Stimulation of Pyrophosphatase Activity in Clostridium pasteurianum by Reductants'

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Many biosynthetic reactions involving adenosine triphosphate (ATP) yield inorganic pyrophosphate as a by-product. Nitrogen fixation by extracts of Clostridium pasteurianum requires ATP as a source of energy and KBH₄, H_2 , or pyruvate as a source of reducing power (Hardy and D'Eustachio, Biochem. Biophys. Res. Comm. 15:314, 1964). While investigating the function of ATP in ammonia synthesis, we found that pyrophosphate accumulated only if the reducing agent was omitted. If one of the above reducing agents was present, neither pyrophosphate formation nor the inhibition of nitrogen fixation by pyrophosphate could be demonstrated. Further studies showed that a variety of reductants produced a 6- to 34-fold stimulation of the enzymatic hydrolysis of pyrophosphate to inorganic phosphate. Although the activity of some mammalian pyrophosphatases is occasionally enhanced by sulfhydryl compounds (Seal and Binkley, J. Biol. Chem. 228:193, 1957; Racker, Proc. Nat. Acad. Sci. U.S. 48:1639, 1962), the stimulation is much less than that observed with the clostridial enzyme. A brief examination of pyrophosphatase activity in several aerobes and anaerobes was done.

 $C.$ pasteurianum ATCC 6013 and $C.$ butyricum were grown anaerobically and harvested, as described by Carnahan and Castle (J. Bacteriol. 75:121, 1958). Bacillus cereus ATCC ⁹¹³⁹ was grown in the modified "G" medium of Stewart and Halvorson (J. Bacteriol. 65:160, 1953). Escherichia coli was grown in nutrient broth (Difco) supplemented with 1% yeast extract and 0.5 % glucose. Azotobacter vinelandii was grown on Burk's nitrogen-free medium (Burk and Lineweaver, J. Bacteriol. **19:**389, 1930). C. kluyveri, Micrococcus lysodeikticus, and baker's yeast were obtained as dried commercial preparations. The clostridia were extracted by use of the method described by Carnahan et al. (Biochim Biophys. Acta 44:520, 1960). The other organisms were suspended in water $(1 g/2 ml)$ and treated in a

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10-ke Raytheon sonic oscillator for 15 min. The disrupted cell suspensions were cooled to ¹ C, clarified by centrifugation at 27,000 \times g for 1 hr, and stored under an atmosphere of $H₂$.

Inorganic pyrophosphatase activity of all extracts was measured by use of a standardized

TABLE 1. Effect of various reductants on pyrophosphatase activity in extracts of Clostridium pasteurianum

Reductant	Amt added	Specific activity*	Stimu- lation
	umoles		
None		0.07	
KBH_4	20	2.36	$34\times$
Cysteine	20	1.65	$24\times$
Dithionite	20	1.43	$20\times$
$\textbf{Ascorbate} \dots \dots \dots \dots$	20	1.39	$20\times$
H_2	1 atm	1.39	$20\times$
	40	1.31	$17\times$
Formamidine sulfinic			
$\text{acid} \ldots \ldots \ldots \ldots \ldots$	20	1.0	$14\times$
Mercaptoethanol	30	0.66	$9\times$
Glutathione	20	0.55	8×
$NADPH1, \ldots, \ldots,$	20	0.55	$8\times$
$NADH_2$	20	$0.50\,$	7х

* Expressed as micromoles of pyrophosphate hydrolyzed per 30 min per milligram of protein. ^t Reduced nicotinamide adenine dinucleotide phosphate.

assay. Reactions were carried out in 2-ml volumes in 25-ml anaerobic flasks equipped with a side arm. The standard reaction mixture contained: tris(hydroxymethyl)aminomethane (Tris) buffer (pH 8.0), 100 μ moles; extract protein, 9 mg; Mg^{++} , 20 μ moles; pyrophosphate, 50 μ moles; reductant, 20 to 40 μ moles. The flasks were evacuated and filled with an atmosphere of argon. After pyrophosphate was tipped from the side arm, the flask was shaken at 30 C for 30 min. The incubation period was stopped by the addition of 1 ml of the reaction mixture to 1 ml of cold 10% trichloroacetic acid. After the precipitated protein was removed by centrifugation, samples of

the supernatant fluid were assayed for inorganic phosphate by the method of Taussky and Schorr (J. Biol. Chem. 202:675, 1953).

The reductants which stimulate pyrophos-

TABLE 2. Effect of reductant on pyrophosphatase activity in extracts of various organisms

	Specific activity*		
Organism	$-KBH_4$	$+KBH_4$	
Clostridium pasteurianum	0.05	2.16	
$C. \; kluyverii \ldots \ldots \ldots \ldots$	0.02	0.19	
$C.$ but yricum	0.04	0.08	
Baker's yeast	0.15	0.08	
Bacillus cereus	4.25	3.97	
$Escherichia coli \ldots \ldots$	$0.56\,$	0.55	
Micrococcus lysodeikticus	0.42	0.42	
$A zotobacter \ vincland iii$	0.13	0.07	
Rat liver	0.68	0.58	
Crystalline yeast pyro-			
$phostase. \ldots \ldots \ldots$	6.13	4.53	

* Expressed as micromoles of pyrophosphate hydrolyzed per 30 min per milligram of protein. phatase activity are listed in decreasing order of effectiveness in Table 1. Potassium borohydride produced a 34-fold stimulation, whereas reduced nicotinamide adenine dinucleotide (NADH2) raised the activity only 7-fold. Preincubation of the extract with KBH_4 or H_2 and immediate evacuation was as effective as a continuous supply of reductant.

Table 2 shows the inorganic pyrophosphatase activity in extracts of several organisms. The three anaerobic clostridia showed a 2- to 40-fold stimulation of pyrophosphate hydrolysis in the presence of KBH₄, whereas extracts from aerobic organisms showed no stimulation. Although this survey of organisms is limited, it suggests that the inorganic pyrophosphatases of anaerobic organisms require reductants for maximal activity, and that those from aerobic organisms do not. This phenomenon of a reductant-stimulated pyrophosphatase in extracts of clostridia should be considered in the study of reactions which eliminate pyrophosphate from ATP, such as amino acid activation.