

## **Supporting Information**



**Figure S1.** Quantification of multimers formed by GPC as function of pH. The monomer fraction was defined as the value of [1 - multimer fraction]. (a) Fraction of multimer formed after 24 h. (b) Time-dependence of multimer formation at a 1:1 Fe<sup>3+</sup>:catechol ratio (legend same as in (a)).

1



**Figure S2.** UV-visible spectra of solutions of  $\text{Fe}^{3+}$  and mPEG-cat. Spectra of  $\text{Fe}^{3+}$  and mPEG-cat in buffers of various pH (top). Spectra of  $\text{Fe}^{3+}$  and mPEG-cat incubated at pH 3 for 2 h before adjustment to pH 7 or 9 (bottom).



**Figure S3.** Rheometry of gels composed of 8cPEGa in unbuffered water. Gel formation with  $Fe^{3+}$ :catechol of 2:3 (red) was studied as a function of time (top), frequency (middle) and strain (bottom); as shown in the time sweep,  $Fe^{3+}$ :catechol of 1:3 (black) does not produce a gel. *G'*: solid line with circles; *G''*: dashed line with triangles (bottom).



**Figure S4.** Time and strain sweep experiments of gels formed at pH 3 composed of 8cPEGa with Fe<sup>3+</sup>:catechol of 1:3 (black), 2:3 (red), and 3:3 (blue). Gel formation was studied as a function of time (top) and, for gels with Fe<sup>3+</sup>:catechol of 2:3 and 3:3, of strain (bottom) for gels with Fe<sup>3+</sup>:catechol of 2:3 and 3:3. G': solid line with circles; G'': dashed line with triangles.



Figure S5. Digital images depicting the stability of hydrogels in water and aqueous EDTA.

Gels were formed at the indicated pH values and Fe<sup>3+</sup>:catechol stoichiometries

5



**Figure S6.** pH-dependent relaxation of two-step hydrogels containing covalent and coordination bonds. These samples were strained to 5 % (top), 10 % (middle), or 20 % (bottom) compression, and the stress was monitored for 100 s.



**Table S1.** Measured pH of reaction between  $Fe^{3+}$  and mPEG-cat

Fe <sup>3+</sup> :mPEG-cat	pH 3	pH 5	pH 7	pH 9
1:3	3.0	5.1	6.6	8.6
2:3	2.7	4.8	6.3	8.2
3:3	2.5	4.7	6.1	8.0

**Table S2.** Gel times associated with one-step cross-linking of 8cPEGa by  $Fe^{3+}$ 

Fe <sup>3+</sup> :catechol	Water	рН 3	pH 5	pH 7	pH 9
1:3	NA	NA	NA	immediate	immediate
2:3	50 s	85 s	NA	immediate	immediate
3:3	20 s	45 s	NA	immediate	immediate

Gel formation assessed by vial inversion method. NA = no gel formed within 60 minutes.

<b>Table 55.</b> Stability of Fe <sup></sup> -8CPECia gets incubated in water of EDTA solution after 24 f	Table S3.	Stability	of Fe <sup>3+</sup> -8cPEGa	a gels incubated	in water or EI	OTA solution	after 24 h
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Solvent	Fe <sup>3+</sup> :catechol	рН 3	рН 5	pH 7	рН 9
	1:3	NA*	Unstable	Mostly Stable	Unstable
Water	2:3	Stable	Mostly Unstable	Mostly Stable	Unstable
	3:3	Stable	Mostly Stable	Stable	Mostly Unstable
100 mM EDTA	1:3	NA	Unstable	Mostly Unstable	Unstable
in Water	2:3	Stable	Unstable	Mostly Unstable	Unstable
in water	3:3	Stable	Unstable	Mostly Unstable	Unstable

1.5 h allowed for gel formation. \*NA: gel does not form within 20 minutes



Table S4. Physical Characterization of Two-Step Hybrid Gels After Equilibration

		рН 3	рН 5	pH 7	pH 9
Color		yellow-brown	black	black	black
Swelling (%)		$159.8\pm3.8$	$108.7 \pm 6.7$ $98.3 \pm 1.3$		88.8 ± 8.3
Initial Modulus (kPa) <sup>a</sup>		8.7 ± 5.6	$18.8\pm4.6$	$47.8\pm6.9$	$39.5\pm2.8$
Initial Modulus (kPa) <sup>b</sup>	5 % strain	$15.8\pm4.7$	$25.8\pm3.6$	$24.4\pm3.8$	$55.4\pm9.4$
	10 % strain	$18.7\pm3.9$	$33.0\pm2.5$	$38.8\pm4.5$	$60.4\pm9.2$
	20 % strain	$21.2\pm4.0$	$37.5\pm2.2$	55.3 ± 3.9	$66.6\pm9.7$
Steady State Modulus (kPa) <sup>c</sup>	5 % strain	$15.4\pm4.4$	$18.7 \pm 2.1$	$12.2 \pm 2.7$	$22.7\pm3.6$
	10 % strain	$17.3\pm3.8$	$24.6\pm1.8$	$19.2 \pm 2.4$	$25.6\pm4.7$
	20 % strain	$19.2\pm3.9$	31.0 ± 1.3	$31.6\pm2.0$	$30.6\pm4.6$
Relaxation (%) <sup>d</sup>	5 % strain	<sup>f</sup>	$13.6 \pm 1.1^{\text{g}}$	$35.9 \pm 1.2^{\text{g}}$	$55.3 \pm 0.7$ <sup>g</sup>
	10 % strain	<sup>f</sup>	$13.3\pm0.9$	$35.5\pm1.6$	$56.1 \pm 1.4$
	20 % strain	<sup>f</sup>	$9.7\pm0.8$	$32.3\pm1.2$	$51.2\pm0.6$
Relaxation Time, τ (s) <sup>e</sup>	5 % strain	<sup>f</sup>	$218\pm56^{\rm g}$	$58.4 \pm 40.4$ <sup>g</sup>	$4.9\pm0.3^{\rm ~g}$
	10 % strain	<sup>f</sup>	$12 \pm 1$	$37.4\pm6.9$	$4.5\pm0.2$
	20 % strain	<sup>f</sup>	$25 \pm 16$	$18.4\pm0.4$	$4.9\pm0.2$
$\alpha^{e}$	5 % strain	<sup>f</sup>	$0.3 \pm 0.02^{\text{g}}$	$0.30 \pm 0.04^{\text{ g}}$	$0.62 \pm 0.03^{\text{g}}$
	10 % strain	<sup>f</sup>	$0.4 \pm 0.05$	$0.35 \pm 0.01$	$0.64 \pm 0.003$
	20 % strain	<sup>f</sup>	$0.38 \pm 0.02$	$0.39 \pm 0.02$	$0.67 \pm 0.004$

<sup>a</sup> initial modulus calculated from slope of linear portion of stress-strain curves (from compression loops) with cross-head speed of 10 mm/min; <sup>b</sup> initial modulus calculated from slope of linear portion of stress-strain curves (from relaxation tests) with cross-head speed of 40 mm/min; <sup>c</sup> steady-state modulus ( $A_I/\epsilon_{100}$ ) calculated from fit of relaxation tests; <sup>d</sup> percent stress relaxed after 100 s; <sup>e</sup> calculated from fit of relaxation tests by expanded exponential model; <sup>f</sup> relaxation data were not calculated for materials equilibrated to pH 3 because of the absence of coordination bonds; <sup>g</sup> larger error for calculations associated with 5 % strain were due to the significance of instrumental drift.