Compounds Which Serve as the Sole Source of Carbon or Nitrogen for Salmonella typhimurium LT-2

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About 600 compounds were screened as possible carbon or nitrogen sources for *Salmonella typhimurium* LT-2. About 100 utilizable compounds were found.

In an attempt to provide new genetic markers in the chromosome of *Salmonella typhimurium*, a search for compounds which can serve as a sole carbon source or as a sole nitrogen source for this organism was undertaken. About 100 utilizable compounds were found. Mapping of the genes concerned with the utilization of these compounds should add considerable detail to the known genetic map (5) on which 200 genes are already localized.

These compounds will also be useful in the investigation of strains carrying deletion mutations if a gene involved in the utilization of one of these compounds is deleted. For example, certain strains, in which part of the leucine operon is deleted, cannot utilize arabinose as a sole source of carbon because the deletion extends into the nearby arabinose operon (2), while a strain with a leucine deletion extending in the other direction cannot utilize a number of other carbon sources (Calvo et al., *unpublished data*). In contrast, we have determined that *his-520*, in which a large deletion extends through the histidine operon and lipopolysaccharide (rfb) regions, grows on all the listed sources of carbon and nitrogen.

The list of compounds should also be useful for taxonomic studies comparing related enteric bacteria. Such studies often make use of smaller lists of compounds (1, 4).

MATERIALS AND METHODS

Compounds were tested for their ability to support the growth of S. *typhimurium* strain LT-2 by a simple auxanographic technique. Minimal salts-agar was prepared by dissolving the following in a liter of distilled

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water: K_2SO_4 , 1.0 g; K_2HPO_4 , 13.5 g; KH_2PO_4 , 4.7 g; $MgSO_4 \cdot 7H_2O$, 0.1 g; agar, 15 g. For testing possible carbon sources, plates containing 30 ml of sterile minimal agar were overlayed with 2 ml of soft agar (0.6% in 0.5% NaCl solution) containing 0.1 ml of an overnight nutrient broth culture of bacteria and 0.1 ml of 0.1 M NH_4Cl. For testing possible nitrogen sources, the 0.1 M NH_4Cl was replaced by 0.2 ml of 50% glucose. When testing the amino acids as nitrogen sources, an additional set of plates was used containing 0.1% citrate as a sole carbon source.

Approximately 2 mg of each compound was placed on the surface of the solidified agar, and growth of the bacteria in the agar was recorded after incubation for 2 days at 37 C. Because the compound diffuses from the original point of deposition, a concentration gradient is established which allows the organism to grow even if certain concentrations are inhibitory. For example, compounds such as lactic acid, tested without prior neutralization, resulted in an inner zone of inhibition surrounded by a region of profuse growth. As controls, compounds were placed on similar plates lacking bacteria. No more than four compounds which gave positive results were each retested on an individual plate.

All chemicals were obtained from commercial sources; their purity was not determined, although several independent batches were tried whenever a positive response was obtained. This auxanographic technique was found to be simple, sensitive, and reliable.

RESULTS AND DISCUSSION

We have found about 100 utilizable carbon (Table 1) and nitrogen (Table 2) sources out of some 600 compounds that were screened. We first tested all of the natural compounds that seemed likely to be utilizable; this yielded most of our 100 compounds. Further screening yielded

Acetic acid	L-Fucose	Guanosine	Oleic acid
N-acetyl-D-glucosamine	Fumaric acid	Inosine	
			Oxaloacetic acid
N-acetyl-D-mannosamine	Galactaric acid (mucic	meso-Inositol	3-Phosphoglyceric acid
Adenosine ^a	acid)	DL-Isocitric acid	L-Proline ^a
L-Alanine ^a	Galactitol (dulcitol)	α -Ketoglutaric acid	Propionic acid
L-Arabinose	D-Galactose	L-Lactic acid	Pyruvic acid
DL-Citramalic acid	D-Glucaric acid (sac-	Lauric acid	L-Rhamnose
Citric acid	charic acid)	L-Lyxose	D-Ribose
L-Cysteine ^a	D-Gluconolactone	L-Malic acid	D-Serine ^a
Cytidine	D-Gluconic acid	D-Maltose	L-Serine ^a
2-Deoxyadenosine	D-Glucosamine	D-Mannitol	D-Sorbitol
2-Deoxycytidine	D-Glucose	D-Mannose	Succinic acid
2-Deoxyguanosine	D-Glucose-6-phosphate	D-Mannosamine	meso-Tartaric acid ^b
2-Deoxy-D-ribose	D-Glucuronic acid	Melibiose	Thymidine
Deoxyuridine	D-Glucuronolactone	L-Methionyl-L-alanine ^a	D-Trehalose
Dihydroxyacetone	DL-glyceric acid	6-Methylaminopurine	Tricarballylic acid
D-Erythrose ^b	Glycerol	riboside	Tridecanoic acid
D-Fructose	α-Glycerophosphate	α -Methyl-D-galactoside	Uridine
D-Fructose-6-phosphate	Glycolic acid	Myristic acid	D-Xylose

TABLE 1. Compounds which serve as sole carbon sources for Salmonella typhimurium LT-2

^a Can serve as the sole source of both carbon and nitrogen.

^b The bacteria will mutate to use this compound.

TABLE 2.	Compounds	which serv	e as sole nitrogen	
sources for Salmonella typhimurium				

Adenosine ^a L-Alanine ^a Ammonium chloride L-Arginine D-Asparagine L-Asparagine D-Aspartic acid L-Aspartic acid L-Cystine ^a L-Cystine	Deoxycytidine D-Glutamic acid L-Glutamic acid L-Glutamine Glutathione (reduced) Glycine L-Homoserine DL-β-Hydroxy glutamic acid L-Methionyl-L-alanine ^a
L-Asparagine	Glycine
L-Aspartic acid	DL- β -Hydroxy glutamic
L-Cystine	L-Methionyl-L-alanine ^a
Cytidine Cytosine	L-Proline ^a D-Serine ^a
Deoxyadenosine	L-Serine ^₄

^a Can serve as the sole source of both carbon and nitrogen.

few new utilizable compounds; therefore, we feel that the list is probably fairly exhaustive in its present form. Table 3 lists the compounds which are not utilizable as sole carbon or nitrogen sources.

The utilization by *Escherichia coli* of a number of carbon sources related to the Krebs cycle has recently been reviewed, with particular emphasis on regulation (3). The pattern of the utilization of fatty acids is of some interest, as there are detailed studies of this in *E. coli* (7). The pattern is fairly similar; the C_4 to C_{10} fatty acids are not utilized in either of the two organisms.

In comparison with organisms such as *Pseudo-monas* (6), *S. typhimurium* is somewhat limited in the range of compounds that can be utilized as a sole carbon source for growth. Thus, phosphoryl-

ated compounds are, in general, poorly utilized. Few of the fatty acids are utilized, amides and ester linkages are generally not hydrolyzed, aromatic rings are not degraded to useful carbon sources, few of the naturally occurring amino acids serve as carbon sources, large molecules are not degraded, and few of the wide variety of available sugars and sugar derivatives are utilized. Some known metabolic intermediates on the list of nonutilizable compounds are presumably not utilized because they do not get into the cells, e.g., *cis*-aconitic acid and ribose-5-phosphate, 6-phosphogluconate, 3-phosphoglycerate, and glucose-1-phosphate.

As noted in a footnote to Table 1, several compounds in the table only support the growth of mutant derivatives of the parent strain. It is likely that by using more bacteria in the inoculum or by employing mutagens, other mutants would be identified which utilized some of the compounds in Table 3. Further study of such clones might provide information on interesting pathways or uptake systems in *Salmonella*.

It is of interest to make a rough estimate of the percentage of the *Salmonella* genome used for the capacity to utilize these various carbon and nitrogen sources. An estimate is made as follows. There are about 4,000 genes in a *Salmonella* (or *E. coli*) if one takes the average size gene to be about 1,000 bases in length. If one allows a permease and about two enzymes for the utilization of each of the compounds involved, that would account for 300 genes. In addition, there may be some utilizable compounds that we have not tested; we allow 100 genes for this. Thus, approximately

	source jor buintor		
Acetamide	δ-Amino-n-valeric acid	D-Chondrosamine (gal-	Formimino glycine
Acetone oxime	Amygdalin	actosamine)	Formyl alanine
N-acetyl-D-alanine	Aniline	Cinnamic acid (trans)	Formyl glycine
N-acetyl-L-alanine	L-Anserine	Citraconate, sodium	Formyl-D-phenylalanine
Acetyl-D-alloisoleucine	Anthranilic acid	L-Citrulline	Formyl-L-phenylalanine
N-acetyl-S-benzyl-DL-	Arabinic acid	Creatine	D-Fructose-1, 6-diphos-
cysteine	D-Arabinose	Creatinine	phate
Acetyl-dihydro-phenyl-	D-Arabitol	Crotonic acid	D-Fucose
alanine ethyl ester	L-Arabitol	Cyanate, potassium	2-Furan pyruvic acid
N-acetyl-galactosamine	D-Araboascorbic acid	DL(+)-Cystathionine,	oxime
N-acetyl-glycine	D-Arabonate, calcium	allo	β-D-Galactose-1-phos-
N-acetyl-L-leucine	Arcaine sulfate	Cysteic acid	phate
e-Acetyl-L-lysine	D-Arginine	3-Deoxyglucose	α -D-Galacturonic acid
N-acetyl-D-methionine	DL-Arterenol	2-Deoxy-D-glucose	Gelatin
N-acetyl-DL-methionine	L-Ascorbic acid	α,β -Diacetyl-diamino-	Gentiobiose
1-Acetyl-3-methyl urea	L-Aspartic diethyl ester	DL-propionic acid	d-Glucoascorbic acid
N-acetyl-D-phenylalanine	Azelaic acid	3,5-Diamino benzoic	Glucoheptonic acid
Acetyl-DL-phenylalanyl	L-Azetidine-2-carboxylic	acid	D-gluco-Heptulose
glycine	acid	2,4-Diaminobutyric acid	L-gluco-Heptulose
Acetylene dicarboxylic	Barbital, sodium	4,5-Diamino-2,6-dihy-	α -D-Glucose pentaacetate
acid	Barbituric acid	droxypyrimidine	β -D-Glucose-1-phosphate
Acetyl-DL-proline	Benzamide	α, ϵ -Diaminopimelic acid	L-Glutamic acid dimethyl
Acetyl-DL-tyrosine	Benzidine	2,6-Diaminopurine	ester
cis-Aconitic acid	Benzilic acid	L-Dihydroorotic acid	L-Glutamic acid- α -ethyl
Adenine	Benzoic acid	Dihydrourocanic acid	ester
Adenosine-3, 5-cyclic	Benzoin	Dihydroxynorephedrine ·	Glutaric acid
phosphate	Benzoin acetate	HCl	DL-Glyceraldehyde
Adenosine-5-phosphate	DL-N-benzoyl-2-alanine	Dihydroxyphenylala-	D-glycero-D-allo-heptose
Adipamide	Benzoyl glycine	nine (DOPA)	D-glycero-D-galacto-hep-
Adipic acid	S-benzyl cysteine	β,β -Dimethyl acrylic acid	tose
Adonitol	Benzyl glycinate	5,5-Dimethyl-1,3-cyclo-	D-glycero-L-galacto-hep-
Agmatine sulfate	Betaine HCl	hexane dione	tose
β -Alanine	Bicarbonate, sodium	3,3-Dimethyl glutaric	D-glycero-D-gluco-hep-
D-Alanine	Bicine	acid	titol
DL- α -Alanine ethyl ester	Biotin	3,5-Dinitro salicylic acid	D-glycero-D-gluco-hep-
DL-Alanylglycylglycine	Biuret	Diphenylamine	tose
Allantoin	Brucine	Dipicolinic acid	D-glycero-L-gluco-heptose
Allantolactone	Bufotenine mono oxalate	Egg albumin	L-glycero-D-gulo-heptitol
D-Allose	hydrate	d-Epinephrine	D-glycero-D-gulo-heptose
D-Altrose	<i>n</i> -Butyl-DL-malate	Erythritol	D-glycero-D-ido-heptose
DL- α -Aminoadipic acid	n-Butyramide	erythro-L-galacto-octitol	β -Glycerophosphoric
p-Aminobenzoic acid	<i>n</i> -Butyric acid	D-erythro-L-gluco-octitol	acid
α -Aminobutyric acid	D-Butyroin	D-erythro-talo-octitol	Glycine amide
DL- β -Aminobutyric acid	Caffeine	D-erythro-L-talo-octose	Glycine ethyl ester
e-Amino-n-carpoic acid	n-Caproic acid	Esculin	Glycine methyl ester
ω -Aminocaprylic acid	Carbamyl glutamic acid	Ethanolamine	Glycocyamine
1-Aminocyclopentane	Carbamyl phosphate	Ethanolamine phosphate	Glycogen
carboxylic acid	N-carbobenzoxy glucos-	L-Ethionine	Glycyl-D-methionine
α -Aminodecanoic acid	amine	Ethyl acetate	Glycylglycine
Aminoguanidine sulfate	Casein	Ethyl anthranilate	Glyoxylic acid
Aminoguanidine bicar-		Ethyl lactate	Guanidine
bonate	Catechol	Ethyl levulinate	α -Guanidobutyric acid
α -Aminolauric acid	D-Cellobiose	N-ethylmaleimide	Guanido-DL-phenylala-
δ-Aminolevulinic acid	Cellotriose	Ethylamine HCl	nine
2-Amino-2-methyl-1,3-	Celtrobiose	Ethylene glycol	Guanidopipecolic acid
propanediol	L- α -Chloro- β -phenyl-	Ethyl itaconate	β-Guanidopropionic acid
DL-2-Amino phenylacetic	propionyl glycine	5-Fluorouracil deoxy-	Guanine
acid	Cholesterol	riboside	D-Gulonolactone
2-Aminopurine	Cholic acid	Folic acid	L-Gulonolactone
2-Aminopyridine	Choline chloride	Formate, sodium	Hemin
	N	<u>.</u>	n

 TABLE 3. Compounds which do not serve either as a sole carbon or a sole nitrogen source for Salmonella typhimurium^a

TABLE 3—Continued

n-Heptylic acid 1-Hexadecanol Hexamethylenetetramine Hippuric acid Histamine diphosphate L-Histidine L-Homocitrulline L-Homocysteic acid Hydrazine sulfate *p*-Hydrazinobenzoic acid Hydrocinnamic acid trans-*β*-Hydromuconic acid *p*-Hydroxybenzoic acid β -Hydroxybutyric acid $D(-)-\beta$ -Hydroxybutyric acid **DL-Hydroxylysine** Hydroxyproline Hydroxy-D-proline, allo Hydroxypyruvic acid 8-Hydroxyquinoline Hypoxanthine L-Iduronic acid barium Imidazole Indole Indole-3-acetic acid 3-(2-Amino-ethyl) indole hydrochloride Inosine-5-phosphate Inulin 3-Iodopropionic acid Isatin Isobutyric acid Isoglutamine DL-Isoleucamine · HCl D-Isoleucine, allo L-Isoleucine L-Isoleucine benzyl ester L-Isoleucine ethyl ester L-Isoleucine methyl ester L-Isoleucyl-L-isoleucine DL-N-isopropyl arterenol Itaconic acid 5-Keto-D-gluconate β -Ketoglutaric acid Kojic acid D-Lactic acid Lactose Lecithin L-Leucinamide D-Leucinamide Leucinamine **D**-Leucine L-Leucine L-Leucine ethyl ester L-Leucine methyl ester D-Leucyl-L-isoleucine Levulinic acid L-Lysine L-Lysine methyl ester **D-Lyxonic** acid **D-Lyxose** Norvalamine · HCl

Maleic acid D-Malic acid Malonamide Malonic acid DL-Mandelic acid L-Mannitol D-Mannoheptulose D-Mannuronic acid **D-Melezitose** Menthol glucuronide Mercaptosuccinic acid meso-Allitol meso-glycero-gulo-Heptitol L(+) meso Lanthionine D-Methionine L-Methionine **DL-Methionine** sulfone DL-Methionine sulfoxide α -Methyl-N-acetyl mannosamine furanoside Methyl amine · HCl Methyl albumin S-methyl-L-cysteine β -Methyl-D-galactoside α -Methyl-D-glucoside B-Methyl-D-glucoside DL- α -Methyl glutamic acid Methyl itaconate α -Methyl-D-mannoside Methyl- α -L-rhamnoside DL- α -Methyl serine DL-O-methyl serine Methyl-*β*-thiogalactoside α -Methyl-D-xyloside β -Methyl-D-xyloside Mevalonic acid 2,3-Monoacetone-Dxylulofuranose trans trans Muconic acid α -Naphthol β -Naphthyl alanine Nicotinamide Nicotinamide adenine dinucleotide Nicotinamide adenine dinucleotide phosphate Nicotinic acid Nitrate, sodium Nitrite, sodium p-Nitrobenzoic acid p-Nitrobenzoyl-L-glutamic acid Nitroguanidophenylalanine β -Nitroguanidopropionic acid 4-Nitroimidazole Norleucamine · HCl **DL-Norleucine** DL-Normetanephrine

D-Norvaline Nucleinate, sodium n-Octanoic acid Orcinol DL-Ornithine HCl Orotic acid Oxalic acid Palmitic acid D-Pantothenic acid **D**-Pantoyl lactone Perseitol Phenethyl- β -D-galactoside Phenol **D**-Phenylalanine L-Phenylalanine Phenylalanine methyl ester Phenylalanyl phenylalanine L-Phenylephrine DL-C-phenyl glycine DL- α -Phenyllactic acid 6-Phosphogluconic acid Phosphoribosyl pyrophosphate DL-O-phosphoserine DL-O-phosphothreonine Phthalic acid Phthalimide α -Picolinic acid Picric acid 2,6-Piperidine dicarboxylic acid Piperonal Polygalacturonic acid Prolamine · HCl **D**-Proline L-Proline methyl ester Propionamide Protocatechuic acid Putrescine 4(p-Nitrobenzyl) pyridine **Pyridoxine** Pyrocatechol Pyruvoyl glycine DL-Pyruvoyl phenylalanine Ouercetin **D-Raffinose** Resorcinol **D-Rhamnose** Riboflavin **D**-Ribonolactone D-Ribose-5-phosphate Salicin Salicylaldehyde Salicylamide Salicylic acid Sedoheptulose anhydride Semicarbazide · HCl DL-Serine ethyl ester

Serotonin Sorbic acid D-Sorbose L-Sorbose Spermidine Spermine Stearamide Stearanilide Stearic acid Succinamide Succinimide Sucrose Sulfanilate, sodium D-Tagaturonic acid **D**-Taloheptulose **D**-Talose Tannic acid **D**-Tartaric acid L-Tartaric acid Taurine Taurocholic acid 2,4,5,6-Tetra-aminopyrimidine sulfate Tetrahydroxysuccinic acid Tetramethylammonium bromide Thiamine Thioglycolate, sodium Thiosemicarbazide Thiourea Threamine D-threo-L-gulo-Octose **D**-Threonine L-Threonine L-Threonine, allo Thymine 2,4,5-Triamino-6-hydroxy pyrimidine sulfate Tripalmitin Tristearin Tryptamine D-Tryptophan L-Tryptophan L-Tryptophan ethyl ester **D**-Tyrosine L-Tyrosine L-Tyrosine-O-sulfate Uracil Urea Uric acid Valeric acid **D-Valine** L-Valine DL-Valine ethyl ester Vanillic acid Vanillin Vanillyl alcohol Xanthine Xanthosine Xanthurenic acid Xylitol L-Xylose

^a Additional compounds not serving as carbon sources are listed in Table 2, and additional compounds not serving as nitrogen sources are listed in Table 1.

10% (400/4,000) of the genome is responsible for the capacity to utilize these compounds.

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