## **Supporting Information**

## Water Networks in Fast Proton Transfer During Catalysis by Human Carbonic Anhydrase $\mathbf{II}^{\dagger}$

Rose Mikulski,<sup>a</sup> Dayne West, <sup>b</sup> Katherine H. Sippel, <sup>b</sup> Balendu Sankara Avvaru,<sup>b</sup>

Mayank Aggarwal,<sup>b</sup> Chingkuang Tu,<sup>a</sup> Robert McKenna,<sup>\*,b</sup> and David N. Silverman<sup>\*,a,b</sup>

<sup>a</sup> Department of Pharmacology, <sup>b</sup> Department of Biochemistry and Molecular Biology, University of Florida, Gainesville, FL 32610, USA

	$(k_{cat})_{hydration}$	$^{D}(k_{cat})_{hydration}$	$(k_{cat}/K_m)_{hydration}$	$(k_{cat})_{dehyd}$	$(k_{cat}/K_m)_{dehyd}$
	(µs <sup>-1</sup> )		$(\mu M^{-1}s^{-1})$	(µs <sup>-1</sup> )	$(\mu M^{-1}s^{-1})$
wild-type	$0.80 \pm 0.03$	3.8 <sup>b</sup>	59 ± 2	$0.24\pm0.02$	$6.7 \pm 0.2$
HCA II		$3.2\pm0.4^{\circ}$			
Y7F-N67Q	$0.67 \pm 0.04$	$3.1 \pm 0.3$	67 ± 4	$0.19 \pm 0.01$	$8.8 \pm 0.3$
HCA II					

**Table S1:** Steady-state constants for the hydration of  $CO_2$  and dehydration of bicarbonate catalyzed by wild-type and Y7F-N67Q HCA II<sup>a</sup>

<sup>a</sup> Measurements of initial velocity in the hydration of  $CO_2$  and dehydration of bicarbonate were carried out on a stopped-flow spectrophotometer (Applied Photophysics SX18.MV) using the initial 5% to 10% of the progress curve following the method of Khalifah (*Journal of Biological Chemistry 246*, 2561-2573 (1971)). The hydration experiments were performed at pH 8.4 using 0.1 to 0.2  $\mu$ M enzyme, 25 mM of CHES buffer, and 70  $\mu$ M thymol blue at 25 °C. The dehydration experiments were performed at pH 6.0 using 0.6 to 1.1  $\mu$ M enzyme, 25 mM MES buffer, and 100  $\mu$ M chlorophenol red at 10 °C. In all experiments, the ionic strength was maintained at 0.2 M using sodium sulfate. Saturated solutions of CO<sub>2</sub> were prepared by bubbling CO<sub>2</sub> gas into water at 25.0 °C (saturating concentration 33.8 mM) and diluting using two coupled, air-tight syringes.

<sup>b</sup> From Steiner et al., *Eur. J. Biochem.* 59, 253-259 (1975)

<sup>c</sup> From Venkatasubban and Silverman, *Biochemistry*, 19, 4984-4989 (1980). This measurement was made using bovine CA II.

## Scheme S1

A double mutant cycle with the values of  $k_B (\mu s^{-1})$  (shown under the variant designation) for proton transfer obtained by <sup>18</sup>O exchange and the corresponding values of changes in estimated energy barriers (shown adjacent to arrows) computed as described by Mildvan et al. (*Archives of Biochemistry and Biophysics 294*, 327-340 (1992)).





**Figure S1.** Active sites of variants of HCA II at pH 8.0. (**A**) N67Q HCA II, and (**B**) Y7F-N67Q HCA II. 2Fo-Fc electron density maps are contoured at 1.2σ. Figure was made using PyMOL (DeLano Scientific).