

Copper isotope fractionation during adsorption and intracellular incorporation by bacteria: Isotopic data

Starting solution for all adsorption experiments $\delta^{65}\text{Cu} = 0.712 \pm 0.14$ unless otherwise noted

n = number of replicate analyses of the same sample, d = number of column replicates included

All adsorption experiments conducted using a 0.01M NaClO₄ electrolyte solution.

Control experiment with 10 mg/L Cu only, no bacteria

Sample ID	F precipitated	pH	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
21.2	0.40	6.44	1.077	2		0.105
21.3	0.01	3.30	0.722	1	2	0.193
21.5	0.05	6.12	0.748	1		
21.7	0.03	4.49	0.673	1		
21.9	0.58	6.67	1.130	2		0.112
21.10.	0.94	7.31	-1.308	2		0.049

***E. coli* adsorption vs. pH, 5g/L bacteria, 10 mg/L Cu**

Sample ID	F adsorbed	pH	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
Cu.8.1.U	0.21	2.79	0.719	1		
Cu.8.3.U	0.35	3.63	1.340	1		
Cu.8.5.U	0.39	3.91	1.307	1		
Cu.8.1.F	0.21	2.85	0.974	1		
Cu.8.3.F	0.24	2.91	1.097	2	2	0.152
Cu.8.6.F	0.84	5.61	1.580	1		
Cu.8.8.F	0.45	3.94	1.891	2		0.124
Cu.8.9.F	0.88	5.90	2.009	1		
Cu.8.10.F	0.43	3.22	1.524	1		

***E. coli* adsorption vs. pH, 15g/L bacteria, 2 mg/L Cu**

Sample ID	F adsorbed	pH	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
4B.1.Cu	0.95	6.45	0.943	1		
4B.2.Cu	0.64	4.22	1.276	1		
4B.3.Cu	0.90	5.30	1.333	1		
4B.4.Cu	0.93	5.03	0.770	1		
4B.5.Cu	0.45	2.49	1.253	1		
4B.6.Cu	0.72	3.76	0.873	1		
4B.7.Cu	0.80	3.96	1.565	1		

4B.8.Cu 0.67 3.14 1.703 1

***B. subtilis* adsorption vs. pH, 5g/L bacteria, 10 mg/L Cu**

Sample ID	F adsorbed	pH	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
20.1	0.43	4.30	0.603	1		
20.2	0.27	2.09	0.629	1		
20.3	0.27	3.15	0.932	3	2	0.184
20.4	0.65	5.30	0.966	3	2	0.164
20.5	0.64	6.57	1.037	2	2	0.085
20.6	0.43	4.44	1.171	2		0.145
20.7	0.50	4.03	0.232	1		
20.8	0.17	2.90	0.704	1		

***E. coli* metal loading experiment, 5g/L bacteria, variable Cu**

Sample ID	F adsorbed	pH	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
12.1.Cu.Load	0.43	4.00	1.046	2		0.070
12.2.Cu.Load	0.26	4.00	1.316	2		0.009
12.3.Cu.Load	0.13	4.00	1.281	2	2	0.289

***B. subtilis* metal loading experiment, 1g/L bacteria, variable Cu (δ^{65} of $\text{Cu}(\text{ClO}_4)_2$ starting soln = 0.88‰)**

Sample ID	F adsorbed	pH	mg/L Cu added	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
31.1	0.19	4.00	4.98	1.254	1		
31.2	0.14	4.00	10.09	1.462	1		
31.3	0.07	4.00	25.49	1.387	1		

Heat-killed *E. coli* adsorption vs. pH, 5g/L bacteria, 10 mg/L Cu(II)

Sample ID	F adsorbed	pH	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
40.2.Cu	0.27	2.80	0.417	2	2	0.108
40.3.Cu	0.38	3.40	0.750	1		
40.4.Cu	0.45	3.90	0.851	2		0.372
40.5.Cu	0.44	4.00	0.816	1		
40.6.Cu	0.45	4.10	0.399	2		0.260
40.8.Cu	0.75	5.11	0.341	2	2	0.014

48 hour kinetics experiment with *E. coli*, 5g/L bacteria, 10 mg/L Cu(II)

Sample ID	F adsorbed	pH	sample time (hours)	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
39.1.Cu	0.22	4.30	0.08	1.265	1		
39.2.Cu	0.23	4.30	0.08	1.361	1		
39.3.Cu	0.28	4.30	0.25	1.811	1		
39.4.Cu	0.32	4.30	0.50	1.019	1		
39.5.Cu	0.36	4.30	1.00	1.370	1		

39.6.Cu	0.37	4.30	1.33	1.229	1		
39.7.Cu	0.52	4.30	25.00	1.325	1		
39.8.Cu	0.55	4.30	48.00	1.618	2		0.131

***E. coli* kinetics and reversibility experiment**
5g/L bacteria, 10 mg/L Cu(II)

Sample ID	F adsorbed	pH	sample time (hours)	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
11.1.Cu.Kin	0.17	4.00	0.08	0.669	2	2	0.098
11.2.Cu.Kin	0.18	4.00	0.08	0.981	2	2	0.150
11.3.Cu.Kin	0.20	4.00	0.25	0.860	2		0.486
11.4.Cu.Kin	0.19	4.00	0.28	0.762	1		
11.6.Cu.Kin	0.18	4.00	0.50	0.511	2		0.072
11.7.Cu.Kin	0.21	4.00	1.00	1.478	3	2	0.063
11.8.Cu.Kin	0.22	4.00	2.00	0.525	2		0.088
11.9.Cu.Kin	0.20	4.00	3.00	0.948	1		
11.10.Cu.Kin	0.44	5.00	3.12	0.903	2		0.170
11.11.Cu.Kin	0.41	5.00	3.28	0.947	1		
11.12.Cu.Kin	0.44	5.00	3.53	0.955	1		
11.14.Cu.Kin	0.45	5.00	5.00	1.028	1		
11.15.Cu.Kin	0.45	5.00	5.00	1.130	1		
11.16.Cu.Kin	0.35	4.00	5.02	0.894	2		0.032
11.17.Cu.Kin	0.31	4.00	5.13	0.672	1		
11.18.Cu.Kin	0.31	4.00	5.45	1.203	1		
11.19.Cu.Kin	0.29	4.00	5.55	1.373	1		
11.20.Cu.Kin	0.28	4.00	5.55	0.862	1		
11.21.Cu.Kin	0.29	4.00	6.00	0.988	1		
11.22.Cu.Kin	0.29	4.00	6.50	0.652	1		
11.23.Cu.Kin	0.35	4.00	24.00	1.179	1		

***B. subtilis* kinetics and reversibility experiment**
bacteria, 10 mg/L Cu(II)

5g/L

Sample ID	F adsorbed	pH	sample time (hours)	$\delta^{65}\text{Cu}$ solution	n	d	2 sigma
30.1	0.42	3.93	0.08	1.034	1		
30.2	0.43	3.93	0.42	1.140	1		
30.3	0.43	3.90	0.87	1.263	2	2	0.047
30.4	0.43	3.89	1.00	1.065	1		
30.5	0.72	4.96	1.08	1.332	1		
30.6	0.72	4.80	1.18	1.452	1		
30.7	0.73	5.05	1.87	1.622	2	2	0.190

30.8	0.65	4.60	3.00	1.267	1	
30.9	0.41	3.90	3.03	1.236	1	
30.10.	0.45	3.90	3.13	1.351	2	0.127

Starting solution (Cu-citrate) for all intracellular incorporation experiments

$\delta^{65}\text{Cu} = -0.451 \pm 0.122$

<i>E. coli</i> intracellular Cu					
	F Cu SOLID	pH	$\delta^{65}\text{Cu}$ SOLID	n	
EC 1	0.01	7.00	-3.045	1	
EC 2	0.02	7.00	-2.493	1	
<i>B. subtilis</i> intracellular Cu					
	F Cu SOLID	pH	$\delta^{65}\text{Cu}$ SOLID	n	
BS 1	0.01	7.00	-1.680	1	
BS 2	0.02	7.00	-1.658	1	
Rio Grande Reservoir consortia intracellular Cu					
	F Cu SOLID	pH	$\delta^{65}\text{Cu}$ SOLID	n	
RGB 1	0.40	7.00	-3.085	1	
Cement Creek consortia intracellular Cu					
	F Cu SOLID	pH	$\delta^{65}\text{Cu}$ SOLID	n	
NB 9/24 2	0.13	7.00	-2.222	2	0.027
NB 9/26 1	0.27	7.00	-1.804	1	

Atomic adsorption standard					
	n	d	$\delta^{65}\text{Cu}$	2 sigma	
AA std	15	5	0.712	0.140	
Copper rod standard					
	n	d	$\delta^{65}\text{Cu}$	2 sigma	
Cu ROD	12	3	0.716	0.166	
Copper-citrate					
	n	d	$\delta^{65}\text{Cu}$	2 sigma	
Cu-citrate	4	2	-0.451	0.122	
Copper perchlorate					
	n	d	$\delta^{65}\text{Cu}$	2 sigma	
Cu(ClO ₄) ₂	3		0.883	0.139	

