

Supporting Information

The Importance of Stereochemically Active Lone Pairs For Influencing Pb(II) and As(III) Protein Binding

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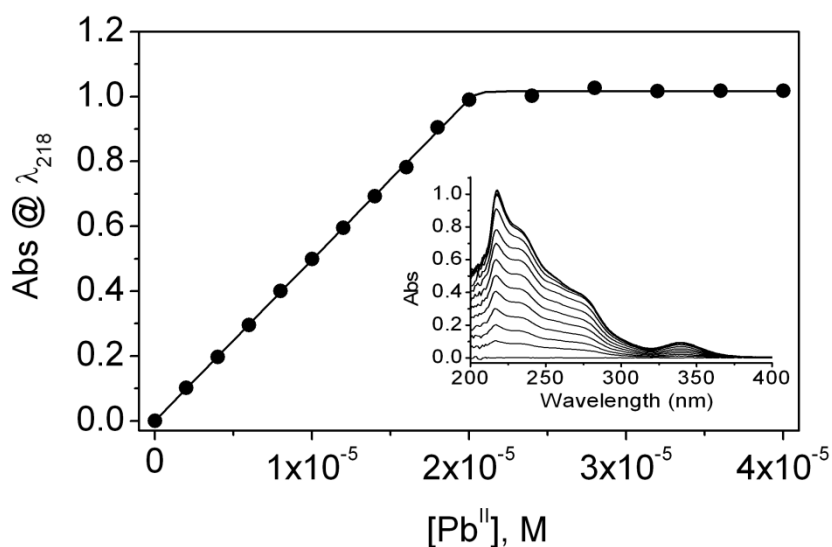


Figure S1. UV/Vis titration of Pb(NO₃)₂ into TRIL12AL16C (60 μM, 100 mM TrisHCl buffer, pH 8.0). The spectral change at 218 nm was plotted as a function of Pb(NO₃)₂ concentration and the plot showed that 1 equivalent of Pb(II) bind per peptide trimer. The solid line represents a nonlinear, least square fitting curve generated using 1:1 binding model by ORIGIN with $K_a = 5.6(\pm 1) \times 10^8 \text{ M}^{-1}$. The inset depicts the UV/Vis spectral change from each addition of 0.1 equivalent of Pb(II).

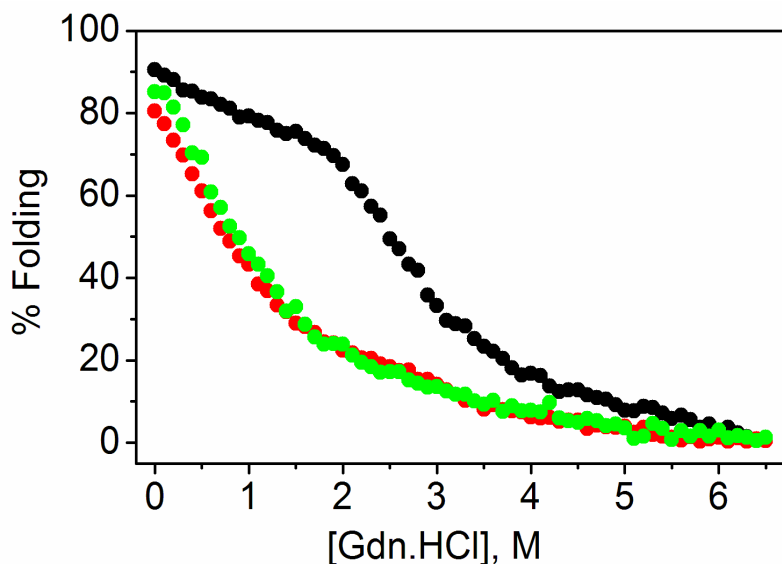


Figure S2. Guanidinium hydrochloride (Gdn.HCl) denaturation curves showing the stability of apo-peptide: **TRIL2WL16C** (black), **TRIL12AL16C** (red), and **TRIL2WL12AL16C** (green). The data were collected at 10 μ M peptide monomer in TrisHCl buffer at pH 8.0. $[\theta]_{222}$ is converted into % folded and plotted against Gdn.HCl concentration.

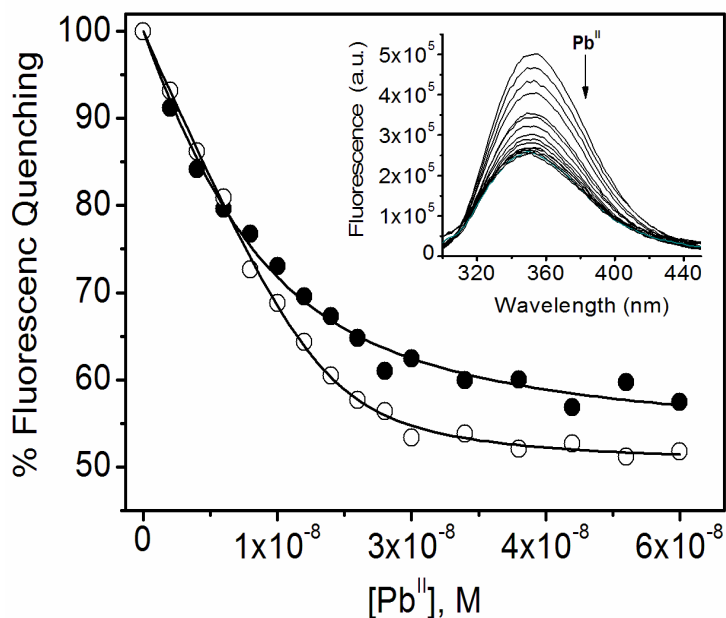


Figure S3. Change in fluorescence intensity upon binding of Pb(II) to the peptide **TRIL2WL16C** (filled circle) and **TRIL2WL12AL16C** (open circle) (30 nM trimer). The Trp was excited at 280 nm and the fluorescence quenching was traced at 354 nm. The curves represent the best fit to Equation 1 by non linear least squares analysis using ORIGIN software. In inset fluorescence quenching spectra were shown for the addition of Pb(II) to **TRIL2WL12AL16C**.

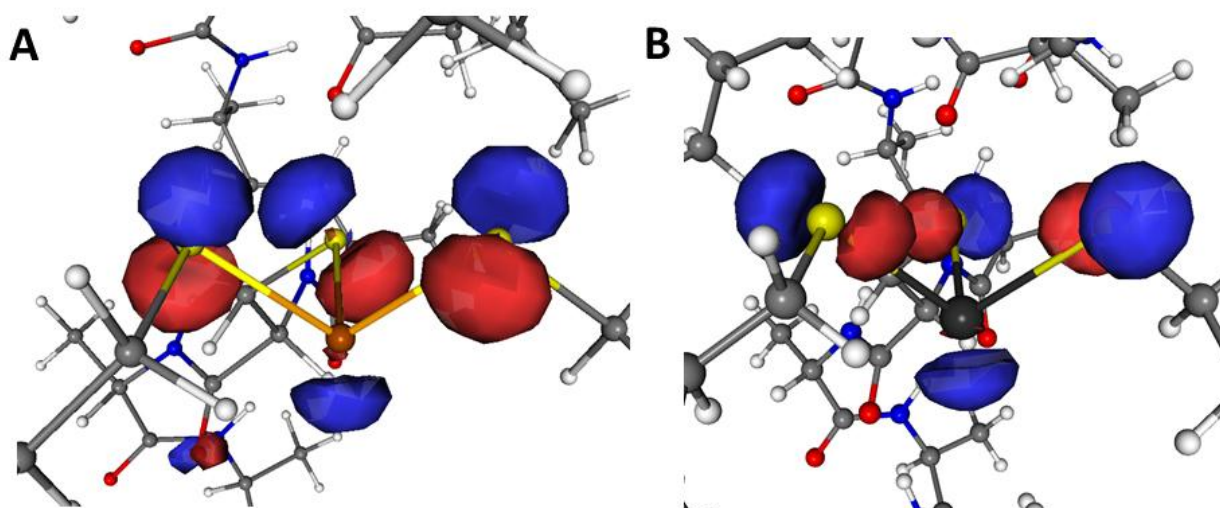


Figure S4. Mapping of isoenergy surfaces associated with the appropriate s and p orbitals for A) As(ACA)_3 and B) Pb(ACA)_3 . The eigenvalue of the displayed orbital is -0.205285 Hartree for the As system, while in the case of the Pb system the associated eigenvalue is -0.09665 Hartree.

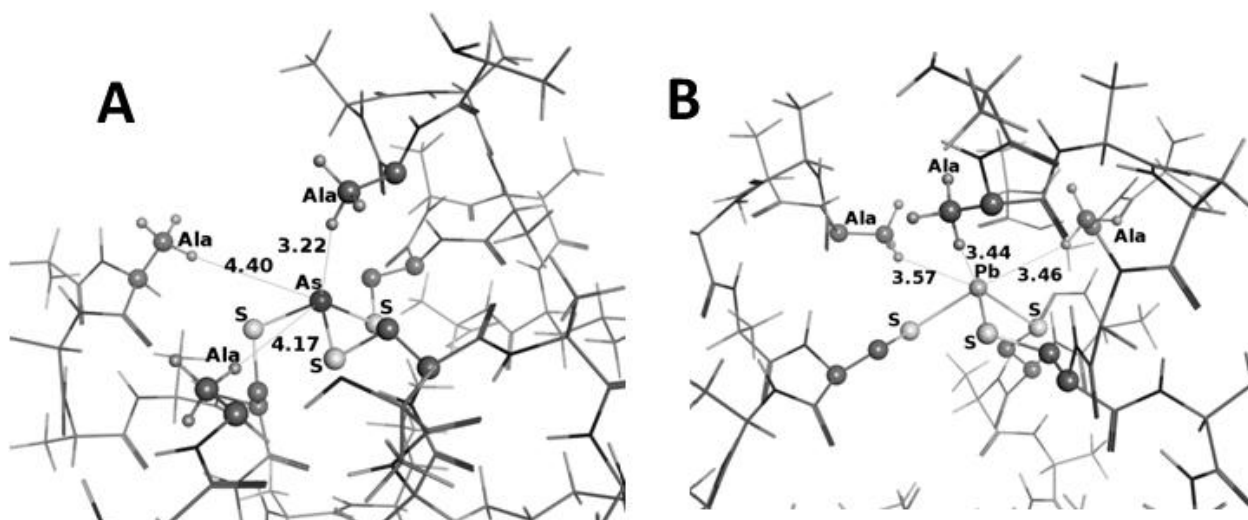


Figure S5. Closest H atom to M ion distances given in Angstrom. A) As(III)(ACA)_3 and B) Pb(II)(ACA)_3 .