## Mutagenicity and Antibacterial Activity of Hydroxamic Acids

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**Received for publication 16 November 1976** 

Hydroxamic acids demonstrated mutagenic and antibacterial activities. These activities appear to be dependent on the hydroxamic acid function and are probably due to the interaction with deoxyribonucleic acid.

Hydroxamic acids have been shown to be effective antifungal and antibacterial agents. Sorbic hydroxamic acid (2) and decanoylhydroxamic acid (3) demonstrated antifungal activities. Salicylhydroxamic acid had in vitro and in vivo antitubercular activity (8). Hase et al. (5) synthesized a series of alkyloxybenzoylhydroxamic acids and fatty acyl hydroxamic acids and found that some of these chemicals possessed antifungal activity. However, none of them was effective against enteric bacteria. Acetohydroxamic acid was an inhibitor of bacterial urease. It also inhibited ammonia and stone formation in the urinary tract during infections (4). The present study investigated the mutagenic and antibacterial activity of several hydroxamic acids with differing acyl groups.

The Ames spot test for chemical mutagenicity (1) was used for the present study. Fifty microliters of sterile dimethyl sulfoxide containing 1 mg of the tested chemical was spotted on the center surface of an agar plate containing Salmonella typhimurium strains TA100 or TA98 (7). The control plates were treated with the solvent only. After incubation at  $37^{\circ}$ C for 2 days, the revertant colonies that appeared were counted, and the clear inhibition zone produced by antibacterial activity was measured. Chemicals inducing more than twice the number of revertants found in the control plate were considered as mutagenic.

In the antibacterial tests, various strains of bacteria (Table 1) were grown in Difco nutrient broth for 6 h at 37°C. The culture was then diluted 100 times with fresh nutrient broth. One-half milliliter of dimethyl sulfoxide containing various amounts of test compounds was added to 4.5 ml of the diluted culture. After incubation at 37°C for 24 h, the minimal concentration for complete growth inhibition was recorded. The concentrations of hydroxamic acids in the medium were 1000, 500, 100, 50, 10, and 1  $\mu$ g/ml. Minimal inhibitory concentrations of 10  $\mu$ g/ml or less were considered as "effective," 50 to 100  $\mu$ g/ml as "slightly effective," and 100  $\mu$ g/ml or greater as "ineffective."

The results of mutation and antibacterial tests were presented in Table 1. All the hydroxamic acids tested except *p*-butoxyphenylacetohydroxamic acid produced clear antibacterial zones and mutations in both S. typhimurium strains TA100 and TA98. It appears that these mutagenic hydroxamic acids may interact with bacterial deoxyribonucleic acid. 2-Naphthoylhydroxamic acid was "effective" against Streptococcus faecalis and Escherichia coli and was "slightly effective" against Bacillus cereus and Proteus morganii. Benzoylhydroxamic acid, salicylhydroxamic acid, and indole-2-carbohydroxamic acid were "slightly effective" against some bacterial strains. p-Butoxyphenylacetohydroxamic acid lacked mutagenicity for S. typhimurium and was virtually ineffective against the tested bacteria.

Thus, hydroxamic acid function appears to be essential for the mutagenic and antibacterial activities of hydroxamic acids, whereas the acyl moiety appears to determine their potency. Although hydroxamic acids have been shown to be effective inhibitors of bacterial and plant urease (6), their antibacterial activity probably is not due to urease inhibition (5). It seems likely that the antibacterial activity is due to their interaction with bacterial deoxyribonucleic acid since hydroxamic acids were shown to have mutagenic and antibacterial effects in this study.

	TABLE 1. Antibacterial and mutagenic activity of hydroxamic acids Minimal	genic activity	y of hydr	xamic a Min	<i>cids</i> imal inhi	bitory co	ncentratio	c acids Minimal inhibitory concentration (µg/ml)		
Hydroxamic acid	Structure	nicity <sup>a</sup>	٩I	H	E	N	v	М	IIA	ШЛ
Benzoylhydroxamic acid	CNHOH	+	1,000	1,000	NA€	500	NA	1,000	100	100
Salicylhydroxamic acid	HOHOHOH	+	500	1,000	1,000	500	NA	AN	500	50
Benzoylaminoacetohydroxamic acid	CONHCH2CNHOH	+	NA	NA	NA	NA	NA	NA	1,000	10
p-Butoxyphenylacetohydroxamic acid	n-Butoxy-CH2CH2CHHOH	I .	NA	NA	NA	NA	NA	NA	NA	500
Indole-2-carbohydroxamic acid	HOHND CINHOH	+	50	500	500	500	1,000	500	500	50
2-Naphthoylhydroxamic acid	OO CONCEPTION	+	100	500	500	100	NA	500	10	10
<ul> <li>* +, Mutagenic for S. typhimurium</li> <li><sup>b</sup> I, Bacillus cereus; II, Corynebacteri cus aureus; VII, Streptococcus faccalis,</li> <li><sup>c</sup> NA, Not active at a dose of 500 µg/ hydroxamic acid was not soluble at 1</li> </ul>	<ul> <li><sup>a</sup> +, Mutagenic for S. typhimurium TA98 and TA100; -, not mutagenic.</li> <li><sup>b</sup> I, Bacillus cereus; III, Corynebacterium xerosis; III, Klebsiella pneumoniae; IV, Proteus morganii; V, Pseudomonas aeruginosa; VI, Staphylococcus faecalis; VIII, Escherichia coli.</li> <li><sup>c</sup> NA, Not active at a dose of 500 µg/ml for p-butoxyphenylacetohydroxamic acid or 1,000 µg/ml for the other compounds. p-Butoxyphenylaceto-hydroxamic acid or 1,000 µg/ml for the other compounds. p-Butoxyphenylaceto-hydroxamic acid was not soluble at 1 mg/ml in the medium.</li> </ul>	IV, <i>Proteus</i> c acid or 1,0	morgani 30 µg/ml	i; V, <i>Psei</i> for the o	<i>idomona</i> ther com	<i>s aerug</i> ı pounds.	inosa; VI p-Butox	l, <i>Staphy</i> yphenyls	lococ- aceto-	

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This investigation was supported by Public Health Service research grant CA 17449 from the National Cancer Institute.

We thank F. Baier for the assistance in the preparation of this manuscript.

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