

Supporting Information

Thiazolylpyrazoles as Δ F508-CFTR Correctors with Improved Hydrophilicity Compared to Bithiazoles

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Formula: C₁₈ H₂₄ N₄ O₃ S₁

Unit cell parameters: a 10.2685(15) b 11.2542(17) c 17.123(3)

alpha 72.768(2) beta 88.794(2) gamma 88.903(2)

space group P-1

9f: summary of data CCDC 767035

Formula: C₂₁ H₂₃ Br₁ N₄ O₃ S₁

Unit cell parameters: a 10.0372(4) b 20.6057(8) c 11.0875(4) beta 107.7350(10)

space group P2₁/c

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Table 1. Calculated and extrapolated logP values of active thiazolopyrazoles **11d/14a/14b/14e/14g/14h/14j** and the bithiazole **1**.

compound	logP	clogP ^{*a}	relative difference (%) ^{*b}
11d	5.52	5.61	1.63593
14a	4.75	4.17	12.28250
14b	4.53	3.82	15.58325
14e	5.1	4.77	6.32385
14g	5.70	5.78	1.42973
14h	4.26	4.12	3.19175
14j	3.46	3.40	1.61200
1	5.71	5.6	1.91409

^{*a} calculated with ChemDraw ultra 11(Cambridge Software)

^{*b} Relative difference=[Absolute value of (logP-clogP)]/logP

Table 2. Logk and logP values of reference compounds (used to establish the logP vs. logk trendline) and thiazolyipyrazole correctors.

Compound	T1 ^{*a}	(T1-T0) ^{*b}	logk ^{*c}	logP
4-chloro-Phenol	20.10	17.08	0.752	2.4 ^{*d}
2,4-dichloro-Phenol	22.16	19.14	0.802	3.21 ^{*d}
3,4,5-trichloro-Phenol	23.61	20.59	0.834	4.36 ^{*d}
pentachlorophenol	25.83	22.81	0.878	5.9 ^{*d}
p, p'-DDT	28.80	25.78	0.931	6.19 ^{*d}
11d	25.48	22.46	0.871	5.52 ^{*e}
14a	24.12	21.1	0.844	4.75 ^{*e}
14b	23.73	20.71	0.836	4.53 ^{*e}
14e	24.71	21.69	0.856	5.09 ^{*e}
14g	25.81	22.79	0.878	5.70 ^{*e}
14h	23.28	20.26	0.827	4.26 ^{*e}
17	22.00	18.98	0.798	3.46 ^{*e}
4a	25.83	22.81	0.878	5.71 ^{*e}
KBr	3.02	0.00		

*^a Retention time on RP-HPLC.

*^b The difference between the retention time T₁ and the void time T₀, which is the retention time of KBr

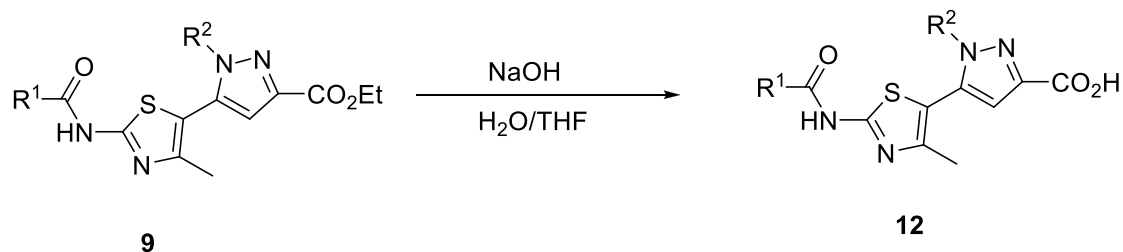
*^c logk is the capacity factor, which equals to $\log [(T_1 - T_0) / T_0]$.

*^d The logP value is found in reference 17.

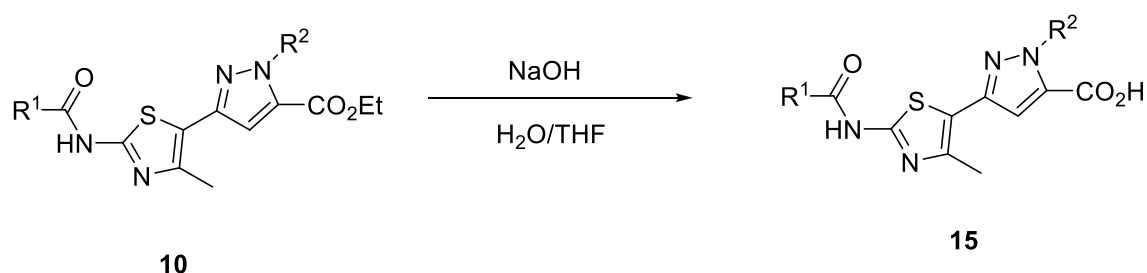
*^e The logP value is extrapolated from the trendline that was established based on the values of the reference compounds, the substituted phenols and p,p'-DDT. The equation of the trendline is $y = 23.304x - 15.151$ and the linear regression square $R^2 = 0.96$.

Preparation of thiazolypyrazole carboxylic acids. None of the thiazolypyrazole carboxylic acids are active in [I] influx cell assay. General procedure for the preparation of acids **12** and **15** as well as representative structural data.

Scheme 1:



Scheme 2:



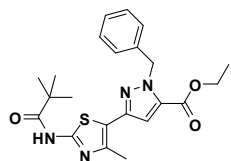
General procedure of saponification of thiazolypyrazole esters: To a 0.2 M solution of the ester (**9** or **10**; 1 equivalent) in 1:1 H₂O/THF was added NaOH (4 equiv). The reaction mixture was stirred at room temperature for 24 h. When the reaction was complete, water, in the same volume as the reaction mixture, was added followed by concentrated HCl (dropwise) to adjust the pH to 3. The product, which precipitated, was collected by filtration and dried under vacuum. The ¹H NMR, ¹³C NMR, and low resolution mass spectrometry data of representative acids **12** and **15** are listed below.

1-(4-Bromophenyl)-5-(4-methyl-2-pivalamidothiazol-5-yl)-1H-pyrazole-3-carboxylic acid (12b). Ester **9a** (R = H and R² = *p*-Br-Ph; 100 mg, 0.2 mmol) was reacted with NaOH (33 mg, 0.8 mmol) by the above general procedure and delivered **12b** as an off-white solid (76 mg, 81%). ¹H NMR (600 MHz, DMSO) δ 13.12 (s, 1H), 12.00 (s, 1H), 7.69 (d, *J* = 8.8, 2H), 7.36 (d, *J* = 8.8, 2H), 7.05 (s, 1H), 2.02 (s, 3H), 1.21 (s, 9H); ¹³C NMR (151 MHz, DMSO) δ 177.58, 163.39, 159.33, 147.57, 145.49, 138.83, 136.07, 132.99, 127.60, 122.27, 112.92, 111.40, 104.99, 39.43, 27.15, 16.22; LC/MS(ESI): cal. [M+H⁺] = 463.04 and [M+2+H⁺] = 465.04, found 462.93 and 464.98.

1-Allyl-3-(4-methyl-2-pivalamidothiazol-5-yl)-1H-pyrazole-5-carboxylic acid (15b). Ester **10a** (R = H and R² = allyl; 95 mg, 0.25 mmol) was reacted with NaOH (40 mg, 1.0 mmol) by the above general procedure and delivered **15b** as an off-white solid (79 mg, 90%). ¹H NMR (600 MHz, DMSO) δ 13.60 (s, 1H), 11.78 (s, 1H), 6.97 (s, 1H), 6.00 (ddd, *J* = 5.5, 10.6, 22.5, 1H), 5.13 (dd, *J* = 3.6, 15.5, 3H), 4.98 (dd, *J* = 1.5, 17.1, 1H), 2.41 (s, 3H), 1.21 (s, 9H); ¹³C NMR (151 MHz, DMSO) δ 177.28, 160.87, 156.73, 143.54, 134.86, 134.70, 117.67, 117.00, 109.40, 53.96, 39.43, 27.26, 17.00; LC/MS(ESI): cal. [M+H⁺] = 349.13, found 349.13

HPLC chromatograms and mass spectra for thiazolylpyrazole correctors

10a (R = H, R² = Bn)



Calculated (cal. for short) [M+H⁺] = 427.18

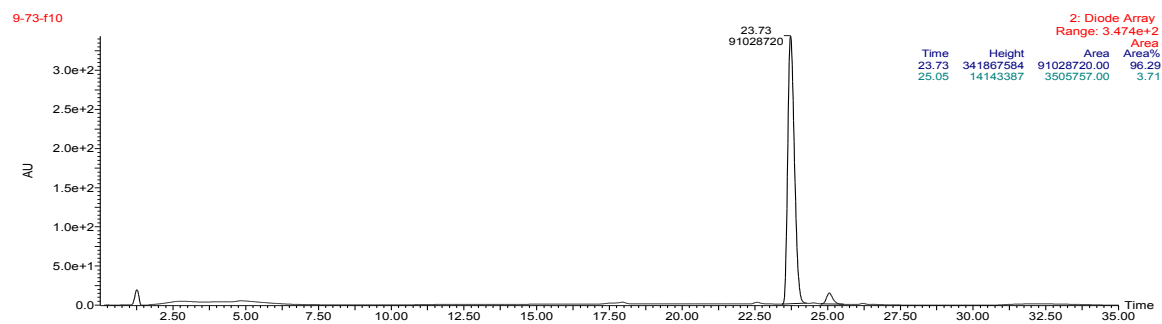


Figure SI-1. HPLC Chromatogram of compound 10a (R = H, R² = Bn)

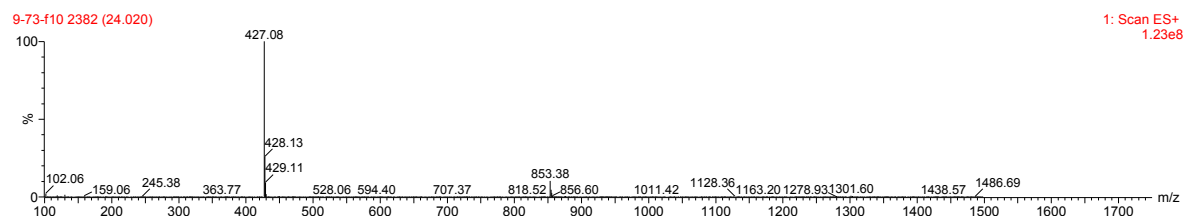
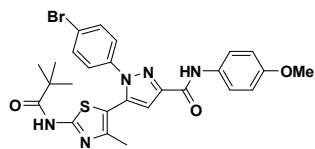


Figure SI-2. Mass spectrum of 10a (R = H, R² = Bn)

11d



cal. $[M+H]^+$ = 568.10

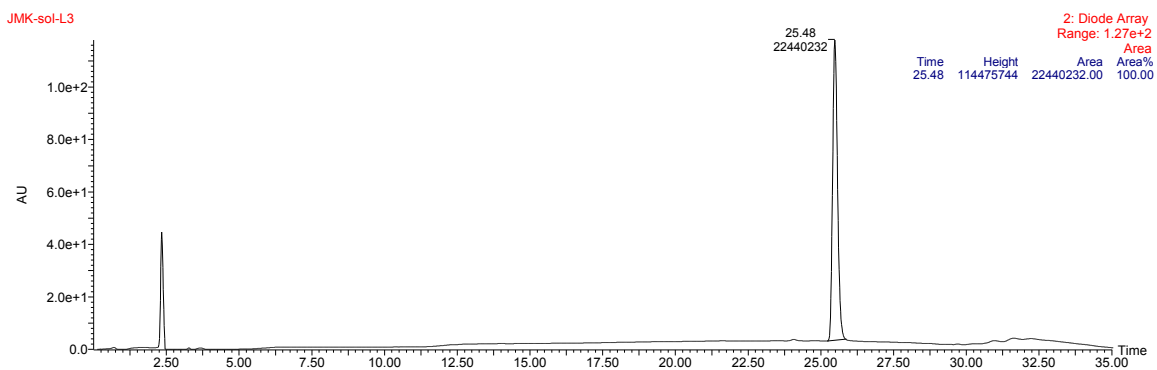


Figure SI-3. HPLC Chromatogram of compound 11d

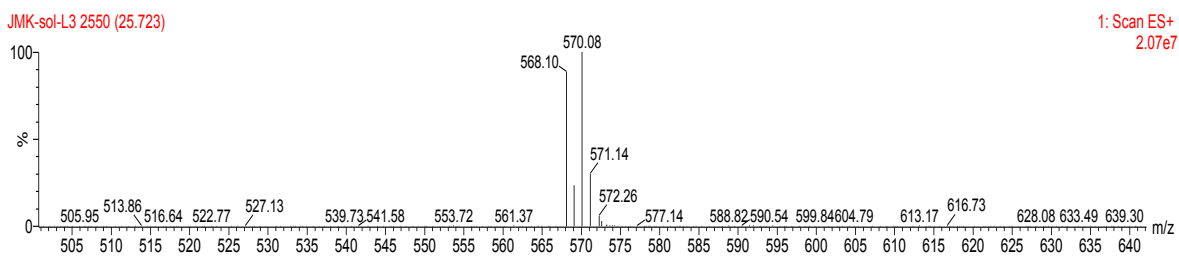
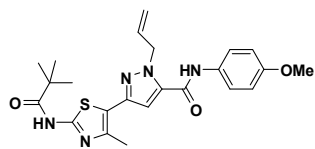


Figure SI-4. Mass spectrum of 11d

14a



cal. $[M+H^+]=454.19$

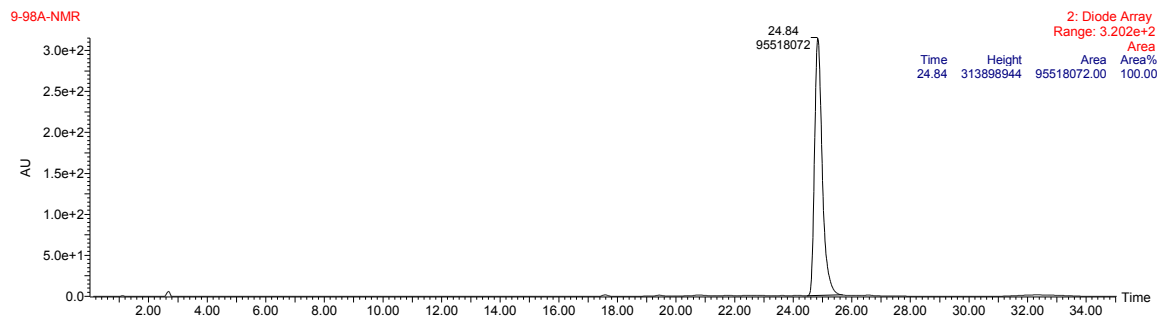


Figure SI-5. HPLC Chromatogram of compound 14a

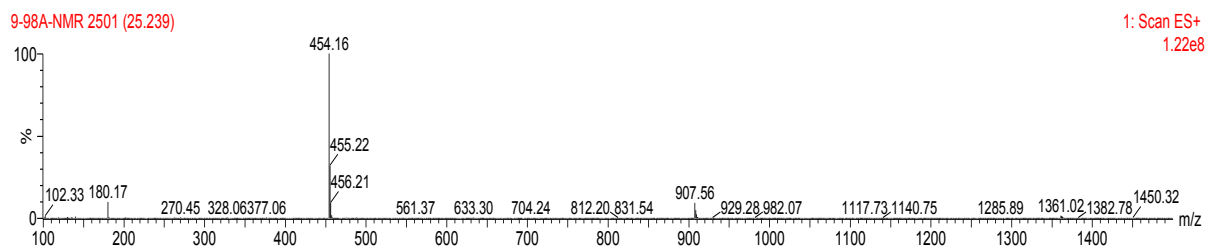
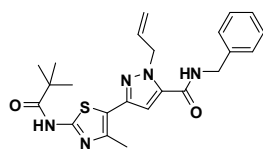


Figure SI-6. Mass spectrum of 14a

14b



cal. $[M+H]^+ = 438.20$

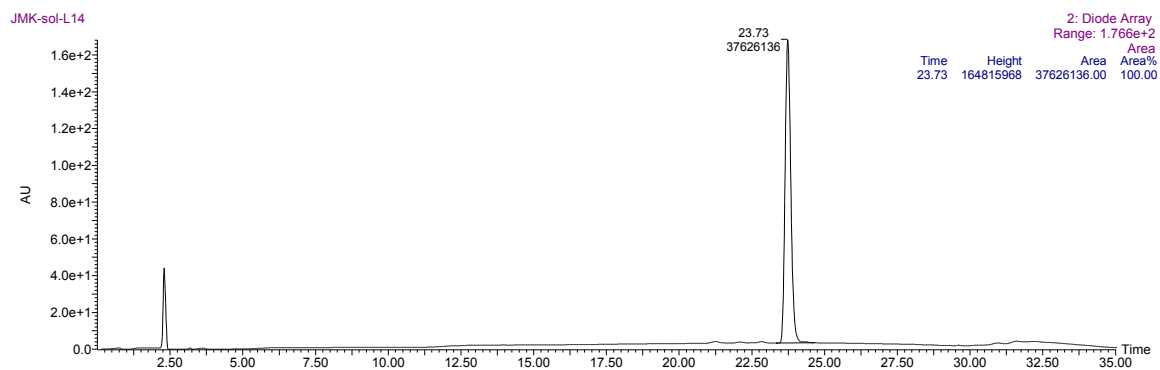


Figure SI-7. HPLC Chromatogram of compound 14b

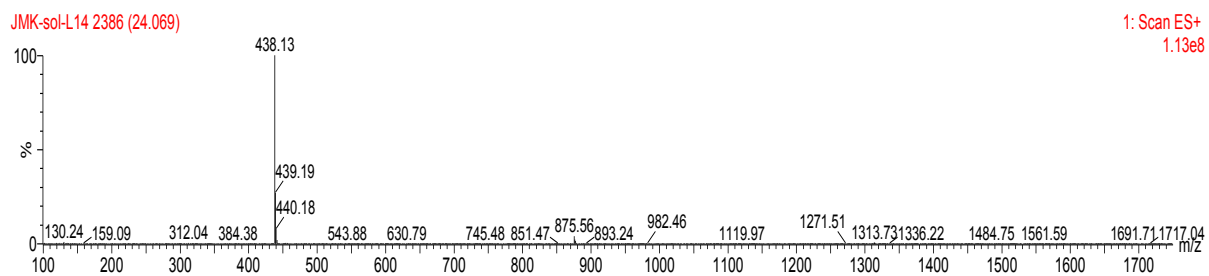
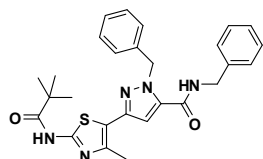


Figure SI-8. Mass spectrum of 14b

14e



cal. $[M+H]^+ = 488.21$

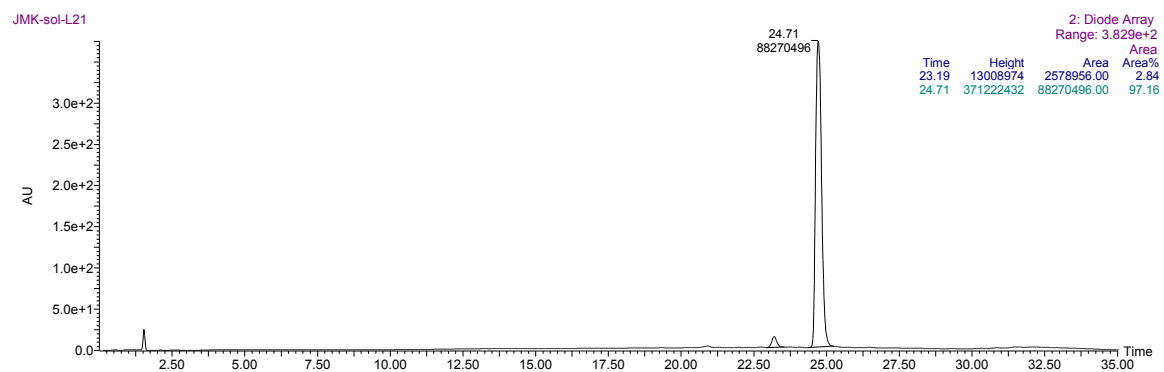


Figure SI-9. HPLC Chromatogram of compound 14e

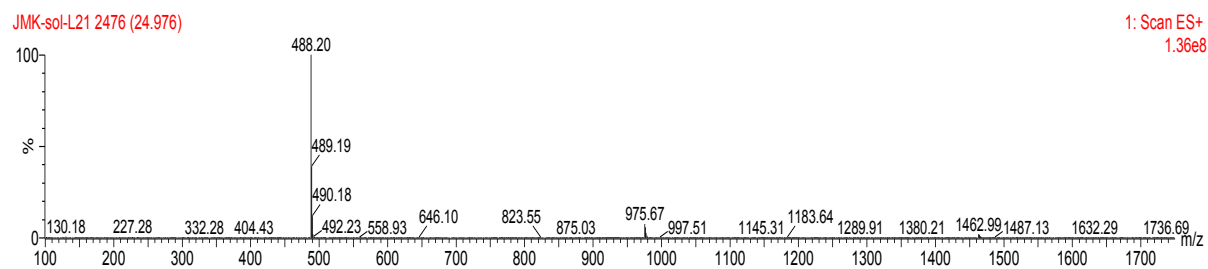
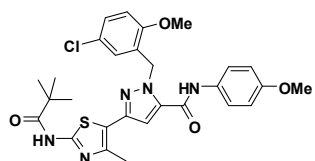


Figure SI-10. Mass spectrum of 14e

14g



cal. $[M+H^+]=568.18$

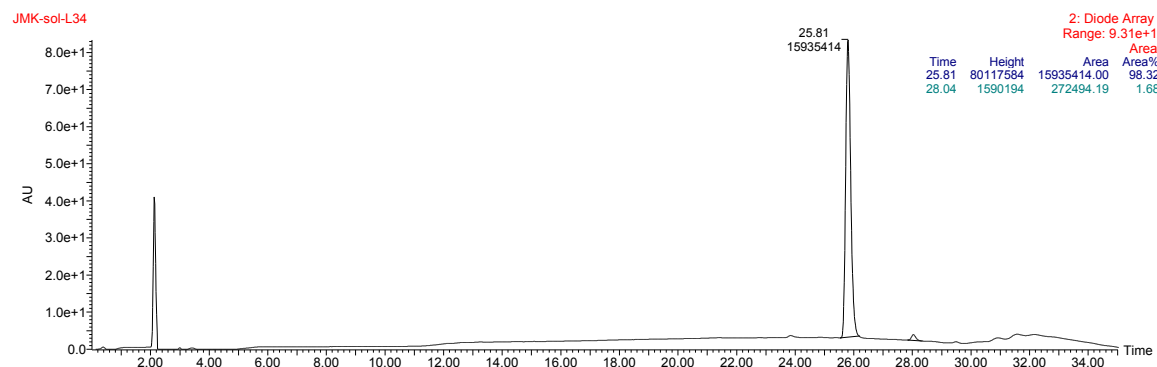


Figure SI-11. HPLC Chromatogram of compound 14g

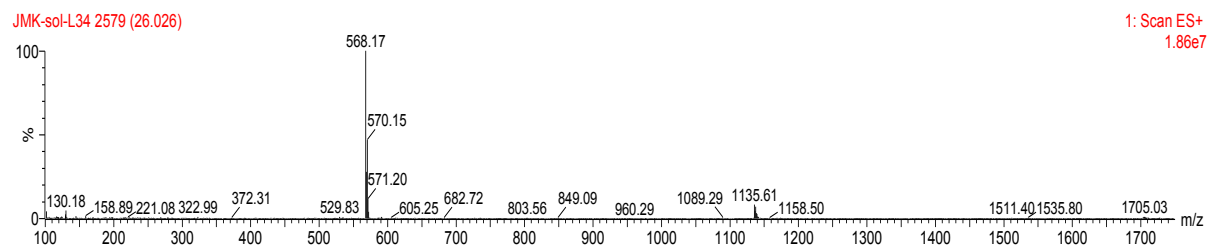
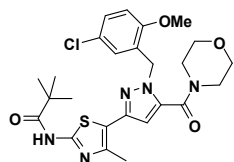


Figure SI-12. Mass spectrum of 14g

14h



cal. $[M+H^+]= 532.18$

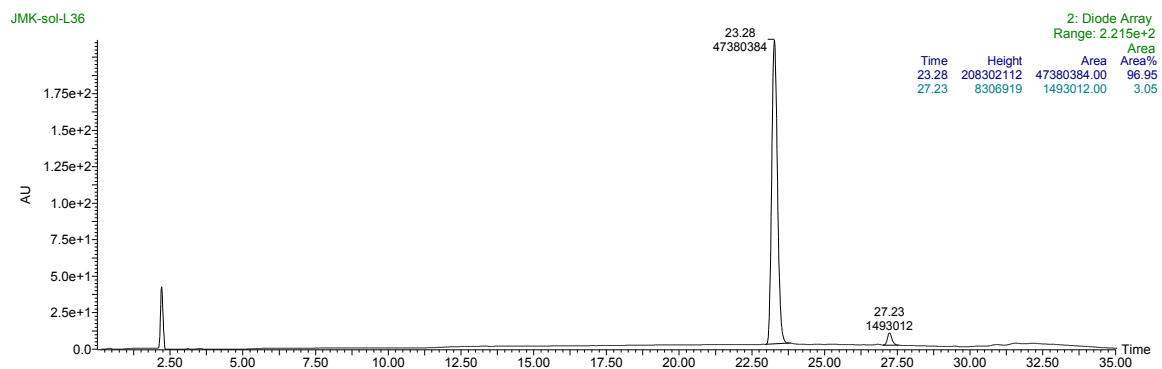


Figure SI-13. HPLC Chromatogram of compound 14h

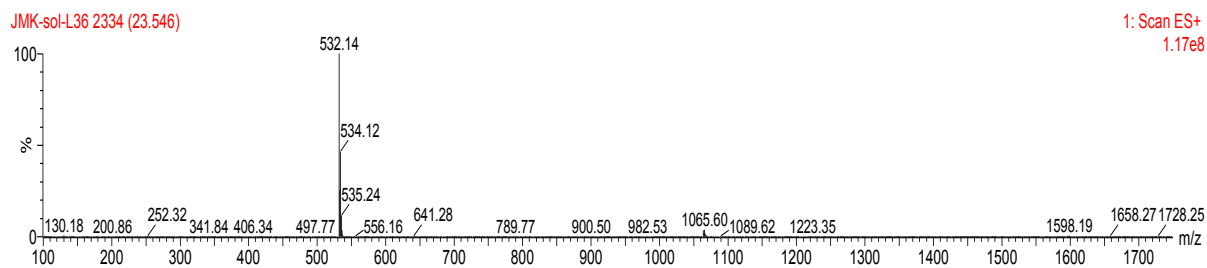


Figure SI-14. Mass spectrum of 14h

