Direct functionalization of (un)protected tetrahydroisoquinolines and isochroman under iron and copper catalysis – two metals, two mechanisms

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1. General

Unless otherwise noted, chemicals were purchased from commercial suppliers and used without further purification. Flash column chromatography was performed on silica gel (40-63µm). For thin layer chromatography (TLC) aluminum coated silica gel was used and signals were visualized with UV light (254nm).

1.1. Caution

Mixing a metal salt and peroxide can cause explosion. See: Jones, A. K.; Wilson, T. E.; Nikam, S. S. In Encyclopedia of Reagents for Organic Synthesis, Paquette, L. A. Ed.; John Wiley & Sons, Inc. 1995, 2, 880.

1.2. Instrumentation

GC-MS runs were performed using a standard capillary column ($30m \times 0.32 mm$ ID), applying the following standardized temperature profile: 2 minutes at 80° C, 10° C/min until 280° C, 15 minutes at 280° C. All samples subjected to HR-MS were analyzed by LC-IT-TOF-MS in only positive ion detection mode upon recording of MS and MS/MS spectra. For the evaluation in the following, only positive ionization spectra were used (where the quasi-molecular ion is the one of [M+H]+), and further data or information were not taken into consideration. Microwave reactions were performed on a BIOTAGE InitiatorTM sixty microwave unit. Melting points were determined using a Kofler-type hot stage microscope and are uncorrected.¹H-NMR and ¹³C-NMR spectra were recorded either on 200 MHz or 400 MHz spectrometer. Chemical shifts are reported as ppm downfield from TMS (tetramethylsilane) as internal standard with multiplicity, number of protons, allocation, and coupling constant(s) in Hertz.



2. Copies of ¹H and ¹³C-NMR spectra of compounds 3a-3s, 5h-5t, 8a-8d, 10a-10f, and 11a-11d















































































































































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