

1 **Supporting Information**

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4 **Impacts of Hematite Nanoparticle Exposure on Biomechanical,**
5 **Adhesive, and Surface Electrical Properties of *E. coli* Cells**

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- 25 **This supporting information includes the following sections:**
- 26 S1. Justification of the loading force magnitude in the AFM force measurement
- 27 S2. SEM analysis (Fig. S1)
- 28 S3. Young's modulus calculation using Hertz model
- 29 S4. Electrophoretic mobility (EPM) measurement and PSD changes over different
- 30 adsorption time (Table S1 and Fig. S2)
- 31 S5. Histogram of surface potential distribution of hematite NPs (Fig. S3)
- 32 S6. More images of *E. coli* cells using KPFM (Fig. S4)
- 33 S7. Correlation between the adsorbed mass of hematite NPs and surface potentials of *E.*
- 34 *coli* cells (Fig. S5)
- 35 References
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37 **S1. Justification of the loading force magnitude in the AFM force measurement**

38 A maximum loading force of 4 nN was chose to push the tip against the cell surface
39 to measure the surface hardness. We intend to measure the indentation for the
40 tip-membrane interactions instead of measuring the interaction forces on the outside layer
41 of the cell (e.g., surface appendages–pili or surface brushes). This is partly because the
42 physical disruption from hematite exposure likely altered the mechanical properties of the
43 cell membrane (2, 6, 7, 9, 11), which is beneath the tightly packed lipopolysaccharide
44 (LPS) molecules and mainly determines the mechanical response. More importantly,
45 through changing the adhesiveness or mechanical properties, the sorption of hematite NPs
46 could disrupt the integrity of the cytoplasm membrane structure (4). Thus, quantifying
47 the membrane property is more relevant in this study. Since Velegol and Logan indicated
48 that in the nonlinear regime the tip contacted the outer membrane but had yet to encounter
49 the stiff peptidoglycan layer (12), we applied a relatively high loading force (~nN) in an
50 effort to engage the tip with the membrane membrane regions, which was also suggested
51 previously (13). As seen from the compliance curve in Fig. 3, the 4-nN loading force led
52 to a linear regime that indicates the tip contact with the inner membrane.

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54 S2. SEM analysis

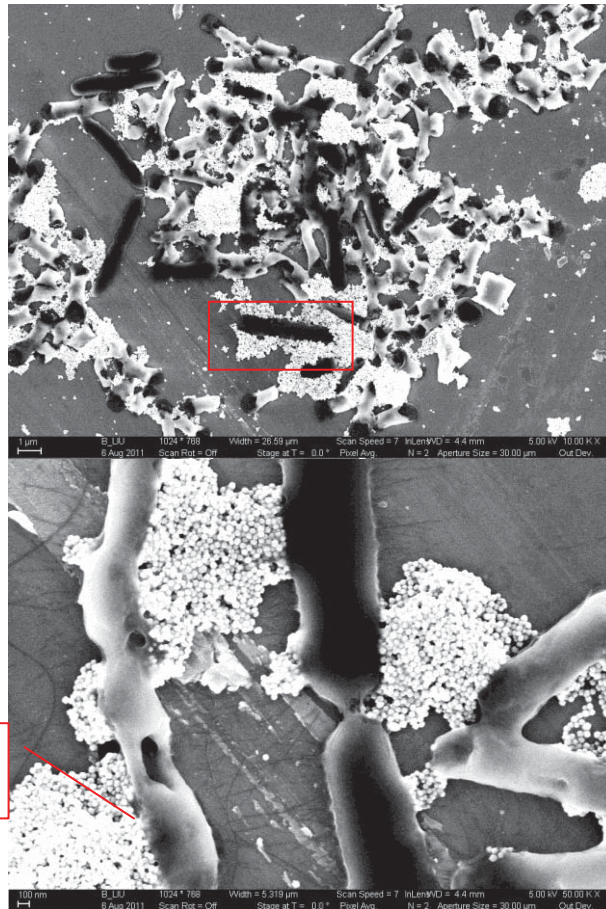
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66 Fig. S1 SEM images of *E. coli* cells with adsorbed hematite NPs. The enlarge area of
67 the red box is also shown in Fig. 1c.

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70 **S3. Young's modulus calculation using Hertz model**

71 The hardness of the cells can be evaluated by the Hertz model as expressed below:

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$$F = \frac{4\sqrt{R_c}}{3} \frac{E}{1-\nu^2} \delta^{3/2} \quad (\text{S1})$$

73 where F is the loading force exerted by the tip on the surface (nN), R_c is the radius of tip
74 curvature (approximately 20 nm as provided by the manufacture), E is the reduced
75 Young's module of *E. coli* cells (Pa), ν is the Poisson's ratio (usually set to 0.5) (3, 10),
76 and δ is the indentation (nm). The calculated Young's module of *E. coli* cells was in the
77 range of 52.7 kPa to 0.38 MPa, which is lower than the reported values of ~25 MPa (1,
78 14). The discrepancy could be ascribed to the uncertainties in the radius of the tip
79 (usually varies with different batches of manufacturing) and in the measurement
80 indentation.

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82 **S4. Electrophoretic mobility (EPM) measurement and PSD changes over different**
83 **sorption time**

84 Because the von Smoluchowski equation is valid for particles with uniform surface
85 charge density, which is not true for bacteria, and moreover, the exact location of the
86 shear plane for *E. coli* cells is hard to determine. Therefore, the EMP measurement is
87 directly presented here in addition to zeta potentials and again, the EMP values of *E. coli*
88 cells were not significantly different over different exposure time. A detailed
89 interpretation of the EPM changes is difficult and beyond the scope of this study.
90 However, it is clear that the gradual increase in EPM is probably caused by the increase
91 of the global electrostatic charge due to the progressive coverage of NPs (5).

92 Table S1 EPM measurements

	Hematite NPs in PBS	<i>E. coli</i> cells in PBS	1 min exposure	5 min exposure	15 min exposure	45 min exposure
EPM, $\mu\text{m/s}/(\text{V/cm})$	-3.78 \pm 2.37	-0.32 \pm 0.00	-2.16 \pm 1.28	-1.73 \pm 1.23	-2.27 \pm 1.3	-1.97 \pm 1.23

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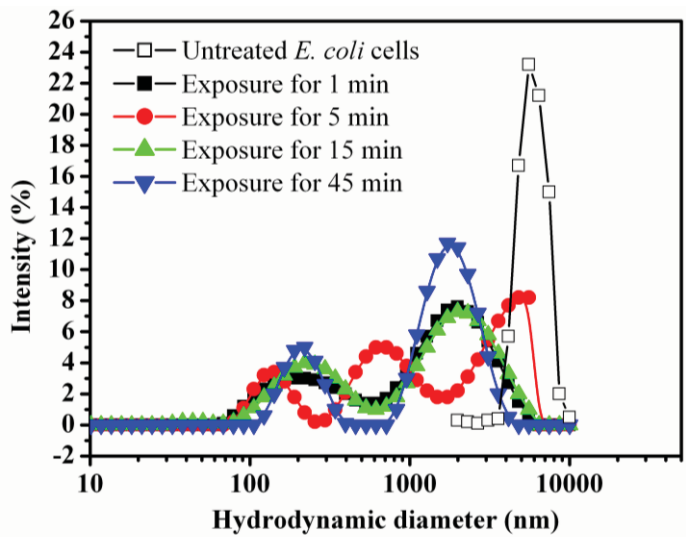


Fig. S2 Modality of PSD of the bacterial suspension at different exposure or adsorption times.

114 **S5. Histogram of surface potential distribution of hematite NPs.**

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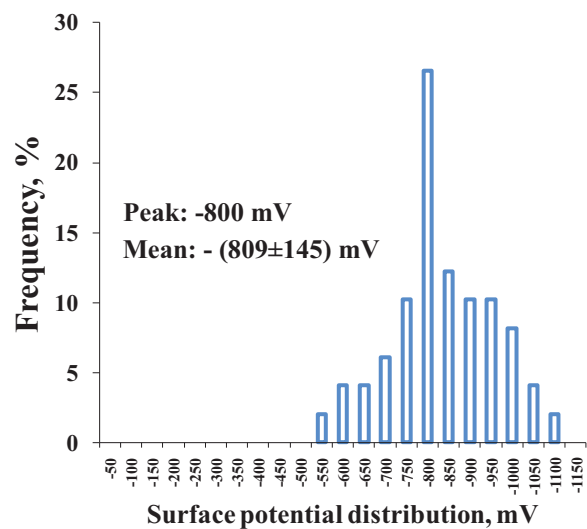
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Figure S3. Histogram of the surface potential distribution measured from 50 NPs randomly chosen from Figure 5b.

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131 S6. More images of *E. coli* cells using KPFM

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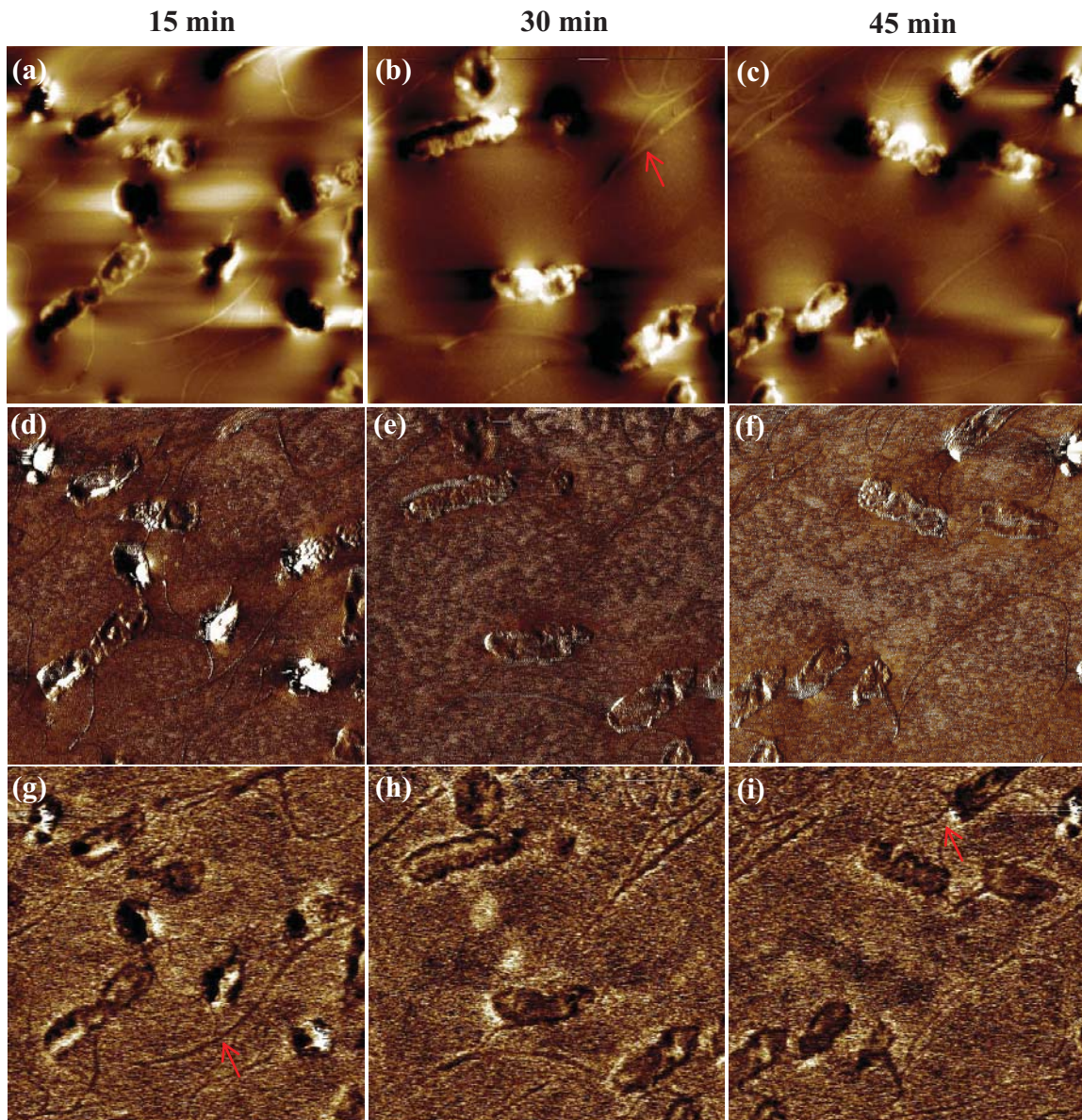
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147 Figure S4. AFM images of *E. coli* cells. (a)-(c) are topography, (d)-(f) are phase, and

148 (g)-(h) are surface potential images. The left column images were taken after drying for

149 approximately 15 min; the middle column, 30 min; and the right column, 45 min. The

150 scan area of each image is $5 \mu\text{m} \times 5 \mu\text{m}$. The red arrows point at a tail-like flagella.

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152 **S7. Correlation between adsorbed mass of hematite NPs and surface potentials of *E.***
153 ***coli* cells**

154 As discussed previously, the surface potential decrease of *E. coli* cells should be
155 caused by the adsorption of hematite NPs. Detailed adsorption kinetics of hematite NPs
156 on *E. coli* cells has been investigated previously (15). Here we further used KPFM to
157 determine the surface potentials of *E. coli* cells by randomly selecting 10 positions in the
158 cell surface at different exposure time and then obtained the average surface potentials.
159 Figure S5 plotted the average surface potentials at different exposure time versus the
160 estimated corresponding mass of hematite NPs per *E. coli* surface area. The surface
161 area that was available for hematite adsorption was estimated as follows: a single *E. coli*
162 cell has 2.5×10^{-13} g volatile suspended solid (VSS) and a volume of 6×10^{-12} m³ (8). In
163 each test tube, approximately 0.1 mg/ml *E. coli* cells were dispersed in a total volume of
164 10 ml suspension; thus, approximately 1 mg *E. coli* cells were present in each test tube,
165 corresponding to 4.0×10^9 cells and 0.024 m² surface area that is available for the
166 adsorption of hematite NPs. With the quantification of hematite loss in the aqueous
167 phase of the 10-ml test tube, we could determine the adsorbed mass of hematite divided
168 by the total surface area (0.024 m²). Apparently, as the hematite NP adsorbed on the
169 cell surfaces, the average surface potential dropped significantly in a non-linear fashion.

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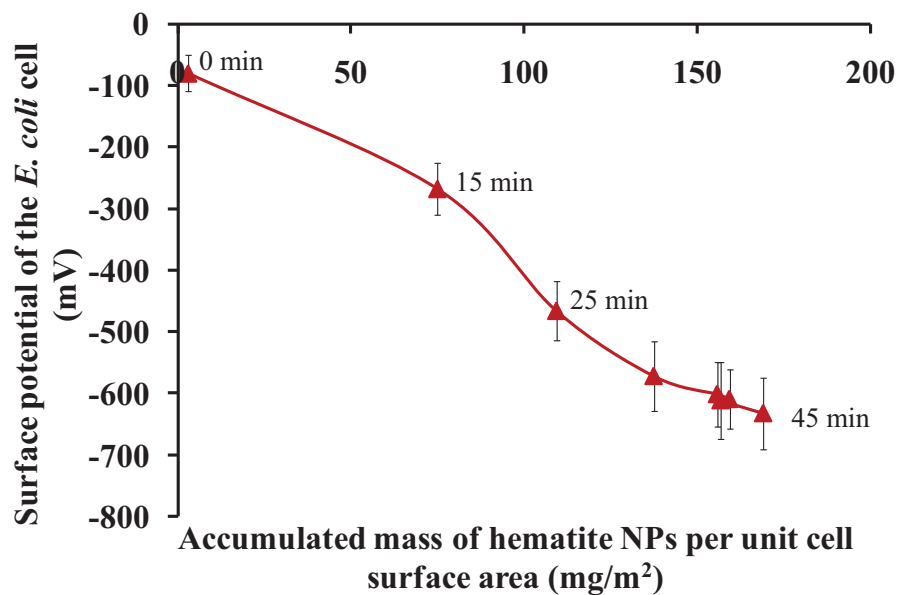


Figure S5. The accumulated mass of hematite NPs on the unit surface area of the cell versus the surface potentials of the *E. coli* cells at different exposure times. All data points are averages from the surface potential measurements on different positions of 15 hematite-treated cells, and error bars represent standard deviation.

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