

Supplementary Data for

THE HEME BINDING PROPERTIES OF GLYCERALDEHYDE-3-PHOSPHATE DEHYDROGENASE

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Fig. S1. Gel filtration of the rGAPDH-heme complex. Panel A: Chromatogram of the GAPDH-heme complex. Fractions were collected between 8 and 65 min and analyzed by UV-vis spectroscopy. Panel B: UV-visible analysis of the fractions. The ratio A_{280}/A_{415} of the complex prior to injection was 3.5. The ratio A_{280}/A_{415} of the eluted fractions containing the highest concentration of heme ranged from 5 to 8.

Fig. S2. In-gel heme staining of apo-and heme-bound GAPDH using the o-dianisidine/ H_2O_2 peroxidase assay. Panel A shows the heme-staining bands corresponding to rGAPDH-heme (5:1) and apo-GAPDH (negative control). Panel B shows different doses of GAPDH-heme, and iNOxSoxy as a positive control for specific heme staining.

Fig. S3. Far-UV CD spectra of apo- and heme-bound GAPDH. Addition of heme to apo-GAPDH produced mild changes in the far-UV CD spectrum, suggesting that heme binding is accompanied by modest conformational changes.

Fig. S4. Spectrophotometric titrations of rabbit GAPDH with increasing doses of hemin. Panel A. Titration of 10 μM GAPDH with hemin (0-10 μM). Panel B: UV-visible spectra of free hemin in buffer.

Fig. S5. Transfer of ferrous heme from GAPDH to apomyoglobin at 25 °C. Heme transfer to apomyoglobin is accompanied by a decrease in absorption at 425 nm (GAPDH-heme) and a concomitant increase at 434 nm (reduced myoglobin).The inset shows the initial and final spectra.

Fig. S6. Reaction of native, anaerobic GAPDH with ferrous heme (Panels A and B), and reaction of GAPDH pre-treated with dithionite and ferrous heme (Panel C). The reaction of GAPDH with ferrous heme (5:1) was slow ($k_{\text{obs}} = 0.0091 \text{ s}^{-1}$ at 10 °C) and incomplete; only 19% of the total available ferrous heme formed a complex with GAPDH after 10 min. No measurable spectral changes were detected for the reaction between dithionite-treated GAPDH and ferrous heme.

Fig. S7. Kinetics of CO binding to two major subpopulations of heme-bound to rGAPDH (2:1, GAPDH:heme, 10 °C). Formation of a Fe(II)-CO complex with a Soret absorption maximum at 410 nm occurs ~4-fold faster than the formation of the 6-coordinate Fe(II)-CO-GAPDH with a characteristic band at 420 nm. The nature of the species absorbing at 410 nm is currently unknown.

Fig. S8. UV-visible spectra of purified human GAPDH (Panel A) or *S. suis* (Panel B) GAPDHs from cultures grown in the presence or absence of the heme synthesis precursor δ-amino levulinic acid.

Fig. S9. Amino acid sequence alignment of *H. sapiens* (Accession P04406.3), *O. cuniculus* (Accession P46406.3) and *S. suis* (Accession AAN86058.1) GAPDHs. The alignment was generated with ClustalW 2.1 software (<http://www.ebi.ac.uk/Tools/msa/clustalw2/>)

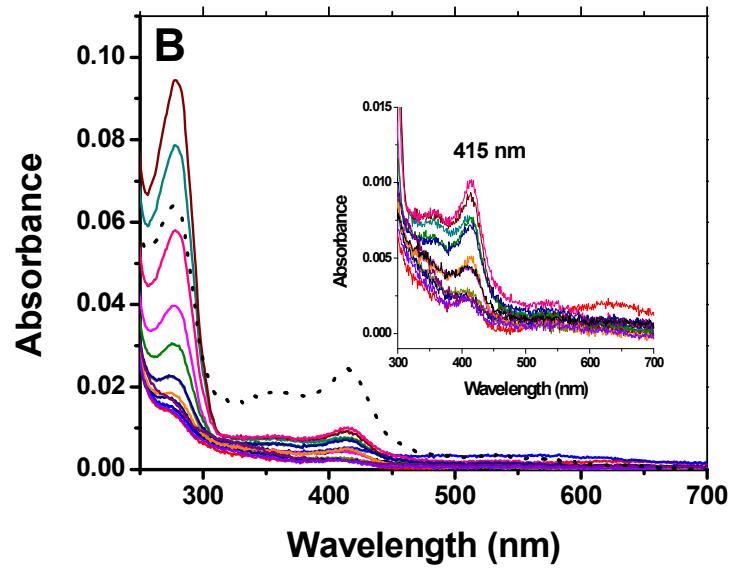
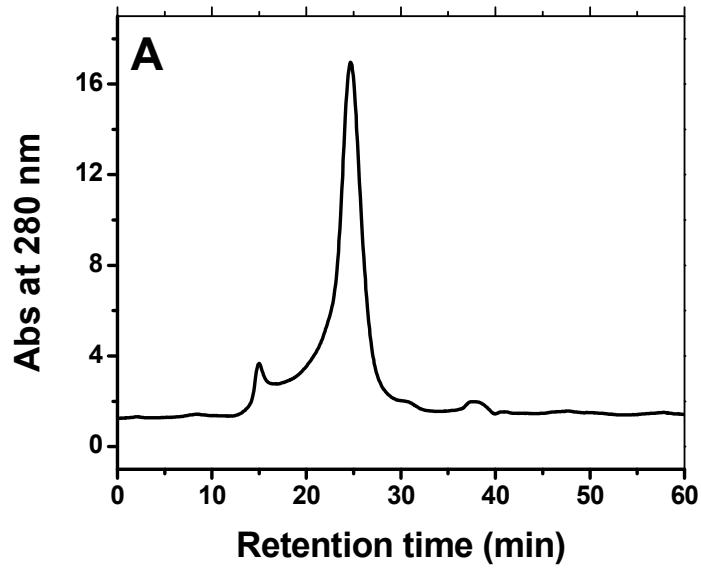


Figure S1

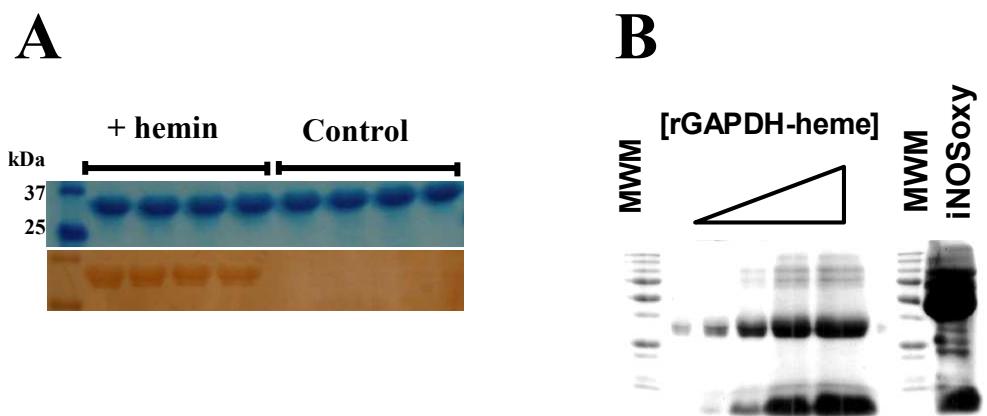


Figure S2

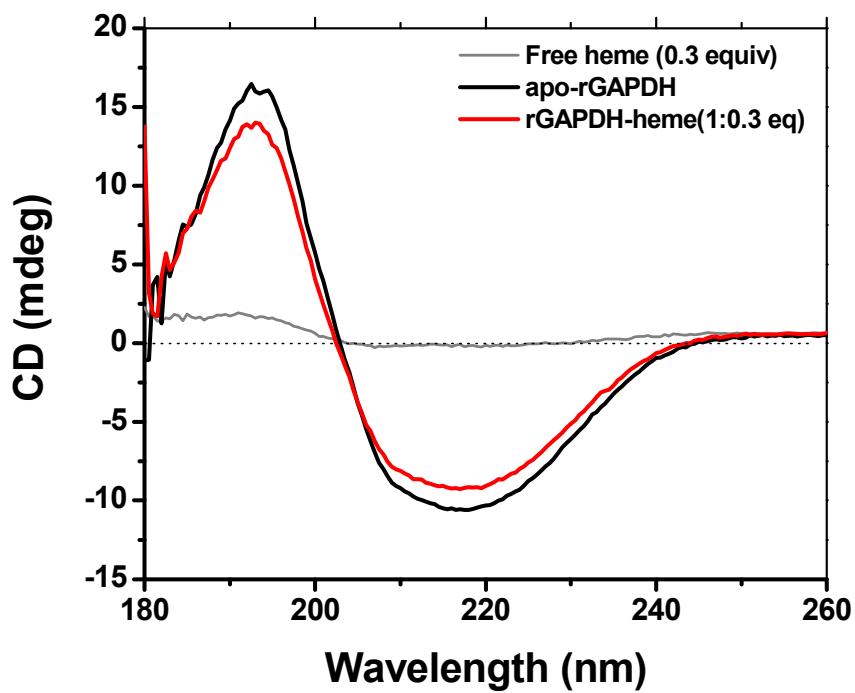


Figure S3

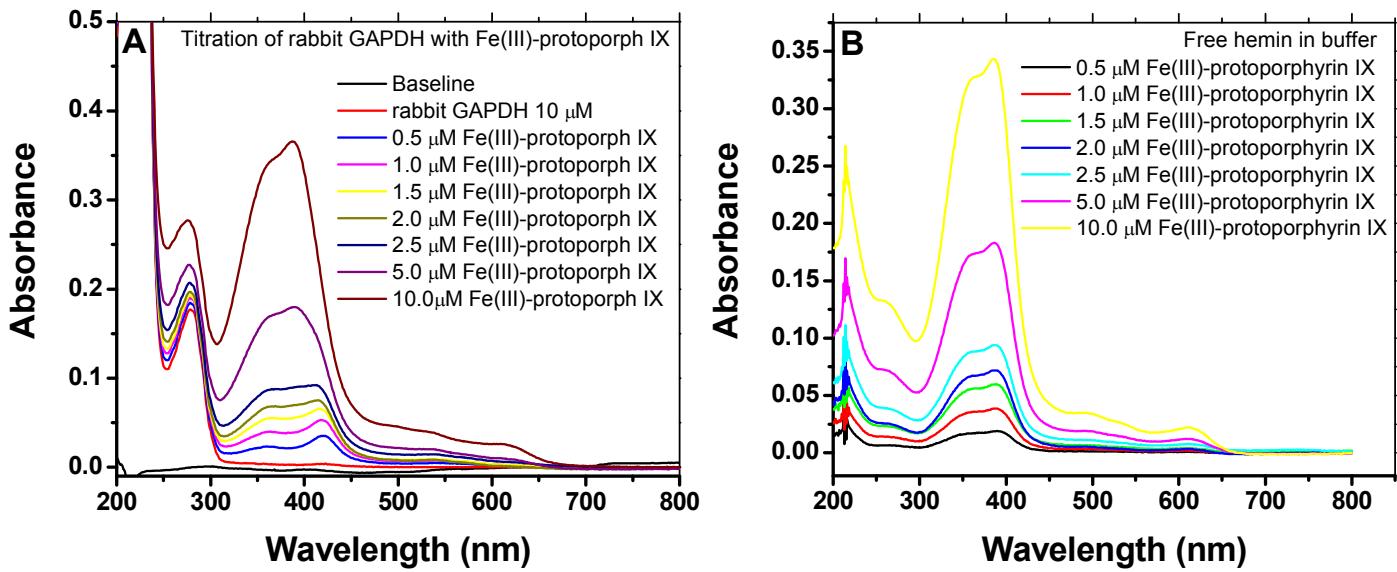


Figure S4

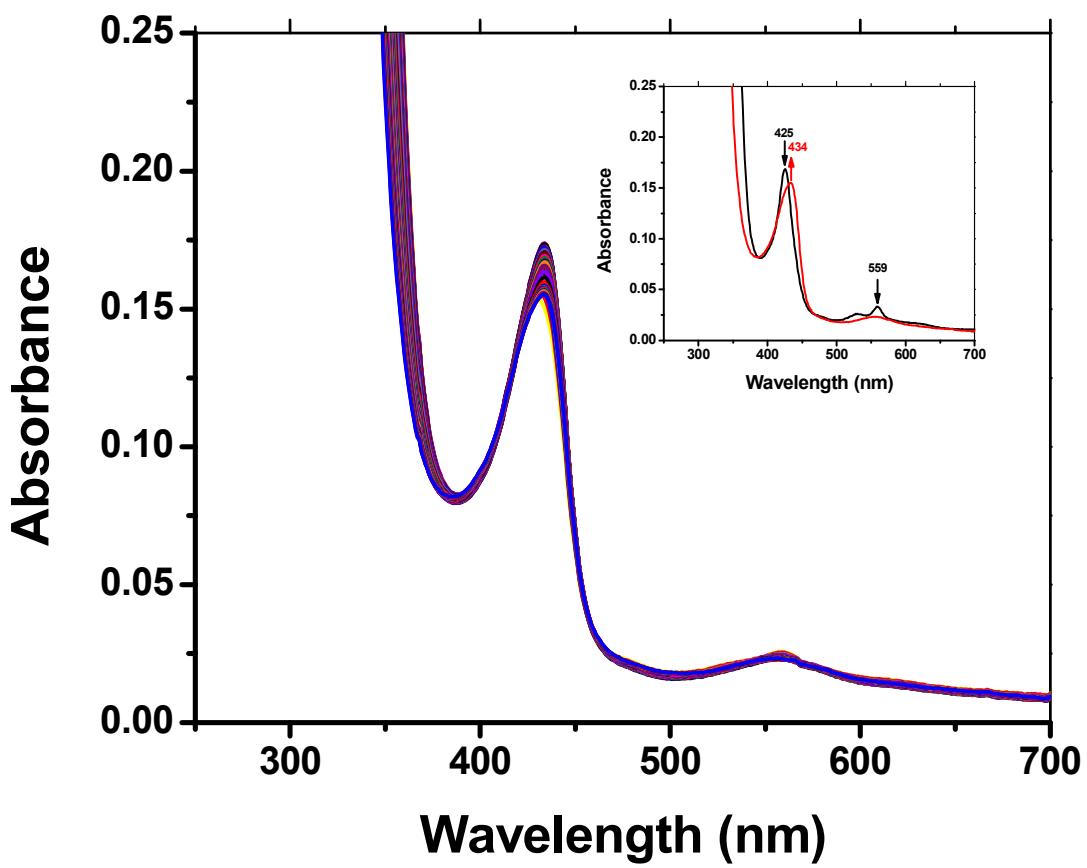


Figure S5

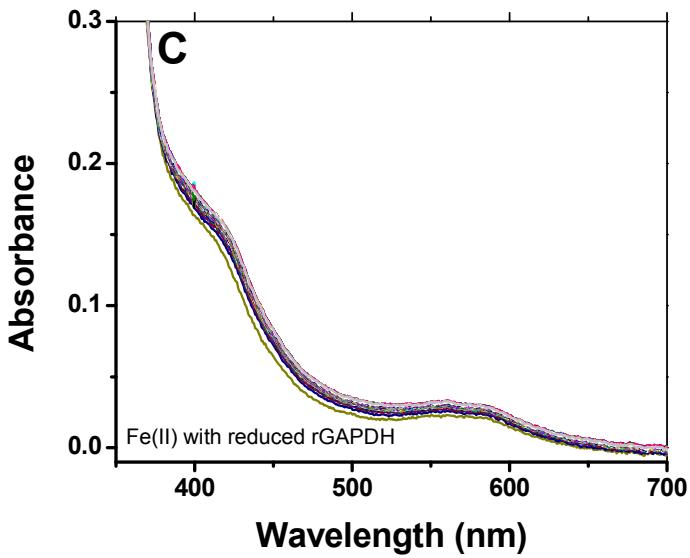
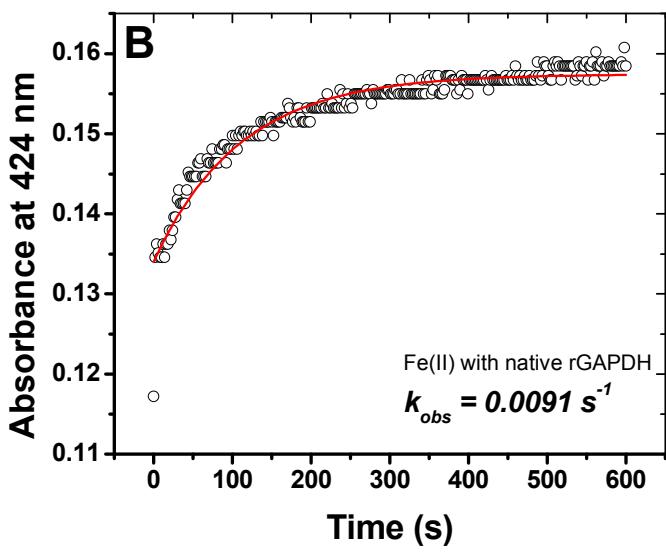
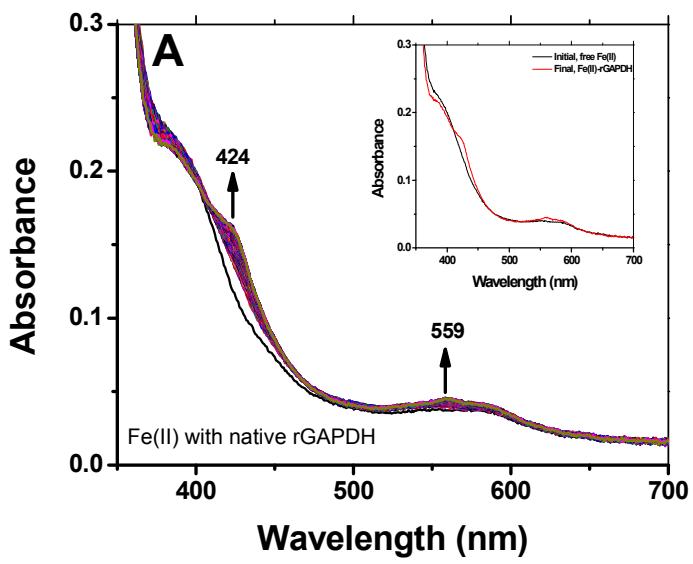


Figure S6

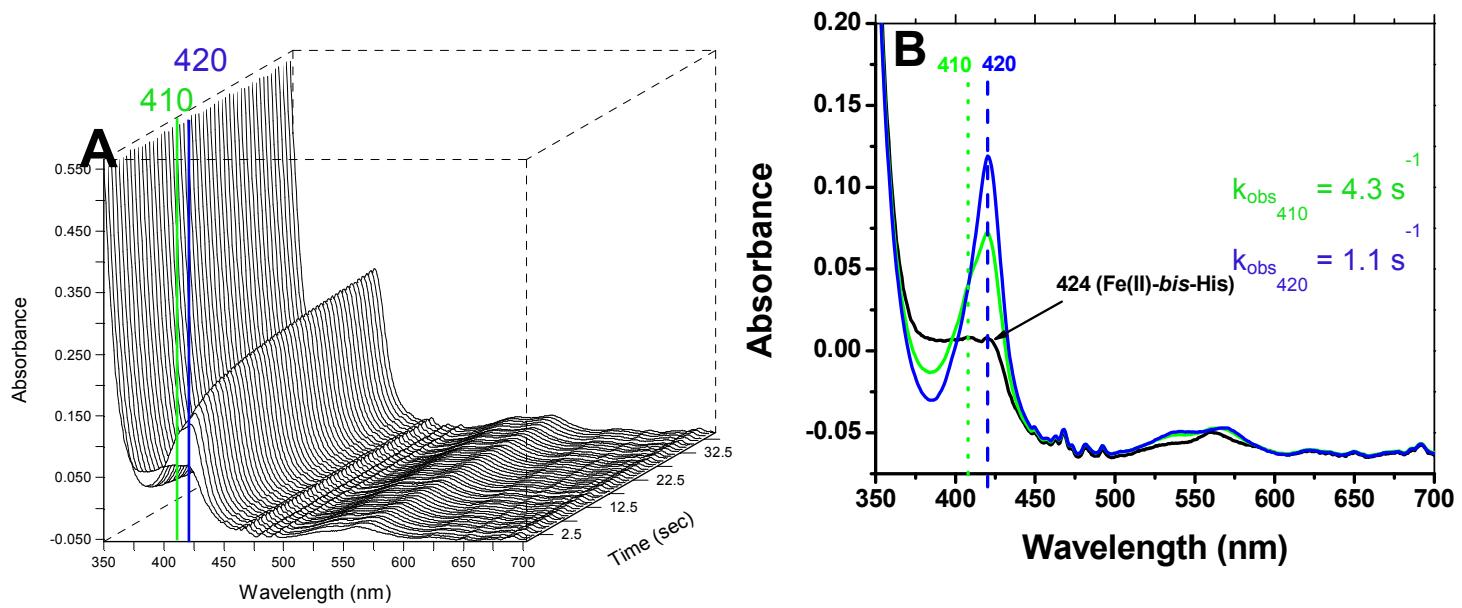


Figure S7

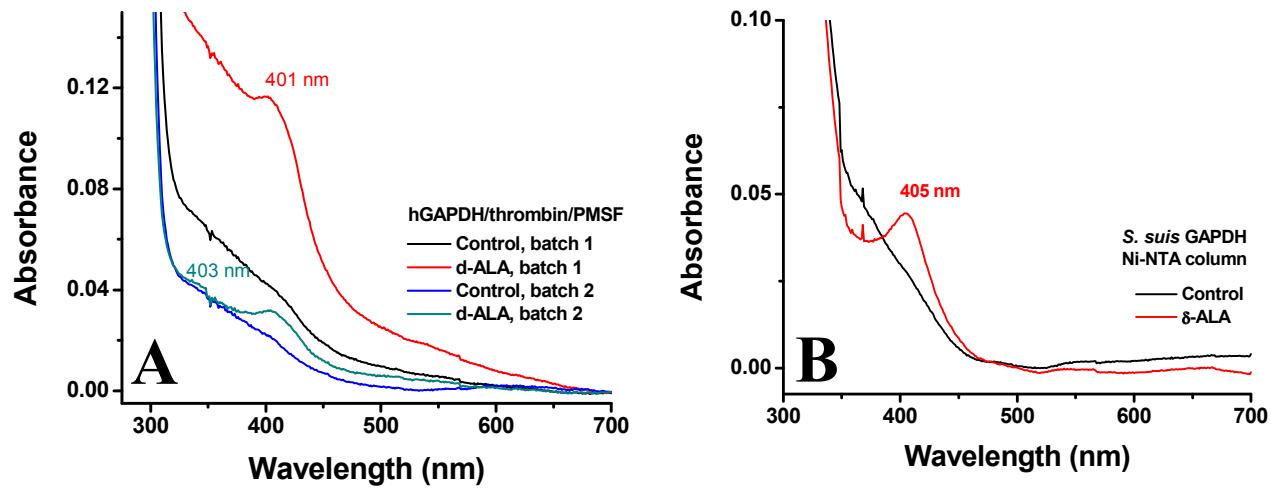


Figure S8

CLUSTAL 2.1 multiple sequence alignment for *H. sapiens*, *O. cuniculus* and *S. suis* GAPDHs (in this order).

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--AITIFQERDPANIKWGDAGAEYVVESTGVFTTMEKAGAHLKG-GAKRV 47
--FVKVSAEREPEGNIDWATDGDIVLEATGFFASKEKAEQHIHANGAKKV 48
.: * : * . : * . * . * : * : * : * : * : * : * : * : *

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----ESFGYTEDQLVSSDIVGISFGSLFDATQTKVIEVDGEQLVKVVSWY 244
: * : * : * : * : . : . * : * : * : * : * : * : * : * : *

MGKVGVNGFGRIGRLVTRAASFNSGKVDI
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DNEFGYSNRVVDLMVHMASKE- 263
DNEMSYTAQLVRTLEYFAKIAK 266
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Figure S9