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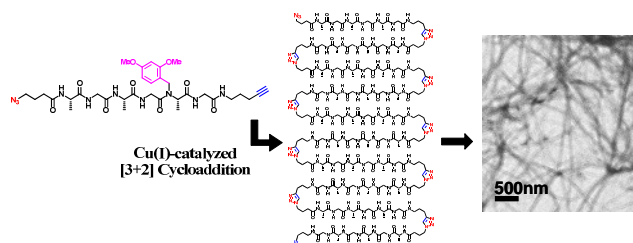
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Ting-Bin Yu, Jane Z. Bai and Zhibin Guan*

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"Click" to Fold: Cycloaddition-Induced Self-Assembly of a Polymer into Well-Defined β -Sheets and Hierarchical Nanofibrils



Cu(I)-catalyzed azide–alkyne cycloaddition polymerization of a peptide monomer induced folding of the formed polymer into well-defined β -sheets which further self assemble into hierarchical nanofibrils. The antiparallel β -sheet structure was confirmed by FTIR, CD and powder X-ray diffraction. TEM and AFM micrographs prove the formation of hierarchical amyloid-like nanofibrils.

**“Click” to Fold: Cycloaddition-Promoted Self-Assembly of a Polymer into Well-Defined
 β -Sheets and Hierarchical Nanofibrils**

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I. General Information.

¹H NMR spectra were taken on 500MHz and 600MHz Bruker instruments. NMR chemical shifts were reported as δ values in ppm relative to TMS or deuterated solvent: CDCl₃ (7.27), DMSO-*d*₆ (2.50), D₂O (4.80), CD₃OD (4.78, 3.31). Data are presented as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), integration and coupling constant(s) in Hz. Multiplets were reported over the range (in ppm) it appeared. ¹³C NMR spectra were taken on 125MHz and 150MHz Bruker instruments. Carbon NMR data were recorded relative to the following solvent signals: CDCl₃ (77.0), DMSO-*d*₆ (39.51), CD₃OD (49.15). All solvents used in reactions were from alumina filtration system. Moisture sensitive reactions were performed under nitrogen atmosphere using flame-dried glassware and standard syringe/septa techniques. Extraction solvents were commercial grade. Flash Chromatography was performed using forced flow of the indicated solvent systems over Fisher silica gel (230 - 400 mesh). All glassware was flame dried before use, and reactions were carried out under nitrogen atmosphere. All buffer solutions were corrected by pH-meter. Infrared (IR) spectra were recorded on a MIDAC Grams/Prospect FT-IR spectrometer using a KBr pellet. Circular dichroism spectroscopy was performed on a Jasco J-810 spectropolarimeter (band width: 1nm, response: 1sec., sensitivity: standard, 1 range: 250-190 nm, data pitch: 0.5 nm, scanning speed: 20 nm/min., accumulation: 5). Mass spectral data (both ES/MS and HRMS) was obtained on a Micromass auto spec spectrometer. MALDI-TOF data was obtained on Applied Biosystems Voyage-DE STR. Gel Permeation Chromatography (GPC) was carried out using an Agilent 1100 Series GPC-SEC Analysis System along with a mixed bed Plgel Mixed-C column from Polymer Labs. The eluent was DMF with 0.1% LiBr and a flow rate of 1.0 mL / min was used. The calibration was performed using poly(ethylene glycol) (PEG)-based molecular weight standards. HPLC analysis was performed using Rainin SD-200 solvent delivery system, UV-1 detector unit and Zobax-80SB C18 reversed phase analytical column (manufactured by Agilent n Varian). The typical flow rate for the analytical analysis was 1 mL/min. In all cases, water and acetonitrile buffers were used with 0.1% TFA. All other chemicals were purchased from Aldrich, ACROS, Novabiochem, or TCI and used without further purification.

II. Synthesis and Characterization

Experimental details of the solid phase synthesis:

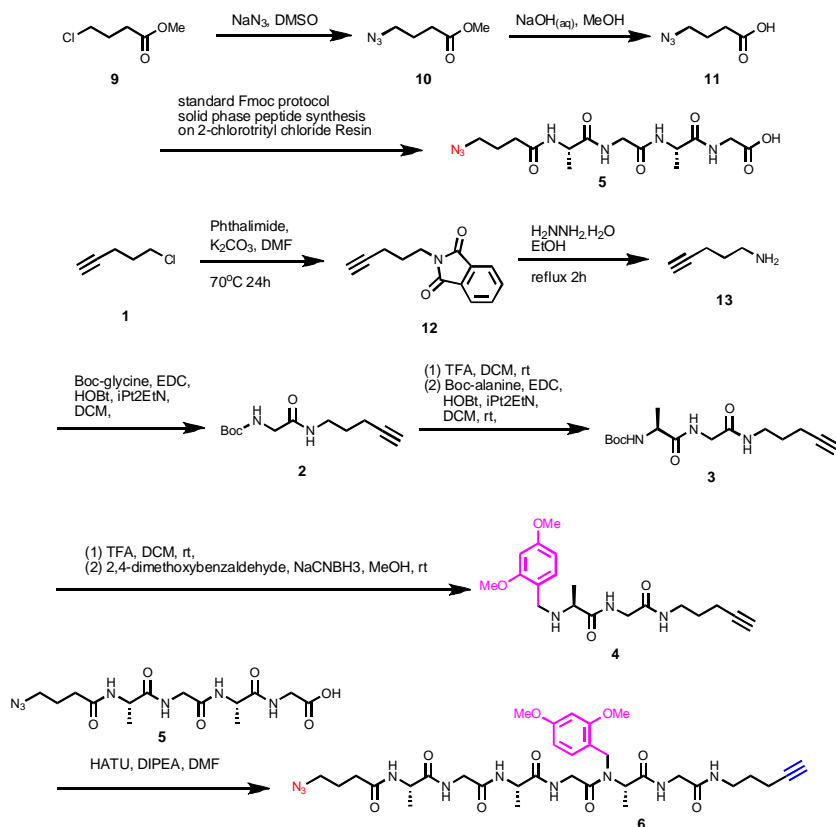
The peptides **5** (Scheme S1) were synthesized by solid phase peptide synthesis (SPPS) following standard Fmoc protocol¹ on 2-chlorotrityl chloride resin using 2-(1H-benzotriazole-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate (HBTU) and/or O-(7-azabenzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate (HATU) as coupling reagents. All the peptides were characterized by analytical HPLC method: 5 - 100% CH₃CN in H₂O gradient over 20 min at 214 nm, 1 mL/min on Agilent Zobax-80SB C18 column.

Standard protocol for SPPS on 2-chlorotrityl chloride resin:¹

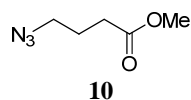
In a 250mL round bottom flask, a suspension of 5.00g 2-chlorotrityl chloride resin (loading: 1.27 mmol/g) and 3.0 equiv of Fmoc-Gly-OH (11.33g, 38.1 mmol) in 50 mL DCM were stirred for 1 hour, followed by addition of 5.0 equiv of *N,N*-diisopropylethylamine (DIPEA). The mixture was stirred for 3 hours at room temperature. The resin was filtered in a solid-phase synthesis reaction vessel and washed with DMF (5 x 50 mL), DCM (5 x 50 mL) and NMP (5 x 50 mL). The resin was then added with capping reagent (DIPEA/MeOH/DCM = 1/2/17) and agitated with N₂ gas for 1 hour. the solution was drained and the resin was washed with DMF (5 x 50 mL), DCM (5 x 50 mL). A preactivated solution of Fmoc-amino acid (50.8 mmol), HBTU (50.8 mmol) and DIPEA (76.2 mmol) in 20 ml NMP was added into the resin for the subsequent couplings. The reaction mixture was agitated with N₂ for 2 hours. The solution was drained and the resin was then washed with DMF (5 x 50 mL), DCM (5 x 50 mL). All the reactions were monitored by the Kaiser test. The Fmoc protecting group was removed with 50 mL of 20% piperidine in NMP twice. The mixture was agitated for 30 minutes, the solution was drained, and the resin was washed with 3 x 50 mL of DMF and 3 x 50 mL of CH₂Cl₂.

After all amino acids were coupled, the resin was treated with a solution containing 10% TFE, 10% AcOH, and 80% DCM for 3 hours at room temperature. The cleavage solution was then transferred to a round bottle flask with hexane and then concentrated in vacuo. Cold ether was added subsequently and white powder precipitated out immediately. The peptide was used without any purification.

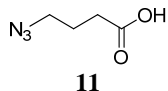
Scheme S1. The Synthesis of monomer **6**.



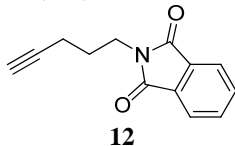
Compounds **10**, **11**, **12**, **13** were prepared according to the literature procedure.^{2,3} Their structures were confirmed by ¹H, ¹³C NMR and mass spectrometry, respectively.



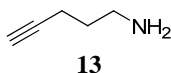
(10) Methyl 4-azidobutyrate. Methyl 4-chlorobutyrate (34.25g, 250 mmole) and sodium azide (22.5g, 346 mmole) were dissolved in DMSO (115 mL) in a 500mL round bottom flask. The mixture solution was heated to 45°C and stirred for 24h. After the mixture was cooled down to room temperature, 200 mL water was added and the mixture was extracted with Et₂O (3 x 100 ml). The combined organic layer was dried over anhydrous MgSO₄, filtered, and concentrated on a rotary evaporator to give **10** as colorless liquid (35.34g, 99%). ¹H NMR (500 MHz, CDCl₃) δ 3.60 (s, 2H), 3.36 (t, *J* = 6.8, 2H), 2.38 (t, *J* = 7.2, 2H), 1.81-1.75 (m, 2H); ¹³C NMR (125MHz, CDCl₃) δ 172.7, 51.4, 49.9, 30.4, 23.8; MS (ESI), *m/z* calcd for [C₅H₉N₃O₂ + H]⁺ = 144.1; found 144.1.



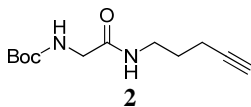
(11) 4-Azidobutyric acid. In a 500 mL flask compound **10** (33.1g 231 mmol) was dissolved in 2N NaOH (461 mL, 2.0eq) solution. A small amount of MeOH was added slowly to make it a homogeneous solution. The mixture was stirred at room temperature for 48 hours, then MeOH was removed under vacuo. The aqueous layer was washed with Et₂O (4 x 50 mL), then acidified with conc. HCl to pH 1. The compound was extracted with Et₂O (4 x 100 ml), and the combined organic fractions were dried over anhydrous MgSO₄, filtered, and concentrated on a rotary evaporator to give **11** as transparent colorless liquid (28.74g, 96%). ¹H NMR (500 MHz, CDCl₃) δ 10.72 (s, 1H), 3.36 (t, *J* = 6.4, 2H), 2.4 (t, *J* = 7.2, 2H), 1.92-1.86 (m, 2H); ¹³C NMR (125MHz, CDCl₃) δ 179.2, 50.4, 30.8, 23.8; MS (ESI), *m/z* calcd for [C₄H₇N₃O₂ + H]⁺ = 130.1; found 130.1.



(12) 1-Phthalimido-4-pentyne. In a 250 mL flask, 5-chloro-1-pentyne (5.13g, 50 mmol), phthalimide (8.83g, 60 mmol), K₂CO₃(5.13g, 50 mmol) and KI (100mg) were suspended in DMF (50 mL). The mixture solution was heated at 70°C for 16 hours. Then the solution was cooled down to room temperature and 50mL H₂O was added into solution. The mixture was extracted by Et₂O (200 mL x 4) and the combined organic layer was dried over anhydrous MgSO₄, filtered, and concentrated on a rotary evaporator. The crude residue was further purified by flash chromatography (DCM) to give **12** as white solid (10.24g, 96%). ¹H NMR (500 MHz, CDCl₃) δ 7.88-7.84 (m, 2H), 7.74-7.71 (m, 2H), 3.81 (t, *J* = 7.5, 2H), 2.28 (td, *J* = 7.0, 2.5, 2H), 1.97-1.92 (m, 3H); ¹³C NMR (125MHz, CDCl₃) δ 168.3, 133.9, 132.1, 82.9, 69.0, 37.1, 27.2, 16.2; MS (ESI), *m/z* calcd for [C₁₃H₁₁NO₂ + H]⁺ = 214.1; found 214.1.

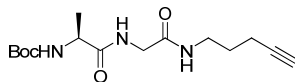


(13) 1-Amino-4-pentyne. In a 250 mL flask was added **12** (7.40g, 34.7 mmol) in EtOH (70 ml) to give a white suspension. Then hydrazine monohydrate was added into solution to give a clean solution. The reaction mixture was heated at 70°C under stirring for 2 hours, during which time much white solid precipitated. After the mixture solution was cooled down to room temperature, 50 mL H₂O was added into solution followed by addition of 2N HCl to adjust its pH to 3.5. The precipitate was filtered and the filtrate was concentrated using a rotary evaporator. The residue was cooled down to 0°C and treated with aq. NaOH (10N, 30 mL). Then the aqueous solution was extracted with DCM (3 x 100 mL). The combined organic layer was dried over anhydrous MgSO₄, filtered, and concentrated on a rotary evaporator to give **13** as green oil (2.16g, 74%). ¹H NMR (500 MHz, CDCl₃) δ 2.77 (t, *J* = 7.0, 2H), 2.22 (td, *J* = 7.0, 2.7, 2H), 1.92 (t, *J* = 2.7, 1H), 1.83 (s, 2H), 1.66-1.62 (m, 2H); ¹³C NMR (125MHz, CDCl₃) δ 83.7, 68.6, 40.7, 31.6, 15.7.



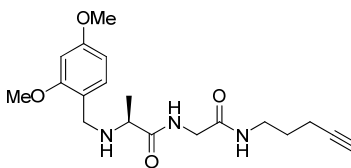
(2) Tert-butyl [2-oxo-2-(pent-4-yn-1-ylamino)ethyl]carbamate. Compound **13** (1.58g, 19.0 mmol) and Boc-Gly-OH (2.28g, 13.0 mmol) were dissolved in 100 mL DCM containing 4.56 mL DIPEA, into which EDC (3.64g, 19.0 mmol) and HOBT (2.57g, 19.0 mmol) were added. The mixture was stirred at room temperature for 12 hours. Then 100 mL of H₂O and 250 mL of EtOAc were added into mixture solution. The combined organic layer was washed with H₂O (100 mL × 1), 1M HCl (100

mL \times 1), sat. NaHCO₃ (100 mL \times 1), and brine, and then dried over anhydrous MgSO₄, filtered, and concentrated on a rotary evaporator to give **2** as white solid (3.06g, 98%). ¹H NMR (500 MHz, CDCl₃) δ 6.51 (s, 1H), 5.27 (s, 1H), 3.77 (d, *J* = 6.3, 2H), 3.39 (q, *J* = 6.3, 2H), 2.26-2.23 (m, 2H), 1.99 (s, 1H), 1.76-1.73 (m, 2H), 1.45 (s, 3H); ¹³C NMR (125MHz, CDCl₃) δ 169.5, 156.1, 83.4, 80.3, 69.2, 44.4, 38.6, 28.3, 27.9, 16.0; HRMS (ESI), *m/z* calcd for [C₁₂H₂₀N₂O₃ + Na]⁺ = 263.1372; found 263.1363.



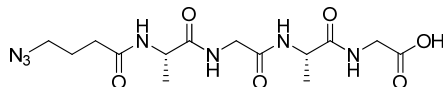
3

(3) Tert-butyl [(2S)-1-oxo-1-[[2-oxo-2-(pent-4-yn-1-ylamino)ethyl]amino]propan-2-yl]carbamate. In a 250 mL flask was added **2** (4.90g, 20.39 mmol) in a TFA/DCM solution (15 mL/10 mL). After stirring the solution at room temperature for 3 hours, TFA/DCM was removed under vacuum. The residue was dissolved in DCM and further evaporated in vacuo. The residue was dried under vacuum for 1 hour. Then the residue and Boc-Alanine were dissolved in DCM (50 mL) and cooled down in a ice bath. DIPEA was added into solution. EDC and HOBt were added into solution subsequently. The mixture was stirred at room temperature for 12 hours. Then 100 mL of H₂O and 250 mL of EtOAc were added into mixture solution. The combined organic layer was washed with H₂O (100 mL \times 1), 1M HCl (100 mL \times 1), sat. NaHCO₃ (100 mL \times 1), and brine, and then dried over anhydrous MgSO₄, filtered, and concentrated on a rotary evaporator. The residue was purified by flash chromatography (MeOH/DCM=1:9) to give **3** as white solid (5.70g, 90%). ¹H NMR (500 MHz, DMSO-*d*₆) δ 8.10 (s, 1H), 7.67 (s, 1H), 7.11 (d, *J* = 6.5, 1H), 6.51 (s, 1H), 6.47 (d, *J* = 8.5, 1H), 3.92-3.90 (m, 1H), 3.68-3.58 (qd, *J* = 16.4, 5.9, 2H), 3.18-3.06 (m, 2H), 2.78 (s, 1H), 2.17-2.13 (m, 2H), 1.60-1.56 (p, *J* = 7.0, 2H), 1.39 (s, 9H), 1.16 (d, *J* = 7.0, 3H); ¹³C NMR (125MHz, CDCl₃) δ 173.2, 168.9, 155.9, 83.4, 80.6, 69.0, 50.94, 43.2, 28.3, 28.0, 17.9, 16.0; HRMS (ESI), *m/z* calcd for [C₁₅H₂₅N₃O₄ + Na]⁺ = 334.1743; found 334.1742.



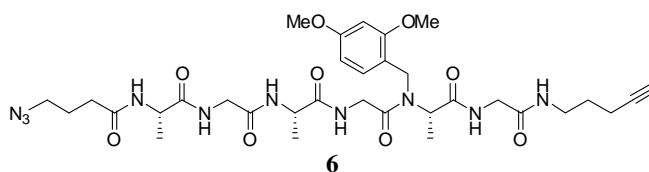
4

(4) (S)-2-(2,4-dimethoxybenzylamino)-N-(2-oxo-2-(pent-4-ynylamino)ethyl) propanamide. In a 100 mL flask, **3** (0.62g, 2.0 mmol) was dissolved in 20 mL TFA/DCM (1:1). The mixture was stirred at room temperature for 2 hours and then the solution concentrated on a rotary evaporator. The residue was added with 2,4-dimethoxybenzaldehyde (0.33g, 2.0 mmol), 0.35mL Et₃N and 10 mL MeOH. Then the mixture was added with NaCNBH₃ (0.15g, 2.4 mmol). The mixture was stirred at room temperature for 48 hours. Then the solution was concentrated on a rotary evaporator. The residue was purified by flash chromatography with EtOAc to give yellow gel-like product 0.60g (1.66 mmol, 83 %) of **4**. ¹H NMR (500 MHz, DMSO-*d*₆) δ 8.07 (s, 1H), 7.87 (s, 1H), 7.18 (d, *J* = 8.5, 1H), 6.51 (s, 1H), 6.47 (d, *J* = 8.5, 1H), 3.74 (d, *J* = 7.0, 4H), 3.70 (d, *J* = 5.5, 2H), 3.55 (q, *J* = 16.4, 2H), 3.13-3.10 (m, 3H), 2.78 (s, 1H), 2.17-2.15 (m, 2H), 1.59-1.56 (m, 2H), 1.14 (d, *J* = 7.0, 3H); ¹³C NMR (125MHz, DMSO-*d*₆) δ 168.5, 160.0, 159.2, 130.3, 104.3, 98.2, 84.0, 71.4, 56.3, 55.4, 55.2, 45.2, 41.8, 28.1, 18.6, 15.3; HRMS (ESI), *m/z* calcd for [C₁₉H₂₇N₃O₄ + H]⁺ = 362.2080; found 362.2074.



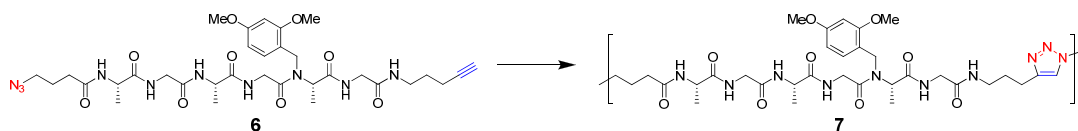
5

(5) (5S,11S)-16-azido-5,11-dimethyl-4,7,10,13-tetraoxo-3,6,9,12-tetraazahexadecan-1-oic acid. Peptide **5** was synthesized according to the standard solid phase peptide synthesis procedure as described earlier on 2-chlorotrityl chloride resin¹. Following cleavage from resin, the final product was purified by precipitation in Et₂O. The yield is 64% according to the initial loading on the resin. ¹H NMR (500 MHz, DMSO-*d*₆) δ 8.15 (t, *J* = 6.6, 3H), 7.92 (d, *J* = 7.0, 1H), 4.31-4.20 (m, 2H), 3.76-3.65 (m, 4H), 3.33-3.30 (t, *J* = 6.8, 2H), 2.20 (t, *J* = 7.0, 2H), 1.76-1.70 (m, 2H), 1.20 (t, *J* = 8.8, 6H); ¹³C NMR (125MHz, DMSO-*d*₆) δ 172.8, 172.4, 171.4, 171.1, 168.4, 50.3, 48.5, 48.0, 42.0, 40.7, 31.9, 31.0, 24.4, 22.1, 18.3, 17.8, 14.0; HRMS (ESI), *m/z* calcd for [C₁₄H₂₃N₇O₆ + Na]⁺ = 408.1608; found 408.1601.



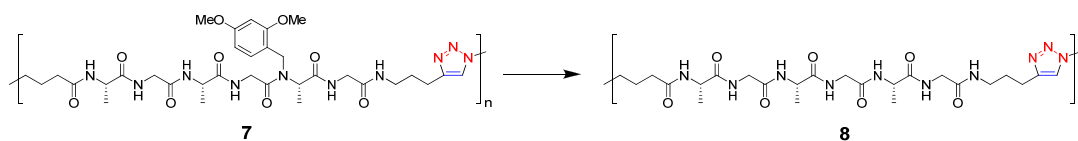
(6) 4-azido-N-((2S,8S,14S)-13-(2,4-dimethoxybenzyl)-8,14-dimethyl-3,6,9,12,15,18-hexaoxo-4,7,10,13,16,19-

hexaazatetracos-23-yn-2-yl)butanamide. In a 250 mL round bottom flask, **4** (0.365g, 1.01 mmol) and peptide **5** (0.39g, 1.01 mmol) were dissolved in 15 mL a DMF:DMSO (95:5) mixed solvent containing 0.265 mL DIPEA. Following addition of HATU (0.461g, 1.2 mmol), the mixture was stirred at room temperature for 48 hours. After addition of 100 mL of H₂O to the completed reaction solution, the mixture was extracted with EtOAc (50 mL × 2). The combined organic layer was washed with H₂O (100 mL × 1), 1M HCl (100 mL × 1), sat. NaHCO₃ (100 mL × 1), and brine, then dried over anhydrous MgSO₄, filtered, and finally concentrated on a rotary evaporator. The residue was purified by flash chromatography with MeOH/DCM (1:9) to give **6** as white solid (349mg, 46%). ¹H NMR (500 MHz, MeOD-*d*₄) δ 7.16 (d, *J* = 8.3, 1H), 6.53 (s, 1H), 6.48 (d, *J* = 8.3, 1H), 4.62 (d, *J* = 15.7, 1H), 4.40–4.37 (m, 2H), 4.22 (s, 2H), 4.17 (d, *J* = 7.2, 1H), 4.05 (d, *J* = 7.2, 1H), 3.89–3.58 (m, 10H), 3.43 (q, *J* = 7.0, 1H), 3.32–3.27 (m, 2H), 3.19–3.14 (m, 1H), 2.28 (t, *J* = 7.2, 2H), 2.16–2.10 (m, 3H), 1.86–1.81 (m, 2H), 1.72–1.70 (m, 2H), 1.36–1.10 (m, 9H); ¹³C NMR (125MHz, DMSO-*d*₆) δ 173.3, 172.7, 172.4, 172.0, 171.5, 171.1, 169.9, 169.2, 168.9, 168.6, 161.8, 160.6, 159.2, 158.2, 132.8, 129.3, 117.4, 105.2, 104.9, 98.8, 98.7, 84.6, 84.5, 71.9, 71.8, 56.1, 55.9, 55.8, 55.7, 55.3, 50.7, 39.0, 48.5, 48.5, 45.6, 44.0, 43.0, 42.4, 41.5, 38.1, 32.4, 31.2, 28.5, 24.8, 18.7, 18.2, 17.1, 15.83, 15.77, 15.0; HRMS (ESI), *m/z* calcd for [C₃₃H₄₈N₁₀O₉ + Na]⁺ = 751.3503; found 751.3494.



Procedure for [2+3] cycloaddition polymerization:

7 was prepared according to a modified literature procedure.⁵ Peptide monomer **6** (182mg, 0.25 mmol), copper acetate (4mg, 0.03 mmol), and 0.25 mL of N₂-degassed DMF were introduced into a small vial. Under stirring the mixture was heated at 80° C in an oil bath for 2 hours. The initial clear solution was transformed into a dark green gel. After cooling down the polymerization mixture with an ice bath, the gel was dissolved with additional DMF and then polymer was precipitated into 0.1 N HCl (20 mL). The precipitate was purified by consecutive three times of centrifugation and redispersion with 0.1 N HCl. A white solid precipitate was finally isolated and dried under vacuum to give 155mg product **7** (yield: 85%). The number average molecular weight (M_n) of polymer is 11500 and weight number average molecular weight (M_w) is 21750 based on GPC characterization. ¹H NMR (500 MHz, *d*₆-DMSO and *d*-TFA) δ 8.37–8.26 (m, 1H), 7.02 (s, 1H), 6.39–6.33(m, 2H), 4.42–4.07 (m, 7H), 3.83–3.20 (m, 11H), 2.73 (s, 2H), 2.17–2.08 (m, 4H), 1.80 (s, 4H), 1.19–1.10 (m, 9H).



Procedure for cleavage of 2,4-dimethoxybenzyl group:

In a 20 mL sample vial, polymer **7** (5mg) was dissolved in 1 mL TFA mixed solution (TFA/TIPS/H₂O=95:2.5:2.5). After stirring at room temperature for 60min, 9 mL of MeOH was introduced slowly into the vial. Cotton-like precipitate appeared overnight which was purified by three Times of centrifugation and redispersion with MeOH to give 3mg of polymer **8**. ¹H NMR (500 MHz, *d*₆-DMSO and *d*-TFA) δ 8.05 (s, 1H), 4.41 (s, 2H), 4.19–4.13 (m, 3H), 4.97–4.70 (m, 6H), 3.12 (s, 2H), 2.69–2.67 (m, 2H), 2.24–2.10 (m, 4H), 1.74 (s, 1H), 1.08–0.82 (m, 9H).

III. ^1H NMR spectra of polymer 7 and 8

After TFA/DCM deprotection, the proton peaks attributed to 2,4-dimethoxybenzyl group completely disappeared (spectra were taken in 50% *d*-TFA and DMSO-*d*6) (Fig. S1).

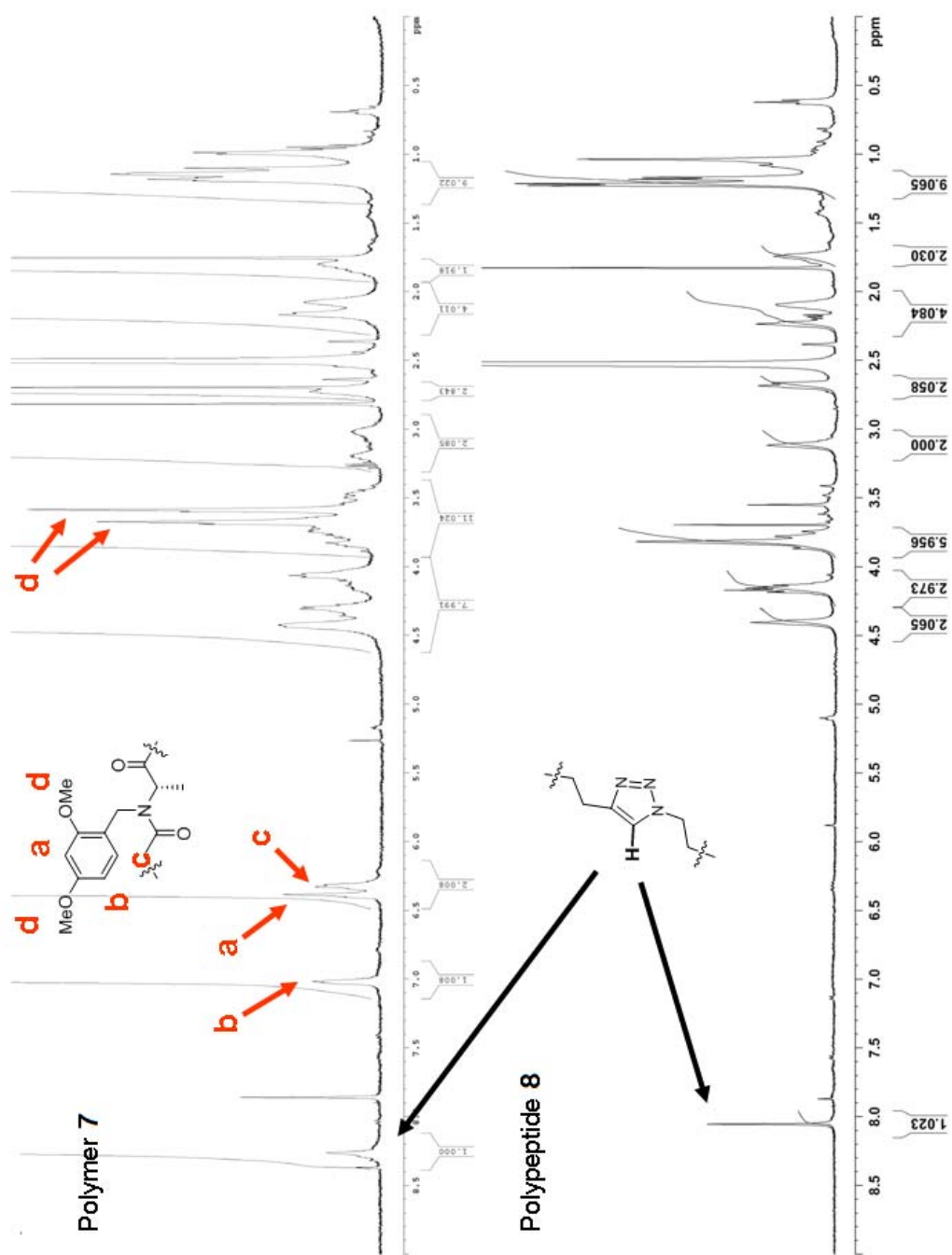


Figure S1. NMR spectra of polymer 7 and 8.

IV. Circular Dichroism (CD) measurement

Following literature procedure, we utilized CD to identify secondary structure for polymer **8**. 250 μL of peptide **8** solution (~ 2 mg/mL in hexafluoroisopropanol (HFIP)) was added in a 1.0-mm length UV cell. Far-UV CD spectra were measured at room temperature on a Jasco J-810 spectropolarimeter (band width: 1nm, response: 1sec., sensitivity: standard, 1 range: 250-190 nm, data pitch: 0.5 nm, scanning speed: 20 nm/min., accumulation: 5). A representative CD spectrum for **8** is shown in Fig. S2. The CD spectra obtained in hexafluoroisopropanol (HFIP) solution exhibit a minima around 206 nm and maxima around 195 nm, which has also been observed in biosynthetic poly(AG)₃YG or poly(AG)₃HG systems.⁶ The unusual shape of spectrum could result from either the formation of π - π stacking in the 1,2,3-triazole ring turn mimics or regular arrangement of β -turn, which might facilitate coupling of the dipole transition moments.^{6a}

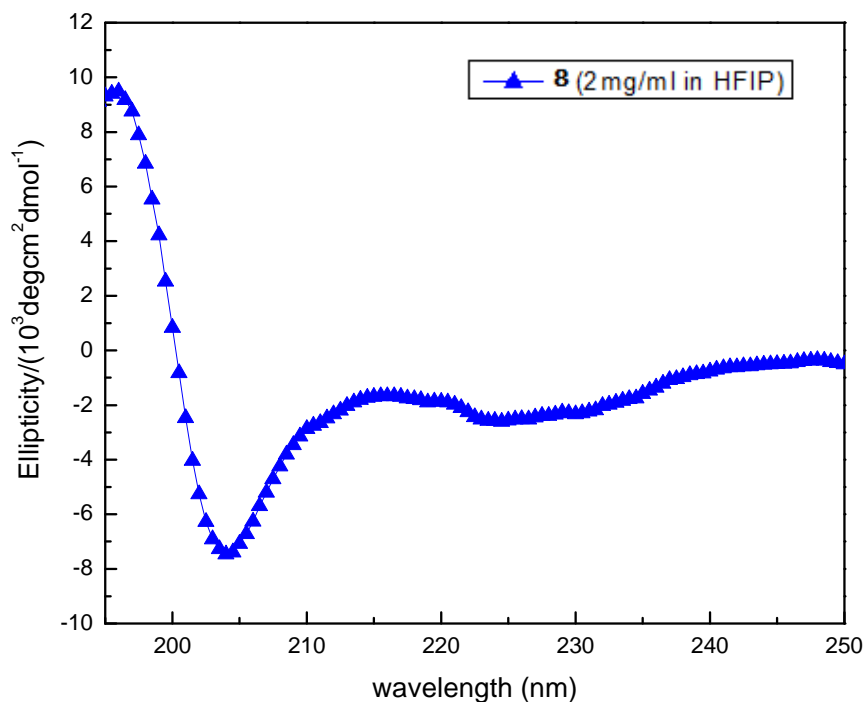


Figure S2. Circular dichroism spectrum of polypeptide **8**.

V. FTIR characterization of polymer 7 and polymer 8

Infrared (IR) spectra were recorded on a MIDAC Grams/Prospect FT-IR spectrometer using a KBr pellet. All powder samples and KBr were ground to reduce the particle size. Then the mixture powder was pressed into a pellet. The IR spectra of polymer **7** revealed a major strong amide I vibration band at ~ 1655 , indicating random coil secondary conformation in peptide (Figure S3a).⁷ The IR spectra of polymer **8**, on the other hand, exhibited strong amide I and II vibrational bands at ~ 1628 cm^{-1} and ~ 1533 cm^{-1} , characteristic of the β -sheet conformation.⁶ In addition, the weak amide I component observed at ~ 1695 cm^{-1} indicates its antiparallel nature (Figure S3b).⁷ The amide I band ~ 1650 cm^{-1} indicates that some fraction of polymer chain has adopted a secondary structure other than that of the antiparallel β -sheet, where may result from breaking of the crystalline structure while physically mixing sample with KBr salt in the sample preparation.

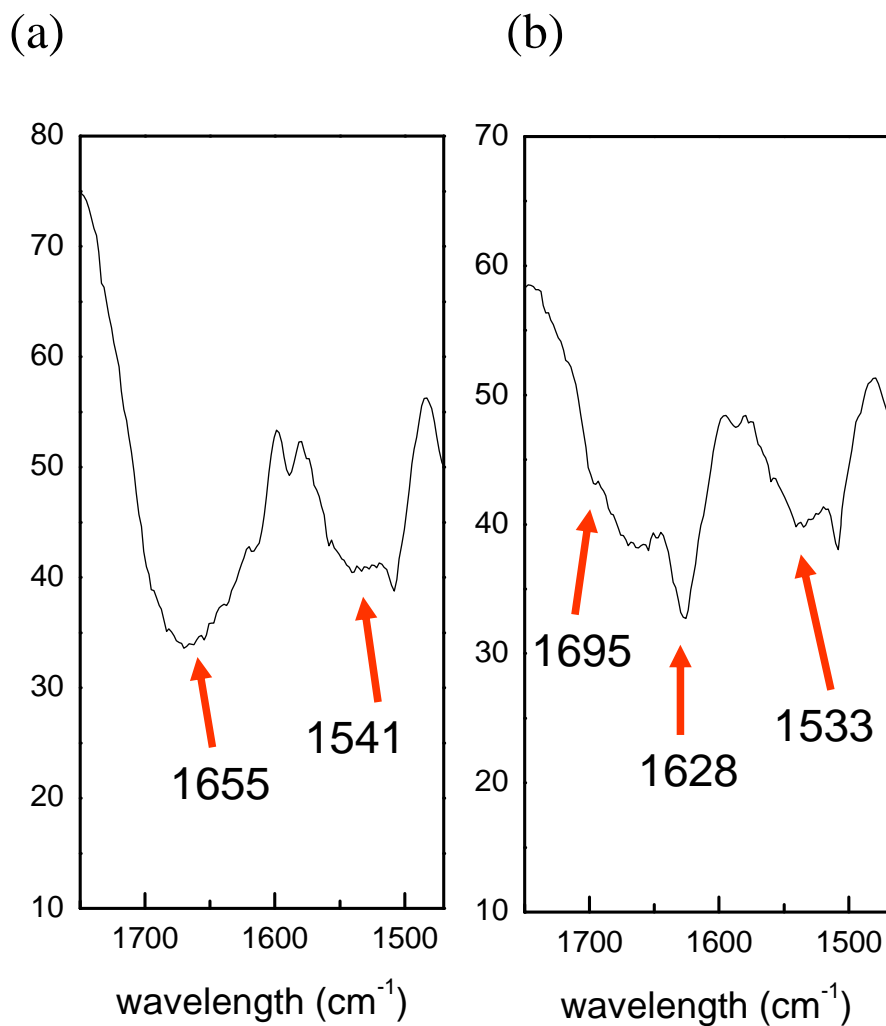


Figure S3. Infrared (IR) spectra of (a) **7** (b) **8**.

VI. Transmission Electron Microscopy (TEM) Analysis

Sample Preparation of Transmission Electron Microscopy: A drop of polymer **8** solutions which crystallized in MeOH/TFA solution with concentrations of 0.5 mg/mL was incubated onto a carbon-coated TEM grid for 1 min, then blotted with filter paper. The grid was floated with 2% uranyl acetate solution. After 60sec incubation, the excess uranyl acetate solution was blotted with filter paper. The sample grid was then dried in vacuo overnight. The investigations were performed on a FEI/Philips CM-20 conventional TEM equipped with an EDAX EDS system operated at an acceleration voltage of 200 kV. Images were recorded either on a CCD camera for bright field imaging. Representative images and histogram of fibril width are shown in Fig. S4 &5, respectively.

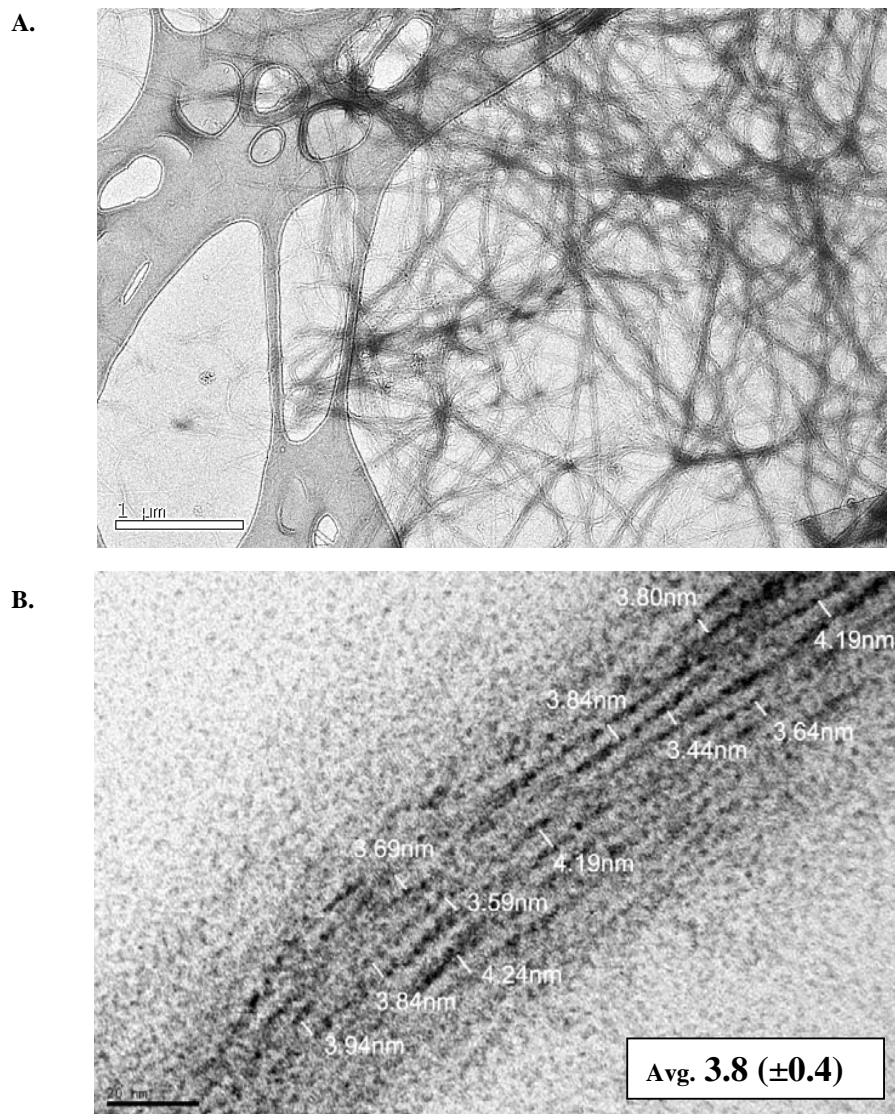


Figure S4. TEM images of fibril formed by polypeptide **8**. (A) Nanofibrils on TEM grid; (B) A zoom up view of one nanofibril. Scale bar on top image: 1 μ m, scale on bottom image: 20nm.

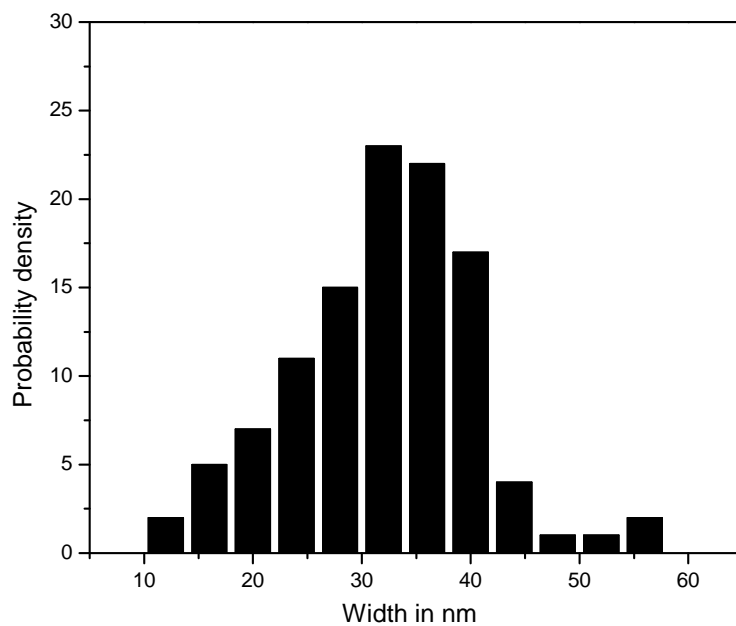


Figure S5. Probability density plot of nanofibril width measured from TEM images of **8**. A total of 110 samplings data were employed in constructing this distribution profile.

VII. Atomic Force Microscopy (AFM) Analysis

Polymer solutions with concentrations of 0.5 mg/mL were spin-coated onto a freshly cleaved mica surface. The morphologies of the samples were analyzed by means of tapping-mode AFM with a Veeco Multimode instrument with a NanoScope IV controller operating in air at room temperature. Height (Figure S6) and phase images were recorded with microfabricated silicon cantilevers (FESP probes, Veeco, Santa Barbara). Set point was chosen at attractive regime. Representative images, height profiles and histogram of height for nanofibrils are shown in Fig. S6-8.



Figure S6. AFM micrograph of fibril formed by polypeptide **8**.

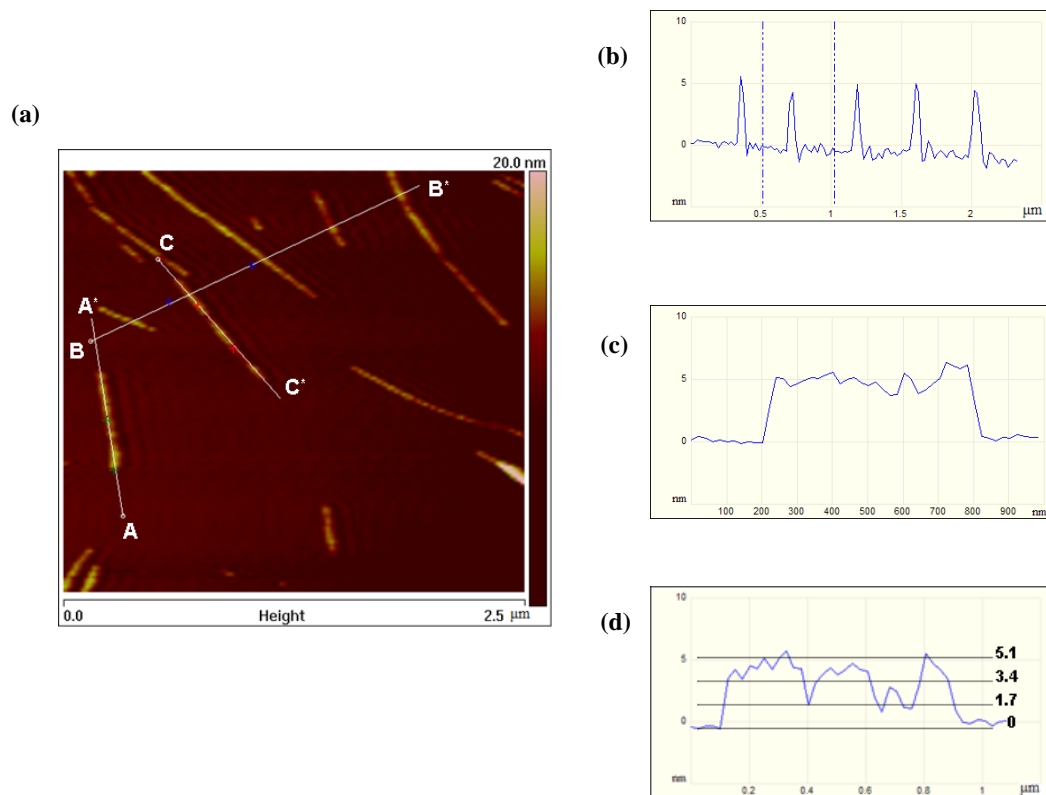


Figure S7. (a) AFM micrograph of fibril formed by polymer **8**. (b) height along A to A* on the fibril. (c) height along B to B* on the fibril. (d) height along C to C* on the fibril.

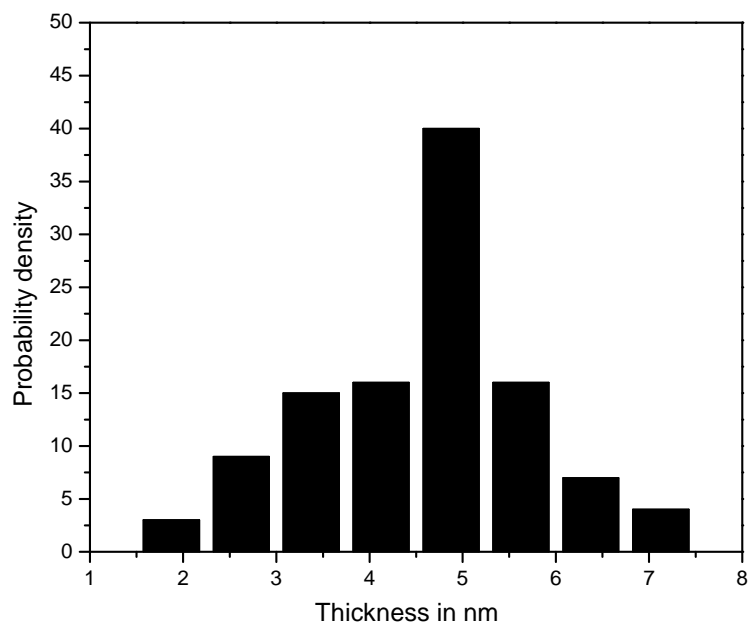


Figure S8. Probability density plot of nanofibril height measured from AFM topographic data of **8**. A total of 110 samplings data were employed in this distribution profile.

VIII. Wide Angle X-Ray Diffraction (WXR) analysis of 8

The powder of sample was prepared by removing the solvent from the crystallized fibril solution. The crystalline powder sample was packed into a thin-walled glass tube. The powder of the samples were analyzed by means of Four Circle Wide Angle X-ray Spectrometer (WAXS) equipped with an OSMIC double focusing multilayer monochromator and a MAR 345 mm diameter image plate detector. The X-ray source: 18 kW Rigaku rotating anode x-ray generator (Cu anode, $\lambda = 1.54 \text{ \AA}$). The major reflections at d spacings of 4.57, 4.22, and 3.64 \AA (Fig. S9), which are consistent with the antiparallel β -sheet d spacings reported for the unoriented fibroin film⁸ and the polymer (AG)_n made by genetic engineering.⁹

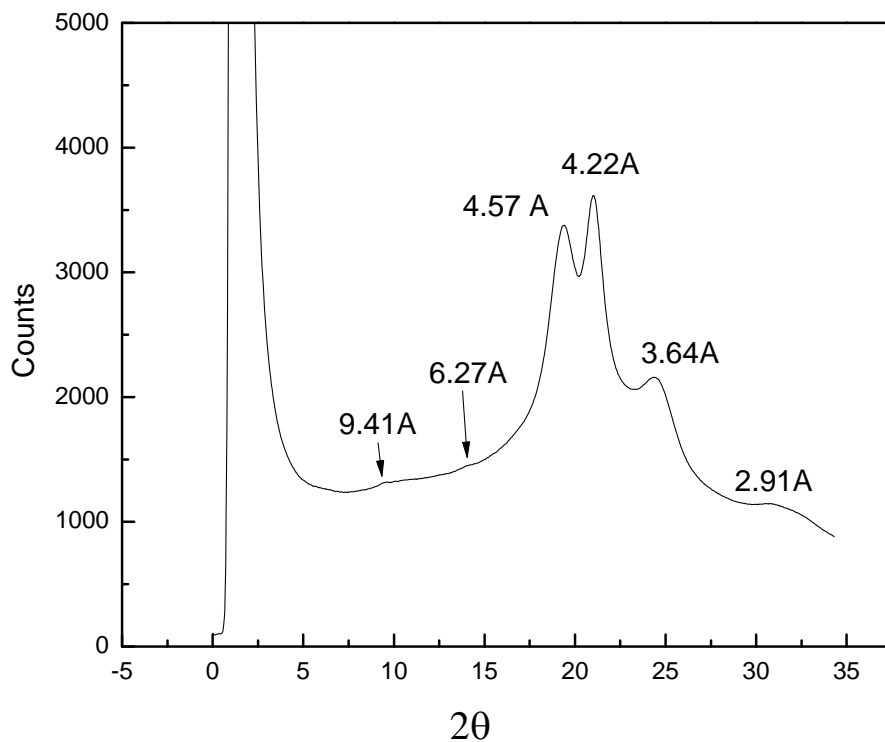


Figure S9. Wide angle X-ray diffraction of solid Polypeptide 8.

IX. Reference

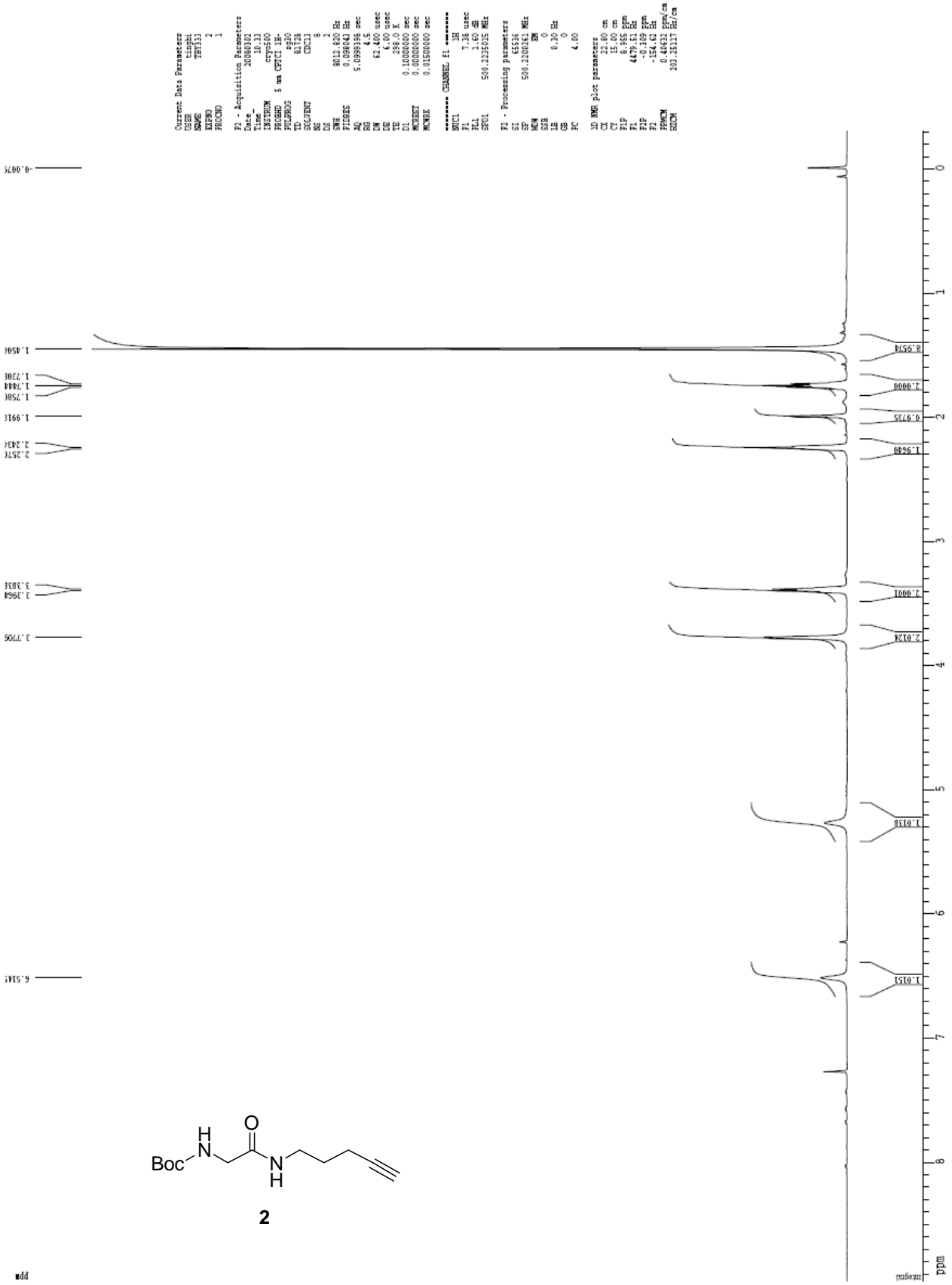
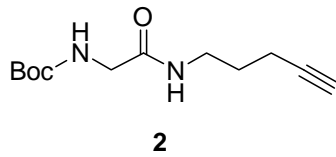
- [1] *NovaBiochem Catalog*, **2003**.
- [2] K. Oh, Z. Guan, *Chem. Comm.* **2006**, 3069.
- [3] (a) T. E. Muller, A. -K. Pleier, *J. Chem. Soc., Dalton Trans.* **1999**, 583.
(b) D. Perez, G. Bures, E. Guitian, L. Castedo, *J. Org. Chem.* **1996**, *61*, 1650.
(c) E. C. Taylor, J. E. Macor, J. L. Pont, *Tetrahedron* **1987**, *43*, 5145.
- [4] N. Khoukhi, M. Vaultier, R. Carrie, *Tetrahedron* **1987**, *43*, 1811.
- [5] M. Van Dijk, K. Mustafa, A. C. Dechesne, C. F. Van Nostrum, ; W. E. Hennink, D. T. S. Rijkers, R. M. J. Liskamp, *Biomacromolecules*, **2007**, *8*, 327.
- [6] (a) N. I. Topilina, S. Higashiya, N. Rana, V. V. Ermolenkov, C. Kossow, A. A. Carlsen, S. C. Ngo, C. C. Wells, E. T. Eisenbraun, K. A. Dunn, I. K. Lednev, R. E. Geer, A. E. Kaloyeros, J. T. Welch, *Biomacromolecules* **2006**, *7*, 1104.
(b) I. K. Lednev, V. V. Ermolenkov, S. Higashiya, L. A. Popova, N. I. Topilina, J. T. Welch, *Biophys. J.* **2006**, *91*, 3805.
(c) S. Higashiya, N. I. Topilina, S. C. Ngo, D. Zagorevskii, J. T. Welch, *Biomacromolecules* **2007**, *8*, 1487.
(d) M. W. T. Werten, A. P. H. A. Moers, T. Vong, H. Zuilhof, J. C. M. van Hest, F. A. de Wolf, *Biomacromolecule* **2008**, *9*, 1705.
(e) J. Safar, P. P. Roller, G. C. Ruben, D. C. Gajdusek, C. J. Jr. Gibbs, *Biopolymers* **1993**, *33*, 1461.
- [7] (a) T. Miyazawa, E. R. Blout, *J. Am. Chem. Soc.* **1961**, *83*, 712.
(b) P. I. Haris, D. Chapman, *Biopolymers* **1995**, *37*, 251.

X. NMR and FTIR spectra

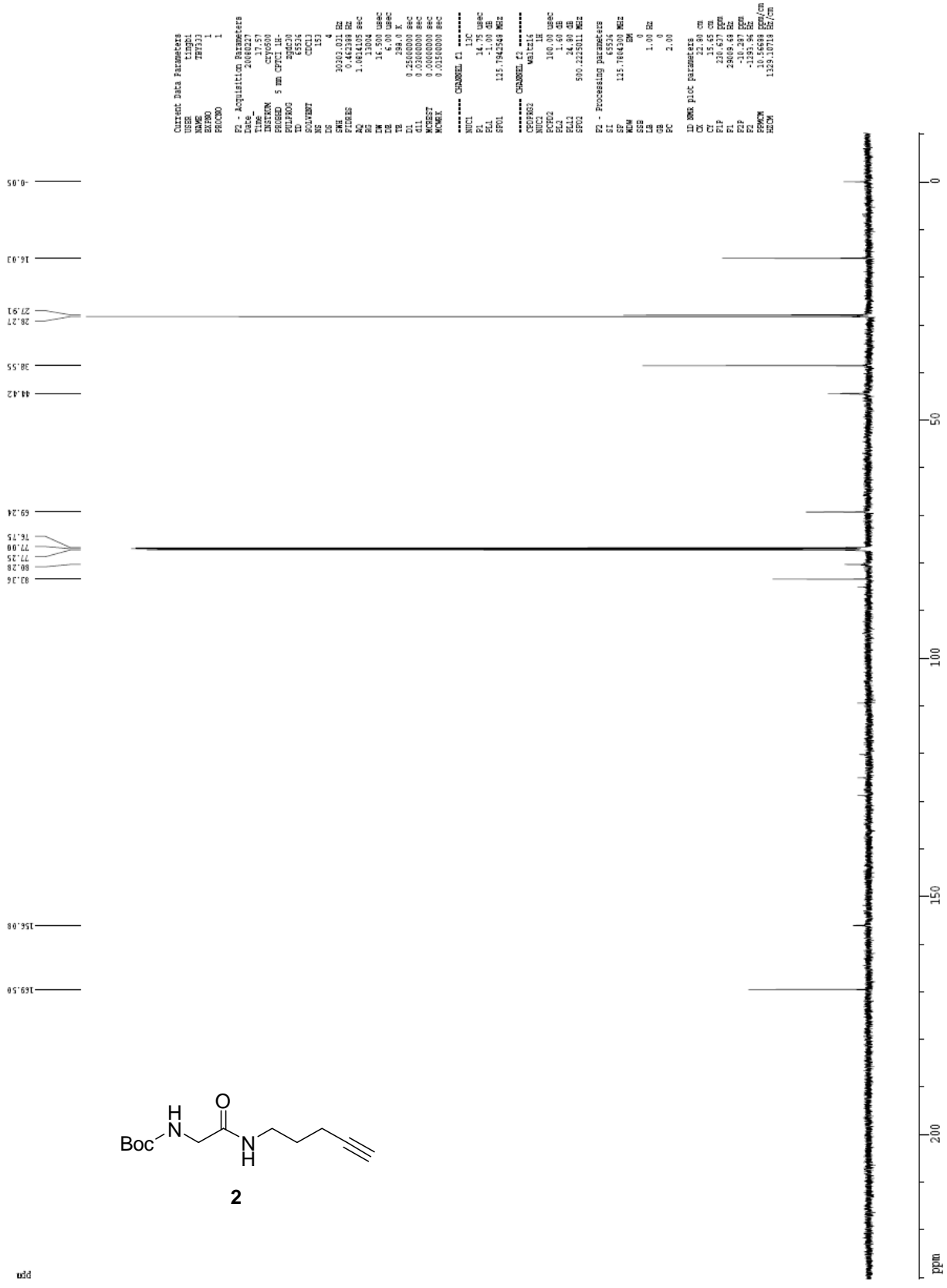
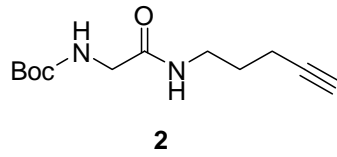
¹H NMR and ¹³C NMR spectra of 2

TBY333-H1-CDCl3
¹H spectrum

ppm

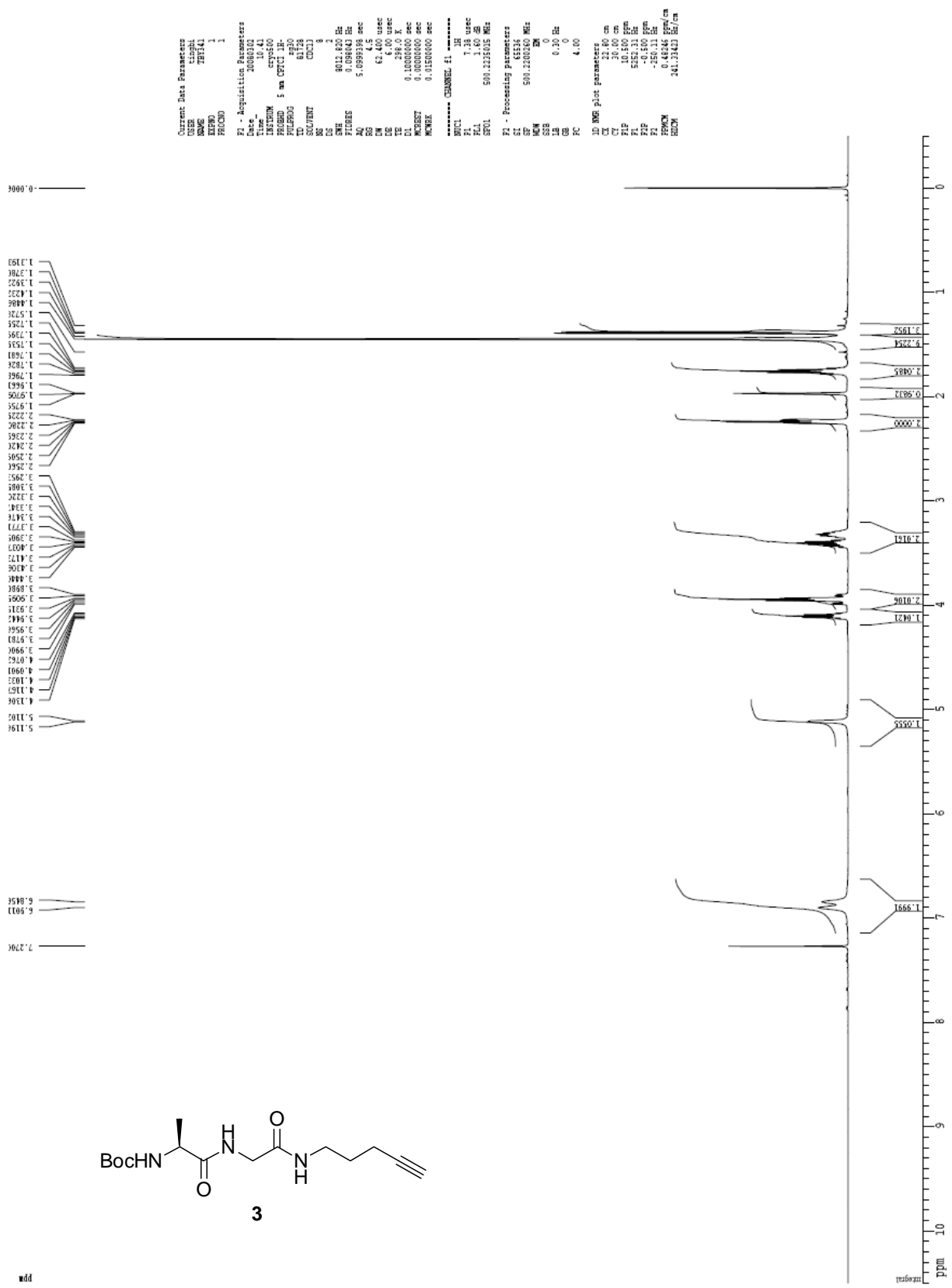
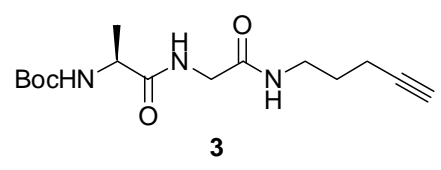


TBY333-Cl3-CDC13
 13C spectrum with 1H decoupling



¹H NMR and ¹³C NMR spectra of **3**

TBY341-H1-CDC13
¹H spectrum
 pdf



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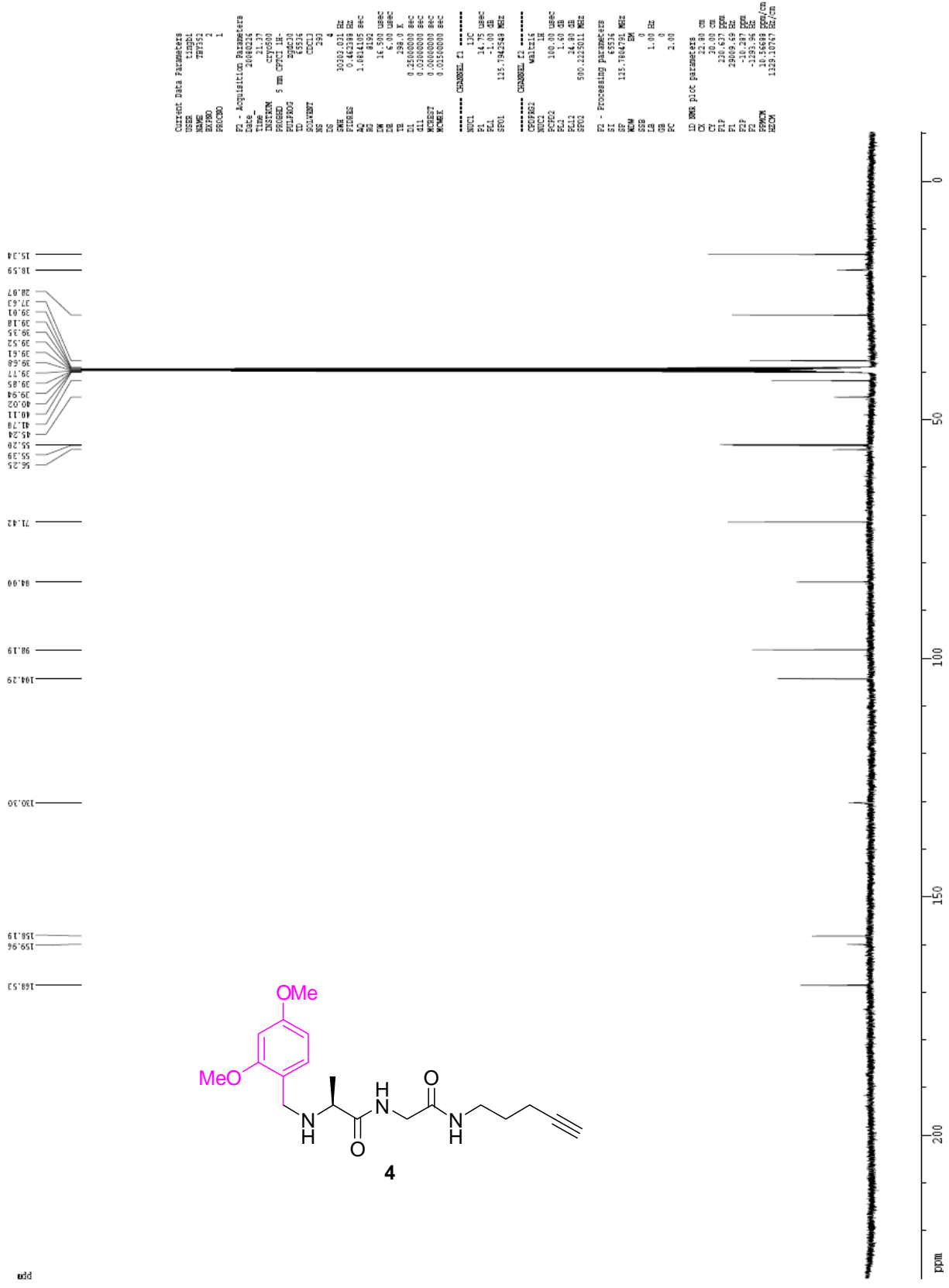
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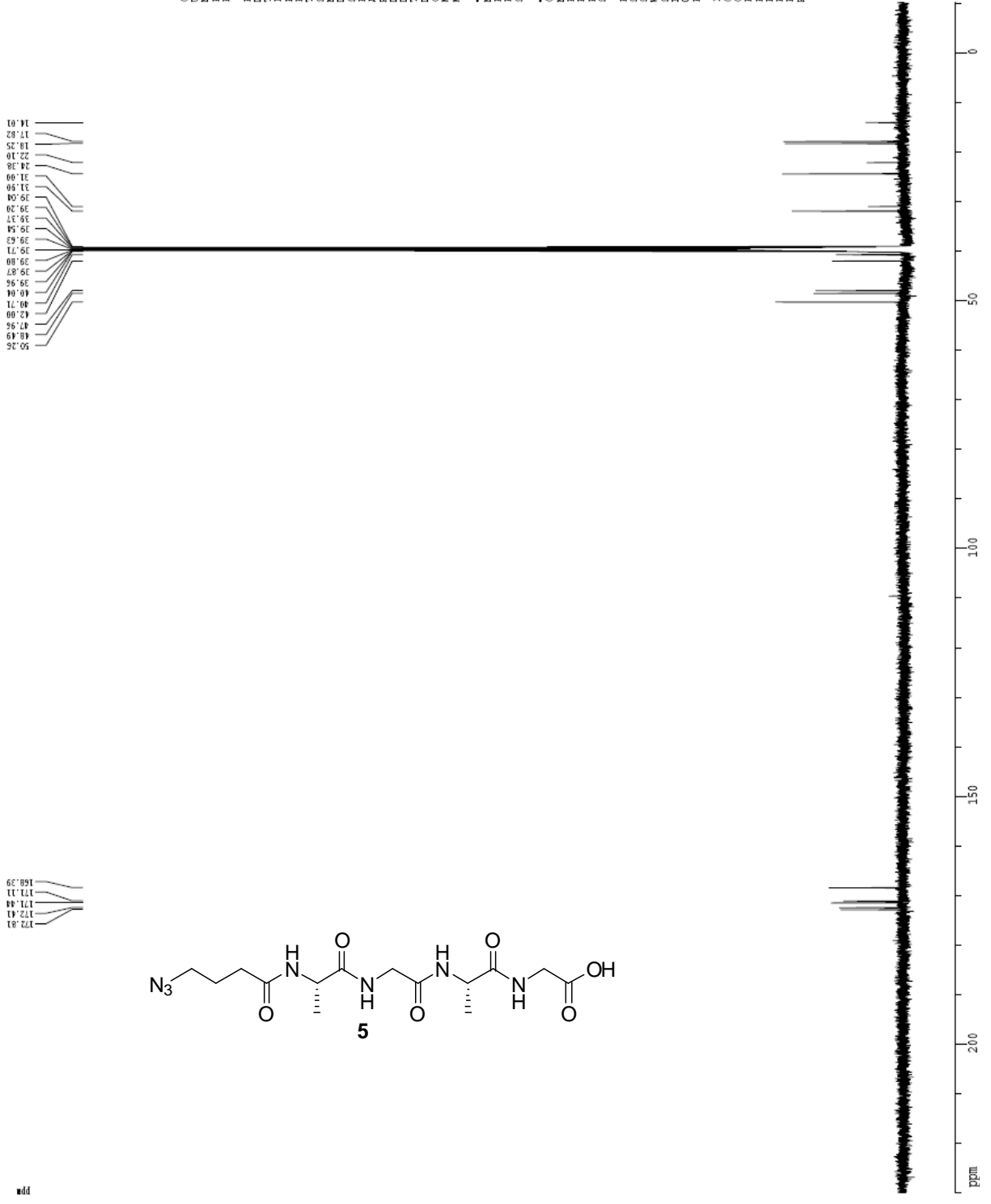
TBY352-Cl3-DMSO
 13C spectrum with 1H decoupling



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1B1382-Cl3-DMSO
 13C spectrum with 1H decoupling



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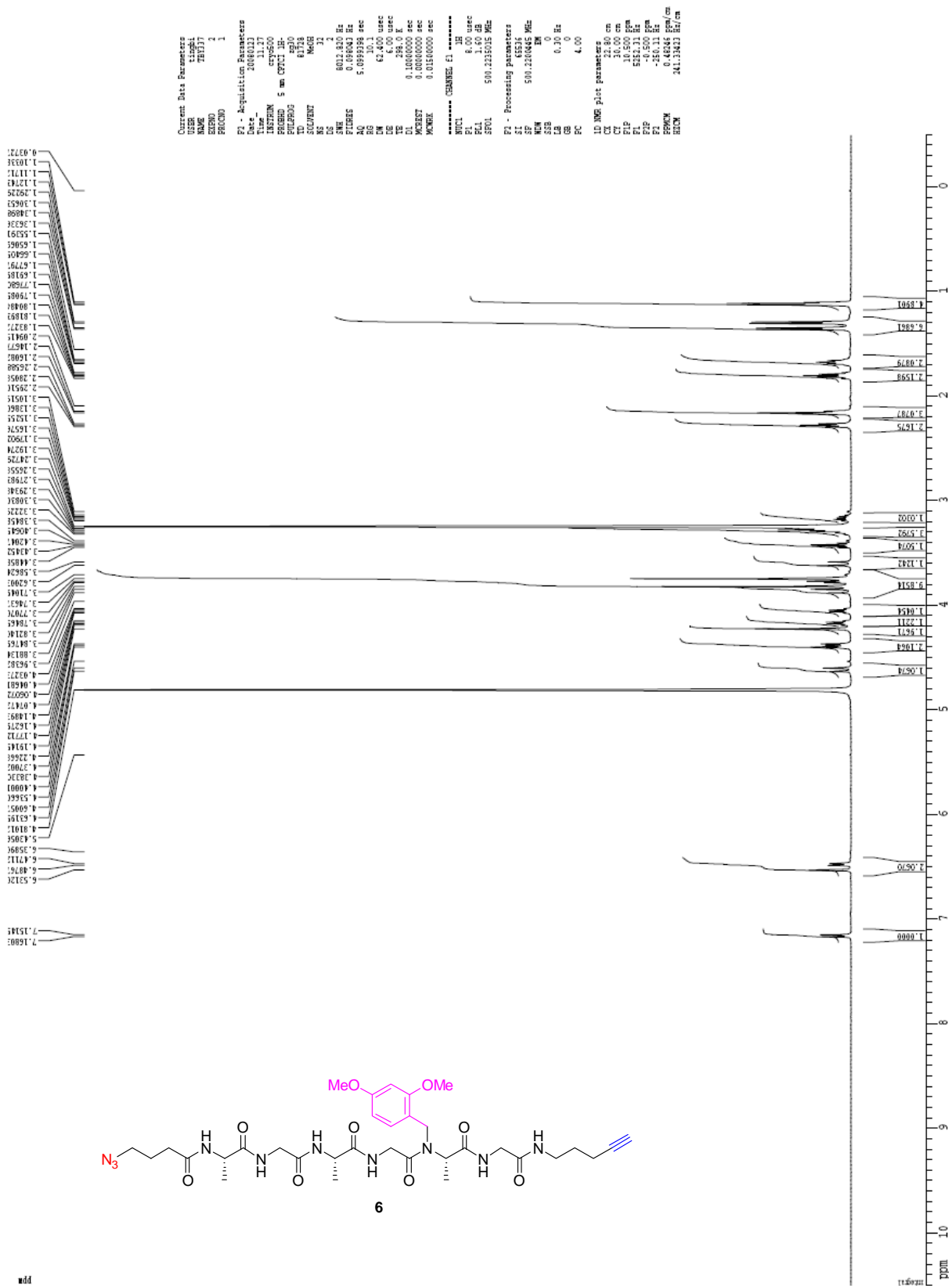
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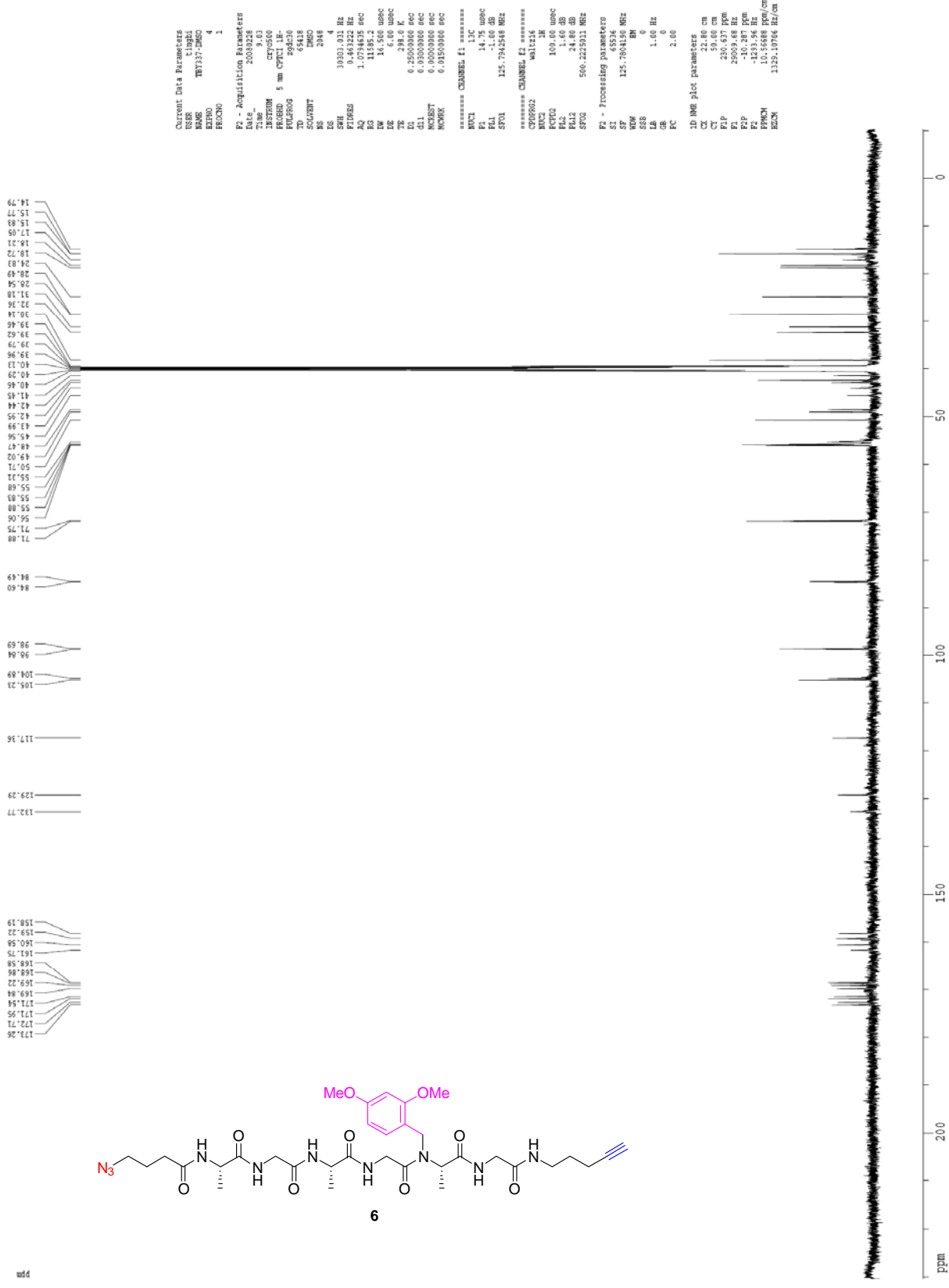
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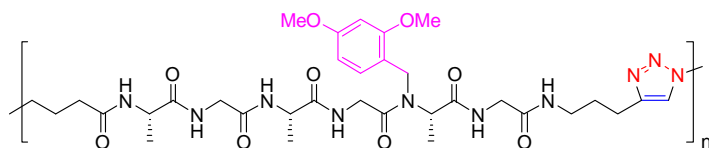
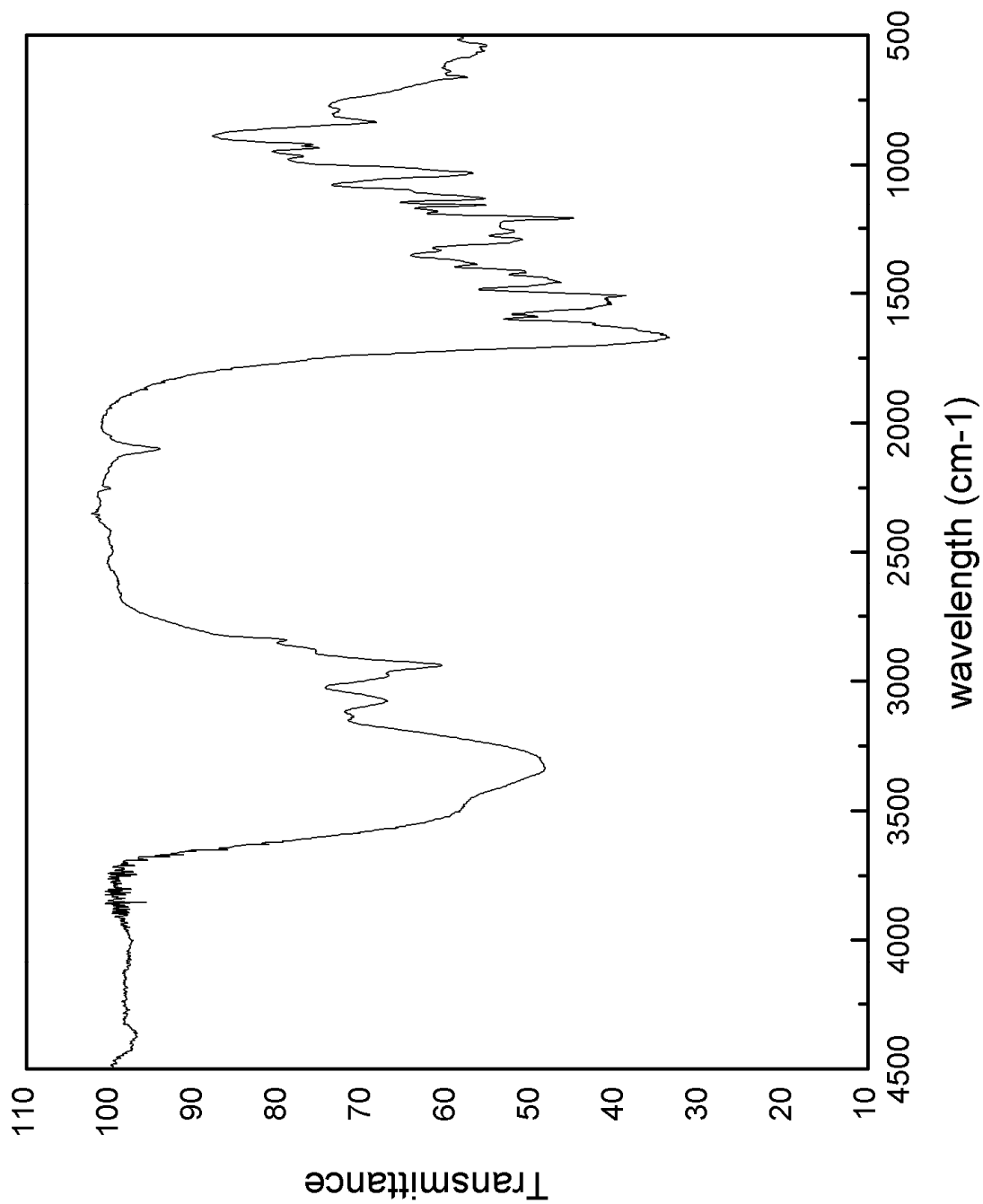
¹H NMR, COSY, ¹³C NMR spectra of **6**

IBY337-1H-NEOH
1H spectrum

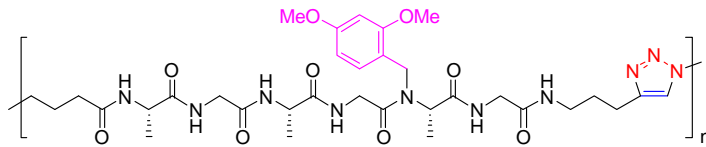


1B1337-Cl3-DMSO
 13C spectrum with 1H decoupling



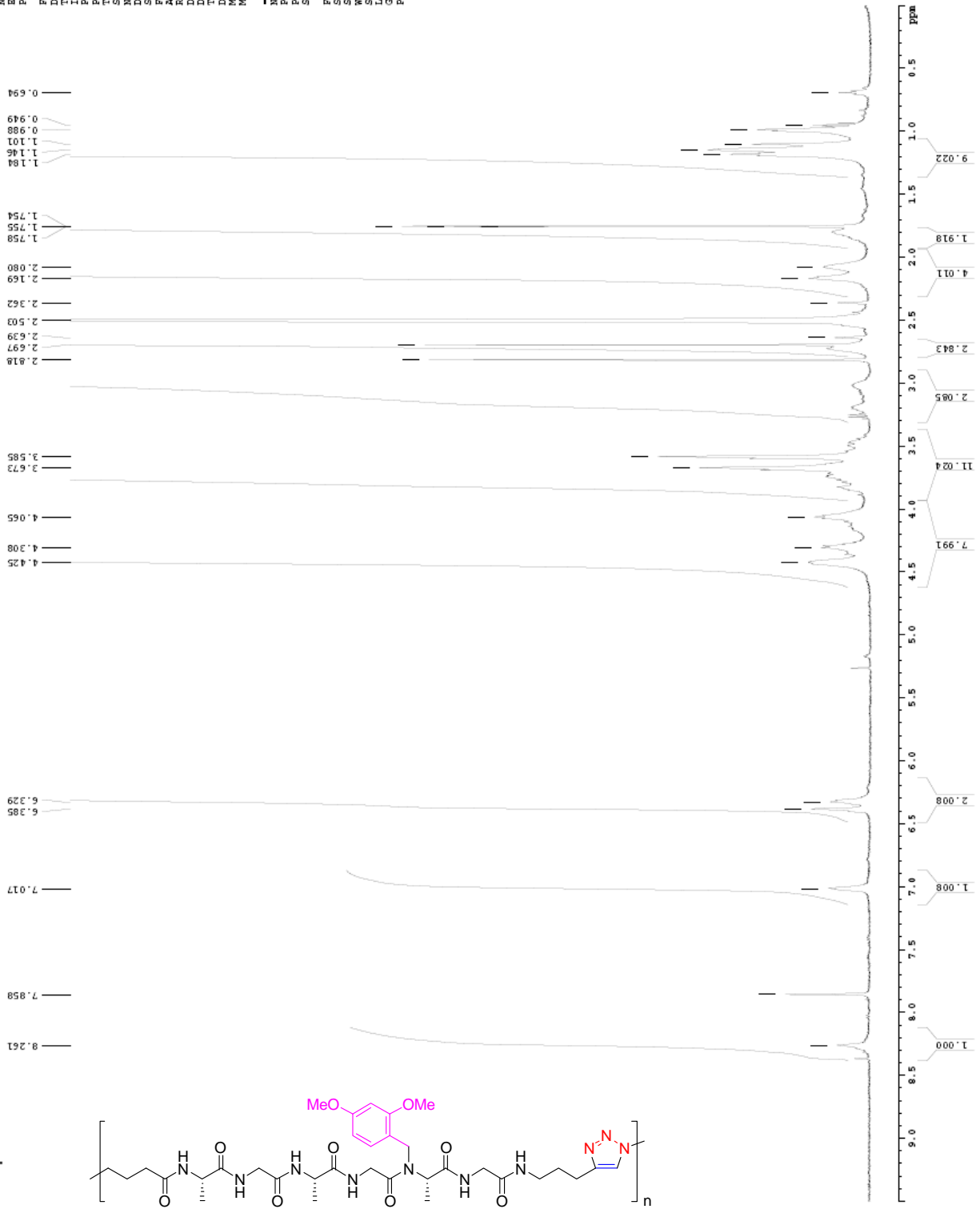


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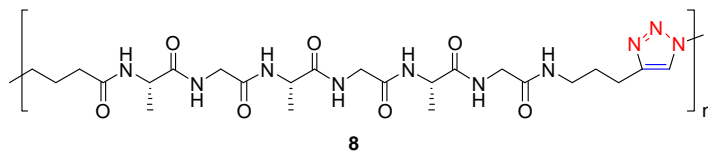
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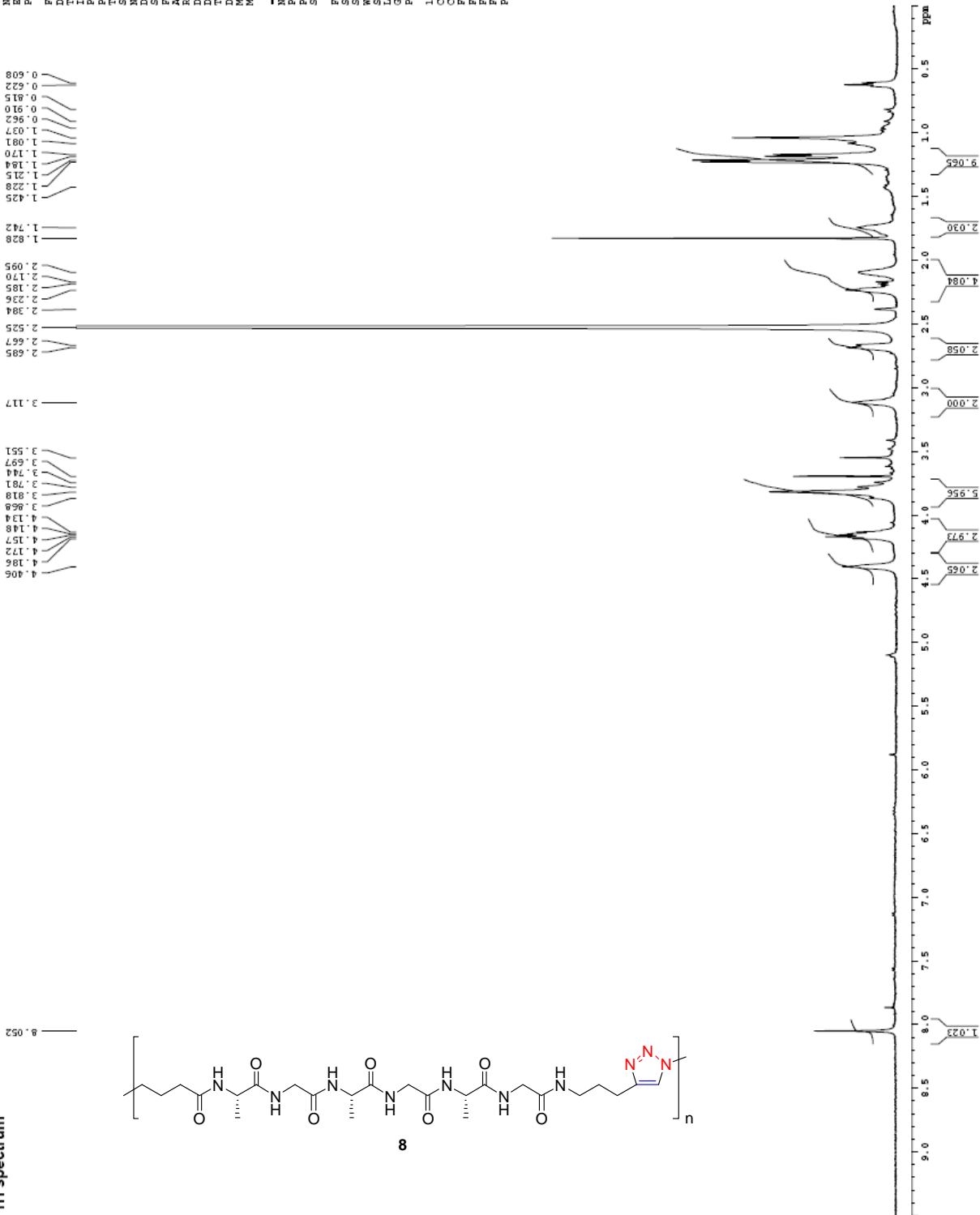
¹H NMR and IR spectra of **8**

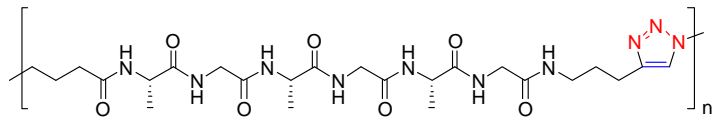
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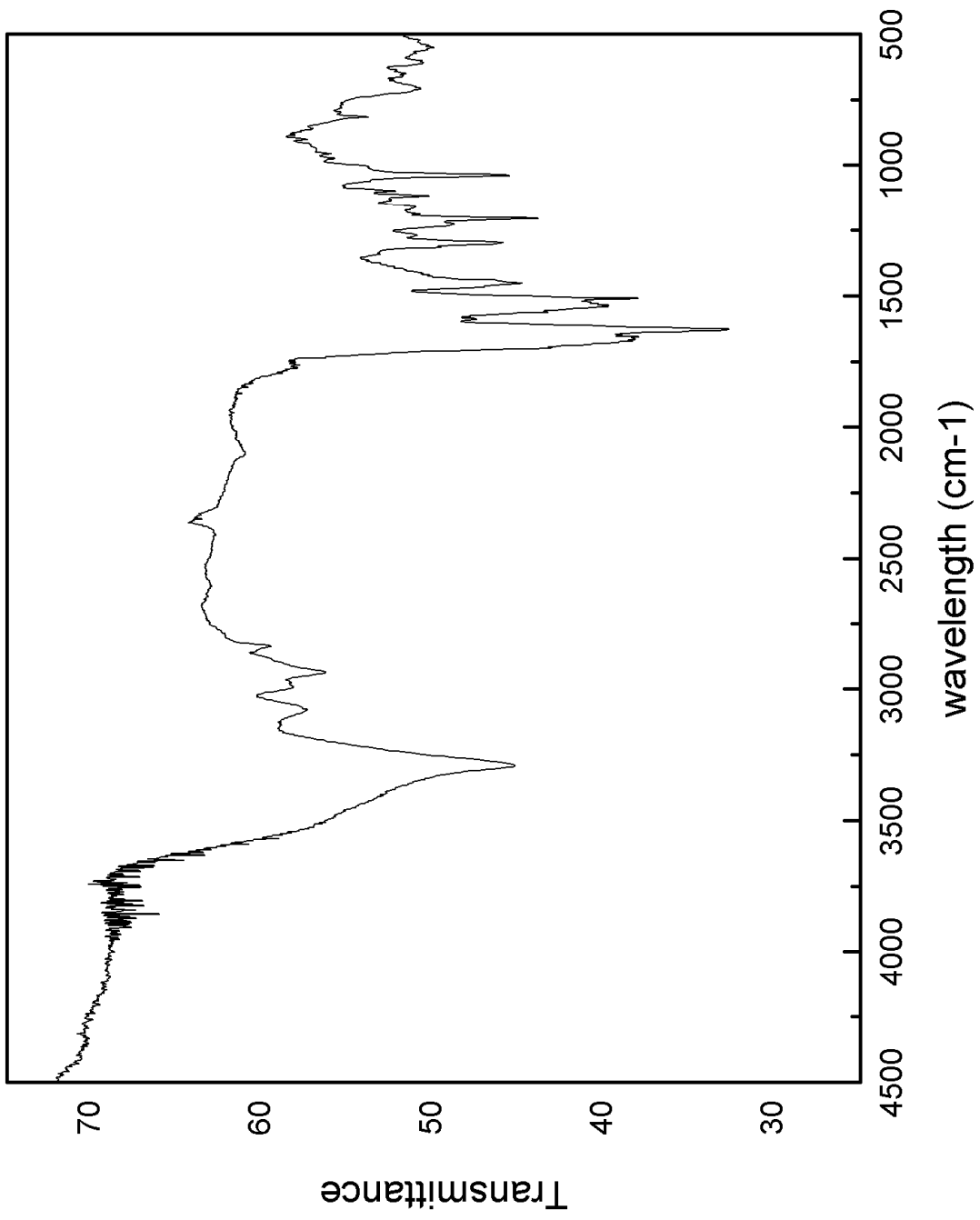
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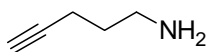
8



¹H NMR and ¹³C NMR spectra of **13**

1B1329-H1
1H spectrum

add



13

