 Hz, C33). ³¹P NMR (161 MHz, 298K, acetone-d₆): δ 91.22. Anal Calc (%) for C₄₂H₅₃N₂O₅PSPd (835.35): C: 60.39; H: 6.39; N: 3.35. Found: C: 61.02; H: 6.81; N: 3.01.

(2-Phenyl)methyl acrylate (8)

A solution of **7b** (10.0 mg, 12 µmol) in 1,1,2,2-tetrachloroethane-*d*₂ (0.45 ml, residual protiated 1,1,2,2-tetrachloroethane was used as internal standard) was heated in a NMR tube at 363 K and occasionally monitored by ¹H NMR spectroscopy over 12 h. Consumption of **7b** was complete after 11 h at 363 K to give in quantitative NMR yield compound **8** identified by 1- and 2-D NMR experiments and by comparison to genuine **8** obtained by Wittig reaction of methyltriphenylphosphonium bromide and benzoyl methyl formate according to a literature procedure (4). ¹H NMR (400 MHz, 298K, 1,1,2,2-tetrachlorethane-*d*₂): δ 7.43-7.36 (m, 5H, Ph),6.35 and 5.93 (d each, ²J_{HH} = 1.4 Hz each, 1:1H, C=CH₂), 3.86 (s, 3H, OCH₃). ¹³C NMR (100 MHz, 298K, 1,1,2,2-tetrachlorethane-*d*₂): δ 167.38 (C_q, CO₂CH₃), 141.00 and 136.65 (C_q each, *ipso*-Ph and C=CH₂), 128.47 and 128.30 (CH each, *ortho*- and *meta*-C, Ph), 128.36 (CH, *para*-C, Ph), 127.33 (CH₂, C=CH₂), 52.50 (CH₃, OCH₃).

X-Ray Diffraction Analyses

X-Ray diffraction analyses were performed at 100 K on a STOE IPDS-II diffractometer equipped with a graphite-monochromated radiation source (λ = 0.71073 Å) and an image plate detection system. Crystals were mounted on a fine glass fiber with silicon grease. The selection, integration and averaging procedure of the measured reflex intensities, the determination of the unit cell dimensions and a least-squares fit of the 2θ values as well as data reduction, LP-correction and space group determination were performed using the X-Area software package delivered with the diffractometer (5). A semiempirical absorption correction was performed. The structures were solved by the Patterson method (SHELXS-97) (6), completed with difference fourier syntheses, and refined with full-matrix least-square using SHELXL-97 (7) minimizing $\omega (F_0^2 - F_c^2)^2$. Weighted R factor (*wR*₂) and the goodness of fit GooF are based on F^2 . Except for complex **6b-LiCl**•(**1.8 acetone**, **0.2 DMSO**), all non-hydrogen atoms were refined with anisotropic displacement parameters. For complex 6b-LiCl•(1.8 acetone, 0.2 DMSO), positional and substitutional disordered solvent molecules were refined isotropically. All hydrogen atoms were treated in a riding model. Graphical output (Ortep plots) were created using ORTEP-3 V2.02 for Windows XP (8).

Table S1. X-Ray diffraction data of complexes 2a-DMSO, 2a-LiCl•(MeOH), 2b-LiCl•(2 THF), 3a, 3b, 4-(MeOH)•(MeOH), and 6b-LiCl•(acetone/DMSO)

	2a-DMSO	2a-LiCl•(MeOH)	2b-LiCl•(2 THF)	3a	3b	4-(MeOH)•(MeOH)	6b-LiCl•(acetone/DMSO)
CCDC no.	792914	793498	792913	792915	792916	792918	792917
emp. formula	$C_{29}H_{39}N_2O_4PPdS_2$	$C_{56}H_{74}Cl_{2}Li_{2}N_{4}O_{8}P_{2}Pd_{2}S_{2} \\$	$C_{82}H_{122}Cl_2Li_2N_4O_{10}P_2Pd_2S_2$	C31H39N2O5PPdS	C37H51N2O5PPdS	C ₂₈ H ₃₁ O ₇ PPdS	C43.79H59ClLiN2O5PPdS1.21
formula weight	681.11	1354.83	1747.48	689.07	773.23	648.96	912.03
[g mol ⁻¹]							
temperature [K]	100 K	100 K	100 K	100 K	100 K	100 K	100 K
crystal system	Monoclinic	monoclinic	triclinic	orthorhombic	monoclinic	Triclinic	monoclinic
space group	P21/c	C2/c	P-1	Pbca	P21/c	P-1	P21/c
a [Å]	9.5065(6)	27.7024(15)	10.462(2)	15.1027(6)	13.2909(7)	8.2486(10)	16.9666(7)
<i>b</i> [Å]	21.7563(9)	10.5058(4)	12.917(3)	16.6125(5)	15.1835(10)	11.8255(14)	10.3382(3)
<i>c</i> [Å]	15.1499(10)	20.3957(12)	16.146(6)	24.1809(8)	22.0751(12)	15.613(2)	25.0043(10)
<i>α</i> [°]	90.00	90.00	101.91(2)	90.00	90.00	69.380(9)	90.00
β [°]	104.582(5)	100.256(5)	99.03(2)	90.00	121.621(4)	77.961(10)	91.511(3)
γ[°]	90.00	80.00	93.739(16)	90.00	90.00	77.384(10)	90.00
Volume [Å ³], Z	3032.5(3), 4	5841.0(5), 4	2097.8(9), 1	6066.8(4), 8	3793.4(4), 4	1376.4(3), 2	4384.3(3), 4
density [calc,g cm ⁻³]	1.492	1.541	1.383	1.509	1.354	1.566	1.382
abs coeff [mm ⁻¹]	0.839	0.890	0.638	0.776	0.629	0.853	0.624
F(000)	1408	2784	916	2848	1616	664	1905
crystal size [mm ³]	0.15 x 0.133 x 0.1	0.24 x 0.1 x 0.1	0.5 x 0.4 x 0.3	0.4 x 0.277 x 0.08	0.35 x 0.24 x 0.12	0.45 x 0.23 x 0.04	0.48 x 0.33 x 0.18
θ range [°]	1.67 - 28.07	4.59 - 25.78	2.43 - 27.32	2.01 - 25.85	1.96 - 27.01	1.86 - 26.85	1.63 - 27.42
hkl-range	-12,12; -28,28; -19,18	-33,33;-12,12;-24,24	-13,13; -16,16; -20,20	-18,18; -20,19; -29,29	-16,16; -19,19; -27,26	-10,10; -14,14; -19,19	-21,21; -12,13; -32,32
no. of rflns collected	36387	35104	27982	69137	50973	18105	61077
no. of ind rflns, $R_{\rm int}$	7279, 0.1045	5505, 0.1532	9362, 0.0632	5766/0.058	8092/0.0799	5791/0.1087	9893/0.0416
abs correction	Integration	integration	integration	integration	integration	integration	integration
weighting parameters a, b ^[A]	0.0355, 0	0.0428, 17.3928	0.0259, 0.9447	0.0216, 15.7103	0.0374, 0	0.0363, 9.7341	0.0348, 7.4081
GOF on F^2	0.980	1.123	1.029	1.135	0.967	1.089	1.021
R_1 , wR_2 (all data)	0.0784/0.0836	0.0588/0.1143	0.0543/0.0672	0.0465/0.0785	0.0634/0.0754	0.0934/0.1590	0.0462/0.0892
$R_1, wR_2 (I > 2\sigma(I))$	0.0447/0.0765	0.0477/0.1095	0.0349/0.0633	0.0365/0.0762	0.0380/0.0704	0.0620/0.1325	0.0357/0.0791
largest diff peak/hole [e/Å ³]	1.088/-0.828	1.480/-1.644	0.411/-0.770	1.346/-0.538	0.405/-0.586	1.236/-1.832	1.760/-1.115
remarks			positional disordered thf molecules		Positional disordered methyl group	positional disordered MeOH molecule	positional and substitutional disordered solvent molecules

^[A]: weighting scheme: $w = 1/[\sigma^2(F_o^2) + (a \cdot P)^2 + b \cdot P)$, $P = [max(F_o^2, 0) + 2F_c^2]/3$.

Figure S1. ORTEP plot of complex **2a-DMSO**. Ellipsoids are shown with 50% probability. Hydrogen atoms are omitted for clarity.



Figure S2. ORTEP plot of **2a-LiCl**•(**MeOH**). Ellipsoids are shown with 50% probability. Hydrogen atoms are omitted for clarity.



Figure S3. ORTEP plot of **2b-LiCl**•(**2 THF**). Ellipsoids are shown with 50% probability. Hydrogen atoms as well as disordered THF molecules are omitted for clarity.



Figure S4. ORTEP plot of **3a**. Ellipsoids are shown with 50% probability. Hydrogen atoms are omitted for clarity.



Figure S5. ORTEP plot of **3b**. Ellipsoids are shown with 50% probability. Hydrogen atoms as well as the disordered methyl position of C37 (C38) are omitted for clarity.



Figure S6. ORTEP plot of **6b-LiCl**•(**1.8 acetone**, **0.2 dmso**). Ellipsoids are shown with 50% probability. Hydrogen atoms as well as positional and substitutional disordered carbon and sulphur positions (O4, O5 are part of the acetone- and/or dmso-molecules, respectively) are omitted for clarity.



Figure S7. ORTEP plot of **4-(MeOH)**•(**MeOH**). Ellipsoids are shown with 50% probability. Hydrogen atoms and one further MeOH molecule are omitted for clarity.



References

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Figure S8: ¹H NMR (400 MHz, 298 K, acetone-*d*₆) of **2a-LiCl**



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Figure S10: ³¹P NMR (161 MHz, 298 K, acetone-*d*₆) of 2a-LiCI





Figure S12: ¹H,¹³C gHSQC NMR (400 MHz, 298 K, acetone- d_6) of **2a-LiCI** (CH and CH₃ red, CH₂ blue)



f1 (ppm)

Figure S13: ¹H,¹³C gHMBC NMR (400 MHz, 298 K, acetone-*d*₆) of 2a-LiCl



f1 (ppm)



Figure S15: ¹³C NMR (100 MHz, 298 K, acetone-d₆) of 2b-LiCl·(0.5 Et₂O)



Figure S16: ³¹P NMR (161 MHz, 298 K, acetone-*d*₆) of 2b-LiCl·(0.5 Et₂O)





f1 (ppm)

Figure S17: ¹H, ¹H COSY NMR (400 MHz, 298 K, acetone-*d*₆) of 2b-LiCl·(0.5 Et₂O)

Figure S18: ¹H,¹H NOEOSY NMR (400 MHz, 298 K, acetone- d_6) of **2b-LiCl·(0.5 Et₂O)** (red cross-peaks: chemical exchange, blue cross-peaks: NOE)



(mqq) 11



Figure S19: ¹H, ¹³₂₂C gHMQC NMR (400 MHz, 298 K, acetone-*d*₆) of 2b-LiCl·(0.5 Et₂O)

f1 (ppm)





f1 (ppm)

Figure S21: ¹H NMR (400 MHz, 298 K, methylene chloride-d₂) of crude **3a** after vaccum-removal of excess methyl acrylate





Figure S22: ¹³C NMR (100 MHz, 298 K, methylene chloride-d₂) of crude 3a after vaccum-removal of excess methyl acrylate

Figure S23: ³¹P NMR (161 MHz, 298 K, methylene chloride-d₂) of crude 3a after vaccum-removal of excess methyl acrylate



Figure S24: ¹H,¹H COSY NMR (400 MHz, 298 K, methylene chloride-*d*₂) of crude **3a** after vaccum-removal of excess methyl acrylate


Figure S25: ¹H,¹³C gHSQC NMR (400 MHz, 298 K, methylene chloride- d_2) of crude **3a** after vaccum-removal of excess methyl acrylate (CH and CH₃: red; CH₂: blue





Figure S26: ¹H,¹³C gHMBC NMR (400 MHz, 298 K, methylene chloride- d_2) of crude **3a** after vaccum-removal of excess methyl acrylate

Figure S27: ¹H NMR (400 MHz, 298 K, methylene chloride- d_2) of crude **3b** after removal of of excess MA under vacuum





Figure S28: ¹³C NMR (100 MHz, 298 K, methylene chloride- d_2) of crude **3b** after removal of of excess MA under vacuum

Figure S29: ³¹P NMR (161 MHz, 298 K, methylene chloride- d_2) of crude **3b** after removal of of excess MA under vacuum



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Figure S30: aliphatic region of ¹H,¹H COSY NMR (400 MHz, 298 K, methylene chloride- d_2) of crude **3b** after removal of of excess MA under vacuum



Figure S31: ¹H,¹³C gHSQC NMR (400 MHz, 298 K, methylene chloride- d_2) of crude **3b** after removal of of excess MA under vacuum, CH and CH₃: red, CH₂: blue



Figure S32: ¹H,¹³C gHMBC NMR (400 MHz, 298 K, methylene chloride-*d*₂) of crude 3b after removal of of excess MA under vacuum



Figure S33: ¹H NMR (400 MHz, 298 K, methylene chloride-d₂) of X-Ray quality 3b after washing with pentane







Figure S3*: ³¹P NMR (161 MHz, 298 K, CD₃OD) of 4-MeOH



Figure S37: Reaction of complex **4-MeOH** and methyl acrylate in CD₃OD: Formation of methyl cinnamate (**5**) and complexes **9** and **10**,**10**[•] [¹H NMR (400 MHz, 298 K, CD₃OD)]



Figure S3, : Reaction of complex **4-MeOH** and methyl acrylate in CD_3OD : Formation of methyl cinnamate (**5**) and complexes **9** and **10,10**^{\cdot} (sample after 2.5 reaction time, removal of volatiles and redissolution in CD_2CI_2)



Figure S' - : Reaction of complex **4-MeOH** and methyl acrylate in CD_3OD : Formation of methyl cinnamate (**5**) and complexes **9** and **10**,**10**^{\cdot} (sample after 2.5 reaction time, removal of volatiles and redissolution in CD_2Cl_2)



Figure S40: ¹H NMR (400 MHz, 298 K, acetone-*d*₆) of **6b-LiCl·(DMSO)**





Figure S4& ³¹P NMR (161 MHz, 298 K, acetone-*d*₆) of 6b-LiCl·(DMSO)



Figure S4' : 1 H, 13 C gHSQC NMR (400 MHz, 298 K, acetone- d_{6}) of **6b-LiCl·(DMSO)**



Figure S4(: ¹H,¹³C gHMBC NMR (400 MHz, 298 K, acetone-*d*₆) of **6b-LiCl·(DMSO)**



Figure S4): ¹H NMR (600 MHz, 298 K, acetone- d_6) of crude **7b** obtained from **6b-LiCI·(DMSO)**, 1.02 equiv AgOTf and 25 equiv methyl acrylate after 60 min at 318 K in methylene chloride- d_2 , removal of all volatiles under vacuum, and redissolution in acetone- d_6 (contains ca. 7 % methyl cinnamate)



Figure S46: ¹³C NMR (150 MHz, 298 K, acetone- d_6) of crude **7b** obtained from **6b-LiCl·(DMSO)**, 1.02 equiv AgOTf and 25 equiv methyl acrylate after 60 min at 318 K in methylene chloride- d_2 , removal of all volatiles under vacuum, and redissolution in acetone- d_6 (contains ca. 7 % methyl cinnamate) C9-11+C23-25+C38-42



Figure S47: ³¹P NMR (161 MHz, 298 K, acetone- d_6) of crude **7b** obtained from **6b-LiCI·(DMSO)**, 1.02 equiv AgOTf and 25 equiv methyl acrylate after 60 min at 318 K in methylene chloride- d_2 , removal of all volatiles under vacuum, and redissolution in acetone- d_6 (contains ca. 7 % methyl cinnamate)

91.22





Figure S4, : ¹H,¹³C gHSQC NMR (600 MHz, 298 K, acetone- d_6) of crude **7b** obtained from **6b-LiCl·(DMSO)**, 1.02 equiv AgOTf and 25 equiv methyl acrylate after 60 min at 318 K in methylene chloride- d_2 , removal of all volatiles under vacuum, and redissolution in acetone- d_6 (contains ca. 7 % methyl cinnamate); CH and CH₃: red; CH₂: blue



Figure S(- : ¹H,¹³C gHMBC NMR (600 MHz, 298 K, acetone- d_6) of crude **7b** obtained from **6b-LiCI·(DMSO)**, 1.02 equiv AgOTf and 25 equiv methyl acrylate after 60 min at 318 K in methylene chloride- d_2 , removal of all volatiles under vacuum, and redissolution in acetone- d_6 (contains ca. 7 % methyl cinnamate)







⁻ S63 -

Figure S5& ³¹P NMR (161 MHz, 298 K, 1,1,2,2-tetrachloroethane-d₂) of isolated 7b



Figure S5': aliphatic region of the ¹H,¹H COSY NMR (400 MHz, 298 K, 1,1,2,2-tetrachloroethane-*d*₂) of isolated 7b



Figure S54: top line:¹H NMR (400 MHz, 298 K, 1,1,2,2-tetrachloroethane- d_2) of isolated **7b** middle line:¹H NMR (400 MHz, 363 K, 1,1,2,2-tetrachloroethane- d_2) of isolated **7b** (after 5 min 363 K) bottom line:¹H NMR (400 MHz, 298 K, 1,1,2,2-tetrachloroethane- d_2) of isolated **7b** (after 12 h 363 K)



Figure S55: time dependent ¹H NMR spectra (400 MHz, 363 K, 1,1,2,2-tetrachloroethane- d_2 , aromatic and olefinic region) indicating complete consumption of **7b** and formation of (2-phenyl)methyl acrylate after ca 11 h at 363 K





Figure S56: ¹H NMR (400 MHz, 298 K, 1,1,2,2-tetrachloroethane-*d*₂) of thermolyzed 7b containing (2-phenyl)methyl acrylate (8)

Figure S57: ¹³C NMR (100 MHz, 298 K, 1,1,2,2-tetrachloroethane- d_2) of thermolyzed **7b** containing (2-phenyl)methyl acrylate (8)



Figure S58: ¹H,¹³C gHSQC NMR (400 MHz, 298 K, 1,1,2,2-tetrachloroethane- d_2) of thermolyzed **7b** containing (2-phenyl)methyl acrylate (**8**); CH and CH₃: red; CH₂: blue



Figure S59: ¹H,¹³C gHMBC NMR (400 MHz, 298 K, 1,1,2,2-tetrachloroethane- d_2) of thermolyzed **7b** containing (2-phenyl)methyl acrylate (**8**)



Theoretical Procedures and DFT results

Computational Details. The Amsterdam Density Functional (ADF) program (9,10) was used. The electronic configuration of the molecular systems was described by a triple- ζ STO basis set on Pd (ADF basis set TZV). Double- ζ STO basis sets, augmented by one polarization function, were used for main group atoms (ADF basis sets DZVP). The inner shells on Pd (including 3d), P and S (including 2p) C and O (1s), were treated within the frozen core approximation. Energies and geometries were evaluated using the local exchange-correlation potential by Vosko et al. (11), augmented in a self-consistent manner with Becke's (12) exchange gradient correction and Perdew's (13,14) correlation gradient correction (BP86 functional). The transition states were approached through a linear transit procedure starting from the coordination intermediates. The forming C-C bond was assumed as reaction coordinate during the linear transit scans. Full transition state searches were started from the maxima along the linear transit paths.

Topographic steric maps. The steric maps have been calculated for the phosphine sulfonato and diazaphospholidine sulfonato ligands in the DFT optimized geometry of the **1-DMSO**, **2a-DMSO** and **2b-DMSO** complexes. The points in space defining the steric map were located with the SambVca package developed by some of us (15). This program analyzes the first coordination sphere around the metal, which is the place where the reaction occurs. It is normally used to calculate the buried volume of a given ligand, which is a number that quantifies the amount of the first coordination sphere of the metal occupied by this ligand. We modified SambVca to understand how the ligand is distributed around the metal, which is the shape of the reactive pocket. We already introduced topographic steric maps to characterize Ru-complexes relevant to olefin metathesis (16), and Rh-complexes active in the asymmetric 1,4-addition of phenylboronic acid to 2-cyclohexenone (17).

To build the steric map, the complex under analysis has been placed with the Pd center at the origin, with the axis bisecting the S-Pd-P angle aligned along the z-axis, and with the S and P atoms in the xz plane at negative z values. After this alignment step the Me and DMSO groups have been removed (because they do not belong to the **1**, **2a** and **2b** ligands) and the first coordination sphere around the metal is analyzed. This sphere, of radius R, is sectioned by a regular 3D cubic mesh of spacing s, which defines cubic voxels v. The distance between the centre of each voxel with all the atoms in the ligand is tested to check if any of the atoms is within a van der Waals distance from the centre of the examined voxel. If no atom is within a van der Waals distance, the examined voxel is marked as a free voxel. Otherwise, the examined voxel is marked as buried.
After all the voxels in the first coordination sphere have been marked as free or buried, for each (x,y) point within the first coordination sphere the program scans the sphere from the top (i.e. away from the ligand) to find at which z value there is the first buried voxel. This procedure results in a surface, defined as $S(x,y) = z_B$, which represents the surface of the ligand that is exposed towards the incoming reactants. In other words, this $S(x,y) = z_B$ surface defines the shape of the reactive pocket. Positive values of z_B indicate that the ligands protrudes in the z > 0 half-sphere, which is the half-sphere where the reacting groups are placed. A schematic representation of the interaction surface between the complex and the substrate is shown in Figure S60.

Finally, the maps are a simple 2D isocontour representation of the surface of interaction $S(x,y) = z_B$. In this work, the radius *R* of the sphere around the metal center was set to 3.5 Å, while for the atoms we adopted the Bondi radii (18) scaled by 1.17, and a mesh of 0.1 Å was used to scan the sphere for buried voxels.



Figure S60. Schematic representation of the interaction surface between the substrate and the complex.

DFT Calculations of reaction pathways

fragment 1



Figure S61. 2,1- and 1,2-insertion of methyl acrylate (MA) into the Pd methyl (Pd-Me) bond of fragment **1**. Energies in kJ mol⁻¹.



Figure S62. 2,1- and 1,2-insertion of methyl acrylate (MA) into the Pd methyl (Pd-Me) bond of fragment **2a**. Energies in kJ mol⁻¹.

fragment **2b** (R = 2,6-di(isopropyl)phenyl)



Figure S63. 2,1- and 1,2-insertion of methyl acrylate (MA) into the Pd methyl (Pd-Me) bond of fragment **2b**. Energies in kJ mol⁻¹.



Figure S64. Geometry of the 1,2- (a, f, j) and 2,1-transition state (b, g, k) for methyl acrylate insertion into the Pd-Me bond of **1** (top line), **2a** (middle line), and **2b** (bottom line). Steric (c, g, l) maps of **1**, **2a** and **2b** with steric isocontour levels in Å.

Cartesian coordinates and energies of the DFT structures









1MA-isom

1MA-trans







1MA-2,1-inser

1MA-2,1-KP

1MA-2,1-TP







1MA-1,2-TP

1MA-1,2-inser

1MA-1,2-KP

1

E = -11.10574744 a.u.

Pd	0.00000	0.00000	0.00000
P	2.25186	0.00000	0.00000
С	3.10754	0.21885	1.60846
С	3.07913	-1.44587	-0.77222
С	2.82483	1.43493	-1.01576
С	2.91451	-0.70461	2.66095
0	2.14909	-1.79059	2.36968
С	1.86023	-2.71104	3.42763
Н	1.20839	-3.47033	2.98388
Н	2.78099	-3.18647	3.79791
Н	1.33662	-2.20465	4.25188
С	3.51318	-0.48664	3.90318
Н	3.36768	-1.19267	4.71744
С	4.29810	0.64750	4.10694
Н	4.75360	0.80836	5.08458
С	4.49066	1.56781	3.08224
Н	5.09039	2.46253	3.24333
С	3.89002	1.35265	1.84402
Н	4.01939	2.08897	1.05219
С	2.61335	-1.91640	-2.02350

0	1.61879	-1.19110	-2.59392
С	1.12139	-1.60781	-3.87181
H	0.34641	-0.88015	-4.13097
H	0.68534	-2.61555	-3.81154
Н	1.92100	-1.58237	-4.62659
С	3.18539	-3.05293	-2.60046
Н	2.83135	-3.42361	-3.56000
С	4.22464	-3.71547	-1.94747
Н	4.66444	-4.59908	-2.41155
С	4.70153	-3.25513	-0.72363
Н	5.52036	-3.76713	-0.21916
С	4.12351	-2.12600	-0.14319
Н	4.49983	-1.75855	0.81175
С	3.95841	1.26531	-1.82225
Н	4.47962	0.30807	-1.82886
С	4.43632	2.30659	-2.61443
Н	5.32245	2.15022	-3.22984
С	3.78289	3.53443	-2.61320
Н	4.15002	4.35513	-3.23078
С	2.65119	3.71714	-1.82195
Н	2.11731	4.66757	-1.80442
С	2.16259	2.68181	-1.02476
S	0.69542	3.05214	0.00000
0	0.32496	4.42832	-0.31985
0	1.12019	2.73932	1.37004
0	-0.33362	2.05949	-0.52977
С	-0.10535	-2.01390	0.35996
Н	0.79785	-2.62203	0.23939
Н	-0.86255	-2.29288	-0.39718
Н	-0.50103	-2.10157	1.38404

1MA-cis

E= -13.69111340 a.u.

0.00000 0.00000 1.41306 -0.97540 2.67146 -0.97730 3.01069 0.00000 2.08800 -0.65758 2.57675 1.36519 4.41577 -0.23536 0.02699 1.69409 1.47821 -0.80012 3.70540 -1.76674 3.49860 -2.56828
1.41306 -0.97540 2.67146 -0.97730 3.01069 0.00000 2.08800 -0.65758 2.57675 1.36519 4.41577 -0.23536 0.02699 1.69409 1.47821 -0.80012 3.70540 -1.76674 3.49860 -2.56828
2.67146 -0.97730 3.01069 0.00000 2.08800 -0.65758 2.57675 1.36519 4.41577 -0.23536 0.02699 1.69409 1.47821 -0.80012 3.70540 -1.76674 3.49860 -2.56828
3.01069 0.00000 2.08800 -0.65758 2.57675 1.36519 4.41577 -0.23536 0.02699 1.69409 1.47821 -0.80012 3.70540 -1.76674 3.49860 -2.56828
2.08800 -0.65758 2.57675 1.36519 4.41577 -0.23536 0.02699 1.69409 1.47821 -0.80012 3.70540 -1.76674 3.49860 -2.56828
2.57675 1.36519 4.41577 -0.23536 0.02699 1.69409 1.47821 -0.80012 3.70540 -1.76674 3.49860 -2.56828
4.41577 -0.23536 0.02699 1.69409 1.47821 -0.80012 3.70540 -1.76674 3.49860 -2.56828
0.026991.694091.47821-0.800123.70540-1.766743.49860-2.56828
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L.21759 -1.79583
2.04447 0.39259
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4.31324 -3.18373
2.07795 -3.21782
0.24609 -1.81134
2.29426 1.21676
0.01473 0.11555
0.98692 -0.32522
1.18340 -0.56396
0.10253 1.38829
1.09353 1.84779
0.76754 2.02919
2.50127 -0.53992
2.34204 -0.15409
3.48977 -0.78920
3.78174 -2.08411
2.94788 -2.74560
L.80497 -2.10470
2.11564 0.86118
4.14835 -0.27020

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2.83698	-3.19732	-3.75367
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4.44957	0.26436	1.96240
4.90237	0.31390	3.28427
4.00936	0.12269	4.33639
2.66030	-0.11013	4.08722
2.20860	-0.14360	2.77098
5.26307	0.40566	0.88317
5.95161	0.50258	3.49966
4.37984	0.16596	5.36110
1.95788	-0.24092	4.90940
1.14912	-0.27644	2.56026
6.59429	0.88742	1.10692
7.01504	1.05458	0.11004
6.57210	1.83538	1.66315
7.19896	0.14134	1.64375
1.31573	-1.18832	-3.97858
0.62190	-0.37147	-4.19744
2.15394	-1.16689	-4.69027
0.79027	-2.15274	-4.03672
-3.05030	0.93387	-1.88852
-3.15436	2.22489	-0.01131
-3.57717	2.07243	-2.61581
-3.67197	1.73456	-3.65157
-4.55227	2.36175	-2.20476
-2.88120	2.91560	-2.53263
-0.93066	-2.40259	0.62481
	4.43594 2.83698 1.77944 4.44957 4.90237 4.00936 2.66030 2.20860 5.26307 5.95161 4.37984 1.95788 1.14912 6.59429 7.01504 6.57210 7.19896 1.31573 0.62190 2.15394 0.79027 -3.05030 -3.15436 -3.57717 -3.67197 -4.55227 -2.88120 -0.93066	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$

1MA-isom

E= -13.67727321 a.u.

Pd	-1.274840	0.061542	0.500571
P	0.830411	-1.721757	0.390923
С	2.299808	-0.910653	-0.415657
С	2.433480	0.488671	-0.496241
S	1.183189	1.615384	0.176184
0	-0.023766	1.398113	-0.730515
0	0.870692	1.091600	1.532279
0	1.738082	2.958773	0.069756
С	1.567773	-2.283153	1.990461
С	0.731864	-3.274051	-0.604888
С	3.544585	1.062973	-1.115758
С	4.537586	0.254908	-1.662880
С	4.417294	-1.130224	-1.594738
С	3.306010	-1.703174	-0.981436
С	-2.706437	0.297606	-0.965649
Н	3.613022	2.149956	-1.153251
H	5.405580	0.709402	-2.141889
H	5.191898	-1.771927	-2.015626
H	3.223351	-2.789519	-0.933931
H	-2.935618	1.370313	-0.976489
Н	-2.145493	-0.018730	-1.851620
С	-2.636420	-1.077768	1.719262
H	-2.031257	-1.565834	2.484916
H	-3.242369	-1.719259	1.076556
С	-2.954180	0.277787	1.875881
H	-3.796676	0.738046	1.354738
С	-2.423484	1.025771	3.048338
Н	-3.592311	-0.322010	-0.785966
С	1.083787	-4.540761	-0.139767
С	0.925394	-5.674149	-0.940995
С	0.406083	-5.540068	-2.224802
С	0.050398	-4.283274	-2.718647
С	0.212019	-3.154459	-1.914032
Н	1.506766	-4.642763	0.859965
Н	1.214236	-6.653906	-0.561316
Н	0.277352	-6.416376	-2.861051

H	-0.348681	-4.193882	-3.726905
0	-0.086468	-1.883721	-2.305183
С	2.815389	-1.832282	2.430115
С	3.307693	-2.171063	3.690916
С	2.545949	-2.973039	4.532545
С	1.291341	-3.429210	4.126410
С	0.796712	-3.075625	2.870460
Н	3.414173	-1.195442	1.780036
Н	4.282213	-1.801362	4.008200
Н	2.917318	-3.246553	5.520657
Н	0.702088	-4.045461	4.801526
0	-0.424499	-3.468335	2.397351
С	-0.480490	-1.670293	-3.664887
Н	-0.574276	-0.585049	-3.772938
Н	0.289986	-2.042779	-4.356293
Н	-1.445833	-2.155111	-3.875104
С	-1.220447	-4.314004	3.235148
Н	-2.129614	-4.522411	2.662270
Н	-0.696906	-5.258178	3.450373
Н	-1.479471	-3.804997	4.175904
0	-1.777536	0.534573	3.956385
0	-2.758844	2.339607	2.963634
С	-2.196522	3.153116	4.023925
Н	-2.541158	4.169980	3.817848
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Н	-1.102189	3.091663	3.991496

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Ρ	2.51266	0.00000	0.00000
С	3.18648	1.63943	-0.56592
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S	0.67348	2.81588	0.00000
0	0.05495	2.00517	-1.12194
0	0.56474	2.02510	1.26848
0	0.25293	4.21011	0.05751
С	3.20330	-0.21488	1.69739
С	3.34183	-1.20400	-1.10596
С	2.97076	4.04785	-0.85402
С	4.25918	4.10967	-1.37718
С	4.99882	2.94065	-1.53057
С	4.46663	1.71921	-1.12530
С	-2.06084	0.32811	0.08855
H	2.35501	4.94130	-0.75425
H	4.67486	5.07100	-1.68121
H	5.99764	2.97581	-1.96661
H	5.04829	0.80831	-1.25073
H	-2.21541	1.14530	-0.62486
H	-2.66104	-0.55743	-0.15216
С	-0.38130	-1.92780	-0.84468
H	0.52496	-2.29375	-1.33567
H	-1.27723	-1.83619	-1.46160
С	-0.48940	-2.06809	0.54988
H	-1.46567	-2.03277	1.03959
С	0.56178	-2.79050	1.31174
H	-2.20555	0.65412	1.12810
С	4.57754	-0.22122	2.03231
С	4.97362	-0.33031	3.36895
С	4.01782	-0.40484	4.38050
С	2.66256	-0.36986	4.07305
С	2.27052	-0.27335	2.73982
0	5.45947	-0.12777	1.00184
Н	6.02890	-0.34160	3.63132
Н	4.34760	-0.47663	5.41759
Н	1.91031	-0.39934	4.86051

Н	1.21066	-0.21331	2.49847
С	4.12508	-2.27070	-0.66638
С	4.62661	-3.21020	-1.56729
С	4.34229	-3.08527	-2.92438
С	3.53962	-2.04229	-3.38940
С	3.02568	-1.11389	-2.48215
Н	4.33212	-2.37941	0.39629
Н	5.23299	-4.03972	-1.20476
Н	4.73536	-3.81121	-3.63713
Н	3.30954	-1.96619	-4.45034
0	2.19247	-0.09157	-2.81154
С	1.85390	0.10705	-4.18778
Н	1.19711	0.98235	-4.19578
Н	2.75537	0.30870	-4.78511
Н	1.32058	-0.76751	-4.58970
С	6.84848	0.00717	1.32115
Н	7.35979	0.12413	0.36008
Н	7.01854	0.89706	1.94472
Н	7.22644	-0.89261	1.82930
0	0.17361	-2.95319	2.60900
0	1.60829	-3.21731	0.85762
С	1.11750	-3.70535	3.41124
Н	0.66138	-3.77268	4.40320
Н	1.26609	-4.70208	2.97698
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E= -13.67046549 a.u.

Pd	0.00000	0.00000	0.00000
P	2.37377	0.00000	0.00000
С	3.08774	1.43951	-0.92804
С	2.47543	2.70726	-0.89898
S	0.92867	3.00877	0.00000
0	-0.02741	2.04493	-0.71381
0	1.17716	2.60090	1.38814
0	0.56215	4.39608	-0.27135
С	3.02277	3.76949	-1.61944
С	4.16853	3.58259	-2.38681
С	4.76515	2.32557	-2.44474
С	4.22788	1.26496	-1.72110
Н	2.51832	4.73452	-1.57333
Н	4.58626	4.41816	-2.94990
Н	5.65238	2.16468	-3.05799
Н	4.70414	0.28651	-1.77040
С	3.09074	0.04614	1.69311
С	3.14867	-1.44575	-0.82601
Н	-2.25175	-1.94295	0.00887
С	-1.56421	-1.49716	0.72979
С	-0.22434	-1.99637	0.68583
Н	-1.98361	-1.26656	1.71070
С	0.05837	-3.03004	-0.34603
С	-2.20934	0.44407	0.00273
Н	-2.16224	0.67450	-1.06887
Н	0.32526	-2.12223	1.62316
Н	-3.16122	-0.02720	0.24988
Н	-2.03322	1.32477	0.63083
С	4.46305	0.25488	1.95958
С	4.91433	0.34214	3.28004
С	4.01347	0.22456	4.33615
С	2.65789	0.02933	4.09114
С	2.20954	-0.04557	2.77530
0	5.28517	0.33707	0.88099
Н	5.96862	0.50669	3.49077
Н	4.38272	0.29894	5.35951
Н	1.94872	-0.03762	4.91515
Н	1.14469	-0.13583	2.56890
С	4.05603	-2.30397	-0.20478

С	4.57948	-3.40737	-0.88072
С	4.18540	-3.65965	-2.19162
С	3.25170	-2.83874	-2.82464
С	2.72612	-1.73702	-2.14473
Н	4.36099	-2.10902	0.82268
Н	5.29670	-4.05964	-0.38268
Н	4.59811	-4.51148	-2.73425
Н	2.93921	-3.05862	-3.84374
0	1.81734	-0.87484	-2.66398
С	1.19435	-1.21304	-3.91111
Н	0.47405	-0.41099	-4.09782
Н	1.93576	-1.23865	-4.72326
Н	0.67066	-2.17529	-3.82159
С	6.62847	0.78797	1.09216
Н	7.05760	0.90330	0.09138
Н	6.63283	1.75727	1.61092
Н	7.21005	0.04619	1.65976
0	-0.48701	-3.11824	-1.43590
0	0.96478	-3.92630	0.12630
С	1.19568	-5.05723	-0.74765
Н	1.98879	-5.63561	-0.26502
Н	0.27533	-5.64947	-0.83916
Н	1.51050	-4.71344	-1.73886

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Pd	0.00000	0.00000	0.00000
P	2.30597	0.00000	0.00000
C	2.94310	1.44173	-0.97388
C	2.29790	2.69474	-0.95077
S	0.77670	2.99532	0.00000
O	-0.19702	2.02971	-0.68454
O	1.07946	2.58674	1.37792
O	0.40125	4.38309	-0.25425
C	2.79203	3.75019	-1.71786
C	3.91165	3.57209	-2.52518
C	4.53786	2.32956	-2.57674
C	4.05657	1.27501	-1.80648
H	2.26601	4.70389	-1.67496
H	4.28530	4.40373	-3.12392
H	5.40435	2.17506	-3.22047
H	4.55601	0.30774	-1.84962
C	3.05090	0.07042	1.67544
C	3.05898	-1.45184	-0.82667
H	-1.90913	-0.28750	0.12571
C	-1.91744	-1.33385	0.62727
C	-0.52592	-1.88576	0.72117
H	-2.50371	-1.90639	-0.11086
C	-0.19752	-2.99103	-0.21063
C	-2.64326	-1.15258	1.96048
H	-2.03815	-0.56073	2.66167
Н	-0.14387	-2.05730	1.73523
Н	-2.83814	-2.13584	2.41472
Н	-3.60404	-0.63758	1.82125
С	4.41929	0.33510	1.90838
C	4.89551	0.44485	3.21828
C	4.02457	0.28823	4.29431
C	2.67463	0.02744	4.08073
C	2.19952	-0.06641	2.77586
О	5.21202	0.44333	0.81003
Н	5.94603	0.65525	3.40556
Н	4.41324	0.38185	5.30891
Н	1.98910	-0.07687	4.92072
H	1.13732	-0.21405	2.59035
C	3.99523	-2.28624	-0.21353
C	4.50319	-3.40373	-0.87664
C	4.06191	-3.69734	-2.16424

С	3.10013	-2.90141	-2.78617
С	2.59223	-1.78194	-2.12174
Н	4.33411	-2.05976	0.79687
Н	5.24328	-4.03627	-0.38683
Н	4.45923	-4.56294	-2.69637
Н	2.75044	-3.15526	-3.78501
0	1.66161	-0.94156	-2.63519
С	0.98602	-1.32895	-3.84150
Н	0.25766	-0.53423	-4.02826
Н	1.69262	-1.38632	-4.68237
Н	0.46809	-2.28649	-3.69023
С	6.53857	0.95591	0.98859
Н	6.94041	1.08332	-0.02194
Н	6.51014	1.92797	1.50142
Н	7.16459	0.24493	1.54819
0	-0.69555	-3.16625	-1.31581
0	0.72434	-3.82622	0.34381
С	1.05848	-4.97484	-0.46991
Н	1.81339	-5.52211	0.10247
Н	0.16423	-5.59097	-0.63365
Н	1.46220	-4.65257	-1.43631

1MA-2,1-TP

E= -13,72149122 a.u.

С	4.10550	1.22622	-1.76727
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С	2.35563	2.68263	-0.94851
С	2.88472	3.72586	-1.70886
С	4.01710	3.52384	-2.49307
С	4.62265	2.27017	-2.52870
P	2.28485	0.00000	0.00000
С	2.96820	-1.48607	-0.83577
С	3.79804	-2.41277	-0.20076
С	4.20121	-3.58151	-0.84820
С	3.76751	-3.83145	-2.14681
С	2.92921	-2.92883	-2.80247
С	2.52238	-1.76219	-2.15142
0	1.69683	-0.83099	-2.69162
С	1.22464	-1.02698	-4.03044
S	0.83636	3.01522	0.00000
0	0.47694	4.40160	-0.28756
Pd	0.00000	0.00000	0.00000
0	-2.18156	-0.59704	0.21515
С	-1.62769	-1.47183	0.93280
0	-2.08370	-1.79664	2.15628
С	-3.21877	-1.00811	2.61220
0	-0.15563	2.04773	-0.65036
С	-0.35532	-2.04529	0.50298
H	-0.41072	-2.41801	-0.53168
С	3.03130	0.02655	1.67520
С	4.41983	0.14903	1.90667
С	4.90564	0.21656	3.21569
С	4.02128	0.16872	4.29171
С	2.65032	0.06408	4.07849
C	2.16757	0.00567	2.77432
H	1.09505	-0.02424	2.58749
0	1.15120	2.63943	1.38454
0	5.21625	0.16927	0.80576
H	5.97259	0.31253	3.40323
H	4.41742	0.22579	5.30617
H	1.95682	0.05173	4.91826
H	4.13451	-2.21981	0.81756
H	4.85278	-4.28927	-0.33640
H	4.07848	-4./3954	-2.66474
H	2.59522	-3.14176	-3.81591
H	2.37896	4.69086	-1.67586
Н	4.42053	4.34692	-3.08423

Н	5.50076	2.10003	-3.15240
Н	4.58346	0.24772	-1.79825
Н	0.58880	-0.16152	-4.23916
Н	2.06611	-1.05229	-4.73824
Н	0.63532	-1.95248	-4.10460
С	0.44463	-2.91774	1.43579
Н	-3.43806	-1.38382	3.61547
Н	-2.94967	0.05432	2.63766
Н	-4.07264	-1.15511	1.94097
С	-0.12829	-4.34086	1.52802
Н	1.47996	-2.98075	1.06660
Н	0.48113	-2.46909	2.44072
Н	0.48473	-4.95875	2.20161
Н	-1.15821	-4.32576	1.91338
Н	-0.13639	-4.82285	0.53818
С	6.60621	0.46387	0.98987
Н	7.03092	0.51086	-0.01809
Н	6.73251	1.43430	1.49109
Н	7.10559	-0.33111	1.56386

1MA-1,2-inser

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Pd	-1.210578	0.762863	0.160249
P	0.899332	1.486117	-0.549308
С	2.246775	0.343145	0.001267
С	2.040641	-1.050085	0.044106
S	0.474919	-1.820500	-0.474649
0	-0.515601	-1.267154	0.551746
0	0.217914	-1.295181	-1.822271
0	0.657268	-3.259633	-0.304056
С	3.050568	-1.891648	0.512866
С	4.259828	-1.363064	0.955512
С	4.463581	0.014851	0.935202
С	3.463598	0.859268	0.461743
н	2.857762	-2.964402	0.526352
н	5.041046	-2.030127	1.322297
н	5.404854	0.437493	1.288122
н	3.630868	1.935943	0.445374
С	1.033053	1.721831	-2.365780
С	1.378810	3.115152	0.163702
н	-3.621107	2.164676	1.339777
С	-3.226112	1.950649	0.344224
С	-2.041467	2.661487	-0.044415
С	-4.344001	1.760981	-0.638150
Н	-1.581532	3.325105	0.695206
С	-3.175054	-0.115859	0.915760
Н	-2.646896	-0.360538	1.846949
Н	-1.964862	3.003852	-1.079859
Н	-4.228987	0.092192	1.108850
Н	-3.062189	-0.885315	0.141166
С	2.251751	1.978473	-3.032842
С	2.273057	2.107386	-4.424756
С	1.092541	1.986557	-5.154889
С	-0.114238	1.728673	-4.512380
С	-0.132186	1.586587	-3.127060
0	3.353744	2.115365	-2.249362
Н	3.207956	2.300237	-4.945966
Н	1.127459	2.088199	-6.240096
Н	-1.032900	1.613266	-5.085660
Н	-1.061962	1.335992	-2.618065
С	1.645617	4.240279	-0.618307
С	1.915608	5.479450	-0.034036
С	1.917310	5.598783	1.352038
С	1.637654	4.495008	2.159152
С	1.358703	3.259577	1.571341
Н	1.641821	4.149867	-1.704303

Η	2.125954	6.342917	-0.664521
Η	2.133096	6.559138	1.821815
Η	1.634061	4.606615	3.241574
0	1.049030	2.133324	2.266963
С	1.089234	2.183838	3.697634
Η	0.849142	1.168855	4.028457
Η	2.094027	2.461156	4.048943
Η	0.342513	2.892032	4.086332
С	4.632313	2.161787	-2.894277
Η	5.368548	2.151974	-2.083955
Η	4.774336	1.278867	-3.533846
Η	4.743929	3.084137	-3.483743
0	-3.908811	1.716967	-1.920122
0	-5.513496	1.664479	-0.308498
С	-4.963504	1.539631	-2.899504
Η	-4.454628	1.512567	-3.866708
Η	-5.496951	0.600622	-2.710310
Η	-5.668246	2.378219	-2.848067

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Pd	-0.915735	0.544577	0.551349
P	0.947839	1.475747	-0.376709
С	2.437599	0.439394	-0.027240
С	2.356599	-0.967321	0.036729
S	0.801616	-1.893663	-0.215389
0	-0.058598	-1.409013	0.950434
0	0.284041	-1.427053	-1.510736
0	1.151239	-3.304348	-0.070786
С	3.499958	-1.713132	0.327924
С	4.715353	-1.080578	0.573971
С	4.794713	0.309218	0.539015
С	3.662746	1.061587	0.241276
Н	3.405700	-2.798435	0.364295
Н	5.599531	-1.677140	0.802398
Н	5.739881	0.812950	0.743878
Н	3.728979	2.148619	0.214803
С	0.835591	1.723424	-2.192001
С	1.335150	3.127401	0.322965
Н	-2.916447	0.347118	1.146065
С	-3.297761	1.350385	0.748177
С	-2.100129	2.211112	0.394721
С	-4.222856	1.061494	-0.432942
Н	-1.786347	2.936090	1.159185
С	-4.072623	1.927366	1.932202
Н	-2.159574	2.672093	-0.600460
С	1.918617	2.173692	-2.981021
С	1.770571	2.294176	-4.365836
С	0.559282	1.964887	-4.970510
С	-0.512238	1.509974	-4.208554
С	-0.360824	1.383795	-2.830521
0	3.060514	2.492931	-2.317376
Н	2.598866	2.639793	-4.979937
Н	0.463743	2.060285	-6.052629
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H	-1.170192	0.980920	-2.224749
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C	1.706797	5.631756	1.525913
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Н	1.491229	6.445983	-0.462923
Н	1.859901	6.602073	1.999668
Н	1.856621	4.566470	3.391159
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С	4.228957	2.781853	-3.094806
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Н	-4.162173	3.430625	4.123287
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0	3.490584	1.102200	0.544348
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Н	-3.672971	2.875524	2.895082
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2a

2aMA-cis

2aMA-isom

2aMA-trans







2aMA-2,1-inser

2a1MA-2,1-KP

2a1MA-2,1-TP







2aMA-1,2-inser 2aMA-1,2-KP

2aMA-1,2-TP

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C	1.51//66	-1./91165	2.82885/
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н	2.546439	-1.406605	2.831511
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11	4.764501	0.429034	-2.211137
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н	0.709504	5.658915	-5.056854
H	0.823632	6.488027	-3.488492
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11 TT	1 1 1 5 5 6 5	4.000407	2.407412
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Н	1.693515	-1.890890	4.971241
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Н	-4.443873	-1.168868	2.111580
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E= -17.41271952 a.u.

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$\begin{array}{cccccc} 1.410734 & -2.763018 & 1.921019 \\ c & 1.077267 & -4.081201 & 2.240925 \\ c & 0.859940 & -5.056549 & 1.266206 \\ c & 0.892822 & -4.652534 & -0.065773 \\ N & 2.058062 & -1.117775 & 0.196970 \\ c & 3.229367 & -1.063118 & -0.703492 \\ c & 2.912473 & -0.059099 & -1.809395 \\ N & 1.914893 & 0.875703 & -1.262453 \\ c & 1.582812 & 1.986403 & -2.114087 \\ c & 2.329416 & 3.173146 & -2.048405 \\ c & 2.065608 & 4.180729 & -2.979928 \\ c & 1.089845 & 4.045234 & -3.965355 \\ c & 0.364398 & 2.853957 & -4.009416 \\ c & 0.599630 & 1.818382 & -3.105523 \\ c & 3.381327 & 3.405349 & -1.005138 \\ c & 0.806395 & 5.157221 & -4.935875 \\ c & 0.806395 & 5.157221 & -4.935875 \\ c & 0.67600 & -6.495684 & 1.638013 \\ c & 1.145903 & -2.985390 & -1.897948 \\ P & 1.259319 & 0.399995 & 0.251226 \\ c & 2.208446 & 1.373600 & 1.540110 \\ c & 3.447913 & 0.900090 & 1.989581 \\ c & 4.198603 & 1.606828 & 2.925341 \\ c & 3.711520 & 2.801558 & 3.447342 \\ c & 2.468782 & 3.276169 & 3.037363 \\ c & 1.719209 & 2.573168 & 2.094116 \\ S & 0.113699 & 3.270291 & 1.606536 \\ 0 & -0.867031 & 2.213235 & 2.100674 \\ Pd & -0.954809 & 0.500675 & 0.767967 \\ c & -1.766279 & -1.131481 & -0.200466 \\ 0 & -0.179942 & 3.308622 & 0.135075 \\ H & 3.830657 & -0.047550 & 1.620310 \\ H & 5.162872 & 1.214384 & 3.250353 \\ H & 4.292689 & 3.360117 & 4.181938 \\ H & 2.051719 & 4.201288 & 3.436572 \\ H & 2.645026 & 5.105333 & -2.926047 \\ H & -0.403388 & 2.720963 & -4.775106 \\ H & 1.018997 & -4.359486 & 3.295698 \\ H & 0.673328 & -5.379037 & -0.851996 \\ \end{array}$	С	1.524901	-2.400204	0.562611
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	С	1.410734	-2.763018	1.921019
C 0.859940 -5.056549 1.266206 C 0.892822 -4.652534 -0.065773 N 2.058062 -1.117775 0.196970 C 3.229367 -1.063118 -0.703492 C 2.912473 -0.059099 -1.809395 N 1.914893 0.875703 -1.262453 C 2.329416 3.173146 -2.048405 C 2.329416 3.173146 -2.048405 C 2.065608 4.180729 -2.979928 C 1.089845 4.045234 -3.965355 C 0.364398 2.853957 -4.009416 C 0.599630 1.818382 -3.105523 C 0.806395 5.157221 -4.935875 C -0.177940 0.539740 -3.226351 C 1.659558 -1.785174 3.032200 C 0.637060 -6.495684 1.638013 C 1.208446 1.373600 1.540110 C 3.711520 2.801558 3.447342 C 2.468782 3.276169 </td <td>С</td> <td>1.077267</td> <td>-4.081201</td> <td>2.240925</td>	С	1.077267	-4.081201	2.240925
C 0.892822 -4.652534 -0.065773 N 2.058062 -1.117775 0.196970 C 3.229367 -1.063118 -0.703492 C 2.912473 -0.059099 -1.809395 N 1.914893 0.875703 -1.262453 C 2.329416 3.173146 -2.048405 C 2.329416 3.173146 -2.048405 C 2.065608 4.180729 -2.979928 C 1.089845 4.045234 -3.965355 C 0.364398 2.853957 -4.009416 C 0.599630 1.818382 -3.105523 C 0.806395 5.157221 -4.935875 C -0.177940 0.539740 -3.226351 C 1.659558 -1.785174 3.032200 C 0.637060 -6.495684 1.638013 C 1.145903 -2.985390 -1.897948 P 1.259319 0.900901 1.989581 C 3.711520 2.801558 3.447342 C 2.468782 3.276169<	С	0.859940	-5.056549	1.266206
N 2.058062 -1.117775 0.196970 C 3.229367 -1.063118 -0.703492 C 2.912473 -0.059099 -1.809395 N 1.914893 0.875703 -1.262453 C 1.582812 1.986403 -2.114087 C 2.329416 3.173146 -2.048405 C 2.065608 4.180729 -2.979928 C 1.089845 4.045234 -3.965355 C 0.364398 2.853957 -4.009416 C 0.599630 1.818382 -3.105523 C 3.381327 3.405349 -1.005138 C 0.806395 5.157221 -4.935875 C -0.177940 0.539740 -3.226351 C 1.659558 -1.785174 3.032200 C 0.637060 -6.495684 1.638013 C 1.29319 0.399995 0.251226 C 2.208446 1.373600 1.540110 C 3.447913	С	0.892822	-4.652534	-0.065773
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	N	2.058062	-1.117775	0.196970
$\begin{array}{cccccc} 2.912473 & -0.059099 & -1.809395 \\ \text{N} & 1.914893 & 0.875703 & -1.262453 \\ \text{C} & 1.582812 & 1.986403 & -2.114087 \\ \text{C} & 2.329416 & 3.173146 & -2.048405 \\ \text{C} & 2.065608 & 4.180729 & -2.979928 \\ \text{C} & 1.089845 & 4.045234 & -3.965355 \\ \text{C} & 0.364398 & 2.853957 & -4.009416 \\ \text{C} & 0.599630 & 1.818382 & -3.105523 \\ \text{C} & 3.381327 & 3.405349 & -1.005138 \\ \text{C} & 0.806395 & 5.157221 & -4.935875 \\ \text{C} & -0.177940 & 0.539740 & -3.226351 \\ \text{C} & 1.659558 & -1.785174 & 3.032200 \\ \text{C} & 0.637060 & -6.495684 & 1.638013 \\ \text{C} & 1.145903 & -2.985390 & -1.897948 \\ \text{P} & 1.259319 & 0.399995 & 0.251226 \\ \text{C} & 2.208446 & 1.373600 & 1.540110 \\ \text{C} & 3.447913 & 0.900090 & 1.989581 \\ \text{C} & 4.198603 & 1.606828 & 2.925341 \\ \text{C} & 3.711520 & 2.801558 & 3.447342 \\ \text{C} & 2.468782 & 3.276169 & 3.037363 \\ \text{C} & 1.719209 & 2.573168 & 2.094116 \\ \text{S} & 0.113699 & 3.270291 & 1.606536 \\ \text{O} & -0.867031 & 2.213235 & 2.100674 \\ \text{Pd} & -0.954809 & 0.500675 & 0.767967 \\ \text{C} & -1.766279 & -1.131481 & -0.200466 \\ \text{O} & -0.029121 & 4.528533 & 2.333092 \\ \text{O} & 0.179932 & 3.308622 & 0.135075 \\ \text{H} & 3.830657 & -0.047550 & 1.620310 \\ \text{H} & 5.162872 & 1.214384 & 3.250353 \\ \text{H} & 4.292689 & 3.360117 & 4.181938 \\ \text{H} & 2.051719 & 4.201288 & 3.436572 \\ \text{H} & -0.403388 & 2.720963 & -4.775106 \\ \text{H} & 1.018997 & -4.359486 & 3.295698 \\ \text{H} & 0.673328 & -5.379037 & -0.851996 \\ \end{array}$	С	3.229367	-1.063118	-0.703492
N 1.914893 0.875703 -1.262453 C 1.582812 1.986403 -2.114087 C 2.329416 3.173146 -2.048405 C 2.065608 4.180729 -2.979928 C 1.089845 4.045234 -3.965355 C 0.364398 2.853957 -4.009416 C 0.599630 1.818382 -3.105523 C 3.381327 3.405349 -1.005138 C 0.806395 5.157221 -4.935875 C -0.177940 0.539740 -3.226351 C 1.659558 -1.785174 3.032200 C 0.637060 -6.495684 1.638013 C 1.145903 -2.985390 -1.897948 P 1.259319 0.399995 0.251226 C 2.208446 1.373600 1.540110 C 3.447913 0.90090 1.989581 C 4.198603 1.606828 2.925341 C 3.711520 2.801558 3.447342 C 2.468782 3.270291	С	2.912473	-0.059099	-1.809395
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	N	1.914893	0.875703	-1.262453
C2.3294163.173146-2.048405C2.0656084.180729-2.979928C1.0898454.045234-3.965355C0.3643982.853957-4.009416C0.5996301.818382-3.105523C3.3813273.405349-1.005138C0.8063955.157221-4.935875C-0.1779400.539740-3.226351C1.659558-1.7851743.032200C0.637060-6.4956841.638013C1.145903-2.985390-1.897948P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C1.766279-1.131481-0.200466O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.77	С	1.582812	1.986403	-2.114087
C2.0656084.180729-2.979928C1.0898454.045234-3.965355C0.3643982.853957-4.009416C0.5996301.818382-3.105523C3.3813273.405349-1.005138C0.8063955.157221-4.935875C-0.1779400.539740-3.226351C1.659558-1.7851743.032200C0.637060-6.4956841.638013C1.145903-2.985390-1.897948P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.77	С	2.329416	3.173146	-2.048405
C1.0898454.045234-3.965355C0.3643982.853957-4.009416C0.5996301.818382-3.105523C3.3813273.405349-1.005138C0.8063955.157221-4.935875C-0.1779400.539740-3.226351C1.659558-1.7851743.032200C0.637060-6.4956841.638013C1.145903-2.985390-1.897948P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C-1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.775106H1.018997-4.3594863.2	С	2.065608	4.180729	-2.979928
C0.3643982.853957-4.009416C0.5996301.818382-3.105523C3.3813273.405349-1.005138C0.8063955.157221-4.935875C-0.1779400.539740-3.226351C1.659558-1.7851743.032200C0.637060-6.4956841.638013C1.145903-2.985390-1.897948P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C-1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.775106H1.018997-4.3594863.295698H0.673328-5.379037-0.	С	1.089845	4.045234	-3.965355
C0.5996301.818382-3.105523C3.3813273.405349-1.005138C0.8063955.157221-4.935875C-0.1779400.539740-3.226351C1.659558-1.7851743.032200C0.637060-6.4956841.638013C1.145903-2.985390-1.897948P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C-1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.775106H1.018997-4.3594863.295698H0.673328-5.379037-0.851996	С	0.364398	2.853957	-4.009416
C3.3813273.405349-1.005138C0.8063955.157221-4.935875C-0.1779400.539740-3.226351C1.659558-1.7851743.032200C0.637060-6.4956841.638013C1.145903-2.985390-1.897948P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C-1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.775106H1.018997-4.3594863.295698H0.673328-5.379037-0.851996	С	0.599630	1.818382	-3.105523
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	С	3.381327	3.405349	-1.005138
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	С	0.806395	5.157221	-4.935875
C1.659558-1.7851743.032200C0.637060-6.4956841.638013C1.145903-2.985390-1.897948P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C-1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.775106H1.018997-4.3594863.295698H0.673328-5.379037-0.851996	С	-0.177940	0.539740	-3.226351
C0.637060-6.4956841.638013C1.145903-2.985390-1.897948P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C-1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.775106H1.018997-4.3594863.295698H0.673328-5.379037-0.851996	С	1.659558	-1.785174	3.032200
C1.145903-2.985390-1.897948P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C-1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.775106H1.018997-4.3594863.295698H0.673328-5.379037-0.851996	С	0.637060	-6.495684	1.638013
P1.2593190.3999950.251226C2.2084461.3736001.540110C3.4479130.9000901.989581C4.1986031.6068282.925341C3.7115202.8015583.447342C2.4687823.2761693.037363C1.7192092.5731682.094116S0.1136993.2702911.606536O-0.8670312.2132352.100674Pd-0.9548090.5006750.767967C-1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.775106H1.018997-4.3594863.295698H0.673328-5.379037-0.851996	С	1.145903	-2.985390	-1.897948
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	P	1.259319	0.399995	0.251226
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	С	2.208446	1.373600	1.540110
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	С	3.447913	0.900090	1.989581
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C	4.198603	1.606828	2.925341
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C	3.711520	2.801558	3.447342
$\begin{array}{cccccccc} C & 1.719209 & 2.573168 & 2.094116 \\ S & 0.113699 & 3.270291 & 1.606536 \\ O & -0.867031 & 2.213235 & 2.100674 \\ Pd & -0.954809 & 0.500675 & 0.767967 \\ C & -1.766279 & -1.131481 & -0.200466 \\ O & -0.029121 & 4.528533 & 2.333092 \\ O & 0.179932 & 3.308622 & 0.135075 \\ H & 3.830657 & -0.047550 & 1.620310 \\ H & 5.162872 & 1.214384 & 3.250353 \\ H & 4.292689 & 3.360117 & 4.181938 \\ H & 2.051719 & 4.201288 & 3.436572 \\ H & 2.645026 & 5.105333 & -2.926047 \\ H & -0.403388 & 2.720963 & -4.775106 \\ H & 1.018997 & -4.359486 & 3.295698 \\ H & 0.673328 & -5.379037 & -0.851996 \\ \end{array}$	С	2.468782	3.276169	3.037363
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	С	1.719209	2.573168	2.094116
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	S	0.113699	3.270291	1.606536
Pd-0.9548090.5006750.767967C-1.766279-1.131481-0.200466O-0.0291214.5285332.333092O0.1799323.3086220.135075H3.830657-0.0475501.620310H5.1628721.2143843.250353H4.2926893.3601174.181938H2.0517194.2012883.436572H2.6450265.105333-2.926047H-0.4033882.720963-4.775106H1.018997-4.3594863.295698H0.673328-5.379037-0.851996	0	-0.867031	2.213235	2.100674
$\begin{array}{cccccc} C & -1.766279 & -1.131481 & -0.200466 \\ O & -0.029121 & 4.528533 & 2.333092 \\ O & 0.179932 & 3.308622 & 0.135075 \\ H & 3.830657 & -0.047550 & 1.620310 \\ H & 5.162872 & 1.214384 & 3.250353 \\ H & 4.292689 & 3.360117 & 4.181938 \\ H & 2.051719 & 4.201288 & 3.436572 \\ H & 2.645026 & 5.105333 & -2.926047 \\ H & -0.403388 & 2.720963 & -4.775106 \\ H & 1.018997 & -4.359486 & 3.295698 \\ H & 0.673328 & -5.379037 & -0.851996 \\ \end{array}$	Pd	-0.954809	0.500675	0.767967
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2b

2bMA-cis

2bMA-isom

2bMA-trans







2bMA-2,1-inser

2bMA-2,1-KP

2bMA-2,1-TP







2bMA-1,2-inser

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2bMA-cis

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Н	0.031828	0.302041	-5.791026

MA E= -2,563445 a.u.

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