SUPPORTING MATERIAL Steric and Electronic Influences on the Torsional Energy Landscape of Retinal

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Computational Methods

QM calculations were performed with Gaussian 03 (1) or Spartan'08 (2) on each model compound. Geometry optimizations used the 6-31G** basis set at the Hartree-Fock level; energies were computed for the optimized structures with second-order Møller-Plesset (MP2) perturbation theory to treat electron correlation. All optimizations were performed to default tolerances and all degrees of freedom were allowed to relax except for specific dihedrals as described below. To examine basis set effects, we computed single point energies for energy minima and maxima of the largest model compounds (MP2 level of theory with a cc-pVTZ basis set), and found only negligible changes in energy barriers of 3% or less. Molecular mechanics (MM) calculations were performed using the program CHARMM (3) without truncation of intermolecular forces. Torsional parameters for the methyl groups in retinal were fit to reproduce the QM energy surfaces described in the main text. Force field parameters for the remaining degrees of freedom were unchanged from those distributed with CHARMM.

Supporting Data

Table S1. Force field parameters, consistent with the CHARMM force field, for methyl rotation in neutral unsaturated alkenes based on the QM results in Figure 2. For comparison, previous MD simulations of retinal typically set the methyl torsion force constants to zero (4), i.e. they were not explicitly parameterized but rather relied solely on the non-bonded terms to determine the potential energy.

Atom types	K_{ϕ}	п	δ
СС2-СС1А-СТ3-НА	0.0201	3	180
CC1A-CC1A-CT3-HA ^a	0.0640	3	180
СС1А-СС1В-СТ3-НА ^b	0.0640	3	0

^{*a*} CC1B-CC1B-CT3-HA uses the same force field parameters as CC1A-CC1A-CT3-HA.

^b CC1B-CC1A-CT3-HA uses the same force field parameters as CC1A-CC1B-CT3-HA.



Figure S1. Protonation of the Schiff base decreases single bond length and increases double bond length. Data for bond lengths are taken from MP2 calculations performed on model compounds 8 (protonated) and 9 (deprotonated).



Figure S2. Comparison of QM (circles) and MM (diamonds) methyl torsion angle energies for C1-Me (equivalent to retinal C5-Me group) in compound **10**. Note that use of current CHARMM parameters for methyl dihedrals accurately reproduces the QM results.

Topologies of model compounds 1, 2, 5, 8, 9, and 10.

All topologies are in CHARMM format.

```
RESI 2MBD
              0.00 ! 2-methyl, 1,3 butadiene
!
GROUP
ATOM C1 CC2
              -0.42 !
ATOM H11 HE2
              0.21 !
ATOM H12 HE2
               0.21 !
ATOM C2 CC1A
               0.00 !
                          H51
                                Н52
                               /
GROUP
                             \setminus
                     1
ATOM C3 CC1B -0.15 ! H11
                             С5-Н5З
ATOM H31 HE1
              0.15 !
                         \backslash
                              /
             -0.42 !
                        C1=C2
ATOM C4 CC2
                                    H41
             0.21 !
ATOM H41 HE2
                         /

ATOM H42 HE2
              0.21 ! H12
                              C3=C4
                              /
GROUP
                    !
                                    /
              -0.27 !
ATOM C5 CT3
                            Н31
                                   H42
АТОМ Н51 НА
               0.09 !
               0.09 !
ATOM H52 HA
              0.09 !
АТОМ Н53 НА
RESI 3MHT
              0.00 ! 3-methyl 1,3,5 hexatriene
GROUP
ATOM C7
         CC2 -0.42 !
              0.21 !
ATOM H71 HE2
                         H192
                                       H101
                                               H121
ATOM H72 HE2
               0.21 !
                                H193
ATOM C8 CC1A -0.15 !
                         \
                                 /
                                        /
ATOM H81 HE1 0.15 !
                         H191-C19
                                        C11=C12
                                \setminus
GROUP
                     1
                                       /
ATOM C9 CC1B 0.00 !
                                 C9=C10
                         H71
                                                H122
ATOM C10 CC1B -0.15 !
                          \
                                 /
                                      C7=C8
ATOM H101 HE1
               0.15 !
                                      H101
                                 \setminus
                            /
GROUP
                     ATOM C11 CC1A -0.15 !
                         H72
                                 H81
ATOM H111 HE1
              0.15 !
             -0.42 !
ATOM C12 CC2
ATOM H121 HE2
               0.21 !
ATOM H122 HE2
               0.21 !
GROUP
ATOM C19 CT3 -0.27 !
ATOM H191 HA
              0.09 !
ATOM H192 HA
               0.09 !
ATOM H193 HA
               0.09 !
RESI 4MOT
               0.00 ! 4-methyloctatetraene-all trans
                     1
GROUP
ATOM C9
         CC2 -0.42 !
ATOM H91 HE2
              0.21 !
ATOM H92 HE2
               0.21 !
GROUP
ATOM C10 CC1B -0.15 !
ATOM H101 HE1 0.15 !
```


 ATOM C11
 CC1A
 -0.15
 !
 H202
 H203
 H151
 H161

 ATOM H111
 HE1
 0.15
 !
 \
 \
 /
 /

 ATOM C12
 CC1A
 -0.15
 !
 H201-C20
 C15=C16

 ATOM H121
 HE1
 0.15
 !
 H111
 C13=C14
 H162

 ATOM C13
 CC1B
 0.00
 !
 \
 /
 \

 ATOM C20
 CT3
 -0.27
 !
 H91
 C11=C12
 H141

 ATOM H201
 HA
 0.09
 !
 /
 \
 \

 ATOM H202
 HA
 0.09
 !
 C9=C10
 H121

 ATOM H203
 HA
 0.09
 !
 /
 \

 GROUP
 !
 H92
 H101
 H21

 ATOM C14 CC1B -0.15 ! ATOM H141 HE1 0.15 ! ATOM C15 CC1A -0.15 ! ATOM H151 HE1 0.15 ! ATOM C16 CC2 -0.42 ! ATOM H161 HE2 0.21 ! ATOM H162 HE2 0.21 ! RESI MAT 0.00 ! 13-methyl fragment of retinal, all-trans ! with net zero charge GROUP ATOM C7 CC2 -0.42 ! ATOM H71 HE2 0.21 ! H71 H81 ATOM H72 HE2 0.21 ! $\$ / ATOM C8 CC1A -0.15 ! C7=C8 H10 ATOM H81 HE1 0.15 ! / \ / H101
 GROUP
 !
 H72
 C9=C10
 H1

 ATOM C9
 CC1B
 0.00
 !
 /
 /
 H121

 ATOM C9
 CC1B
 0.00 !
 /
 /
 /

 ATOM C19
 CT3
 -0.27 !
 H191--C19
 C11=C12
 H141

 ATOM H191
 HA
 0.09 !
 /
 /
 /

 ATOM H192
 HA
 0.09 !
 H192 |
 H111
 C13=C14
 H161

 ATOM H193
 HA
 0.09 !
 H193
 /
 /
 /

 GROUP
 !
 H193
 /
 /
 /
 /

 ATOM C10
 CC1B
 -0.15 !
 H201--C20
 C15=1NZ-C16--H162

 ATOM H101
 HE1
 0.15 !
 H202 |
 H151
 H163

 ATOM C11
 CC1A
 -0.15 !
 H203
 H203
 H163

 ATOM H111 HE1 0.15 ATOM C12 CC1A -0.15 ATOM H121 HE1 0.15 GROUP ATOM C13 CC1B 0.00 ATOM C20 CT3 -0.27 ATOM H201 HA 0.09 АТОМ Н202 НА 0.09 ATOM H203 HA 0.09 GROUP ATOM C14 CC1B -0.15 !changed ATOM H141 HE1 0.15 ATOM C15 CC1A 0.23 !changed ATOM H151 HE1 0.15 ATOM 1NZ NS1 -0.60 !NS1 for deprotonated Schiff Base ATOM C16 CT3 -0.05 !added ATOM H161 HA 0.09 !added ATOM H162 HA 0.09 !added ATOM H163 HA 0.09 !added

RESI MATH		1.00	! 13-methyl fragment of retinal, all-trans
CROUP			! with net +1 overall charge
ATTOM C7	CC2	_0 /2	
ATOM U71	СС2 ЦЕ 2	0.42	: I H71 H81
ATOM H72	HE2	0.21	
ATOM C8	CC1A	-0 15	
ATOM H81	HE1	0.15	
GROUP	111111	0.10	H_{12} $C_{9=C10}$ H_{121}
ATOM C9	CC1B	0 00	
ATOM C19	СТЗ	-0 27	H191C19 $C11=C12$ $H141$
ATOM H191	НД	0.27	
ATOM H192	НД	0.09	· // / / / / / / / / / / / / / / / / /
ATOM H193	НЪ	0.09	I H103 / \ /
GROUP	117.1	0.05	$H^{2}(1) = -C^{2}(1)$
ATOM C10	CC1B	-0 15	
ATOM H101	HE1	0.15	. , , , , , , , , , , , , , , , , , , ,
ATOM C11		-0 15	H203 / \
АТОМ H111	HE1	0.15	н162 н163
ATOM C12		-0 15	11102 11105
ATOM 012	HE1	0.15	
GROUP		0.10	
ATOM C13	CC1B	0 00	
ATOM C20	CP	-0 27	
ATOM H201	НД	0.27	
ATOM H202	НΔ	0.09	
ATOM H203	НД	0.09	
GROUP	117.1	0.05	
ATOM C14	CC1B	-0 15	Ichanged
ATOM 014 ATOM H141	HE1	0.15	: changed
ATOM C15		0.13	Ichanged
ATOM H151	HR1	0.20	
ATOM 1NZ	NS2	-0 40	! NS2 for protonated Schiff Base
ATOM 2HZ1	HC	0 38	. Not for proconacea bonifi babe
ATOM C16	CTT 3	0.18	ladded
ATOM H161	НД	0.10	Ladded
ATOM H162	НД	0.09	Ladded
ATOM H163	НА	0.09	Ladded
111011 11105	111 1	0.05	·uuuuu
RESI RTC5		0.00	! retinal C5-methyl fragment.
			! nomenclature from PDB based on retinol
!			
GROUP			
ATOM C1	СТ	0.00	!
ATOM C2	CT2	-0.18	
ATOM H21	HA	0.09	
ATOM H22	HA	0.09	
АТОМ СЗ	CT2	-0.18	!
АТОМ НЗ1	HA	0.09	!
ATOM H32	HA	0.09	!
ATOM C4	CT2	-0.18	!
ATOM H41	HA	0.09	!
ATOM H42	HA	0.09	!
ATOM C5	CC1A	0.00	!
ATOM C6	CC1A	0.00	!
GROUP			!
ATOM C7	CC1B	-0.15	!

ATOM	H71	HE1	0.15	!		
ATOM	C8	CC1B	-0.15	!		
ATOM	H81	HE1	0.15	!		
ATOM	С9	CC1A	0.00	!		
ATOM	C10	CC1A	-0.42	!		
ATOM	H101	HE1	0.21	!		
ATOM	H102	HE1	0.21	!	Н162 Н163 Н171 Н172	
GROUE	2			!	/	
ATOM	C16	CT3	-0.27	!	Н161-С16 С17-Н173	H191 H192
ATOM	H161	HA	0.09	!	\setminus /	\setminus /
ATOM	H162	HA	0.09	!	H21 C1 H71 H81	C19
ATOM	H163	HA	0.09	!	$ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$	/ \
GROUE	2			!	H22-C2 C6C7====C8	С9 н193
ATOM	C17	CT3	-0.27	!		
ATOM	H171	HA	0.09	!	H31-C3 C5 H181	C10-H101
ATOM	H172	HA	0.09	!	/ \ / \ /	
ATOM	H173	HA	0.09	!	H32 C4 C18-H182	H102
GROUE	2			!	/ \ \	
ATOM	C18	CT3	-0.27	!	H41 H42 H183	
ATOM	H181	HA	0.09	!		
ATOM	H182	HA	0.09	!		
ATOM	H183	HA	0.09	!		
GROUI	2			!		
ATOM	C19	CT3	-0.27	!		
ATOM	H191	HA	0.09	!		
ATOM	H192	HA	0.09	!		
ATOM	H193	HA	0.09	!		

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