#### Concise Total Synthesis and Stereochemical Revision of (+)-Naseseazines A and B. Regioselective Arylative Dimerization of Diketopiperazine Alkaloids.

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**General Procedures.** All reactions were performed in oven-dried or flame-dried round-bottom flasks. The flasks were fitted with rubber septa and reactions were conducted under a positive pressure of argon. Gas-tight syringes with stainless steel needles or cannulae were used to transfer air- and moisture-sensitive liquids. Where necessary (so noted), reactions were performed in Schlenk tubes fitted with a PTFE stopcock. Flash column chromatography was performed as described by Still et al. using granular silica gel (60 Å pore size, 40–63  $\mu$ m, 4–6% H<sub>2</sub>O content, Zeochem).<sup>1</sup> Analytical thin layer chromatography (TLC) was performed using glass plates pre-coated with 0.25 mm 230–400 mesh silica gel impregnated with a fluorescent indicator (254 nm). TLC plates were visualized by exposure to short wave ultraviolet light (254 nm) and an aqueous solution of ceric ammonium molybdate (CAM) followed by heating on a hot plate (~250 °C). Organic solutions were concentrated at 29–30 °C on rotary evaporators capable of achieving a minimum pressure of ~2 torr.

**Materials.** Commercial reagents and solvents were used as received with the following exceptions: dichloromethane, acetonitrile, tetrahydrofuran, methanol, pyridine, toluene, and triethylamine were purchased from J.T. Baker (Cycletainer<sup>TM</sup>) and were purified by the method of Grubbs et al. under positive argon pressure.<sup>2</sup> Nitroethane was distilled over calcium hydride and stored over 4 Å molecular sieves. 1-Ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride was purchased from Oakwood Products, Inc.; *N*-hydroxybenzotriazole was purchased from Aroz Technologies, LLC; 6-bromo-3-formylindole was purchased from Frontier Scientific, Inc.; (+)-1,2-bis((2*S*,5*S*)-2,5-diethylphospholano)benzene(1,5-cyclooctadiene)rhodium(I) trifluoromethanesulfonate was purchased from Strem Chemicals, Inc.; and (±)-Boc- $\alpha$ -phosphonoglycine trimethyl ester as well as all amino acid derivatives were purchased from Chem-Impex International, Inc. All other solvents and chemicals were purchased from Sigma–Aldrich.

Instrumentation. Proton nuclear magnetic resonance (<sup>1</sup>H NMR) spectra were recorded with Varian inverse probe INOVA-500 and Varian INOVA-500 spectrometers, are reported in parts per million on the  $\delta$  scale, and are referenced from the residual protium in the NMR solvent (CDCl<sub>3</sub>:  $\delta$  7.24 (CHCl<sub>3</sub>), DMSO-d<sub>6</sub>: δ 2.50 (DMSO-d<sub>5</sub>)). Data are reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, t = triplet, m = multiplet), coupling constant(s) in Hertz, integration, assignment]. Carbon-13 nuclear magnetic resonance (<sup>13</sup>C NMR) spectra were recorded with a Varian INOVA-500 spectrometer, are reported in parts per million on the  $\delta$  scale, and are referenced from the carbon resonances of the solvent (CDCl<sub>3</sub>:  $\delta$  77.23, DMSO-d<sub>6</sub>:  $\delta$  39.51). Data are reported as follows: chemical shift (assignment). Fluorine-19 nuclear magnetic resonance (<sup>19</sup>F NMR) spectra were recorded with a Varian Mercury 300 spectrometer, are reported in parts per million on the  $\delta$ scale, and are referenced from the fluorine resonance of neat trichlorofluoromethane (CFCl<sub>3</sub>:  $\delta$  0). Data are reported as follows: chemical shift (assignment). Infrared data (IR) were obtained with a Perkin-Elmer 2000 FTIR, and are reported as follows: frequency of absorption (cm<sup>-1</sup>), intensity of absorption (s = strong, m = medium, w = weak, br = broad). Optical rotations were measured on a Jasco-1010 polarimeter. UV-Vis spectrophotometric data were acquired on a Varian Cary 50 Bio UV-Vis spectrophotometer. Preprativ HPLC was performed on a Waters system with the 1525 Binary HPLC Pump, 2489 UV/Vis Dtctor, 3100 Mass Detector, System Fluidics Organizer, and 2767 Sample Manager components. We are grateful to Dr. Li Li for obtaining the mass spectrometric data at the Department of Chemistry's Instrumentation Facility, Massachusetts Institute of Technology. High resolution mass spectra (HRMS) were recorded on a Bruker Daltonics APEXIV 4.7 Tesla FT-ICR-MS using either an electrospray (ESI) or direct analysis in real time (DART) ionization source.

<sup>&</sup>lt;sup>1</sup> Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. **1978**, 43, 2923.

<sup>&</sup>lt;sup>2</sup> Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. Organometallics **1996**, 15, 1518.

**Positional Numbering System.** At least three numbering schemes for diketopiperazine alkaloids exist in the literature.<sup>3</sup> In assigning the <sup>1</sup>H and <sup>13</sup>C NMR data of all intermediates en route to our total syntheses of (+)-naseseazines A and B, we wished to employ a uniform numbering scheme. For ease

of direct comparison, particularly between early intermediates and advanced compounds, the numbering scheme used by Barrow for (+)-WIN-64821 (using positional numbers 1–19) was optimal and is used in this document. This numbering system is also consistent with that employed in the isolation paper of (+)naseseazines A and B.<sup>4</sup> In key instances, the products are accompanied by the numbering system as shown below.



**Biogenetic Hypothesis.** We consider a biosynthetic hypothesis in which dimerization occurs at an advanced stage using well-elaborated tryptophan systems, such as cyclodipeptides, to be plausible.<sup>5</sup> Furthermore, a likely dimerization scheme would invoke a cationic mechanism (radical or non-radical) in contrast to that which is suggested in the isolation report.<sup>4</sup> Two contiguous or non-contiguous single-electron oxidation events of an indole core accompanied by a Friedel–Crafts addition of another tryptophan indole could give rise to a number of dimer cognates whose regioisomeric constitution would be controlled by the structure of an enzyme-binding pocket.



<sup>&</sup>lt;sup>3</sup> (a) Von Hauser, D.; Weber, H. P.; Sigg, H. P. *Helv. Chim. Acta* **1970**, *53*, 1061. (b) Barrow, C. J.; Cai, P.; Snyder, J. K.; Sedlock, D. M.; Sun, H. H.; Cooper, R. *J. Org. Chem.* **1993**, *58*, 6016. (c) Springer, J. P.; Büchi, G.; Kobbe, B.; Demain, A. L.; Clardy, J. *Tetrahedron Lett.* **1977**, *28*, 2403.

<sup>&</sup>lt;sup>4</sup> Raju, R.; Piggott, A. M.; Conte, M.; Aalbersberg, W. G. L.; Feussner, K.; Capon, R. J. Org. Lett. **2009**, *11*, 3862.

<sup>&</sup>lt;sup>5</sup> For hypotheses on dimeric cyclotryptamine biosynthesis, see: (a) Woodward, R. B.; Yang, N. C.; Katz, T. J.; Clark, V. M.; Harley-Mason, J.; Ingleby, R. F. J.; Sheppard, N. *Proc. Chem. Soc.* **1960**, 76, and (b) Robinson, R.; Teuber, H. J. *Chem. Ind.* **1954**, 783. For labeling and biosynthetic studies, see: (c) Kirby, G. W.; Shah, S. W.; Herbert, E. J. *J. Chem. Soc. C* **1969**, 1916, (d) O'Donovan, D. G.; Keogh, M. F. *J. Chem. Soc. C* **1966**, 1570, and (e) Schutte, H. R.; Maier, B. *Arch. Pharm.* **1965**, *298*, 459. For reviews, see: (f) Kim, J.; Movassaghi, M. *Chem. Soc. Rev.* **2009**, *38*, 3035, (g) Schmidt, M. A.; Movassaghi, M. *Synlett* **2008**, 313, and (h) Steven, A.; Overman, L. E. *Angew. Chem. Int. Ed.* **2007**, *46*, 5488.

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#### Benzyl 6-bromo-3-formyl-1H-indole-1-carboxylate (S1):

Triethylamine (1.98 mL, 14.2 mmol, 1.50 equiv) was added via syringe to a solution of 6bromo-3-formylindole (2.13 g, 9.49 mmol, 1 equiv) and 4-dimethylaminopyridine (116 mg, 949  $\mu$ mol, 0.100 equiv) in dichloromethane (25 mL) at 23 °C. Benzyl chloroformate (1.74 mL, 12.3 mmol, 1.30 equiv) was added dropwise to the solution via syringe. After 1 h, another portion of benzyl chloroformate (267  $\mu$ L, 1.90 mmol, 0.20 equiv) was added via syringe. After 40 min, the reaction mixture was diluted with dichloromethane (100 mL) and washed with saturated aqueous sodium bicarbonate solution (100 mL). The aqueous layer was further extracted with dichloromethane (2 × 25 mL). The combined organic layers were washed with aqueous hydrogen chloride (1 N, 100 mL) and the resulting aqueous layer was extracted with dichloromethane (2 × 25 mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. Benzyl 6-bromo-3-formyl-1*H*-indole-1-carboxylate (**S1**) (3.40 g, 100%) was obtained as a white solid. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 10.01 (s, 1H, C <sub>10</sub> <b>H</b> ), 8.35 (s, 1H, C <sub>8</sub> <b>H</b> ), 8.17 (s, 1H, C <sub>2</sub> <b>H</b> ), 8.09 (d, $J = 8.4$ , 1H, C <sub>5</sub> <b>H</b> ), 7.50–7.38 (m, 5H, Ph <sub>Cbz</sub> - <b>H</b> ), 7.38–7.50 (m, 1H, C <sub>6</sub> <b>H</b> ), 5.48 (s, 2H, Ph <sub>Cbz</sub> C <b>H</b> <sub>2</sub> ).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	δ 185.6 ( $C_{10}$ ), 149.9 ( $C=O_{Cbz}$ ), 136.7 ( $C_9$ ), 136.1 ( $C_2$ ), 134.2 ( $Ph_{Cbz}$ - <i>ipso</i> -C), 129.5 ( $Ph_{Cbz}$ -C), 129.2 ( $Ph_{Cbz}$ -C), 129.1 ( $Ph_{Cbz}$ -C), 128.5 ( $C_6$ ), 125.0 ( $C_4$ ), 123.5 ( $C_5$ ), 122.0 ( $C_3$ ), 120.3 ( $C_7$ ), 118.5 ( $C_8$ ), 70.3 ( $Ph_{Cbz}$ CH <sub>2</sub> ).
FTIR (thin film) cm <sup>-1</sup> :	1753 (s), 1679 (s), 1343 (m), 1224 (s), 1092 (m).
HRMS (DART) $(m/z)$ :	calc'd for C <sub>17</sub> H <sub>13</sub> BrNO <sub>3</sub> [M+H] <sup>+</sup> : 358.0073, found: 358.0077.
TLC (20% ethyl acetate in hexanes), Rf:	0.47 (UV, CAM).

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# (Z)-Benzyl 6-bromo-3-(2-((*tert*-butoxycarbonyl)amino)-3-methoxy-3-oxoprop-1-en-1-yl)-1*H*-indole-1-carboxylate (29):

1,8-Diazabicycloundec-7-ene (1.70 mL, 11.4 mmol, 1.20 equiv) was added dropwise to a solution of carboxaldehyde **S1** (3.40 g, 9.49 mmol, 1 equiv) and ( $\pm$ )-Boc- $\alpha$ -phosphonoglycine trimethyl ester (3.39 g, 11.4 mmol, 1.20 equiv) in dichloromethane (32 mL) at 0 °C. The ice-water bath was then removed and the reaction mixture was allowed to warm to 23 °C. After 4 h, the reaction mixture was diluted with dichloromethane (100 mL) and washed with aqueous hydrogen chloride (1 N, 100 mL). The resulting aqueous layer was extracted with dichloromethane (2 × 25 mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The resulting oil was purified by flash column chromatography on silica gel (eluent: 20% ethyl acetate in hexanes) to afford the enamide **29** (4.85 g, 97%) as a beige foam. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 8.27 (br-s, 1H, $C_8$ <b>H</b> ), 7.74 (br-s, 1H, $C_2$ <b>H</b> ), 7.46–7.33 (m, 1H, $C_{12}$ <b>H</b> ), 7.46–7.33 (m, 1H, $C_5$ <b>H</b> ),
	$7.46-7.33 \text{ (m, 5H, Ph}_{Cbz}-H\text{)}, 7.30 \text{ (dd, } J = 1.8, 8.4,$
	1H, $C_6$ <b>H</b> ), 6.44 (s, 1H, $N_{10}$ <b>H</b> ), 5.38 (s, 2H,
	Ph <sub>Cbz</sub> C <b>H</b> <sub>2</sub> ), 3.82 (s, 3H, OC <b>H</b> <sub>3</sub> ), 1.35 (s, 9H,
	$C(CH_3)_3).$
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\delta$ 165.7 ( <b>C</b> <sub>13</sub> ), 152.7 ( <b>C</b> =O <sub>Boc</sub> ), 150.0 ( <b>C</b> =O <sub>Cbz</sub> ),
	135.4 ( <b>C</b> <sub>9</sub> ), 134.5 (Ph <sub>Cbz</sub> - <i>ipso</i> - <b>C</b> ), 129.0 (Ph <sub>Cbz</sub> - <b>C</b> ),
	128.8 (Ph <sub>Cbz</sub> -C), 128.8 (Ph <sub>Cbz</sub> -C), 128.3 (C <sub>11</sub> ),
	127.1 ( $\mathbf{C}_2$ ), 126.7 ( $\mathbf{C}_6$ ), 124.2 ( $\mathbf{C}_4$ ), 121.4 ( $\mathbf{C}_{12}$ ),
	$120.2 (C_5), 118.9 (C_7), 118.3 (C_8), 114.8 (C_3), 81.0$
	$(C(CH_3)_3)$ , 69.5 $(Ph_{Cbz}CH_2)$ , 52.6 $(OCH_3)$ , 28.1
	$(C(\mathbf{CH}_3)_3).$
FTIR (thin film) cm <sup>-1</sup> :	3324 (br, m), 2979 (m), 1723 (s), 1244 (s), 759 (m).
HRMS (DART) $(m/z)$ :	calc'd for $C_{25}H_{25}BrN_2O_6 [M]^+$ : 528.0891, found: 528.0877.
TLC (20% ethyl acetate in hexanes), Rf:	0.31 (UV, CAM).

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#### $N^{\alpha}$ -Boc- $N^{\text{in}}$ -Cbz-6-bromotryptophan methyl ester (+)-30:

Dichloromethane (7.5 mL) and methanol (7.5 mL) were sequentially added via syringe to a Fischer-Porter tube charged with enamide **29** (4.00 g, 7.56 mmol, 1 equiv) and (+)-1,2-bis((2*S*,5*S*)-2,5-diethylphospholano)benzene(1,5-cyclooctadiene)rhodium(I) trifluoromethanesulfonate (100 mg, 138 µmol, 1.83 mol%). The reaction vessel was charged with hydrogen gas (80 psi) and then discharged (4 cycles). The vessel was charged a final time with hydrogen gas (80 psi) then sealed. After 9.5 h, the Fischer-Porter tube was depressurized and the reaction mixture was concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 25% ethyl acetate in hexanes) to afford bromotryptophan derivative (+)-**30** (3.88 g, 96.6 %, >99% ee)<sup>6</sup> as a white foam. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 8.35 (br-s, 1H, C <sub>8</sub> H), 7.48–7.30 (m, 1H, C <sub>2</sub> H), 7.48–7.30 (m, 1H, C <sub>5</sub> H), 7.48–7.30 (m, 1H, C <sub>6</sub> H), 7.48–7.30 (m, 5H, Ph <sub>Cbz</sub> -H), 5.41 (app-s, 2H, Ph <sub>Cbz</sub> CH <sub>2</sub> ), 5.11 (d, $J = 7.8$ , 1H, N <sub>10</sub> H), 4.60 (app- q, $J = 5.8$ , 1H, C <sub>11</sub> H), 3.64 (s, 3H, OCH <sub>3</sub> ), 3.21 (dd, $J = 5.5$ , 14.8, 1H, C <sub>12</sub> H <sub>a</sub> ), 3.10 (dd, $J = 5.5$ , 14.8, 1H, C <sub>12</sub> H <sub>b</sub> ), 1.38 (s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$ \begin{split} &\delta \ 172.3 \ (\mathbf{C}_{13}), \ 155.2 \ (\mathbf{C}=\mathbf{O}_{Boc}), \ 150.4 \ (\mathbf{C}=\mathbf{O}_{Cbz}), \\ &136.2 \ (\mathbf{C}_{9}), \ 135.0 \ (\mathbf{Ph}_{Cbz}\text{-}\textit{ipso-}\mathbf{C}), \ 129.6 \ (\mathbf{C}_{4}), \ 129.0 \\ &(\mathbf{Ph}_{Cbz}\text{-}\mathbf{C}), \ 129.0 \ (\mathbf{Ph}_{Cbz}\text{-}\mathbf{C}), \ 128.7 \ (\mathbf{Ph}_{Cbz}\text{-}\mathbf{C}), \ 126.4 \\ &(\mathbf{C}_{6}), \ 124.2 \ (\mathbf{C}_{2}), \ 120.3 \ (\mathbf{C}_{5}), \ 118.8 \ (\mathbf{C}_{7}), \ 118.6 \\ &(\mathbf{C}_{8}), \ 116.2 \ (\mathbf{C}_{3}), \ 80.3 \ (\mathbf{C}(\mathbf{CH}_{3})_{3}), \ 69.2 \ (\mathbf{Ph}_{Cbz}\mathbf{CH}_{2}), \\ &53.7 \ (\mathbf{C}_{11}), \ 52.6 \ (\mathbf{OCH}_{3}), \ 28.4 \ (\mathbf{C}(\mathbf{CH}_{3})_{3}), \ 27.9 \\ &(\mathbf{C}_{12}). \end{split} $
FTIR (thin film) cm <sup>-1</sup> :	3369 (m), 2978 (m), 1742 (s), 1715 (s), 1247 (s).
HRMS (DART) $(m/z)$ :	calc'd for $C_{25}H_{27}BrN_2O_6 [M]^+$ : 530.1047, found: 530.1071.
$\left[\alpha\right]_{D}^{24}$ :	$+32 (c = 0.19, \text{CHCl}_3).$
TLC (25% ethyl acetate in hexanes), Rf:	0.37 (UV, CAM).

<sup>&</sup>lt;sup>6</sup> In order to determine the enantiopurity, bromotryptophan derivative (+)-**30** was hydrogenated using palladium on carbon in acetic acid to remove the *N*-carboxybenzyl and aryl bromide functional groups. The product was then compared against commercially available enantioenriched *N*-Boc-L-Trp-OMe and *N*-Boc-D-Trp-OMe standards. Chiral HPLC analysis [Chiralpak IC column; 1.0 mL/min; 15% isopropanol in hexanes;  $t_R(N$ -Boc-L-Trp-OMe) = 10.6 min,  $t_R(N$ -Boc-D-Trp-OMe) = 13.5 min] showed the enantiopurity to be >99% ee.

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#### Bromoindole diketopiperazine (-)-31:

Trifluoroacetic acid (15 mL) was added via syringe to a solution of  $N^{\alpha}$ -Boc- $N^{\text{in}}$ -Cbz-L-6bromotryptophan methyl ester ((+)-30) (3.83 g, 7.21 mmol, 1 equiv) in dichloromethane (50 mL) at 23 °C. After 1 h, the brown solution was concentrated under reduced pressure to afford a viscous brown residue, which was dissolved in dichloromethane (100 mL) and cooled to 0 °C in an ice-water bath. Triethylamine (4.52 mL, 32.5 mmol, 4.50 equiv) was added dropwise via syringe. Nhydroxybenzotriazole (1.46 g, 10.8 mmol, 1.50 equiv) and  $N^{\alpha}$ -Boc-L-proline (3.10 g, 14.4 mmol, 2.00 equiv) were sequentially added to the solution. After 2 min, upon dissolution of the  $N^{\alpha}$ -Boc-Lproline, 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrogen chloride (2.07 g, 10.8 mmol, 1.50 equiv) was added and the reaction mixture was allowed to warm to 23 °C. After 1 h, the reaction mixture was diluted with dichloromethane (60 mL) and washed with aqueous hydrogen chloride solution (1 N, 250 mL). The aqueous layer was extracted with dichloromethane ( $2 \times 100$  mL). The combined organic layers were washed with saturated aqueous sodium bicarbonate solution (250 mL) and the resulting aqueous layer was extracted with dichloromethane  $(2 \times 25 \text{ mL})$ . The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The resulting off-white foam was then dissolved in dichloromethane (50 mL) and trifluoroacetic acid (15 mL) was added dropwise to the solution. After 1 h, the solution was concentrated under reduced pressure. The viscous residue was dissolved in methanol (200 mL) and cooled to 0 °C. Ammonium hydroxide (28-30% ammonia, 10 mL) was added dropwise and the reaction mixture was allowed to warm to 23 °C. After 22 h, the reaction mixture was cooled to -78°C and filtered. The white precipitate was washed with methanol ( $2 \times 50$  mL), which had been cooled to 0 °C in an ice-water bath, and was dried under reduced pressure at 23 °C to afford diketopiperazine (-)-31 (2.68 g, 74.9%, 2-steps) as a fluffy white solid. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, DMSO- <i>d</i> <sub>6</sub> , 20 °C):	$\delta$ 8.21 (s, 1H, C <sub>8</sub> H), 8.13 (s, 1H, N <sub>10</sub> H), 7.66 (d, J
	$= 8.4, 1H, C_5H$ ), 7.61 (s, 1H, C <sub>2</sub> H), 7.53 (d, J =
	$6.8, 2H, Ph_{Cbz}-o-H), 7.47-7.37 (m, 1H, C_6H),$
	7.47–7.37 (m, 3H, Ph <sub>Cbz</sub> - <b>H</b> ), 5.46 (app-s, 2H,
	$Ph_{Cbz}CH_2$ , 4.42 (app-t, $J = 5.2, 1H, C_{11}H$ ), 4.11
	$(app-t, J = 8.5, 1H, C_{15}H), 3.38-3.24 (m, 2H,$
	$C_{19}$ <b>H</b> ), 3.17 (dd, $J = 5.2, 15.3, 1$ H, $C_{12}$ H <sub>a</sub> ), 3.05
	$(dd, J = 5.2, 15.3, 1H, C_{12}H_b), 2.10-2.00 (m, 1H,$
	$C_{17}H_a$ ), 1.80–1.63 (m, 2H, $C_{18}H$ ), 1.61–1.47 (m,
	$1\mathrm{H},\mathrm{C}_{17}\mathrm{H}_{\mathrm{b}}$ ).
<sup>13</sup> C NMR (125.8 MHz, DMSO- <i>d</i> <sub>6</sub> , 20 °C):	δ 169.3 ( <b>C</b> <sub>16</sub> ), 165.1 ( <b>C</b> <sub>13</sub> ), 149.9 ( <b>C</b> =O <sub>Cbz</sub> ), 135.3
	$(C_9)$ , 135.2 (Ph <sub>Cbz</sub> - <i>ipso</i> -C), 129.6 (C <sub>4</sub> ), 128.7
	(Ph <sub>Cbz</sub> -C), 128.6 (Ph <sub>Cbz</sub> -C), 128.4 (Ph <sub>Cbz</sub> -C), 125.6
	$(\mathbf{C}_6), 124.9 (\mathbf{C}_2), 121.7 (\mathbf{C}_5), 117.3 (\mathbf{C}_7), 117.1$
	(C <sub>8</sub> ), 116.8 (C <sub>3</sub> ), 68.6 (Ph <sub>Cbz</sub> CH <sub>2</sub> ), 58.5 (C <sub>15</sub> ), 54.3
	$(\mathbf{C}_{11}), 44.7 (\mathbf{C}_{19}), 27.8 (\mathbf{C}_{17}), 24.8 (\mathbf{C}_{12}), 22.0 (\mathbf{C}_{18}).$

FTIR (thin film) $cm^{-1}$ :	3584 (s), 1737 (s), 1668 (s), 1434 (m), 1247 (m), 1084 (w).
HRMS (DART) $(m/z)$ :	calc'd for $C_{24}H_{23}BrN_3O_4 [M+H]^+$ : 496.0866, found: 496.0870.
$\left[\alpha\right]_{D}^{24}$ :	-33 (c = 0.09, DMSO).
TLC (30% acetone in dichloromethane), Rf:	0.31 (UV, CAM).

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#### Pinacol boronic ester (-)-32:

A 25-mL round bottom flask was charged with diketopiperazine (–)-**31** (500 mg, 1.01 mmol, 1 equiv), aminobiphenyl(XPhos)palladium chloride precatalyst complex (39.6 mg, 50.4  $\mu$ mol, 5.00 mol%), XPhos (72.2 mg, 151  $\mu$ mol, 15.0 mol%), bis(pinacolato)diboron (769 mg, 3.03 mmol, 3.00 equiv), and tribasic potassium phosphate (643 mg, 3.03 mmol, 3.00 equiv). The flask was then evacuated and charged with argon (3 cycles). Dimethyl sulfoxide (10 mL) was introduced to the flask via syringe and the initial suspension was warmed to 60 °C. After 2.5 h, the black solution was cooled to 23 °C, diluted with ethyl acetate (125 mL), and washed with saturated aqueous sodium bicarbonate solution (250 mL). The resulting aqueous layer was extracted with ethyl acetate (2 × 125 mL), and the combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude reaction mixture was then purified by flash column chromatography on silica gel (eluent: 3% acetone in ethyl acetate) to afford pinacol boronic ester (–)-**32** (356 mg, 65.0%) as a white foam. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 8.61 (br-s, 1H, C <sub>8</sub> H), 7.65 (d, $J = 7.8$ , 1H, C <sub>6</sub> H), 7.59 (s, 1H, C <sub>2</sub> H), 7.51 (d, $J = 7.9$ , 1H, C <sub>5</sub> H), 7.42 (d, $J = 7.1$ , 2H, Ph <sub>Cbz</sub> - <i>o</i> -H), 7.36–7.26 (m, 3H, Ph <sub>Cbz</sub> -H), 6.79 (s, 1H, N <sub>10</sub> H), 5.35 (app-s, 2H, Ph <sub>Cbz</sub> CH <sub>2</sub> ), 4.25 (app-d, $J = 7.2$ , 1H, C <sub>11</sub> H), 3.84 (app-t, $J = 7.1$ , 1H, C <sub>15</sub> H), 3.56 (dd, $J = 2.7$ , 15.2, 1H, C <sub>12</sub> H <sub>a</sub> ), 3.53–3.37 (m, 2H, C <sub>19</sub> H), 2.90 (dd, $J =$ 9.5, 15.3, 1H, C <sub>12</sub> H <sub>b</sub> ), 2.17–2.05 (m, 1H, C <sub>17</sub> H <sub>a</sub> ), 1.88–1.64 (m, 1H, C <sub>17</sub> H <sub>b</sub> ), 1.88–1.64 (m, 2H, C <sub>18</sub> H), 1.29 (s, 12H, OC(CH <sub>3</sub> ) <sub>2pinacol</sub> ).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\begin{split} &\delta \ 169.8 \ (\mathbf{C}_{16}), \ 165.0 \ (\mathbf{C}_{13}), \ 150.3 \ (\mathbf{C}=\mathbf{O}_{\text{Cbz}}), \ 135.3 \\ &(\mathbf{C}_{9}), \ 135.0 \ (\text{Ph}_{\text{Cbz}}\text{-}ipso\text{-}\mathbf{C}), \ 132.3 \ (\mathbf{C}_{4}), \ 129.0 \ (\mathbf{C}_{6}), \\ &128.6 \ (\text{Ph}_{\text{Cbz}}\text{-}\mathbf{C}), \ 128.6 \ (\text{Ph}_{\text{Cbz}}\text{-}\mathbf{C}), \ 128.4 \ (\text{Ph}_{\text{Cbz}}\text{-}\mathbf{C}), \\ &125.4 \ (\mathbf{C}_{2}), \ 125.4 \ (\mathbf{C}_{7}), \ 121.8 \ (\mathbf{C}_{8}), \ 118.2 \ (\mathbf{C}_{5}), \\ &116.1 \ (\mathbf{C}_{3}), \ 83.7 \ (\text{OC}(\text{CH}_{3})_{2pinacol}), \ 68.7 \ (\text{Ph}_{\text{Cbz}}\text{CH}_{2}), \\ &58.9 \ (\mathbf{C}_{15}), \ 54.3 \ (\mathbf{C}_{11}), \ 45.3 \ (\mathbf{C}_{19}), \ 28.0 \ (\mathbf{C}_{17}), \ 26.1 \\ &(\mathbf{C}_{12}), \ 24.8 \ (\text{OC}(\text{CH}_{3})_{2pinacol}), \ 22.5 \ (\mathbf{C}_{18}). \end{split}$
FTIR (thin film) $cm^{-1}$ :	2978 (m), 1738 (s), 1673 (s), 1432 (s), 1354 (s).
HRMS (DART) $(m/z)$ :	calc'd for C <sub>30</sub> H <sub>35</sub> BN <sub>3</sub> O <sub>6</sub> [M+H] <sup>+</sup> : 544.2613, found: 544.2613.
$\left[\alpha\right]_{D}^{23}$ :	$-63 (c = 0.2, \text{CHCl}_3).$
TLC (3% acetone in ethyl acetate), Rf:	0.20 (UV, CAM).

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#### Potassium trifluoroborateindole (-)-33:

An aqueous solution of potassium hydrogen fluoride (4.5 M, 4.73 mL, 21.3 mmol, 20.0 equiv) was added dropwise to a solution of pinacol boronic ester (-)-**32** (578 mg, 1.06 mmol, 1 equiv) in methanol (15 mL) at 23 °C. After 1.5 h, the turbid solution was concentrated to dryness under reduced pressure. The white solid residue was suspended in a dichloromethane and hexanes solution (1:1, 50 mL) and filtered over Celite to remove the pinacol byproduct. After further washing with a dichloromethane and hexanes solution (1:1, 50 mL), the solids were extracted with acetone (80 mL) and the filtrate was concentrated under reduced pressure to afford potassium 6-trifluoroborateindole (-)-**33** (489 mg, 87.8%) as a beige solid. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, acetone- <i>d</i> <sub>6</sub> , 20 °C):	δ 8.40 (s, 1H, C <sub>8</sub> H), 7.54 (d, $J = 7.3$ , 2H, Ph <sub>Cbz</sub> - $o$ - H), 7.52 (s, 1H, C <sub>2</sub> H), 7.45 (d, $J = 7.7$ , 1H, C <sub>6</sub> H), 7.43–7.39 (m, 3H, Ph <sub>Cbz</sub> -H), 7.36 (d, $J = 7.5$ , 1H, C <sub>5</sub> H), 6.79 (s, 1H, N <sub>10</sub> H), 5.44 (app-s, 2H, Ph <sub>Cbz</sub> CH <sub>2</sub> ), 4.41 (dd, $J = 3.6$ , 8.3, 1H, C <sub>11</sub> H), 4.10 (app-t, $J = 8.1$ , 1H, C <sub>15</sub> H), 3.54–3.43 (m, 1H, C <sub>12</sub> H <sub>a</sub> ), 3.54–3.43 (m, 1H, C <sub>19</sub> H <sub>a</sub> ), 3.42–3.33 (m, 1H, C <sub>19</sub> H <sub>b</sub> ), 2.96 (dd, $J = 8.3$ , 15.5, 1H, C <sub>12</sub> H <sub>b</sub> ), 2.18–2.07 (m, 1H, C <sub>17</sub> H <sub>a</sub> ), 1.85–1.70 (m, 1H,
<sup>13</sup> C NMR (125.8 MHz, acetone- <i>d</i> <sub>6</sub> , 20 °C) <sup>7</sup> :	$\begin{array}{l} C_{17}\mathbf{H}_{b}, 1.85-1.70 \ (m, 2H, C_{18}\mathbf{H}). \\ \delta \ 170.1 \ (\mathbf{C}_{16}), 166.1 \ (\mathbf{C}_{13}), 151.3 \ (\mathbf{C}_{Cbz}=\mathbf{O}), 136.9 \\ (Ph_{Cbz}\text{-}ipso\text{-}\mathbf{C}), 136.5 \ (\mathbf{C}_{9}), 129.3 \ (Ph_{Cbz}\text{-}\mathbf{C}), 129.0 \\ (Ph_{Cbz}\text{-}\mathbf{C}), 129.0 \ (Ph_{Cbz}\text{-}\mathbf{C}), 129.0 \ (\mathbf{C}_{4}), 127.7 \ (\mathbf{C}_{6}), \\ 123.0 \ (\mathbf{C}_{2}), 118.5 \ (\mathbf{C}_{8}), 117.6 \ (\mathbf{C}_{3}), 117.5 \ (\mathbf{C}_{5}), 68.4 \\ (Ph_{Cbz}\mathbf{CH}_{2}), 59.5 \ (\mathbf{C}_{15}), 55.2 \ (\mathbf{C}_{11}), 45.6 \ (\mathbf{C}_{19}), 28.7 \\ (\mathbf{C}_{17}), 26.6 \ (\mathbf{C}_{12}), 23.0 \ (\mathbf{C}_{18}). \end{array}$
<sup>19</sup> F NMR (282.4 MHz, acetone- <i>d</i> <sub>6</sub> , 20 °C):	$\delta - 141.0 (C_6 BF_3 K).$
FTIR (thin film) cm <sup>-1</sup> :	3583 (br, s), 1726 (m), 1666 (s), 1422 (m), 1255 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{24}H_{22}BF_3N_3O_4[M-K]^-: 484.1661$ , found: 484.1643.
$[\alpha]_{D}^{21}$ :	-42 (c = 0.14, acetone).

<sup>&</sup>lt;sup>7</sup> C<sub>7</sub> is not observed in the <sup>13</sup>C NMR spectrum. For other examples of unobservable carbon atoms *ipso* to the trifluoroborate functional group, see: Molander, G.; Argintaru, O. A.; Aron, I.; Dreher, S. D. *Org. Lett.* **2010**, *12*, 5783.

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#### <u>Alanine diketopiperazine (–)-(9):</u>

Trifluoroacetic acid (30 mL) was added via syringe to a solution of  $N^{\alpha}$ -Boc- $N^{\text{in}}$ -Cbz-Ltryptophan methyl ester<sup>8</sup> (9.0 g, 19.9 mmol, 1 equiv) in dichloromethane (100 mL) at 23 °C. After 1 h, the brown solution was concentrated under reduced pressure to afford a viscous brown residue, which was dissolved in dichloromethane (200 mL) and cooled to 0 °C in an ice-water bath. Triethylamine (12.5 mL, 89.5 mmol, 4.50 equiv) was added dropwise via syringe.  $N_{-}$ hydroxybenzotriazole (4.03 g, 29.8 mmol, 1.50 equiv) and  $N^{\alpha}$ -Boc-L-alanine (7.53 g, 39.8 mmol, 2.00 equiv) were sequentially added to the resulting solution. After 2 min, upon dissolution of the  $N^{\alpha}$ -Boc-L-alanine, 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrogen chloride (5.72 g, 29.8 mmol, 1.50 equiv) was added and the reaction mixture was allowed to warm to 23 °C. After 13.5 h, aqueous hydrogen chloride solution (1 N, 500 mL) was added and the aqueous layer was extracted with dichloromethane  $(2 \times 100 \text{ mL})$ . The combined organic layers were washed with saturated aqueous sodium bicarbonate solution (500 mL) and the resulting aqueous layer was extracted with dichloromethane ( $2 \times 100$  mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The resulting off-white foam was then dissolved in dichloromethane (100 mL) and trifluoroacetic acid (30 mL) was added dropwise to the solution. After 2 h, the solution was concentrated under reduced pressure. The viscous residue was dissolved in methanol (300 mL) and cooled to 0 °C. Ammonium hydroxide (28–30% ammonia, 10.5 mL) was added dropwise and the reaction mixture was allowed to warm to 23 °C. After 23 h, the white precipitate was filtered and washed with methanol ( $2 \times 50$  mL), which had been cooled to 0 °C in an ice-water bath. The white powder was dried under reduced pressure at 23 °C to afford diketopiperazine (-)-9 (6.7 g, 89%, 2-steps). Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, DMSO- <i>d</i> <sub>6</sub> , 20 °C):	δ 8.17 (s, 1H, N <sub>14</sub> <b>H</b> ), 8.10 (s, 1H, N <sub>10</sub> <b>H</b> ), 8.06 (d, $J = 8.1, 1H, C_8$ <b>H</b> ), 7.67 (d, $J = 7.8, 1H, C_5$ <b>H</b> ), 7.51 (d, $J = 6.3, 2H, Ph_{Cbz}$ - <i>o</i> - <b>H</b> ), 7.51 (s, 1H, C <sub>2</sub> <b>H</b> ), 7.43 (app-t, $J = 7.0, 2H, Ph_{Cbz}$ - <i>m</i> - <b>H</b> ), 7.38 (t, $J = 7.1, 1H, Ph_{Cbz}$ - <i>p</i> - <b>H</b> ), 7.32 (app-t, $J = 7.3, 1H, C_7$ <b>H</b> ), 7.25 (app-t, $J = 7.9, 1H, C_6$ <b>H</b> ), 5.47 (app-s, 2H, Ph_{Cbz}CH <sub>2</sub> ), 4.22 (app-t, $J = 4.3, 1H, C_{11}$ <b>H</b> ), 3.71 (q, $J = 7.0, 1H, C_{15}$ <b>H</b> ), 3.20 (dd, $J = 4.4, 14.7, 1H, C_{13}$ <b>H</b> <sub>2</sub> ), 3.03 (dd, $J = 4.6, 14.6, 1H, C_{12}$ <b>H</b> <sub>b</sub> ), 0.66 (d, J) = 0.05 (d), 0.0
<sup>13</sup> C NMR (125.8 MHz, DMSO- <i>d</i> <sub>6</sub> , 20 °C):	$J = 7.0, 3H, C_{17}H).$ $\delta 168.0 (C_{16}), 166.6 (C_{13}), 150.1 (C=O_{Cbz}), 135.4 (Ph_{Cbz}-ipso-C), 134.6 (C_9), 130.6 (C_4), 128.7 (Ph_{Cbz}-C), 128.5 (Ph_{Cbz}-C), 128.2 (Ph_{Cbz}-C), 124.6 (C_7), 124.3 (C_2), 122.8 (C_6), 120.0 (C_5), 116.2$

<sup>&</sup>lt;sup>8</sup> Prepared from commercially available *N*<sup>u</sup>-Boc-tryptophan methyl ester in one step: Kiso, Y.; Inai, M.; Kitagawa, K.; Akita, T. *Chem. Lett.* **1983**, *5*, 739.

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	$(\mathbf{C}_3), 114.4 (\mathbf{C}_8), 68.3 (Ph_{Cbz}\mathbf{C}H_2), 54.4 (\mathbf{C}_{11}), 49.7 (\mathbf{C}_{15}), 27.9 (\mathbf{C}_{12}), 19.5 (\mathbf{C}_{17}).$
FTIR (thin film) cm <sup>-1</sup> :	3584 (m), 3050 (br, s), 1739 (m), 1678 (s), 1453 (m), 1244 (m).
HRMS (DART) $(m/z)$ :	calc'd for $C_{22}H_{20}N_3O_4$ [M–H] <sup>-</sup> : 390.1459, found: 390.1458.
$\left[\alpha\right]_{D}^{24}$ :	-12 (c = 0.17, DMSO).
TLC (5% methanol in dichloromethane), Rf:	0.59 (UV, CAM).

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#### **Tetracyclic bromide A (+)-11:**

Pyridinium tribromide (4.24 g, 13.3 mmol, 2.00 equiv) was added to a suspension of diketopiperazine (–)-9 (2.50 g, 6.62 mmol, 1 equiv) in 2,2,2-trifluoroethanol (132 mL) at 23 °C. After 20 min, once the white solids have fully dissolved, a solution of saturated aqueous sodium thiosulfate (100 mL) was added. The reaction mixture was diluted with dichloromethane (200 mL) and washed with saturated aqueous sodium thiosulfate (200 mL). The resulting aqueous solution was further extracted with dichloromethane (2 × 150 mL) and the combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude reaction mixture was then purified by flash column chromatography on silica gel (eluent: 60% ethyl acetate in hexanes) to afford tetracyclic bromide (+)-11 (2.08 g, 69.2%) as a white foam. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 7.65 (d, $J = 8.1$ , 1H, C <sub>8</sub> H), 7.44 (d, $J = 7.2$ , 2H, Ph <sub>Cbz</sub> - <i>o</i> -H), 7.40 (d, $J = 7.6$ , 1H, C <sub>5</sub> H), 7.37 (app-t, J = 7.0, 2H, Ph <sub>Cbz</sub> - <i>m</i> -H), 7.32 (t, $J = 7.2$ , 1H, Ph <sub>Cbz</sub> - <i>p</i> -H), 7.24 (app-t, $J = 7.6$ , 1H, C <sub>7</sub> H), 7.06 (app-t, $J = 7.5$ , 1H, C <sub>6</sub> H), 6.32 (s, 1H, C <sub>2</sub> H), 6.21 (s, 1H, N <sub>14</sub> H), 5.42 (d, $J = 12.2$ , 1H, Ph <sub>Cbz</sub> CH <sub>a</sub> ), 5.32 (d, $J$
	= 12.2, 1H, $Pn_{Cbz}CH_b$ ), 4.32 (dd, $J = 3.5$ , 10.1, 1H, $C_{11}H$ ), 3.93 (q, $J = 6.7$ , 1H, $C_{15}H$ ), 3.69 (dd, $J = 3.5$ , 14.1, 1H, $C_{12}H_a$ ), 3.06 (dd, $J = 10.2$ , 14.2, 1H, $C_{12}H_b$ ), 1.29 (d, $J = 6.8$ , 3H, $C_{17}H$ ).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$ \begin{split} &\delta \ 168.9 \ (\mathbf{C}_{13}), \ 167.3 \ (\mathbf{C}_{16}), \ 152.9 \ (\mathbf{C}=\mathbf{O}_{Cbz}), \ 139.4 \\ &(\mathbf{C}_9), \ 135.8 \ (\mathbf{Ph}_{Cbz}\text{-}ipso\text{-}\mathbf{C}), \ ), \ 132.7 \ (\mathbf{C}_4), \ 130.9 \ (\mathbf{C}_7), \\ &128.6 \ (\mathbf{Ph}_{Cbz}\text{-}\mathbf{C}), \ 128.3 \ (\mathbf{Ph}_{Cbz}\text{-}\mathbf{C}), \ 128.2 \ (\mathbf{Ph}_{Cbz}\text{-}\mathbf{C}), \\ &124.7 \ (\mathbf{C}_6), \ 124.6 \ (\mathbf{C}_5), \ 116.8 \ (\mathbf{C}_8), \ 84.6 \ (\mathbf{C}_2), \ 68.3 \\ &(\mathbf{Ph}_{Cbz}\mathbf{CH}_2), \ 60.3 \ (\mathbf{C}_3), \ 58.0 \ (\mathbf{C}_{11}), \ 51.4 \ (\mathbf{C}_{15}), \ 37.6 \\ &(\mathbf{C}_{12}), \ 14.8 \ (\mathbf{C}_{17}). \end{split} $
FTIR (thin film) $cm^{-1}$ :	3273 (br-m), 1720 (s), 1692 (s), 1408 (m), 752 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{22}H_{21}BrN_3O_4 [M+H]^+: 470.0710$ , found 470.0692.
$\left[\alpha\right]_{D}^{25}$ :	$+71 (c = 0.09, CHCl_3).$
TLC (60% ethyl acetate in hexanes), Rf:	0.20 (UV, CAM).

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#### (-)-N<sup>in</sup>, N<sup>in</sup>'-Dicarboxybenzyl Naseseazine A (15):

Silver(I) hexafluoroantimonate (73.1 mg, 213 µmol, 5.00 equiv) was added as a solid to a solution of tetracyclic bromide (+)-**11** (20.0 mg, 42.5 µmol, 1 equiv), potassium 6-trifluoroborateindole (-)-**33** (33.4 mg, 63.8 µmol, 1.50 equiv), and 18-crown-6 (56.2 mg, 213 µmol, 5.00 equiv) in nitroethane (2 mL) at 23 °C. After 1 h, aqueous hydrogen chloride (2 N, 2 mL) was added. After stirring for 5 min, the reaction mixture was diluted with dichloromethane (60 mL) and washed with aqueous hydrogen chloride (2 N, 60 mL). The resulting aqueous layer was further extracted with dichloromethane (2 × 30 mL) and the combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 8% methanol in ethyl acetate) to afford (-)- $N^{in}$ , $N^{int}$ -dicarboxybenzyl naseseazine A (**15**) (19.3 mg, 56.2%) as a white solid. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	$\delta$ 8.11 (br-s, 1H, C <sub>8</sub> H), 7.62 (d, J = 7.8, 1H, C <sub>8</sub> H),
	7.46 (s, 1H, $C_2$ <b>H</b> ), 7.40 (d, $J$ = 7.0, 2H, $Ph_{Cbz}$ - $o$ - <b>H</b> ),
	7.37–7.28 (m, 5H, Ph <sub>Cbz</sub> -H), 7.37–7.28 (m, 1H,
	$C_5$ <b>H</b> ), 7.28–7.20 (m, 3H, Ph <sub>Cbz</sub> - <b>H</b> ), 7.15(app-t, J =
	7.6, 1H, $C_7$ <b>H</b> ), 7.05 (d, $J = 7.2$ , 1H, $C_5$ <b>H</b> ), 6.93
	$(app-t, J = 7.4, 1H, C_6H), 6.90 (d, J = 8.4, 1H),$
	$C_6$ <b>H</b> ), 6.41 (s, 1H, N <sub>10</sub> <b>H</b> ), 6.28 (s, 1H, N <sub>14</sub> <b>H</b> ), 6.28
	(s, 1H, C <sub>2</sub> H), 5.38–5.21 (m, 4H, Ph <sub>Cbz</sub> CH <sub>2</sub> ), 4.29
	$(d, J = 9.5, 1H, C_{11}H), 4.18 (br-s, 1H, C_{11}H), 4.01$
	$(app-t, J = 7.5, 1H, C_{15}H), 3.91 (q, J = 6.8, 1H)$
	$C_{15}$ <b>H</b> ), 3.64–3.47 (m, 2H, $C_{19}$ <b>H</b> ), 3.64–3.47 (m,
	1H, $C_{12}H_a$ ), 3.30 (d, $J = 12.9$ , 1H, $C_{12}H_a$ ), 2.85 (dd,
	$J = 10.7, 15.5, 1H, C_{12}H_b$ , 2.81–2.70 (m, 1H,
	$C_{12}\mathbf{H}_{b}$ ), 2.33–2.21 (m, 1H, $C_{17}\mathbf{H}_{a}$ ), 2.04–1.91 (m,
	1H, C <sub>17</sub> <b>H</b> <sub>b</sub> ), 2.04–1.91 (m, 1H, C <sub>18</sub> H <sub>a</sub> ), 1.91–1.78
	$(m, 1H, C_{18}H_b), 1.30 (d, J = 6.7, 3H, C_{17}H).$
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\delta$ 170.0 ( <b>C</b> <sub>16</sub> ), 169.6 ( <b>C</b> <sub>13</sub> ), 167.7 ( <b>C</b> <sub>16</sub> ), 165.2 ( <b>C</b> <sub>13</sub> ),
	$153.2 (C_{Cbz}=0), 150.6 (C_{Cbz}=0), 140.0 (C_9), 139.3$
	$(\mathbf{C}_{7'}), 136.3 (Ph_{Cbz}-ipso-\mathbf{C}), 135.0 (\mathbf{C}_4), 135.0 (\mathbf{C}_9),$
	135.0 (Ph <sub>Chz</sub> - <i>ipso</i> - <b>C</b> ), 129.5 ( <b>C</b> <sub>4</sub> ), 129.2 ( <b>C</b> <sub>7</sub> ), 129.1
	(Ph <sub>Cbz</sub> -C), 129.0 (Ph <sub>Cbz</sub> -C), 129.0 (Ph <sub>Cbz</sub> -C), 128.6
	(Ph <sub>Cbz</sub> -C), 128.2 (Ph <sub>Cbz</sub> -C), 128.2 (Ph <sub>Cbz</sub> -C), 125.3
	$(\mathbf{C}_5), 124.7 (\mathbf{C}_2), 124.3 (\mathbf{C}_6), 121.4 (\mathbf{C}_6), 119.5$
	$(\mathbf{C}_{5}), 117.0 (\mathbf{C}_{8}), 116.1 (\mathbf{C}_{3}), 113.2 (\mathbf{C}_{8}), 83.9 (\mathbf{C}_{2}),$

69.1 (Ph<sub>Chz</sub>CH<sub>2</sub>), 68.1 (Ph<sub>Chz</sub>CH<sub>2</sub>), 59.3 (C<sub>15</sub>), 58.9

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	$(\mathbf{C}_3), 58.5 (\mathbf{C}_{11}), 54.2 (\mathbf{C}_{11'}), 51.9 (\mathbf{C}_{15}), 45.7 (\mathbf{C}_{19'}), 35.4 (\mathbf{C}_{12}), 28.4 (\mathbf{C}_{17'}), 26.4 (\mathbf{C}_{12'}), 22.8 (\mathbf{C}_{18'}), 15.3 (\mathbf{C}_{17}).$
FTIR (thin film) cm <sup>-1</sup> :	3226 (br, m), 1688 (s), 1399 (m), 1308 (m), 751 (m).
HRMS (DART) $(m/z)$ :	calc'd for $C_{46}H_{43}N_6O_8$ [M+H] <sup>+</sup> : 807.3137 found: 807.3116.
$\left[\alpha\right]_{D}^{24}$ :	$-42 (c = 0.38, \text{CHCl}_3).$
TLC (7.5% methanol in ethyl acetate), Rf:	0.11 (UV, CAM).

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#### (+)-Naseseazine A (1):

Palladium on charcoal (10% w/w, 1.0 mg, 905 nmol, 0.10 equiv) was added to a solution of  $(-)-N^{in},N^{int}$ -dicarboxybenzyl naseseazine A (**15**) (7.3 mg, 9.05 µmol, 1 equiv) in acetic acid (1 mL) at 23 °C. A stream of hydrogen gas was passed through the solution for 2 min by discharge of a balloon equipped with a needle extending into the reaction mixture. After stirring the solution for 9 h under an atmosphere of hydrogen gas, the solution was filtered over Celite. The solids were further extracted with methanol and the combined filtrates were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 5% methanol, 45% tetrahydrofuran, 50% dichloromethane) to afford (+)-naseseazine A (**1**) (4.1 mg, 80%) as a white solid. Structural assignments were made using additional information from gCOSY, HSQC, HMBC, and ROESY experiments.

<sup>1</sup> H NMR (500 MHz, methanol- $d_4$ , 20 °C):	$\delta$ 7.57 (d, J = 8.4, 1H, C <sub>5</sub> H), 7.40 (d, J = 1.3, 1H,
	$C_8$ <b>H</b> ), 7.12 (s, 1H, $C_2$ <b>H</b> ), 7.05 (app-dt, $J = 1.3, 7.9$ ,
	1H, $C_7$ H), 7.03 (dd, $J = 1.7, 8.4, 1H, C_6$ H), 6.85
	$(dd, J = 0.6, 7.4, 1H, C_5H), 6.69 (d, J = 8.0, 1H)$
	$C_8$ <b>H</b> ), 6.68 (app-dt, $J = 0.9, 7.4, 1$ H, $C_6$ <b>H</b> ), 5.83 (s,
	1H, C <sub>2</sub> H), 4.65 (dd, $J = 7.4, 9.2, 1H, C_{11}H$ ), 4.39
	$(dt, J = 1.1, 4.7, 1H, C_{11}H), 4.16 (dq, J = 1.4, 6.9),$
	1H, $C_{15}$ H), 3.98 (ddd, $J = 1.7, 6.3, 10.9, 1$ H,
	$C_{15}$ <b>H</b> ), 3.42 (app-dt, $J = 8.3, 11.7, 1$ <b>H</b> , $C_{19}$ <b>H</b> <sub>a</sub> ),
	$3.30-3.20 \text{ (m, 1H, C_{19}H_b)}, 3.30-3.20 \text{ (m, 2H,}$
	$C_{12}$ <b>H</b> ), 3.30–3.20 (m, 1H, $C_{12}$ <b>H</b> <sub>a</sub> ), 2.59 (dd, $J = 9.9$ ,
	13.6, 1H, $C_{12}H_{b}$ ), 1.96 (app-ddt, $J = 2.0, 7.2, 12.5,$
	1H, $C_{17}$ H <sub>a</sub> ), 1.72–1.61 (m, 1H, $C_{18}$ H <sub>a</sub> ), 1.47–1.39
	$(m, 1H, C_{18}H_b), 1.38 (d, J = 7.1, 3H, C_{17}H),$
	$0.96-0.85 (m, 1H, C_{17}\mathbf{H}_{b}).$
<sup>1</sup> H NMR (500 MHz, DMSO- <i>d</i> <sub>6</sub> , 20 °C):	$\delta$ 10.81 (s, 1H, N <sub>1</sub> <b>H</b> ), 8.20 (s, 1H, N <sub>14</sub> <b>H</b> ), 7.71 (s,
	1H, $N_{10}$ H), 7.56 (d, $J = 8.4$ , 1H, $C_5$ H), 7.28 (s, 1H,
	$C_{2}$ <b>H</b> ), 7.19 (d, $J = 1.8$ , 1H, $C_{2}$ <b>H</b> ), 7.04–6.95 (m,
	1H, C <sub>7</sub> H), 7.04–6.95 (m, 1H, C <sub>6</sub> H), 6.83 (d, $J =$
	7.3, 1H, C <sub>5</sub> H), 6.75 (d, $J = 2.8$ , 1H, N <sub>1</sub> H), 6.62 (d,
	$J = 8.0, 1H, C_8H$ , 6.58 (app-t, $J = 7.4, 1H, C_6H$ ),
	5.65 (d, $J = 2.8$ , 1H, C <sub>2</sub> H), 4.62 (app-t, $J = 8.7$ , 1H,
	$C_{11}$ <b>H</b> ), 4.28 (app-t, $J = 4.9, 1$ <b>H</b> , $C_{11}$ <b>H</b> ), 4.14 (q, $J =$
	6.5, 1H, $C_{15}H$ ), 4.06 (app-t, $J = 7.8$ , 1H, $C_{15}H$ ),

 $3.41-3.29 \text{ (m, 1H, C}_{19}\mathbf{H}_{a}$ ),  $3.30-3.18 \text{ (m, 1H, C}_{19}\mathbf{H}_{b}$ )  $3.30-3.18 \text{ (m, 1H, C}_{12}\mathbf{H}_{a}$ ), 3.14-3.00 (m, m)

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	1H, $C_{12}H_a$ ), 3.14–3.00 (m, 1H, $C_{12}H_b$ ), 2.41 (dd, $J = 10.0, 13.6, 1H, C_{12}H_b$ ), 2.03–1.94 (m, 1H, $C_{17}H_a$ ), 1.74–1.56 (m, 2H, $C_{18}H$ ), 1.47–1.36 (m, 1H, $C_{17}H_b$ ), 1.23 (d, $J = 6.9, 3H, C_{17}H$ ).
<sup>13</sup> C NMR (125.8 MHz, methanol- <i>d</i> <sub>4</sub> , 20 °C):	$ \begin{split} &\delta \ 172.7 \ (\mathbf{C}_{13}), \ 170.9 \ (\mathbf{C}_{16}), \ 170.9 \ (\mathbf{C}_{16}), \ 167.5 \ (\mathbf{C}_{13}), \\ &149.3 \ (\mathbf{C}_{9}), \ 138.1 \ (\mathbf{C}_{9}), \ 137.3 \ (\mathbf{C}_{7}), \ 136.1 \ (\mathbf{C}_{4}), \\ &129.6 \ (\mathbf{C}_{7}), \ 127.8 \ (\mathbf{C}_{4}), \ 126.6 \ (\mathbf{C}_{2}), \ 125.1 \ (\mathbf{C}_{5}), \\ &120.6 \ (\mathbf{C}_{6}), \ 120.5 \ (\mathbf{C}_{5}), \ 119.7 \ (\mathbf{C}_{6}), \ 111.2 \ (\mathbf{C}_{8}), \\ &110.4 \ (\mathbf{C}_{8}), \ 109.7 \ (\mathbf{C}_{3}), \ 87.3 \ (\mathbf{C}_{2}), \ 61.4 \ (\mathbf{C}_{3}), \ 60.5 \\ &(\mathbf{C}_{11}), \ 60.2 \ (\mathbf{C}_{15}), \ 57.4 \ (\mathbf{C}_{11}), \ 52.4 \ (\mathbf{C}_{15}), \ 46.1 \ (\mathbf{C}_{19}), \\ &39.9 \ (\mathbf{C}_{12}), \ 29.3 \ (\mathbf{C}_{17}), \ 29.3 \ (\mathbf{C}_{12}), \ 22.7 \ (\mathbf{C}_{18}), \ 15.4 \ (\mathbf{C}_{17}). \end{split} $
<sup>13</sup> C NMR (125.8 MHz, DMSO- <i>d</i> <sub>6</sub> , 20 °C):	$ \begin{split} &\delta \ 170.0 \ (\mathbf{C}_{13}), \ 169.1 \ (\mathbf{C}_{16}), \ 168.6 \ (\mathbf{C}_{16}), \ 165.5 \ (\mathbf{C}_{13}), \\ &148.1 \ (\mathbf{C}_{9}), \ 135.9 \ (\mathbf{C}_{9}), \ 135.7 \ (\mathbf{C}_{7}), \ 134.4 \ (\mathbf{C}_{4}), \\ &127.9 \ (\mathbf{C}_{7}), \ 126.1 \ (\mathbf{C}_{4}), \ 125.0 \ (\mathbf{C}_{2}), \ 123.6 \ (\mathbf{C}_{5}), \\ &119.2 \ (\mathbf{C}_{5}), \ 118.0 \ (\mathbf{C}_{6}), \ 117.8 \ (\mathbf{C}_{6}), \ 109.3 \ (\mathbf{C}_{8}), \\ &109.2 \ (\mathbf{C}_{8}), \ 109.1 \ (\mathbf{C}_{3}), \ 85.0 \ (\mathbf{C}_{2}), \ 59.3 \ (\mathbf{C}_{3}), \ 58.5 \\ &(\mathbf{C}_{15}), \ 58.4 \ (\mathbf{C}_{11}), \ 55.2 \ (\mathbf{C}_{11}), \ 50.4 \ (\mathbf{C}_{15}), \ 44.6 \ (\mathbf{C}_{19}), \\ &38.8 \ (\mathbf{C}_{12}), \ 27.7 \ (\mathbf{C}_{17}), \ 25.7 \ (\mathbf{C}_{12}), \ 21.9 \ (\mathbf{C}_{18}), \ 14.8 \\ &(\mathbf{C}_{17}). \end{split} $
FTIR (thin film) cm <sup>-1</sup> :	3584 (s), 3272 (br, s), 2924 (w), 1667 (s), 1453 (w), 1412 (m), 1346 (w), 1307 (w).
UV (CH <sub>3</sub> OH) $\lambda_{max}$ (log $\epsilon$ ):	229 (4.67), 285 (3.91).
HRMS (DART) $(m/z)$ :	calc'd for $C_{30}H_{31}N_6O_4$ [M+H] <sup>+</sup> : 539.2401, found: 539.2392.
$[\alpha]_{D}^{24}$ :	+123 ( $c = 0.12$ , CH <sub>3</sub> OH). <sup>9</sup>

TLC (5% methanol, 47.5% tetrahydrofuran, 47.5% dichloromethane), Rf: 0.15 (UV, CAM).

<sup>&</sup>lt;sup>9</sup> Literature value:  $[\alpha]_D^{23} = +139$  (*c* 0.10, CH<sub>3</sub>OH), see Raju, R.; Piggott, A. M.; Conte, M.; Aalbersberg, W. G. L.; Feussner, K.; Capon, R. J. *Org. Lett.* **2009**, *11*, 3862.



#### Proline diketopiperazine (-)-(10):

Trifluoroacetic acid (30 mL) was added via syringe to a solution of  $N^{\alpha}$ -Boc- $N^{\text{in}}$ -Cbz-Ltryptophan methyl ester<sup>8</sup> (7.10 g, 15.7 mmol, 1 equiv) in dichloromethane (100 mL) at 23 °C. After 1 h, the brown solution was concentrated under reduced pressure to afford a viscous brown residue, which was dissolved in dichloromethane (160 mL) and cooled to 0 °C in an ice-water bath. Triethylamine (9.85 mL, 70.7 mmol, 4.50 equiv) was added dropwise via syringe. Nhydroxybenzotriazole (3.18 g, 23.6 mmol, 1.50 equiv) and  $N^{\alpha}$ -Boc-L-proline (6.76 g, 31.4 mmol, 2.00 equiv) were sequentially added to the solution. After 2 min, upon dissolution of the  $N^{\alpha}$ -Boc-Lproline, 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrogen chloride (4.51 g, 23.6 mmol, 1.50 equiv) was added and the reaction mixture was allowed to warm to 23 °C. After 1 h and 45 min, another portion of 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrogen chloride (1.50 g, 7.85 mmol, 0.500 equiv) was added. After 1.5 h, aqueous hydrogen chloride solution (1 N, 500 mL) was added and the aqueous layer was extracted with dichloromethane ( $2 \times 50$  mL). The combined organic layers were washed with saturated aqueous sodium bicarbonate solution (500 mL) and the resulting aqueous layer was extracted with dichloromethane  $(2 \times 50 \text{ mL})$ . The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced The resulting off-white foam was then dissolved in dichloromethane (100 mL) and pressure. trifluoroacetic acid (30 mL) was added dropwise to the solution at 23 °C. After 1 h, the solution was concentrated under reduced pressure. The viscous residue was dissolved in methanol (400 mL) and cooled to 0 °C. Ammonium hydroxide (28-30% ammonia, 15 mL) was added dropwise and the reaction mixture was allowed to warm to 23 °C. After 14 h, another portion of ammonium hydroxide (28–30% ammonia, 10 mL) was added dropwise via syringe. After 23 h, the reaction mixture was cooled to -78 °C and filtered. The white precipitate was washed with methanol (2 × 50 mL), which had been cooled to 0 °C in an ice-water bath, and was dried under reduced pressure to afford diketopiperazine (-)-10 (4.63 g, 70.6%, 2-steps) as a fine white powder. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>, 20 °C):

<sup>13</sup>C NMR (125.8 MHz, DMSO-*d*<sub>6</sub>, 20 °C):

δ 8.11 (s, 1H, N<sub>10</sub>H), 8.07 (d,  $J = 8.0, 1H, C_8H$ ), 7.67 (d,  $J = 7.7, 2H, C_5H$ ), 7.60 (s, 1H, C<sub>2</sub>H), 7.52 (d,  $J = 7.7, 1H, Ph_{Cbz}$ -*o*-H), 7.46–7.37 (m, 3H, Ph<sub>Cbz</sub>-H), 7.33 (app-dt,  $J = 0.9, 7.7, 1H, C_7H$ ), 7.27 (app-dt,  $J = 0.9, 7.5, 1H, C_6H$ ), 5.46 (s, 2H, Ph<sub>Cbz</sub>CH<sub>2</sub>), 4.42 (app-t,  $J = 5.2, 1H, C_{11}H$ ), 4.12 (app-t,  $J = 8.2, 1H, C_{15}H$ ), 3.40–3.26 (m, 2H, C<sub>19</sub>H), 3.22 (dd,  $J = 4.2, 15.3, 1H, C_{12}H_a$ ), 3.05 (dd,  $J = 5.6, 15.3, 1H, C_{12}H_b$ ), 2.09–1.99 (m, 1H, C<sub>17</sub>H<sub>a</sub>), 1.78–1.62 (m, 2H, C<sub>18</sub>H), 1.62–1.48 (m, 1H, C<sub>17</sub>H<sub>b</sub>).

δ 169.3 (**C**<sub>16</sub>), 165.2 (**C**<sub>13</sub>), 150.1 (**C**<sub>Cbz</sub>=**O**), 135.4 (**Ph**<sub>Cbz</sub>-*ipso*-**C**), 134.6 (**C**<sub>9</sub>), 130.4 (**C**<sub>4</sub>), 128.6 (**Ph**<sub>Cbz</sub>-**C**), 128.5 (**Ph**<sub>Cbz</sub>-**C**), 128.3 (**Ph**<sub>Cbz</sub>-**C**), 124.5 (**C**<sub>7</sub>), 124.0 (**C**<sub>2</sub>), 122.7 (**C**<sub>6</sub>), 119.7 (**C**<sub>5</sub>), 116.9

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	$(C_3)$ , 114.5 $(C_8)$ , 68.2 $(Ph_{Cbz}CH_2)$ , 58.5 $(C_{15})$ , 54.3 $(C_{11})$ , 44.7 $(C_{19})$ , 27.8 $(C_{17})$ , 24.9 $(C_{12})$ , 22.0 $(C_{18})$ .
FTIR (thin film) cm <sup>-1</sup> :	3233 (br, m), 1732 (s), 1667 (s), 1455 (m), 1399 (m), 1249 (m).
HRMS (DART) $(m/z)$ :	calc'd for C <sub>24</sub> H <sub>24</sub> N <sub>3</sub> O <sub>4</sub> [M+H] <sup>+</sup> : 418.1761, found: 418.1764.
$\left[\alpha\right]_{D}^{24}$ :	-49 (c = 0.23, DMSO).
TLC (40% acetone in dichloromethane), Rf:	0.45 (UV, CAM).

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#### Tetracyclic bromide B (+)-12:

Pyridinium tribromide (461 mg, 1.44 mmol, 1.20 equiv) was added to a suspension of diketopiperazine (–)-10 (500 mg, 1.20 mmol, 1 equiv) in 2,2,2-trifluoroethanol (10 mL) at 23 °C. After 1 h, the reaction mixture was poured into a saturated aqueous sodium thiosulfate solution (60 mL) and extracted with dichloromethane (60 mL). The resulting aqueous layer was further extracted with dichloromethane ( $2 \times 30 \text{ mL}$ ) and the combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 75% ethyl acetate in hexanes) to afford tetracyclic bromide (+)-12 (304 mg, 51.1%) as a beige foam. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 7.64 (d, $J = 8.2$ , 1H, C <sub>8</sub> <b>H</b> ), 7.48–7.39 (m, 1H, C <sub>5</sub> <b>H</b> ), 7.48–7.39 (m, 2H, Ph <sub>Cbz</sub> - <i>o</i> - <b>H</b> ), 7.34 (app-t, $J = 7.0$ , 2H, Ph <sub>Cbz</sub> - <i>m</i> - <b>H</b> ), 7.29 (t, $J = 7.3$ , 1H, Ph <sub>Cbz</sub> - <i>p</i> - <b>H</b> ), 7.23 (app-t, $J = 7.6$ , 1H, C <sub>7</sub> <b>H</b> ), 7.07 (app-t, $J = 7.3$ , 1H, C <sub>6</sub> <b>H</b> ), 6.29 (s, 1H, C <sub>2</sub> <b>H</b> ), 5.41 (d, $J = 12.2$ , 1H, Ph <sub>Cbz</sub> C <b>H</b> <sub>a</sub> ), 5.30 (d, $J = 12.2$ , 1H, Ph <sub>Cbz</sub> C <b>H</b> <sub>b</sub> ), 4.35 (dd, $J = 3.4$ , 14.3, 1H, C <sub>12</sub> <b>H</b> <sub>a</sub> ), 4.00 (app-t, $J = 7.9$ , 1H, C <sub>15</sub> <b>H</b> ), 3.78 (dd, $J = 3.4$ , 14.3, 1H, C <sub>12</sub> <b>H</b> <sub>a</sub> ), 3.37–3.27 (m, 1H, C <sub>19</sub> <b>H</b> <sub>a</sub> ), 3.18–3.07 (m, 1H, C <sub>19</sub> <b>H</b> <sub>b</sub> ), 3.03 (dd, $J = 10.1$ , 14.0, 1H, C <sub>12</sub> <b>H</b> <sub>b</sub> ), 2.21–2.09 (m, 1H, C <sub>17</sub> <b>H</b> <sub>a</sub> ), 2.09–1.96 (m, 1H, C <sub>17</sub> <b>H</b> <sub>b</sub> ), 1.80, 1.69 (m, 2H, C, <b>H</b> )
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\begin{split} & (\mathbf{C}_{17}\mathbf{H}_{b}), 1.80-1.68 \text{ (m, 2H, } \mathbf{C}_{18}\mathbf{H}). \\ & \delta 167.2 \text{ (C}_{13}), 164.3 \text{ (C}_{16}), 152.8 \text{ (C}_{Cbz}=O), 139.4 \\ & (\mathbf{C}_{9}), 135.8 \text{ (Ph}_{Cbz}\text{-}ipso\text{-}C), 132.7 \text{ (C}_{4}), 130.8 \text{ (C}_{7}), \\ & 128.6 \text{ (Ph}_{Cbz}\text{-}C), 128.3 \text{ (Ph}_{Cbz}\text{-}C), 128.2 \text{ (Ph}_{Cbz}\text{-}C), \\ & 124.9 \text{ (C}_{5}), 124.7 \text{ (C}_{6}), 116.6 \text{ (C}_{8}), 84.6 \text{ (C}_{2}), 68.3 \\ & (\text{Ph}_{Cbz}\text{CH}_{2}), 60.6 \text{ (C}_{3}), 60.5 \text{ (C}_{15}), 59.2 \text{ (C}_{11}), 45.1 \\ & (\mathbf{C}_{19}), 37.5 \text{ (C}_{12}), 27.3 \text{ (C}_{17}), 23.3 \text{ (C}_{18}). \end{split}$
FTIR (thin film) $cm^{-1}$ :	1722 (s), 1679 (s), 1479 (m), 1410 (m), 751 (m).
HRMS (ESI) $(m/z)$ :	calc'd for C <sub>24</sub> H <sub>23</sub> BrN <sub>3</sub> O <sub>4</sub> [M+H] <sup>+</sup> : 496.0866, found: 496.0877.
$[\alpha]_{D}^{23}$ :	+63 ( $c = 0.11$ , CHCl <sub>3</sub> ).
TLC (75% ethyl acetate in hexanes), Rf:	0.18 (UV, CAM).

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#### (-)-N<sup>in</sup>, N<sup>in</sup>'-Dicarboxybenzyl Naseseazine B (16):

Silver(I) hexafluoroantimonate (69.2 mg, 202 µmol, 5.00 equiv) was added as a solid to a solution of tetracyclic bromide (+)-**12** (20.0 mg, 40.3 µmol, 1 equiv), potassium 6-trifluoroborateindole (-)-**33** (31.6 mg, 60.4 µmol, 1.50 equiv), and 18-crown-6 (53.3 mg, 202 µmol, 5.00 equiv) in nitroethane (2 mL) at 23 °C. After 1 h, aqueous hydrogen chloride (2 N, 2 mL) was added. After stirring for 5 min, the reaction mixture was diluted with ethyl acetate (60 mL) and washed with brine (3 × 60 mL). The organic layer was dried over anhydrous sodium sulfate, was filtered, and was concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: gradient, 5% methanol in ethyl acetate→5% methanol in dichloromethane) to afford (-)- $N^{in}$ , $N^{int}$ -dicarboxybenzyl naseseazine B (**16**) (16.7 mg, 49.8%) as a white solid. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	$\delta$ 8.14 (br-s, 1H, C <sub>8</sub> · <b>H</b> ), 7.66 (d, J = 8.0, 1H, C <sub>8</sub> <b>H</b> ),
	7.49 (s, 1H, $C_2$ H), 7.46–7.32 (m, 1H, $C_5$ H),
	7.46–7.32 (m, 7H, Ph <sub>Cbz</sub> -H), 7.32–7.25 (m, 1H,
	$C_5$ <b>H</b> ), 7.32–7.25 (m, 3H, Ph <sub>Cbz</sub> - <b>H</b> ), 7.18 (app-t, $J =$
	7.5, 1H, $C_7$ <b>H</b> ), 7.07–6.97 (m, 1H, $C_6$ <b>H</b> ), 7.07–6.97
	$(m, 1H, C_6 H), 6.38 (s, 1H, C_2 H), 5.82 (s, 1H,$
	$N_{10}$ <b>H</b> ), 5.41–5.22 (m, 4H, Ph <sub>Cbz</sub> C <b>H</b> <sub>2</sub> ), 4.37 (d, <i>J</i> =
	7.1, 1H, $C_{11}$ H), 4.26 (dd, $J = 2.3, 10.2, 1H, C_{11}$ H),
	4.04 (app-t, $J = 8.0, 1H, C_{15}H$ ), 4.00 (app-t, $J =$
	7.9, 1H, $C_{15}$ H), 3.65–3.47 (m, 2H, $C_{19}$ H),
	$3.65-3.47 (m, 1H, C_{12}H_a), 3.65-3.47 (m, 1H,$
	$C_{12}H_{a}$ , 3.41–3.31 (m, 1H, $C_{19}H_{a}$ ), 3.20–3.10 (m,
	1H, $C_{19}H_b$ ), 2.91–2.78 (m, 1H, $C_{12}H_b$ ), 2.91–2.78
	$(m, 1H, C_{12}H_b), 2.30-2.23 (m, 1H, C_{17}H_a),$
	$2.23-2.15 (m, 1H, C_{17}H_a), 2.14-2.04 (m, 1H,$
	$C_{17}H_b$ ), 2.03–1.90 (m, 1H, $C_{17}H_b$ ), 2.03–1.90 (m,
	1H, C <sub>18</sub> <b>H</b> <sub>a</sub> ), 1.90–1.66 (m, 2H, C <sub>18</sub> <b>H</b> ), 1.90–1.66
	$(\mathbf{m}, \mathbf{1H}, \mathbf{C}_{18}\mathbf{H}_{\mathrm{b}}).$
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\delta$ 169.5 ( <b>C</b> <sub>16</sub> ), 167.9 ( <b>C</b> <sub>16</sub> ), 165.5 ( <b>C</b> <sub>13</sub> ), 165.1 ( <b>C</b> <sub>13</sub> ),
	$153.2 (\mathbf{C}_{Cbz}=\mathbf{O}), 150.5 (\mathbf{C}_{Cbz}=\mathbf{O}), 139.8 (\mathbf{C}_{0}), 139.5$
	$(\mathbf{C}_{7'}), 136.4 (Ph_{Cbz}-ipso-\mathbf{C}), 135.0 (Ph_{Cbz}-ipso-\mathbf{C}),$
	$135.0 (C_4), 135.0 (C_9), 129.1 (C_7), 129.1 (C_4),$
	129.1 (Ph <sub>Cbz</sub> -C), 129.0 (Ph <sub>Cbz</sub> -C), 129.0 (Ph <sub>Cbz</sub> -C),
	128.6 (Ph <sub>Cbz</sub> -C), 128.3 (Ph <sub>Cbz</sub> -C), 128.2 (Ph <sub>Cbz</sub> -C),
	$125.5 (C_5), 124.9 (C_2), 124.4 (C_6), 121.4 (C_6),$

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119.5 (**C**<sub>5'</sub>), 116.8 (**C**<sub>8</sub>), 115.8 (**C**<sub>3'</sub>), 113.3 (**C**<sub>8'</sub>),

83.6 ( $C_2$ ), 69.3 ( $Ph_{Cbz}CH_2$ ), 68.2 ( $Ph_{Cbz}CH_2$ ), 61.1 ( $C_{15}$ ), 59.7 ( $C_{11}$ ), 59.4 ( $C_{15'}$ ), 59.2 ( $C_3$ ), 54.2 ( $C_{11'}$ ),

	45.7 ( $\mathbf{C}_{19}$ ), 45.2 ( $\mathbf{C}_{19}$ ), 35.4 ( $\mathbf{C}_{12}$ ), 28.5 ( $\mathbf{C}_{17}$ ), 27.6 ( $\mathbf{C}_{17}$ ), 26.6 ( $\mathbf{C}_{12}$ ), 23.6 ( $\mathbf{C}_{18}$ ), 22.8 ( $\mathbf{C}_{18}$ ).
FTIR (thin film) cm <sup>-1</sup> :	1721 (s), 1679 (s), 1399 (s), 1249 (m), 751 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{48}H_{45}N_6O_8 [M+H]^+$ : 833.3293, found: 833.3283.
$\left[\alpha\right]_{D}^{25}$ :	$-40 (c = 0.18, \text{CHCl}_3).$
TLC (5% methanol in ethyl acetate), Rf:	0.09 (UV, CAM).

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#### (+)-Naseseazine B (2):

Palladium on charcoal (10% w/w, 2.1 mg, 1.98 µmol, 0.25 equiv) was added to a solution of dicarboxybenzyl naseseazine B (-)-(16) (6.6 mg, 7.92 µmol, 1 equiv) in acetic acid (1 mL) at 23 °C. A stream of hydrogen gas was passed through the solution for 2 min by discharge of a balloon equipped with a needle extending into the reaction mixture. After stirring the solution for 3.5 h under an atmosphere of hydrogen gas, the solution was filtered over Celite. The solids were further extracted with methanol and the combined filtrates were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 5% methanol, 45% tetrahydrofuran, 50% dichloromethane) to afford (+)-naseseazine B (2) (3.6 mg, 80%) as a white solid. Structural assignments were made using additional information from gCOSY, HSQC, HMBC, and ROESY experiments.

<sup>1</sup> H NMR (500 MHz, methanol- $d_{1}$ , 20 °C):	$\delta$ 7.57 (d. J = 7.4, 1H, C <sub>5</sub> <b>H</b> ), 7.41 (d. J = 1.3, 1H,
	$C_{\rm o}$ <b>H</b> ), 7.12 (s. 1H, $C_{\rm o}$ <b>H</b> ), 7.06 (app-dt, $J = 1.2, 7.9$ .
	1H. $C_{2}$ H), 7.01 (dd, $J = 1.6, 8.5, 1$ H, $C_{4}$ H), 6.83
	$(d, J = 7.2, 1H, C_{e}H), 6.69 (d, J = 7.7, 1H, C_{e}H),$
	6.68 (app-t, $J = 7.3$ , 1H, C <sub>2</sub> H), 5.84 (s, 1H, C <sub>2</sub> H),
	4.71 (app-t, $J = 8.6, 1H, C_{11}H$ ), 4.39 (app-t, $J =$
	4.6, 1H, $C_{11}$ H), 4.29 (app-t, $J = 7.7, 1H, C_{15}$ H),
	$3.98 (ddd, J = 1.4, 6.3, 10.7, 1H, C_{15}H), 3.52-3.37$
	$(m, 2H, C_{19}H), 3.52-3.37 (m, 1H, C_{19}H_{a}),$
	$3.35-3.21$ (m, 2H, $C_{12}$ H), $3.35-3.21$ (m, 1H,
	$C_{12}H_a$ , 3.35–3.21 (m, 1H, $C_{19}H_b$ ), 2.57 (dd, $J =$
	10.2, 13.7, 1H, $C_{12}H_b$ ), 2.30–2.22 (m, 1H, $C_{17}H_a$ ),
	$2.15-2.04 (m, 1H, C_{17}H_b), 2.04-1.87 (m, 2H,$
	$C_{18}$ <b>H</b> ), 2.04–1.87 (m, 1H, $C_{17}$ <b>H</b> <sub>a</sub> ), 1.73–1.60 (m,
	1H, $C_{18}$ <b>H</b> <sub>a</sub> ), 1.49–1.40 (m, 1H, $C_{18}$ <b>H</b> <sub>b</sub> ), 0.97–0.87
	$(m, 1H, C_{17'}H_b).$
<sup>1</sup> H NMR (500 MHz, DMSO- $d_6$ , 20 °C):	$\delta$ 10.81 (s, 1H, N <sub>1</sub> H), 7.71 (s, 1H, N <sub>10</sub> H), 7.57 (d,
	$J = 8.4, 1H, C_{\circ}H$ , 7.30 (d, $J = 1.3, 1H, C_{\circ}H$ ), 7.19
	$(d, J = 2.2, 1H, C_2H), 7.00 \text{ (app-dt, } J = 1.3, 7.9,$
	1H, $C_7$ H), 6.99 (d, $J = 8.6, 1H, C_6$ H), 6.82–6.76
	$(m, 1H, C_5H), 6.82-6.76 (m, 1H, N_1H), 6.60 (d, J)$
	$= 7.6, 1H, C_8H$ , 6.58 (app-dt, $J = 1.0, 7.5, 1H$ ,
	$C_6$ <b>H</b> ), 5.68 (d, $J = 3.2, 1$ <b>H</b> , $C_2$ <b>H</b> ), 4.72 (app-t, $J =$
	8.2, 1H, $C_{11}$ <b>H</b> ), 4.35 (app-t, $J = 7.9$ , 1H, $\overline{C_{15}}$ <b>H</b> ),
	4.29 (app-t, $J = 5.1, 1$ H, C <sub>11</sub> H), 4.06 (app-t, $J =$

 $8.7, 1H, C_{15}H$ , 3.41-3.30 (m,  $2H, C_{19}H$ ), 3.41-

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	3.30 (m, 1H, $C_{19}H_a$ ), 3.29–3.24 (m, 1H, $C_{19}H_b$ ), 3.22 (dd, $J = 4.6$ , 14.6, 1H, $C_{12}H_a$ ), 3.13 (dd, $J =$ 7.4, 13.6, 1H, $C_{12}H_a$ ), 3.05 (dd, $J = 5.8$ , 14.8, 1H, $C_{12}H_b$ ), 2.36 (dd, $J = 10.5$ , 13.6, 1H, $C_{12}H_b$ ), 2.21– 2.11 (m, 1H, $C_{17}H_a$ ), 2.03–1.91 (m, 1H, $C_{17}H_a$ ), 2.03–1.91 (m, 1H, $C_{17}H_b$ ), 1.91–1.77 (m, 2H, $C_{18}H$ ), 1.75–1.56 (m, 2H, $C_{18}H$ ), 1.48–1.36 (m, 1H, $C_{17}H_b$ ).
<sup>13</sup> C NMR (125.8 MHz, methanol- <i>d</i> <sub>4</sub> , 20 °C):	$ \begin{split} &\delta \ 170.9 \ (\mathbf{C}_{16}), \ 170.3 \ (\mathbf{C}_{16}), \ 168.6 \ (\mathbf{C}_{13}), \ 167.5 \ (\mathbf{C}_{13}), \\ &149.3 \ (\mathbf{C}_{9}), \ 138.1 \ (\mathbf{C}_{9}), \ 137.2 \ (\mathbf{C}_{7}), \ 136.2 \ (\mathbf{C}_{4}), \\ &129.6 \ (\mathbf{C}_{7}), \ 127.8 \ (\mathbf{C}_{4}), \ 126.6 \ (\mathbf{C}_{2}), \ 125.1 \ (\mathbf{C}_{5}), \\ &120.6 \ (\mathbf{C}_{6}), \ 120.6 \ (\mathbf{C}_{5}), \ 119.8 \ (\mathbf{C}_{6}), \ 111.3 \ (\mathbf{C}_{8}), \\ &110.5 \ (\mathbf{C}_{8}), \ 109.8 \ (\mathbf{C}_{3}), \ 87.1 \ (\mathbf{C}_{2}), \ 61.9 \ (\mathbf{C}_{15}), \ 61.9 \\ &(\mathbf{C}_{3}), \ 61.7 \ (\mathbf{C}_{11}), \ 60.2 \ (\mathbf{C}_{15}), \ 57.3 \ (\mathbf{C}_{11}), \ 46.4 \ (\mathbf{C}_{19}), \\ &46.1 \ (\mathbf{C}_{19}), \ 39.7 \ (\mathbf{C}_{12}), \ 29.3 \ (\mathbf{C}_{17}), \ 29.3 \ (\mathbf{C}_{12}), \ 28.6 \\ &(\mathbf{C}_{17}), \ 24.4 \ (\mathbf{C}_{18}), \ 22.7 \ (\mathbf{C}_{18'}). \end{split} $
<sup>13</sup> C NMR (125.8 MHz, DMSO- <i>d</i> <sub>6</sub> , 20 °C):	$ \begin{split} \delta \ &169.1 \ (\mathbf{C}_{16}), \ &167.9 \ (\mathbf{C}_{16}), \ &165.9 \ (\mathbf{C}_{13}), \ &165.5 \ (\mathbf{C}_{13}), \\ &148.1 \ (\mathbf{C}_{9}), \ &135.9 \ (\mathbf{C}_{9}), \ &135.6 \ (\mathbf{C}_{7}), \ &134.6 \ (\mathbf{C}_{4}), \\ &127.9 \ (\mathbf{C}_{7}), \ &126.1 \ (\mathbf{C}_{4}), \ &125.0 \ (\mathbf{C}_{2}), \ &123.4 \ (\mathbf{C}_{5}), \\ &119.1 \ (\mathbf{C}_{5}), \ &117.9 \ (\mathbf{C}_{6}), \ &117.8 \ (\mathbf{C}_{6}), \ &109.2 \ (\mathbf{C}_{8}), \\ &109.2 \ (\mathbf{C}_{8}), \ &109.2 \ (\mathbf{C}_{3}), \ &84.8 \ (\mathbf{C}_{2}), \ &59.9 \ (\mathbf{C}_{15}), \ &59.8 \\ &(\mathbf{C}_{3}), \ &59.5 \ (\mathbf{C}_{11}), \ &58.4 \ (\mathbf{C}_{15}), \ &55.2 \ (\mathbf{C}_{11}), \ &44.6 \ (\mathbf{C}_{19}), \\ &44.6 \ (\mathbf{C}_{19}), \ &38.7 \ (\mathbf{C}_{12}), \ &27.6 \ (\mathbf{C}_{17}), \ &27.0 \ (\mathbf{C}_{17}), \ &25.6 \\ &(\mathbf{C}_{12}), \ &22.9 \ (\mathbf{C}_{18}), \ &21.9 \ (\mathbf{C}_{18}). \end{split} $
FTIR (thin film) cm <sup>-1</sup> :	3295 (br, s), 2954 (w), 2882 (w), 1661 (s), 1426 (m), 1343 (w), 1312 (w), 1245 (w), 750 (m).
UV (CH <sub>3</sub> OH) $\lambda_{max}$ (log $\epsilon$ ):	201 (4.95), 203 (4.98), 204 (4.92), 229 (4.67), 285 (3.92).
HRMS (DART) $(m/z)$ :	calc'd for $C_{32}H_{33}N_6O_4$ [M+H] <sup>+</sup> : 565.2558, found: 565.2546.
$[\alpha]_{D}^{24}$ :	+101 ( $c = 0.23$ , CH <sub>3</sub> OH). <sup>10</sup>
TLC (5% methanol, 45% tetrahydrofuran, 50% d	lichloromethane), Rf: 0.19 (UV, CAM).

<sup>&</sup>lt;sup>10</sup> Literature value:  $[\alpha]_D^{23} = +95$  (c 0.08, CH<sub>3</sub>OH), Raju, R.; Piggott, A. M.; Conte, M.; Aalbersberg, W. G. L.; Feussner, K.; Capon, R. J. *Org. Lett.* **2009**, *11*, 3862.

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(-)-N<sup>in</sup>, N<sup>in</sup>'-Dicarboxybenzyl *iso*-Naseseazine A (13):

Silver(I) hexafluoroantimonate (73.0 mg, 213 µmol, 5.00 equiv) was added as a solid to a solution of tetracyclic bromide (+)-**11** (20.0 mg, 42.5 µmol, 1 equiv) and proline diketopiperazine (-)-**10** (26.6 mg, 63.8 µmol, 1.50 equiv) in nitroethane (2 mL) at 23 °C. After 2 h, the reaction mixture was diluted with dichloromethane (60 mL) and washed with brine (60 mL). The resulting aqueous layer was extracted with dichloromethane ( $2 \times 30 \text{ mL}$ ) and the combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: gradient, 5% methanol in ethyl acetate→5% methanol in dichloromethane) to afford a regioisomeric mixture of (-) $N^{in}$ , $N^{int}$ -dicarboxybenzyl naseseazine A (**15**) and (-)- $N^{in}$ , $N^{int}$ -dicarboxybenzyl *iso*-naseseazine A (**13**) (18.1 mg, 53%, (-)-**15**:(-)-**13**, 1.4:1) as a white solid.

Regioisomers (-)-15 and (-)-13 were separated for the purpose of full and independent characterization by preparative HPLC [Waters X-Bridge preparative HPLC column, C18, 5  $\mu$ m, 19 × 250 mm; 20.0 mL/min; gradient, 55% $\rightarrow$ 65% acetonitrile in water, 15 min;  $t_{\rm R}((-)$ -15) = 4.6 min,  $t_{\rm R}((-)$ -13) = 5.7 min]. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

#### (-)-N<sup>in</sup>,N<sup>in</sup>'-Dicarboxybenzyl iso-Naseseazine A (13):

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, 20 °C):

 $\delta 8.03$  (s, 1H, C<sub>8</sub>H), 7.65 (d, J = 8.2, 1H, C<sub>8</sub>H), 7.49 (s, 1H,  $C_{5}$ H), 7.43 (d, J = 7.8, 2H,  $Ph_{Chz}$ -o-H),  $7.41-7.25 \text{ (m, 8H, Ph}_{Chz}$ -**H**), 7.41-7.25 (m, 1H, $C_{2}H$ , 7.41–7.25 (m, 1H,  $C_{5}H$ ), 7.18 (app-dt, J =1.1, 8.0, 1H,  $C_7$ H), 7.13 (d, J = 8.9, 1H,  $C_7$ H), 7.01  $(app-dt, J = 0.8, 7.5, 1H, C_6H), 6.42 (s, 1H, C_2H),$ 5.85 (s, 1H,  $N_{10}$ H), 5.81 (s, 1H,  $N_{14}$ H), 5.40 (s, 2H,  $Ph_{Cbz}$ CH<sub>2</sub>), 5.37 (d, J = 12.4, 1H,  $Ph_{Cbz}$ CH<sub>a</sub>), 5.31  $(d, J = 12.4, 1H, Ph_{Cbz}CH_b), 4.41 (dd, J = 3.5, 9.8),$ 1H,  $C_{11}$ H), 4.23 (dd, J = 2.7, 9.4, 1H,  $C_{11}$ H), 4.03-3.95 (m, 1H, C<sub>15</sub>H), 4.03-3.95 (m, 1H,  $C_{15}$ **H**), 3.59–3.38 (m, 1H,  $C_{12}$ **H**<sub>a</sub>), 3.59–3.38 (m, 2H,  $C_{19}$ H), 3.59–3.38 (m, 1H,  $C_{12}$ H<sub>a</sub>), 2.94 (dd, J  $= 10.0, 13.8, 1H, C_{12}H_{\rm b}), 2.84 \, (dd, J = 9.9, 15.3)$ 1H,  $C_{12}H_{h}$ ), 2.30–2.19 (m, 1H,  $C_{17}H_{a}$ ), 1.94–1.77  $(m, 2H, C_{18}H), 1.94-1.77 (m, 1H, C_{17}H_{b}), 1.33 (d,$  $J = 6.9, 3H, C_{17}H$ ).

<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$ \begin{split} &\delta \ 169.5 \ (\mathbf{C}_{16}), \ 169.5 \ (\mathbf{C}_{13}), \ 167.8 \ (\mathbf{C}_{16}), \ 165.0 \ (\mathbf{C}_{13}), \\ &153.3 \ (\mathbf{C}_{Cbz}=\mathbf{O}), \ 150.5 \ (\mathbf{C}_{Cbz}=\mathbf{O}), \ 139.7 \ (\mathbf{C}_{9}), \ 137.2 \\ &(\mathbf{C}_{6}), \ 136.3 \ (\mathbf{C}_{Cbz}-ipso-\mathbf{C}), \ 135.4 \ (\mathbf{C}_{4}), \ 135.1 \ (\mathbf{C}_{Cbz}-ipso-\mathbf{C}), \ 134.9 \ (\mathbf{C}_{9}), \ 130.2 \ (\mathbf{C}_{4}), \ 129.2 \ (\mathbf{C}_{7}), \ 129.1 \\ &(\mathrm{Ph}_{Cbz}-\mathbf{C}), \ 129.0 \ (\mathrm{Ph}_{Cbz}-\mathbf{C}), \ 128.8 \ (\mathrm{Ph}_{Cbz}-\mathbf{C}), \ 128.7 \\ &(\mathrm{Ph}_{Cbz}-\mathbf{C}), \ 128.3 \ (\mathrm{Ph}_{Cbz}-\mathbf{C}), \ 128.3 \ (\mathrm{Ph}_{Cbz}-\mathbf{C}), \ 128.3 \\ &(\mathbf{C}_{5'}), \ 125.1 \ (\mathbf{C}_{5}), \ 124.5 \ (\mathbf{C}_{6}), \ 123.3 \ (\mathbf{C}_{7'}), \ 117.1 \\ &(\mathbf{C}_{8}), \ 116.3 \ (\mathbf{C}_{8'}), \ 116.0 \ (\mathbf{C}_{3'}), \ 115.9 \ (\mathbf{C}_{2}), \ 83.6 \ (\mathbf{C}_{2}), \\ &69.3 \ (\mathrm{Ph}_{Cbz}\mathbf{CH}_2), \ 68.2 \ (\mathrm{Ph}_{Cbz}\mathbf{CH}_2), \ 59.3 \ (\mathbf{C}_{15'}), \ 58.6 \\ &(\mathbf{C}_{11}), \ 58.5 \ (\mathbf{C}_{3}), \ 54.2 \ (\mathbf{C}_{11'}), \ 52.0 \ (\mathbf{C}_{15}), \ 45.6 \ (\mathbf{C}_{19}), \\ &35.2 \ (\mathbf{C}_{12}), \ 28.4 \ (\mathbf{C}_{17'}), \ 26.8 \ (\mathbf{C}_{12'}), \ 22.7 \ (\mathbf{C}_{18'}), \ 15.3 \ (\mathbf{C}_{17}). \end{split}$
FTIR (thin film) cm <sup>-1</sup> :	3251 (br, m), 1720 (s), 1688 (s), 1401 (m), 1310 (m), 751 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{46}H_{43}N_6O_8$ [M+H] <sup>+</sup> : 807.3137 found: 807.3115.
$\left[\alpha\right]_{D}^{24}$ :	$-20 (c = 0.26, \text{CHCl}_3).$
TLC (8% methanol in ethyl acetate), Rf:	0.06 (UV, CAM).

# (-)-N<sup>in</sup>,N<sup>in</sup>'-Dicarboxybenzyl Naseseazine A (15):

Please see page S14 for the full characterization data for  $(-)-N^{in}, N^{in}$ -dicarboxybenzyl naseseazine A (15).

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#### (+)-iso-Naseseazine A (17):

Palladium on charcoal (10% w/w, 6.0 mg, 5.67  $\mu$ mol, 0.25 equiv) was added to a solution of (–)- $N^{in}$ , $N^{in}$ -dicarboxybenzyl *iso*-naseseazine A (**13**) (18.3 mg, 22.7  $\mu$ mol, 1 equiv) in acetic acid (1 mL) at 23 °C. A stream of hydrogen gas was passed through the solution for 2 min by discharge of a balloon equipped with a needle extending into the reaction mixture. After stirring the solution for 20 h under an atmosphere of hydrogen gas, the solution was filtered over Celite. The solids were further extracted with methanol and the combined filtrates were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 7.5% methanol, 42.5% tetrahydrofuran, 50% dichloromethane) to afford (+)-*iso*-naseseazine A (**17**) (6.4 mg, 52.4%) as a white solid. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

$\delta$ 7.71 (s, 1H, C <sub>5</sub> H), 7.32 (d, J = 8.6, 1H, C <sub>8</sub> H),
7.10 (s, 1H, $C_2$ , H), 7.06 (app-dt, $J = 1.0, 7.7, 1$ H,
$C_7$ <b>H</b> ), 7.01 (dd, $J = 1.7, 8.6, 1$ H, $C_7$ <b>H</b> ), 6.90 (d, $J =$
7.2, 1H, C <sub>5</sub> H), 6.72–6.65 (m, 1H, C <sub>8</sub> H), 6.72–6.65
$(m, 1H, C_6H), 5.87 (s, 1H, C_2H), 4.97-4.78 (m,$
1H, $C_{11}$ H), 4.38 (app-t, $J = 4.1, 1$ H, $C_{11}$ H), 4.23
$(dq, J = 1.3, 7.0, 1H, C_{15}H), 3.88 (ddd, J = 1.3, 6.1,$
11.0, 1H, $C_{15}$ H), 3.44 (dd, $J = 4.6$ , 14.6, 1H,
$C_{12}H_a$ , 3.34 (dd, $J = 7.2$ , 13.6, 1H, $C_{12}H_a$ ),
$3.25-3.15 \text{ (m, 1H, C}_{12}\mathbf{H}_{b}\text{)}, 3.25-3.15 \text{ (m, 1H,}$
$C_{19}$ , $H_a$ ), 3.06 (app-dt, $J = 4.1, 10.3, 1H, C_{19}$ , $H_b$ ),
2.52 (dd, $J = 10.7, 13.7, 1H, C_{12}H_{b}$ ), 1.89–1.80 (m,
1H, $C_{17}$ , $H_a$ ), 1.62–1.48 (m, 1H, $C_{18}$ , $H_a$ ), 1.40 (d, $J =$
$6.9, 3H, C_{17}H$ ), $1.22-1.12$ (m, $1H, C_{18}H_b$ ),
$0.59-0.47 (m, 1H, C_{17}H).$
$\delta$ 173.1 ( <b>C</b> <sub>13</sub> ), 171.0 ( <b>C</b> <sub>16</sub> ), 170.2 ( <b>C</b> <sub>16</sub> ), 167.3 ( <b>C</b> <sub>13</sub> ),
149.3 ( $C_0$ ), 136.8 ( $C_0$ ), 136.5 ( $C_4$ ), 134.2 ( $C_6$ ),
$129.5 (C_7), 128.6 (C_4), 126.8 (C_7), 125.1 (C_5),$
$123.2 (C_7), 120.6 (C_6), 117.5 (C_5), 113.0 (C_8),$
111.2 ( $C_8$ ), 109.8 ( $C_3$ ), 87.3 ( $C_2$ ), 61.5 ( $C_3$ ), 60.8
$(\mathbf{C}_{11}), 60.1 (\mathbf{C}_{15}), 57.9 (\mathbf{C}_{11}), 52.4 (\mathbf{C}_{15}), 45.9 (\mathbf{C}_{19}),$
$40.3 (C_{12}), 30.2 (C_{12}), 29.2 (C_{17}), 22.3 (C_{18}), 15.3$
$(C_{17}).$
3353 (br-s), 2924 (w), 1662 (s), 1466 (w), 1424
(m), 1345 (w), 1308 (m), 750 (m).

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# HRMS (DART) (m/z):calc'd for $C_{30}H_{31}N_6O_4 [M+H]^+$ : 539.2401,<br/>found: 539.2416. $[\alpha]_D^{-24}$ :+110 (c = 0.18, CH<sub>3</sub>OH).

TLC (7.5% methanol, 42.5% tetrahydrofuran, 50% dichloromethane), Rf: 0.23 (UV, CAM).

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(-)-N<sup>in</sup>, N<sup>in</sup>'-Dicarboxybenzyl *iso*-Naseseazine B (14):

Silver(I) hexafluoroantimonate (69.2 mg, 202 µmol, 5.00 equiv) was added as a solid to a solution of tetracyclic bromide (+)-**12** (20.0 mg, 40.3 µmol, 1 equiv) and proline diketopiperazine (-)-**10** (25.2 mg, 60.4 µmol, 1.50 equiv) in nitroethane (2 mL) at 23 °C. After 2 h, the reaction mixture was diluted with dichloromethane (60 mL) and washed with brine (60 mL). The resulting aqueous layer was extracted with dichloromethane ( $2 \times 30 \text{ mL}$ ) and the combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: gradient, 5% methanol in ethyl acetate  $\rightarrow$ 7.5% methanol in dichloromethane) to afford a regioisomeric mixture of (-)- $N^{in}$ , $N^{int}$ -dicarboxybenzyl naseseazine B (**16**) and (-)- $N^{in}$ , $N^{int}$ -dicarboxybenzyl *iso*-naseseazine B (**14**) (15.7 mg, 46.8%, (-)-**16**:(-)-**14**, 1.4:1) as a white solid.

Regioisomers (-)-16 and (-)-14 were separated for the purpose of full and independent characterization by preparative HPLC [Waters X-Bridge preparative HPLC column, C18, 5  $\mu$ m, 19 × 250 mm; 20.0 mL/min; gradient, 55% $\rightarrow$ 65% acetonitrile in water, 15 min;  $t_{\rm R}((-)$ -16) = 5.5 min,  $t_{\rm R}((-)$ -14) = 6.6 min]. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

#### (-)-N<sup>in</sup>,N<sup>in</sup>'-Dicarboxybenzyl iso-Naseseazine B (14):

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, 20 °C):

 $\delta$  8.04 (br-s, 1H, C<sub>8</sub>H), 7.65 (d, J = 8.0, 1H, C<sub>8</sub>H), 7.48 (s, 1H, C<sub>2</sub>H), 7.43 (dd,  $J = 1.1, 7.6, 2H, Ph_{Chz^{-1}}$ **H**), 7.41–7.34 (m, 5H, Ph<sub>Cbz</sub>-**H**), 7.34–7.25 (m, 1H,  $C_5H$ ), 7.34–7.25 (m, 1H,  $C_5H$ ), 7.34–7.25 (m, 3H,  $Ph_{Chz}$ -**H**), 7.21–7.14 (m, 1H, C<sub>7</sub>**H**), 7.21–7.14 (m, 1H,  $C_7$ H), 7.04 (app-dt,  $J = 0.7, 7.5, 1H, C_6$ H), 6.42 (s, 1H, C<sub>2</sub>H), 5.70 (s, 1H, N<sub>10</sub>H), 5.40 (s, 2H,  $Ph_{Cbz}CH_2$ , 5.36 (d, J = 12.4, 1H,  $Ph_{Cbz}CH_a$ ), 5.30  $(d, J = 12.3, 1H, Ph_{Cbz}CH_{b}), 4.44 (dd, J = 3.5, 9.8),$ 1H,  $C_{11}$ H), 4.23 (dd, J = 2.3, 9.6, 1H,  $C_{11}$ H), 4.06  $(app-t, J = 7.9, 1H, C_{15}H), 4.00 (app-t, J = 7.5, 1H)$  $C_{15}H$ , 3.60 (dd,  $J = 3.8, 13.9, 1H, C_{12}H_{a}$ ),  $3.58-3.45 \text{ (m, 2H, C_{19}H)}, 3.43 \text{ (dd, } J = 3.0, 15.2,$  $1H, C_{12}H_{a}, 3.40-3.32 (m, 1H, C_{19}H_{a}), 3.20-3.11$  $(m, 1H, C_{19}H_{b}), 2.91 (dd, J = 10.0, 13.8, 1H)$  $C_{12}H_{\rm b}$ ), 2.85 (dd,  $J = 10.1, 15.3, 1H, C_{12}H_{\rm b}$ ),  $2.32-2.16 (m, 1H, C_{17}H_a), 2.32-2.16 (m, 1H,$  $C_{17}H_{a}$ , 2.16–2.04 (m, 1H,  $C_{17}H_{b}$ ), 1.98–1.65 (m,

 $2H, C_{18} \textbf{H}), 1.98 {-} 1.65 \ (m, 2H, C_{18} \textbf{H}), 1.98 {-} 1.65$ 

	$(m, 1H, C_{17'}H_b).$
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\begin{split} &\delta \ 169.4 \ (\mathbf{C}_{16}), \ 168.0 \ (\mathbf{C}_{16}), \ 165.5 \ (\mathbf{C}_{13}), \ 164.9 \ (\mathbf{C}_{13}), \\ &153.3 \ (\mathbf{C}_{\text{Cbz}}=\mathbf{O}), \ 150.5 \ (\mathbf{C}_{\text{Cbz}}=\mathbf{O}), \ 139.7 \ (\mathbf{C}_{9}), \ 137.3 \\ &(\mathbf{C}_{6}), \ 136.3 \ (\text{Ph}_{\text{Cbz}}\text{-}\textit{ipso}\text{-}\mathbf{C}), \ 135.3 \ (\mathbf{C}_{4}), \ 135.0 \ (\mathbf{C}_{9}), \\ &134.9 \ (\text{Ph}_{\text{Cbz}}\text{-}\textit{ipso}\text{-}\mathbf{C}), \ 130.1 \ (\mathbf{C}_{4}), \ 129.1 \ (\mathbf{C}_{7}), \ 129.1 \\ &(\text{Ph}_{\text{Cbz}}\text{-}\mathbf{C}), \ 129.0 \ (\text{Ph}_{\text{Cbz}}\text{-}\mathbf{C}), \ 128.8 \ (\text{Ph}_{\text{Cbz}}\text{-}\mathbf{C}), \ 128.7 \\ &(\text{Ph}_{\text{Cbz}}\text{-}\mathbf{C}), \ 128.2 \ (\text{Ph}_{\text{Cbz}}\text{-}\mathbf{C}), \ 128.2 \ (\text{Ph}_{\text{Cbz}}\text{-}\mathbf{C}), \ 128.3 \\ &(\mathbf{C}_{5}), \ 125.2 \ (\mathbf{C}_{2}), \ 124.5 \ (\mathbf{C}_{6}), \ 123.4 \ (\mathbf{C}_{7}), \ 116.8 \\ &(\mathbf{C}_{8}), \ 116.3 \ (\mathbf{C}_{8}), \ 115.9 \ (\mathbf{C}_{5}), \ 115.9 \ (\mathbf{C}_{3}), \ 83.5 \ (\mathbf{C}_{2}), \\ &69.3 \ (\text{Ph}_{\text{Cbz'}}\text{-}\mathbf{CH}_{2}), \ 68.2 \ (\text{Ph}_{\text{Cbz}}\text{-}\mathbf{CH}_{2}), \ 61.1 \ (\mathbf{C}_{15}), \ 59.7 \\ &(\mathbf{C}_{11}), \ 59.3 \ (\mathbf{C}_{15}), \ 58.7 \ (\mathbf{C}_{3}), \ 54.2 \ (\mathbf{C}_{11}), \ 45.6 \ (\mathbf{C}_{19}), \\ &45.2 \ (\mathbf{C}_{19}), \ 35.2 \ (\mathbf{C}_{12}), \ 28.4 \ (\mathbf{C}_{17}), \ 27.6 \ (\mathbf{C}_{17}), \ 26.8 \\ &(\mathbf{C}_{12}), \ 23.6 \ (\mathbf{C}_{18}), \ 22.7 \ (\mathbf{C}_{18}). \end{split}$
FTIR (thin film) cm <sup>-1</sup> :	3584 (m), 2954 (m), 1720 (s), 1677 (s), 1400 (s), 1254 (m), 750 (m).
HRMS (ESI) $(m/z)$ :	calc'd for C <sub>48</sub> H <sub>45</sub> N <sub>6</sub> O <sub>8</sub> [M+H] <sup>+</sup> : 833.3293, found: 833.3291.
$[\alpha]_{D}^{24}$ :	$-32 (c = 0.39, \text{CHCl}_3).$
TLC (8% methanol in ethyl acetate), Rf:	0.05 (UV, CAM).

# (-)-N<sup>in</sup>,N<sup>in</sup>'-Dicarboxybenzyl Naseseazine B (16):

Please see page S21 for the full characterization data for  $(-)-N^{in}, N^{in}$ -dicarboxybenzyl naseseazine B (16).

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#### (+)-iso-Naseseazine B (18):

Palladium on charcoal (10% w/w, 4.4 mg, 4.11 µmol, 0.25 equiv) was added to a solution of  $(-)-N^{in},N^{int}$ -dicarboxybenzyl *iso*-naseseazine B (**14**) (13.7 mg, 16.5 µmol, 1 equiv) in acetic acid (1 mL) at 23 °C. A stream of hydrogen gas was passed through the solution for 2 min by discharge of a balloon equipped with a needle extending into the reaction mixture. After stirring the solution for 18 h under an atmosphere of hydrogen gas, the solution was filtered over Celite. The solids were further extracted with methanol and the combined filtrates were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 7.5% methanol, 42.5% tetrahydrofuran, 50% dichloromethane) to afford (+)-*iso*-naseseazine B (**18**) (5.7 mg, 61.4%) as a white solid. Structural assignments were made using additional information from gCOSY, HSQC, HMBC, and nOe experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	$\delta$ 8.28 (s, 1H, N <sub>1</sub> · <b>H</b> ), 7.59 (s, 1H, C <sub>5</sub> · <b>H</b> ), 7.33 (d, J
	$= 8.6, 1H, C_8H$ , 7.14–7.09 (m, 1H, C <sub>2</sub> H),
	7.14–7.09 (m, 1H, $C_7$ <b>H</b> ), 7.06 (app-dt, $J = 0.9, 7.6$ ,
	1H, $C_7$ H), 6.84 (d, $J = 7.4$ , 1H, $C_5$ H), 6.68 (app-t, $J$
	$= 7.4, 1H, C_6H$ , 6.63 (d, $J = 8.0, 1H, C_8H$ ), 5.87
	$(d, J = 3.2, 1H, C_2H), 5.76 (s, 1H, N_{10}H), 5.29 (d, J)$
	$J = 3.1, 1H, N_1H, 4.67 (dd, J = 7.3, 9.5, 1H)$
	$C_{11}H$ , 4.33 (dd, $J = 3.0, 10.7, 1H, C_{11}H$ ), 4.18
	(app-t, $J = 8.2, 1$ H, C <sub>15</sub> H), 4.08 (app-t, $J = 7.8, 1$ H,
	$C_{15}H$ , 3.66 (dd, $J = 3.5, 14.9, 1H, C_{12}H_a$ ),
	3.63-3.43 (m, 2H, C <sub>19</sub> <b>H</b> ), $3.63-3.43$ (m, 2H,
	$C_{19}$ <b>H</b> ), 3.31 (dd, $J = 7.1, 13.9, 1$ H, $C_{12}$ <b>H</b> <sub>a</sub> ), 2.99
	$(dd, J = 10.5, 15.0, 1H, C_{12}H_b), 2.74 (dd, J = 10.5,$
	13.8, 1H, $C_{12}H_b$ ), 2.36–2.22 (m, 1H, $C_{17}H_a$ ),
	$2.36-2.22 \text{ (m, 1H, C}_{17}\mathbf{H}_{a}\text{)}, 2.22-2.08 \text{ (m, 1H,}$
	$C_{17}H_{b}$ ), 2.08–1.76 (m, 2H, $C_{18}H$ ), 2.08–1.76 (m,
	2H, $C_{18}$ H), 2.08–1.76 (m, 1H, $C_{17'}$ H <sub>b</sub> ).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\delta = 169.5 (\mathbf{C}_{12}) = 168.2 (\mathbf{C}_{12}) = 166.2 (\mathbf{C}_{12}) = 165.4 (\mathbf{C}_{12})$
	$147.3 (C_{e}) 135.8 (C_{e}) 134.7 (C_{e}) 134.3 (C_{e})$
	$128.7 (C_7), 127.0 (C_4), 124.4 (C_7), 124.3 (C_5), 128.7 (C_7), 127.0 (C_4), 124.4 (C_7), 124.3 (C_5), 124.3 (C_7), 124$
	$123.1 (\mathbf{C}_{77}), 119.9 (\mathbf{C}_{6}), 115.9 (\mathbf{C}_{57}), 112.4 (\mathbf{C}_$
	110.3 ( $C_{21}$ ), 109.8 ( $C_{2}$ ), 86.2 ( $C_{2}$ ), 60.8 ( $C_{11}$ ), 60.7
	$(\mathbf{C}_{15}), 60.3 (\mathbf{C}_{3}), 59.4 (\mathbf{C}_{15}), 55.2 (\mathbf{C}_{11'}), 45.6$
	$(\mathbf{C}_{19/19}), 45.4 (\mathbf{C}_{19/19}), 38.6 (\mathbf{C}_{12}), 28.5 (\mathbf{C}_{17}), 27.8$
	$(\mathbf{C}_{17}), 27.1 (\mathbf{C}_{12}), 23.5 (\mathbf{C}_{18}), 22.7 (\mathbf{C}_{18}).$

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FTIR (thin film) $cm^{-1}$ :	3306 (br, s), 2925 (w), 1660 (s), 1428 (m), 1341 (w), 1312 (w), 1215 (w), 750 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{32}H_{33}N_6O_4 [M+H]^+$ : 565.2558, found: 565.2544.
$\left[\alpha\right]_{D}^{24}$ :	+43 ( $c = 0.15$ , CHCl <sub>3</sub> ).

TLC (5% methanol, 45% tetrahydrofuran, 50% dichloromethane), Rf: 0.18 (UV, CAM).

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#### 3-(2-Thienyl)tetracycle (-)-19 and 3-(3-Thienyl)tetracycle (-)-20:

A solution of silver(I) hexafluoroantimonate (0.20 M, 425  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane was added via syringe to a solution of tetracyclic bromide (+)-**11** (20.0 mg, 42.5  $\mu$ mol, 1 equiv) and thiophene (6.7  $\mu$ L, 85  $\mu$ mol, 2.0 equiv) in dichloromethane (2 mL) at 23 °C. After 1 h, brine (200  $\mu$ L) was added to the solution. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 20% acetone in dichloromethane) to afford a mixture of regioisomeric 3-(thienyl)tetracycles (Run 1: 15.3 mg, 76.0%, (-)-**19**:(-)-**20**, 6.2:1; Run 2: 15.6 mg, 77.5%, (-)-**19**:(-)-**20**, 6.4:1) as a clear film.

Regioisomers (–)-19 and (–)-20 were separated for the purpose of full and independent characterization by preparative HPLC [Waters X-Bridge preparative HPLC column, C18, 5 µm, 19 × 250 mm; 20.0 mL/min; 50% acetonitrile in water;  $t_{\rm R}((-)-19) = 8.4$  min,  $t_{\rm R}((-)-20) = 8.7$  min]. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

#### 3-(2-Thienyl)tetracycle (-)-19:

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 7.68 (d, $J = 8.3$ , 1H, C <sub>8</sub> H), 7.38 (d, $J = 7.9$ , 2H, Ph <sub>Cbz</sub> - $o$ -H), 7.35–7.27 (m, 3H, Ph <sub>Cbz</sub> -H), 7.35–7.27 (m, 1H, C <sub>5</sub> H), 7.24 (app-dt, $J = 1.3$ , 7.8, 1H, C <sub>7</sub> H), 7.17 (dd, $J = 1.2$ , 5.1, 1H, C <sub>5</sub> H), 7.06 (app-dt, $J = 1.0$ , 7.6, 1H, C <sub>6</sub> H), 6.84 (dd, $J = 3.6$ , 5.1, 1H, C <sub>4</sub> H), 6.63 (dd, $J = 1.2$ , 3.6, 1H, C <sub>3</sub> H), 6.22 (s, 1H, C <sub>2</sub> H), 5.46 (s, 1H, N <sub>14</sub> H), 5.38 (d, $J = 12.3$ , 1H, Ph <sub>Cbz</sub> CH <sub>a</sub> ), 5.28 (d, $J = 12.3$ , 1H, Ph <sub>Cbz</sub> CH <sub>b</sub> ), 4.40 (dd, $J = 3.5$ , 14.0, 1H, C <sub>12</sub> H <sub>a</sub> ), 2.93 (dd, $J = 10.1$ , 13.9, 1H, C <sub>12</sub> H <sub>b</sub> ), 1.33 (d, $J = 6.9$ , 3H, C <sub>17</sub> H).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\begin{split} &\delta \ 169.3 \ (\mathbf{C}_{13}), \ 167.7 \ (\mathbf{C}_{16}), \ 153.2 \ (\mathbf{C}_{Cbz} = \mathbf{O}), \ 145.3 \\ &(\mathbf{C}_{2}), \ 139.9 \ (\mathbf{C}_{9}), \ 136.3 \ (\mathbf{Ph}_{Cbz} - ipso - \mathbf{C}), \ 134.6 \ (\mathbf{C}_{4}), \\ &129.6 \ (\mathbf{C}_{7}), \ 128.7 \ (\mathbf{Ph}_{Cbz} - \mathbf{C}), \ 128.3 \ (\mathbf{Ph}_{Cbz} - \mathbf{C}), \ 128.2 \\ &(\mathbf{Ph}_{Cbz} - \mathbf{C}), \ 127.3 \ (\mathbf{C}_{4}), \ 125.3 \ (\mathbf{C}_{5}), \ 124.9 \ (\mathbf{C}_{5}), \\ &124.9 \ (\mathbf{C}_{3}), \ 124.4 \ (\mathbf{C}_{6}), \ 117.1 \ (\mathbf{C}_{8}), \ 83.9 \ (\mathbf{C}_{2}), \ 68.1 \\ &(\mathbf{Ph}_{Cbz} \mathbf{CH}_{2}), \ 58.4 \ (\mathbf{C}_{11}), \ 56.3 \ (\mathbf{C}_{3}), \ 51.9 \ (\mathbf{C}_{15}), \ 35.5 \\ &(\mathbf{C}_{12}), \ 15.3 \ (\mathbf{C}_{17}). \end{split}$
FTIR (thin film) cm <sup>-1</sup> :	3260 (br, m), 1712 (s), 1689 (s), 1409 (m), 1311 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{26}H_{24}N_3O_4S [M+H]^+: 474.1482$ , found: 474.1470.
$\left[\alpha\right]_{D}^{24}$ :	$-35 (c = 0.33, \text{CHCl}_3).$

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TLC (ethyl acetate), Rf:

0.47 (UV, CAM).

## <u>3-(3-Thienyl)tetracycle (–)-20:</u>

$\begin{split} &\delta~7.66~(\mathbf{d},J=8.1,1\mathbf{H},\mathbf{C}_{8}\mathbf{H}),7.37~(\mathbf{dd},J=1.8,8.3,\\ &2\mathbf{H},\mathbf{Ph}_{\mathrm{Cbz}}\text{-}o\text{-}\mathbf{H}),7.35\text{-}7.30~(\mathbf{m},3\mathbf{H},\mathbf{Ph}_{\mathrm{Cbz}}\text{-}\mathbf{H}),7.29\\ &(\mathbf{d},J=7.3,1\mathbf{H},\mathbf{C}_{5}\mathbf{H}),7.26~(\mathbf{dd},J=3.0,5.1,1\mathbf{H},\\ &\mathbf{C}_{4}\mathbf{H}),7.21~(\mathrm{app}\text{-}\mathrm{dt},J=1.3,7.8,1\mathbf{H},\mathbf{C}_{7}\mathbf{H}),7.05\\ &(\mathrm{app}\text{-}\mathrm{dt},J=1.0,7.5,1\mathbf{H},\mathbf{C}_{6}\mathbf{H}),6.90~(\mathrm{dd},J=1.4,5.1,1\mathbf{H},\mathbf{C}_{5}\mathbf{H}),6.80~(\mathrm{dd},J=1.4,3.0,1\mathbf{H},\mathbf{C}_{2}\mathbf{H}),\\ &6.24~(\mathrm{s},1\mathbf{H},\mathbf{C}_{2}\mathbf{H}),5.54~(\mathrm{s},1\mathbf{H},\mathbf{N}_{14}\mathbf{H}),5.40~(\mathrm{d},J=12.4,1\mathbf{H},\mathbf{Ph}_{\mathrm{Cbz}}\mathbf{CH}_{a}),5.28~(\mathrm{d},J=12.3,1\mathbf{H},\\ &\mathbf{Ph}_{\mathrm{Cbz}}\mathbf{CH}_{b}),4.37~(\mathrm{dd},J=3.4,10.1,1\mathbf{H},\mathbf{C}_{11}\mathbf{H}),3.99\\ &(\mathrm{q},J=6.9,1\mathbf{H},\mathbf{C}_{15}\mathbf{H}),3.47~(\mathrm{dd},J=3.4,13.8,1\mathbf{H},\\ &\mathbf{C}_{12}\mathbf{H}_{a}),2.84~(\mathrm{dd},J=10.1,13.9,1\mathbf{H},\mathbf{C}_{12}\mathbf{H}_{b}),1.33\\ &(\mathrm{d},J=6.9,3\mathbf{H},\mathbf{C}_{17}\mathbf{H}). \end{split}$
$ \begin{split} &\delta \ 169.3 \ (\mathbf{C}_{13}), \ 167.7 \ (\mathbf{C}_{16}), \ 153.2 \ (\mathbf{C}_{Cbz} = \mathbf{O}), \ 142.3 \\ &(\mathbf{C}_{3'}), \ 139.9 \ (\mathbf{C}_{9}), \ 136.3 \ (\mathbf{Ph}_{Cbz} \text{-} \textit{ipso-C}), \ 134.8 \ (\mathbf{C}_{4}), \\ &129.3 \ (\mathbf{C}_{7}), \ 128.7 \ (\mathbf{Ph}_{Cbz} \text{-} \mathbf{C}), \ 128.3 \ (\mathbf{Ph}_{Cbz} \text{-} \mathbf{C}), \ 128.3 \\ &(\mathbf{Ph}_{Cbz} \text{-} \mathbf{C}), \ 127.5 \ (\mathbf{C}_{4'}), \ 125.5 \ (\mathbf{C}_{5'}), \ 124.8 \ (\mathbf{C}_{5}), \\ &124.4 \ (\mathbf{C}_{6}), \ 121.5 \ (\mathbf{C}_{2'}), \ 117.1 \ (\mathbf{C}_{8}), \ 83.0 \ (\mathbf{C}_{2}), \ 68.2 \\ &(\mathbf{Ph}_{Cbz} \mathbf{CH}_{2}), \ 58.4 \ (\mathbf{C}_{11}), \ 56.2 \ (\mathbf{C}_{3}), \ 51.9 \ (\mathbf{C}_{15}), \ 34.4 \\ &(\mathbf{C}_{12}), \ 15.3 \ (\mathbf{C}_{17}). \end{split} $
3268 (br, m), 1709 (s), 1689 (s), 1481 (m), 1409 (m).
calc'd for $C_{26}H_{24}N_3O_4S [M+H]^+: 474.1482$ , found: 474.1465.
$-17 (c = 0.21, \text{CHCl}_3).$
0.43 (UV, CAM).

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#### <u>3-(2-Thienyl)tetracycle (–)-19 and 3-(3-Thienyl)tetracycle (–)-20:</u>

A Schlenk tube was charged with tetracyclic bromide (+)-11 (20.0 mg, 42.5 µmol, 1 equiv), potassium 3-thiophenetrifluoroborate (16.2 mg, 85.1 µmol, 2.00 equiv), and 18-crown-6 (22.4 mg, 85.1 µmol, 2.00 equiv). Nitroethane (2 mL) was then introduced via syringe. Upon dissolution of all solid components, the reaction mixture was cooled to -78 °C. A solution of silver(I) hexafluoroantimonate (0.17 M, 500 µL, 85.1 µmol, 2.00 equiv) in nitroethane at -78 °C was then introduced to the Schlenk tube via cannula. The vessel was subsequently fitted with a PTFE screw cap, sealed, and introduced into a cold bath at -45 °C. After 12 h, aqueous hydrogen chloride (2 N, 4 mL) was added. The reaction mixture was diluted with ethyl acetate (60 mL) and washed with brine  $(3 \times 60 \text{ mL})$ . The organic layer was dried over anhydrous sodium sulfate, was filtered, and was concentrated under reduced pressure. The residue was diluted with dichloromethane (30 mL) and washed with brine (30 mL). The resulting aqueous layer was extracted with dichloromethane ( $2 \times 30$ mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 90% ethyl acetate in hexanes) to afford a mixture of regioisomeric 3-(thienyl)tetracycles (Run 1: 10.4 mg, 51.6%, (-)-19:(-)-20, 1:17; Run 2: 9.5 mg, 47%, (-)-**19**:(-)-**20**, 1:17) as a clear film.

#### 3-(2-Thienyl)tetracycle (-)-19:

Please see page S33 for the full characterization data for 3-(2-thienyl)tetracycle (-)-19.

#### 3-(3-Thienyl)tetracycle (-)-20:

Please see page S34 for the full characterization data for 3-(3-thienyl)tetracycle (-)-20.

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#### <u>3-Allyltetracycle (–)-21:</u>

A solution of silver(I) hexafluoroantimonate (0.20 M, 425  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane was added via syringe to a solution of tetracyclic bromide (+)-**11** (20.0 mg, 42.5  $\mu$ mol, 1 equiv) and allyltributylstannane (26.1  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane (2 mL) at 23 °C. After 1 h, brine (200  $\mu$ L) was added to the solution. The crude reaction mixture was then purified by flash column chromatography on silica gel (eluent: 20% acetone in dichloromethane) to afford 3-allyl tetracycle (-)-**21** (Run 1: 11.1 mg, 60.5%; Run 2: 10.9 mg, 59.4%) as a clear film. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	$\begin{split} &\delta 7.61 \ (\mathrm{d}, J = 7.0, 1\mathrm{H}, \mathrm{C_8H}), 7.42 \ (\mathrm{d}, J = 8.3, 2\mathrm{H}, \\ &\mathrm{Ph_{Cbz}}\text{-}o\text{-}\mathbf{H}), 7.37\text{-}7.27 \ (\mathrm{m}, 3\mathrm{H}, \mathrm{Ph_{Cbz}}\text{-}\mathbf{H}), 7.21\text{-}7.15 \\ &(\mathrm{m}, 1\mathrm{H}, \mathrm{C_7H}), 7.21\text{-}7.15 \ (\mathrm{m}, 1\mathrm{H}, \mathrm{C_3H}), 7.01 \ (\mathrm{app-} \\ &\mathrm{dt}, J = 0.9, 7.5, 1\mathrm{H}, \mathrm{C_6H}), 5.93 \ (\mathrm{s}, 1\mathrm{H}, \mathrm{C_2H}), 5.84 \\ &(\mathrm{br}\text{-}\mathrm{s}, 1\mathrm{H}, \mathrm{N_{14}H}), 5.47\text{-}5.40 \ (\mathrm{m}, 1\mathrm{H}, \\ &\mathrm{CH_2CH=CH_2}), 5.41 \ (\mathrm{d}, J = 12.3, 1\mathrm{H}, \mathrm{Ph_{Cbz}}\mathrm{CH_a}), \\ &5.27 \ (\mathrm{d}, J = 12.3, 1\mathrm{H}, \mathrm{Ph_{Cbz}}\mathrm{CH_b}), 5.03\text{-}4.94 \ (\mathrm{m}, \\ &2\mathrm{H}, \mathrm{CH_2CH=CH_2}), 4.19 \ (\mathrm{dd}, J = 3.9, 10.1, 1\mathrm{H}, \\ &\mathrm{C_{11}H}), 3.93 \ (\mathrm{q}, J = 6.8, 1\mathrm{H}, \mathrm{C_{15}H}), 3.03 \ (\mathrm{dd}, J = \\ &4.0, 13.8, 1\mathrm{H}, \mathrm{C_{12}H_a}), 2.44\text{-}2.28 \ (\mathrm{m}, 2\mathrm{H}, \\ &\mathrm{CH_2CH=CH_2}), 2.44\text{-}2.28 \ (\mathrm{m}, 1\mathrm{H}, \mathrm{C_{12}H_b}), 1.30 \ (\mathrm{d}, \\ &J = 6.8, 3\mathrm{H}, \mathrm{C_{17}H}). \end{split}$
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\begin{split} &\delta \ 169.5 \ (\mathbf{C}_{13}), \ 167.6 \ (\mathbf{C}_{16}), \ 153.3 \ (\mathbf{C}_{\text{Cbz}}\text{=}\text{O}), \ 140.3 \\ &(\mathbf{C}_{9}), \ 136.4 \ (\text{Ph}_{\text{Cbz}}\text{-}\textit{ipso}\text{-}\text{C}), \ 135.0 \ (\mathbf{C}_{4}), \ 132.2 \\ &(\text{CH}_2\text{CH}\text{=}\text{CH}_2), \ 128.9 \ (\mathbf{C}_7), \ 128.7 \ (\text{Ph}_{\text{Cbz}}\text{-}\text{C}), \ 128.4 \\ &(\text{Ph}_{\text{Cbz}}\text{-}\text{C}), \ 128.3 \ (\text{Ph}_{\text{Cbz}}\text{-}\text{C}), \ 124.2 \ (\mathbf{C}_5), \ 124.1 \ (\mathbf{C}_6), \\ &119.9 \ (\text{CH}_2\text{CH}\text{=}\text{CH}_2), \ 116.8 \ (\mathbf{C}_8), \ 81.1 \ (\mathbf{C}_2), \ 68.1 \\ &(\text{Ph}_{\text{Cbz}}\text{CH}_2), \ 58.1 \ (\mathbf{C}_{11}), \ 54.7 \ (\mathbf{C}_3), \ 51.9 \ (\mathbf{C}_{15}), \ 42.4 \\ &(\text{CH}_2\text{CH}\text{=}\text{CH}_2), \ 32.8 \ (\mathbf{C}_{12}), \ 15.4 \ (\mathbf{C}_{17}). \end{split}$
FTIR (thin film) cm <sup>-1</sup> :	3264 (br, m), 1687 (s), 1481 (m), 1410 (m), 1328 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{25}H_{26}N_3O_4$ [M+H] <sup>+</sup> : 432.1918, found: 432.1906.
$\left[\alpha\right]_{D}^{24}$ :	$-37 (c = 0.26, \text{CHCl}_3).$
TLC (ethyl acetate), Rf:	0.34 (UV, CAM).
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### <u>3-Allyltetracycle (–)-21:</u>

A solution of silver(I) hexafluoroantimonate (0.20 M, 425  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane was added via syringe to a solution of tetracyclic bromide (+)-**11** (20.0 mg, 42.5  $\mu$ mol, 1 equiv) and allyltrimethylsilane (13.5  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane (2 mL) at 23 °C. After 1 h, brine (200  $\mu$ L) was added to the solution. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 20% acetone in dichloromethane) to afford 3-allyl tetracycle (-)-**21** (Run 1: 9.5 mg, 52%; Run 2: 9.6 mg, 52%) as a clear film.

Please see page S36 for the full characterization data for 3-allyltetracycle (-)-21.

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### 3-Acetonyltetracycle (–)-22:

A solution of silver(I) hexafluoroantimonate (0.20 M, 425  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane was added via syringe to a solution of tetracyclic bromide (+)-**11** (20.0 mg, 42.5  $\mu$ mol, 1 equiv) and (isopropenyloxy)trimethylsilane (85% purity, 16.6  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane (2 mL) at 23 °C. After 1 h, brine (200  $\mu$ L) was added to the solution. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 20% acetone in dichloromethane) to afford 3-acetonyltetracycle (-)-**22** (Run 1: 17.3 mg, 90.9%; Run 2: 17.2 mg, 90.4%) as a clear film. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 7.60 (app-s, 1H, $C_8$ <b>H</b> ), 7.41 (d, $J = 7.1, 2H$ , Ph <sub>Cbz</sub> - $o$ - <b>H</b> ), 7.34 (app-t, $J = 6.8, 2H$ , Ph <sub>Cbz</sub> - $m$ - <b>H</b> ), 7.32–7.27 (m, 1H, Ph <sub>Cbz</sub> - $p$ - <b>H</b> ), 7.19 (d, $J = 7.5, 1H$ , C <sub>5</sub> <b>H</b> ), 7.16 (app-t, $J = 7.8, 1H, C_7$ <b>H</b> ), 6.99 (app-t, $J = 7.5, 1H, C_6$ <b>H</b> ), 6.29 (br-s, 1H, N <sub>14</sub> <b>H</b> ), 6.12 (s, 1H, C <sub>2</sub> <b>H</b> ), 5.39 (d, $J = 12.3, 1H, Ph_{Cbz}CH_a$ ), 5.27 (d, $J = 12.3, 1H, Ph_{Cbz}CH_b$ ), 4.42 (dd, $J = 5.3, 9.8, 1H$ , C <sub>11</sub> <b>H</b> ), 3.97 (q, $J = 6.7, 1H, C_{15}$ <b>H</b> ), 2.98 (dd, $J = 5.4, 13.5, 1H, C_{14}$ , $C_{14}$ , $D_{14}$ ,
	$CH_2C(=O)CH_3$ , 2.52 (d, $J = 10.4$ , 1H, $CH_2C(=O)CH_3$ ), 2.65 (d, $J = 18.4$ , 1H, $CH_2C(=O)CH_3$ ), 2.52 (dd, $J = 10.1$ , 13.5, 1H, $C_{12}H_b$ ), 2.04 (s, 3H, $CH_2C(=O)CH_3$ ), 1.31 (d, $J = 6.8$ , 3H, $C_{17}H$ ).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$\begin{split} &\delta \ 206.0 \ (\mathrm{CH}_2\mathbf{C}(=\mathrm{O})\mathrm{CH}_3), \ 170.0 \ (\mathbf{C}_{13}), \ 167.7 \ (\mathbf{C}_{16}), \\ &153.6 \ (\mathbf{C}_{\mathrm{Cbz}}=\mathrm{O}), \ 139.8 \ (\mathbf{C}_9), \ 136.2 \ (\mathrm{Ph}_{\mathrm{Cbz}}\text{-}ipso\mathrm{-C}), \\ &135.9 \ (\mathbf{C}_4), \ 129.0 \ (\mathbf{C}_7), \ 128.7 \ (\mathrm{Ph}_{\mathrm{Cbz}}\text{-C}), \ 128.5 \\ &(\mathrm{Ph}_{\mathrm{Cbz}}\text{-C}), \ 128.3 \ (\mathrm{Ph}_{\mathrm{Cbz}}\text{-C}), \ 124.4 \ (\mathbf{C}_6), \ 124.3 \ (\mathbf{C}_5), \\ &116.9 \ (\mathbf{C}_8), \ 80.5 \ (\mathbf{C}_2), \ 68.2 \ (\mathrm{Ph}_{\mathrm{Cbz}}\mathrm{CH}_2), \ 58.2 \ (\mathbf{C}_{11}), \\ &51.9 \ (\mathbf{C}_{15}), \ 51.4 \ (\mathbf{C}_3), \ 49.2 \ (\mathrm{CH}_2\mathrm{C}(=\mathrm{O})\mathrm{CH}_3), \ 33.4 \\ &(\mathbf{C}_{12}), \ 31.0 \ (\mathrm{CH}_2\mathrm{C}(=\mathrm{O})\mathrm{CH}_3), \ 15.6 \ (\mathbf{C}_{17}). \end{split}$
FTIR (thin film) cm <sup>-1</sup> :	3271 (br, m), 1712 (s), 1688 (s), 1481(m), 1410 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{25}H_{26}N_3O_5[M+H]^+$ : 448.1867, found: 448.1846.
$\left[\alpha\right]_{D}^{24}$ :	$-59 (c = 0.52, \text{CHCl}_3).$
TLC (ethyl acetate), Rf:	0.23 (UV, CAM).

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#### <u>3-(Methyl-α,α-dimethylacetate)tetracycle (–)-23:</u>

A solution of silver(I) hexafluoroantimonate (0.20 M, 425  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane was added via syringe to a solution of tetracyclic bromide (+)-**11** (20.0 mg, 42.5  $\mu$ mol, 1 equiv) and methyl trimethysilyl dimethylketene acetal (17.2  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane (2 mL) at 23 °C. After 1 h, brine (200  $\mu$ L) was added to the solution. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 20% acetone in dichloromethane) to afford 3-(methyl- $\alpha$ , $\alpha$ -dimethylacetate)tetracycle (-)-**23** (Run 1: 18.4 mg, 88.2%; Run 2: 18.8 mg, 89.9%) as a clear film. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	$ \begin{split} &\delta~7.69~(\mathrm{app\text{-}s},1\mathrm{H},\mathrm{C_8}\mathbf{H}),7.38~(\mathrm{m},2\mathrm{H},\mathrm{Ph}_{\mathrm{Cbz}}\text{-}o\text{-}\mathbf{H}),\\ &7.32~(\mathrm{app\text{-}t},J=7.5,2\mathrm{H},\mathrm{Ph}_{\mathrm{Cbz}}\text{-}m\text{-}\mathbf{H}),7.29\text{-}7.25~(\mathrm{m},1\mathrm{H},\mathrm{Ph}_{\mathrm{Cbz}}\text{-}p\text{-}\mathbf{H}),7.20~(\mathrm{app\text{-}t},J=7.2,1\mathrm{H},\mathrm{C_7}\mathbf{H}),\\ &7.18~(\mathrm{d},J=7.6,1\mathrm{H},\mathrm{C_5}\mathbf{H}),7.01~(\mathrm{app\text{-}dt},J=0.9,\\ &7.6,1\mathrm{H},\mathrm{C_6}\mathbf{H}),6.45~(\mathrm{s},1\mathrm{H},\mathrm{C_2}\mathbf{H}),6.17~(\mathrm{br\text{-}s},1\mathrm{H},\\ &\mathrm{N_{14}}\mathbf{H}),5.58\text{-}5.08~(\mathrm{br\text{-}m},2\mathrm{H},\mathrm{Ph}_{\mathrm{Cbz}}\mathrm{C}\mathbf{H}_2),4.38\text{-}4.27~(\mathrm{br\text{-}m},1\mathrm{H},\mathrm{C_{11}}\mathbf{H}),3.95~(\mathrm{q},J=6.2,1\mathrm{H},\mathrm{C_{15}}\mathbf{H}),3.63~(\mathrm{s},3\mathrm{H},\mathrm{OCH_3}),2.91\text{-}2.75~(\mathrm{br\text{-}m},1\mathrm{H},\mathrm{C_{12}}\mathbf{H_a}),2.71\text{-}\\ &2.57~(\mathrm{br\text{-}m},1\mathrm{H},\mathrm{C_{12}}\mathbf{H_b}),1.32~(\mathrm{d},J=6.8,3\mathrm{H},\mathrm{C_{17}}\mathbf{H}),\\ &1.26~(\mathrm{s},3\mathrm{H},\mathrm{C}(\mathrm{C}\mathbf{H}_3)_2),0.88~(\mathrm{s},3\mathrm{H},\mathrm{C}(\mathrm{C}\mathbf{H}_3)_2). \end{split}$
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	$ \begin{split} &\delta \ 176.6 \ ({\bf CO}_2{\bf CH}_3), \ 170.1 \ ({\bf C}_{13}), \ 167.7 \ ({\bf C}_{16}), \ 153.1 \\ &({\bf C}_{\rm Cbz}{=}{\bf O}), \ 141.2 \ ({\bf C}_9), \ 136.5 \ ({\bf Ph}_{\rm Cbz}{-}ipso{-}{\bf C}), \ 133.0 \\ &({\bf C}_8), \ 129.3 \ ({\bf C}_7), \ 128.6 \ ({\bf Ph}_{\rm Cbz}{-}{\bf C}), \ 128.2 \ ({\bf Ph}_{\rm Cbz}{-}{\bf C}), \\ &128.1 \ ({\bf Ph}_{\rm Cbz}{-}{\bf C}), \ 126.3 \ ({\bf C}_5), \ 123.9 \ ({\bf C}_6), \ 117.2 \ ({\bf C}_4), \\ &80.3 \ ({\bf C}_2), \ 68.1 \ ({\bf Ph}_{\rm Cbz}{\bf CH}_2), \ 58.7 \ ({\bf C}_3), \ 58.2 \ ({\bf C}_{11}), \\ &52.4 \ ({\bf OCH}_3), \ 51.9 \ ({\bf C}_{15}), \ 47.7 \ ({\bf C}({\bf CH}_3)_2), \ 33.4 \\ &({\bf C}_{12}), \ 21.7 \ ({\bf C}({\bf CH}_3)_2), \ 21.7 \ ({\bf C}({\bf CH}_3)_2), \ 15.7 \ ({\bf C}_{17}). \end{split} $
FTIR (thin film) cm <sup>-1</sup> :	3266 (br, m), 1717 (s), 1688 (s), 1482 (m), 1410 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{27}H_{30}N_3O_6[M+H]^+$ : 492.2129, found: 492.2137.
$\left[\alpha\right]_{D}^{24}:$	$-52 (c = 0.29, \text{CHCl}_3).$
TLC (ethyl acetate), Rf:	0.38 (UV, CAM).

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### 3-(4-Tolyl)tetracycle (–)-24:

A solution of silver(I) hexafluoroantimonate (0.20 M, 425  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in dichloromethane was added via syringe to a solution of tetracyclic bromide (+)-**11** (20.0 mg, 42.5  $\mu$ mol, 1 equiv) and toluene (9.1  $\mu$ L, 85  $\mu$ mol, 2.0 equiv) in dichloromethane (2 mL) at 23 °C. After 1 h, brine (200  $\mu$ L) was added to the solution. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 20% acetone in dichloromethane) to afford 3-(4-tolyl)tetracycle (-)-**24** (Run 1: 11.1 mg, 54.2%; Run 2: 9.3 mg, 45%) as a clear film. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 7.64 (d, $J = 7.9$ , 1H, C <sub>8</sub> H), 7.37 (d, $J = 7.5$ , 2H, Ph <sub>Cbz</sub> - $o$ -H), 7.35–7.26 (m, 3H, Ph <sub>Cbz</sub> -H), 7.35–7.26 (m, 1H, C <sub>5</sub> H), 7.18 (app-t, $J = 7.9$ , 1H, C <sub>7</sub> H), 7.07–7.00 (m, 2H, C <sub>3</sub> H), 7.07–7.00 (m, 2H, C <sub>2</sub> H), 7.07–7.00 (m, 1H, C <sub>6</sub> H), 6.34 (s, 1H, C <sub>2</sub> H), 5.85 (s, 1H, N <sub>14</sub> H), 5.41 (d, $J = 12.4$ , 1H, Ph <sub>Cbz</sub> CH <sub>a</sub> ), 5.28 (d, $J = 12.3$ , 1H, Ph <sub>Cbz</sub> CH <sub>b</sub> ), 4.37 (dd, $J = 3.4$ , 10.1, 1H, C <sub>11</sub> H), 3.98 (q, $J = 6.8$ , 1H, C <sub>15</sub> H), 3.49 (dd, $J = 3.4$ , 13.8, 1H, C <sub>12</sub> H <sub>a</sub> ), 2.85 (dd, $J = 10.3$ , 13.9, 1H, C <sub>12</sub> H <sub>b</sub> ), 2.26 (s, 3H, ArCH <sub>3</sub> ), 1.33 (d, $J = 6.8$ , 3H, C <sub>17</sub> H).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	δ 169.6 ( $C_{13}$ ), 167.7 ( $C_{16}$ ), 153.1 ( $C_{Cbz}$ =O), 139.8 ( $C_{9}$ ), 138.8 ( $C_{4'}$ ), 137.5 ( $C_{1'}$ ), 136.3 ( $Ph_{Cbz}$ - <i>ipso</i> -C), 135.1 ( $C_{4}$ ), 129.8 ( $C_{2'}$ ), 129.1 ( $C_{7}$ ), 128.7 ( $Ph_{Cbz}$ -C), 128.2 ( $Ph_{Cbz}$ -C), 128.2 ( $Ph_{Cbz}$ -C), 128.7 ( $C_{3'}$ ), 125.1 ( $C_{5}$ ), 124.3 ( $C_{6}$ ), 117.0 ( $C_{8}$ ), 83.3 ( $C_{2}$ ), 68.1 ( $Ph_{Cbz}$ CH <sub>2</sub> ), 58.6 ( $C_{11}$ ), 58.3 ( $C_{3}$ ), 51.9 ( $C_{15}$ ), 35.0 ( $C_{12}$ ), 21.1 (ArCH <sub>3</sub> ), 15.3 ( $C_{17}$ ).
FTIR (thin film) $cm^{-1}$ :	3268 (br, m), 1712 (s), 1689 (s), 1409 (m), 1313 (m).
HRMS (ESI) $(m/z)$ :	calc'd for C <sub>29</sub> H <sub>28</sub> N <sub>3</sub> O <sub>4</sub> [M+H] <sup>+</sup> : 482.2074, found: 482.2056.
$\left[\alpha\right]_{D}^{24}$ :	$-6.4 (c = 0.28, \text{CHCl}_3).$
TLC (20% acetone in dichloromethane), Rf:	0.22 (UV, CAM).

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#### 3-(4-Tolyl)tetracycle (-)-24 and 3-(2-Tolyl)tetracycle (-)-25:

A Schlenk tube was charged with tetracyclic bromide (+)-**11** (20.0 mg, 42.5 µmol, 1 equiv), potassium *o*-tolyltrifluoroborate (16.8 mg, 85.1 µmol, 2.00 equiv), and 18-crown-6 (22.4 mg, 85.1 µmol, 2.00 equiv). Nitroethane (2 mL) was then introduced via syringe. Upon dissolution of all solid components, the reaction mixture was cooled to -78 °C. A solution of silver(I) hexafluoroantimonate (0.17 M, 500 µL, 85.1 µmol, 2.00 equiv) in nitroethane at -78 °C was then introduced to the Schlenk tube via cannula. The vessel was subsequently fitted with a PTFE screw cap, sealed, and introduced into a cold bath at -45 °C. After 12 h, aqueous hydrogen chloride (2 N, 4 mL) was added. The reaction mixture was diluted with ethyl acetate (60 mL) and washed with brine (3 × 60 mL). The organic layer was dried over anhydrous sodium sulfate, was filtered, and was concentrated under reduced pressure. The residue was diluted with dichloromethane (30 mL) and washed with brine (30 mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 90% ethyl acetate in hexanes) to afford a mixture of regioisomeric 3-(tolyl)tetracycles (Run 1: 5.1 mg, 25%, (-)-24:(-)-25, 1:4.2; Run 2: 5.4 mg, 26%, (-)-24:(-)-25, 1:3.7) as a clear film.

Regioisomers (–)-24 and (–)-25 were separated for the purpose of full and independent characterization by preparative HPLC [Waters X-Bridge preparative HPLC column, C18, 5 µm, 19 × 250 mm; 20.0 mL/min; 50% acetonitrile in water;  $t_{\rm R}((-)-25) = 11.5 \text{ min}, t_{\rm R}((-)-24) = 12.4 \text{ min}]$ . Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

#### 3-(4-Tolyl)tetracycle (–)-24:

Please see page S40 for the full characterization data for 3-(4-tolyl)tetracycle (-)-24.

### 3-(2-Tolyl)tetracycle (-)-25:

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, 20 °C):

δ 7.66 (d, J = 8.0, 1H, C<sub>8</sub>**H**), 7.34–7.22 (m, 5H, Ph<sub>Cbz</sub>-**H**), 7.34–7.22 (m, 1H, C<sub>5</sub>**H**), 7.34–7.22 (C<sub>7</sub>**H**), 7.19 (d, J = 7.4, 1H, C<sub>6</sub>**H**), 7.16 (app-dt, J = 1.0, 7.4, 1H, C<sub>5</sub>**H**), 7.10 (app-dt, J = 1.0, 7.5, 1H, C<sub>6</sub>**H**), 7.01 (app-dt, J = 1.2, 7.5, 1H, C<sub>4</sub>**H**), 6.85 (d, J = 7.8, 1H, C<sub>3</sub>**H**), 6.50 (s, 1H, C<sub>2</sub>**H**), 5.62 (s, 1H, N<sub>14</sub>**H**), 5.33 (d, J = 12.4, 1H, Ph<sub>Cbz</sub>C**H**<sub>a</sub>), 5.24 (d, J = 12.4, 1H, Ph<sub>Cbz</sub>C**H**<sub>b</sub>), 4.50 (dd, J = 5.7, 9.6, 1H, C<sub>11</sub>**H**), 4.02 (q, J = 6.8, 1H, C<sub>15</sub>**H**), 3.26 (dd, J = 5.8, 14.4, 1H, C<sub>12</sub>**H**<sub>a</sub>), 3.04 (dd, J = 9.7, 14.3, 1H, C<sub>12</sub>**H**<sub>b</sub>), 2.40 (s, 3H, ArC**H**<sub>3</sub>), 1.34 (d, J = 6.9, 3H, C<sub>17</sub>**H**). Concise Total Synthesis and Stereochemical Revision of (+)-Naseseazines A and B. Regioselective Arylative Dimerization of Diketopiperazine Alkaloids. Justin Kim and Mohammad Movassaghi\* Page S42/S118

<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	δ 169.6 ( $C_{13}$ ), 168.0 ( $C_{16}$ ), 153.3 ( $C_{Cbz}$ =O), 140.1 ( $C_{9}$ ), 137.9 ( $C_{2}$ ), 136.3 ( $Ph_{Cbz}$ - <i>ipso</i> -C), 135.8 ( $C_{1}$ ), 135.5 ( $C_{4}$ ), 133.2 ( $C_{6}$ ), 129.2 ( $C_{7}$ ), 128.7 ( $Ph_{Cbz}$ -C), 128.2 ( $C_{3}$ ), 128.2 ( $C_{5}$ ), 128.2 ( $Ph_{Cbz}$ -C), 128.1 ( $Ph_{Cbz}$ -C), 126.4 ( $C_{4}$ ), 125.9 ( $C_{5}$ ), 124.2 ( $C_{6}$ ), 117.2 ( $C_{8}$ ), 82.6 ( $C_{2}$ ), 68.0 ( $Ph_{Cbz}$ CH <sub>2</sub> ), 59.1 ( $C_{3}$ ), 59.0 ( $C_{11}$ ), 52.1 ( $C_{15}$ ), 36.0 ( $C_{12}$ ), 21.9 (ArCH <sub>3</sub> ), 15.5 ( $C_{17}$ ).
FTIR (thin film) cm <sup>-1</sup> :	1689 (s), 1481 (m), 1405 (m), 1329 (m), 751 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{29}H_{28}N_3O_4 [M+H]^+$ : 482.2074, found: 482.2066.
$\left[\alpha\right]_{D}^{24}$ :	$-69 (c = 0.13, \text{CHCl}_3).$
TLC (ethyl acetate), Rf:	0.44 (UV, CAM).

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#### 3-(4-Methoxyphenyl)tetracycle (–)-S2 and 3-(2-Methoxyphenyl)tetracycle (–)-26:

A Schlenk tube was charged with tetracyclic bromide (+)-11 (20.0 mg, 42.5 µmol, 1 equiv), potassium 2-methoxytrifluoroborate (18.2 mg, 85.1 µmol, 2.00 equiv), and 18-crown-6 (22.4 mg, 85.1 μmol, 2.00 equiv). Nitroethane (2 mL) was then introduced via syringe. Upon dissolution of all A solution of silver(I) solid components, the reaction mixture was cooled to -78 °C. hexafluoroantimonate (0.17 M, 500 µL, 85.1 µmol, 2.00 equiv) in nitroethane at -78 °C was then introduced to the Schlenk tube via cannula. The vessel was subsequently fitted with a PTFE screw cap, sealed, and introduced into a cold bath at -45 °C. After 12 h, aqueous hydrogen chloride (2 N, 4 mL) was added. The reaction mixture was diluted with ethyl acetate (60 mL) and washed with brine  $(3 \times 60 \text{ mL})$ . The organic layer was dried over anhydrous sodium sulfate, was filtered, and was concentrated under reduced pressure. The residue was diluted with dichloromethane (30 mL) and washed with brine (30 mL). The resulting aqueous layer was extracted with dichloromethane ( $2 \times 30$ mL). The combined organic layers were dried over anhydrous sodium sulfate, were filtered, and were concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 90% ethyl acetate in hexanes) to afford a mixture of regioisomeric 3-(methoxyphenyl)tetracycles (Run 1: 11.9 mg, 56.2%, (-)-S2:(-)-26, 1:2.2; Run 2: 11.1 mg, 52.5%, (-)-**S2**:(-)-**26**, 1:2.7) as a clear film.

Regioisomers (–)-S2 and (–)-26 were separated for the purpose of full and independent characterization by preparative HPLC [Waters X-Bridge preparative HPLC column, C18, 5 µm, 19 × 250 mm; 20.0 mL/min; gradient, 40% $\rightarrow$ 60% acetonitrile in water, 20 min;  $t_{\rm R}(S2) = 12.5$  min,  $t_{\rm R}(26) = 14.6$  min]. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

#### <u>3-(4-Methoxyphenyl)tetracycle (–)-S2:</u>

<sup>1</sup> H NMR (500 MHz, $CDCl_3$ , 20 °C):	$\delta$ 7.65 (d, $J$ = 7.9, 1H, C <sub>8</sub> H), 7.37 (dd, $J$ = 1.1, 8.2,
	2H, Ph <sub>Cbz</sub> - <i>o</i> - <b>H</b> ), 7.35–7.26 (m, 3H, Ph <sub>Cbz</sub> - <b>H</b> ), 7.35–
	7.26 (m, 1H, $C_5$ <b>H</b> ), 7.20 (app-dt, $J = 1.2, 8.1, 1$ H,
	$C_7$ <b>H</b> ), 7.07–7.02 (m, 1H, $C_6$ <b>H</b> ), 7.07–7.02 (m, 2H,
	$C_{3}$ <b>H</b> ), 6.76 (d, $J = 8.9, 2$ H, $C_{2}$ <b>H</b> ), 6.31 (s, 1H,
	$C_2$ <b>H</b> ), 5.42 (s, 1H, N <sub>14</sub> <b>H</b> ), 5.41 (d, <i>J</i> = 12.2, 1H,
	$Ph_{Cbz}CH_{a}$ ), 5.28 (d, $J = 12.3$ , 1H, $Ph_{Cbz}H_{b}$ ), 4.38
	$(dd, J = 3.4, 10.1, 1H, C_{11}H), 4.00 (q, J = 6.8, 1H),$
	$C_{15}$ <b>H</b> ), 3.73 (s, 3H, OC <b>H</b> <sub>3</sub> ), 3.50 (dd, <i>J</i> = 3.4, 13.9,
	1H, $C_{12}H_a$ ), 2.84 (dd, $J = 10.2, 13.9, 1H, C_{12}H_b$ ),
	1.33 (d, $J = 6.8, 3H, C_{17}H$ ).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>2</sub> , 20 °C):	$\delta$ 169.3 ( <b>C</b> <sub>12</sub> ), 167.7 ( <b>C</b> <sub>16</sub> ), 159.0 ( <b>C</b> <sub>17</sub> ), 153.2
	$(\mathbf{C}_{Ch_2}=\mathbf{O}), 139.8 (\mathbf{C}_0), 136.4 (Ph_{Ch_2}-ipso-\mathbf{C}), 135.1$
	$(\mathbf{C}_{4})$ , 133.7 $(\mathbf{C}_{4})$ , 129.1 $(\mathbf{C}_{7})$ , 128.7 $(\mathbf{Ph}_{C+}-\mathbf{C})$ .

 $(C_4)$ , 133.7  $(C_4)$ , 129.1  $(C_7)$ , 128.7  $(Ph_{Cbz}$ -C), 128.7  $(Ph_{Cbz}$ -C), 128.3  $(Ph_{Cbz}$ -C), 127.1  $(C_3)$ , 125.2 Concise Total Synthesis and Stereochemical Revision of (+)-Naseseazines A and B. Regioselective Arylative Dimerization of Diketopiperazine Alkaloids. Justin Kim and Mohammad Movassaghi\* Page S44/S118

1709 (s), 1688 (s), 1409 (m), 1312 (m), 1254 (m).
calc'd for $C_{29}H_{28}N_3O_5[M+H]^+$ : 498.2023, found: 498.2032.
$-13 (c = 0.070, \text{CHCl}_3).$
0.40 (UV, CAM).
δ 7.61 (d, $J = 7.5$ , 1H, C <sub>8</sub> H), 7.36 (d, $J = 7.1$ , 1H, C <sub>5</sub> H), 7.32–7.20 (m, 1H, C <sub>7</sub> H), 7.32–7.20 (m, 5H, Ph <sub>Cbz</sub> -H), 7.32–7.20 (m, 1H, C <sub>5</sub> H), 7.13 (app-t, $J =$ 7.5, 1H, C <sub>6</sub> H), 6.92 (d, $J = 8.0$ , 1H, C <sub>6</sub> H), 6.75 (app-t, $J = 7.4$ , 1H, C <sub>4</sub> H), 6.66 (dd, $J = 1.6$ , 7.8, 1H, C <sub>3</sub> H), 6.45 (s, 1H, C <sub>2</sub> H), 5.90–5.75 (m, 1H, N <sub>14</sub> H), 5.36 (d, $J = 12.5$ , 1H, Ph <sub>Cbz</sub> CH <sub>a</sub> ), 5.19 (d, $J =$ 12.5, 1H, Ph <sub>Cbz</sub> CH <sub>b</sub> ), 4.52 (dd, $J = 5.8$ , 9.9, 1H, C <sub>11</sub> H), 4.05 (q, $J = 6.8$ , 1H, C <sub>15</sub> H), 3.89 (s, 3H, OCH <sub>3</sub> ), 3.12 (dd, $J = 5.8$ , 13.5, 1H, C <sub>12</sub> H <sub>a</sub> ), 3.01 (dd, $J = 10.1$ , 13.5, 1H, C <sub>12</sub> H <sub>b</sub> ), 1.36 (d, $J = 6.9$ , 3H, C <sub>17</sub> H).
δ 170.1 ( $C_{13}$ ), 167.6 ( $C_{16}$ ), 156.8 ( $C_{1'}$ ), 153.4 ( $C_{Cbz}$ =O), 140.5 ( $C_{9}$ ), 136.4 ( $Ph_{Cbz}$ - <i>ipso</i> -C), 135.4 ( $C_{4}$ ), 129.4 ( $C_{5'}$ ), 129.2 ( $C_{3'}$ ), 129.1 ( $C_{7}$ ), 128.8 ( $C_{2'}$ ), 128.6 ( $Ph_{Cbz}$ -C), 128.1 ( $Ph_{Cbz}$ -C), 127.9 ( $Ph_{Cbz}$ -C), 126.8 ( $C_{5}$ ), 124.2 ( $C_{6}$ ), 120.8 ( $C_{4'}$ ), 117.5 ( $C_{8}$ ), 111.5 ( $C_{6'}$ ), 82.1 ( $C_{2}$ ), 67.9 ( $Ph_{Cbz}$ CH <sub>2</sub> ), 58.7 ( $C_{11}$ ), 56.8 ( $C_{3}$ ), 55.6 (OCH <sub>3</sub> ), 52.0 ( $C_{15}$ ), 36.3 ( $C_{12}$ ), 15.9 ( $C_{17}$ ).
1712 (s), 1688 (s), 1482 (m), 1409 (m), 1326 (m).
calc'd for $C_{29}H_{28}N_3O_5[M+H]^+$ : 498.2023, found: 498.2033.
$-152 (c = 0.15, CHCl_3).$
0.33 (UV, CAM).

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### 3-(trans-Styryl)tetracycle (-)-27:

A solution of silver(I) hexafluoroantimonate (0.20 M, 425  $\mu$ L, 85.1  $\mu$ mol, 2.00 equiv) in nitroethane was added via syringe to a solution of tetracyclic bromide (+)-**11** (20.0 mg, 42.5  $\mu$ mol, 1 equiv), potassium *trans*-styryltrifluoroborate (17.9 mg, 85.1  $\mu$ mol, 2.00 equiv), and 18-crown-6 (22.4 mg, 85.1  $\mu$ mol, 2.00 equiv) in nitroethane (2 mL) at 23 °C. After 1 h, aqueous hydrogen chloride (2 N, 4 mL) was added. The reaction mixture was diluted with ethyl acetate (60 mL) and washed with brine (3 × 60 mL). The organic layer was dried over anhydrous sodium sulfate, was filtered, and was concentrated under reduced pressure. The crude reaction mixture was purified by flash column chromatography on silica gel (eluent: 20% acetone in dichloromethane). The fractions containing the desired product were combined and were concentrated under reduced pressure. The resulting residue was loaded onto a silica gel column and purified by flash column chromatography (eluent: 90% ethyl acetate in hexanes) to afford 3-(*trans*-styryl)tetracycle (–)-**27** (Run 1: 12.6 mg, 60.0%; Run 2: 12.0 mg, 57.2%) as a white foam. Structural assignments were made using additional information from gCOSY, HSQC, and HMBC experiments.

<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 20 °C):	δ 7.68 (d, $J = 8.0, 1H, C_8H$ ), 7.41 (d, $J = 7.1, 2H$ , Ph <sub>Cbz</sub> - $o$ - <b>H</b> ), 7.33–7.18 (m, 1H, C <sub>7</sub> <b>H</b> ), 7.33–7.18 (m, 1H, C <sub>5</sub> <b>H</b> ), 7.33–7.18 (m, 8H, Ar <b>H</b> ), 7.08 (app- t, $J = 7.5, 1H, C_6H$ ), 6.25 (d, $J = 16.0, 1H$ , C <b>H=</b> CHPh), 6.09 (d, $J = 14.4, 1H, CH=$ C <b>H</b> Ph), 6.07 (s, 1H, C <sub>2</sub> <b>H</b> ), 5.77 (br-s, 1H, N <sub>14</sub> <b>H</b> ), 5.39 (d, $J = 12.3, 1H, Ph_{Cbz}$ C <b>H</b> <sub>a</sub> ), 5.29 (d, $J = 12.3, 1H, Ph_{Cbz}$ C <b>H</b> <sub>a</sub> ), 5.29 (d, $J = 3.1, 10.0, 1H, C_{11}H$ ), 3.97 (q, $J = 6.8, 1H, C_{15}H$ ), 3.28 (dd, $J = 3.3, 13.9, 1H$ , C <sub>12</sub> <b>H</b> <sub>a</sub> ), 2.63 (dd, $J = 10.2, 13.8, 1H, C_{12}H_b$ ), 1.32 (d, $J = 6.8, 3H, C_{17}H$ ).
<sup>13</sup> C NMR (125.8 MHz, CDCl <sub>3</sub> , 20 °C):	δ 169.5 (C <sub>13</sub> ), 167.7 (C <sub>16</sub> ), 153.4 (C <sub>Cbz</sub> =O), 140.4 (C <sub>9</sub> ), 136.3 (Ph <sub>Cbz</sub> - <i>ipso</i> -C), 136.1 (Ph <sub>styrene</sub> - <i>ipso</i> -C), 133.1 (C <sub>4</sub> ), 131.7 (CH=CHPh), 129.4 (CH=CHPh), 129.3 (Ph <sub>Cbz</sub> -C), 128.8 (Ph <sub>Cbz</sub> -C), 128.7 (Ph <sub>Cbz</sub> -C), 128.4 (Ph <sub>Cbz</sub> -C), 128.3 (Ph <sub>Cbz</sub> -C), 128.2 (Ph <sub>Cbz</sub> -C), 126.8 (C <sub>7</sub> ), 125.2 (C <sub>5</sub> ), 124.3 (C <sub>6</sub> ), 116.9 (C <sub>8</sub> ), 81.9 (C <sub>2</sub> ), 68.2 (Ph <sub>Cbz</sub> CH <sub>2</sub> ), 58.3 (C <sub>11</sub> ), 57.2 (C <sub>3</sub> ), 51.9 (C <sub>15</sub> ), 33.3 (C <sub>12</sub> ), 15.3 (C <sub>17</sub> ).
FTIR (thin film) cm <sup>-1</sup> :	3262 (br, m), 1712 (s), 1690 (s), 1409 (m), 1315 (m).
HRMS (ESI) $(m/z)$ :	calc'd for $C_{30}H_{28}N_3O_4$ [M+H] <sup>+</sup> : 494.2074, found: 494.2084.
$\left[\alpha\right]_{D}^{24}$ :	$-23 (c = 0.081, \text{CHCl}_3).$
TLC (ethyl acetate), Rf:	0.44 (UV, CAM).

Assignment <sup>11</sup>	<b>Raju et al. Report</b> <sup>4,12</sup>	This Work
5	(+)-Naseseazine A (1)	(+)-Naseseazine A (1)
	<sup>1</sup> H NMR, 600 MHz, methanol- $d_4$	<sup>1</sup> H NMR, 500 MHz, methanol- $d_4$ , 20 °C
N1	-	-
C2	5.83 (s)	5.83 (s)
C3	-	-
C4	-	-
C5	6.85 (d, J = 7.4)	6.85 (dd, J = 0.6, 7.4)
C6	6.67 (t, J = 7.5)	6.68  (app-dt,  J = 0.9, 7.4)
C7	7.05 (t, $J = 7.2$ )	7.05  (app-dt,  J = 1.3, 7.9)
C8	6.69 (d, J = 7.6)	6.69 (d, J = 8.0)
C9	-	-
N10	-	-
C11	$4.64 (\mathrm{dd}, J = 7.4, 8.4)$	4.65 (dd, <i>J</i> = 7.4, 9.2)
C12	3.26 (m)	3.30-3.20 (m)
012	2.59 (dd, J = 10.2, 13.7)	2.59 (dd, <i>J</i> = 9.9, 13.6)
C13	-	-
N14	-	-
C15	4.15 (q, J = 6.9)	4.16 (dq, J = 1.4, 6.9)
C16	-	-
C17	1.38 (d, J = 6.9)	1.38 (d, $J = 7.1$ )
N1'	-	-
C2'	7.11 (s)	7.12 (s)
C3'	-	-
C4'	-	-
C5'	7.57 (d, J = 8.4)	7.57 (d, J = 8.4)
C6'	7.02 (d, J = 8.4)	7.03 (dd, J = 1.7, 8.4)
C7'	-	-
C8'	7.40 (s)	7.40 (d, $J = 1.3$ )
C9'	-	-
N10'	-	-
C11'	4.39  (br-t,  J = 4.5)	4.39  (app-dt,  J = 1.1, 4.7)
C12'	3.30 (m)	3.30-3.20 (m)
	3.28 (m)	
C13'	-	-
N14'	-	-
C15'	3.97 (dd, J = 6.6, 10.8)	3.98 (ddd, J = 1.7, 6.3, 10.9)
C16'	-	-
C17'	1.97 (m)	1.96 (app-ddt, $J = 2.0, 7.2, 12.5$ )
	<u> </u>	0.96-0.85 (m)
C18'	1.66 (m)	1.72 - 1.61 (m)
-	1.43 (m)	1.47–1.39 (m)
C19'	3.42 (dt, J = 8.1, 11.8)	3.42 (app-dt, $J = 8.3, 11.7$ )
	3.24 (m)	3.30–3.20 (m)

# Table S1. Comparison of our data for (+)-Naseseazine A (1) with literature:

 <sup>&</sup>lt;sup>11</sup> Please see page S3 for the positional numbering system used in this report.
<sup>12</sup> The reference points for the residual protium and carbon resonances of the NMR solvent were not listed.

<b>Assignment</b> <sup>11</sup>	<b>Raju et al. Report</b> <sup>4,12</sup>	This Work
8	(+)-Naseseazine A (1)	(+)-Naseseazine A (1)
	<sup>1</sup> H NMR, 600 MHz, DMSO- $d_6$	<sup>1</sup> H NMR, 500 MHz, DMSO- $d_6$ , 20 °C
N1	6.72 (d, J = 2.8)	6.75 (d, J = 2.8)
C2	5.65 (d, J = 2.8)	5.65 (d, J = 2.8)
C3	-	-
C4	-	-
C5	6.82 (d, J = 7.3)	6.83 (d, J = 7.3)
C6	6.57 (dt, J = 0.7, 7.5)	6.58  (app-t,  J = 7.4 )
C7	7.00 (t, J = 7.2)	7.04–6.95 (m)
C8	6.61 (d, J = 7.8)	6.62 (d, J = 8.0)
C9	-	-
N10	-	-
C11	$4.60  (\mathrm{dd}, J = 8.4, 9.9)$	4.62  (app-t,  J = 8.7)
C12	3.08 (dd, J = 7.6, 13.6)	3.14–3.00 (m)
012	2.41 (dd, $J = 9.9, 13.6$ )	2.41 (dd, J = 10.0, 13.6)
C13	-	-
N14	8.17 (s)	8.20 (s)
C15	4.13 (q, J = 6.9)	4.14 (q, J = 6.5)
C16	-	-
C17	1.21 (d, J = 6.9)	1.23 (d, J = 6.9)
N1'	10.79 (d, <i>J</i> = 1.9)	10.81 (s)
C2'	7.17 (d, J = 2.2)	7.19 (d, <i>J</i> = 1.8)
C3'	-	-
C4'	-	-
C5'	7.55 (d, J = 8.3)	7.56 (d, J = 8.4)
C6'	6.98 (d, J = 8.2)	7.04–6.95 (m)
C7'	-	-
C8'	7.27 (d, J = 1.2)	7.28 (s)
C9'	-	-
N10'	7.68 (s)	7.71 (s)
C11'	4.27 (br-t, J = 5.2)	4.28  (app-t,  J = 4.9 )
C12'	3.21 (dd, J = 4.8, 15.0)	3.30–3.18 (m)
~~~~	3.03 (dd, J = 5.8, 14.9)	3.14–3.00 (m)
C13'	-	-
N14'	-	-
C15'	4.06 (dd, J = 7.5, 10.0)	4.06  (app-t,  J = 7.8)
C16'	-	-
C17'	1.96 (m)	2.03–1.94 (m)
	<u>1.40 (m)</u>	1.47–1.36 (m)
C18'	1.6/(m)	1.74–1.56 (m)
	<u>1.61 (m)</u>	
C19'	3.34 (m)	3.41–3.29 (m)
	3.25 (m)	3.30 - 3.18 (m)

# Table S2. Comparison of our data for (+)-Naseseazine A (1) with literature:

Assignment <sup>11</sup>	<b>Raju et al. Report</b> <sup>4,12</sup>	This Work	Chemical Shift Difference
	(+)-Naseseazine A (1)	(+)-Naseseazine A (1)	$\Delta\delta$
	<sup>13</sup> C NMR, 151 MHz	<sup>13</sup> C NMR, 126 MHz	$\delta$ (this work) – $\delta$ (Raju et.
	methanol- $d_4$	methanol- $d_4$ , 20 °C	al. report)
C2	87.1	87.3	0.2
C3	61.2	61.4	0.2
C4	135.8	136.1	0.3
C5	124.9	125.1	0.2
C6	120.2	120.6	0.4
C7	129.2	129.6	0.4
C8	110.9	111.2	0.3
C9	149.1	149.3	0.2
C11	60.2	60.5	0.3
C12	39.7	39.9	0.2
C13	172.6	172.7	0.1
C15	52.1	52.4	0.3
C16	170.6	170.9	0.3
C17	15.2	15.4	0.2
C2'	126.2	126.6	0.4
C3'	109.5	109.7	0.2
C4'	127.6	127.8	0.2
C5'	120.3	120.5	0.2
C6'	119.5	119.7	0.2
C7'	137.2	137.3	0.1
C8'	110.1	110.4	0.3
C9'	137.9	138.1	0.2
C11'	57.2	57.4	0.2
C12'	29.0	29.3	0.3
C13'	167.3	167.5	0.2
C15'	60.0	60.2	0.2
C16'	170.6	170.9	0.3
C17'	29.0	29.3	0.3
C18'	22.5	22.7	0.2
C19'	45.8	46.1	0.3

# Table S3. Comparison of our data for (+)-Naseseazine A (1) with literature:

Assignment <sup>11</sup>	<b>Raju et al. Report</b> <sup>4,12</sup>	This Work	
0	(+)-Naseseazine B (2)	(+)-Naseseazine B (2)	
	<sup>1</sup> H NMR, 600 MHz, methanol- $d_4$	<sup>1</sup> H NMR, 500 MHz, methanol- $d_4$ , 20 °C	
N1	-	-	
C2	5.85 (s)	5.84 (s)	
C3	-	-	
C4	-	-	
C5	6.84 (dt, J = 0.9, 7.2)	6.83 (d, <i>J</i> = 7.2)	
C6	6.68 (t, J = 7.6)	6.68  (app-t,  J = 7.3 )	
C7	7.06 (dt, $J = 1.3, 7.6$ )	7.06  (app-dt,  J = 1.2, 7.9)	
C8	6.69 (d, J = 7.6)	6.69 (d, <i>J</i> = 7.7)	
C9	-	-	
N10	-	-	
C11	4.75 (dd, J = 8.7, 10.2)	4.71 (app-t, $J = 8.6$ )	
C12	3.27 (m)	3.35-3.21 (m)	
012	2.59 (dd, J = 10.2, 13.8)	2.57 (dd, J = 10.2, 13.7)	
C13	-	-	
N14	-	-	
C15	4.33 (dd, J = 7.1, 9.5)	4.29 (app-t, $J = 7.7$ )	
C16	-	-	
C17	2.28 (m)	2.30-2.22 (m)	
017	2.11 (m)	2.15-2.04 (m)	
C18	2.00 (m)	2.04–1.87 (m)	
010	1.95 (m)	2.04–1.87 (m)	
C19	3.49 (m)	3.52–3.37 (m)	
	3.44 (m)	3.52-3.37 (m)	
N1'	-	-	
C2'	7.12 (s)	7.12 (s)	
C3'	-	-	
C4'	-	-	
C5'	7.58 (d, J = 8.4)	7.57 (d, J = 7.4)	
C6'	7.03  (dd, J = 1.8, 8.4)	7.01  (dd, J = 1.6, 8.5)	
C7'	-	-	
<u>C8'</u>	7.41 (d, $j = 1.4$ )	7.41 (d, <i>J</i> =1.3)	
C9'	-	-	
N10'	-	-	
C11'	4.40  (br-t,  J = 4.7 )	4.39  (app-t,  J = 4.6 )	
C12'	3.32 (m)	3.35–3.21 (m)	
	3.28 (m)	3.35–3.21 (m)	
C13'	-	-	
N14'	-	-	
C15'	3.99 (ddd, J = 1.6, 6.6, 11.4)	3.98 (ddd, J = 1.4, 6.3, 10.7)	
C16'	-	-	
C17'	1.97 (m)	2.04–1.87 (m)	
	0.92 (m)	0.97-0.87 (m)	
C18'	1.67 (m)	1.73–1.60 (m)	
	<u>1.44 (m)</u>	<u>1.49–1.40 (m)</u>	
C19'	3.43 (m)	3.52–3.37 (m)	
017	3.24 (m)	3.35-3.21 (m)	

# Table S4. Comparison of our data for (+)-Naseseazine B (2) with literature:

Assignment <sup>11</sup>	<b>Raju et al. Report</b> <sup>4,12</sup>	This Work
0	(+)-Naseseazine B (2)	(+)-Naseseazine B (2)
	<sup>1</sup> H NMR, 600 MHz, DMSO- $d_6$	<sup>1</sup> H NMR, 500 MHz, DMSO- $d_6$ , 20 °C
N1	6.75 (d, J = 3.2)	6.82–6.76 (m)
C2	5.65 (d, J = 3.2)	5.68 (d, J = 3.2)
C3	-	-
C4	-	_
C5	6.82 (d, J = 7.3)	6.82–6.76 (m)
C6	6.58 (dt, J = 0.9, 7.4)	6.58 (app-dt, $J = 1.0, 7.5$ )
C7	7.00 (t, J = 7.2)	7.00 (app-dt, $J = 1.3, 7.9$ )
C8	6.61 (d, J = 7.8)	6.60 (d, J = 7.6)
C9	-	-
N10	-	-
C11	4.71 (dd, J = 7.5, 10.5)	4.72  (app-t,  J = 8.2 )
<u> </u>	3.12 (dd, J = 7.5, 13.6)	3.13 (dd. J = 7.4, 13.6)
C12	2.36 (dd, J = 10.5, 13.6)	2.36 (dd, J = 10.5, 13.6)
C13	-	-
N14	_	-
C15	4.34 (t, $J = 8.0$ )	4.35 (app-t, J = 7.9)
C16	-	-
G1 <b>5</b>	2.15 (m)	2.21–2.11 (m)
C17	1.95 (m)	2.03–1.91 (m)
C18	1.85 (m)	1.91–1.77 (m)
C19	3.34 (m)	3.41–3.30 (m)
N1'	10.80 (d, J = 1.8)	10.81 (s)
C2'	7.18 (d, J = 2.2)   7.19 (d, J = 2.2)	
C3'	-	-
C4'	_	-
C5'	7.56 (d, J = 8.4)	7.57 (d, J = 8.4)
C6'	6.98 (dd, J = 1.5, 8.4)	6.99 (d, J = 8.6)
C7'	-	-
C8'	7.29 (d, J = 1.5)	7.30 (d, J = 1.3)
C9'	-	-
N10'	7.67 (s)	7.71 (s)
C11'	4.27 (br-t, J = 5.1)	4.29  (app-t,  J = 5.1 )
0121	3.20 (dd, J = 4.5, 15.0)	3.22 (dd, J = 4.6, 14.6)
C12	$3.04 (\mathrm{dd}, J = 6.0, 15.0)$	$3.05 (\mathrm{dd}, J = 5.8, 14.8)$
C13'	-	-
N14'	-	-
C15'	4.06 (dd, J = 7.6, 9.0)	4.06  (app-t,  J = 8.7)
C16'	-	-
C17!	1.97 (m)	2.03–1.91 (m)
	1.41 (m)	1.48–1.36 (m)
C19!	1.68 (m)	175 156 (m)
CIð	1.62 (m)	1./3–1.30 (M)
C10!	3.34 (m)	3.41–3.30 (m)
C19 <sup>-</sup>	3.24 (m)	3.29–3.24 (m)

## Table S5. Comparison of our data for (+)-Naseseazine B (2) with literature:

Assignment <sup>11</sup>	<b>Raju et al. Report</b> <sup>4,12</sup>	This Work	Chemical Shift Difference
_	(+)-Naseseazine B (2)	(+)-Naseseazine B (2)	$\Delta\delta$
	<sup>13</sup> C NMR, 151 MHz	<sup>13</sup> C NMR, 126 MHz	$\delta$ (this work) – $\delta$ (Raju et.
	methanol- $d_4$	methanol- $d_4$ , 20 °C	al. report)
C2	86.8	87.1	0.3
C3	61.7	61.9	0.2
C4	136.0	136.2	0.2
C5	124.8	125.1	0.3
C6	120.3	120.6	0.3
C7	129.1	129.6	0.5
C8	111.0	111.3	0.3
C9	149.0	149.3	0.3
C11	61.3	61.7	0.4
C12	39.5	39.7	0.2
C13	168.4	168.6	0.2
C15	61.8	61.9	0.1
C16	170.2	170.3	0.1
C17	28.3	28.6	0.3
C18	24.1	24.4	0.3
C19	45.9	46.4	0.5
C2'	126.1	126.6	0.5
C3'	109.5	109.8	0.3
C4'	127.6	127.8	0.2
C5'	120.3	120.6	0.3
C6'	119.4	119.8	0.4
C7'	136.9	137.2	0.3
C8'	110.3	110.5	0.2
C9'	137.9	138.1	0.2
C11'	57.0	57.3	0.3
C12'	29.2	29.3	0.1
C13'	167.3	167.5	0.2
C15'	59.9	60.2	0.3
C16'	170.7	170.9	0.2
C17'	29.1	29.3	0.2
C18'	22.4	22.7	0.3
C19'	45.8	46.1	0.3

# Table S6. Comparison of our data for (+)-Naseseazine B (2) with literature:

### Marfey's Analysis of (+)-Naseseazine B (2).

Marfey's analysis was performed on synthetic (+)-naseseazine B (**2**) using the method described in the isolation report.<sup>4</sup> (+)-Naseseazine B (**2**) (350  $\mu$ g, 620 nmol, 1 equiv) was dissolved in aqueous hydrochloric acid (6 N, 1.40 mL) and heated to 100 °C in a sealed vial. After 12 h, the reaction mixture was allowed to cool to 23 °C then concentrated under reduced pressure. A solution of 1-fluoro-2,4-dinitrophenyl-5-L-alanine amide (L-FDAA) in acetone (1% w/v, 700  $\mu$ L) and aqueous sodium bicarbonate (1 M, 140  $\mu$ L) were sequentially added to the residue and the resulting solution was heated to 37 °C. After 1 h, the reaction mixture was allowed to cool to 23 °C and aqueous hydrochloric acid (1 M, 140  $\mu$ L) was added. The solution was then diluted with acetonitrile (5.67 mL) and passed through a syringe filter (Acrodisc GHP, 13 mm diameter, 0.45  $\mu$ m pore size). An aliquot of this solution (10  $\mu$ L) was diluted in water (450  $\mu$ L) and analyzed by HPLC.

A standard solution of L-Ala-NH<sub>2</sub>-DNP-L-Pro was prepared from L-proline. A solution of L-FDAA in acetone (1% w/v, 100  $\mu$ L) and aqueous sodium bicarbonate (1 M, 20  $\mu$ L) were sequentially added to an aqueous solution of L-proline (50 mM, 50  $\mu$ L). The solution was then heated to 37 °C. After 1 h, the reaction mixture was allowed to cool to 23 °C and aqueous hydrochloric acid (1 M, 20  $\mu$ L) was added. The solution was then diluted with isopropanol (810  $\mu$ L) and passed through a syringe filter (Acrodisc GHP, 13 mm diameter, 0.45  $\mu$ m pore size). An aliquot of this solution (1  $\mu$ L) was diluted in water (450  $\mu$ L) and analyzed by HPLC.

A standard solution of L-Ala-NH<sub>2</sub>-DNP-D-Pro was prepared analogously from D-proline.

HPLC conditions: Waters X-Bridge analytical HPLC column, C18, 5  $\mu$ m, 4.6 × 250 mm; 2.00 mL/min; gradient, 0%  $\rightarrow$  30% acetonitrile in water, 30 min;  $t_R$ (L-Ala-NH<sub>2</sub>-DNP-L-Pro) = 13.1 min,  $t_R$ (L-Ala-NH<sub>2</sub>-DNP-D-Pro) = 16.0 min. HPLC traces were obtained using UV detection (340 nm) and product identity was further corroborated with HPLC traces of mass spectrometric data (ESI+). See Figure S1 for the HPLC traces of L-FDAA derived (+)-naseseazine B hydrolysate, the corresponding amino acid standards, and their co-injection. The HPLC analysis confirms the presence of L-proline residues in the synthetic samples of (+)-naseseazine B (2) and is consistent with our use of L-proline building blocks en route to the natural product. Marfey's analysis thus lends further support to our stereochemical reassignment of (+)-naseseazine B (2).



Concise Total Synthesis and Stereochemical Revision of (+)-Naseseazines A and B. Regioselective Arylative Dimerization of Diketopiperazine Alkaloids. Justin Kim and Mohammad Movassaghi\*



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Injection Date : Seq. Line : 1 Sample Name : Location : Acq. Operator : Inj : 1 Acq. Method : Analysis Method :



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## Scan Analysis Report

Sample Name: Naseseazine A				
Peak Table Peak Style Peak Threshold Range		Peaks 0.0100 600.0nm to 200.0nm		
Wavelength (nm)	Abs			
285.0 229.0	0.243 1.376	-		



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## Scan Analysis Report

## Sample Name: Naseseazine B

Peak Table			
Peak Style		Peaks	
Peak Threshold		0.0100	
Range		600.0nm t	o 200.0nm
Wavelength (nm)	Abs		

285.0	0.335
229.0	1.905
204.4	3.350
202.6	3.848
201.0	3.600







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