

## Supporting Information

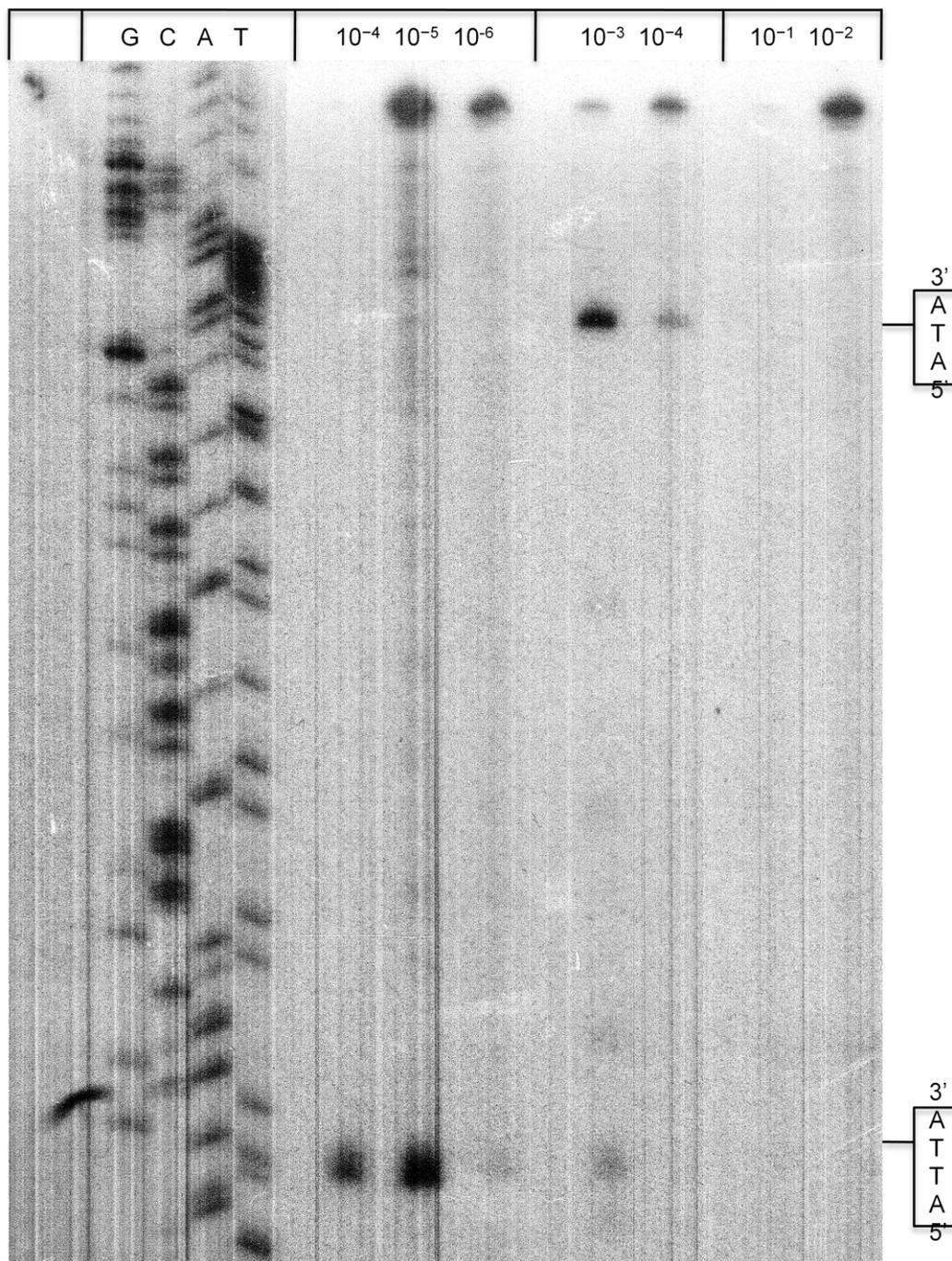
# A Novel, Unusually Efficacious Duocarmycin Carbamate Prodrug That Releases No Residual Byproduct

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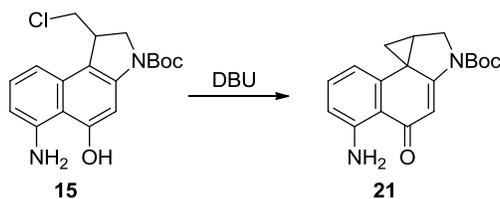
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**Figure S1.** Thermally induced strand cleavage of w794 DNA; DNA-agent incubation at 23 °C for 48 h, removal of unbound agent by EtOH precipitation, and 30 min of thermolysis (100 °C) followed by 8% denaturing PAGE and autoradiography. Lane 1, control DNA; lanes 2–5, Sanger G, C, A, and T sequencing reactions; lanes 6–8, (+)-**1** ( $1 \times 10^{-4}$  to  $1 \times 10^{-6}$ ); lanes 9–10, (-)-**1** ( $1 \times 10^{-3}$  to  $1 \times 10^{-4}$ ); lanes 11–12, (+)-**6** ( $1 \times 10^{-1}$  to  $1 \times 10^{-2}$ ).



***N*-tert-Butyloxycarbonyl-5-amino-1,2,9,9a-tetrahydrocyclopropa[*c*]benzo[*e*]indole-4-one** (***N*-Boc-ACBI, 21**). Compound **15** (4 mg, 11.4  $\mu\text{mol}$ ) in 0.2 mL of acetonitrile was treated with 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU, 6  $\mu\text{L}$ , 0.043 mmol). The reaction mixture was allowed to stir at room temperature for 90 min. After 90 min, the solvent was evaporated under reduced pressure and the residue was purified by PTLC ( $\text{SiO}_2$ , 50% EtOAc/hexanes) to provide **21** (3.5 mg, 100% yield) as an orange oil.  $^1\text{H}$  NMR (acetone- $d_6$ , 600 MHz)  $\delta$  7.25 (br, 1H), 7.12 (t,  $J = 7.8$  Hz, 1H), 6.59 (d,  $J = 8.4$  Hz, 1H), 6.12 (d,  $J = 7.2$  Hz, 1H), 4.00–3.94 (m, 2H), 2.85 (m, 1H), 1.53 (s, 9H), 1.45–1.40 (m, 2H).  $^{13}\text{C}$  NMR (acetone- $d_6$ , 150 MHz)  $\delta$  191.0, 159.6, 153.1, 152.8, 144.3, 134.0, 116.3, 115.2, 109.9, 109.5, 85.5, 54.2, 35.0, 31.9, 29.0, 25.5. ESI-TOF HRMS  $m/z$  313.1553 ( $\text{M}+\text{H}^+$ ,  $\text{C}_{18}\text{H}_{20}\text{N}_2\text{O}_3$  requires 313.1547).

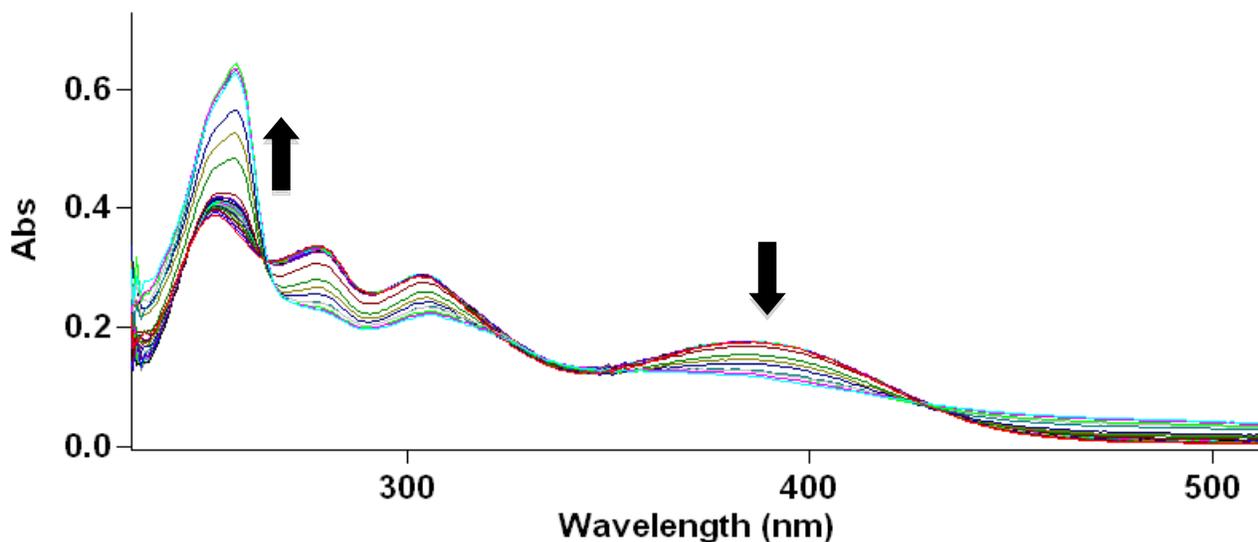
**Solvolysis of 21.** Compound **21** was dissolved in  $\text{CH}_3\text{OH}$  (1.5 mL). The  $\text{CH}_3\text{OH}$  solution was mixed with aqueous buffer (pH 2, 1.5 mL). The buffer contained 4:1:20 (v:v:v) 1.0 M citric acid, 0.2 M  $\text{Na}_2\text{HPO}_4$ , and  $\text{H}_2\text{O}$ , respectively. After mixing, the solvolysis solutions were stoppered and kept at 25  $^\circ\text{C}$  in the dark. The UV spectrum of the solutions was measured 3–4 times in the first two days and once a day for 2–4 weeks. The UV monitoring was continued until no further change was detectable. The long-wavelength absorption at 380 nm and short-wavelength absorption at 255 nm were monitored. The solvolysis rate constant and half-life were calculated from the data recorded at the short wavelength (255 nm) from the least square treatment of the slopes of plots of time versus  $\ln [(A_{\text{Final}} - A_{\text{Initial}})/(A_{\text{Final}} - A)]$ .

pH 1 buffer: 10 M citric acid: 0.2 M  $\text{Na}_2\text{HPO}_4$ :  $\text{H}_2\text{O}$  (4:1:20)

$$t_{1/2} = 8.52 \text{ h}, k = 1.5 \times 10^{-5} \text{ s}^{-1}$$

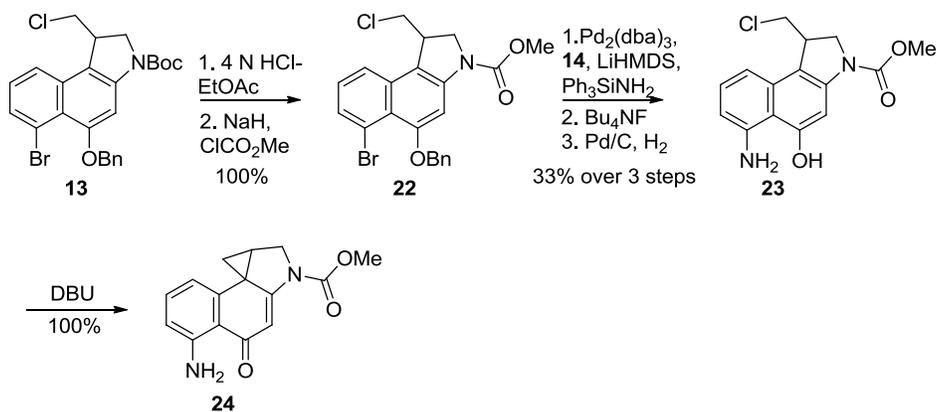
pH 2 buffer: 1.0 M citric acid: 0.2 M  $\text{Na}_2\text{HPO}_4$ :  $\text{H}_2\text{O}$  (4:1:20)

$$t_{1/2} = 40.3 \text{ h}, k = 5 \times 10^{-6} \text{ s}^{-1}$$



**Figure S2.** UV-visible spectra of **21** in 50% CH<sub>3</sub>OH-aqueous buffer (pH 2) recorded at various time intervals.

**Scheme S1**



**Methyl 5-(Benzyloxy)-6-bromo-1-(chloromethyl)-1H-benzo[e]indole-3(2H)-carboxylate (22).**

Compound **13** (50 mg, 0.099 mmol) was dissolved in 4 N HCl in EtOAc (2.0 mL) and the mixture was allowed to stir at room temperature for 15 min. The solvent was removed under a stream of nitrogen and the residue was redissolved in anhydrous DMF (0.9 mL) and the solution was cooled to 0 °C. Once cooled, 60% NaH in mineral oil (9.5 mg, 0.23 mmol) was added and the reaction mixture was allowed to stir at 0 °C for 30 min. Methyl chloroformate (36 μL, 0.47 mmol) was added and the solution was warmed to room temperature. After 4 h, the reaction mixture was quenched with the addition of saturated aqueous NH<sub>4</sub>Cl and diluted with ethyl acetate. The organic layer was washed with H<sub>2</sub>O, saturated aqueous NaCl, then dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. Flash chromatography (SiO<sub>2</sub>, 2 × 8 cm, 15% EtOAc/hexanes elution) provided compound **22** (45 mg, 100%)

as a off white solid.  $^1\text{H}$  NMR (acetone- $d_6$ , 600 MHz)  $\delta$  7.95 (br, 1H), 7.80 (d,  $J = 8.4$  Hz, 1H), 7.65 (d,  $J = 7.2$  Hz, 3H), 7.41 (t,  $J = 7.8$  Hz, 2H), 7.32 (m, 2H), 5.33 (s, 2H), 4.23 (dd,  $J = 2.4, 11.4$  Hz, 1H), 4.18 (m, 1H), 4.11 (m, 1H), 3.96 (dd,  $J = 3, 10.8$  Hz, 1H), 3.83 (s, 3H), 3.71 (dd,  $J = 10.8, 8.4$  Hz, 1H).  $^{13}\text{C}$  NMR (acetone- $d_6$ , 150 MHz)  $\delta$  157.6, 154.8, 143.9, 138.4, 134.8, 132.6, 130.1, 129.9, 129.6, 129.3, 124.2, 121.6, 119.2, 117.2, 100.4, 72.5, 54.1, 53.9, 48.4, 43.2. IR (film)  $\nu_{\text{max}}$  2951, 2357, 1614, 1361, 1133  $\text{cm}^{-1}$ . ESI-TOF HRMS  $m/z$  460.0317 ( $\text{M}+\text{H}^+$ ,  $\text{C}_{22}\text{H}_{19}\text{BrClNO}_3$  requires 460.0310).

**Methyl 6-Amino-1-(chloromethyl)-5-hydroxy-1H-benzo[e]indole-3(2H)-carboxylate (23).** See compound **15** for procedure. Compound **23** was isolated as a tan solid (33% over 3 steps).  $^1\text{H}$  NMR (THF- $d_8$ , 600 MHz)  $\delta$  7.39 (br, 1H), 7.07 (t,  $J = 7.8$  Hz, 1H), 6.75 (d,  $J = 7.8$  Hz, 1H), 6.30 (d,  $J = 7.8$  Hz, 1H), 4.19 (d,  $J = 11.4$  Hz, 1H), 4.03 (t,  $J = 9$  Hz, 1H), 3.90 (dd,  $J = 2.4, 10.8$  Hz, 1H), 3.83 (m, 1H), 3.76 (s, 3H), 3.39 (t,  $J = 10.2$  Hz, 1H).  $^{13}\text{C}$  NMR (THF- $d_8$ , 150 MHz)  $\delta$  158.6, 154.1, 148.9, 142.0, 134.3, 129.3, 114.4, 112.0, 110.0, 107.3, 98.2, 53.4, 52.7, 47.2, 43.6. IR (film)  $\nu_{\text{max}}$  3197, 1672, 1584, 1483, 1411, 1336, 1150, 748  $\text{cm}^{-1}$ . ESI-TOF HRMS  $m/z$  307.0845 ( $\text{M}+\text{H}^+$ ,  $\text{C}_{15}\text{H}_{15}\text{ClN}_2\text{O}_3$  requires 307.0844).

**Methyl 5-Amino-1,2,9,9a-tetrahydro-1H-benzo[e]cyclopropa[c]indol-4-one-2-carboxylate (24).** See compound **21** for procedure. Compound **24** was isolated as a bright yellow solid (70%).  $^1\text{H}$  NMR (acetone- $d_6$ , 600 MHz)  $\delta$  7.22 (br, 1H), 7.13 (t,  $J = 7.8$  Hz, 1H), 6.60 (d,  $J = 8.4$  Hz, 1H), 6.13 (d,  $J = 6.6$  Hz, 1H), 4.05–3.99 (m, 2H), 3.80 (s, 3H), 2.89–2.86 (m, 1H), 1.46 (d,  $J = 6$  Hz, 2H).  $^{13}\text{C}$  NMR (acetone- $d_6$ , 150 MHz)  $\delta$  190.2, 159.7, 153.8, 152.0, 143.5, 133.3, 115.5, 114.4, 109.3, 108.8, 53.4, 53.2, 34.1, 31.0, 24.9. IR (film)  $\nu_{\text{max}}$  3441, 1730, 1600, 1537, 1440, 1291,  $\text{cm}^{-1}$ . ESI-TOF HRMS  $m/z$  271.1077 ( $\text{M}+\text{H}^+$ ,  $\text{C}_{15}\text{H}_{14}\text{N}_2\text{O}_3$  requires 271.1077).

Solvolysis of **24**:

pH 1 buffer: 10 M citric acid: 0.2 M  $\text{Na}_2\text{HPO}_4$ :  $\text{H}_2\text{O}$  (4:1:20)

$$t_{1/2} = 11.6 \text{ h}, k = 1.8 \times 10^{-5} \text{ sec}^{-1}$$

pH 2 buffer: 1.0 M citric acid: 0.2 M  $\text{Na}_2\text{HPO}_4$ :  $\text{H}_2\text{O}$  (4:1:20)

$$t_{1/2} = 57.6 \text{ h}, k = 3.3 \times 10^{-6} \text{ sec}^{-1}$$

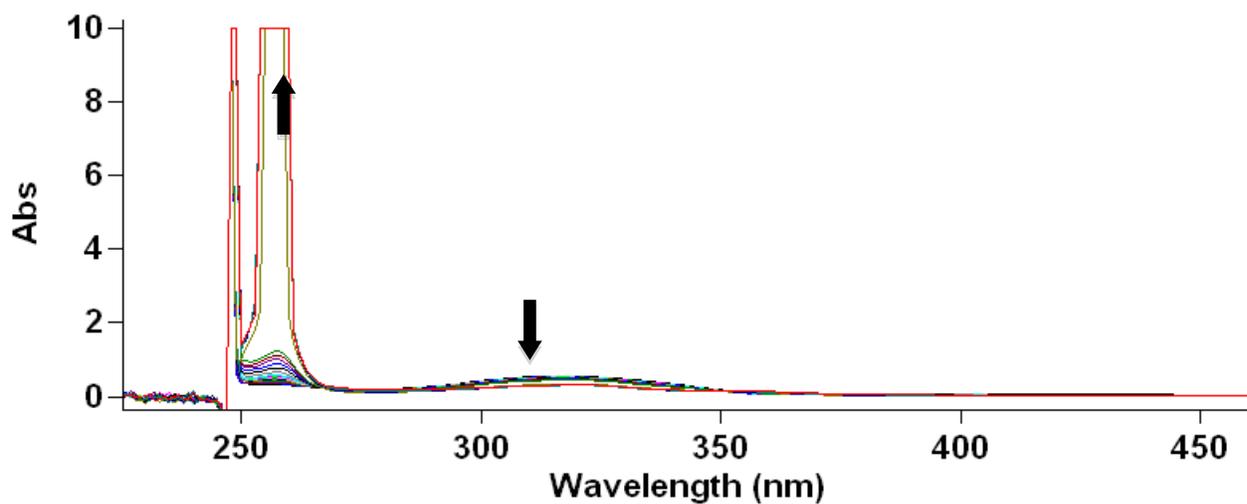


Figure S3. UV-visible spectra of **24** in 50% CH<sub>3</sub>OH-aqueous buffer (pH 1) recorded at various time intervals.

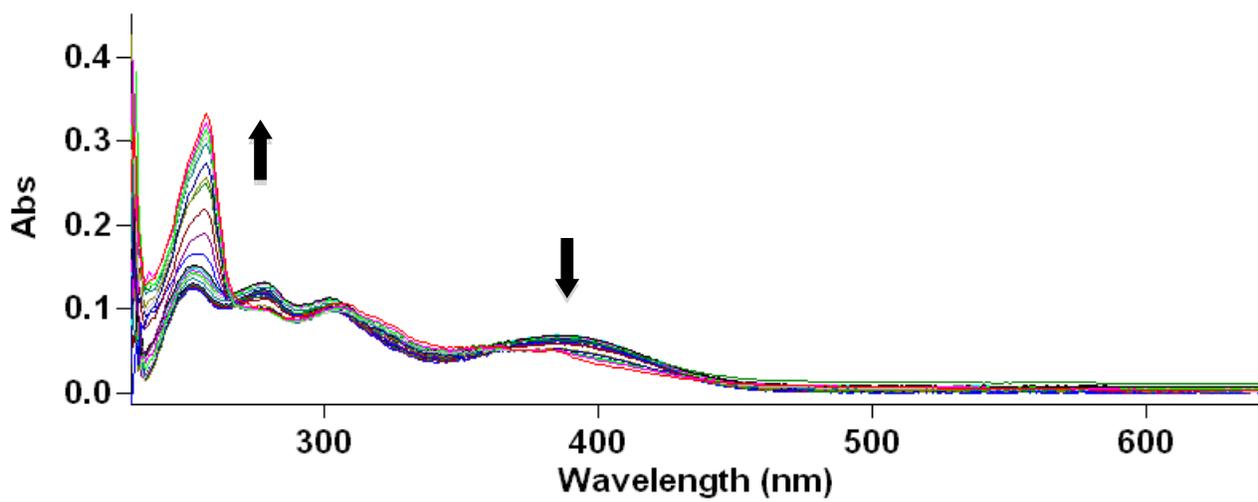


Figure S4. UV-visible spectra of **24** in 50% CH<sub>3</sub>OH-aqueous buffer (pH 2) recorded at various time intervals.