Marine sulfate-reducing bacteria cause serious corrosion of iron under electroconductive biogenic mineral crust

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Table S1. Compilation of corrosion rates recorded for (anoxic) natural and engineered environments, and for laboratory cultures of sulfate-reducing bacteria.

Location	Type of environment	Corrosion rate (mm yr ⁻¹) a	Method ^b	Reference
Bohai Bay, China	Marine sediment	0.03 - 0.09	Weight loss	Li (2009)
Draugen Oil Field, Norwegian North Sea	Produced water re-injection system; untreated produced water	0.12	Electrochemical ^c	Vik et al. (2007)
Tonnenlegerbucht, German North Sea	Marine anoxic sediment	0.12 - 0.28	Weight loss	This study
South Korea	Gas transmission pipelines; disbonded coating	0.33 - 0.47	Pit measurement	Li et al. (2000)
Gullfaks Oil Field, Norwegian North Sea	Water injection system; biocide treatment	0.09 - 0.75	Weight loss	Boedtker et al. (2008)
Culture	Type of energy source, medium ^d	Corrosion rate (mm yr ^{-1) e}	Method	Reference
Desulfovibrio sp.	Lactate-based, freshwater	0.007	Weight loss	Hardy and Brown (1984)
Desulfovibrio vulgaris, Woolwich	Lactate-based, freshwater	0.024	Weight loss	Gaylarde (1992)
Desulfovibrio desulfuricans, New Jersey	Lactate-based, marine	0.094	Weight loss	Beech et al. (1994)
Desulfovibrio desulfuricans	Lactate-based, brackish	0.285	Weight loss	Bell and Lim (1981)
SRB bioreactor	Lactate-based, brackish	0.408	Weight loss	Hubert et al. (2005)
Desulfovibrio vulgaris, Hildenborough	H ₂ , freshwater ^f	0.500	Electrochemical ^g	Pankhania et al. (1986)
Sterile artificial seawater	Fe ⁰ lithotrophy, marine	0.010	Weight loss	This study
Deculfonile informa	Fe ⁰ lithotrophy, marine	0.012	Weight loss	This study
Desulfopila inferna	re innonopriy, marine		•	,
'Desulfopila corrodens', strain IS4	Fe ⁰ lithotrophy, marine	0.311	Weight loss	This study

a. Corrosion rate in natural and engineered environments. The recorded range is given. Corrosion must not necessarily be influenced by SRB in these systems.

b. Method of corrosion rate determination. Weight loss is considered the most accurate technique.

 $[\]textbf{c.}\, \textbf{A}\, \text{linear polarisation resistance}\,\, (\text{LPR})\,\, \text{probe was used}.$

d. Energy source for SRB metabolism is indicated. All cultures contained iron or mild steel. 'Fe⁰ lithotrophy' indicates that metallic iron was the only available source of reducing equivalents. Salt content of media (freshwater, brackish, marine) is also designated.

e. Corrosion rate in cultures of SRB. The maximal reported value is given. The large range of corrosion rates for SRB with external energy source is attributed to the sometimes protective properties of formed FeS films (see text).

f. H₂ from previous polarisation experiment was available in these cultures.

g. Calculated from slope of cathodic polarisation curve by overvoltage-intercept method.