

## Supporting Information

# Molecular Mechanism of Monodisperse Colloidal Tin Doped Indium Oxide Nanocrystals by a Hot-injection Approach

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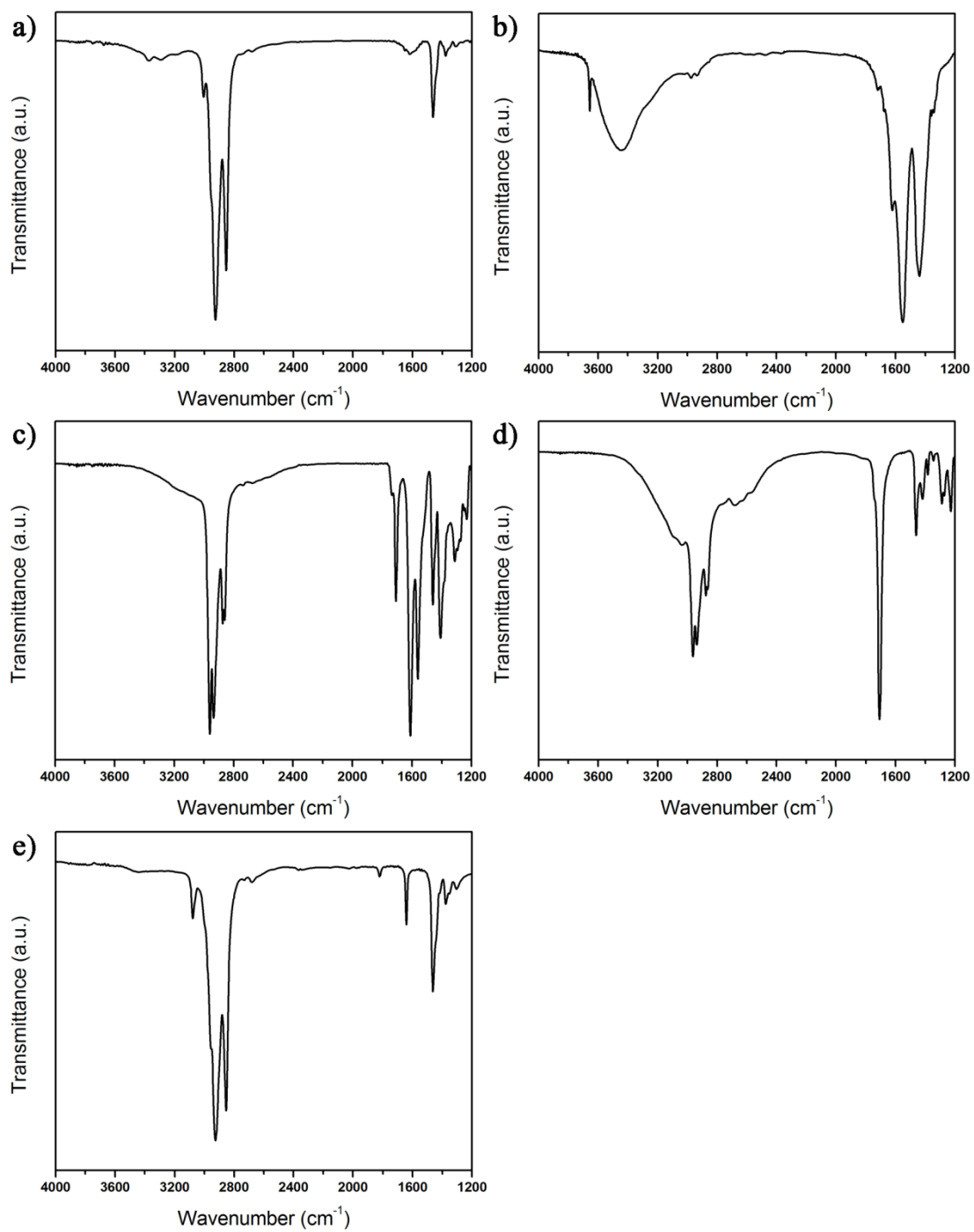
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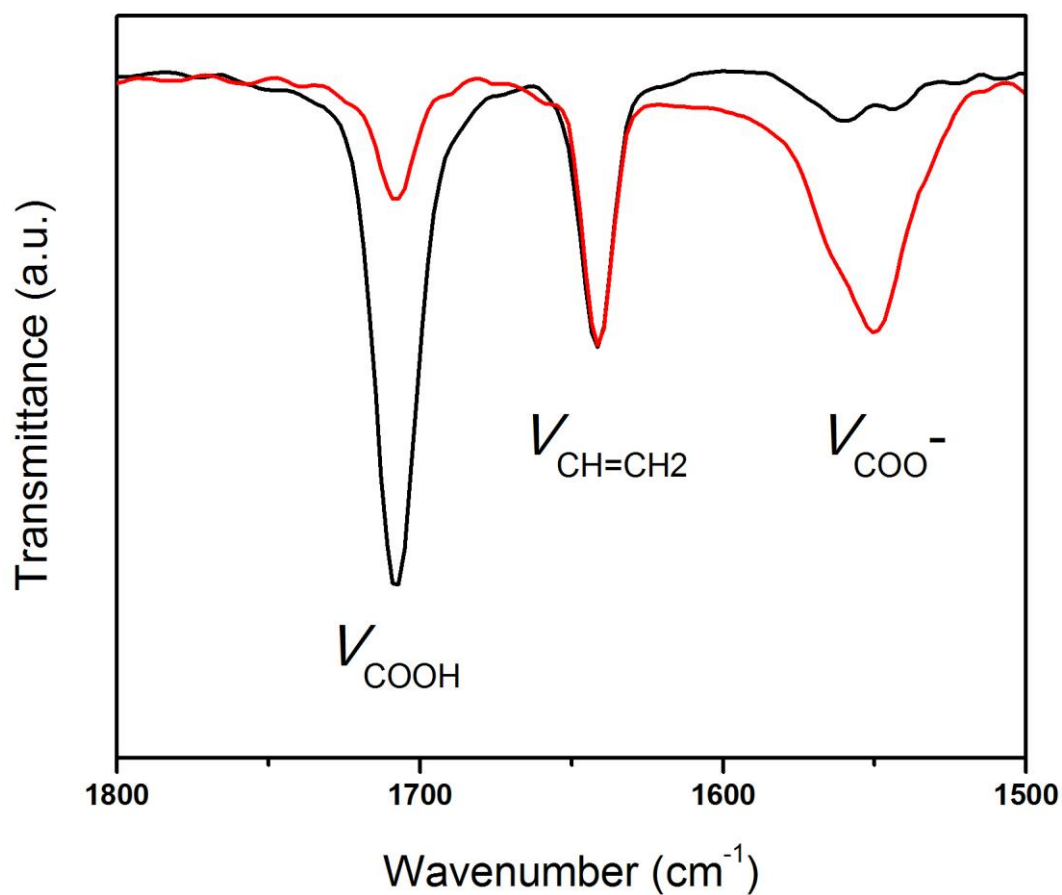
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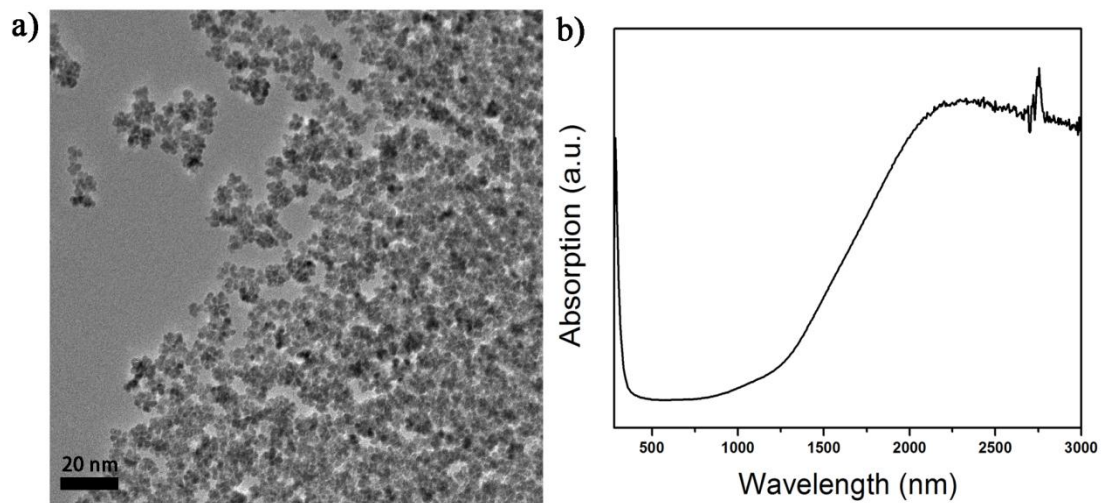
*& [yez@zju.edu.cn](mailto:yez@zju.edu.cn) (Prof. Zhizhen Ye)*



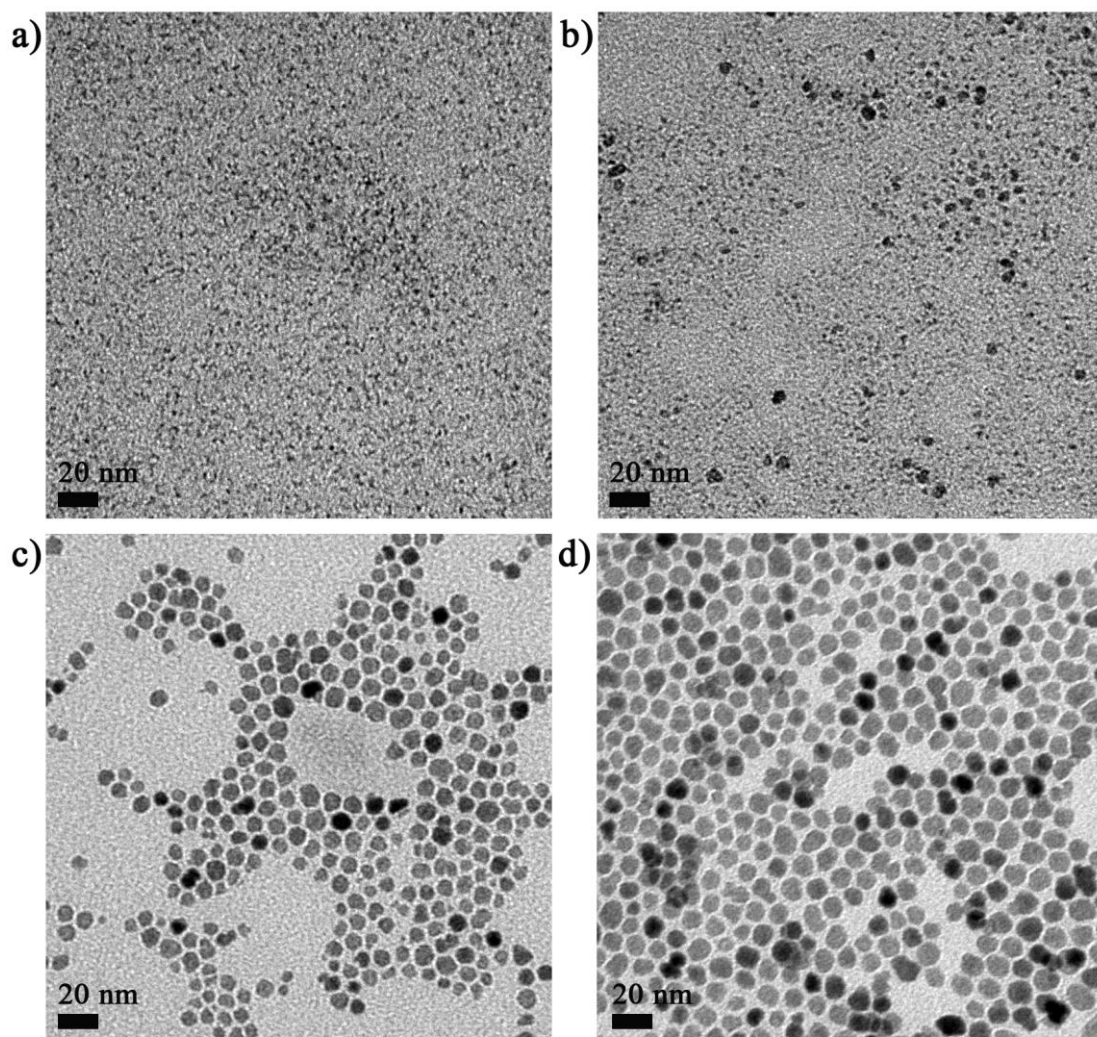
**Figure S1 FTIR spectra of a) oleylamine, b) indium acetate, c) tin(II) 2-ethylhexanate, d) 2-ethylhexanoic acid and e) ODE, respectively.**



**Figure S2 FTIR spectra of the mixture of indium acetate (1.2 mmol), 2-ethylhexanoic acid (3.6 mmol) and ODE (10 ml) before (25 °C) and after ligand exchange at 290 °C for one hour. The proportion of 2-ethylhexanoate to the acetate groups was determined by estimating the remaining 2-ethylhexanoic acid in solution derived from the calculations of the absorption peak area in the corresponding FTIR spectra.**

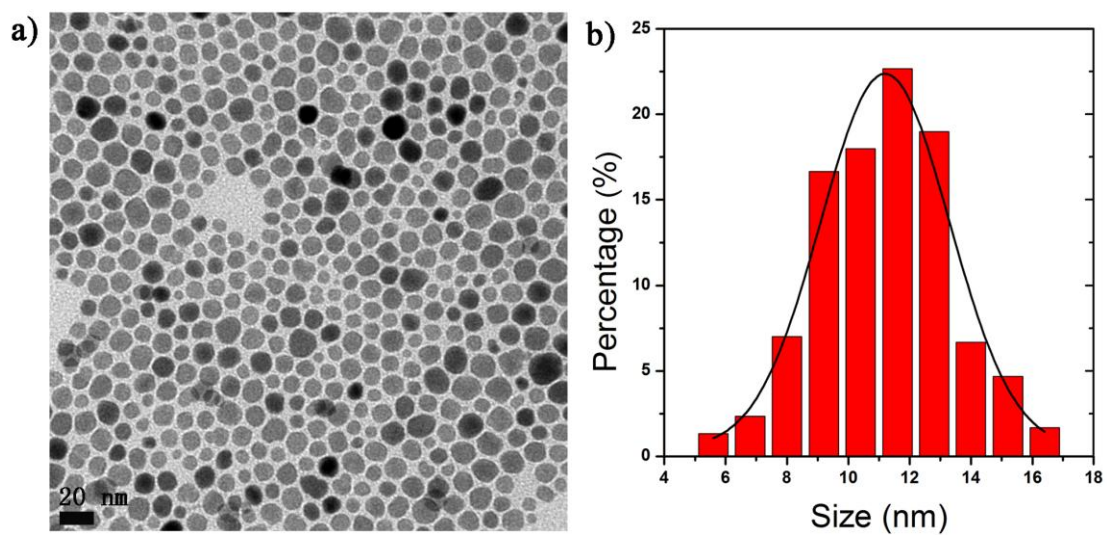


**Figure S3 ITO nanoflowers from the reactions using *n*-octanoic acid in the reagents. a) a typical TEM image and b) UV-Vis-NIR spectrum.**

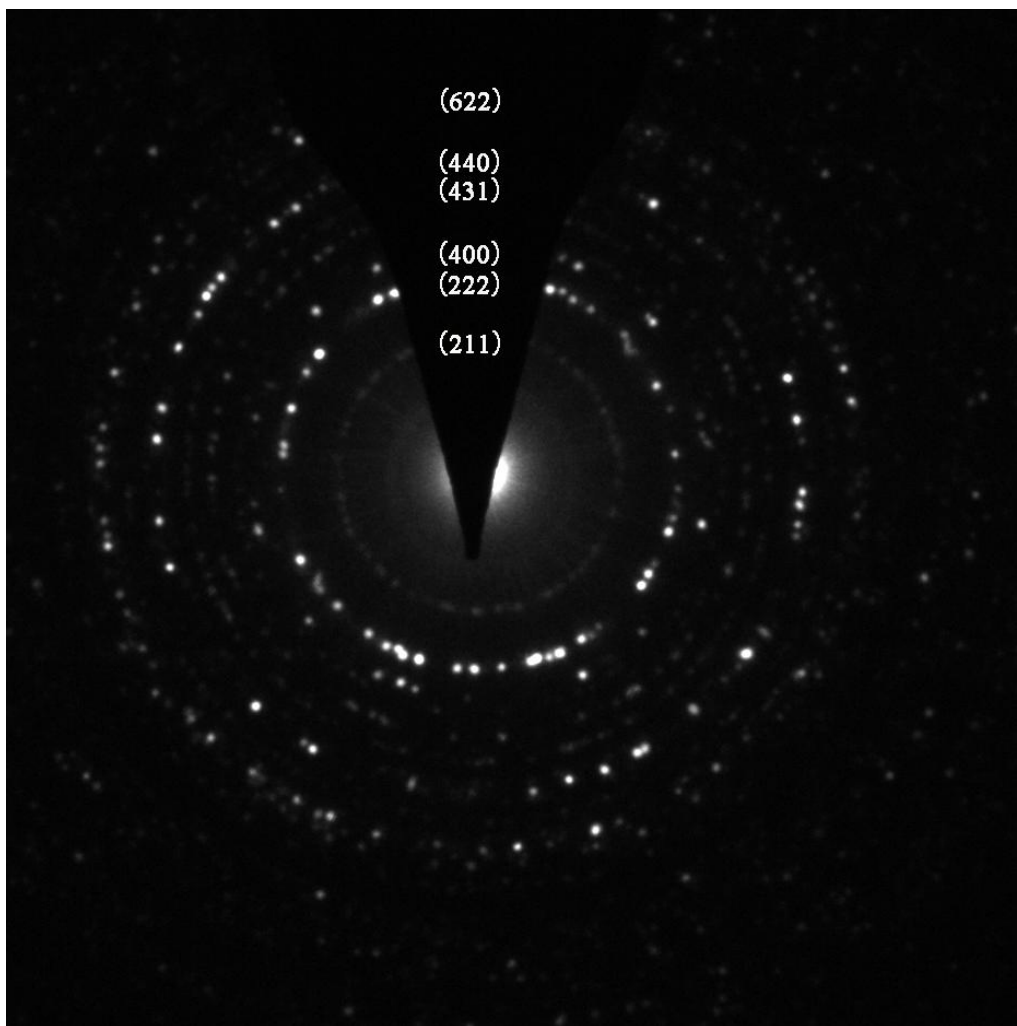


**Figure S4 Temporal evolution of the morphologies of the ITO nanocrystals.**

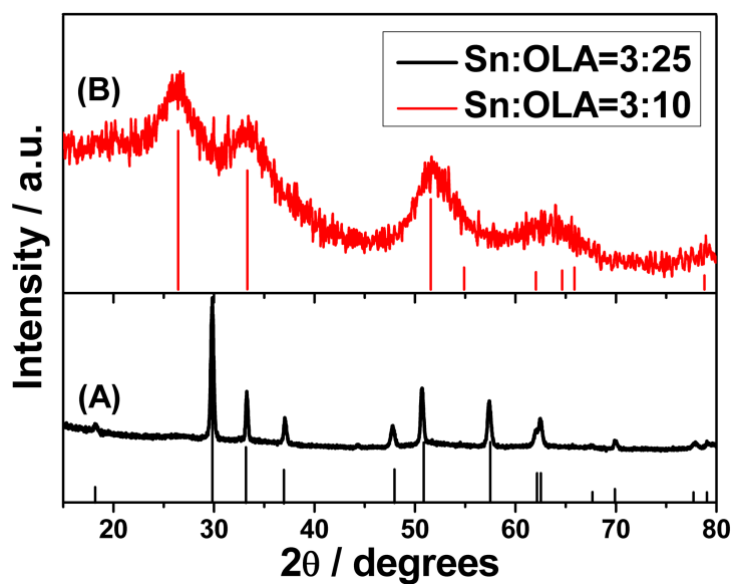
a)-d)TEM images were recorded from the nanocrystals in the aliquots which were taken from the reaction flask at 3, 8, 20 and 30 min, respectively.



**Figure S5 ITO nanocrystals (10 mol.% of tin precursor) obtained by the Masayuki method.** a) a typical TEM image and b) the corresponding histogram showing the size distribution of  $11.2 \pm 2.1$  nm.

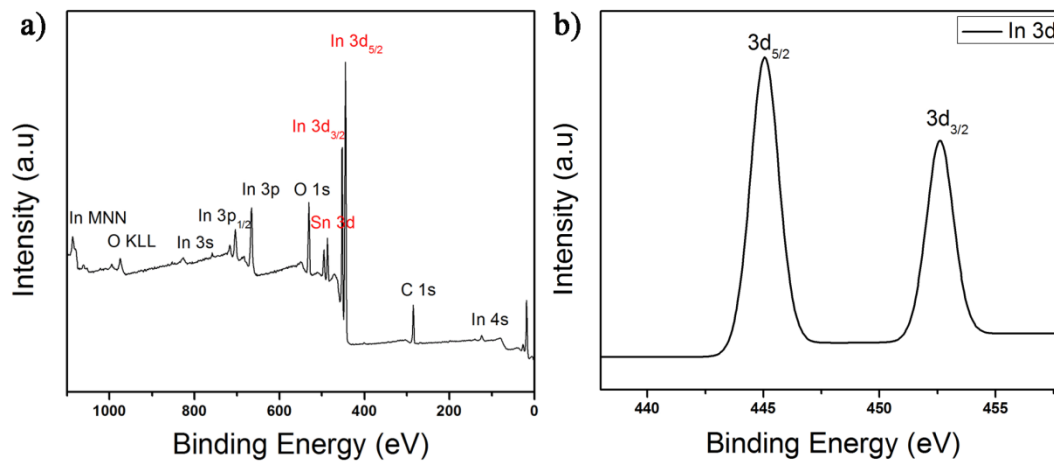


**Figure S6** Electron diffraction pattern of the ITO nanocrystals.



**Figure S7** XRD patterns of the products prepared by aminolysis of different amounts of tin(II) 2-ethylhexanoate in oleylamine at 280 °C for 1 h. The results suggest that both (A) SnO and (B) SnO<sub>2</sub> (JCPDS patterns of PDF#06-0395 and PDF#41-1445) may be obtained depending on the reaction conditions.





**Figure S8 XPS spectra of the ITO nanocrystals from the hot-injection approach.**

a) wide scan of the ITO nanocrystals and b) narrow scan of the In 3d peaks.