Supplementary Material For:

Computational Design of Enone-Binding Proteins with Catalytic Activity for the Morita-Baylis-Hillman Reaction

Sinisa Bjelic^{1,#}, Lucas G. Nivon^{1,#}, Nihan Çelebi-Ölçüm^{2,3}, Gert Kiss², Carolyn F. Rosewall⁴, Helena M. Lovick⁴, Erica L. Ingalls⁴, Jasmine Lynn Gallaher¹, Jayaraman Seetharaman⁵, Scott Lew⁵, Gaetano Thomas Montelione⁵, John Francis Hunt⁵, Forrest Edwin Michael⁴, K. N. Houk², David Baker^{1,6,*}

Supplementary Methods

Theozyme setup

The theozyme was constructed with the substrates stacking on top of each other to exploit the maximum interaction between cyclohexenone and 4-nitrobenzaldehyde. The theozyme contained three stereocenters, with the nucleophile attacking the beta carbon of cyclohexenone to create an (S) configuration at the anomeric carbon. The alpha carbon of the cyclohexenone was in an (S) configuration and the carbon of the aldehyde in an (R) configuration.

Design filtering

After initial filtering by scaffold and energy terms, designs were also ranked by other factors such as the number of unsatisfied hydrogen bonds, surface complementarity, and active-site rigidity upon repacking without the ligand. Filtered designs were inspected manually to select the best designs for testing. Those final designs were adjusted by manual inspection to place backing-up interactions to the catalytic interactions (backing-up residues often lie outside of the designable shell in Rosetta), or to place a specific residue where Rosetta was not able to clearly distinguish a best residue at a given position. Missing residues from the original construct used in expression of the scaffold were added and a glycine-serine dipeptide was appended before the C-terminal Hexa-His tag.

Gene synthesis and mutagenesis, protein expression and purification

Final designed genes were synthesized by Genscript (Genscript, USA Inc. Piscataway NJ) and cloned into a pET29b+ vector using NdeI and XhoI cleavage sites. This places a leucine-glutamate-hexa-histidine tag at the C-terminus of each polypeptide chain. Genes were transformed into BL21(DE3) pLysS *E. coli* cells and grown in LBautoinduction medium for 6 hours at 37 deg. C followed by 20-24 hours of induction at 18 deg. C. The lysate was filtered and passed over Qiagen Ni-NTA columns, followed by elution in a 250 mM imidazole solution. Proteins were concentrated and dialyzed into PBS (Phosphate buffer saline; 1X, pH 7.5) buffer for subsequent testing.

The Kunkel mutagenesis protocol was used for single point and saturation mutagenesis (*1*) and mutant gene variants were confirmed by sequencing (Genewiz, Inc.).

Synthesis of 3, 4 and aldol side-product 5

All reactions were performed under nitrogen atmosphere using flame-dried glassware. Infrared spectra were measured on a Perkin Elmer Spectrum RX I Spectrometer. Small-molecular mass spectra were collected on a JEOL HX-110 Mass Spectrometer with FAB or electron impact ionization, or a Bruker Esquire 1100 Liquid Chromatograph - Ion Trap Mass Spectrometer. Column chromatography was performed using silica gel (Sorbent Technologies, 60 Å, 230-400 mesh). NMR spectra were recorded on Bruker AV-300 or AV-500 spectrometers. ¹H NMR chemical shifts (δ) are reported in parts per million (ppm) downfield of TMS and are referenced relative to TMS (0.00 ppm) or residual protonated CHCl₃ (7.26 ppm). ¹³C NMR chemical shifts (δ) are reported in parts per million (ppm) relative to the carbon resonance of CDCl₃ (77.0 ppm).

Materials. THF and CH₂Cl₂ were degassed and dried on solvent columns of neutral alumina. All other commercial reagents were used as received. Deuterated solvents were purchased from Cambridge Isotope Laboratories, Inc., stored over 4Å molecular sieves, and were used without further purification.

4-Formyl-*N***-(prop-2-ynyl)benzamide (1)**. 4-Carboxylbenzaldehyde (2.164 g, 14.4 mmol), propargylammonium chloride (1.313 g, 14.1 mmol), 1-hydroxybenzotriazole (1.9 g, 14.1 mmol) *N*,*N***-**diisopropylcarbodiimide (2.33 mL, 14.1 mmol) and triethylamine (2.0 mL, 14.1 mmol) were combined in dichloromethane and stirred at room temperature. The resulting mixture was diluted with dicholoromethane and washed with 1M HCl, 1M NaHCO₃ and water, dried over MgSO₄ and filtered and concentrated. The product was purified by column chromatography EtOAc/Hex followed by recrystallization from EtOAc/Hex obtaining a white solid.

¹H NMR (500 MHz, CDCl₃): δ 10.08 (s, 1H), 7.97-7.93(m, 4H), 6.43 (br s, 1 H), 4.28 (dd, J = 5.0 Hz, 2.4 Hz, 2H), 2.31(t, J = 2.4 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 191.4, 165.9, 138.8, 138.5, 129.9, 127.8, 78.9, 72.3, 29.9. GC/MS (CI, m/z): 187(21), 158(35), 133(100), 105(52), 77(53), 51(38), 39(15). FTIR (KBr, cm⁻¹): 3314, 3242, 2833, 2732, 2117, 1740, 1690, 1643, 1572, 1542, 1499, 1420, 1388, 1352, 1323, 1298, 1258, 1209, 1182, 1154, 1051, 1016, 986, 920, 850, 798, 757, 709, 683.



4-(1-Hydroxybut-2-enyl)-*N*-(**prop-2-ynyl**)**benzamide** (**2**). In an oven dried 250 mL 2necked rbf under N₂, 4-formyl-N-(prop-2-ynyl)benzamide (**1**, 0.317 g, 1.7 mmol) was dissolved in THF (75 mL). The solution was then cooled to 0 $^{\circ}$ C, and propenylmagnesium bromide (0.5 M, 10.5 mL) was added dropwise over 30 min. The mixture was immediately quenched at 0 $^{\circ}$ C by adding water, and was then extracted with ethyl acetate (2 X 75 mL). The organic layers were combined and washed with water (25 mL), and sat. NH₄Cl (25 mL), then dried over MgSO₄, filtered and concentrated. The product was obtained as a thick oil (105 mg, 27% yield) of a 3:2 mixture of *E/Z* isomers after column chromatography (1-2% MeOH/DCM).

¹H NMR (500 MHz, CDCl₃, observed as a 3:2 mixture of isomers): δ 7.77 (d, *J* = 8.8 Hz, 4H, both), 7.46 (t, *J* = 9.0 Hz, 4H, both), 6.43 (s, 2H, both), 5.77 (dq, *J* = 15.0, 6.5 Hz, 1H, major), 5.75 – 5.60 (m, 3H, both), 5.20 (d, *J* = 7.0 Hz, 1H, major), 4.24 (dd, *J* = 5.0, 2.5 Hz, 4H, both), 2.29 (t, *J* = 2.5 Hz, 2H, both), 2.26 (br s, 2H, both), 1.83 (d, *J* = 6.9 Hz, 3H, minor), 1.72 (d, *J* = 6.4 Hz, 3H, major). ¹³C NMR (125 MHz, CDCl₃, observed as a 3:2 mixture of isomers): δ 166.9(both), 147.6(minor), 147.3(major), 133.2(major), 132.7(major), 132.4(minor), 128.3(major), 127.2(minor), 127.15(major), 127.1(minor), 126.3(major), 13.4(minor). GC/MS (CI, m/z): 229(1), 210(57), 157,(54), 128(100), 77(16), 51(16). FTIR (thin film, cm⁻¹): 3295, 2916, 2100, 1642, 1612, 1570, 1542, 1500, 144, 1421, 1353, 1305, 1152, 1047, 967, 920, 858.



(*E*)-(4-But-2-enoyl-*N*-(prop-2-ynyl)benzamide (3). In a 4 dram vial under N_2 2 (98 mg, 0.43 mmol) was dissolved in 10 mL of dichloromethane. Dess-Martin periodinane (182

mg, 0.43 mmol) was added to the mixture and was allowed to stir overnight. The mixture was diluted with dichloromethane (20 mL) and then washed with saturated NaHCO₃ (2 X 50 mL), water (50 mL) and dried over MgSO₄, filtered and concentrated. The crude material was a 1:1 mixture of E/Z isomers, which were separated by column chromatography (1:4 EtOAc/Hex) obtaining the (*E*) isomer as white solid (32 mg, 33% yield).

¹H NMR (500 MHz, CDCl₃): δ , 7.96(d, J = 8.2 Hz, 2H), 7.87(d, J = 8.3 Hz, 2H), 7.10(dq, J = 15.5, 7.0 Hz, 1H), 6.89(dd, J = 15.5, 1.5 Hz, 1H), 6.54 (br s, 1 H), 4.28 (dd, J = 5.0 Hz, 2.4 Hz, 2H), 2.31 (t, J = 2.45 Hz, 1H), 2.02(dd, J = 7.0, 1.5 Hz, 3H) ¹³C NMR (125 MHz, CDCl₃): δ 190.1, 166.2, 146.3, 140.5, 137.1, 128.7, 127.4, 127.3, 79.2, 72.1, 29.9, 18.7. GC/MS (CI, m/z): 227(44), 173(100), 115(31), 76(29), 69(52). FTIR (KBr, cm⁻¹): 3568, 3280, 3237, 2947, 2123, 1670, 1637, 1617, 1560, 1533, 1499, 1437, 145, 1354, 1336, 1291, 1224, 1156, 1106, 1015, 993, 964, 923, 875, 823, 765, 697, 668, 647.



2-(Hydroxy(4-nitrophenyl)methyl)cyclohex-2-enone (4): Prepared as previously reported (2), spectral data matches literature values. ¹H NMR (300 MHz, CDCl₃): δ 8.20 (d, J = 8.7 Hz, 2H), 7.57 (d, J = 8.7 Hz, 2H), 6.87 (t, J = 3.9 Hz, 1H), 5.63 (d, J = 5.7 Hz, 1H), 3.68 (d, J = 5.7 Hz, 1H), 2.46 (m, 4H), 2.01 (m, 2H).



6-(Hydroxy(4-nitrophenyl)methyl)cyclohex-2-enone (5): In a flame-dried 25 mL round-bottomed flask diisopropylamine (0.3 mL, 2.2 mmol) and THF (10 mL) were combined and cooled to -78 °C. Then *n*-butyllithium (2.2 M, 0.9 mL, 2 mmol) was added slowly. The solution was stirred at -78 °C for 30 min then 2-cyclohexenone (0.19 mL, 2 mmol) was added. The reaction mixture was again stirred at -78 °C for 30 min. Then 4nitrobenzaldehyde (0.302 g, 2 mmol) was added. After one minute, sat. NH₄Cl was added to quench the reaction. The mixture was extracted with $Et_2O(3X)$. The organic layers were combined and washed with water, sat. NaCl, dried (MgSO₄), and concentrated. The product was obtained as a 3:1 mixture of diastereomers which were partially separable by column chromatography (EtOAc/Hexanes). Spectral data matched literature values (3). Anti diastereomer (major); ¹H NMR (300 MHz, CDCl₃): δ 8.20 (d, 2 H, J = 8.7 Hz), 7.55 (d, 2 H, J = 8.7 Hz), 7.06 (m, 1 H), 6.08 (d, 1 H, J = 9.9 Hz), 4.97 (m, 1 H), 2.8-2.5 (m, 1 H)H), 2.5-2.2 (m, 2 H), 1.7-1.5 (m, 2 H). Syn diastereomer (minor); ¹H NMR (300 MHz, $CDCl_3$): 8.24 (dd, 2 H, J = 8.4, 1.8 Hz), 7.54 (d, 2 H, J = 8.4 Hz), 7.00 (m, 1 H), 6.13 (dd, 1 H, J = 9.6, 3.0 Hz), 6.10 (t, 1 H, J = 1.2 Hz), 2.91 (d, 1 H, J = 4.8 Hz), 2.70 (m, 1)H), 2.5-2.2 (m, 2 H), 2.00 (m, 1 H), 1.62 (m, 1 H).

Molecular Dynamics Simulations

Molecular dynamics (MD) were performed with Amber 11 (4) using explicit solvent and periodic boundaries to investigate the dynamical behavior of BH25 and

BH32 proteins. Simulations were run for 20-50 ns, or for one microsecond where specified.

System Preparation. Simulation systems were set up by placing the protein, including the co-crystallized water molecules from the scaffold, at the center of the simulation box and solvating the protein with TIP3P (5) water molecules ensuring a solvent layer of 10 Å around the protein. This resulted in the addition of ~10,000 – 20,000 solvent molecules depending on the scaffold and a system size of ~72,000 atoms for BH25 and ~32,500 atoms for BH32. The systems were neutralized by addition of explicit counter ions. All systems were parameterized using the Stony Brook modification of the Amber 99 force field (6). The parameters for the substrates, cyclohexenone and 4-nitrobenzaldehyde, were generated with the antechamber module of Amber 11 (4) using the general Amber force field (GAFF) (7) with partial charges set to fit the electrostatic potential generated at HF/6-31G(d) level of theory by RESP (8). The charges were calculated according to the Merz-Singh-Kollman scheme (9, 10) using Gaussian 03 (11). For the parameterization of the alkoxide intermediate bound to the nucleophilic cysteine

(int2, where $Nu = \psi$), assuring a total charge of -1.0 for the unit. LEaP module of Amber 11 was used to split the unit into two, and to generate separate libraries for the alkoxide intermediate (ALK) and the non-standard cysteine residue (CYC).

The systems were initially minimized for the positions of water molecules and ions, with harmonic restraints of 150 kcal/mol applied to the solute. Initial minimization was followed by an unrestrained minimization of all atoms. The systems were heated gently in six steps of 50 K for 50 ps (from 0 K to 300 K) at constant volume with a time step of 1 fs. Each system was equilibrated for 2 ns with a 2 fs time step in the NVT ensemble at 300 K using the Langevin equilibration scheme. The systems were then equilibrated for 2 ns with a 2 fs time step at a constant pressure of 1 atm. Harmonic

restraints of 30 kcal/mol were applied to the solute during the heating and equilibration stages, and water molecules were triangulated using the SHAKE algorithm.

Production MD. Multiple 20 - 50 ns production MD simulations were performed for each system (with and without the substrate bound to the active site) using PMEMD (*12*) in the isothermal-isobaric ensemble (NPT) with a time step of 2 fs. Long-range effects were modeled using the particle-mesh-Ewald method (*13*). This general MD protocol has been recently described for the evaluation and ranking of enzyme designs (*14*).

For microsecond production runs of BH32 we used DESRES's Anton special purpose machine (*15*) at the Pittsburgh Supercomputing Center.

Trajectory analysis. Geometries and velocities were saved every 0.2 ps, resulting in a total of 100,000 frames from each production run. Post-MD data extraction and analysis was performed using the ptraj module of Amber 11.

Crystallization, data collection and structure determination

The crystals of SeMet-BH32 were grown in two different conditions by mixing 1 μ l of protein sample with 1 μ l of reservoir solution consisting of 0.1M PBS buffer PH 7.5 25% PEG 3350. Condition two is the same as condition one plus the addition of 0.1M cyclohexenone. The crystals were obtained by the hanging drop vapor diffusion method. The crystals of SeMet-BH25 N43Y were obtained by mixing 1 μ l of protein sample with 1 μ l of reservoir solution consisting of 1.44M potassium acetate, 50mM MES, PH 6 by the micro-batch under-oil method. Both crystals were grown at 18°C, cryo-protected with 20% glycerol and flash-cooled in liquid nitrogen. Diffraction data sets were collected at

the peak of the selenium K edge on a single crystal using the beam line X4A with a Quantum 4R detector at the National Synchrotron Light Source (NSLS) at Brookhaven National Laboratory. Data were integrated and scaled with the HKL2000 package (Otwinowski and Minor, 1997). Matthew's coefficient calculations indicated one molecule per asymmetric unit in the monoclinic space group for BH32 and three molecules per symmetric unit in the tetragonal space group for BH25 N43Y.

The structures of BH32 and BH25 N43Y were solved by the single-wavelength anomalous dispersion (SAD) phasing method by SHELX (*16*) using a SeMet-substituted crystal. An experimental electron density map was obtained using ShelxD. After phase refinement we constructed an initial model with resolve (*17*) extended the model using ARP/wARP (*18*) and refined it with Refmac (*19*) and CNS (*20*). Model building was performed using Coot (*21*). Several cycles of simulated annealing and minimization were carried out using the CNS program package (*20*). The R-free was calculated based on 10% of randomly selected data excluding from the refinement. Structure validation was performed with PROCHECK (*22*). Residues in the loop region 169–177 and 231–238 in BH25 N43Y are not defined in the electron density map and are assumed to be disordered. The crystallographic statistics for data collection and refinement are summarized in Table S1.

Analysis of active mutants

We can rationalize improvements in activity in the optimized sequences based on the crystal structures. For BH25 we identified the variant N43Y as the most active point mutant - it was predicted to hydrogen bond with the oxyanion of Int1, and the crystal structure supports this (Figure 4B). Other more active mutants include (with structural

justification): W164Y (creates room for the W166 stacking on 4-nitrobenzaldehyde), G312M (better packing), Y129F (create a more hydrophobic pocket for the nitro group which does not form hydrogen bonds in water).

For BH32 we identified a number of variants with slightly higher activity than the wild-type design: S124A (intended to create a more hydrophobic pocket for the nitro group), S9H (remove hydrogen bonding residue from nitro-group pocket), S91V (Original intended hydrogen-bond may be too long, testing a replacement with hydrophobic packing). The most active point-mutant N14I was tested after examining the crystal structure - the backbone moves too far in the crystal structure for N14 to form the intended hydrogen bond, and the mutation instead forms a hydrophobic pack from the new backbone position. The S9G mutant creates a more hydrophobic pocket for the nitro group. The MD simulation correctly predicts many of the sidechain shifts observed in the crystal structure, but we could not identify simple local mutations to repair these deficiencies and get higher activity.

Appendix A. Composite transition s	state – a superposition	of the transition state	(from QM and MD)
and int2			

HETATM	1	C1	LG1	х	1	26.285	45.600	23.012	1.00	0.00	
HETATM	2	01	LG1	х	1	23.471	44.469	21.679	1.00	0.00	
HETATM	3	C2	LG1	х	1	25.504	44.500	23.726	1.00	0.00	
HETATM	4	02	LG1	х	1	23.398	44.667	22.276	1.00	0.00	
HETATM	5	C3	LG1	х	1	24.199	43.557	22.275	1.00	0.00	
HETATM	6	03	LG1	х	1	25.729	42.801	25.316	1.00	0.00	
HETATM	7	C4	LG1	х	1	24.415	43.845	22.785	1.00	0.00	
HETATM	8	04	LG1	х	1	27.223	38.764	19.257	1.00	0.00	
HETATM	9	C5	LG1	х	1	27.556	45.063	22.379	1.00	0.00	
HETATM	10	05	LG1	х	1	27.381	40.257	17.776	1.00	0.00	
HETATM	11	C6	LG1	х	1	28.390	44.284	23.374	1.00	0.00	
HETATM	12	06	LG1	х	1	26.031	43.356	25.744	1.00	0.00	
HETATM	13	C7	LG1	х	1	27.640	43.117	23.985	1.00	0.00	
HETATM	14	07	LG1	х	1	22.166	46.667	24.146	1.00	0.00	
HETATM	15	C8	LG1	х	1	26.239	43.475	24.381	1.00	0.00	
HETATM	16	C9	LG1	х	1	27.031	39.884	18.868	1.00	0.00	
HETATM	17	C10	LG1	x	1	26.217	40.075	18.757	1.00	0.00	
HETATM	18	C11	LG1	x	1	26.319	40.824	19.739	1.00	0.00	
HETATM	19	C12	LG1	x	1	26.082	42.136	19.334	1.00	0.00	
HETATM	20	C13	LG1	x	1	25.389	43.031	20.158	1.00	0.00	
HETATM	21	C14	LG1	x	1	24,913	42.630	21,402	1.00	0.00	
HETATM	22	C15	T.G1	x	1	25 154	41 322	21 819	1 00	0 00	
HETATM	23	C16	LG1	x	1	25.852	40.430	20.999	1.00	0.00	
HETATM	24	C17	LG1	x	1	25.743	41.023	19.771	1.00	0.00	
HETATM	25	C18	T.G1	x	1	25 582	42 377	19 467	1 00	0 00	
HETATM	26	C19	T.G1	x	1	25 141	43 284	20 435	1 00	0 00	
HETATM	27	C20	T.G1	x	1	24 856	42 856	21 727	1 00	0 00	
HETATM	28	C21	T.G1	x	1	25 000	41 507	22 036	1 00	0 00	
HETATM	29	C22	T.G1	x	1	25 444	40 601	21 067	1 00	0 00	
HETATM	30	C23	T.G1	x	1	26 361	38 918	19 063	1 00	0 00	
HETATM	31	C24	LG1	x	1	26 457	40 480	17 653	1 00	0 00	
HETATM	32	H1	T.G1	x	1	25 651	45 975	22 230	1 00	0 00	
HETATM	33	H2	T.G1	x	1	27 322	44 421	21 548	1 00	0 00	
HETATM	34	н3	LG1	x	1	28 138	45 867	21 967	1 00	0 00	
HETATM	35	нΔ	T.G1	x	1	29 277	43 922	22 904	1 00	0 00	
HETATM	36	H5	LG1	x	1	28.720	44.941	24,153	1.00	0.00	
HETATM	37	н6	T.G1	x	1	27 586	42 314	23 281	1 00	0 00	
HETATM	38	H7	T.G1	x	1	28 169	42 751	24 834	1 00	0 00	
HETATM	39	н8	T.G1	x	1	24 771	44 905	24 397	1 00	0 00	
HETATM	40	н9	LG1	x	1	23 651	43 088	23 072	1 00	0 00	
HETATM	41	H10	LG1	x	1	23 924	43 293	23 564	1 00	0.00	
HETATM	42	H11	T.G1	x	1	26 432	42 470	18 388	1 00	0 00	
HETATM	43	H12	LG1	x	1	25 231	44 029	19 828	1 00	0 00	
HETATM	44	H13	LG1	x	1	24 806	40 994	22 769	1 00	0 00	
HETATM	45	H14	LG1	x	1	26 024	39 434	21 332	1 00	0 00	
HETATM	46	H15	LCI	x	1	25 804	42 736	18 492	1 00	0 00	
HETATM	47	H16	LC1	x	1	25 038	44 314	20 190	1 00	0 00	
HETATM	48	H17	LG1	x	1	24 778	41 160	23 017	1 00	0 00	
HETATM	49	H18	LCI	x	1	25 554	39 575	21 327	1 00	0 00	
IIIIIAIM	-1.7	1110	цет	~	-	20.004	59.515	21.32/	1.00	0.00	

Figure S1. Product accumulation time course for BH32 and variants.



Figure S2. Overlay of representative MD snapshots (30-40 ns, in blue) and the wild-type designs (in gray). The enone substrate (in yellow) is docked in its designed orientation. The nucleophilic cysteine (C39) of BH25 (A) remains preorganized through a strong H-bond network between residues K285, D313 and R363. For BH32, catalytic base E46 is highly solvent exposed (B), and E46-water interactions strongly compete with E46-H23 dyad. Catalytic residues designed to donate H-bonds to the enolate intermediate in BH25 (C) and in BH32 (D) are engaged in alternative binding patterns generating a bottleneck for the later steps of the reaction.



Figure S3. MD on the covalently bound alkoxide intermediate (Int2) for design BH25.

(A) The design with docked intermediate; (B) an MD snapshot after 20 ns.



Figure S4. (A) Nucleophilic dyad for BH25. Overlay of an MD snapshot at 22 ns MD (in blue) and the wild-type BH25 design (in gray). The nucleophilic cysteine (C39) remains preorganized through a strong H-bond network between residues K285, D313 and R363. (B) The plot of distances C39^{HG}-D313^{OD} versus C39^{HG}-K285^{NZ} shows that C39 binds to D313 and K285 in a triangular fashion. The relative populations of strong/moderate H-bond configurations suggest a more tightly bound C39^{HG}-D313^{OD} compared to C39^{HG}-K285^{NZ}.



Figure S6. Electron density map 2fofc superimposed on the structure of BH32 with the crystal structure in orange and the design in pink. Residue H23 and the backing up D46 are shown.



Figure S7. Alignment of BH25 and BH32 designs with X-ray structures and original scaffold, respectively. BH25 is a dimer and red denotes designed residues, which are part of the designed active site. Purple is the second active site, yellow is a large structural change compared to the design and green is the improved variant N42Y.

BH25

	I-> Chain A
lftx 3uw6 BH25	NDFHRDTWAEVDLDAIYDNVENLRRLLPDDTHIMAWVKANAYGHGDVQVA NDFHRDTWAEVDLDAIYDNVENLRRLLPDDTHIMASVCGNAYGHGDVQVA NDFHRDTWAEVDLDAIYDNVENLRRLLPDDTHIMASVCGNAYGHGDVQVA
1.0	
litx 3uw6	RTALEAGASRLAVAFLDEALALREKGIEAPILV <mark>I</mark> GASRPADAALAAQQRI RTALEAGASRLAVAFLDEALALREKGIEAPILV <mark>I</mark> GASRPADAALAAQQRI
BH25	RTALEAGASRLAVAFLDEALALREKGIEAPILV <mark>I</mark> GASRPADAALAAQQRI ***********************************
lftx	ALTVFRSDWLEEASALYSGPFPIHFHL <mark>K</mark> MDTGMG <mark>H</mark> LGVKDEEETKRIVAL
3uw6 BH25	ALTVFRSDWLEEASALYSGPFPIHFHLYMDTGMGSLGVKDEEETKRIVAL ALTVFRSDWLEEASALYSGPFPIHFHL <mark>Y</mark> MDTGMG <mark>S</mark> LGVKDEEETKRIVAL *****************
lftx	IERHPHFVLEGL <mark>Y</mark> T <mark>H</mark> FATADEVNTDYFSYQYTRFLHMLEWLPSRPPLVH <mark>C</mark>
3uw6 BH25	IERHPHFVLEGLWINFATYFSYQYTRFLHMLEWLPSRPPLVH <mark>C</mark> IERHPHFVLEGLWINFATADEVNTDYFSYQYTRFLHMLEWLPSRPPLVH <mark>C</mark> *************
lftx	ANSAASLRFPDRTFNMV <mark>R</mark> FGIAMYGLAPSPGIKPLLPYPLKEAFSLHSRL
3uw6 BH25	ANSAASLRFPDRTFNMVOFGIAMYGLAPSLPYPLKEAFSLHSRL ANSAASLRFPDRTFNMVOFGIAMYGLAPSPGIKPLLPYPLKEAFSLHSRL

lftx 3uw6	VHVKKLQPGEKVS <mark>Y</mark> GATYTAQTEEWIGTIPIGY <mark>A</mark> DGWLRRLQHFHVLVDG VHVKKLOPGEKVSEGATYTAOTEEWIGTIPIGYKDGWLRBLOHEHVLVDG
BH25	VHVKKLQPGEKVSFGATTTAGTEEWIGTTFIGTADGWLKKLQHFHVLVDG ************************************
1fty	
3uw6	QKAPIVGRI <mark>LG</mark> D <mark>M</mark> CMIRLPGPLPVGTKVTLIGRQGDKVISIDDVARHLET
BH25	QKAPIVGRI <mark>CM</mark> DQCMIRLPGPLPVGTKVTLIGRQGDEVISIDDVARHLET ************************************
1fty	<mark> -> Chain B</mark> INVEVOCTISYRVORIFERHKRIMEVRNAINDEHRDTWAEVDLDAIV
3uw6	INYEVPCTISYRVPRIFFRHKRIMEVRNAIGRGNDFHRDTWAEVDLDAIY
BH25	INYEVPCTISYRVPRIFFRHKRIMEVRNAINDFHRDTWAEVDLDAIY ************************************
lftx	DNVENLRRLLPDDTHIMA <mark>V</mark> V <mark>KA</mark> NAYGHGDVQVARTALEAGASRLAVAFLD
3uw6 BH25	DNVENLRRLLPDDTHIMA <mark>S</mark> VCCNAYGHGDVQVARTALEAGASRLAVAFLD DNVENLRRLLPDDTHIMA <mark>V</mark> VKANAYGHGDVOVARTALEAGASRLAVAFLD
-	***************************************
lftx	EALALREKGIEAPILV <mark>L</mark> GASRPADAALAAQQRIALTVFRSDWLEEASALY

3uw6	EALALREKGIEAPILV <mark>T</mark> GASRPADAALAAQQRIALTVFRSDWLEEASALY
BH25	EALALREKGIEAPILV <mark>L</mark> GASRPADAALAA <u>QQ</u> RIALTVFRSDWLEEASALY *************
1ftx	SGPFPIHFHLKMDTGMGRLGVKDEEETKRIVALTERHPHFVLEGLYTHFA
311W6	SGPEPIHEHL <mark>Y</mark> MDTGMG <mark>S</mark> LGVKDEEETKRIVALIERHPHEVLEGL <mark>WTW</mark> FA
BH25	SGPFPIHFHL <mark>K</mark> MDTGMG <mark>R</mark> LGVKDEEETKRIVALIERHPHFVLEGL <mark>YTH</mark> FA
21120	********* ****** *********************
lftx	TADEVNTDYFSYQYTRFLHMLEWLPSRPPLVHCANSAASLRFPDRTFNMV
3uw6	TYFSYQYTRFLHMLEWLPSRPPLVHCANSAASLRFPDRTFNMV
BH25	TADEVNTDYFSYQYTRFLHMLEWLPSRPPLVHCANSAASLRFPDRTFNMV
	* *************************************
lftx	<mark>R</mark> FGIAMYGLAPSPGIKPLLPYPLKEAFSLHSRLVHVKKLQPGEKVS <mark>Y</mark> GAT
3uw6	<mark>Q</mark> FGIAMYGLAPSLPYPLKEAFSLHSRLVHVKKLQPGEKVS <mark>H</mark> GAT
BH25	<mark>R</mark> FGIAMYGLAPSPGIKPLLPYPLKEAFSLHSRLVHVKKLQPGEKVS <mark>H</mark> GAT

lftx	YTAQTEEWIGTIPIGY <mark>A</mark> DGWLRRLQHFHVLVDGQKAPIVGRI <mark>CM</mark> D <mark>Q</mark> CMIR
3uw6	YTAQTEEWIGTIPIGY <mark>K</mark> DGWLRRLQHFHVLVDGQKAPIVGRI <mark>LG</mark> D <mark>M</mark> CMIR
BH25	YTAQTEEWIGTIPIGY <mark>K</mark> DGWLRRLQHFHVLVDGQKAPIVGRI <mark>LG</mark> D <mark>M</mark> CMIR
	*************** ***********************
lftx	LPGPLPVGTKVTLIGRQGDEVISIDDVARHLETINYEVPCTISYRVPRIF
3uw6	LPGPLPVGTKVTLIGRQGDKVISIDDVARHLETINYEVPCTISYRVPRIF
BH25	LPGPLPVGTKVTLIGRQGDEVISIDDVARHLETINYEVPCTISYRVPRIF

lftx	FRHKRIMEVRNAI
3uw6	FRHKRIMEVRNAIGRG
BH25	FRHKRIMEVRNAI
	* * * * * * * * * * * * *

BH32

1x42 BH32	MIRAVFFDEVGTLESVEGEAKTHLKIMEEVLGDYPLNPKTLLDEYEKLTR MIRAVFFDSIGTLNSVEGEAKSHLKIMEEVLGDYPLNPKTLLDEYEKLTR ******* :*** **** **:*****************
1x42 BH32	EAFSNYAGKPYRPIRDIEEVMRKLAEKYGFKYPEN <mark>FWEIHLRMFORY</mark> GE EAFSNYAGKPYRPIRDIEEVMRKLAEKYGFKYPEN <mark>FWEISLRMSORY</mark> GE ************************************
1x42 BH32	LYPEVVEVLKSLKGKYHVGMITDSD <mark>TEYLMAH</mark> LDALGIKDLFDSITTSEE LYPEVVEVLKSLKGKYHVGMITDSD TEQAMAH LDALGIKDLFDSITTSEE **********************************
1x42 BH32	AGFFKPHPRIFELALKKAGVKGEEAVYVGDNPVKDCGGSKNLGMTSILLD AGFFKPHPRIFELALKKAGVKGEEAVYVGDNPVKDCGGSKNLGMTSILLD ***********************************
1x42 BH32	RKGEKREFWDKCDFIVSDLREVIKIVDELNGQ RKGEKREFWDKCDFIVSDLREVIKIVDELNGQ *****

Protein Name	BH32	BH32	BH25 N43Y
Space group	P21	P21	P41212
Molecules per	1	1	3
asymmetric unit			
<i>V</i> м (Å ³ Da ⁻¹)	1.59	2.36	2.82
Unit Cell (Å,º)	a=33.606,b=67.	a=34.334,b=	a=b=112.345,c=23
	579,c=48.193,α	71.367 c=	7.059,α=β=γ=90.0°
	=90,β=109.30γ	52.926,α=90,	
	=90.0°	β= 104.83	
		,γ=90	
Wavelength(Å)	0.979	0.979	0.979
Resolution (Å)	50-1.59(1.59-	50-2.3(2.3-	50-2.3(2.3-2.41)
	1.63) ^a	2.38)	
Temperature(K)	100	100	100
Unique reflections	27144	10847	85640
Mean I/σ(I)	22.8	16.1	23.1
Sigma Cutoff	0	0	0
Completeness	91.0(69.0)	99.1(100.0)	99.5(96.1)
Redundancy	2.0(1.5)	7.5(6.0)	10.0
R _{merge} #	0.040(0.128)	0.082(0.179)	0.070(0.058)
R _{cryst} +	0.200	0.238	0.215
R _{free} *	0.248	0.279	0.259
RMSD			
Bond lengths (Å)	0.021	0.008	0.010
Bond angles (o)	2.17	1.20	1.30
No. of residues	230	230	1113
No. of ions		1	

Table S1. Summary of crystal parameters, data collection and refinement.

 $R_{\text{cryst}} = \sum_{hkl} ||F_{\text{obs}}| ||F_{\text{cnlc}}|| / \sum_{hkl} |F_{\text{obs}}|$ * R_{free} is calculated in same manner as R_{cryst} except that it uses 10% of the reflection data omitted from refinement.

^a Values in parentheses are for the highest resolution bin.

Table S2. Annotation of sidechain dihedral angles in BH25 design. Dihedrals are measured to chain C in 3UW6 X-ray structure. In general only small deviations are observed between different chains in the unit cell. In cases where there is a definite difference between the sidechain chi angels additional values are reported. Sidechain chis are reported in degrees and rounded of to whole numbers. ^{*a*} Dihedral measured to CD1 and ^{*b*} to OE1, respectively. ^{*c*} No density for the sidechain. ^{*d*} Dihedral is to CD1 in chain B.

3UW6	$\chi_{\scriptscriptstyle 1}$		χ	2	$\chi_{_3}$		7	\mathcal{U}_4
numbering	Nbb-Ca	a-Cb-						
	Co	g						
3UW6	3UW6	BH25						
S37	71	-179						
C39	50	72						
G40	_							
¥43	_							
T85	60	-58						
Y129	-180	-176	85ª	17 ^ª				
S136	79	55						
W164	60	65	-67 ^a	-72ª				
W166	167	-170	-77 ^ª	-105 ^ª				
Q219	-75	-178	-178	-179	53 [°]	3		
F265	N.D. ^c							
K285	-73	-60	-149	179	-83	-178	-72	180
L311	-64	-61	-37 ^a	-42				
			-117 ^d					
G312	—							

Q314	63	64	170	-170	98	78	

Table S3. Annotation of sidechain dihedrals in the BH32 design. Dihedrals are measuredrelative to the X-ray structure 3U26.

3U26	$\chi_{\scriptscriptstyle 1}$		χ	2	λ	3	7	χ_4
numbering	Nbb-Ca	a-Cb-						
	Co	9						
3U26	3U26	BH32	3U26	BH32	3U26	BH32	3U26	BH32
S9	65	179						
L10	-53	-177						
L14	63	-177	133	23				
A19	-	-						
Н23	-85	-72	68	179				
E46	-70	-75	-58	-67	-55	-51		
L64	-89	-68	-180	-179				
L68	-60	-62	173	175				
S91	-58	-172						
S95	-62	-66						
Q128	-72	-145	-179	-107	-62	55		
A129	-	-						
F132	176	-179	82	71				

REFERENCES

- 1. Kunkel, T. A. (1985) Rapid and Efficient Site-Specific Mutagenesis without Phenotypic Selection, *Proc. Natl. Acad. Sci. U. S. A.* 82, 488-492.
- 2. Luo, S. Z., Wang, P. G., and Cheng, J. P. (2004) Remarkable rate acceleration of imidazole-promoted Baylis-Hillman reaction involving cyclic enones in basic water solution, *J. Org. Chem.* 69, 555-558.
- 3. Kataoka, T., Iwama, T., Tsujiyama, S., Iwamura, T., and Watanabe, S. (1998) The chalcogeno-Baylis-Hillman reaction: A new preparation of allylic alcohols from aldehydes and electron-deficient alkenes, *Tetrahedron 54*, 11813-11824.
- Case, D. A., Darden, T. A., Cheatham, I. T. E., Simmerling, C. L., Wang, J., Duke, R. E., Luo, R., Walker, R. C., Zhang, W., Merz, K. M., Roberts, B., Wang, B., Hayik, S., Roitberg, A., Seabra, G., Kolossváry, I., Wong, K. F., Paesani, F., Vanicek, J., Liu, J., Wu, X., Brozell, S. R., Steinbrecher, T., Gohlke, H., Cai, Q., Ye, X., Wang, J., Hsieh, M. J., Cui, G., Roe, D. R., Mathews, D. H., Seetin, M. G., Sagui, C., Babin, V., Luchko, T., Gusarov, S., Kocalenko, A., and Kollman, P. A. (2010) AMBER 11, University of California, San Francisco.
- Jorgensen, W. L., Chandrasekhar, J., Madura, J. D., Impey, R. W., and Klein, M. L. (1983) Comparison of Simple Potential Functions for Simulating Liquid Water, *Journal of Chemical Physics 79*, 926-935.
- 6. Wang, J. M., Cieplak, P., and Kollman, P. A. (2000) How well does a restrained electrostatic potential (RESP) model perform in calculating conformational energies of organic and biological molecules?, *Journal of Computational Chemistry 21*, 1049-1074.
- 7. Wang, J. M., Wolf, R. M., Caldwell, J. W., Kollman, P. A., and Case, D. A. (2004) Development and testing of a general amber force field, *Journal of Computational Chemistry 25*, 1157-1174.
- 8. Bayly, C. I., Cieplak, P., Cornell, W. D., and Kollman, P. A. (1993) A Well-Behaved Electrostatic Potential Based Method Using Charge Restraints for Deriving Atomic Charges - the Resp Model, *Journal of Physical Chemistry* 97, 10269-10280.
- 9. Besler, B. H., Merz, K. M., and Kollman, P. A. (1990) Atomic Charges Derived from Semiempirical Methods, *Journal of Computational Chemistry 11*, 431-439.
- 10. Singh, U. C., and Kollman, P. A. (1984) An Approach to Computing Electrostatic Charges for Molecules, *Journal of Computational Chemistry 5*, 129-145.
- Frisch, M. J., Trucks, G. W., Schlegel, H. B., Scuseria, G. E., Robb, M. A., Cheeseman, J. R., Montgomery, J. J. A., Vreven, T., Kudin, K. N., Burant, J. C., Millam, J. M., Iyengar, S. S., Tomasi, J., Barone, V., Mennucci, B., Cossi, M., Scalmani, G., Rega, N., Petersson, G. A., Nakatsuji, H., Hada, M., Ehara, M., Toyota, K., Fukuda, R., Hasegawa, J., Ishida, M., Nakajima, T. b., Honda, Y., Kitao, O., Nakai, H., Klene, M., Li, X., Knox, J. E., Hratchian, H. P., Cross, J. B., Bakken, V., Adamo, C., Jaramillo, J., Gomperts, R., Stratmann, R. E., Yazyev, O.,

Austin, A. J., Cammi, R., Pomelli, C., Ochterski, J. W., Ayala, P. Y., Morokuma, K., Voth, G. A., Salvador, P., Dannenberg, J. J., Zakrzewski, V. G., Dapprich, S., Daniels, A. D., Strain, M. C., Farkas, O., Malick, D. K., Rabuck, A. D., Raghavachari, K., Foresman, J. B., Ortiz, J. V., Cui, Q., Baboul, A. G., Clifford, S., Cioslowski, J., Stefanov, B. B., Liu, G., Liashenko, A., Piskorz, P., Komaromi, I., Martin, R. L., Fox, D. J., Keith, T., Al-Laham, M. A., Peng, C. Y., Nanayakkara, A., Challacombe, M., Gill, P. M. W., Johnson, B., Chen, W., Wong, M. W., Gonzalez, C., and Pople, J. A. (2004) Gaussian 03, Revision C.02, *Gaussian, Inc., Wallingford CT*.

- 12. Duke, R. E., and Pedersen, L. G. (2003) PMEMD, *University of North Carolina, Chapel Hill.*
- 13. Darden, T., York, D., and Pedersen, L. (1993) Particle Mesh Ewald an N.Log(N) Method for Ewald Sums in Large Systems, *Journal of Chemical Physics 98*, 10089-10092.
- 14. Kiss, G., Rothlisberger, D., Baker, D., and Houk, K. N. (2010) Evaluation and ranking of enzyme designs, *Protein Sci.* 19, 1760-1773.
- 15. Shaw, D. E. (2009) *Proceedings of the ACM/IEEE Conference on Supercomputing (SC09), ACM Press, New York.*
- 16. Sheldrick, G. M. (2008) A short history of SHELX, *Acta Crystallographica Section A* 64, 112-122.
- 17. Terwilliger, T. C. (2003) Automated main-chain model building by template matching and iterative fragment extension, *Acta Crystallographica Section D-Biological Crystallography 59*, 38-44.
- 18. Perrakis, A., Morris, R., and Lamzin, V. S. (1999) Automated protein model building combined with iterative structure refinement, *Nat. Struct. Biol. 6*, 458-463.
- 19. Murshudov, G. N., Vagin, A. A., and Dodson, E. J. (1997) Refinement of macromolecular structures by the maximum-likelihood method, *Acta Crystallographica Section D-Biological Crystallography* 53, 240-255.
- 20. Brunger, A. T., Adams, P. D., Clore, G. M., DeLano, W. L., Gros, P., Grosse-Kunstleve, R. W., Jiang, J. S., Kuszewski, J., Nilges, M., Pannu, N. S., Read, R. J., Rice, L. M., Simonson, T., and Warren, G. L. (1998) Crystallography & NMR system: A new software suite for macromolecular structure determination, *Acta Crystallographica Section D-Biological Crystallography* 54, 905-921.
- 21. Emsley, P., and Cowtan, K. (2004) Coot: model-building tools for molecular graphics, *Acta Crystallographica Section D-Biological Crystallography* 60, 2126-2132.
- 22. Lovell, S. C., Davis, I. W., Adrendall, W. B., de Bakker, P. I. W., Word, J. M., Prisant, M. G., Richardson, J. S., and Richardson, D. C. (2003) Structure validation by C alpha geometry: phi,psi and C beta deviation, *Proteins-Structure Function and Genetics 50*, 437-450.