

Supporting Information

***In situ*-generated metal oxide catalyst during CO oxidation reaction transformed from redox-active metal-organic framework supported palladium nanoparticles**

Jin Yeong Kim,^a Mingshi Jin,^b Kyung Joo Lee,^a Jae Yeong Cheon,^c Sang Hoon Joo,^{c*} Ji Man Kim,^{b*} and Hoi Ri Moon^{a*}

^a Interdisciplinary School of Green Energy, Ulsan National Institute of Science and Technology (UNIST), Ulsan 689-798, Republic of Korea. E-mail: hoirimoon@unist.ac.kr

Fax: +82-52-217-2019; Tel: +82-52-217-2928

^b Department of Chemistry, BK21 School of Chemical Materials Science and Department of Energy Science, Sungkyunkwan University, Suwon 440-746, Republic of Korea. E-mail: jimankim@skku.edu

Fax: +82-31-290-7075; Tel: +82-31-290-7074

^c School of Nano-Bioscience and Chemical Engineering, Ulsan National Institute of Science and Technology (UNIST), Ulsan 689-798, Republic of Korea. E-mail: shjoo@unist.ac.kr

Fax: +82-52-217-2019; Tel: +82-52-217-2522

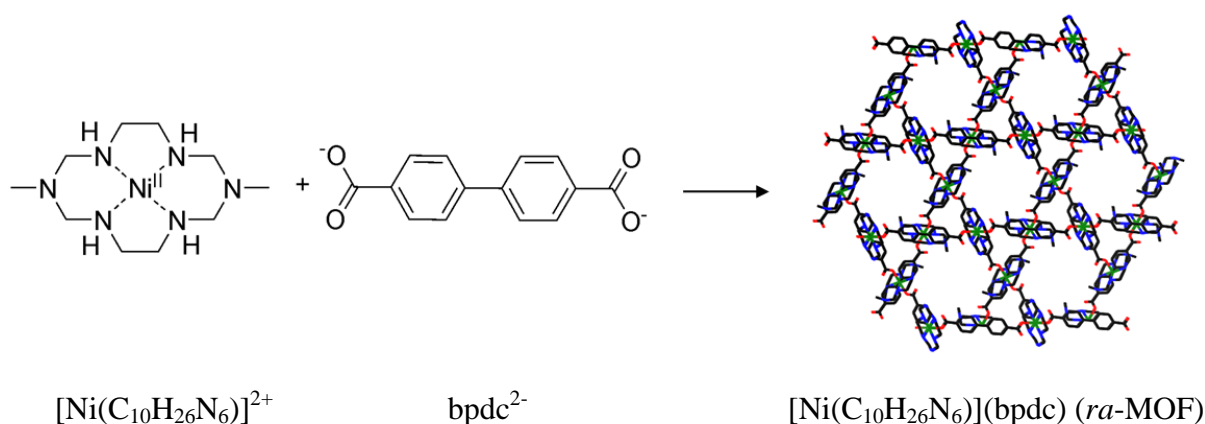


Figure S1. Synthetic scheme of *ra*-MOF by self-assembly of [Ni(C₁₀H₂₆N₆)]²⁺ and bpd²⁻

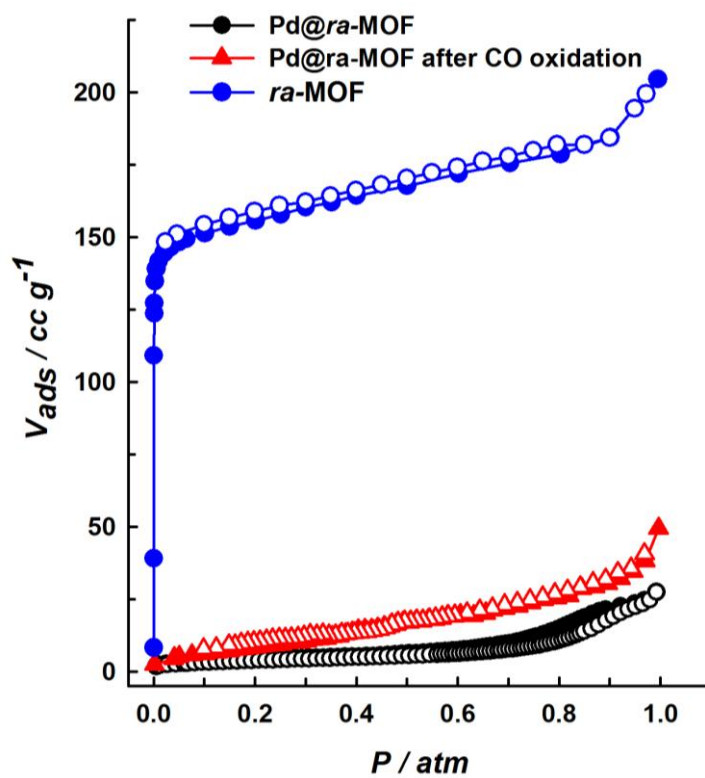


Figure S2. Comparison of N₂ sorption data for *ra*-MOF and Pd@*ra*-MOF before and after CO oxidation reaction

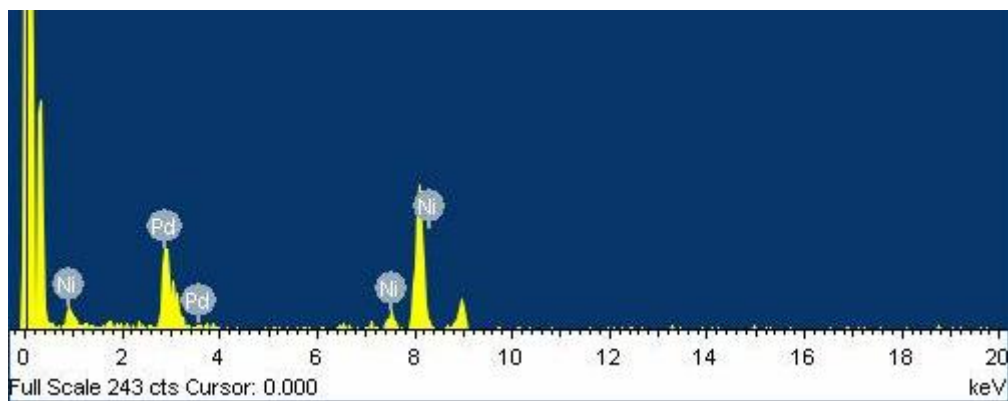


Figure S3. EDS result of Pd@*ra*-MOF

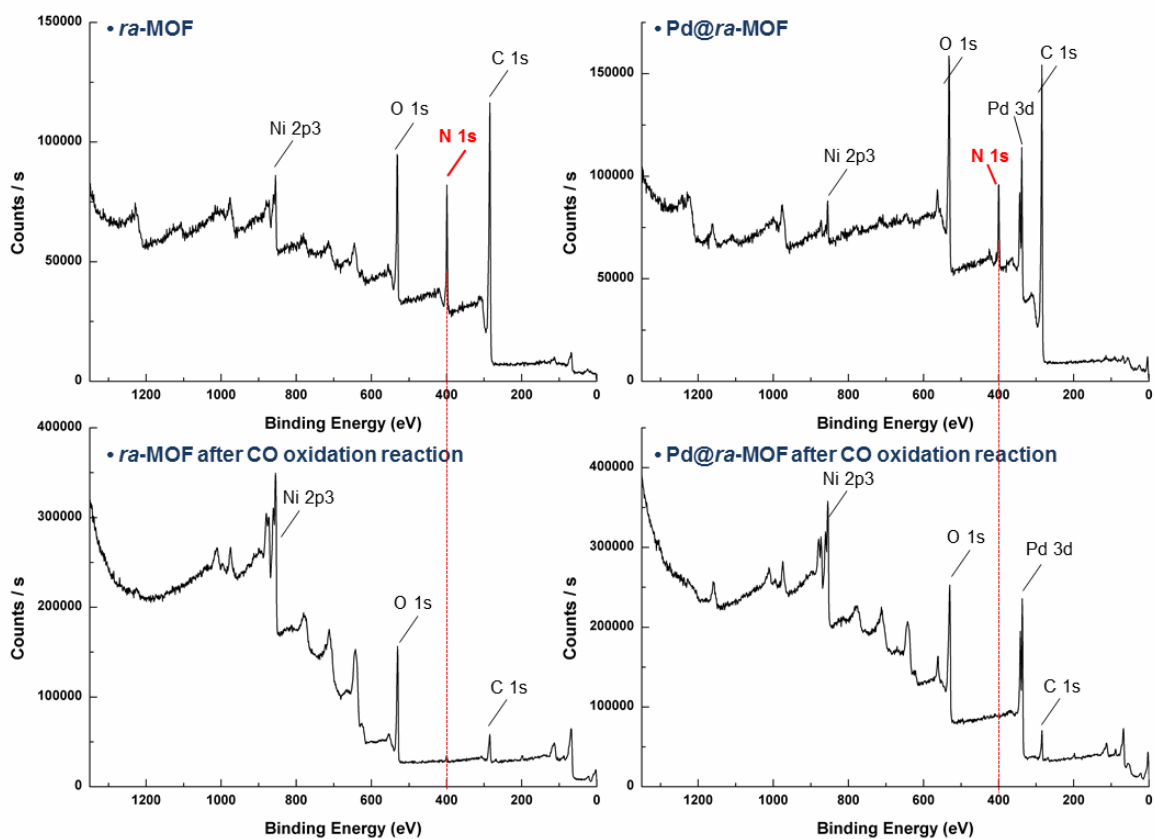


Figure S4. XPS wide scan spectra for *ra*-MOF and Pd@*ra*-MOF before and after CO oxidation reaction, respectively.

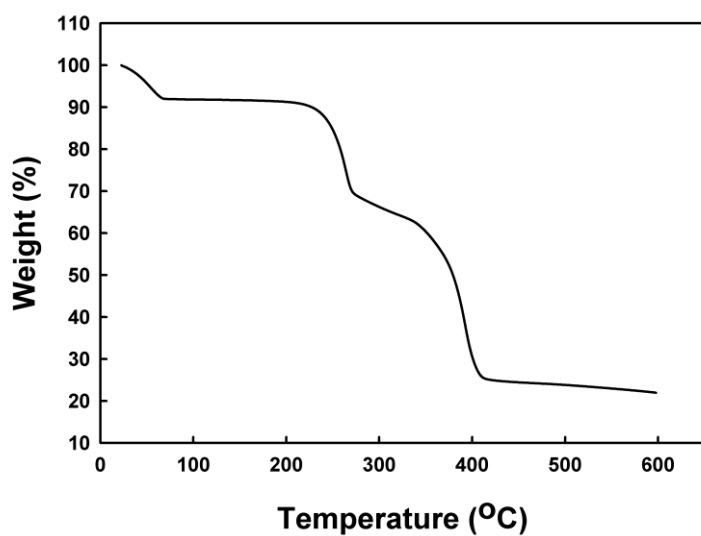


Figure S5. TGA trace of *ra*-MOF

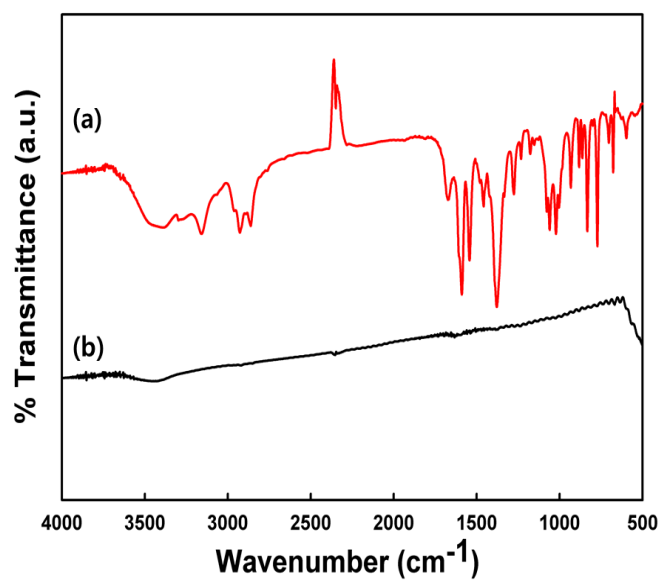


Figure S6. Comparison of IR spectra for *ra*-MOF (a) before and (b) after CO oxidation reaction upto 500 °C.

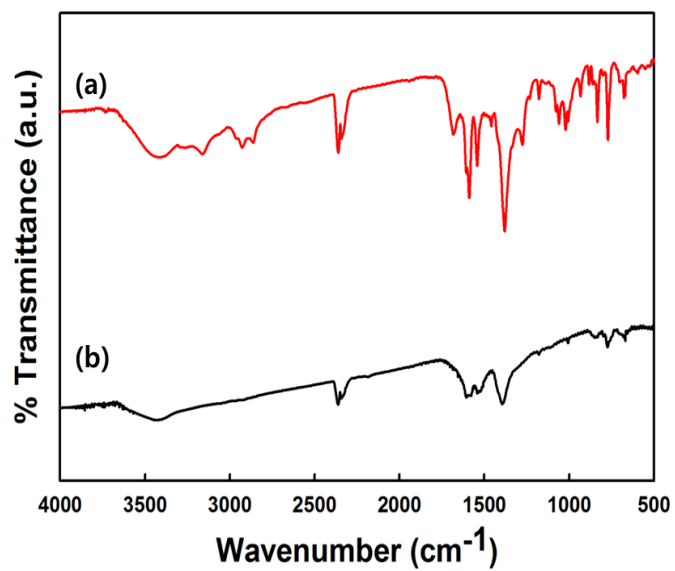


Figure S7. Comparison of IR spectra for Pd@*ra*-MOF (a) before and (b) after CO oxidation reaction upto 300 °C.