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Metalloporphyrin Mixed-Valence π -Cation Radicals: $[\text{Fe}(\text{oxoOEC}^{\bullet/2})(\text{Cl})]_2\text{SbCl}_6$, Structure, Magnetic Properties, and Near-IR Spectra

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Table S1. Complete Crystallographic Details for $[\text{Fe}(\text{oxoOEC}^{*/2})(\text{Cl})]_2\text{SbCl}_6$

formula	$\text{C}_{37}\text{H}_{46}\text{Cl}_6\text{FeN}_4\text{OSb}_{0.50}$
FW, amu	892.20
a , Å	10.333(2)
b , Å	12.555(3)
c , Å	15.849(3)
α , deg	83.64(3)
β , deg	77.53(3)
γ , deg	87.19(3)
V , Å ³	1994.6(7)
space group	$P\bar{1}$
Z	2
D_c , g/cm ³	1.486
F(000)	915
μ , mm ⁻¹	1.146
crystal dimensions, mm	$0.2 \times 0.15 \times 0.02$
radiation	MoK α , $\bar{\lambda} = 0.71073$ Å
temperature, K	127(2)
diffractometer	Enraf-Nonius Fast Area Detector
θ range for collected data, deg	2.20–25.50
index range	$-12 \leq h \leq 12$ $-10 \leq k \leq 15$ $-19 \leq l \leq 19$
total data collected	14683
absorption correction	Empirical (DIFFABS)
relative transmission coefficients (I)	1.0 and 0.3477
unique data	7319 ($R_{\text{int}} = 0.1391$)
unique observed data [$I > 2\sigma(I)$]	4843
refinement method	Full-matrix least-squares on F^2
data/restraints/parameters	7319/0/453
goodness-of-fit (based on F^2)	1.030
final R indices [$I > 2\sigma(I)$]	$R_1 = 0.0983$, $wR_2 = 0.2397$
final R indices (all data)	$R_1 = 0.1633$, $wR_2 = 0.2935$

Table S2. Atomic Coordinates and Equivalent Isotropic Displacement Parameters (\AA^2) for $[\text{Fe}(\text{oxoOEC}^{*/2})(\text{Cl})]_2\text{SbCl}_6^a$

atom	x	y	z	$U(\text{eq})$
Fe(1)	0.86665(13)	0.20352(11)	0.07576(9)	0.0401(4)
Cl(1)	0.6892(2)	0.3098(2)	0.07664(17)	0.0484(6)
O(1)	0.5883(8)	-0.0728(6)	0.3321(5)	0.058(2)
N(1)	0.9778(8)	0.2896(6)	0.1358(5)	0.0379(17)
N(2)	0.8062(8)	0.1111(6)	0.1975(5)	0.0377(17)
N(3)	0.8331(8)	0.0686(6)	0.0191(5)	0.0407(18)
N(4)	0.9945(7)	0.2511(6)	-0.0386(5)	0.0393(18)
C(a1)	1.0615(9)	0.3705(8)	0.0996(6)	0.043(2)
C(a2)	0.9688(11)	0.2905(8)	0.2245(7)	0.048(2)
C(a3)	0.8110(9)	0.1408(8)	0.2744(6)	0.041(2)
C(a4)	0.7281(10)	0.0203(9)	0.2080(7)	0.049(3)
C(a5)	0.7534(10)	-0.0170(8)	0.0582(7)	0.045(2)
C(a6)	0.8606(9)	0.0580(8)	-0.0685(7)	0.045(2)
C(a7)	1.0015(10)	0.2163(9)	-0.1187(6)	0.044(2)
C(a8)	1.0713(10)	0.3405(9)	-0.0546(7)	0.049(3)
C(b1)	1.1084(10)	0.4226(8)	0.1634(7)	0.044(2)
C(b2)	1.0506(9)	0.3702(7)	0.2425(6)	0.039(2)
C(b3)	0.7321(11)	0.0730(8)	0.3506(6)	0.045(2)
C(b4)	0.6698(10)	-0.0034(9)	0.3023(7)	0.049(2)
C(b5)	0.7344(11)	-0.0836(9)	-0.0089(7)	0.050(2)
C(b6)	0.7991(10)	-0.0383(9)	-0.0846(7)	0.048(2)
C(b7)	1.0856(10)	0.2823(9)	-0.1864(6)	0.045(2)
C(b8)	1.1264(10)	0.3635(8)	-0.1464(7)	0.047(2)
C(m1)	0.8850(10)	0.2245(8)	0.2883(6)	0.043(2)
C(m2)	0.7048(9)	-0.0355(8)	0.1468(6)	0.042(2)
C(m3)	0.9388(11)	0.1258(8)	-0.1308(7)	0.047(2)
C(m4)	1.1010(9)	0.3957(9)	0.0097(6)	0.044(2)
C(11)	1.2064(10)	0.5099(9)	0.1442(7)	0.052(3)
C(12)	1.3483(8)	0.4684(9)	0.1153(6)	0.068(3)
C(21)	1.0755(8)	0.3840(7)	0.3290(6)	0.048(2)
C(22)	1.1633(13)	0.2934(10)	0.3622(8)	0.066(3)
C(31)	0.8251(12)	0.0080(10)	0.4032(7)	0.055(3)
C(32)	0.7588(13)	-0.0709(11)	0.4782(9)	0.072(4)
C(41)	0.6211(12)	0.1351(9)	0.4056(7)	0.057(3)

Table S2. Continued

atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
C(42)	0.5294(12)	0.1997(10)	0.3521(9)	0.064(3)
C(51)	0.6524(10)	-0.1808(8)	0.0125(8)	0.050(2)
C(52)	0.5029(11)	-0.1581(9)	0.0325(8)	0.058(3)
C(61)	0.8036(11)	-0.0758(10)	-0.1728(7)	0.055(3)
C(62)	0.6774(14)	-0.0556(11)	-0.2070(9)	0.068(3)
C(71)	1.1222(12)	0.2630(11)	-0.2813(8)	0.066(3)
C(72)	1.0563(13)	0.3420(10)	-0.3386(8)	0.063(3)
C(81)	1.2214(11)	0.4514(10)	-0.1870(8)	0.058(3)
C(82)	1.3643(12)	0.4244(11)	-0.1839(9)	0.068(4)
Sb	0.5000	0.5000	0.5000	0.0527(3)
Cl(3)	0.6928(3)	0.4197(2)	0.4183(2)	0.0638(8)
Cl(4)	0.5744(3)	0.6719(2)	0.4373(2)	0.0591(7)
Cl(5)	0.3881(3)	0.4909(3)	0.3859(2)	0.0681(8)
C(1)	0.6982(12)	0.2185(10)	0.6281(11)	0.076(4)
Cl(11)	0.6832(4)	0.2511(3)	0.7448(2)	0.0773(9)
Cl(12)	0.8478(4)	0.1387(4)	0.6066(3)	0.0886(12)

^a*U*(eq) is defined as one third of the trace of the orthogonalized \mathbf{U}_{ij} tensor. The estimated standard deviations of the least significant digits are given in parentheses.

Table S3. Bond Lengths for $[\text{Fe}(\text{oxoOEC}^{\bullet/2})(\text{Cl})]_2\text{SbCl}_6^a$

bond	length (Å)	bond	length (Å)
Fe(1)–N(4)	2.047(8)	Sb–Cl(4)#1	2.364(3)
Fe(1)–N(1)	2.053(8)	Sb–Cl(3)#1	2.375(3)
Fe(1)–N(3)	2.080(8)	Sb–Cl(3)	2.375(3)
Fe(1)–N(2)	2.124(8)	C(1)–Cl(12)	1.790(12)
Fe(1)–Cl(1)	2.213(3)	C(1)–Cl(11)	1.911(16)
O(1)–C(b4)	1.225(12)	C(m1)–H(m1)	0.9500
N(1)–C(a1)	1.360(12)	C(m2)–H(m2)	0.9500
N(1)–C(a2)	1.390(13)	C(m3)–H(m3)	0.9500
N(2)–C(a3)	1.325(13)	C(m4)–H(m4)	0.9500
N(2)–C(a4)	1.403(13)	C(11)–H(1a)	0.9900
N(3)–C(a6)	1.376(13)	C(11)–H(1b)	0.9900
N(3)–C(a5)	1.392(12)	C(12)–H(1c)	0.9800
N(4)–C(a8)	1.375(13)	C(12)–H(1d)	0.9800
N(4)–C(a7)	1.376(12)	C(12)–H(1e)	0.9800
C(a1)–C(m4)	1.399(13)	C(21)–H(2a)	0.9900
C(a1)–C(b1)	1.439(14)	C(21)–H(2b)	0.9900
C(a2)–C(m1)	1.401(14)	C(22)–H(2c)	0.9800
C(a2)–C(b2)	1.432(14)	C(22)–H(2d)	0.9800
C(a3)–C(m1)	1.393(14)	C(22)–H(2e)	0.9800
C(a3)–C(b3)	1.505(13)	C(31)–H(3a)	0.9900
C(a4)–C(m2)	1.326(15)	C(31)–H(3b)	0.9900
C(a4)–C(b4)	1.490(15)	C(32)–H(3c)	0.9800
C(a5)–C(m2)	1.383(14)	C(32)–H(3d)	0.9800
C(a5)–C(b5)	1.473(15)	C(32)–H(3e)	0.9800
C(a6)–C(m3)	1.375(14)	C(41)–H(4a)	0.9900
C(a6)–C(b6)	1.463(15)	C(41)–H(4b)	0.9900
C(a7)–C(m3)	1.386(14)	C(42)–H(4c)	0.9800
C(a7)–C(b7)	1.436(13)	C(42)–H(4d)	0.9800
C(a8)–C(m4)	1.388(15)	C(42)–H(4e)	0.9800
C(a8)–C(b8)	1.446(14)	C(51)–H(5a)	0.9900
C(b1)–C(b2)	1.376(14)	C(51)–H(5b)	0.9900
C(b1)–C(11)	1.490(14)	C(52)–H(5c)	0.9800
C(b2)–C(21)	1.478(13)	C(52)–H(5d)	0.9800
C(b3)–C(41)	1.522(16)	C(52)–H(5e)	0.9800

Table S3. Continued

bond	length (Å)	bond	length (Å)
C(b3)–C(b4)	1.534(14)	C(61)–H(6a)	0.9900
C(b3)–C(31)	1.551(15)	C(61)–H(6b)	0.9900
C(b5)–C(b6)	1.323(15)	C(62)–H(6c)	0.9800
C(b5)–C(51)	1.485(15)	C(62)–H(6d)	0.9800
C(b6)–C(61)	1.515(15)	C(62)–H(6e)	0.9800
C(b7)–C(b8)	1.385(15)	C(71)–H(7a)	0.9900
C(b7)–C(71)	1.513(16)	C(71)–H(7b)	0.9900
C(b8)–C(81)	1.507(14)	C(72)–H(7c)	0.9800
C(11)–C(12)	1.523(14)	C(72)–H(7d)	0.9800
C(21)–C(22)	1.538(14)	C(72)–H(7e)	0.9800
C(31)–C(32)	1.522(15)	C(81)–H(8a)	0.9900
C(41)–C(42)	1.547(17)	C(81)–H(8b)	0.9900
C(51)–C(52)	1.528(15)	C(82)–H(8c)	0.9800
C(61)–C(62)	1.517(17)	C(82)–H(8d)	0.9800
C(71)–C(72)	1.507(18)	C(82)–H(8e)	0.9800
C(81)–C(82)	1.509(17)	C(1)–H(1)	0.9900
Sb–Cl(5)	2.363(3)	C(1)–H(2)	0.9900
Sb–Cl(5)#1)	2.363(3)		
Sb–Cl(4)	2.364(3)		

^aThe estimated standard deviations of the least significant digits are given in parentheses.

Table S4. Bond Angles for $[\text{Fe}(\text{oxoOEC}^{\cdot/2})(\text{Cl})]_2\text{SbCl}_6^a$

angle	degree	angle	degree
N(4)–Fe(1)–N(1)	87.1(3)	Cl(4)#1–Sb–Cl(3)#1	90.26(10)
N(4)–Fe(1)–N(3)	87.3(3)	Cl(5)–Sb–Cl(3)	90.38(12)
N(1)–Fe(1)–N(3)	152.2(3)	Cl(5)#1–Sb–Cl(3)	89.62(13)
N(4)–Fe(1)–N(2)	155.4(3)	Cl(4)–Sb–Cl(3)	90.26(10)
N(1)–Fe(1)–N(2)	87.0(3)	Cl(4)#1–Sb–Cl(3)	89.74(10)
N(3)–Fe(1)–N(2)	86.8(3)	Cl(3)#1–Sb–Cl(3)	180.0
N(4)–Fe(1)–Cl(1)	104.9(2)	Cl(12)–C(1)–Cl(11)	104.1(8)
N(1)–Fe(1)–Cl(1)	101.5(2)	C(b1)–C(11)–H(1a)	109.1
N(3)–Fe(1)–Cl(1)	106.2(2)	C(12)–C(11)–H(1a)	109.1
N(2)–Fe(1)–Cl(1)	99.7(2)	C(b1)–C(11)–H(1b)	109.1
C(a1)–N(1)–C(a2)	103.8(8)	C(12)–C(11)–H(1b)	109.1
C(a1)–N(1)–Fe(1)	128.0(6)	H(1a)–C(11)–H(1b)	107.8
C(a2)–N(1)–Fe(1)	127.5(6)	C(11)–C(12)–H(1c)	109.5
C(a3)–N(2)–C(a4)	110.1(8)	C(11)–C(12)–H(1d)	109.5
C(a3)–N(2)–Fe(1)	125.9(6)	H(1c)–C(12)–H(1d)	109.5
C(a4)–N(2)–Fe(1)	122.6(6)	C(11)–C(12)–H(1e)	109.5
C(a6)–N(3)–C(a5)	105.7(8)	H(1c)–C(12)–H(1e)	109.5
C(a6)–N(3)–Fe(1)	125.9(6)	H(1d)–C(12)–H(1e)	109.5
C(a5)–N(3)–Fe(1)	126.9(7)	C(b2)–C(21)–H(2a)	109.0
C(a8)–N(4)–C(a7)	105.0(8)	C(22)–C(21)–H(2a)	109.0
C(a8)–N(4)–Fe(1)	125.9(7)	C(b2)–C(21)–H(2b)	109.0
C(a7)–N(4)–Fe(1)	127.8(6)	C(22)–C(21)–H(2b)	109.0
N(1)–C(a1)–C(m4)	122.7(9)	H(2a)–C(21)–H(2b)	107.8
N(1)–C(a1)–C(b1)	112.7(9)	C(21)–C(22)–H(2c)	109.5
C(m4)–C(a1)–C(b1)	124.5(9)	C(21)–C(22)–H(2d)	109.5
N(1)–C(a2)–C(m1)	123.9(9)	H(2c)–C(22)–H(2d)	109.5
N(1)–C(a2)–C(b2)	111.7(9)	C(21)–C(22)–H(2e)	109.5
C(m1)–C(a2)–C(b2)	124.4(10)	H(2c)–C(22)–H(2e)	109.5
N(2)–C(a3)–C(m1)	125.4(9)	H(2d)–C(22)–H(2e)	109.5
N(2)–C(a3)–C(b3)	114.6(9)	C(32)–C(31)–H(3a)	108.2
C(m1)–C(a3)–C(b3)	119.9(9)	C(b3)–C(31)–H(3a)	108.2
C(m2)–C(a4)–N(2)	127.9(9)	C(32)–C(31)–H(3b)	108.2
C(m2)–C(a4)–C(b4)	124.1(10)	C(b3)–C(31)–H(3b)	108.2
N(2)–C(a4)–C(b4)	107.9(9)	H(3a)–C(31)–H(3b)	107.4

Table S4. Continued

angle	degree	angle	degree
C(m2)–C(a5)–N(3)	123.8(9)	C(31)–C(32)–H(3c)	109.5
C(m2)–C(a5)–C(b5)	126.9(9)	C(31)–C(32)–H(3d)	109.5
N(3)–C(a5)–C(b5)	109.3(9)	H(3c)–C(32)–H(3d)	109.5
C(m3)–C(a6)–N(3)	124.6(9)	C(31)–C(32)–H(3e)	109.5
C(m3)–C(a6)–C(b6)	125.4(10)	H(3c)–C(32)–H(3e)	109.5
N(3)–C(a6)–C(b6)	109.9(8)	H(3d)–C(32)–H(3e)	109.5
N(4)–C(a7)–C(m3)	123.3(9)	C(b3)–C(41)–H(4a)	108.9
N(4)–C(a7)–C(b7)	111.6(9)	C(42)–C(41)–H(4a)	108.9
C(m3)–C(a7)–C(b7)	125.0(9)	C(b3)–C(41)–H(4b)	108.9
N(4)–C(a8)–C(m4)	124.2(9)	C(42)–C(41)–H(4b)	108.9
N(4)–C(a8)–C(b8)	111.2(9)	H(4a)–C(41)–H(4b)	107.7
C(m4)–C(a8)–C(b8)	124.4(10)	C(41)–C(42)–H(4c)	109.5
C(b2)–C(b1)–C(a1)	105.6(9)	C(41)–C(42)–H(4d)	109.5
C(b2)–C(b1)–C(11)	128.7(9)	H(4c)–C(42)–H(4d)	109.5
C(a1)–C(b1)–C(11)	125.6(9)	C(41)–C(42)–H(4e)	109.5
C(b1)–C(b2)–C(a2)	106.2(9)	H(4c)–C(42)–H(4e)	109.5
C(b1)–C(b2)–C(21)	128.2(9)	H(4d)–C(42)–H(4e)	109.5
C(a2)–C(b2)–C(21)	125.4(9)	C(b5)–C(51)–H(5a)	108.6
C(a3)–C(b3)–C(41)	113.4(9)	C(52)–C(51)–H(5a)	108.6
C(a3)–C(b3)–C(b4)	99.8(8)	C(b5)–C(51)–H(5b)	108.6
C(41)–C(b3)–C(b4)	108.5(9)	C(52)–C(51)–H(5b)	108.6
C(a3)–C(b3)–C(31)	110.8(9)	H(5a)–C(51)–H(5b)	107.6
C(41)–C(b3)–C(31)	114.0(9)	C(51)–C(52)–H(5c)	109.5
C(b4)–C(b3)–C(31)	109.4(9)	C(51)–C(52)–H(5d)	109.5
O(1)–C(b4)–C(a4)	123.9(10)	H(5c)–C(52)–H(5d)	109.5
O(1)–C(b4)–C(b3)	129.0(10)	C(51)–C(52)–H(5e)	109.5
C(a4)–C(b4)–C(b3)	107.2(9)	H(5c)–C(52)–H(5e)	109.5
C(b6)–C(b5)–C(a5)	107.2(10)	H(5d)–C(52)–H(5e)	109.5
C(b6)–C(b5)–C(51)	130.6(10)	C(b6)–C(61)–H(6a)	108.4
C(a5)–C(b5)–C(51)	122.2(10)	C(62)–C(61)–H(6a)	108.4
C(b5)–C(b6)–C(a6)	107.8(10)	C(b6)–C(61)–H(6b)	108.4
C(b5)–C(b6)–C(61)	126.3(10)	C(62)–C(61)–H(6b)	108.4
C(a6)–C(b6)–C(61)	125.8(9)	H(6a)–C(61)–H(6b)	107.5
C(b8)–C(b7)–C(a7)	106.1(9)	C(61)–C(62)–H(6c)	109.5

Table S4. Continued

angle	degree	angle	degree
C(b8)–C(b7)–C(71)	128.3(9)	C(61)–C(62)–H(6d)	109.5
C(a7)–C(b7)–C(71)	125.7(9)	H(6c)–C(62)–H(6d)	109.5
C(b7)–C(b8)–C(a8)	106.0(9)	C(61)–C(62)–H(6e)	109.5
C(b7)–C(b8)–C(81)	127.9(10)	H(6c)–C(62)–H(6e)	109.5
C(a8)–C(b8)–C(81)	125.7(10)	H(6d)–C(62)–H(6e)	109.5
C(a3)–C(m1)–C(a2)	126.5(10)	C(72)–C(71)–H(7a)	108.9
C(a4)–C(m2)–C(a5)	127.2(10)	C(b7)–C(71)–H(7a)	108.9
C(a6)–C(m3)–C(a7)	127.6(10)	C(72)–C(71)–H(7b)	108.9
C(a8)–C(m4)–C(a1)	126.9(9)	C(b7)–C(71)–H(7b)	108.9
C(b1)–C(11)–C(12)	112.6(10)	H(7a)–C(71)–H(7b)	107.7
C(b2)–C(21)–C(22)	113.0(8)	C(71)–C(72)–H(7c)	109.5
C(32)–C(31)–C(b3)	116.3(10)	C(71)–C(72)–H(7d)	109.5
C(a3)–C(m1)–H(m1)	116.7	H(7c)–C(72)–H(7d)	109.5
C(a2)–C(m1)–H(m1)	116.7	C(71)–C(72)–H(7e)	109.5
C(a4)–C(m2)–H(m2)	116.4	H(7c)–C(72)–H(7e)	109.5
C(a5)–C(m2)–H(m2)	116.4	H(7d)–C(72)–H(7e)	109.5
C(a6)–C(m3)–H(m3)	116.2	C(b8)–C(81)–H(8a)	108.6
C(a7)–C(m3)–H(m3)	116.2	C(82)–C(81)–H(8a)	108.6
C(a8)–C(m4)–H(m4)	116.5	C(b8)–C(81)–H(8b)	108.6
C(a1)–C(m4)–H(m4)	116.5	C(82)–C(81)–H(8b)	108.6
C(b3)–C(41)–C(42)	113.3(10)	H(8a)–C(81)–H(8b)	107.6
C(b5)–C(51)–C(52)	114.6(9)	C(81)–C(82)–H(8c)	109.5
C(b6)–C(61)–C(62)	115.4(10)	C(81)–C(82)–H(8d)	109.5
C(72)–C(71)–C(b7)	113.3(11)	H(8c)–C(82)–H(8d)	109.5
C(b8)–C(81)–C(82)	114.5(10)	C(81)–C(82)–H(8e)	109.5
Cl(5)–Sb–Cl(5)#1	180.0	H(8c)–C(82)–H(8e)	109.5
Cl(5)–Sb–Cl(4)	89.36(11)	H(8d)–C(82)–H(8e)	109.5
Cl(5)#1–Sb–Cl(4)	90.64(11)	Cl(12)–C(1)–H(1)	110.9
Cl(5)–Sb–Cl(4)#1	90.64(11)	Cl(11)–C(1)–H(1)	110.9
Cl(5)#1–Sb–Cl(4)#1	89.36(11)	Cl(12)–C(1)–H(2)	110.9
Cl(4)–Sb–Cl(4)#1	180.0	Cl(11)–C(1)–H(2)	110.9
Cl(5)–Sb–Cl(3)#1	89.62(13)	H(1)–C(1)–H(2)	109.0
Cl(5)#1–Sb–Cl(3)#1	90.38(12)		
Cl(4)–Sb–Cl(3)#1	89.74(10)		

^aThe estimated standard deviations of the least significant digits are given in parentheses.

Table S5. Anisotropic Displacement Parameters (\AA^2) for $[\text{Fe}(\text{oxoOEC}^{\cdot/2})(\text{Cl})]_2\text{SbCl}_6^a$

atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Fe(1)	0.0370(7)	0.0464(8)	0.0377(8)	0.0022(6)	-0.0114(5)	-0.0073(6)
Cl(1)	0.0430(13)	0.0515(14)	0.0529(15)	-0.0043(11)	-0.0159(10)	-0.0001(11)
O(1)	0.065(5)	0.060(5)	0.049(4)	0.000(4)	-0.011(4)	-0.021(4)
N(1)	0.046(5)	0.038(4)	0.030(4)	0.001(3)	-0.012(3)	-0.004(3)
N(2)	0.040(4)	0.044(4)	0.033(4)	-0.003(3)	-0.015(3)	-0.003(3)
N(3)	0.035(4)	0.044(4)	0.043(5)	0.004(3)	-0.012(3)	-0.005(3)
N(4)	0.033(4)	0.045(4)	0.040(4)	0.004(3)	-0.013(3)	-0.005(3)
C(a1)	0.037(5)	0.054(6)	0.042(5)	-0.004(4)	-0.016(4)	-0.007(4)
C(a2)	0.053(6)	0.038(5)	0.058(7)	-0.010(5)	-0.019(5)	0.002(4)
C(a3)	0.034(5)	0.048(5)	0.042(5)	0.007(4)	-0.012(4)	-0.003(4)
C(a4)	0.042(6)	0.051(6)	0.050(6)	0.010(5)	-0.009(4)	0.006(5)
C(a5)	0.039(5)	0.042(5)	0.052(6)	-0.002(4)	-0.009(4)	0.004(4)
C(a6)	0.037(5)	0.047(6)	0.053(6)	-0.003(5)	-0.012(4)	-0.003(4)
C(a7)	0.042(5)	0.057(6)	0.035(5)	0.000(4)	-0.013(4)	-0.004(4)
C(a8)	0.036(5)	0.066(7)	0.044(6)	0.010(5)	-0.013(4)	-0.006(5)
C(b1)	0.046(6)	0.044(5)	0.044(6)	0.003(4)	-0.021(4)	-0.001(4)
C(b2)	0.039(5)	0.039(5)	0.041(5)	-0.002(4)	-0.016(4)	0.006(4)
C(b3)	0.062(6)	0.048(6)	0.027(5)	0.002(4)	-0.015(4)	-0.005(5)
C(b4)	0.047(6)	0.048(6)	0.050(6)	-0.003(5)	-0.007(4)	-0.003(5)
C(b5)	0.056(6)	0.052(6)	0.048(6)	-0.010(5)	-0.019(5)	0.002(5)
C(b6)	0.044(6)	0.058(6)	0.042(6)	0.003(5)	-0.011(4)	-0.008(5)
C(b7)	0.044(5)	0.055(6)	0.039(5)	-0.007(4)	-0.009(4)	-0.009(4)
C(b8)	0.039(5)	0.052(6)	0.046(6)	0.010(5)	-0.008(4)	-0.011(4)
C(m1)	0.054(6)	0.042(5)	0.036(5)	-0.002(4)	-0.016(4)	0.000(4)
C(m2)	0.035(5)	0.049(6)	0.047(6)	-0.009(4)	-0.018(4)	-0.005(4)
C(m3)	0.058(6)	0.049(6)	0.038(5)	-0.012(4)	-0.019(4)	0.001(5)
C(m4)	0.037(5)	0.058(6)	0.036(5)	0.007(4)	-0.011(4)	-0.013(4)
C(11)	0.051(6)	0.056(6)	0.052(6)	0.004(5)	-0.023(5)	-0.018(5)
C(12)	0.052(7)	0.081(9)	0.077(9)	-0.007(7)	-0.025(6)	-0.013(6)
C(21)	0.056(6)	0.041(5)	0.051(6)	-0.008(4)	-0.023(5)	0.001(5)
C(22)	0.077(9)	0.069(8)	0.062(8)	-0.007(6)	-0.037(6)	0.000(6)
C(31)	0.064(7)	0.062(7)	0.043(6)	0.001(5)	-0.021(5)	-0.008(6)
C(32)	0.073(9)	0.077(9)	0.067(8)	0.020(7)	-0.029(6)	-0.020(7)
C(41)	0.068(8)	0.055(7)	0.047(6)	-0.002(5)	-0.009(5)	-0.014(6)

Table S5. Continued

atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
C(42)	0.055(7)	0.062(7)	0.074(8)	-0.013(6)	-0.013(6)	0.005(6)
C(51)	0.045(6)	0.046(6)	0.061(7)	-0.011(5)	-0.013(5)	-0.004(4)
C(52)	0.058(7)	0.046(6)	0.073(8)	-0.006(5)	-0.018(6)	-0.015(5)
C(61)	0.060(7)	0.060(7)	0.045(6)	-0.009(5)	-0.011(5)	-0.006(5)
C(62)	0.083(9)	0.070(8)	0.060(8)	-0.014(6)	-0.031(6)	0.003(7)
C(71)	0.055(7)	0.074(8)	0.064(8)	-0.003(6)	0.000(5)	-0.019(6)
C(72)	0.072(8)	0.066(8)	0.048(7)	0.002(5)	-0.012(5)	-0.005(6)
C(81)	0.052(7)	0.069(7)	0.050(7)	0.002(5)	-0.006(5)	-0.017(6)
C(82)	0.054(7)	0.082(9)	0.065(8)	0.021(7)	-0.016(6)	-0.019(6)
Sb	0.0543(7)	0.0491(6)	0.0497(6)	-0.0035(4)	0.0012(4)	-0.0096(5)
Cl(3)	0.0561(17)	0.0584(17)	0.0672(19)	-0.0072(13)	0.0099(13)	-0.0067(13)
Cl(4)	0.0585(17)	0.0487(15)	0.0647(18)	-0.0013(12)	-0.0012(13)	-0.0121(12)
Cl(5)	0.078(2)	0.0656(19)	0.0633(19)	-0.0063(14)	-0.0175(15)	-0.0128(15)
C(1)	0.039(6)	0.056(7)	0.135(13)	-0.025(8)	-0.012(7)	0.003(5)
Cl(11)	0.077(2)	0.079(2)	0.074(2)	-0.0157(17)	-0.0091(16)	0.0043(17)
Cl(12)	0.072(2)	0.114(3)	0.089(3)	-0.043(2)	-0.0219(18)	0.004(2)

^aThe estimated standard deviations of the least significant digits are given in parentheses. The anisotropic displacement factor exponent takes the form: $-2 \pi [h^2 U_{11} + \dots + 2 h k a^* b^* U_{12}]$.

Table S6. Hydrogen Atom Coordinates and Equivalent Isotropic Displacement Parameters (\AA^2) for $[\text{Fe}(\text{oxoOEC}^{•/2})(\text{Cl})]_2\text{SbCl}_6^a$

atom	x	y	z	$U(\text{eq})$
H(m1)	0.8780	0.2382	0.3469	0.052
H(m2)	0.6484	-0.0949	0.1659	0.050
H(m3)	0.9513	0.1087	-0.1890	0.056
H(m4)	1.1538	0.4572	-0.0093	0.053
H(1a)	1.1997	0.5472	0.1968	0.062
H(1b)	1.1843	0.5627	0.0979	0.062
H(1c)	1.3708	0.4160	0.1609	0.102
H(1d)	1.4089	0.5283	0.1048	0.102
H(1e)	1.3565	0.4340	0.0617	0.102
H(2a)	0.9896	0.3867	0.3711	0.057
H(2b)	1.1189	0.4533	0.3260	0.057
H(2c)	1.1195	0.2248	0.3674	0.099
H(2d)	1.1775	0.3071	0.4192	0.099
H(2e)	1.2490	0.2906	0.3212	0.099
H(3a)	0.8910	-0.0322	0.3627	0.066
H(3b)	0.8744	0.0594	0.4267	0.066
H(3c)	0.7055	-0.0313	0.5242	0.107
H(3d)	0.8268	-0.1148	0.5013	0.107
H(3e)	0.7014	-0.1173	0.4575	0.107
H(4a)	0.6607	0.1852	0.4365	0.069
H(4b)	0.5672	0.0842	0.4500	0.069
H(4c)	0.5813	0.2528	0.3099	0.096
H(4d)	0.4591	0.2364	0.3912	0.096
H(4e)	0.4898	0.1508	0.3213	0.096
H(5a)	0.6764	-0.2237	0.0634	0.060
H(5b)	0.6746	-0.2248	-0.0370	0.060
H(5c)	0.4801	-0.1114	0.0795	0.087
H(5d)	0.4565	-0.2258	0.0504	0.087
H(5e)	0.4763	-0.1225	-0.0196	0.087
H(6a)	0.8246	-0.1538	-0.1695	0.066
H(6b)	0.8770	-0.0397	-0.2153	0.066
H(6c)	0.6062	-0.0984	-0.1697	0.102
H(6d)	0.6925	-0.0761	-0.2664	0.102
H(6e)	0.6520	0.0206	-0.2072	0.102

Table S6. Continued

atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> (eq)
H(7a)	1.2196	0.2674	-0.3016	0.079
H(7b)	1.0970	0.1896	-0.2873	0.079
H(7c)	0.9599	0.3397	-0.3180	0.094
H(7d)	1.0800	0.3232	-0.3985	0.094
H(7e)	1.0861	0.4144	-0.3365	0.094
H(8a)	1.1936	0.5164	-0.1568	0.069
H(8b)	1.2149	0.4688	-0.2484	0.069
H(8c)	1.3694	0.3904	-0.1259	0.103
H(8d)	1.4152	0.4903	-0.1963	0.103
H(8e)	1.4011	0.3751	-0.2273	0.103
H(1)	0.6211	0.1779	0.6228	0.092
H(2)	0.7045	0.2848	0.5874	0.092

^a*U*(eq) is defined as one third of the trace of the orthogonalized \mathbf{U}_{ij} tensor the estimated standard deviations of the least significant digits are given in parentheses.