Molecular Dynamics Investigation of Cl[−] **and Water Transport through a**

Eukaryotic CLC Transporter

Mary Hongying Cheng and Rob D. Coalson

Department of Chemistry, University of Pittsburgh

Pittsburgh, PA 15260

SI Methods

Simulation systems

Fig. S1: Initial configurations for MD simulation A) System 3; B) System 4; C) System 5; D) System 6; E) System 7; and F) System 8. Y515 and S165 are colored in green and yellow, respectively. Protonated and deprotonated E210 are colored in pink and purple, respectively. Binding Cl[−] ions are colored in red. The channel z axis runs from extracellular to intracellular side. In all plots, extracellular region is at top of plot, and intracellular region is at the bottom.

MD simulation protocols

Simulations were carried out using NAMD (1) following standard MD procedure. Simulation protocols included periodic boundary conditions, water wrapping, hydrogen atoms constrained via SHAKE, and evaluation of long-range electrostatic forces via the Particle Mesh Ewald (PME) algorithm (2). The bonded interactions and the short-range non-bonded interactions were calculated at every time-step (2 fs), and electrostatic interactions were calculated every two time-steps (4 fs). The cutoff distance for non-bonded interactions was 12 Å. A smoothing function was employed for the van der Waals interactions at a distance of 10 Å. The pair-list of the nonbonded interactions was calculated every 20 time-steps with a pair-list distance of 13.5Å.

Adaptive biasing force calculations

The single ion potential of mean force (PMF) profiles for transporting a Cl[−] ion from S_{ext} to S_{cen} and S_{int} were calculated along the transport path using the adaptive biasing force (ABF) (3, 4) method implemented in NAMD (1). The ABF calculations of the PMF were carried out in different windows along the ion transport path. The length of each ABF window was 2.0 Å. The initial configurations for each ABF window were either taken from snapshots of our control MD simulations that have the target ion positioned within the window, or obtained from a 100ps equilibration simulation that had the target ion positioned within the window. ABF calculations of the PMF were continued for 6~8 ns within each window. The average force acting on the target ion was accumulated in 0.1 Å sized bins. The boundary force constant was set to be 20 Kcal/mol. The biasing force was applied only after the accumulation of 1000 samples (successive simulation time steps) in individual bins.

SII Results and Discussion

Fig. S2: (A) Orientation of the GLU_{ex} gate E210 in the x-ray crystal structure. The GLU_{ex} gate is characterized by two dihedral angles χ 1=49.6° (defined by N CA CB and CG atoms) and γ 2=90.8° (defined by CA CB CG and CD atoms; B) typical orientation of γ 1=80±20° and γ 2= -65±20° in subunit A of System A_dB_d ; C) typical orientation of χ 1=50±20° and χ 2= 80±20° in subunit B of System A_dB_d ; (D) typical orientation of χ 1= 180° and χ 2= 180° for the protonated E210 in System A_nB_n . For E210, blue, cyan and red represent nitrogen, carbon and oxygen atoms, respectively. Hydrogen atoms are not shown in the figures.

Fig. S3: Comparison of CmCLC after 80 ns MD simulations with (A) protonated E210 (A_pB_p system); and (B) deprotonated E210 (A_dB_d system). Channel orientation is the same as in Fig. 1. Red and orange balls represent Cl[−] ions in the crystal structure and coming from the internal solution, respectively. Protonated and deprotonated E210 are colored in pink and purple, respectively. In our simulations, Cl[−] ions which originally occupied the S_{int} sites in the X-ray structure escaped to the internal solution within 40 ns in both A_dB_d and A_pB_p simulation systems. After an additional 30 ns of MD simulation, other Cl[−] ions migrated from the intracellular solution and occupied the S_{int} sites again. Extracellular region is at top of plot, and intracellular region is at the bottom.

Fig. S4: A) Waters within 12 Å of the GLU_{ex} gate (E210) at 80 ns in the A_pB_p system (protonated E210) without external potential. A continuous water wire starting from the intracellular K171 and extending to the GLU_{ex} gate quickly formed and remained stable throughout the simulations. This continuous water path has been previously proposed as the proton transport pathway (5-7). The external binding site S_{ext} is highly hydrophobic, F256 and V431, or P432 and V431 (not shown) can block the water path from the extracellular solution to the S_{ext} site, leading to a dehydrated Cl[−] binding to the S_{ext} site for the majority of the 80 ns of MD simulation; B) A continuous water path from the extracellular solution to the GLU_{ex} gate was formed within 4ns once an external potential of -120 mV was applied. Waters are shown in CPK format.

Fig. S5: MD snapshots of A_pB_p system at A) $t = 0$ ns, B) $t = 10$ ns, C) $t = 50$ ns and D) $t = 70$ ns. Residues S165, protonated E210 and Y515 are shown in yellow, pink and green, respectively. Within 50 ns the GLU_{ex} gate swung from the S_{cen} site to the extracellular solution and fully opened for Cl transport (C and D).

Fig. S6: Comparison of waters within 12 Å of the deprotonated GLU_{ex} gate (E210) in A) A_dB_d system; B) System 4; and C) Y265A mutant channel. The deprotonated GLUex gate is shown in purple. In System 4, the GLU_{ex} gate was deprotonated after a simulation of 80 ns in the A_pB_p system. Y265A mutation was performed on the A_dB_d system.

Fig. S7: Comparison of the radial distribution function (RDF) of the permeant Cl[−] ion and oxygen atoms of water in bulk water solution (blue line) and inside the permeation path lined by Y515 and S165 (red line). More than 50% dehydration of the first hydration shell was observed when the Cl[−] was transferred from bulk solution to the location indicated in the inset figure, which shows the permeant Cl[−] with its first hydration shell in the constriction formed by Y515 and S165.

Fig. S8: Time evolution of dihedral angles χ 1 (black) and χ 2 (gray) of the GLU_{ex} gate in (A) subunit A and (B) subunit B of System 3. Once initially protonated E210 was deprotonated, dihedral angles of χ 1 and χ 2 changed from initial values $\pm 180^\circ \pm 20^\circ$ to the values $70^\circ \pm 20^\circ$ and $-65^\circ \pm 15^\circ$. The orientation change of the GLU_{ex} gate facilitated the translocation of Cl[−] from the S_{cen} to the S_{int} site.

Fig. S9: Binding of two Cl[−] ions (cyan spheres) to the S_{cen} site when E210 is protonated. One Cl[−] ion binds near the backbone of the GLU_{ex} gate and the other either binds to A) a site lined by the hydroxyl groups from Y515 and E210 (termed site A); or B) a site lined by the hydroxyl groups from Y515 and S165 (termed site B). Extracellular region is at top of plot, and intracellular region is at the bottom. Black arrows illustrate putative Cl[−] paths found in our simulations.

Supporting References:

- 1. Phillips, J. C., R. Braun, W. Wang, J. Gumbart, E. Tajkhorshid, E. Villa, C. Chipot, R. D. Skeel, L. Kale, and K. Schulten. 2005. Scalable molecular dynamics with NAMD. *J Comput Chem* 26:1781-1802.
- 2. Darden, T., D. York, and L. Pedersen. 1993. Particle Mesh Ewald an N.Log(N) Method for Ewald Sums in Large Systems. *J Chem Phys* 98:10089-10092.
- 3. Darve, E., D. Rodriguez-Gomez, and A. Pohorille. 2008. Adaptive biasing force method for scalar and vector free energy calculations. *J Chem Phys* 128:144120.
- 4. Chipot, C., and J. Hénin. 2005. Exploring the free-energy landscape of a short peptide using an average force. *J Chem Phys* 123: 244906.
- 5. Wang, D., and G. A. Voth. 2009. Proton transport pathway in the CIC Cl⁻/H⁺ antiporter. *Biophys J* 97:121-131.
- 6. Kieseritzky, G., and E. W. Knapp. 2011. Charge transport in the ClC-type chlorideproton anti-porter from Escherichia coli. *J Biol Chem* 286:2976-2986.
- 7. Kuang, Z. F., U. Mahankali, and T. L. Beck. 2007. Proton pathways and H⁺/Cl⁻ stoichiometry in bacterial chloride transporters. *Proteins: Struct., Func., and Bioinform* 68:26-33.