

# Generating Singlet Oxygen Bubbles: A New Mechanism for Gas-Liquid Oxidations in Water

Dorota Bartusik,<sup>1</sup> David Aebisher,<sup>1</sup> BiBi Ghafari,<sup>2</sup> Alan M. Lyons\*<sup>2</sup> and Alexander Greer\*<sup>1</sup>

<sup>1</sup>Department of Chemistry, Brooklyn College, City University of New York, Brooklyn, New York 11210

<sup>2</sup>Department of Chemistry, College of Staten Island, City University of New York, Staten Island, New York 10314

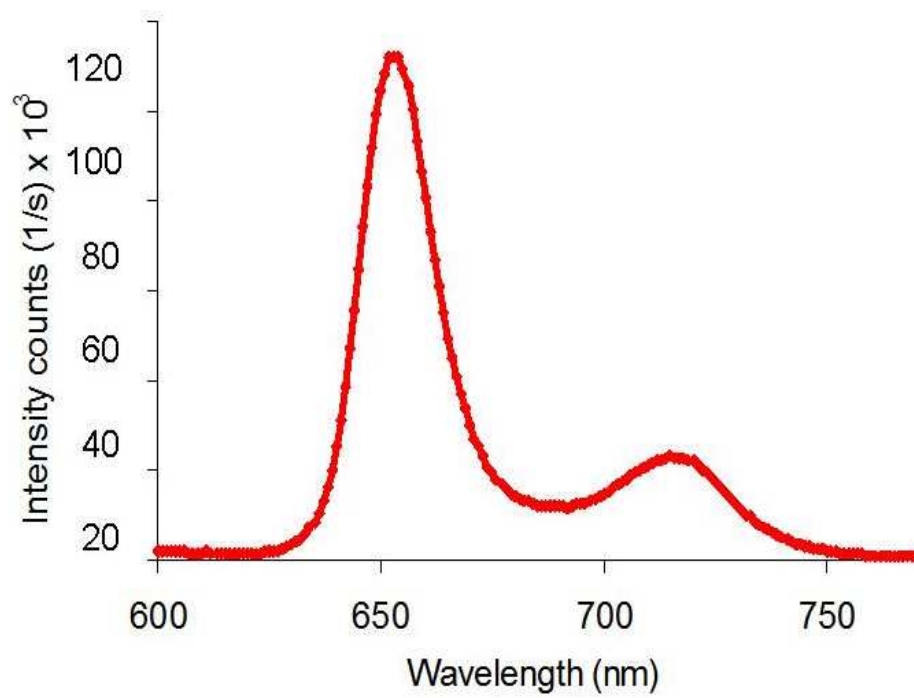
E-mail: alan.lyons@csi.cuny.edu; agreer@brooklyn.cuny.edu

## *Supporting Information*

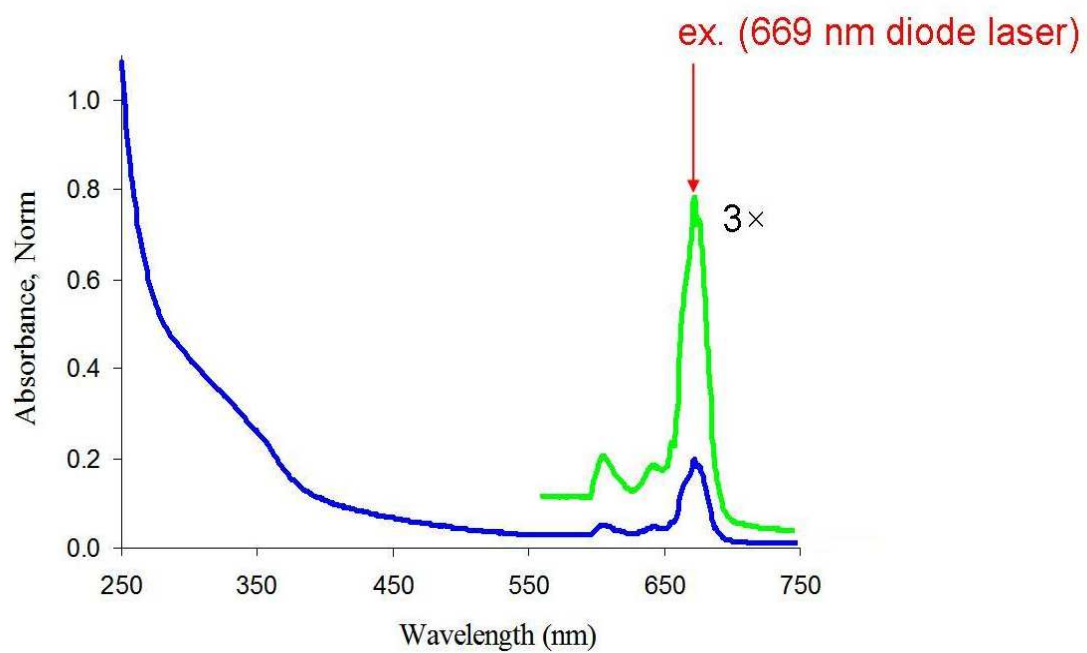
### **Table of Contents**

Page

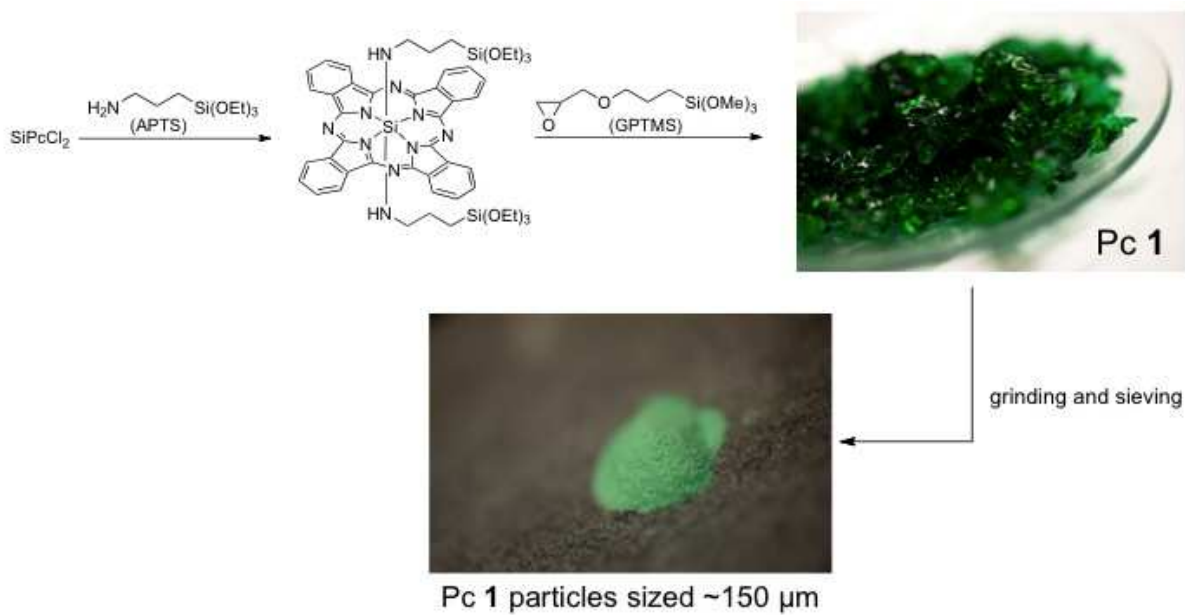
1	Table of Contents
2	Figure S1: Fluorescence spectrum of a thin sheet of Pc 1.
3	Figure S2: UV-VIS spectrum of the <sup>1</sup> O <sub>2</sub> sensitizing glass Pc 1.
4	Figure S3: Hybrid <sup>1</sup> O <sub>2</sub> sensitizing (phthalocyanine entrapped sol-gel) glass.
5	Figure S4: Loading of sensitizer particles into the chamber of a device.
6	Figure S5: Images of the SMA receptacle with oxygen inlet.
7	Figure S6: Photomicrographs of the polyethylene membranes.
8	Figure S7: Luminescence of singlet oxygen in D <sub>2</sub> O and in air.
9	Figure S8: Image of the loss of singlet oxygen in bubbles that reach the air interface.
10	Table S1: Device dimensions and membrane characteristics.
11	Table S2: Calculated particle surface area based on loading.



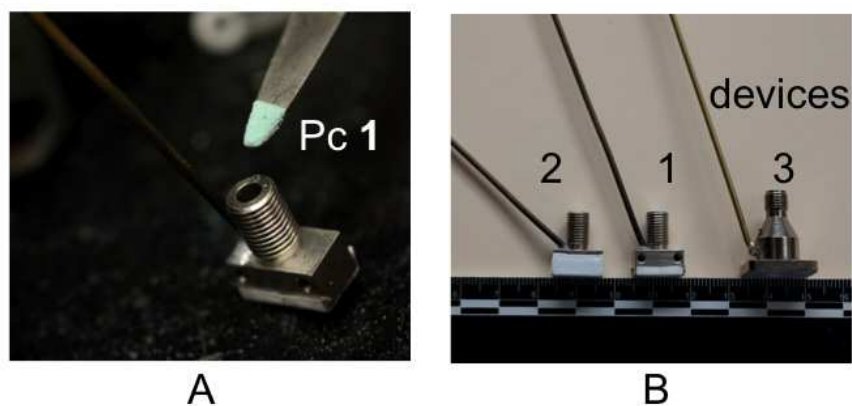
**Figure S1.** Fluorescence spectrum ( $\lambda_{\text{ex}} = 250$  up to 400 nm) of a sheet of Pc **1**, ~0.5 mm in thickness.



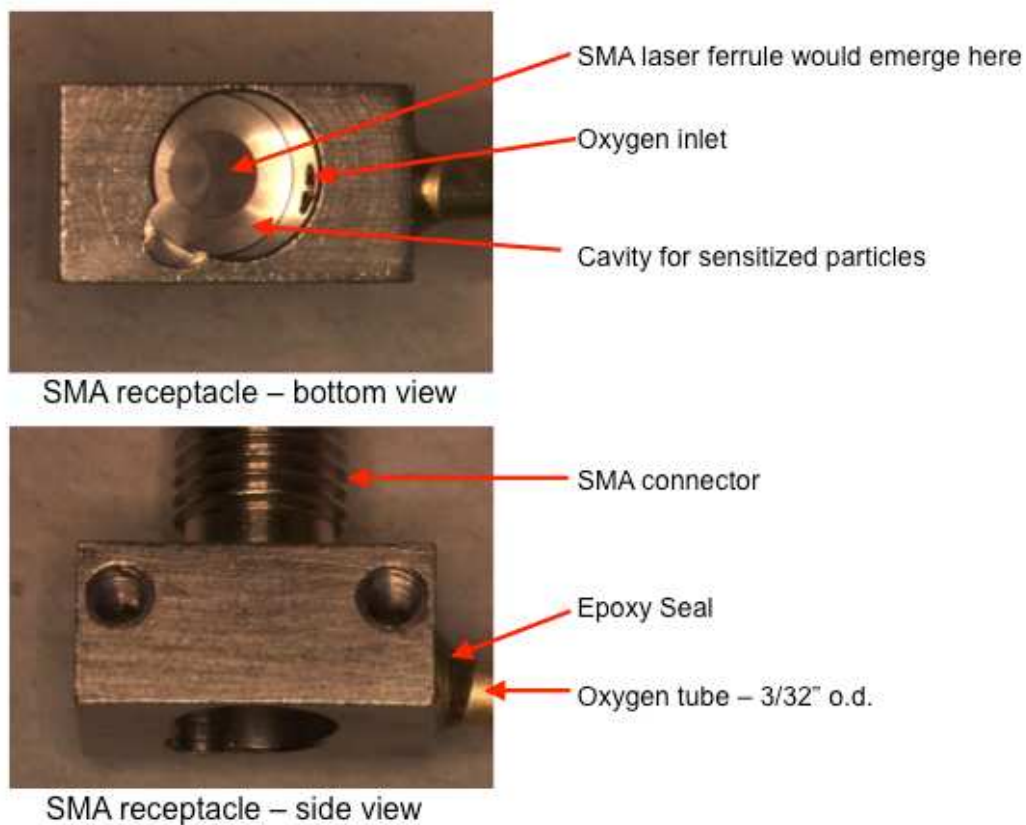
**Figure S2.** UV-VIS spectrum of the  $^1\text{O}_2$  sensitizing glass Pc 1. The red arrow shows where the 669 nm laser line of the diode laser overlaps the Q-band.



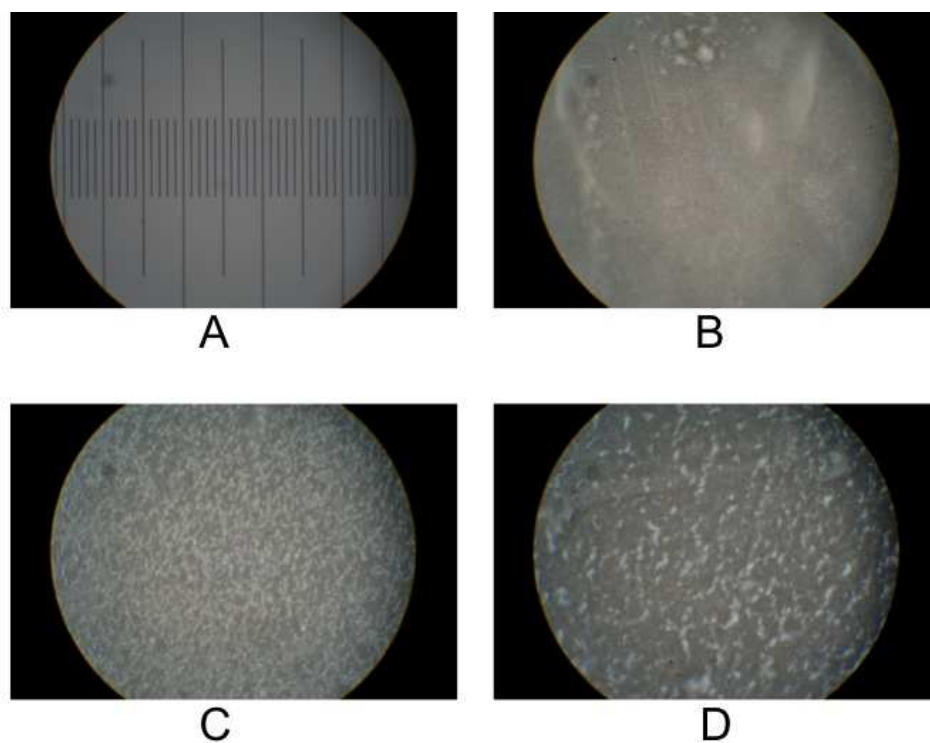
**Figure S3.** Schematic of the synthetic approach used as well as photos showing the hybrid  $^1\text{O}_2$  sensitizing (phthalocyanine entrapped sol-gel) glass, and the fine powder achieved after grinding.



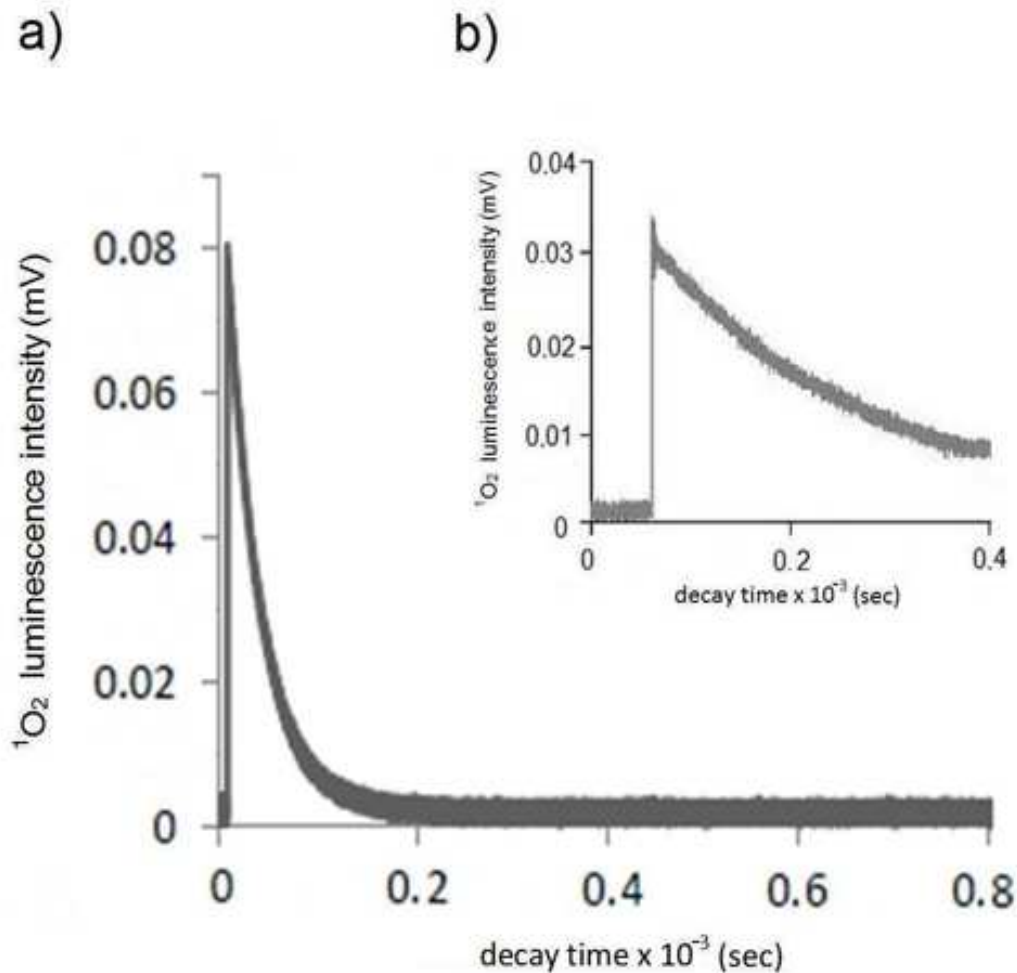
**Figure S4.** These photos show (A) the loading of  $\sim 150\ \mu\text{m}$  Pc 1 sensitizer particles into the chamber of device 2 via a spatula, and (B) the three devices (without optical fiber) placed above a centimeter-scaled ruler.



**Figure S5.** Images of the SMA receptacle with oxygen inlet.

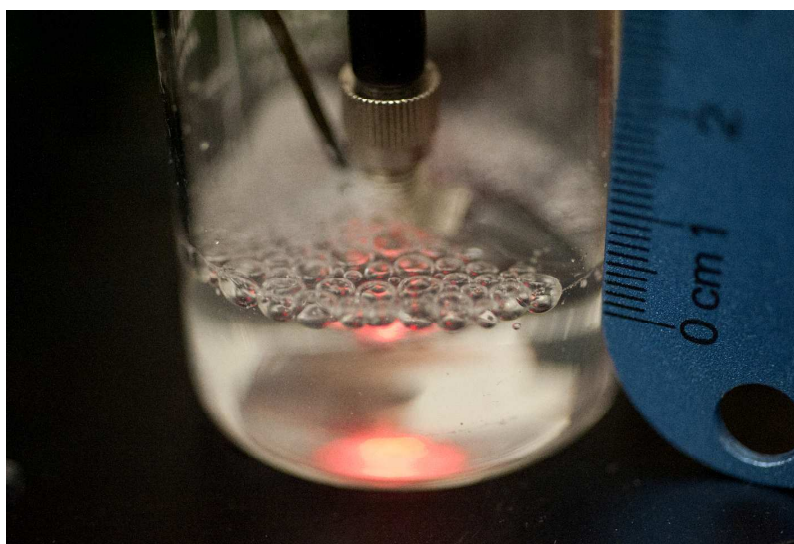


**Figure S6.** Photomicrographs of the polyethylene membranes: (A) micrometer calibration (1 unit =0.007368 mm =7.37  $\mu\text{m}$ ); (B) 0.05  $\mu\text{m}$  pore size membrane; (C) 0.22  $\mu\text{m}$  pore size membrane; and (D) 0.44  $\mu\text{m}$  pore size membrane. The individual pores were too small to resolve and see under the low-power magnification.



**Figure S7.** Luminescence from singlet oxygen at 1270 nm were observed with 355-nm pulsed irradiation through 35 mg of sensitizer particles loaded into device 1 with flowing oxygen (60 mL/min). First-order decay kinetics were observed and fitted to the equation  $[\text{luminescence}_{1270}(t) = A \times (\exp^{-t/\tau})]$ , where  $1/k_{\text{obs}} = \tau(^1\text{O}_2)$  lifetime. The lifetime of singlet oxygen in (a) 3 mL  $\text{D}_2\text{O}$  was  $60 \pm 3 \mu\text{s}$  (average of 6 experiments), and (b) in air was  $\sim 1.1$  ms (estimated from 3 experiments).





**Figure S8.** Loss of singlet oxygen in bubbles that reach the air interface from device 2 loaded with 35 mg sensitizer particles. The O<sub>2</sub> flow rate was 60 mL/min.

**Table S1. Device Dimensions and Membrane Characteristics**

Device	Chamber Diameter (mm)	Chamber Height (mm)	Membrane Pore Size ( $\mu\text{m}$ )	Membrane Thickness ( $\mu\text{m}$ )	Capillary Pressure (PSI)
1	5.7	5.3	0.05	70	108
2	5.7	5.3	0.22	90	25
3	10	10	0.44	150	12

**Table S2. Calculated Particle Surface Area Based on Loading**

Quantity of Pc 1 loaded into devices (mg)	Total surface area of sensitizer particles 1 (mm <sup>2</sup> /mg)	Total number of sensitizer particles 1
0	0	0
1	30	420
3	90	1260
10	300	4200
35	1050	14,700
50	1500	21,000
75	2250	31,500

<sup>a</sup> 150±30 μm sensitizer particles.