

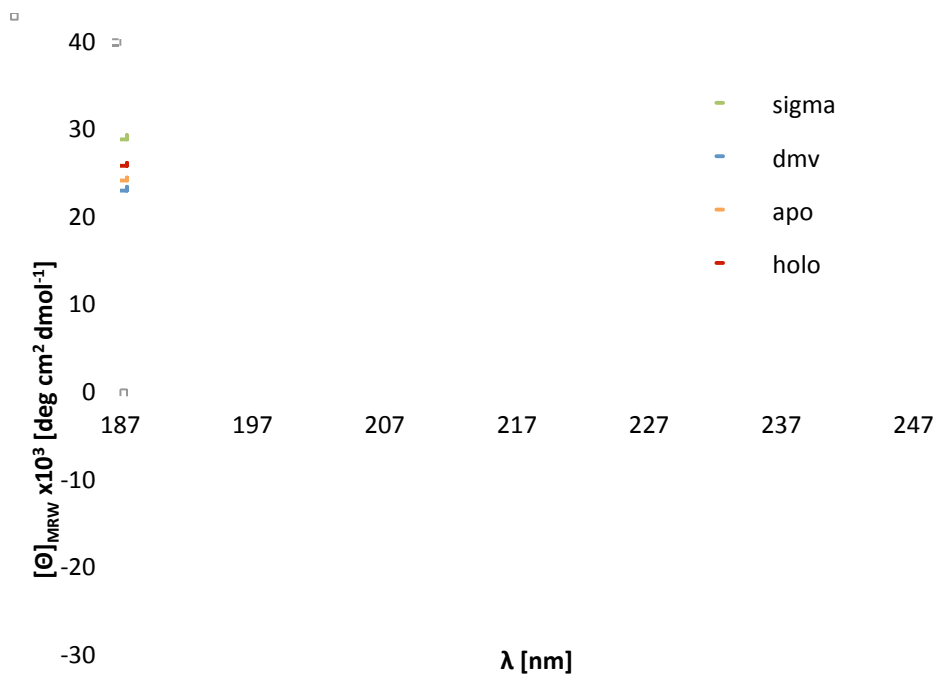
## A high-throughput method for the quantification of iron saturation in lactoferrin preparations

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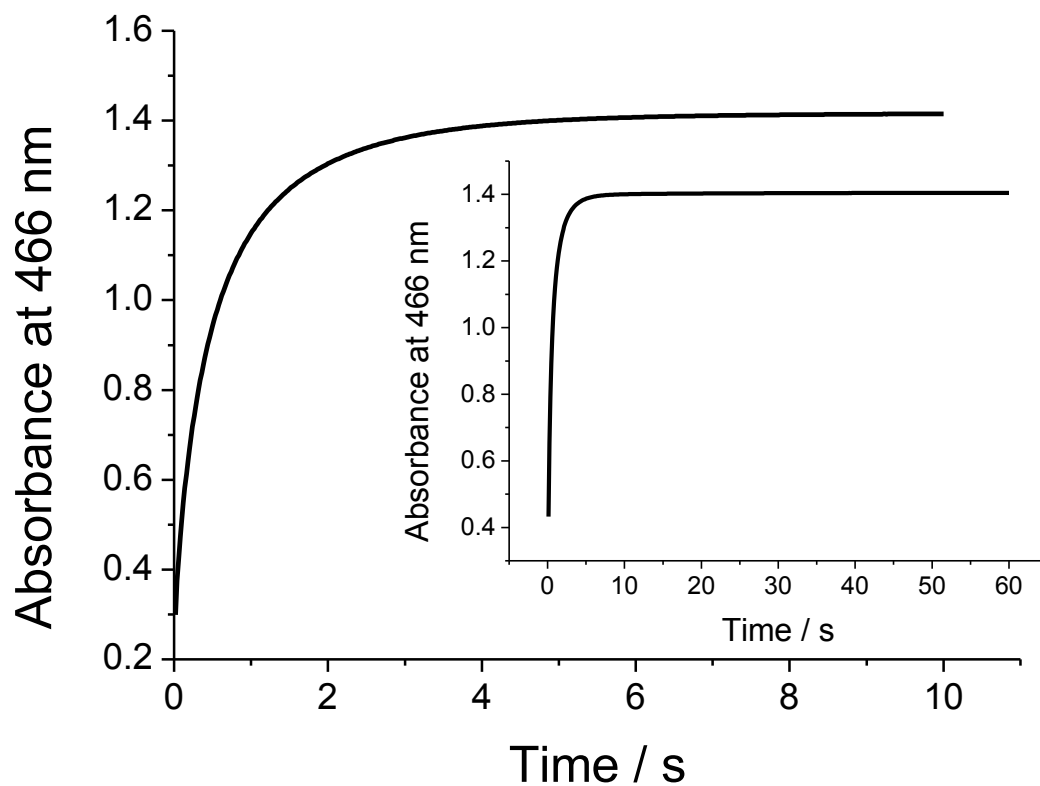
**Table S1.** Comparison of the absorbance coefficients values for apo- and hololactoferrin based on literature data

	$\lambda$ [nm]	$\varepsilon$ [ $M^{-1}cm^{-1}$ ]	Ref.
apolactoferrin		109000	[1]
	280	98400*	[2]
		101600*	[3]
	279	88500	[4]
	278	107000	[5]
hololactoferrin	280	135000	[1]
		112000*	[6]
	465	2800	[5]
		4640*	[6]
	466	2300	[7]
	470	2300	[8]

\* recalculated based on information about A1% or A0.1% for denoted wavelength



**Fig. S1.** Circular dichroism spectra of native lactoferrin obtained from Sigma-Aldrich (green) and DMV Int. (blue) companies as well as the apo- (< 2% of iron saturation) and holo-Lf (> 65% iron saturation) produced by us



**Fig. S2.** The increase in the absorbance monitored at 466 nm upon mixing of native lactoferrin (Lf) with a mixture of  $\text{Fe}(\text{NO}_3)_3$  and nitrilotriacetic acid (NTA). Experimental conditions:  $[\text{Lf}] = 0,32 \text{ mM}$ ,  $[\text{Fe}^{3+}] = 1,28 \text{ mM}$ ,  $[\text{NTA}] = 1,28 \text{ mM}$ ,  $[\text{NaCl}] = 150 \text{ mM}$ ,  $50 \text{ mM}$  Tris-HCl buffer,  $\text{pH} = 7.4$ ,  $T = 25 \text{ }^\circ\text{C}$

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