

# Cubane-Type Co<sub>4</sub>S<sub>4</sub> Clusters: Synthesis, Redox Series, and Magnetic Ground States

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## Supporting Information

Figure S-1: Magnetic data for [Co(Pr<sup>*i*</sup><sub>2</sub>NH<sub>2</sub>Me<sub>2</sub>)<sub>2</sub>(S<sup>*t*</sup>Bu)<sub>2</sub>] (**1**).

Figure S-2: Magnetic data for [Co(Pr<sup>*i*</sup><sub>2</sub>NH<sub>2</sub>Me<sub>2</sub>)<sub>2</sub>Cl<sub>2</sub>] (**2**).

Figure S-3: Energy levels and Boltzmann population for [Co<sub>4</sub>S<sub>4</sub>(Pr<sup>*i*</sup><sub>2</sub>NH<sub>2</sub>Me<sub>2</sub>)<sub>4</sub>] (**5**).

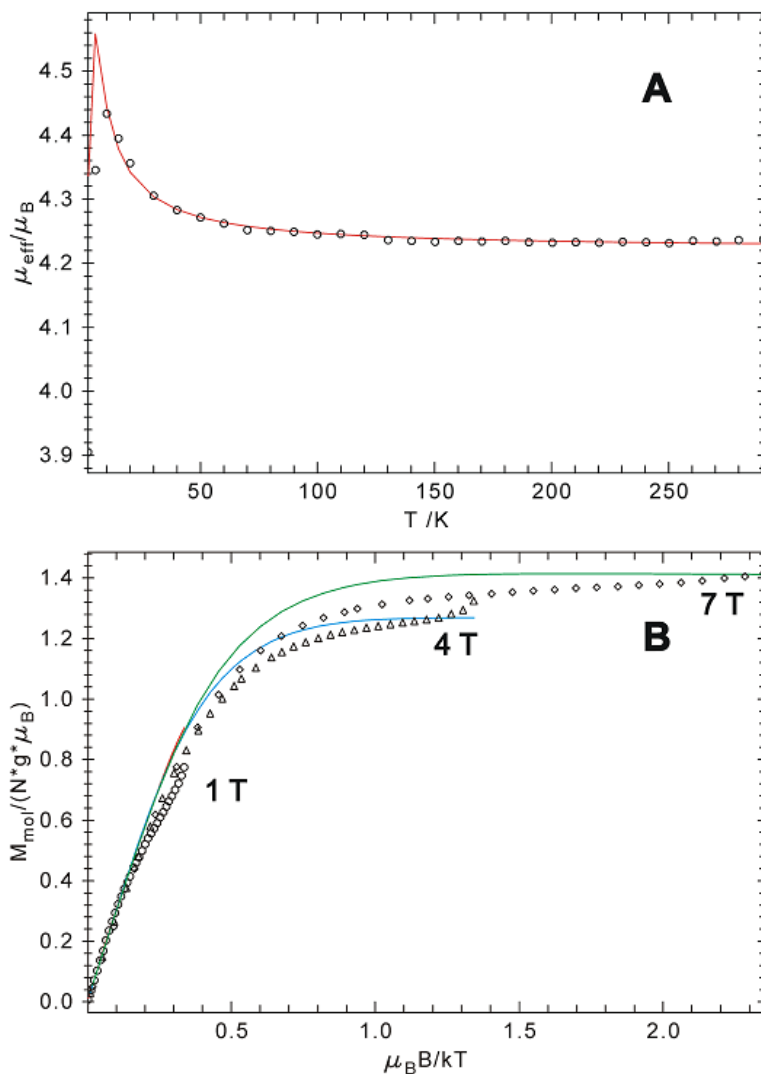
Figure S-4: Structure depiction for [Co<sub>8</sub>S<sub>8</sub>(PPr<sup>*i*</sup><sub>3</sub>)<sub>6</sub>] (**7**) and [Co<sub>8</sub>S<sub>8</sub>(PPr<sup>*i*</sup><sub>3</sub>)<sub>6</sub>]<sup>1+</sup> (**8**).

Table S-1. Crystal Data and Summary of Data Collection and Refinement for **1**, **3**, [**4**](BF<sub>4</sub>), **5**, [**6**](BPh<sub>4</sub>), **7**, and [**8**](BF<sub>4</sub>).

Table S-2. Selected Interatomic Distances (Å) and Angles (deg) of [Co<sub>4</sub>S<sub>4</sub>(PPr<sup>*i*</sup><sub>3</sub>)<sub>4</sub>] and [Co<sub>4</sub>S<sub>4</sub>(Pr<sup>*i*</sup><sub>2</sub>NH<sub>2</sub>Me<sub>2</sub>)<sub>4</sub>].

Table S-3. Selected Interatomic Distances (Å) and Angles (deg) of [Co<sub>4</sub>S<sub>4</sub>(PPr<sup>*i*</sup><sub>3</sub>)<sub>4</sub>]<sup>1+</sup> and [Co<sub>4</sub>S<sub>4</sub>(Pr<sup>*i*</sup><sub>2</sub>NH<sub>2</sub>Me<sub>2</sub>)<sub>4</sub>]<sup>1+</sup>.

**Figure S-1:** (A) Temperature dependence of the effective magnetic moment, and (B) multi-field variable temperature measurement of  $[\text{Co}(\text{Pr}^{\text{f}}_2\text{NHCMe}_2)_2(\text{SBU}^{\text{f}})_2]$  (**1**). The lines in (A) are spin Hamiltonian simulations for  $S = 3/2$  with the following parameters: (A)  $g_{\text{Co}} = 2.186$ ,  $D_{\text{Co}} = 0$ , and  $\Theta_{\text{W}} = 0.8$  K. (B): The Brillouin function for  $S = 3/2$  calculated with  $g_{\text{Co}} = 2.$ ,  $D_{\text{Co}} = 0$ ,  $\Theta_{\text{W}} = 0$ .

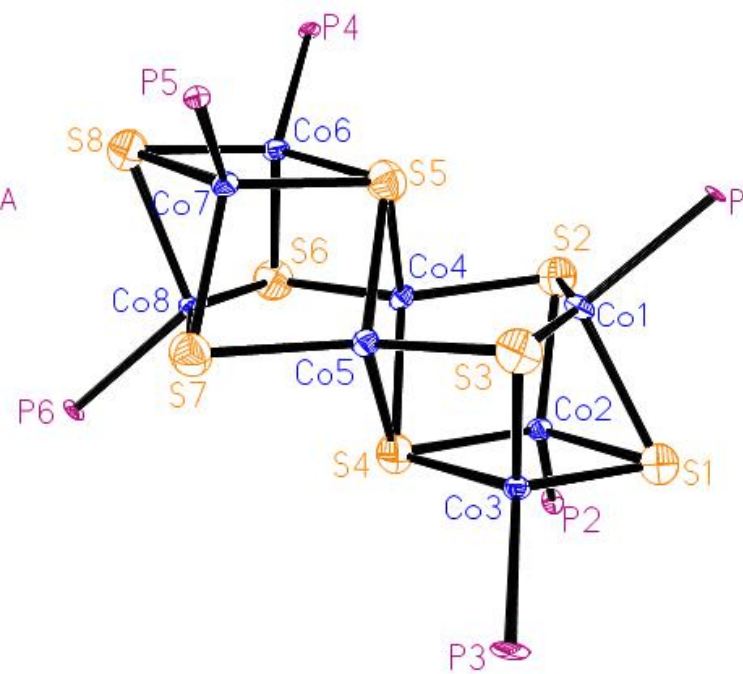
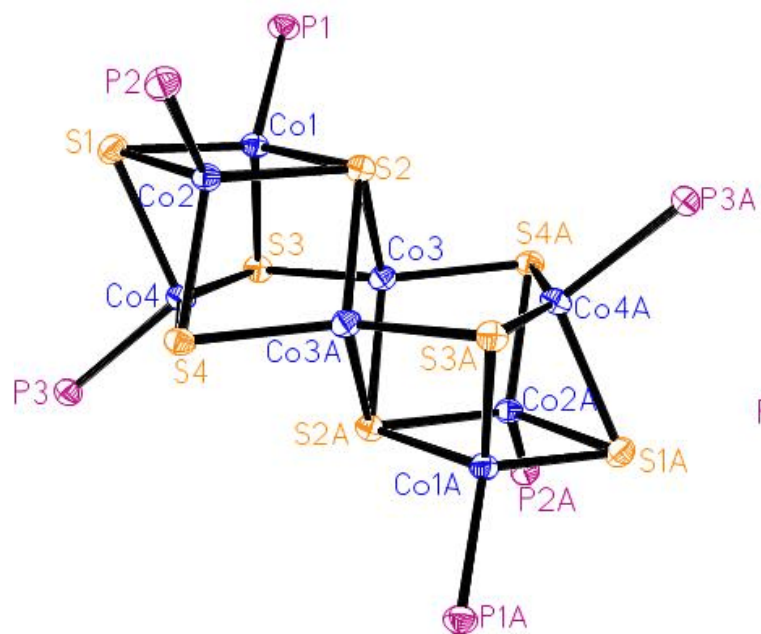
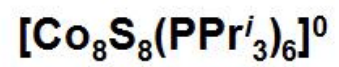


**Figure S-2:**

(A) Temperature dependence of the effective magnetic moment, and (B) multi-field variable temperature measurement of  $[\text{Co}(\text{Pr}^i_2\text{NHCM}_e_2)_2\text{Cl}_2]$  (**2**). The lines are spin Hamiltonian simulations for  $S = 3/2$  with parameters for (A):  $g_{\text{Co}} = 2.18$ ,  $D_{\text{Co}} = 2.3 \text{ cm}^{-1}$ , and  $\Theta_W = 0.7 \text{ K}$ . (B):  $g = 2.18$ ,  $D_{\text{Co}} = 4.3 \text{ cm}^{-1}$ , and  $\Theta_W = 1.0 \text{ K}$ . The values of  $D_{\text{Co}}$  and  $\Theta_W$  in (B) are optimized independently from those in (A).



**Figure S-3.** (A) Energies of the four total-spin manifolds for  $[\text{Co}_4\text{S}_4(\text{Pr}^i_2\text{NHCMe}_2)_4]$  (**5**) as function of an applied field, and calculated with the parameter used for the simulation in Figure 8 ( $S_i = 3/2$ ,  $i = 1-4$ ;  $J = -420 \text{ cm}^{-1}$ ,  $J' = -100 \text{ cm}^{-1}$ ,  $g_1 = 2.17$ ,  $g_{2-4} = 2.0$ ,  $D_i = 0$ ). (B) Boltzmann population of the lowest level as function of temperature.



**Figure S-4:** Molecular structures of  $[\text{Co}_8\text{S}_8(\text{PPr}^i_3)_6]$  (**7**) and  $[\text{Co}_8\text{S}_8(\text{PPr}^i_3)_6]^{1+}$  (**8**). Isopropyl groups are omitted for clarity

**Table S-1.** Crystal Data and Summary of Data Collection and Refinement for **1**, **3**, **[4]**(BF<sub>4</sub>), **5**, **[6]**(BPh<sub>4</sub>), **7**, and **[8]**(BF<sub>4</sub>).<sup>a,b</sup>

	<b>1</b>	<b>3</b>	<b>[4]</b> (BF <sub>4</sub> )	<b>5</b>	<b>[6]</b> (BPh <sub>4</sub> )	<b>7</b>	<b>[8]</b> (BF <sub>4</sub> )
formula	C <sub>30</sub> H <sub>58</sub> Co N <sub>4</sub> S <sub>2</sub>	C <sub>36</sub> H <sub>84</sub> Co <sub>4</sub> P <sub>4</sub> S <sub>4</sub>	C <sub>36</sub> H <sub>84</sub> BCo <sub>4</sub> F <sub>4</sub> P <sub>4</sub> S <sub>4</sub>	C <sub>44</sub> H <sub>80</sub> Co <sub>4</sub> N <sub>8</sub> S <sub>4</sub>	C <sub>68</sub> H <sub>100</sub> B CO <sub>4</sub> N <sub>8</sub> S <sub>4</sub>	C <sub>54</sub> H <sub>126</sub> Co <sub>8</sub> P <sub>6</sub> S <sub>8</sub>	C <sub>54</sub> H <sub>126</sub> B Co <sub>8</sub> F <sub>4</sub> P <sub>6</sub> S <sub>8</sub>
fw	597.85	990.76	1091.68	1085.12	1404.33	1689.29	1776.10
T, K	193(2)	130(2)	193(2)	105(2)	100(2)	193(2)	193(2)
crystal system	triclinic	monoclinic	orthorhombic	monoclinic	triclinic	triclinic	triclinic
space group	<i>P1</i>	<i>C2/c</i>	<i>I-42m</i>	<i>Cc</i>	<i>P-1</i>	<i>P-1</i>	<i>P-1</i>
a, Å	9.890(1)	19.057(3)	12.442(1)	14.321(3)	9.507(1)	12.3518(6)	13.5596(7)
b, Å	10.303(1)	12.780(2)	12.442(1)	20.380(5)	18.56(1)	12.4877(6)	20.897(1)
c, Å	10.437(1)	39.863(6)	16.575(2)	18.726(4)	20.32(1)	13.9495(7)	41.467(2)
α, deg	106.082(2)	90	90	90	85.581(2)	73.960(1)	86.085(1)
β, deg	99.487(2)	92.948(3)	90	96.093(4)	87.589(3)	84.740(1)	81.220(1)
γ, deg	117.495(2)	90	90	90	88.381(2)	64.229(1)	88.473(1)
V, Å <sup>3</sup>	851.3(1)	9695(3)	2565.8(4)	5434(2)	3571(2)	1861.2(2)	11584(1)
Z	1	8	2	4	2	1	6
d <sub>calcd</sub> , g/cm <sup>3</sup>	1.166	1.357	1.413	1.326	1.306	1.507	1.528
2θ range, deg	4.3 to 50.0	2.0-50.0	4.1-50.0	3.4-50.0	4.0-50.0	3.6-50.0	2.7-50.0
GOF (F <sup>2</sup> )	0.937	1.070	1.009	1.027	1.034	1.027	1.047
R1 <sup>b</sup>	0.0439, <sup>d</sup> 0.0520 <sup>e</sup>	0.0749, <sup>d</sup> 0.0919 <sup>e</sup>	0.0255, <sup>d</sup> 0.0260 <sup>e</sup>	0.0200, <sup>d</sup> 0.0205 <sup>e</sup>	0.0447, <sup>d</sup> 0.0597 <sup>e</sup>	0.0344, <sup>d</sup> 0.0393 <sup>e</sup>	0.0538, <sup>d</sup> 0.0785 <sup>e</sup>
wR2 <sup>c</sup>	0.0870, <sup>d</sup> 0.0913 <sup>e</sup>	0.1981, <sup>d</sup> 0.2109 <sup>e</sup>	0.0672, <sup>d</sup> 0.0677 <sup>e</sup>	0.0489, <sup>d</sup> 0.0492 <sup>e</sup>	0.1117, <sup>d</sup> 0.1215 <sup>e</sup>	0.0893, <sup>d</sup> 0.0940 <sup>e</sup>	0.1430, <sup>d</sup> 0.1589 <sup>e</sup>

<sup>a</sup> Collected using Mo Kα radiation (λ = 0.71073 Å). <sup>b</sup> R1 = Σ[(F<sub>o</sub> - F<sub>c</sub>)] / Σ(F<sub>o</sub>). <sup>c</sup> wR2 = {Σ[w(F<sub>o</sub><sup>2</sup> - F<sub>c</sub><sup>2</sup>)<sup>2</sup> / Σ[w(F<sub>o</sub><sup>2</sup>)<sup>2</sup>]}<sup>1/2</sup>. <sup>d</sup> I > 2\_σ(I). <sup>e</sup> All data.



**Table S-2.** Selected Interatomic Distances (Å) and Angles (deg) of [Co<sub>4</sub>S<sub>4</sub>(PPr<sup>i</sup>)<sub>4</sub>] and [Co<sub>4</sub>S<sub>4</sub>(Pr<sup>i</sup><sub>2</sub>NHCMe<sub>2</sub>)<sub>4</sub>]

<b>[Co<sub>4</sub>S<sub>4</sub>(PPr<sup>i</sup>)<sub>4</sub>]</b>					
Co1-S1	2.223(2)	Co1-Co2	2.618(1)	S-Co-S	103.8(1)-106.39(7)
Co1-S2	2.229(2)	Co1-Co3	2.604(1)	Co-S-Co	71.24(6)-72.00(6)
Co1-S3	2.232(2)	Co1-Co4	2.604(1)		
Co2-S1	2.231(2)	Co2-Co3	2.602(1)		
Co2-S2	2.236(2)	Co2-Co4	2.600(1)		
Co2-S4	2.233(2)	Co3-Co4	2.592(2)		
Co3-S2	2.232(2)	mean of 6	2.603[9]		
Co3-S3	2.218(3)				
Co3-S4	2.223(2)	Co2-P2 <sup>a</sup>	2.235(2)		
Co4-S1	2.220(2)				
Co4-S3	2.214(3)	P2-Co2-S1 <sup>a</sup>	114.6(1)		
Co4-S4	2.210(2)	P2-Co2-S2	112.7(1)		
mean of 122.223[8]		P2-Co2-S4	112.5(2)		
<b>[Co<sub>4</sub>S<sub>4</sub>(Pr<sup>i</sup><sub>2</sub>NHCMe<sub>2</sub>)<sub>4</sub>]<sup>0</sup></b>					
Co1-S1	2.264(2)	Co1-Co2	2.723(1)	S-Co-S	103.17(3)-105.26(3)
Co1-S2	2.256(1)	Co1-Co3	2.677(1)	Co-S-Co	72.45(2)-74.13(3)
Co1-S3	2.240(1)	Co1-Co4	2.678(1)		
Co2-S1	2.253(1)	Co2-Co3	2.711(1)		
Co2-S2	2.262(1)	Co2-Co4	2.679(1)		
Co2-S4	2.260(1)	Co3-Co4	2.680(1)		
Co3-S1	2.250(1)	mean of 6	2.69[2]		
Co3-S3	2.246(1)				
Co3-S4	2.245(1)	Co1-C1	1.988(2)		
Co4-S2	2.245(1)	Co2-C2	2.000(2)		
Co4-S3	2.254(1)	Co3-C3	1.973(2)		
Co4-S4	2.274(1)	Co4-C4	1.979(2)		
mean of 122.25[1]	mean of 4		1.99[1]		

<sup>a</sup>Data for non-disordered P2 atom.

**Table S-3.** Selected Interatomic Distances (Å) and Angles (deg) of  $[\text{Co}_4\text{S}_4(\text{PPr}^i_3)_4]^{1+}$  and  $[\text{Co}_4\text{S}_4(\text{Pr}^i_2\text{NHCMe}_2)_4]^{1+}$

$[\text{Co}_4\text{S}_4(\text{PPr}^i_3)_4]^{1+}$				
Co1-S1	2.206(1)	S1-Co1-S1A	105.20(4)	
Co1-S1C	2.207(1)	S1-Co1-S1C	104.94(3)	
Co1-Co1A	2.612(1)	Co1-S1-Co1C	72.59(3)	
Co1-Co1B	2.606(1)	Co1-S1-Co1B	72.44(3)	
Co1-P1	2.257(1)	P1-Co1-S1	113.18(3)	
		P1-Co1-S1C	114.49(4)	
$[\text{Co}_4\text{S}_4(\text{Pr}^i_2\text{NHCMe}_2)_4]^{1+}$				
Co1-S1	2.221(1)	Co1-Co2	2.661(1)	S-Co-S103.09(3)-105.69(3)
Co1-S2	2.209(1)	Co1-Co3	2.682(1)	Co-S-Co 72.53(3)-74.87(3)
Co1-S4	2.223(1)	Co1-Co4	2.622(1)	
Co2-S1	2.231(1)	Co2-Co3	2.684(1)	
Co2-S2	2.227(1)	Co2-Co4	2.663(1)	
Co2-S3	2.260(1)	Co3-Co4	2.658(1)	
Co3-S2	2.204(1)	mean of 6	2.66[2]	
Co3-S3	2.232(1)			
Co3-S4	2.238(1)	Co1-C1	1.977(3)	
Co4-S1	2.204(1)	Co2-C12	1.984(3)	
Co4-S3	2.227(1)	Co3-C23	1.984(3)	
Co4-S4	2.210(1)	Co4-C34	1.964(3)	
mean of 12	2.22[2]	mean of 4	1.977[9]	