SI APPENDIX

Carboxylic acid reductase is a versatile enzyme for the conversion of fatty acids into fuels and chemical commodities.

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Methods

Strains and plasmids

The genes encoding for *M. marinum* CAR and *B. Subtilis* Sfp were synthesized with codonoptimization for *E. coli* (Genscript, USA), while *tesA*, *slr1192* and *ahr* were amplified from *E. coli* BL21(DE3) and *Synechocystis* sp. PCC 6803 genomes using the following PCR program: 95 °C for 2 min; 40 cycles of 95 °C for 30 s, 56 °C for 30 s and 72 °C for 1 min; and 72 °C for 20 min. Genetic inserts along with T7-based commercial vectors (Novagen) were cut with the appropriate restriction enzymes (New England Biolabs) and ligated using T4 DNA ligase (Fermentas) (Tables S3-S5). Multiple genes were assembled in artificial operons as described previously (1). Plasmids were used to transform *E. coli* BL21 (DE3) to generate the strains listed in Table S6. Table S7 summarizes the gene-products used to engineer the recombinant pathways.

Protein expression and purification

Overnight LB-grown pre-cultures of BL21(DE3) were used to inoculate 20 ml cultures of Overnight ExpressTM Instant TB Medium (Novagen) at 2% (v/v). Cultures were incubated overnight (18-24 h, 30 $^{\circ}$ C, 250 rpm). Cells were pelleted and resuspended in a lysis buffer containing lysozyme (2 mg/ml) and 2% (v/v) hexane, and incubated for 30 min at room temperature with gentle inversion. The insoluble debris was centrifuged (17,000*g*, 5 min) and the supernatant was applied to a microfuge spin column pre-filled with 200μ l His-Select[®] Nickel Affinity Gel (Sigma Aldrich). The column was washed five times with 0.5 ml 0.1 M Tris-HCl (pH 7.5, 21 $^{\circ}$ C) and the recombinant his-tagged protein eluted with 100 µl 0.4 M imidazole, prepared in 50 mM Tris-HCl buffer (pH 7.5, 21 $^{\circ}$ C). Protein recovery was estimated using the Bio-Rad Protein Assay (Biorad).

Enzyme characterization

All enzyme reactions were performed in triplicates and monitored at 340 nm for up to 15 min in 96-well microplates (Tecan M200 Affinity). Control reactions without addition of the purified enzyme were also included to take into account any background oxidation of NADPH. For the CAR assay, a 100 μl reaction volume typically contained the following components: CAR_{his} (0-10 μ g/ml), 1 mM NADPH, 1 mM ATP, 10 mM MgCl₂ and 0.5 mM fatty acids. For K_m and V_{max} determinations, the various (co)substrates: fatty acids, NADPH and ATP were prepared at 11 different concentrations with the concentration range specified in the figure legends. Based on the initial reaction rates, the apparent K_m and V_{max} values were determined using the enzyme kinetics module of SigmaPlot (Systat Software, San Jose, CA). For the AHR assay, reactions typically contained the following components: $YjgB_{his}$ (1 or 10 μ g/ml), 1 mM NADPH and 0.5 mM C₄-C₁₂ aldehydes or C₄-C₁₂ alcohols. For qualitative confirmation of aldehyde (in the case of CARhis) and alcohol synthesis (in the case of YigB_{his}), reaction mixtures (500 μ l) were mixed with an organic solvent (75 μ l) and the organic phase analyzed by GC-MS, as described below.

In vivo **production of fatty alcohols and alkanes**

Strains were cultivated either in Overnight Express™ Instant TB Medium (Novagen) or in a defined minimal medium containing: M9 salts (Sigma Aldrich); BME vitamins (Sigma Aldrich); 100 mM potassium phosphate buffer (pH 7.5, 21 $^{\circ}$ C); 2 mM MgSO4; 0.1 mM CaCl₂; micronutrient mix consisting of 10 nM FeSO₄, 3 μ M (NH₄)₆Mo₇O₂₄, 0.4 mM boric acid, 30 μM CoCl₂, 15 μM CuSO₄, 80 μM MnCl₂ and 10 μM ZnSO₄; 2% (w/v) glucose; appropriate antibiotics, 50 μ g/ml ampicillin and/or 50 μ g/ml spectinomycin; and 50 μ M IPTG. Cultures (2-5ml) were incubated for up to 48 h with shaking at 180 - 200 rpm in 50 ml sterile Falcon tubes. For total fatty alcohol and alkane quantification, 100 µl cell culture was vigorously mixed with 200 μ l acetone, microfuged $(17,000g, 5 \text{ min})$ and the resulting supernatant analyzed by GC-MS as described below. Glucose levels were quantified at 340nm, based on the reduction of NAD⁺ catalyzed by glucose-6-phosphate dehydrogenase.

In vitro **conversion of fatty acids to fatty alcohols and alkanes**

In vitro alkane synthesis was carried out as described previously (2) with the addition of CAR_{his} (100 µg/ml), 1 mM NADPH, 1 mM ATP and 0.5 mM fatty acid (C₄-C₁₆); C₁₄-C₁₆ fatty acid substrates were added as suspensions due to poor solubility. For *in vitro* fatty alcohol formation, ADC_{his} was replaced with Ahr_{his} (10 μ g/ml) and NADH, Nphenylmethazonium methosulphate (PMS) & ferrous ammonium sulphate were omitted. All tubes were incubated in a heat block at 30 $^{\circ}$ C for up to 4 h without shaking. For analyte extraction, reaction mixtures were terminated either by the addition of acetone or chloroform. The organic phase/supernatant was analyzed by GC-MS as described below.

In vitro **conversion of fatty acids to fatty alcohols and alkanes**

For *in vitro* alkane synthesis, reactions were performed in 500 µ reaction volumes: 50 mM, Tris-HCl (pH 7.5, 21^oC), CAR_{his} (100 µg/ml), ADC_{his} (200 µg/ml), 1 mM NADPH, 1 mM ATP, 10 mM $MgCl₂$, 20 µM PMS, 1 mM NADH, 20 µM ferrous ammonium sulphate and 0.5 mM fatty acid substrates ranging from C_4 (butyric acid) to C_{16} (hexadecanoic acid). The protocol was modified according to ADC assays described previously (2). For *in vitro* fatty alcohol formation, ADC_{his} was replaced with YjgB_{his} (10 μ g/ml) and the following components were omitted: NADH, PMS & ferrous ammonium sulphate. Due to their hydrophobicity, the C_{14} and C_{16} fatty acid substrates were added to the assays as suspensions. All reactions were performed in a heat block at 30 $^{\circ}$ C for up to 4 h without shaking, and terminated either by addition of acetone or chloroform. The supernatant or organic phase was analyzed by GC-MS as described below. Given that conversion was dependent on two NAPH-requiring reactions, fatty alcohol formation was also monitored by following the oxidation of NADPH at 340 nm.

In vitro **conversion of TAGs to aldehydes**

Reactions were performed in a 1 ml volume containing the following: 50mM Tris-HCl (pH 7.5, 21^oC), CAR_{his} (300 µg/ml), lipase (100 µg/ml), 1 mM NADPH, 1 mM ATP, 10 mM $MgCl₂$ and 1 mM suspensions or emulsions of the commercially purified TAGs: $C₈$ TAG (glyceryl trioctanoate) and C_{12} TAG (glyceryl tridodecanoate). All tubes were incubated in a heat block at 30 \degree C for up to 4 h. Reactions were terminated by vigorously mixing with an equal of chloroform. The lower organic phase was analyzed directly by size-exclusion HPLC as described below. For reactions containing C_{12} TAG, samples were concentrated by evaporation of solvent in a Genevac $(30 °C, 10 min)$ prior to HPLC analysis.

Lipase-mediated *in vivo* **formation of fatty alcohols**

A 1 ml preinduced cell culture (OD ~10) of the strain, PC-Ahr_{his}, encoding for the CAR_{his} and Ahrhis enzymes was centrifuged (7000*g*, 10 min). The cell pellet was resuspended in an equal volume of 100 mM potassium phosphate buffer (pH 7.5 , 21° C) together with three distinct sources of TAGs: (i) harvested cells of the cell wall-less *Chlamydomonas reinhardtii* strain cc406, (ii) palm oil (Afroase) and (iii) coconut oil (Biona Organic). Cells were supplemented with 100mM glucose and incubated at 30 $^{\circ}$ C with 100 µl samples taken at 30 min, 1 h, 2 h and 5 h. Fatty alcohols were analyzed by GC-MS as described below.

Hydrocarbon analysis by GC-MS

Metabolite analysis was performed with an Agilent 7890A gas chromatograph equipped with a 5975 mass spectrometry detector. All samples $(1 \mu l)$ were analyzed in splitless injection mode with the inlet temperature set at 300 °C and passed through an Equity-1 fused silica capillary column (Supelco) (30 m x 320 μ m x 1 μ m) at a flow rate of 1.9 ml/min, using helium as the carrier gas. The oven was initially held at 45 °C for 2.5 min and ramped up to 300 °C at a rate of 20 °C/min. Data was acquired within the 25–350 m/z range. For analyte identification, fragmentation patterns and retention times of the analytes were compared with the NIST mass spectral library and commercially available standards of fatty alcohols, acids and alkanes. A standard curve for quantification was prepared with commercial preparations of fatty alcohols and alkanes.

Hydrocarbon analysis by HPLC-size exclusion chromatography

For separation and detection of the TAGs, fatty acids and aldehydes, size-exclusion HPLC analysis was carried out using an Agilent 1200 Series HPLC module coupled to an Agilent refractive index detector. For analyte separation, samples (10 µl) were injected and passed through 3 size-exclusion columns (300 x 7.8 mm x 5 or 10 µm; Phenomenex) of different pore sizes; 50, 100 and 500 Å; and connected in series, at a flow rate 0.8 ml/min. The chromatography was performed in isocratic mode using super-purity grade chloroform (Romil) as the mobile phase with column and refraction index detector temperatures set at 40 °C. The retention times of the analytes were confirmed using commercial standards.

Stoichiometric evaluation

The stoichiometric conversion efficiency (percent conversion, mole product per mole glucose) was calculated by two independent methods, as described in detail below:

A maximum potential yield of C_{12} fatty alcohol per glucose is 0.33 and 0.288, respectively, without biomass formation. All stoichiometric evaluations were made in minimal media. Where an error is given in the text it represents the standard error (SEM, n=2-4).

The experimentally measured distribution of fatty alcohols in all TPC-Ahr strains was highly similar regardless of cultivation conditions, as summarized below:

The molar demand for glucose that is required for each of the main fatty alcohols, identified in the TPC-Ahr strain, was estimated based on the assumption that approximately 3 mole of $CO₂$ is released per mole of catabolized glucose through substrate oxidation. The molar demand for glucose was related to the distribution of fatty alcohols that was repeatedly observed; this is summarized in the Table below:

The analysis suggests that the maximum potential molar yield of fatty alcohols with the TPC-Ahr strains is 30% (mole of fatty alcohols per mole of glucose), however, under these conditions there would be no biomass formation. Since fatty alcohol production was studied under exponential batch growth conditions, the same analysis was also repeated with the assumption that 25% of all glucose is used for biomass-formation. The maximum yield of fatty alcohols is then 22.4%.

An alternative approach to estimate the maximum potential yield of fatty alcohols is to carry out the so-called 'flux balance analysis'. The maximum potential molar yield was calculated for C¹² alcohol with a stoichiometric model of *iJR*904 *E. coli* to which the TPC pathway was added. Within the *E. coli i*JR904 stoichiometric model (3), the following three reactions were added:

TES: C_{12} -ACP + H₂O = C_{12} fatty acid + ACP

CAR: C_{12} fatty acid + NADPH + H⁺ + ATP = C_{12} aldehyde + NADP⁺ + AMP + PPi

AHR: C_{12} aldehyde + NADPH $\rightarrow C_{12}$ alcohol + NADP⁺

The *iJR904* (3) and *iAF1260* (4) models do not include any pathways for C_{14} -ACP and C_{16} -ACP formation. The calculation of the potential conversion rate for all fatty alcohols is therefore estimated according to C_{12} fatty alcohol synthesis even though the C_{14} and C_{16} alcohols are produced with and without mono-unsaturation.

After loading the *iJR904* GlcMM model into COBRA Toolbox 1.3.3 (5), the three reactions are added accordingly:

modelTes = addReaction(model,'TesA',{'ddcaACP[c]','h2o[c]','dodecanoate[c]','ACP[c]'},[-1 -1 1 1], false);

 modelCAR $=$ addReaction(modelTes,'CAR',{'dodecanoate[c]','nadph[c]','h[c]','atp[c]','dodecanal[c]','nadp[c]','amp[c]','ppi[c]'},[-1 -1 -1 -1 1 1 1 1], false);

 \Box modelAHR $=$

addReaction(modelCAR,'AHR',{'dodecanal[c]','nadph[c]','dodecanol[c]','nadp[c]'},[-1 -1 1 1], false);

Then, an exchange reaction is also added for C_{12} alcohol:

modelTPC = addReaction(modelAHR,'dodecanolEX',{'dodecanol[c]'},[-1], true);

In order to obtain the maximum potential conversion rate, we changed the objective function from biomass to 'dodecanolEX', which represents the rate of accumulation of C_{12} fatty alcohol given that C_{12} fatty alcohol does not excrete from wild-type *E. coli* cultivated in minimal media, and optimize, as follows:

```
modelTPCobjC12 = changeObjective(modelTPC, 'dodecanoIEX');solution = optimizeCbModel(modelTPCobjC12,'max',false,false)
```
A solution of f: 1.7266 was obtained using glpk solver.

Since the maximum glucose uptake rate is 6 mmol per gram dry weight per h, the maximum potential molar conversion efficiency is:

 $1.7266/6 = 0.288$ mol C₁₂ fatty alcohol per mole glucose. Under such conditions, there would be no biomass formation as the robustness analysis shows above.

If the above analysis is carried out also for the cyanobacterial pathway,

AAR: C_{12} -ACP + NADPH \rightarrow C_{12} aldehyde + NADP + ACP

with the following reactions added to COBRA:

modelSchirmer = addReaction(model,'ACR',{'ddcaACP[c]','nadph[c]','dodecanal[c]','ACP[c]','nadp[c]'},[-1 -1 1 1 1], false);

 modelAHR2 $=$ addReaction(modelSchirmer,'AHR',{'dodecanal[c]','nadph[c]','dodecanol[c]','nadp[c]'},[-1 -1 1 1], false);

modelTPC2 = addReaction(modelAHR2,'dodecanolEX',{'dodecanol[c]'},[-1], true);

robustnessAnalysis(modelTPC2, 'dodecanolEX');

we obtain an f-value of 1.7600, 1.9% greater potential yield than the CAR pathway, without any biomass formation.

In order to provide insight into the relative distribution of fluxes in the optimized models, the fluxes for key enzymes in central carbon metabolism and the major excretion products were extracted from the optimized solution. The values are plotted on the next page, with enzyme names as given for the *i*JR904 *E. coli* genome-scale model (3). The main difference between the two pathways is the additional ATP requirement of the CAR-dependent pathway. Notably, the simulation is unrealistic as there is no biomass-formation. Still, the below bar graph suggests that the task of reducing $NADP⁺$ is shifted slightly away from the proton gradient dependent transhydrogenase PntAB (THD2) towards the pentose phosphate pathway (as shown by the ratio of flux through PGI (phosphoglucoseisomerase) vs. G6PDH2r (glucose-6-phosphate dehydrogenase, Zwf)). As enhanced flux through the oxidative pentose phosphate pathway results in increased loss of carbon, in the form of $CO₂$, there is less fatty alcohol produced.

All estimates of Gibbs free energy changes under standard conditions were obtained using the eQuilibrator online platform (6).

Fig. S1 SDS-PAGE of purified preparations of CAR_{his}, Ahr_{his} and ADC_{his}. Proteins were separated on 12% (w/v) acrylamide gel and stained with Coomassie Blue (7). Lane 1, molecular-mass markers (in kDa); lane 2, crude fraction of BL21(DE3); lane 3, crude fraction of PC; lane 4, crude fraction of Ahr_{his}; lane 5, crude fraction of ADC_{his}; lane 6, his-tag eluate fraction from BL21(DE3); lane 7, purified CAR_{his}; lane 8, purified crude Ahr_{his}; and lane 9, purified ADC_{his}. Purified recombinant proteins are marked with an arrow and their theoretical molecular weights indicated in brackets.

Fig. S2 NADPH specificity of CARhis. Reactions were carried out in 50 mM Tris-HCl (pH 7.5, 21^oC) containing purified CAR_{his} (0.25 µg/ml), 2 mM benzoic acid, 10 mM MgCl₂ and 1mM ATP with either 1 mM NADPH or 1 mM NADH as the source of reductant. Absorbance was monitored at 340 nm at 30 s intervals over a 10 min period.

Fig. S3 Kinetic analysis of CARhis in the presence of benzoic acid. Reactions were carried out in the presence of varying concentrations of **(***A***)** benzoic acid (0.2-0.5 mM), **(***B***)** NADPH (250-1000 μ M) and (C) ATP (0.05-3 mM). The K_m and V_{max} were determined from nonlinear regression plots, using the enzyme kinetics module of Sigmaplot (Systat Software, San Jose, CA). Results are means \pm SEM of triplicate experiments.

Fig. S4 Effect of pH on CARhis activity. Reactions were carried out in 50 mM Tris-HCl buffer (21°C) containing purified CARhis, 2 mM benzoic acid, 1 mM ATP and 1 mM NADPH within the pH range of 7 to 8.5. The initial rate of CAR activity was determined from the rate of NADPH oxidation at 340 nm. Results are means \pm SEM of duplicate experiments.

Fig. S5 Thermostability of CAR_{his}. In the absence of reducing agents, CAR_{his} was incubated without shaking at three different temperatures; 25° C, 30° C and 37° C; for up to 96 h in 50 mM Tris-HCl buffer (pH 7.5 , 21° C). Samples were taken at specific time intervals and the initial rate of CAR activity determined in the presence of 2 mM benzoic acid, 1 mM ATP, 10 $mM MgCl₂$ and 1 mM NADPH. Results are means \pm SEM. of duplicate experiments.

Retention time (min)

Retention time (min)

H

I

G

Fig. S6 GC-MS confirmation of aldehyde synthesis by CARhis. Reactions were carried out in the presence of CAR_{his} (100-300 μ g/ml), 1 mM NADPH, 1 mM ATP, 10 mM MgCl₂ and 0.5mM C_4 - C_{18} fatty acids $(A - K)$. Reactions were incubated at 30 °C for up to 4 h or in the case of the longer chain substrates ($\geq C_{16}$) up to 12 h. Reactions were mixed with chloroform and the organic phase analyzed by GC-MS (see Methods). For clarity, only the peak representing the aldehyde is shown. Dashed lines represent chromatograms without the addition of CARhis. Aldehyde formation was confirmed by comparing the mass spectrum of the analyte (in red) against reference standards (in blue) from the NIST spectral database. The match factors and reverse match factors quantitatively describe the spectral match between a sample spectrum and the library spectrum, and are indicated in brackets. The reverse match factor is obtained by excluding peaks that are present in the sample spectrum but not the library spectrum. Both values are derived from a modified cosine of the angle between the spectra, otherwise known as the normalized dot product. A value of 900 or greater signifies an excellent match; 800–900, a good match; and 700–800, a fair match.

C12 fatty acid (µM)

0 20 40 60 80 100

0.00

Fig. S7. Kinetic analysis of CAR_{his} based on C_4 to C_{12} saturated fatty acid substrates. Reactions were carried out in 50mM Tris-HCl buffer containing CAR (0.5 to 4 μ g/ml), 1 mM NADPH and fatty acid substrates ranging from 0 to 20 mM. For the C_4 substrate, the buffering capacity was increased to 100mM. Reactions were monitored at 340nm at 30 s intervals. The apparent K_m and V_{max} values were determined from non-linear regression plots using the enzyme kinetics module of SigmaPlot (Systat Software, San Jose, CA). The saturated fatty acids evaluated were: **(A)** C_4 fatty acid (0.5-20 mM), **(B)** C_6 fatty acid (0.2-5) mM), **(C)** C₈ fatty acid (0.1-2.5 mM), **(D)** C₁₀ fatty acid (10-750 µM) and **(E)** C₁₂ fatty acid (5-100 μ M). Results are means \pm SEM of triplicate experiments.

B

Strain

A

Fig. S8. Activity and expression levels of AHR in crude soluble fractions. **(***A***)** The crude fractions were analyzed by SDS-PAGE for expression levels of AHR. Lane 1, molecularmass markers (in kDa); lane 2, crude fraction of TesA; lane 3, crude fraction of TPC; lane 4, crude fraction of TPC-slr1192; lane 5, crude fraction of TPC-slr1192his; lane 6, crude fraction of TPC-Ahr; lane 7, crude fraction of TPC-Ahrhis; and lane 8, molecular-mass markers (in kDa). Proteins bands representing the CAR and AHR enzymes are indicated by red arrows. **(***B***)** AHR activity was determined in crude fractions of all fatty alcohol-producing strains using C_{12} aldehyde as the substrate. The TesA strain was included as a negative control. Results are means \pm SEM of duplicate experiments.

Fig. S9. Amino acid sequence alignment of *E. coli* Ahr and *Synechocystis* sp. PCC 6803 slr1192. Sequences were aligned using ClustalW 1.8 [\(http://www.ebi.ac.uk/Tools/msa/\)](http://www.ebi.ac.uk/Tools/msa/) (8) and formatted using Boxshade 3.21 [\(http://www.ch.embnet.org/software/BOX_form.html\)](http://www.ch.embnet.org/software/BOX_form.html). Dark-shaded boxes depict identical amino acids, while lighter-shaded boxes signify amino acid residues with similar properties. Conserved amino acid residues for zinc binding are highlighted in red.

Fig. S10. NADPH specificity of Ahrhis. Reactions were carried out in 50 mM Tris-HCl (pH 7.5, 21°C) buffer containing Ahr_{his} (1 µg /ml), 0.5 mM NADPH or NADH and 0.5 mM C₆ aldehyde. Absorbance was monitored at 340 nm at 15 s intervals over a 15 min period.

Fig. S11. Substrate specificity of Ahr_{his}. Reactions were carried out in 50mM Tris-HCl buffer containing Ahr_{his} (1 µg or 10 µg /ml), 0.5 mM NAD(P)H or NAD(P)⁺ with either **(***A*) 0.5mM aldehydes (C_4 - C_{12}) or **(***B*) 0.5 mM alcohols (C_4 - C_{12}). Initial rates for both reactions were determined from the rate of NAD(P)H oxidation or NAD(P)⁺ reduction. Results are means \pm SEM of duplicate experiments.

B

A C₄ alcohol (899, 916)

Retention time (min)

Retention time (min)

Fig. S12. GC-MS verification of alcohol synthesis by Ahr_{his}. Reactions were carried out in 50mM Tris-HCl buffer containing Ahr_{his} (10-30 μ g/ml), 1 mM NADPH and 0.5mM C₄-C₁₂ aldehydes $(A-E)$. Reactions were incubated at 30 °C for up to 2 h and terminated by addition of chloroform. The organic phase analyzed by GC-MS (see Methods). For clarity, only the peak representing the alcohol is shown. Dashed lines represent chromatograms without the addition of the enzyme. Alcohol formation was confirmed by comparing the mass spectrum of the analyte (in red) against reference standards (in blue) from the NIST spectral database. The match factors and reverse match factors are indicated in brackets. Refer to **Fig. S6** for explanation of (reverse) match factors.

Fig. S13. GC-MS confirmation of enzyme-mediated formation of alcohols and alkanes. Alcohol $(A-F)$ and alkane formation $(G-F)$ were verified by comparing the mass spectrum of the analyte (in red) against the reference standards (in blue) from the NIST spectral database. The match factors and reverse match factors are indicated in brackets. Refer to **Fig. S6** for explanation of (reverse) match factors.

Retention time (min)

Fig. S14. Conversion of TAGs to fatty acids and aldehydes. Reactions were performed in 50 mM Tris-HCl (pH 7.5, 21° C) containing 2 mM NADPH, 2 mM ATP, 10 mM MgCl₂ and 0.5 mM C₈-TAG (glyceryl trioctanoate) in the presence of (A) lipase and CAR enzymes (B) lipase only (0.1 mg/ml) and **(C)** CAR only (0.3 mg/ml). A further reaction using **(D)** C₁₂-TAG (glyceryl trilaurate) as the substrate was also carried out.

Fig. S15. *In vivo* synthesis of fatty alcohols using palm oil as substrate. Harvested cells from a preinduced culture of PC-Ahrhis, were resuspended in 50 mM potassium phosphate buffered medium, supplemented with glucose and lipase (1 mg/ml) and incubated at 30 °C with shaking at 150 rpm for up to 5 h in the presence of palm oil. Asterisk denotes fatty acid.

Table S1: Key features of the fatty acyl-ACP/TAG to fatty alcohol-producing systems. Enzyme abbreviations according to the legend of Figure 1.

*Calculation of G values under standard conditions performed with eQuilibrator (4)

**Green box=detectable levels of fatty alcohols, grey box = undetected fatty alcohols

Table S2: Substrate specificity and turnovers of the metabolic enzymes required for fatty-acyl ACP to fatty alcohol conversion

*Green box=observed activity, blue box= undetected activity, grey box = untested substrates

**Partially purified enzyme.

***Purified ACR2 from *Marinobacter aquaeolei* VT8

Table S3. List of the primers and templates used for gene(s) amplification.

^aRestriction enzyme sites are underlined and italicized

Abbreviations used: FP, forward primer; RP, reverse primer

Table S4. Summary of the vector backbone and gene(s) inserts used for plasmid construction.

^a Restriction enzyme cuts are specified in brackets.

Table S5. List of the plasmids used in the study including their properties.

Table S6. List of *E. coli* **strains engineered in the study.**

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