

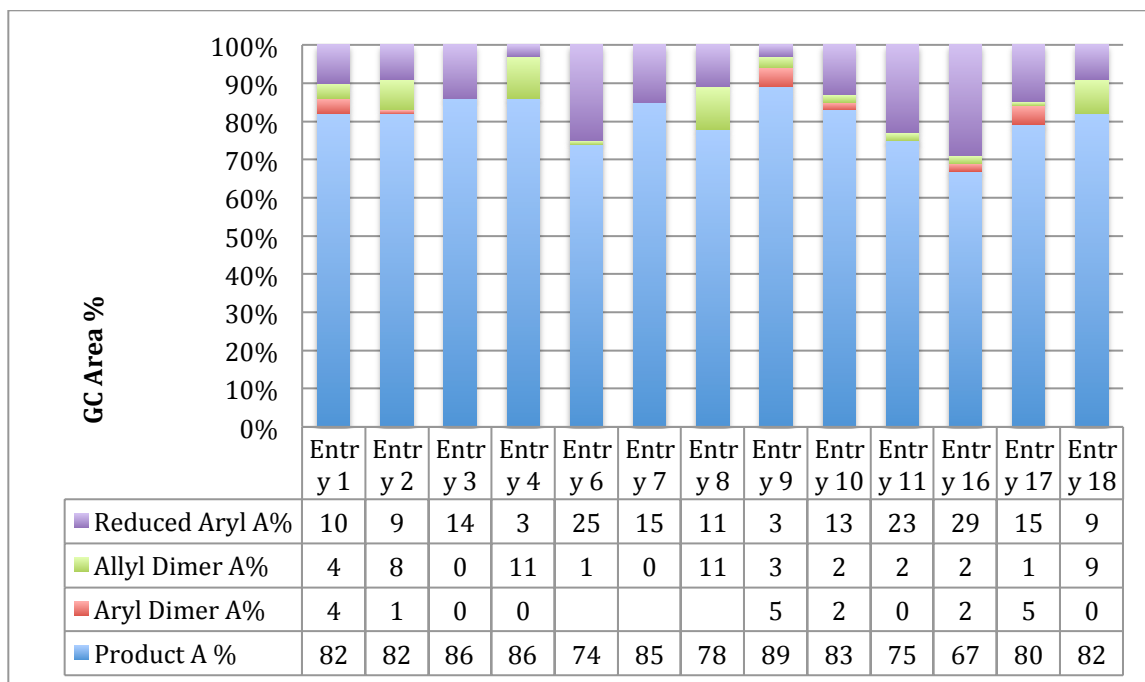
# Direct Cross-coupling of Organic Halides with Allylic Acetates

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## Supporting Information

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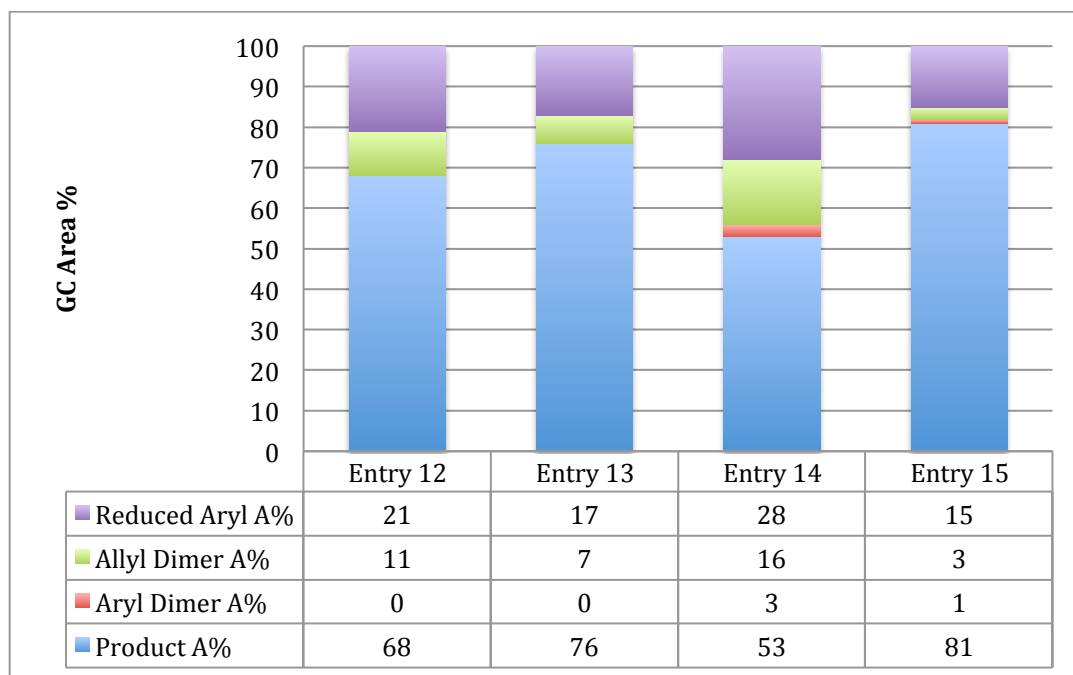
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**Table S1:** Selectivity Data for the Coupling Reactions of Iodoarenes with Cinnamyl Acetate in Table 2<sup>a</sup>

<sup>a</sup>GC area% ratios are uncorrected. The high molecular weight of both the product and the aryl dimer for entry 5 prevented GC analysis. The similarly, the high MW for the aryl dimer of entries 6, 7 & 8 prevented detection by GC.

Entry	Ar-X	product	yield <sup>b</sup>
1	I	1 mmol scale, set up in glovebox	88
2	I	10 mmol scale, set up on benchtop	81
3	I	R = C(O)Me ( <b>3b</b> )	71
4	I	R = CHO ( <b>3c</b> )	70
6	I	R = NHC(O)CF <sub>3</sub> ( <b>3e</b> )	64
7	I	R = CH <sub>2</sub> OTBS ( <b>3f</b> )	80
8	I	R = NMe <sub>2</sub> ( <b>3g</b> )	55
9	I	R = Me ( <b>3h</b> )	86
10	I	R = OMe ( <b>3i</b> )	83
11	I	R = Br ( <b>3j</b> )	64
16	I		78
		<b>(3n)</b>	
17	I	R = CN ( <b>3o</b> )	86
18	I	R = OMe ( <b>3p</b> )	80

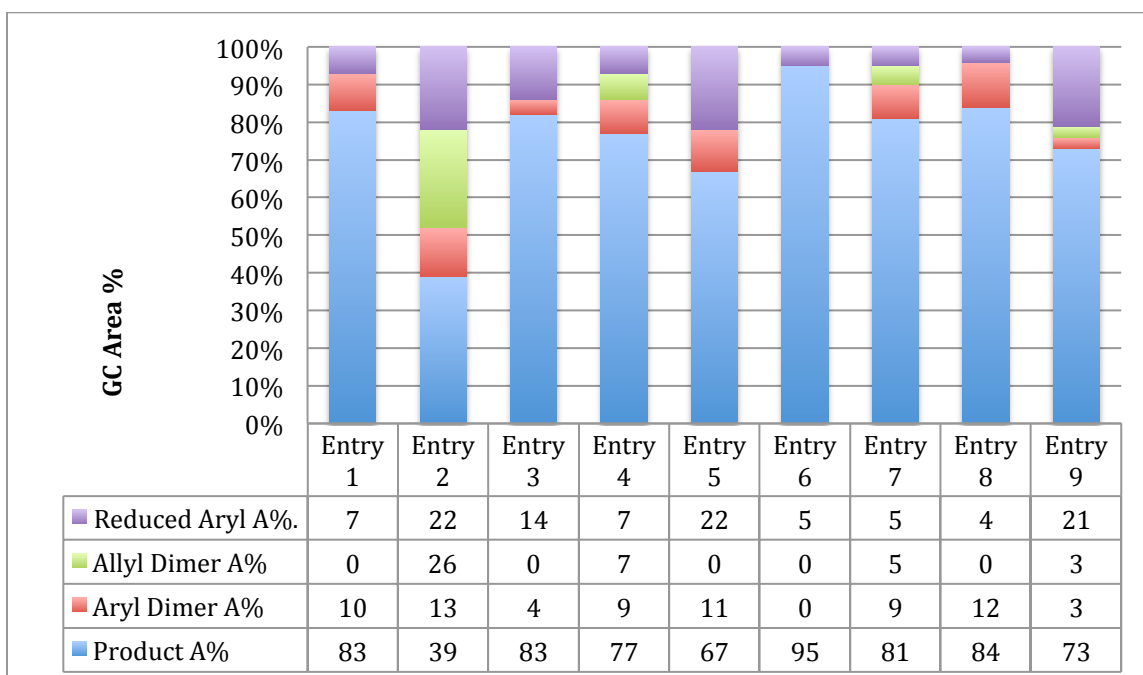
<sup>b</sup> Isolated yields from Table 2.

**Table S2:** Selectivity Data for the Coupling Reactions of Bromoarenes with Cinnamyl Acetate in Table 2<sup>a</sup><sup>a</sup>GC area% ratios are uncorrected.

Entry	Ar-X	product	yield <sup>b</sup>
12	Br	R = CO <sub>2</sub> Me ( <b>3k</b> )	65
13	Br	R = C(O)Me ( <b>3b</b> )	48
14	Br	R = CF <sub>3</sub> ( <b>3l</b> )	51
15	Br	R = CN ( <b>3m</b> )	77

<sup>b</sup> Isolated yields from Table 2.

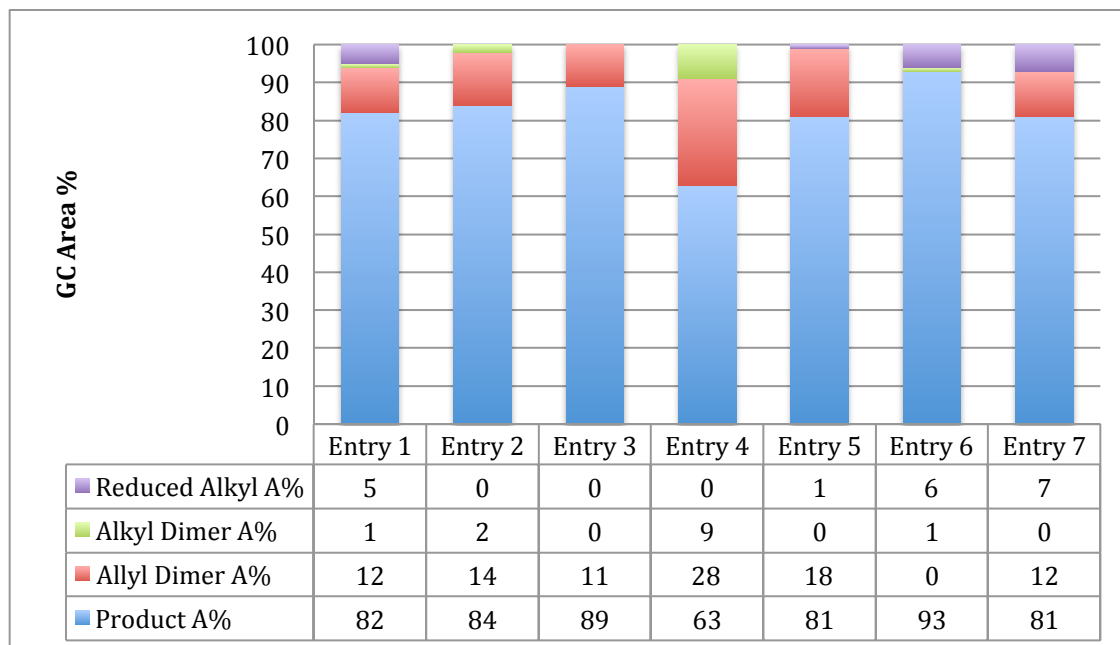
**Table S3:** Selectivity Data for the Coupling Reactions of Substituted Allylic Acetates with Iodoarenes in Table 3<sup>a</sup>



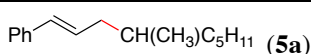
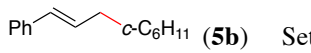
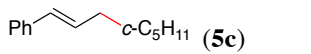
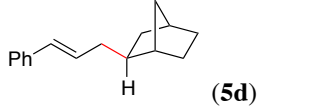
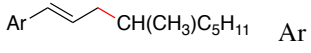

<sup>a</sup>GC area% ratios are uncorrected.

Entry	allylic acetate	Product	yield <sup>b</sup>
1			81
2			52
3			75
4			97
5			55
6			65
7			52
8			80
9			73

<sup>b</sup> Isolated yields from Table 3.

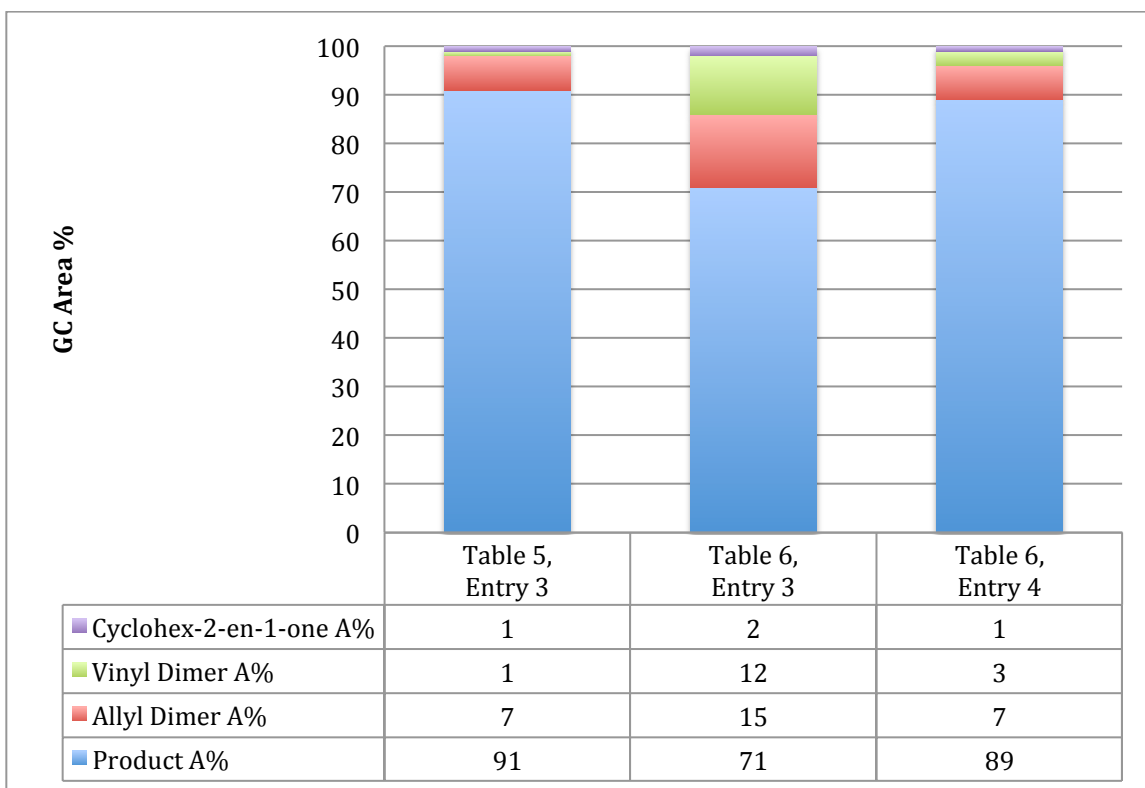
**Table S4:** Selectivity Data for the Coupling Reactions of Allylic Acetates with Alkyl Bromides in Table 4<sup>a</sup>

<sup>a</sup>GC area% ratios are uncorrected. All entries except entry 3 are an average of 2 runs. The alkyl reduction products of entries 2-4 were too low boiling and would elute with solvent, preventing detection by GC analysis. The allyl dimer for entry 6 was too high boiling for GC analysis on our instrument.

Entry	product	yield <sup>b</sup>
1	 <b>(5a)</b>	79
2	 <b>(5b)</b> Set up in glovebox	88
3	Set up in the benchtop <sup>c</sup>	90
4	 <b>(5c)</b>	68
5	 <b>(5d)</b>	78
6	 <b>(5e)</b> Ar = <i>p</i> -MeO-C <sub>6</sub> H <sub>4</sub>	79
7	 <b>(5f)</b> Ar = <i>p</i> -F <sub>3</sub> C-C <sub>6</sub> H <sub>4</sub>	66

<sup>b</sup> Isolated yields from Table 4.

**Table S5:** Selectivity Data for the Coupling Reactions of Allylic Substrates with a Vinyl Bromide in Tables 5 and 6<sup>a</sup>

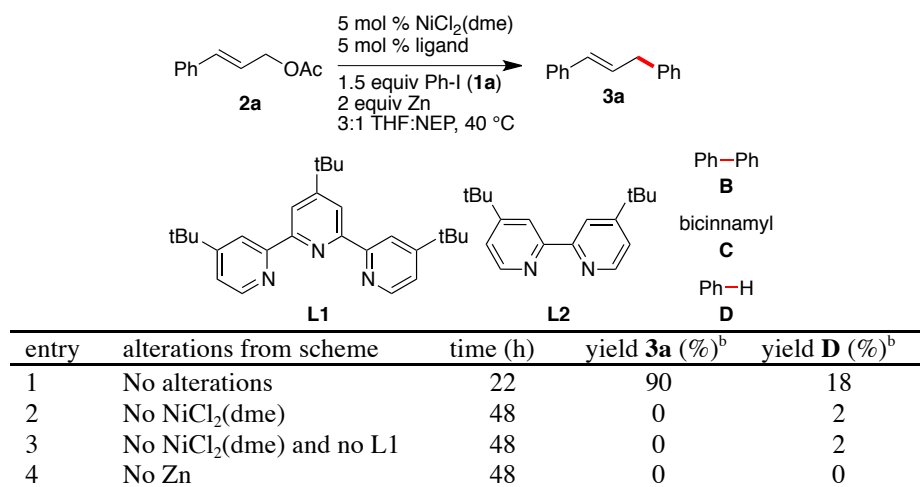


<sup>a</sup> GC area% ratios are uncorrected.

Entry	Starting Material	Product	Yield <sup>b</sup>
Table 5, Entry 3	 <b>(2c)</b>	 <b>(7a)</b>	78
Table 6, Entry 3	 <b>(2m)</b>	 <b>(7b)</b>	51
Table 6, Entry 4	 <b>(2n)</b>	 <b>(7c)</b>	67

<sup>b</sup> Isolated yields from Tables 5 and 6.

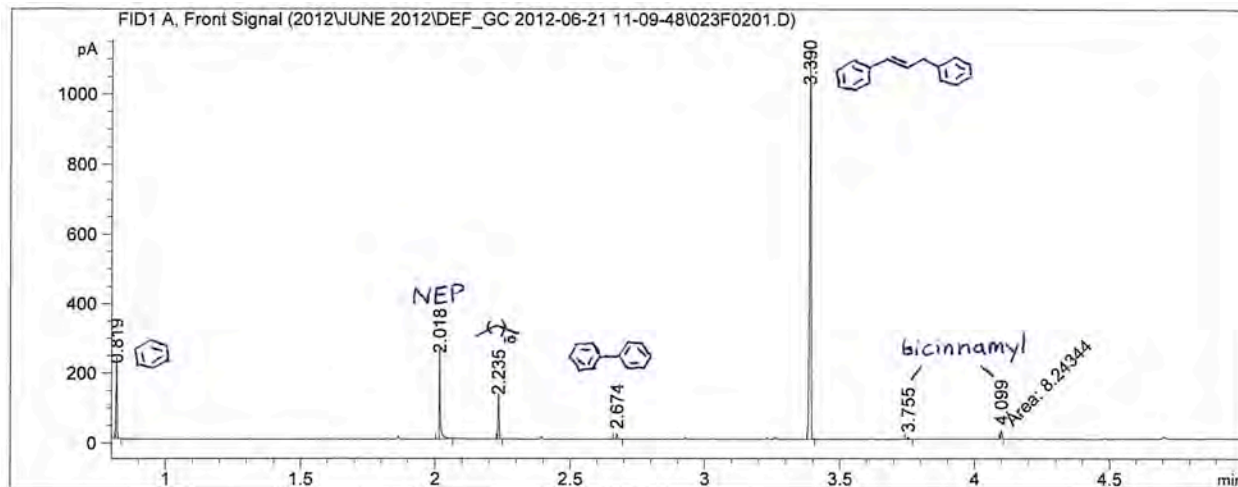
**Table S6:** Control reactions for the coupling of cinnamyl acetate with iodobenzene<sup>a</sup>



<sup>a</sup> Reactions were run on a 0.5 mmol scale in 1 ml of 3:1 THF:NEP. <sup>b</sup> Corrected GC yields vs. an internal standard (dodecane). Amount of **4** and **5** produced was negligible in all cases but the standard reaction.

## II. Example GC Yield Calculation

GC chromatogram from Table 1, Entry 1 (with terpyridine **L1**):



=====  
 Area Percent Report  
 =====

Sorted By : Retention Time  
 Multiplier: : 1.0000  
 Dilution: : 1.0000  
 Do not use Multiplier & Dilution Factor with ISTDs

Signal 1: FID1 A, Front Signal

Peak #	RetTime [min]	Sig	Type	Area [pA*s]	Height [pA]	Area %
1	0.819	1	BB	64.71152	201.13582	10.38658
2	2.018	1	BB	72.92727	231.00208	11.70526
3	2.235	1	BB	37.65450	112.79506	6.04377
4	2.674	1	BB	5.44882	14.77481	0.87457
5	3.390	1	BB	431.64377	1054.72852	69.28137
6	3.755	1	BB	2.40078	5.21727	0.38534
7	4.099	1	MM	8.24344	24.97758	1.32312

Totals : 623.03010 1644.63112

$$\text{mmol Product} = \frac{(\text{Area Product}) \times (\text{mmol Internal Standard})}{(\text{Area Internal Standard}) \times (\text{Correction Factor})}$$

$$\text{mmol Product} = \frac{431.64377 \times 0.044}{37.65450 \times 1.12}$$

$$\text{mmol Product} = 0.45$$

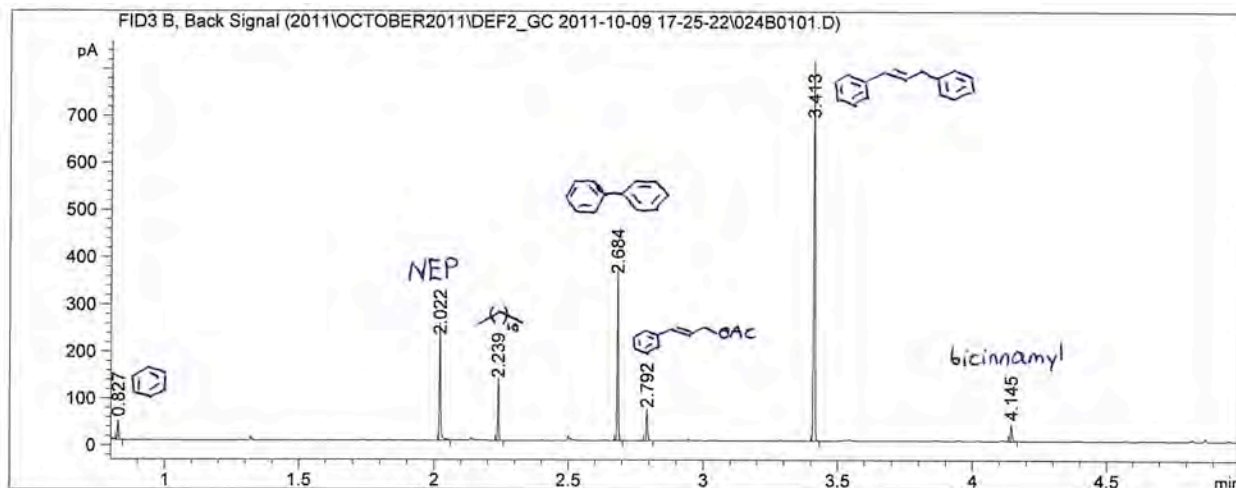
$$\text{GC Yield} = \frac{\text{mmol Product}}{\text{mmol Theoretical}} \times 100\%$$



$$GC\ Yield = \left(\frac{0.45}{0.50}\right) \times 100\%$$

$$GC\ Yield = 90\%$$

GC chromatogram from Table 1, Entry 2 (with bipyridine **L2**):



=====  
 Area Percent Report  
 =====

Sorted By : Retention Time  
 Multiplier: : 1.0000  
 Dilution: : 1.0000  
 Do not use Multiplier & Dilution Factor with ISTDs

Signal 1: FID3 B, Back Signal

Peak #	RetTime [min]	Sig	Type	Area [pA*s]	Height [pA]	Area %
1	0.827	1	VB	14.26214	38.55714	2.38767
2	2.022	1	BB	79.97160	214.97862	13.38831
3	2.239	1	BB	38.76766	123.18441	6.49022
4	2.684	1	BB	114.61913	344.47751	19.18877
5	2.792	1	BB	23.92602	59.99182	4.00553
6	3.413	1	BB	309.75629	675.46704	51.85733
7	4.145	1	BB	16.02118	34.36098	2.68216

Totals : 597.32401 1491.01752

Similar to above, the calculated GC Yield is 62%.