

# Organometallic Palladium Complexes with a Water-Soluble Iminophosphorane Ligand as Potential Anticancer Agents

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**Supporting Information**

1.  $^1\text{H}$  and  $^{13}\text{C}\{^1\text{H}\}$  NMR spectra for compounds 2 and 3.

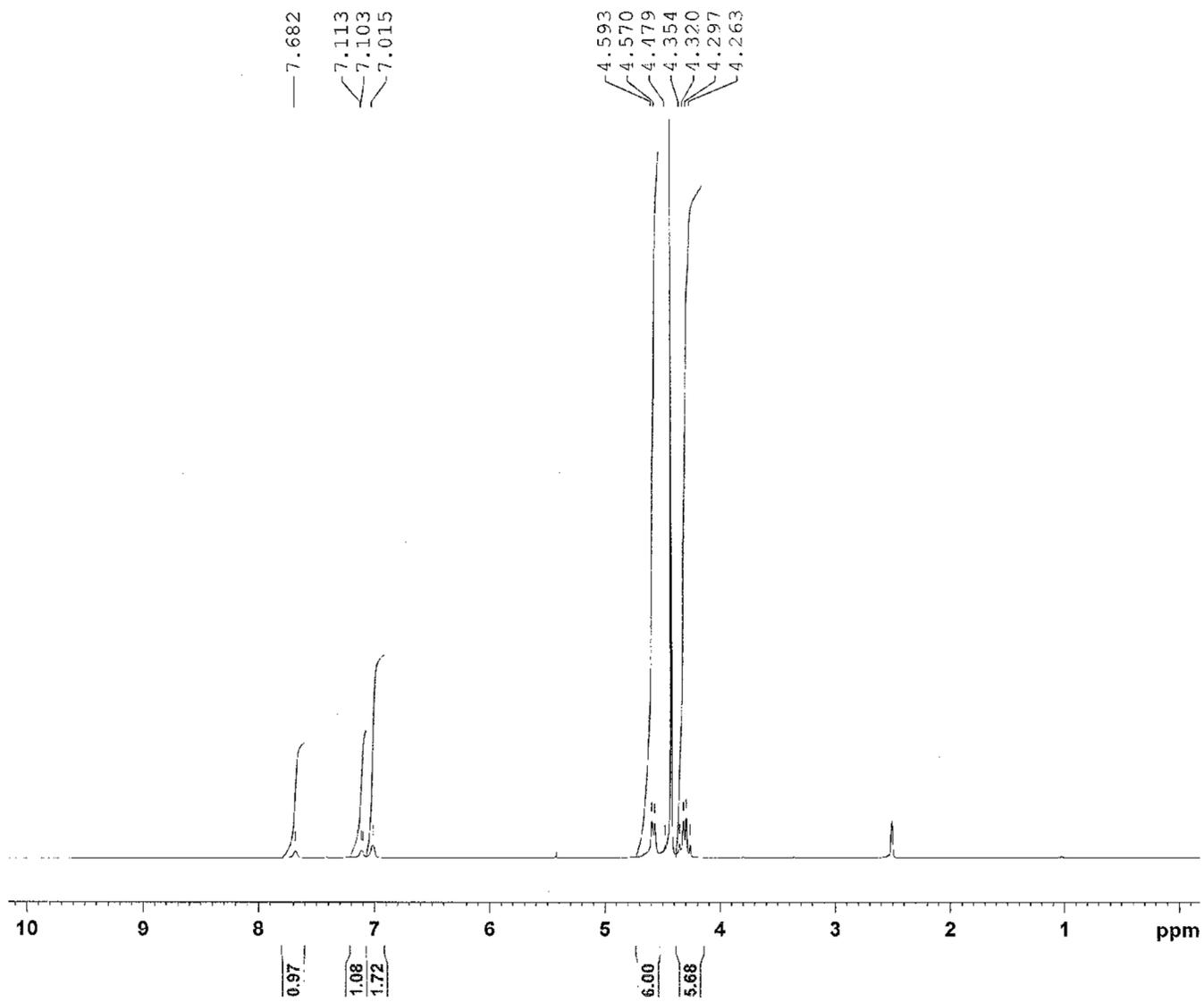
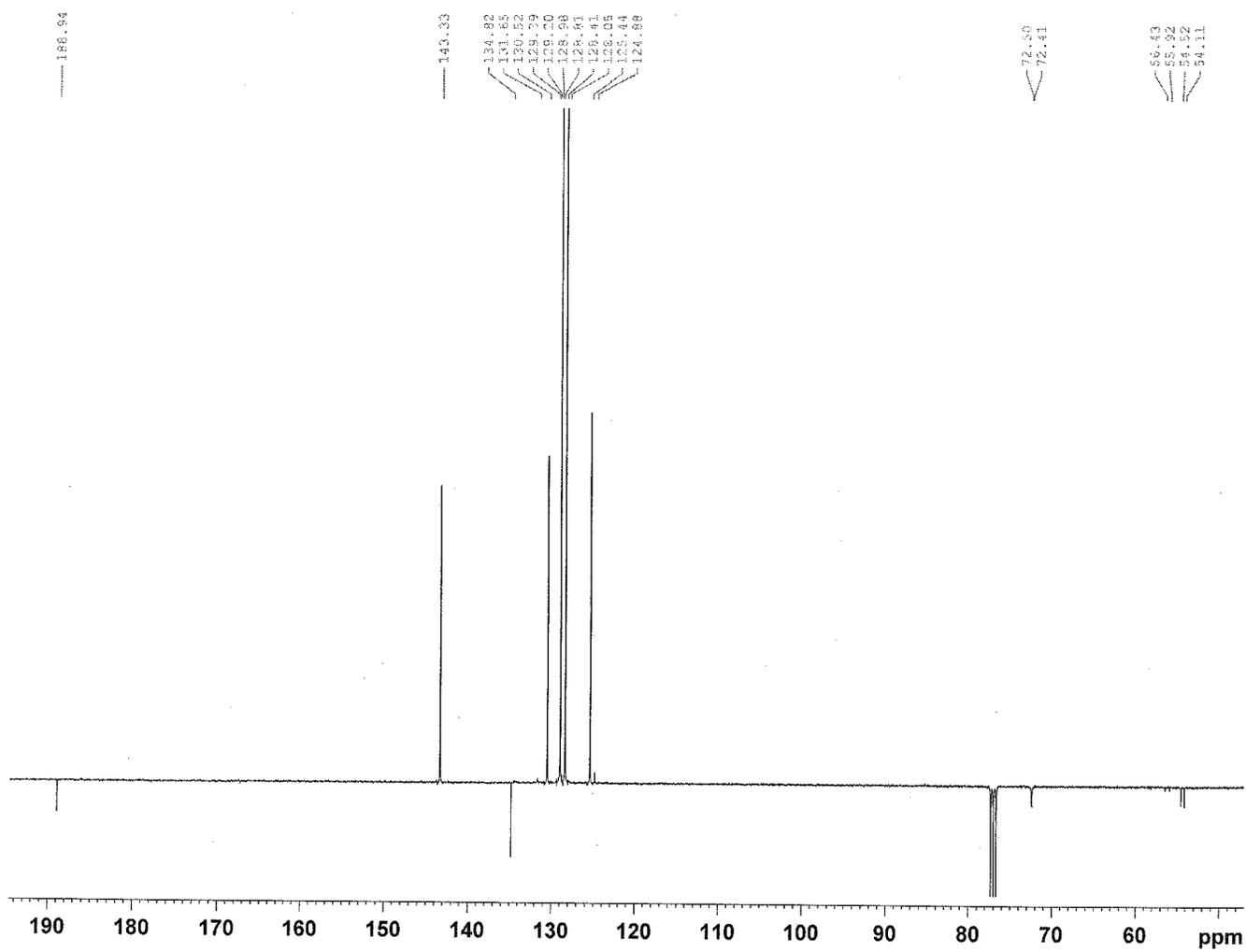


Figure S1.  $^1\text{H}$  NMR spectra for compound 2.



**Figure S2.**  $^{13}\text{C}\{^1\text{H}\}$  NMR spectra for compound 2.

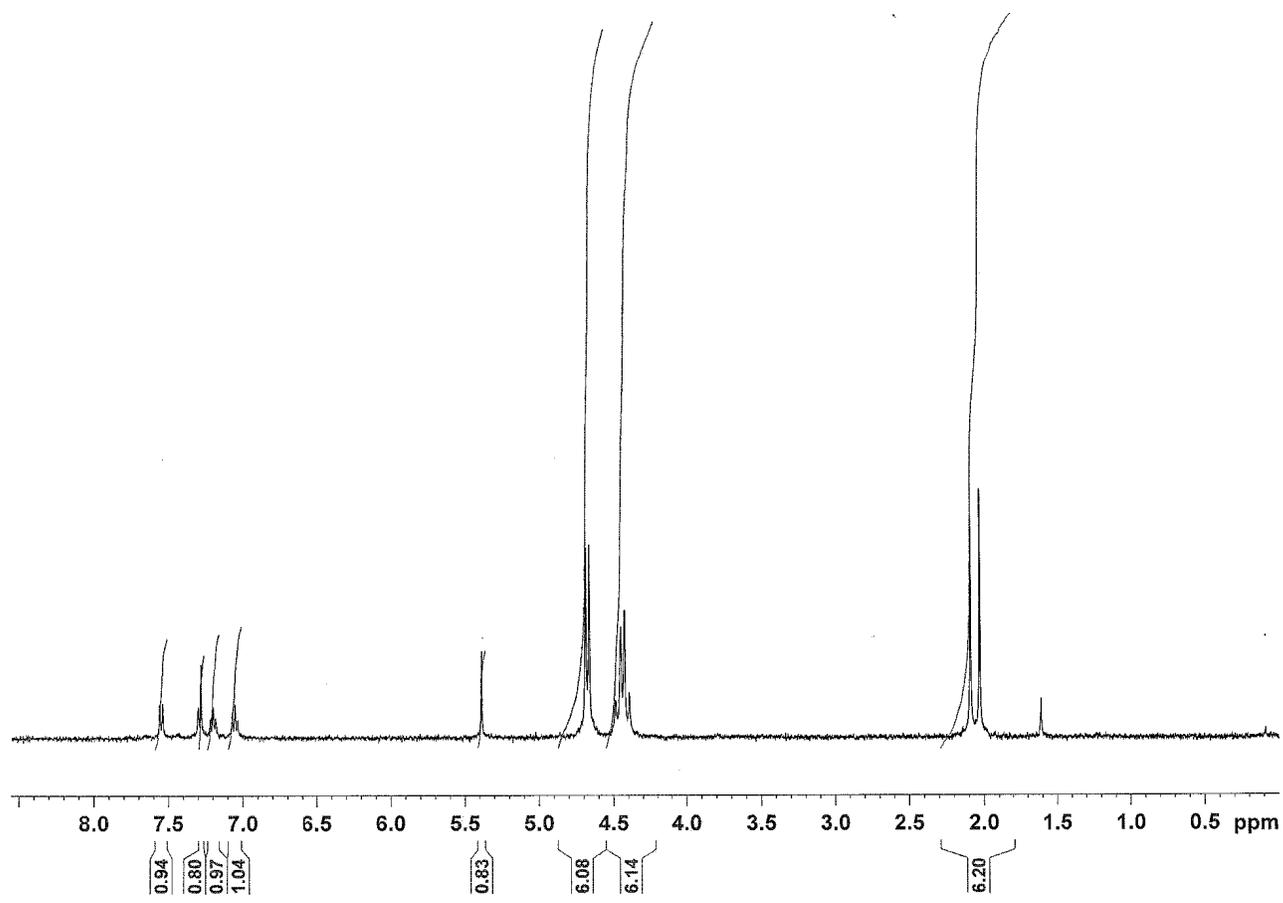


Figure S3. <sup>1</sup>H NMR spectra for compound 3.

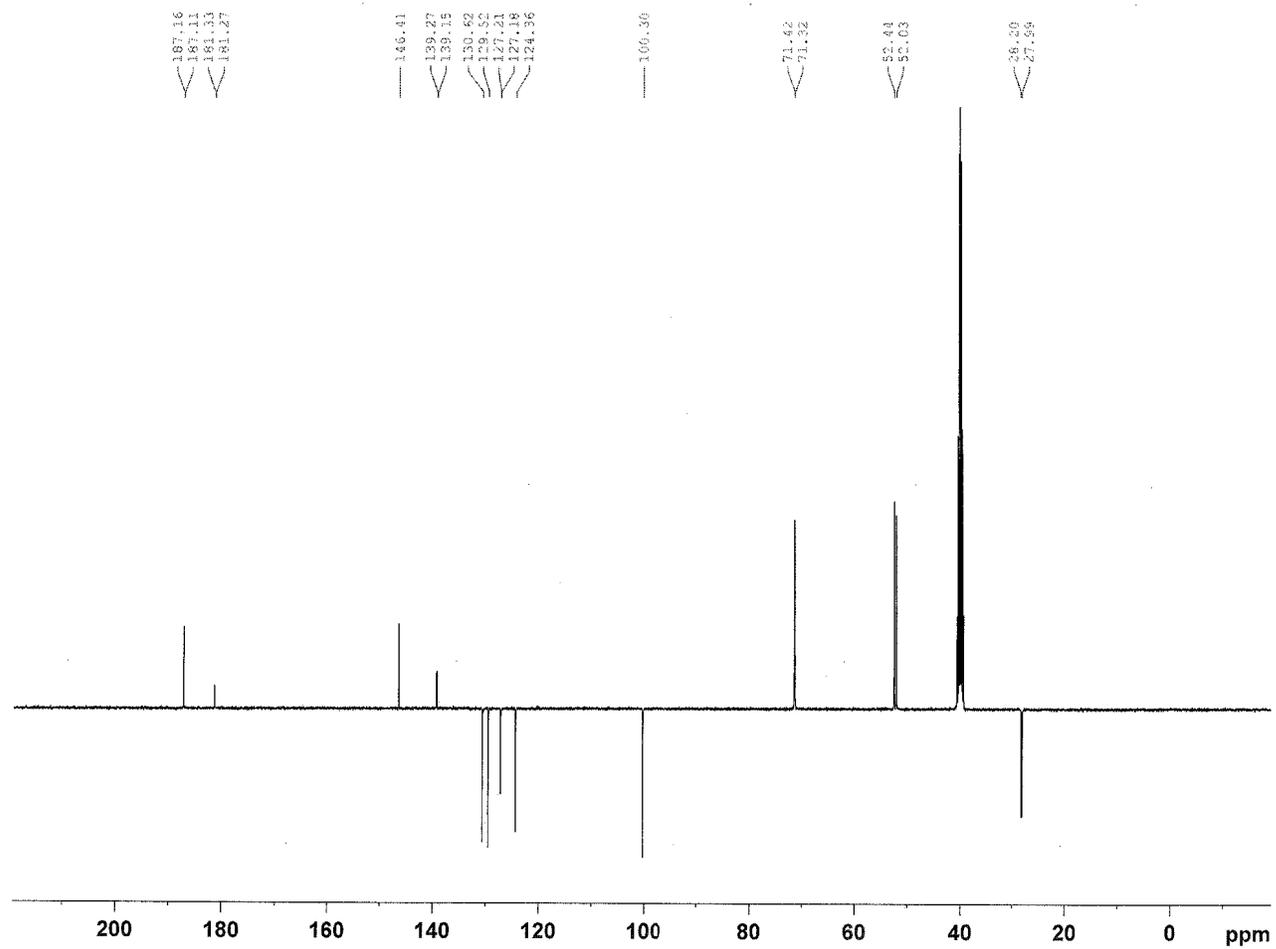


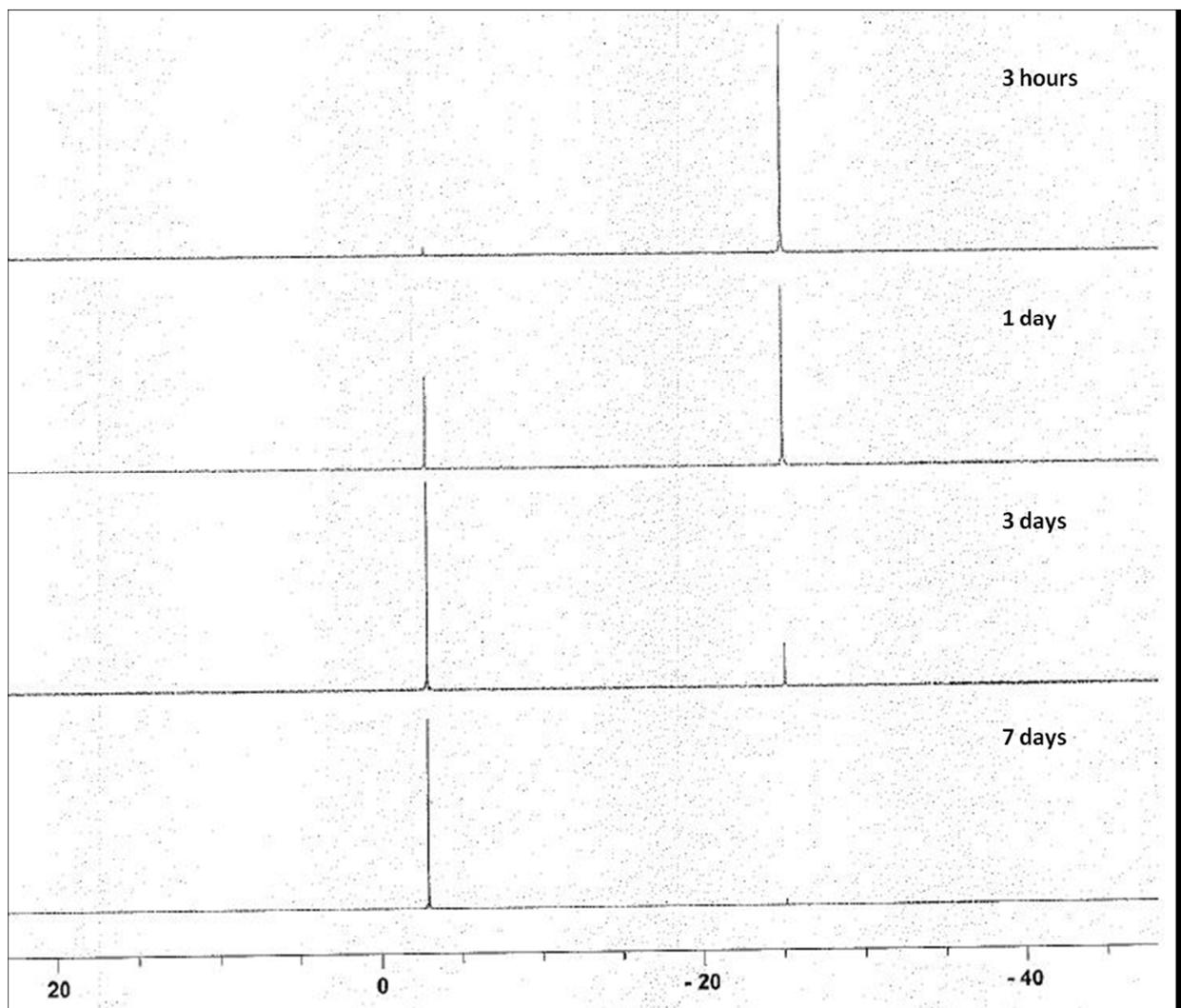
Figure S4.  $^{13}\text{C}\{^1\text{H}\}$  NMR spectra for compound 3.

**2. Stability of ligand 1 and compounds 2-8 in d<sup>6</sup>-DMSO solution and mixtures d<sup>6</sup>-DMSO:D<sub>2</sub>O (50:50) overtime assessed by <sup>31</sup>P{<sup>1</sup>H} NMR spectroscopy.**

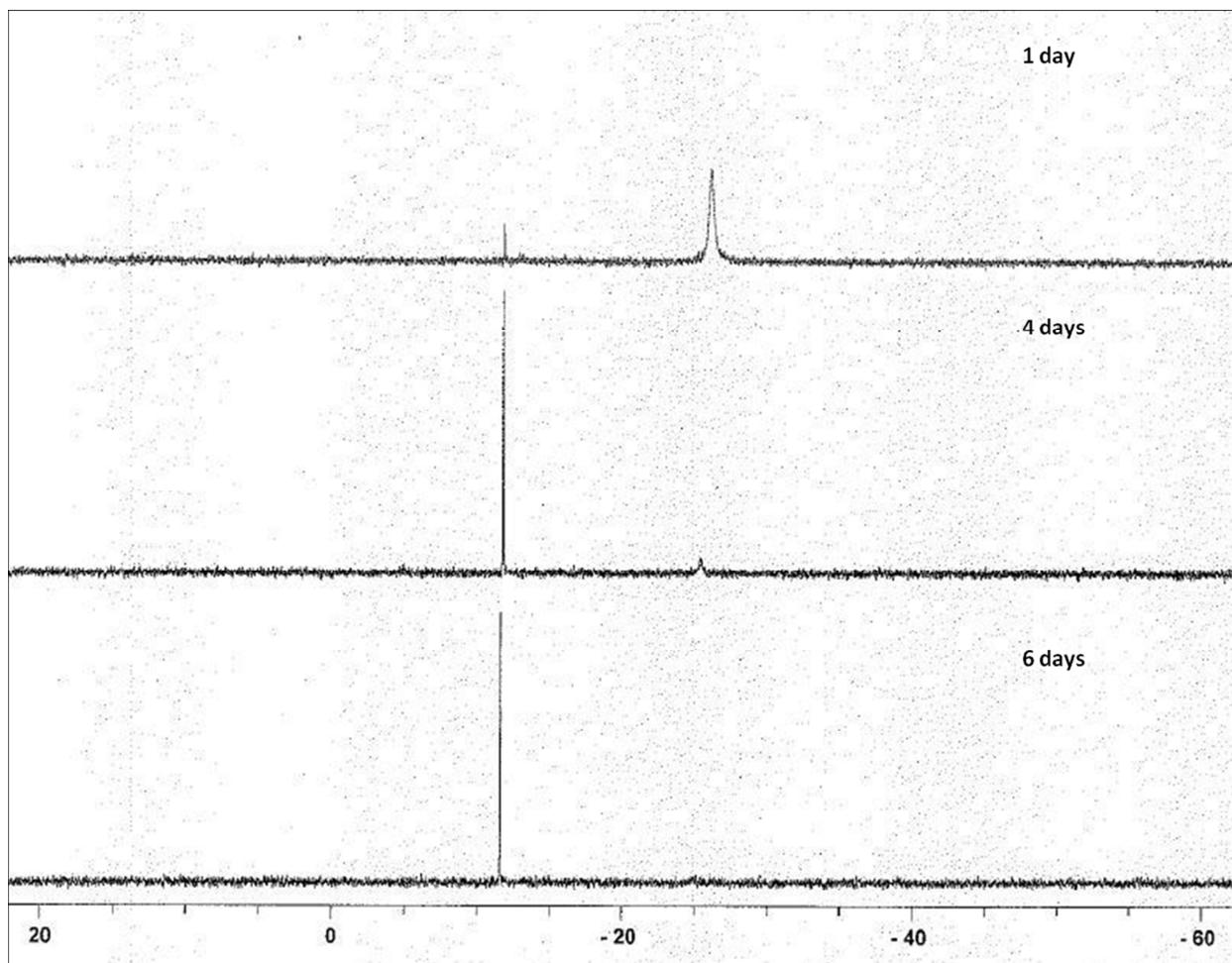
**Table S1.** Stability of compounds 2-8 in d<sup>6</sup>-DMSO solution and mixtures d<sup>6</sup>-DMSO:D<sub>2</sub>O (50:50) overtime assessed by <sup>31</sup>P{<sup>1</sup>H} NMR spectroscopy.

	d <sup>6</sup> -DMSO						d <sup>6</sup> -DMSO/D <sub>2</sub> O				
	1 day	4 days	1 week	2 weeks	Half life (50%)		1	4	1 week	2 weeks	Half life (50%)
<b>1</b>											
<b>2</b>	>99%	>99%	>99%		months	<b>2</b>	88%	75%		50%	2 weeks
<b>3</b>	>99%		>99%	>99%	months	<b>3</b>	>99%	>99%	>99%		months
<b>4</b>			>99%	>99%	months	<b>4</b>	>99%		>99%		months
<b>5</b>		43%	0%		~4 days	<b>5</b>	83%	45%	0%		~4 days
<b>6</b>		94%		93%	months	<b>6</b>	92%	86%			~1 month
<b>7</b>	>95%	90%			~1 month	<b>7</b>	85%	65%	50%		~1 week
<b>8</b>	64%	23%	0%		2 days	<b>8</b>	90%	75%	65%		~2 weeks

Half life of ligand **1** (assessed by <sup>31</sup>P{<sup>1</sup>H} NMR spectroscopy) was: months in d<sup>6</sup>-DMSO; 2 weeks in d<sup>6</sup>-DMSO:D<sub>2</sub>O (50:50) and 2 days in D<sub>2</sub>O.



**Figure S5.** Selected  $^{31}\text{P}\{^1\text{H}\}$  NMR spectra showing the decomposition of ligand **1** in  $\text{D}_2\text{O}$  ( $\delta$ -25.1 (s)) overtime. The peak at -3.4 ppm corresponds to TPA=O in  $\text{D}_2\text{O}$ .



**Figure S6.** Selected  $^{31}\text{P}\{^1\text{H}\}$  NMR spectra showing the decomposition of compound **5** in  $d^6$ -DMSO: $\text{D}_2\text{O}$  (50:50) ( $\delta$  -27.0 (s)) overtime. The peak at -12.0 ppm corresponds to TPA=O in  $d^6$ -DMSO: $\text{D}_2\text{O}$  (50:50).

### 3. Crystal data and structure refinement for complex 3.

**Table S2.** Crystal data and structure refinement for complex 3.

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formula	C <sub>18</sub> H <sub>24</sub> N <sub>4</sub> O <sub>3</sub> PPd
Fw	481.78
T [K]	293(2) K
$\lambda$ (MoK $\alpha$ )[Å]	0.71073
crystal system	monoclinic,
Space group	P2(1)/c
$a$ [Å]	9.5820(19)
$b$ [Å]	20.496(4)
$c$ [Å]	10.647(2)
$\alpha$ [°]	90 deg
$\beta$ [°]	106.92(3)
$\chi$ [°]	90 deg.
$V$ [Å] <sup>3</sup>	2000.5(7)
$Z$	4
$D_{\text{calcd}}$ (g cm <sup>-3</sup> )	1.600
$\mu$ (mm <sup>-1</sup> )	1.033
$R(F_o)^a$	0.0376 [for 3411 data with $F_o^2 > 2\sigma(F_o^2)$ ]
$R_w(F_o^2)^b$	0.0968

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$$^a R(F_o) (= R1) = \Sigma ||F_o| - |F_c|| / \Sigma |F_o|. \quad ^b R_w(F_o^2) (= wR2) = \{ \Sigma [ w(F_o^2 - F_c^2)^2 ] / \Sigma [ w(F_o^2)^2 ] \}^{1/2}$$

