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### Towards a Structural Basis for the Relationship Between Blood Group and the Severity of El Tor Cholera\*\*

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#### Synthesis of oligosaccharides 7 and 8

Recently, we developed an efficient synthesis of protected Lewis-y tetrasaccharide 6,<sup>[1]</sup> which proved suitable for further elaboration into the target tetrasaccharide 7 and pentasaccharide 8. Thus, hydrolysis of the acetonide group in 6 and selective orthoestermediated acetylation of diol S1 provided acceptor alcohol S2. Galactosylation of S2 using thioglycoside donor S4 gave pentasaccharide S3 in 70% yield. A five-step deprotection sequence allowed conversion of compounds S2 and S3 into tetrasaccharide 7 and pentsaccharide 8, which were isolated in good yield as the methyl carbamates. While the methyl carbamate modifications were unexpected, both NMR spectroscopy and mass spectrometry of the final compounds were consistent with these structures.



**Scheme S1.** Reagents and conditions: a) 80% AcOH (aq), 80 °C, 85%; b) (i)  $(EtO)_3CH_3$ , p-toluenesulfonic acid, DMF, (ii) 80% AcOH (aq), 72% over two steps; c) **6**, NIS, TfOH, MS 4Å, CH<sub>2</sub>Cl<sub>2</sub>, -40 °C, 70%; d) (i) H<sub>2</sub>NCH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>, n-BuOH, 80 °C, (ii) Ac<sub>2</sub>O, C<sub>5</sub>H<sub>5</sub>N, (iii) NaOMe, MeOH, (iv) TBAF, THF, (v) H<sub>2</sub>, 20% Pd(OH)<sub>2</sub>-C, MeOH; the five step deprotection sequence provided tetrasaccharide **7** in 53%, and pentasaccharide **8** in 50% yield from **S2** and **S3**, respectively.

#### **General methods**

All solvents were dried using a PureSolv purification system. All solvents used for flash chromatography were GPR grade, except hexane and ethyl acetate, when HPLC grade was used. All concentrations were performed *in vacuo*, unless otherwise stated. All reactions were performed in oven dried glassware under a N<sub>2</sub>(g) atmosphere, unless otherwise stated. Analytical TLC was performed on silica gel 60-F<sup>254</sup> (Merck) with detection by fluorescence and/or charring following immersion in a 5% H<sub>2</sub>SO<sub>4</sub>/methanol solution, unless otherwise stated. Optical rotations were measured at the sodium D-line with an Optical Activity AA-1000 polarimeter. [ $\alpha$ ]<sub>D</sub> values are given in units of 10<sup>-1</sup> deg cm<sup>2</sup> g<sup>-1</sup>. Infrared spectra were recorded on a Perkins-Elmer Spectrum One FT-IR spectrometer. Electrospray (ES+) ionisation mass spectra were obtained on a Bruker HCT-Ultra mass spectrometer, and high resolution ES+ were performed on a Bruker Daltonics MicroTOF mass spectrometer.



Tetrasaccharide 6<sup>[1]</sup> (2.2 g, 1.45 mmol) was dissolved in 80% AcOH (150 mL) and stirred at 80 °C for 2 h. The solution was evaporated to give an oil which was purified over SiO<sub>2</sub> using hexane-EtOAc (2:1) as eluent to furnish diol **S1** as a yellow oil (1.85 g, 85%);  $[\alpha]_D^{25}$  -15 (c 1.5, CHCl<sub>3</sub>); IR: 3429, 2931, 1715, 1454, 1386, 1099, 1052, 737, 697 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.74-7.65 (m, 4 H, Ar-H), 7.41-7.17 (m, 45 H, Ar-H), 7.11 (t, J = 7.5 Hz, 1 H, Ar-H), 7.02 (t, J = 7.5 Hz, 2 H, Ar-H), 6.93 (d, J = 7.5 Hz, 2 H, Ar-H), 5.04-4.99 (m, 3 H, H-1<sub>A</sub>, H-1<sub>D</sub>, PhCH), 4.96 (d, J = 11.6 Hz, 1 H, PhCH), 4.91 (d, J = 11.6 Hz, 1 H, PhCH), 4.83-4.72 (m, 8 H, H-1<sub>B</sub>, H-1<sub>C</sub>, H-5<sub>C</sub>, 4 × PhCH), 4.70-4.61 (m, 4 H, H-3<sub>A</sub>, 3 × PhCH), 4.58 (d, J = 11.6 Hz, 1 H, PhCH), 4.49 (d, J = 11.4 Hz, 1 H, PhCH), 4.47-4.41 (m, 4 H, H-2<sub>A</sub>, 3 × PhCH), 4.31 (t, J = 9.4 Hz, 1 H, H-4<sub>A</sub>), 4.23 (d, J = 12.2 Hz, 1 H, PhCH), 4.13 (dd, J = 10.0, 3.3 Hz, 1H, H-2<sub>D</sub>), 4.08-4.06 (br d, J = 11.2 Hz, 1 H, H-6<sub>a</sub>A), 4.01 (dd, J = 10.2, 2.6 Hz, H-3<sub>C</sub>), 3.89 (br s, 1 H, H-4<sub>B</sub>), 3.85-3.69 (m, 7 H, H-6<sub>bA</sub>, H-6<sub>aB</sub>, H-6<sub>bB</sub>, H-2<sub>C</sub>, H-4<sub>C</sub>, H-3<sub>D</sub>, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH), 3.51-3.42 (m, 6 H, H-5A, OCH<sub>2</sub>, H-2<sub>B</sub>, H-3<sub>B</sub>, H-5<sub>B</sub>, H-4<sub>D</sub>, H-2<sub>C</sub>,  $OC_{H_2}CH_2CH_2NH$ ) 3.31 (q, J = 6.4 Hz, 1 H, H-5<sub>D</sub>), 3.07-3.05 (m, 2 H,  $OCH_2CH_2CH_2NH$ ), 2.14 (s, 1H, OH), 1.70-1.60 (m, 3 H, OH, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH), 1.09 (d, J = 6.4 Hz, 3 H, 6<sub>C</sub>), 1.05 (s, 9 H, SiC(CH<sub>3</sub>)<sub>3</sub>), 0.81 (d, J = 6.4 Hz, 3 H,  $6_D$ ); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  168.6, 168.1, 156.4, 139.0-123.0, 100.8 (2 C), 98.0, 97.5, 81.3, 79.6, 79.1, 78.2, 77.6, 77.3, 75.5, 75.1, 75.0, 74.8, 74.5, 73.7, 73.4, 73.3, 72.9, 72.6, 72.4, 72.2 (2 C), 68.0, 67.9, 67.4, 66.5, 66.4, 66.2, 61.7, 56.5, 38.5, 29.6, 26.9, 19.5, 17.1, 16.9; ESI-MS: C<sub>108</sub>H<sub>118</sub>N<sub>2</sub>NaO<sub>22</sub>Si requires m/z 1845.7842; found: m/z 1845.7855 [M+Na]<sup>+</sup>.

# 3-(Benzyloxycarbonylamino)propyl 4-O-[4-O-acetyl-6-O-benzyl-2-O-(2,3,4-tri-O-benzyl- $\alpha$ -L-fucopyranosyl)- $\beta$ -D-galactopyranosyl]-6-O-*t*-butyldiphenylsilyl-2-deoxy-2-phthalimido-3-O-(2,3,4-tri-O-benzyl- $\alpha$ -L-fucopyranosyl)- $\beta$ -D-glucopyranoside (S2)



Triethyl orthoacetate (0.3 mL, 1.75 mmol) and p-toluenesulfonic acid (0.3 g, 1.75 mmol) were added to a solution of diol S1 (1.6 g, 0.877 mmol) in dry DMF (15 mL). The reaction mixture was stirred at room temperature for 2 h. The solvents were removed under reduced pressure and a solution of the intermediate orthoester in 80% aqueous AcOH (50 mL) was stirred at room temperature for 1 h. The reaction mixture was evaporated to dryness and the product was purified over SiO2 using hexane-EtOAc (4:1) as eluant to furnish acetate **S2** as a colourless syrup (1.2 g, 72%);  $[\alpha]_D^{25}$  -35 (c 1.5, CHCl<sub>3</sub>); IR: 3428, 2932, 1716, 1651, 1453, 1387, 1098, 750, 698 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.74-7.66 (m, 4 H, Ar-H), 7.34-7.17 (m, 45 H, Ar-H), 7.12 (t, J = 7.3 Hz, 1 H, Ar-H), 7.04 (t, J = 7.3 Hz, 2 H, Ar-H), 6.94 (m, 2 H, Ar-H), 5.39 (d, J = 3.6 Hz, 1 H, H-4<sub>B</sub>), 5.05-4.99 (m, 3 H, H-1<sub>A</sub>, H-1<sub>D</sub>, PhCH), 4.95 (d, J = 11.6 Hz, 1 H, PhCH), 4.88-4.86 (m, 2 H, H-1<sub>B</sub>, PhCH), 4.81 (d, J = 11.7 Hz, 1 H, PhCH), 4.76-4.65 (m, 7 H, H-1<sub>c</sub>, H-5<sub>c</sub>, H-3<sub>A</sub>, 4 × PhCH), 4.61 (d, *J* = 11.6 Hz, 1 H, PhCH), 4.58 (d, J = 11.6 Hz, 1 H, PhCH), 4.53 (d, J = 11.7 Hz, 1 H, PhCH), 4.48 (d, J = 11.4 Hz, 1 H, PhCH), 4.45 (dd, J = 8.5, 10.5 Hz, 1 H, H-2<sub>A</sub>), 4.40 (brd, J = 11.9 Hz, 2 H, 2 × PhCH), 4.33-4.27 (m, 3 H, H-4<sub>A</sub>, H-4<sub>D</sub>, PhCH), 4.20 (d, *J* = 12.2 Hz, 1 H, PhCH), 4.12-4.09  $(m, 2 H, H-6_{aA}, H-2_D), 3.93 (dd, J = 2.5, 10.2 Hz, 1 H, H-3_C), 3.87-3.77 (m, 4 H, H-6_{bA}, H-2_C)$ H-3<sub>D</sub>, OC<u>*H*</u><sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH), 3.64 (ddd, *J* = 1.6, 3.6, 9.8 Hz, 1 H, H-3<sub>B</sub>), 3.62-3.54 (m, 3 H, H- $6_{aB}$ , H- $6_{bB}$ , H- $4_{C}$ ), 3.47-3.39 (m, 4 H, H- $5_{A}$ , H- $2_{B}$ , H- $5_{D}$ , OC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH), 3.32 (t, J = 8.4 Hz, 1 H, H-5<sub>B</sub>), 3.09-3.05 (m, 2H, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH), 1.73-1.58 (m, 5 H, OH,  $OCH_2CH_2CH_2NH$ ,  $COCH_3$ ), 1.11 (d, J = 6.4 Hz, 3 H, H-6<sub>c</sub>), 1.06 (s, 9 H, SiC(CH<sub>3</sub>)<sub>3</sub>), 0.89 (d, J = 6.4 Hz, 3 H, H-6<sub>D</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  169.8, 168.6, 168.1, 156.4, 138.7-123.7, 100.9, 100.6, 98.0, 97.5, 80.8, 79.9, 79.1, 77.7, 77.5, 77.2, 75.5, 74.8, 74.7, 74.6, 74.3, 73.9, 73.4, 72.9 (2 C), 72.8, 72.4, 72.3, 71.7, 69.0, 67.9, 67.2, 66.5 (2 C), 66.2, 61.5, 56.5, 38.4, 29.6, 26.8, 20.8, 19.5, 17.1, 16.8; ESI-MS: C<sub>110</sub>H<sub>120</sub>N<sub>2</sub>NaO<sub>23</sub>Si requires m/z 1887.7833; found: m/z 1887.7846 [M+Na]<sup>+</sup>.

3-(Benzyloxycarbonylamino)propyl 4-O-[4-O-acetyl-6-O-benzyl-3-O-(2,3,4,6-tetra-O-benzyl- $\alpha$ -D-galactopyranosyl)-2-O-(2,3,4-tri-O-benzyl- $\alpha$ -L-fucopyranosyl)- $\beta$ -D-galactopyranosyl]-6-O-t-butyldiphenylsilyl-2-deoxy-2-phthalimido-3-O-(2,3,4-tri-O-benzyl- $\alpha$ -L-fucopyranosyl)- $\beta$ -D-glucopyranoside (S3)



A solution of tetrasaccharide S2 (1.00 g, 0.54 mmol) and thioglycoside donor S4 (406 mg, 0.64 mmol) in anhydrous  $CH_2CI_2$  (30 mL) was strirred with MS-4 Å (500 mg) under N<sub>2</sub> at room temperature for 1 h. The reaction mixture was cooled to -40 °C and Niodosuccinimide (172 mg, 0.77 mmol) and TfOH (30 µL. 0.18 mmol) were added in succession. The reaction mixture was stirred at same temperature for 1 h before dilution with CH<sub>2</sub>Cl<sub>2</sub> (50 mL). The reaction mixture was filtered through a Celite<sup>®</sup> bed and the organic layer was washed with 5% aq.  $Na_2S_2O_3$ , satd.  $NaHCO_3$  and water, dried ( $Na_2SO_4$ ) and evaporated to dryness. The crude product was purified over SiO<sub>2</sub> using hexane-EtOAc (6:1) as eluent to furnish pentasaccharide **S3** (900 mg, 70%) as a yellow oil;  $\left[\alpha\right]_{D}^{25}$  + 39 (c 1.5, CHCl<sub>3</sub>); IR : 2931, 1715, 1454, 1386, 1238, 1095, 751, 697 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.69 (d, J = 7.0 Hz, 2 H, Ar-H), 7.58 (d, J = 7.0 Hz, 2 H, Ar-H), 7.43 (d, J = 7.4 Hz, 2 H, Ar-H), 7.33 (d, J = 7.4 Hz, 2 H, Ar-H), 7.29-7.02 (m, 50 H, Ar-H), 6.96-6.87 (m, 8 H, Ar-H), 6.82-6.76 (m, 5 H, Ar-H), 6.71-6.67 (m, 4 H, Ar-H), 5.60 (d, J = 3.8 Hz, 1 H, H-1<sub>D</sub>), 5.58 (d, J = 3.0 Hz, 1 H, H-4<sub>B</sub>), 5.46 (d, J = 3.0 Hz, 1 H, H-1<sub>E</sub>), 5.00 (brs, 2 H, 2 × PhCH), 4.93 (d, J = 11.2 Hz, 1 H, PhCH), 4.92 (d, J = 11.5 Hz, 1 H, PhCH), 4.86 (d, J = 7.8 Hz, 1 H, H- $1_{B}$ ), 4.84 (d, J = 8.5 Hz, 1 H, H- $1_{A}$ ), 4.80 (d, J = 9.1 Hz, 1 H, PhCH), 4.71-4.65 (m, 5 H, H-1<sub>C</sub>, H-5<sub>C</sub>, 3 × PhCH), 4.63-4.29 (m, 15 H, H-3<sub>A</sub>, H-2<sub>A</sub>, 13 × PhCH), 4.25-4.11 (m, 6 H, H-4<sub>A</sub>, H-6<sub>aA</sub>, H-5<sub>D</sub>, 3 × PhCH), 4.08 (dd, J = 2.6, 10.2 Hz, 1 H, H-3<sub>E</sub>), 4.04-3.96 (m, 4 H, H-4<sub>E</sub>, H-5<sub>E</sub>, H-2<sub>D</sub>, PhCH), 3.84 (dd, J = 2.4, 10.4 Hz, 1 H, H-3<sub>C</sub>), 3.81 (dd, J = 3.0, 10.2 Hz, 1 H, H-2<sub>E</sub>), 3.79-3.71 (m, 4 H, H-2<sub>B</sub>, H-3<sub>B</sub>, H-6<sub>aE</sub>, OC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH), 3.69 (dd, J = 3.6, 10.4 Hz, 1 H,  $H-2_{C}$ ), 3.62 (dd, J = 2.5, 10.4 Hz, 1 H,  $H-3_{D}$ ), 3.60-3.57 (m, 2 H,  $H-6_{bA}$ ,  $H-6_{bE}$ ), 3.48 (br s, 1 H, H-4<sub>C</sub>), 3.45-3.34 (m, 4 H, H-6<sub>aB</sub>, H-6<sub>bB</sub>, H-4<sub>D</sub>, OC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH), 3.29-3.25 (m, 2 H, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH), 3.15 (dd, J = 7.8, 9.3 Hz, 1 H, H-5<sub>B</sub>), 3.05 (br d, J = 9.7, 1 H, H-5<sub>A</sub>), 1.77-1.70 (m, 2 H, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH), 1.67 (s, 3 H, COCH<sub>3</sub>), 1.22 (d, J = 6.3 Hz, 3 H, H-6<sub>D</sub>), 1.15 (d, J = 6.3 Hz, 3 H, H-6<sub>c</sub>), 0.81 (s, 9 H, SiC(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  169.5,

168.8, 168.3, 157.5, 138.7-123.7 (Ar-C), 100.3, 98.5, 98.3 (2 C), 97,9, 80.1, 79.4 (2 C), 78.6, 78.2, 77.6, 77.3, 76.8, 76.1, 75.9, 75.7, 75.3, 75.0 (2 C), 74.2 (2 C), 73.9, 73.8 (2 C), 73.4, 73.3 (2 C), 73.2, 73.0, 72.8, 72.7, 71.5, 70.5, 69.5, 68.6, 68.5, 67.5, 67.1, 66.6, 65.7, 61.6, 56.7, 38.5, 29.6, 26.7, 21.2, 19.4, 17.0, 16.9; ESI-MS: *m/z* 2411.7 [M+Na]<sup>+</sup>.

3-(Methyloxycarbonylamino)propyl  $\alpha$ -L-fucopyranosyl-(1 $\rightarrow$ 2)- $\beta$ -D-galactopyranosyl-(1 $\rightarrow$ 4)-[ $\alpha$ -L-fucopyranosyl-(1 $\rightarrow$ 3)]-2-acetamido-2-deoxy- $\beta$ -D-glucopyranoside (7)



A solution of tetrasaccharide S1 (450 mg, 0.25 mmol) and ethylene diamine (0.2 mL) in nbutanol (5 mL) was heated to 90 °C for 8 h. The solvents were removed under reduced pressure and a solution of the crude product was re-dissolved in acetic anhydride-pyridine (2 mL; 1:1 v/v) and kept at room temperature for 3 h. The reaction mixture was evaporated and co-evaporated with toluene before treatment with 0.1 M NaOMe in MeOH (5 mL) for 3 h at room temperature. The solution was neutralized using Dowex 50W X-8 (H+) resin, filtered, and concentrated under reduced pressure. The crude mixture was redissolved in tetrabutylammonium fluoride in THF (1 M, 3 mL) and the reaction mixture was stirred at rt for 6 h. The crude product was purified over SiO<sub>2</sub> using hexane-EtOAc (3:1) as eluant. Finally, a solution of the product in CH<sub>3</sub>OH (2 mL) was added 20% Pd(OH)<sub>2</sub>-C (150 mg) and the reaction mixture was allowed to stir at room temperature under a positive pressure of hydrogen for 7 h. The mixture was filtered through a Celite bed and evaporated to dryness before final purification on a Sephadex LH-20 column using CH<sub>3</sub>OH–H<sub>2</sub>O (8:1 v/v) as eluant to give tetrasaccharide **7** (95 mg, 53%) as a glassy solid;  $\left[\alpha\right]_{D}^{25}$  +9.2 (*c* 1.0,D2O); IR: 3440, 2926, 2372, 1429, 1377, 1030, 679 cm<sup>-1</sup>; ESI-MS: C<sub>31</sub>H<sub>54</sub>N<sub>2</sub>NaO<sub>21</sub> requires *m/z* 813.3214; found: *m/z* 813.3221 [M+Na]<sup>+</sup>. See Table S1 for NMR data.

3-(Methyloxycarbonylamino)propyl  $\alpha$ -D-galactopyranosyl-(1 $\rightarrow$ 3)-[ $\alpha$ -L-fucopyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D-galactopyranosyl-(1 $\rightarrow$ 4)-[ $\alpha$ -L-fucopyranosyl-(1 $\rightarrow$ 3)]-2-acetamido-2-deoxy- $\beta$ -D-glucopyranoside (8)



A solution of pentasaccharide S3 (800 mg, 0.33 mmol) and ethylene diamine (0.2 mL) in nbutanol (10 mL) was heated to 90 °C for 8 h. The solvents were removed under reduced pressure and a solution of the crude product was re-dissolved in acetic anhydride-pyridine (2 mL; 1:1 v/v) and kept at room temperature for 3 h. The reaction mixture was evaporated and co-evaporated with toluene before treatment with 0.1 M NaOMe in MeOH (5 mL) for 3 h at room temperature. The solution was neutralized using Dowex 50W X-8 (H+) resin, filtered, and concentrated under reduced pressure. The crude mixture was redissolved in tetrabutylammonium fluoride in THF (1 M, 4 mL) and the reaction mixture was stirred at rt for 6 h. The crude product was purified over SiO<sub>2</sub> using hexane-EtOAc (4:1) as eluant. Finally, a solution of the product in CH<sub>3</sub>OH (2 mL) was added 20% Pd(OH)<sub>2</sub>-C (150 mg) and the reaction mixture was allowed to stir at room temperature under a positive pressure of hydrogen for 7 h. The mixture was filtered through a Celite bed and evaporated to dryness before final purification on a Sephadex LH-20 column using CH<sub>3</sub>OH–H<sub>2</sub>O (8:1 v/v) as eluant to give pentasaccharide 8 (150 mg, 50%) as a glassy solid;  $[\alpha]_D = +32.7$  (c 1.0, D<sub>2</sub>O); IR: 3446, 2916, 2382, 1419, 1387, 1060, 679 cm<sup>-1</sup>; ESI-MS: C<sub>36</sub>H<sub>64</sub>N<sub>2</sub>NaO<sub>26</sub> requires m/z 975.3739; found: m/z 975.3751 [M+Na]<sup>+</sup>. See Table S1 for NMR data.

**Protein sequences:** LTBh and CTB were expressed from *Vibrio sp.60* clones harbouring plasmids PMMB68<sup>[2]</sup> and pATA13,<sup>[3]</sup> respectively, that were kindly provided by Professor T. R. Hirst, Australian National University. The primary structure of each protein was confirmed by subjecting the plasmids to DNA sequencing. The LTBh gene is that derived from an *E. coli* H74-114 isolate of human origin.<sup>[2]</sup> Its sequence is identical to that found in the 2O2L.pdb crystal structure published by Krengel and co-workers apart from the final residue (Asn103Lys).<sup>[4]</sup> The CTB gene is identical to that for the toxin from *V. cholerae* O1 El Tor biotype and O139 serogroup.<sup>[5]</sup> This sequence differs from the toxin derived from the *V. cholerae* O1 classical biotype, at two residues (Tyr18His and Ile47Thr).<sup>[5a]</sup> The 3CHB.pdb CTB crystal structure published by Hol and co-workers depicts CTB typical of the classical biotype,<sup>[6]</sup> but with one additional mutation (Arg94His). The I47T mutant was prepared by Quickchange mutagenesis using a plasmid harbouring the El Tor CTB gene. The I47T mutation was confirmed by gene sequencing and also electrospray mass spectrometry of the expressed protein. The amino acid sequences for the mature 103-residue proteins are shown in Figure S1.

	01	11	21	31
LTBh (H74-114)	APQSITELCS	EYHNTQIYTI	NDKILSYTES	MAGKREMVII
LTBh (202L.pdb)	APQSITELCS	EYHNTQIYTI	NDKILSYTES	MAGKREMVII
CTB (classical)	TPQNITDLCA	EYHNTQI <mark>H</mark> TL	NDKIFSYTES	LAGKREMAII
CTB (3CHB.pdb)	TPQNITDLCA	EYHNTQI <mark>H</mark> TL	NDKIFSYTES	LAGKREMAII
CTB (El Tor)	TPQNITDLCA	EYHNTQI <mark>Y</mark> TL	NDKIFSYTES	LAGKREMAII
CTB (El Tor I47T)	TPQNITDLCA	EYHNTQI <mark>Y</mark> TL	NDKIFSYTES	LAGKREMAII
HYB (3EFX.pdb)	APQNITELCS	EYHNTQIYTI	NDKILSYTES	LAGKREMAII
	41	51	61	71
LTBh (H74-114)	TFKSGATFQV	EVPGSQHIDS	QKKAIERMKD	TLRITYLTET
LTBh (202L.pdb)	TFKSGATFQV	EVPGSQHIDS	QKKAIERMKD	TLRITYLTET
CTB (classical)	TFKNGA <mark>T</mark> FQV	EVPGSQHIDS	QKKAIERMKD	TLRIAYLTEA
CTB (3CHB.pdb)	TFKNGA <mark>T</mark> FQV	EVPGSQHIDS	QKKAIERMKD	TLRIAYLTEA
CTB (El Tor)	TFKNGA <mark>I</mark> FQV	EVPGSQHIDS	QKKAIERMKD	TLRIAYLTEA
CTB (El Tor I47T)	TFKNGA <mark>T</mark> FQV	EVPGSQHIDS	QKKAIERMKD	TLRIAYLTEA
HYB (3EFX.pdb)	TFKNGATFQV	EVPGSQHIDS	QKKAIERMKD	TLRIAYLTEA
1	81	91 1	L01	
LTBh (H74-114)	KIDKLCVWNN	KTPNSIAAIS	ME <mark>N</mark>	
LTBh (202L.pdb)	KIDKLCVWNN	KTPNSIAAIS	ME <mark>K</mark>	
CTB (classical)	KVEKLCVWNN	KTP <mark>H</mark> AIAAIS	MAN	
CTB (3CHB.pdb)	KVEKLCVWNN	KTP <mark>R</mark> AIAAIS	MAN	
CTB (El Tor)	KVEKLCVWNN	KTP <mark>H</mark> AIAAIS	MAN	
CTB (El Tor I47T)	KVEKLCVWNN	KTP <mark>H</mark> AIAAIS	MAN	
HYB (3EFX.pdb)	KVEKLCVWNN	KTPNSIAAIS	MAN	

**Figure S1.** Amino acid sequences of LTBh and CTB variants. LTBh (H74-114) and CTB (EI Tor) were used in the current study. Differences between the CTB sequences are highlighted in green and differences between the LTBh sequences are highlighted in blue.

Protein expression: A single bacterial colony of the appropriate clone was used to inoculate a flask of high salt LB growth media (100 mL). The starter culture was incubated at 30 °C for 20 h before an aliquot (20 mL) was added to 10 x 1 L growth media. The cultures were then incubated at 30 °C until the OD<sub>600</sub> had reached ~0.5, at which time IPTG (120 mg) was added to each flask to induce protein over-expression. After incubating for a further 20 h, the cells were isolated by centrifugation at 38,000 x g for 20 min. The bacterial pellet was discarded and sodium hexametaphosphate (2.5 g L<sup>-1</sup>) was added to the supernatant and the pH was adjusted to 3.5 by addition of HCI. The solution was then stored at 4 °C for 4 days to allow the protein to precipitate. Alternatively, the protein was recovered by ammonium sulfate precipitation. The suspension was centrifuged at 16,000 x g for 45 min, before discarding the supernatant and the protein pellet was resuspended in TRIS buffer (20 mL). This solution was centrifuged at 110,500 x g for 30 min. The supernatant was collected and filtered through a 0.45 µm filter (Sartorius Minisart). The protein solution was purified on a lactose affinity column (Sigma) and eluted with TEAN buffer (pH 7.0: 50 mM Tris, 200 mM NaCl, 3 mM NaN<sub>3</sub>, 1 mM EDTA) containing 300 mM lactose. The purity of the protein was determined to be >95% by SDS-PAGE.

**Isothermal titration calorimetry:** Protein solutions were dialysed into phosphate buffered saline (pH 7.4) using SnakeSkin® pleated dialysis tubing (Thermo Scientific) with 7000 MWCO. The buffer solution was changed three times to ensure effective dialysis. ITC experiments were performed using either a VP-ITC or a iTC200calorimeter (GE Healthcare), with a cell volume of 1.409 mL and 0.2028 mL, respectively. All titrations were conducted at 25 °C using a CTB or LTBh subunit concentration of 100  $\mu$ M (i.e. 20  $\mu$ M pentamer concentration). Ligands **7** and **8** were dissolved in the buffer from the final dialysis step. Titrations, typically run in duplícate, comprised 25-30 injections of 8  $\mu$ L at 4 minute (or 2  $\mu$ L at 2 minute) intervals. Separate titrations of ligands into the same buffer were used to subtract the heat of dilution from the integrated data prior to curve fitting in Origin using the one site model. The binding stoichiometry (n) was fixed at 1.0 for low c-value titrations as described by Turnbull and Daranas.<sup>[7]</sup>



Figure S2. ITC titrations using the lle47Thr El Tor CTB mutant. (a) titration of 32 mM tetrasaccharide 7 into 120  $\mu$ M B-subunit (black) with corresponding dilution experiment (red) and control titration of 180 mM D<sub>2</sub>O into water (blue). (b) Titration of 32 mM tetrasaccharide 7 into 120  $\mu$ M B-subunit (black) overlaid with a competition titration experiment in which 32 mM tetrasaccharide 7 was titrated into a mixture of 120  $\mu$ M B-subunit and 5.5 mM pentasaccharide 8.

Figure S2 shows titrations using the IIe47Thr EI Tor CTB mutant protein. The positive heat of dilution in panel (a) arises from a slight buffer mismatch that was a consequence of freeze-drying the oligosaccharide from a  $D_2O$  NMR sample which leads to the hydroxyl group protons exchanging for deuterium. The control titration curve shown in blue represents the amount of heat that is released when a corresponding concentration of  $D_2O$  is diluted in water. This minor buffer mismatch has no impact on the processed ITC data (Fig. S2a lower panel).

Tetrasaccharide 7 was found to bind with  $K_d = 1.0 \pm 0.3$  mM,  $\Delta H = 2.3 \pm 0.8$  kcal/mol.

When tetrasaccharide **7** is titrated into a mixture of pentasaccharide **8** and the mutant protein the enthalpy changes decrease (green line Fig S2b) relative to a titration in the absence of pentasaccharide **8** (black line Fig S2b). This result is consistent with the STD-NMR experiments which suggest that both oligosaccharides can bind in the same site on the protein. However, the small difference in the ITC curves (Fig S2) was insufficient to allow quantitative determination of the affinity and enthalpy change for binding of the pentasaccharide.

**NMR sample preparation:** Protected carbohydrate intermediates were dissolved in deuterated chloroform. Final unprotected carbohydrates were dissolved in  $D_2O$  (99.97%, Eurisotop, France). Protein-ligand complexes were dissolved in deuterated phosphate buffered saline (PBS) at pH 7.4 to give a protein concentration of 60  $\mu$ M (or 30  $\mu$ M for the I47T mutant). A 500:1 excess ratio of ligand was used.

**NMR data acquisition and analysis:** <sup>1</sup>H NMR spectra for protected carbohydrates were acquired at 500 MHz on a Bruker Avance 500 instrument or at 300 MHz on a Bruker Avance 300 instrument. <sup>13</sup>C NMR spectra for protected carbohydrates were acquired at 75 MHz on a Bruker Avance 300 instrument. Chemical shifts are given in parts per million downfield from tetramethylsilane. Monosaccharide residues are labelled A-E as indicated in the structures accompanying each synthetic procedure. All NMR measurements on final unprotected carbohydrates were performed on a Bruker AVANCE III 600 MHz spectrometer equipped with a 5 mm TCI cryogenic probe using 3 mm NMR tubes. The temperature was set at 300 K for all experiments. Topspin 3.0 was used for data acquisition, processing and analysis. Standard pulse sequences from the Bruker library were used. All experiments were acquired with 2048 points in the F2 dimension and 256 to 512 increments in the F1 dimension. HMBC was optimized for a long range coupling constant of 7 Hz. TOCSY mixing time was set to 60 ms with an excitation bandwith of 8000 Hz. Zero Quantum filtered 1D-TOCSY were acquired using Gaussian selective pulses and a mixing time of 300 ms was used in the NOESY experiment.

All STD experiments, were acquired using a train of Gaussian shaped pulses of 48 ms, spaced by 1.0 ms delays, with an excitation bandwidth of 55 Hz. Saturation times of 5.0 s and a 10 ms  $T1 \rho$  filter ( $\gamma B_1/2\pi = 6000$  Hz) was used to remove residual protein signals. On-resonance irradiation was performed both at -1 ppm and 7.5 ppm and off-resonance irradiation was set at 30.0 ppm. Appropriate blank experiments were also performed to assure the absence of direct irradiation on the ligand. Analysis of the STD data was done using the CLEAN-STD approach in order to avoid power spillover in those halves of the spectra closer to the on-resonance RF irradiation frequency.

	Lewis	-у 7			BLew	ris-y <b>8</b>	
			Multiplicity. J				
Position	$\delta_{H}$	$\delta_{C}$	(Hz)	Position	$\delta_{H}$	$\delta_{C}$	Multiplicity. J (Hz)
H1-GlcNAc-A	4.39	100.9	d, 8.0	H1-GlcNAc-A	4.39	100.8	d, 8.0
H2-GIcNAc-A	3.77	55.9		H2-GIcNAc-A	3.76	56.0	
H3-GIcNAc-A	3.48	74.8		H3-GIcNAc-A	3.63	74.8	
H4-GIcNAc-A	3.80	73.3		H4-GIcNAc-A	3.80	73.9	
H5-GIcNAc-A	3.34	75.5		H5-GIcNAc-A	3.33	75.5	
H6-GIcNAc-A	3.72	59.9		H6-GIcNAc-A	3.73	59.9	
H6'-GIcNAc-A	3.91	59.9	dd, 1.6, 12.0	H6'-GIcNAc-A	3.93	59.9	dd, 1.6, 12.0
H1-Gal-B	4.39	100.2	d, 8.0	H1-Gal-B	4.46	100.3	d, 7.6
H2-Gal-B	3.54	76.2		H2-Gal-B	3.77	72.7	
H3-Gal-B	3.74	73.5		H3-Gal-B	3.86	76.1	dd, 2.0, 9.5
H4-Gal-B	3.75	68.7		H4-Gal-B	4.15	63.1	
H5-Gal-B	3.71	74.8		H5-Gal-B	3.50	74.5	
H6-Gal-B	3.60	61.4		H6-Gal-B	3.60	61.3	
H1-Fuc-C	4.99	98.6	d, 3.9	H1-Fuc-C	4.99	98.5	d, 3.7
H2-Fuc-C	3.58	67.6		H2-Fuc-C	3.58	67.6	
H3-Fuc-C	3.80	69.1		H3-Fuc-C	3.80	69.0	
H4-Fuc-C	3.69	71.9		H4-Fuc-C	3.66	72.0	
H5-Fuc-C	4.77	66.8	q, 6.3	H5-Fuc-C	4.77	66.8	q, 6.6
H6-Fuc-C	1.13	15.4	d, 6.3	H6-Fuc-C	1.13	15.5	d, 6.6
H1-Fuc-D	5.16	99.3	d, 3.1	H1-Fuc-D	5.17	98.9	d, 4.1
H2-Fuc-D	3.68	68.2		H2-Fuc-D	3.68	67.7	
H3-Fuc-D	3.68	69.7		H3-Fuc-D	3.60	70.0	
H4-Fuc-D	3.71	71.6		H4-Fuc-D	3.71	71.7	
H5-Fuc-D	4.15	66.8	q, 6.2	H5-Fuc-D	4.22	66.8	dd, 6.1, 6.5
H6-Fuc-D	1.16	15.4	q, 6.2	H6-Fuc-D	1.18	15.3	d, 6.5
O-CH <sub>2</sub>	3.49	67.8		H1-Gal-E	5.13	92.9	d, 3.1
	3.81	67.8		H2-Gal-E	3.78	69.4	
CH <sub>2</sub>	1.63	28.7	q, 6.5	H3-Gal-E	3.77	68.0	
N-CH <sub>2</sub>	3.04	37.3	dddd, 6.7	H4-Gal-E	3.86	69.2	
Acetyl-Me	1.92	22.1	S	H5-Gal-E	4.11	71.0	dd, 6.1.6.1
Acetyl-CO		174.4		H6-Gal-E	3.62	61.1	d, 6.1
Carbamate-OMe	3.53	52.3		O-CH <sub>2</sub>	3.50	67.8	
Carbamate-CO		159.2			3.82	67.8	
				CH <sub>2</sub>	1.63	28.7	t, 6.5
				N-CH <sub>2</sub>	3.04	37.3	dddd, 6.5
				Acetyl-Me	1.92	22.1	S
				Acetyl-CO		174.3	
				Carbamate-OMe	3.54	52.3	
				Carbamate-CO		159.2	

Fable S1. Chemical shifts of	of Lewis-y (7) and	BLewis-y (8) in $D_2O$ .
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LTB	Bh-BLewis-y LTBh-Lewis-y CTB-I		B-Lewis-y		
400.0		400.0		400.0	
100.0	H5-GICNAC-A	100.0	H5-GICNAC-A	100.0	H5-GICNAC-A
55.9	H6-GIcNAc-A	89.5	H6-GIcNAc-A	70.5	H6-GIcNAc-A
110.1	H1-GlcNAc-A	140.8	H1-GlcNAc-A/	131.2	H1-GIcNAc-A/
			H1-Gal-B		H1-Gal-B
63.8	H1-Gal-B				
59.7	H4-Gal-B				
78.3	H1-Fuc-C	87.7	H1-Fuc-C	96.0	H1-Fuc-C
47.9	H5-Fuc-C	62.6	H5-Fuc-C	59.2	H5-Fuc-C
33.1	H6-Fuc-C	31.8	H6-Fuc-C	38.8	H6-Fuc-C
60.2	H1-Fuc-D	58.3	H1-Fuc-D	95.3	H1-Fuc-D
62.3	H5-Fuc-D	64.2	H5-Fuc-D	74.0	H5-Fuc-D
50.1	H6-Fuc-D	32.8	H6-Fuc-D	45.9	H6-Fuc-D
54.8	H1-Gal-E				
52.4	H5-Gal-E				
29.6	CH <sub>2</sub>	31.7	CH <sub>2</sub>	29.8	CH <sub>2</sub>
33.2	N-CH <sub>2</sub>	38.1	N-CH <sub>2</sub>	39.6	N-CH <sub>2</sub>
79.8	Acetyl	61.8	Acetyl	71.7	Acetyl

**Table S2.** Normalized STD amplification factors for the LTBh-BLewis-y. LTBh-Lewis-y and CTB-<br/>Lewis-y complexes.



**Fig S3.** Normalized STD amplification factors for the LTBh-BLewis-y. LTBh-Lewis-y and CTB-Lewis-y complexes.

CTB-I47T-BLewis-y		CTB-I	CTB-I47T -Lewis-y		
100.0	H5-GIcNAc-A	100.0	H5-GlcNAc-A		
55.4	H6-GIcNAc-A	74.2	H6-GlcNAc-A		
114.2	H1-GIcNAc-A	129.5	H1-GIcNAc-A/		
			H1-Gal-B		
66.9	H1-Gal-B				
57.2	H4-Gal-B				
83.7	H1-Fuc-C	97.2	H1-Fuc-C		
60.9	H5-Fuc-C	91.5	H5-Fuc-C		
31.6	H6-Fuc-C	26.0	H6-Fuc-C		
61.7	H1-Fuc-D	75.1	H1-Fuc-D		
62.5	H5-Fuc-D	69.8	H5-Fuc-D		
39.8	H6-Fuc-D	25.7	H6-Fuc-D		
51.1	H1-Gal-E				
44.7	H5-Gal-E				
23.0	CH <sub>2</sub>	20.7	$CH_2$		
27.7	N-CH <sub>2</sub>	27.2	N-CH <sub>2</sub>		
64.8	Acetyl	50.3	Acetyl		

**Table S3.** Normalized STD amplification factors for the CTB-I47T-BLewis-y and CTBI47T-Lewis-y complexes.



**Fig S4.** Normalized STD amplification factors for the CTB-I47T-BLewis-y and CTBI47T-Lewis-y complexes.



**Figure S5:** <sup>1</sup>H NMR spectrum of Lewis-y tetrasaccharide **7**.





Figure S7: HSQC-edited of Lewis-y tetrasaccharide 7.



Figure S8: DQF-COSY of Lewis-y tetrasaccharide 7.



FigureS9: TOCSY of Lewis-y tetrasaccharide 7.



Figure S10: HSQC-TOCSY of Lewis-y tetrasaccharide 7.







Figure S13: Expansion of the HMBC spectrum of Lewis-y tetrasaccharide 7.



Figure S14: NOESY of Lewis-y tetrasaccharide 7.



Figure S15: <sup>1</sup>H NMR spectrum of BLewis-y pentasaccharide 8.



Figure S16: <sup>13</sup>C NMR spectrum of BLewis-y pentasaccharide 8.



Figure S17: DQF-COSY of BLewis-y pentasaccharide 8.



Figure S18: HSQC-edited of BLewis-y pentasaccharide 8.



Figure S19: HSQC-TOCSY of BLewis-y pentasaccharide 8.



Figure S20: Expansion of the HSQC-TOCSY of BLewis-y pentasaccharide 8.



Figure S21: HMBC of BLewis-y pentasaccharide 8.



Figure S22: Expansion of the HMBC of BLewis-y pentasaccharide 8.



**Figure S23:** Stack-plot of standard <sup>1</sup>H (black) and 1D-TOCSY with selective excitation on anomeric protons of the BLewis-y pentasaccharide **8**: Fuc-D, Gal-E, Fuc-C and Gal-B and GlcNAc-A subunits from top to bottom. The TOCSY spectra were acquired with RF irradiations at 5.17 ppm (red) , 5.13 ppm (green), 4.99 ppm (purple), 4.47 ppm (yellow) and 4.40 ppm (orange), respectively.



**Figure S24:** Stack-plot of standard <sup>1</sup>H (black) and 1D-TOCSY with selective excitation on H5 protons of the BLewis-y pentasaccharide **8**: Fuc-C, Fuc-D, Gal-E and Gal-B subunits from top to bottom (The OCH<sub>2</sub> protons of the side chain are overlapped with the Gal-B H5 proton). The TOCSY spectra were acquired with RF irradiations at 4.76 ppm (red) , 4.23 ppm (green), 4.10 ppm (purple) and 3.49 ppm (yellow), respectively.



**Figure S25:** Stack-plot of standard <sup>1</sup>H (black) and 1D-TOCSY with selective excitation on H4-Gal-B and H6-GlcNAc-A protons of the BLewis-y pentasaccharide **8** subunits (from top to bottom). The TOCSY spectra were acquired with RF irradiations at 4.17 ppm (red) , 3.91 ppm (green), respectively.



**Figure S26:** Standard <sup>1</sup>H (bottom) and STD-NMR spectra (top) of the CTB - Lewis-y **7** complex. The STD-NMR spectra were acquired with RF irradiations at -1 ppm (green) and 7.5 ppm (red). Receiver gain and number of scans were identical in both STD experiments.



**Figure S27:** Standard <sup>1</sup>H (bottom) and STD-NMR spectra (top) of the LTBh - Lewis-y **7** complex. The STD-NMR spectra were acquired with RF irradiations at -1 ppm (green) and 7.5 ppm (red). Receiver gain and number of scans were identical in both STD experiments.



**Figure S28:** Standard <sup>1</sup>H (bottom) and STD-NMR spectra (top) of the LTBh - BLewis-y **8** complex. The STD-NMR spectra were acquired with RF irradiations at -1 ppm (green) and 7.5 ppm (red). Receiver gain and number of scans were identical in both STD experiments.



**Figure S29:** Standard <sup>1</sup>H (bottom) and STD-NMR spectra (top) of the CTB - BLewis-y **8** complex. The STD-NMR spectrum was acquired with a RF irradiation at 7.5 ppm (red).

#### Comparison of the crystal structure of LTBh with models of El Tor CTB

All CTB crystal structures in the Protein Data Bank are for classical biotype CTB with a point mutation at position 94 (His94Arg). Therefore models of EI Tor CTB were constructed using the crystal structure of either classical biotype CTB (3CHB.pdb)<sup>[6]</sup> or a CTB-LTBh hybrid protein (3EFX.pdb)<sup>[8]</sup> by substitution of appropriate residues in the binding pocket. In each case three mutations were made. The models were used purely for a visual comparison with the complex of LTBh and A-Lewis-y **3b** (2O2L.pdb) in order to identify which amino acids might be responsible for preventing B-Lewis-y pentasaccharide **8** binding to EI Tor CTB.<sup>[4]</sup> Ligand **8** used in our studies differs from compound **3b** in two regards: **3b** has a glucose at the reducing terminus and a GalNAc at the non-reducing terminus, whereas **8** has a GlcNAc at the reducing terminus and a Gal at the non-reducing terminus. A model for ligand **8** was constructed using the coordinates for **3b** in the 2O2L.pdb structure. The coordinates for ligand **3b** in models shown below is taken from the 3EFX.pdb structure. The relocation of the acetamide group in converting **3b** into **8** is not predicted to make any unfavourable steric clash with the proteins.



Ligand **8** (green) with EI Tor CTB model derived from 3CHB.pdb (yellow) and LTBh 2O2L.pdb (blue)



Ligand **3b** (pink) with EI Tor CTB model derived from 3CHB.pdb (yellow) and LTBh 2O2L.pdb (blue)



Ligand 8 (green) with EI Tor CTB model derived from 3EFX.pdb (yellow) and LTBh 2O2L.pdb (blue)



Ligand **3b** (pink) with EI Tor CTB model derived from 3EFX.pdb (yellow) and LTBh 2O2L.pdb (blue)

Figure S30. Models depicting a possible binding mode for pentasaccharide 8









#### References

- [1] P. K. Mandal, W. B. Turnbull, *Carbohydr. Res.* **2011**, *346*, 2113-2120.
- [2] M. Sandkvist, T. R. Hirst, M. Bagdasarian, J. Bacteriol. 1987, 169, 4570-4576.
- [3] A. T. Aman, S. Fraser, E. A. Merritt, C. Rodigherio, M. Kenny, M. Ahn, W. G. J. Hol, N. A. Williams, W. I. Lencer, T. R. Hirst, *Proc. Natl. Acad. Sci. USA* **2001**, *98*, 8536-8541.
- [4] A. Holmner, G. Askarieh, M. Ökvist, U. Krengel, J. Mol. Biol. 2007, 371, 754-764.
- [5] a) O. Olsvik, J. Wahlberg, B. Petterson, M. Uhlen, T. Popovic, I. K. Wachsmuth, P. I. Fields, J. Clin.
- Microbiol. 1993, 31, 22-25; b) M. Lebens, J. Holmgren, FEMS Microbiol. Lett. 1994, 117, 197-202.
- [6] E. A. Merritt, P. Kuhn, S. Sarfaty, J. L. Erbe, R. K. Holmes, W. G. J. Hol, J. Mol. Biol. 1998, 282, 1043-1059.
- [7] W. B. Turnbull, A. H. Daranas, J. Am. Chem. Soc. 2003, 125, 14859-14866.
- [8] A. Holmner, M. Lebens, S. Teneberg, J. Angstrom, M. Ökvist, U. Krengel, *Structure* **2004**, *12*, 1655-1667.