

High-Density Three-Dimension Graphene Macroscopic Objects for High-Capacity Removal of Heavy Metal Ions

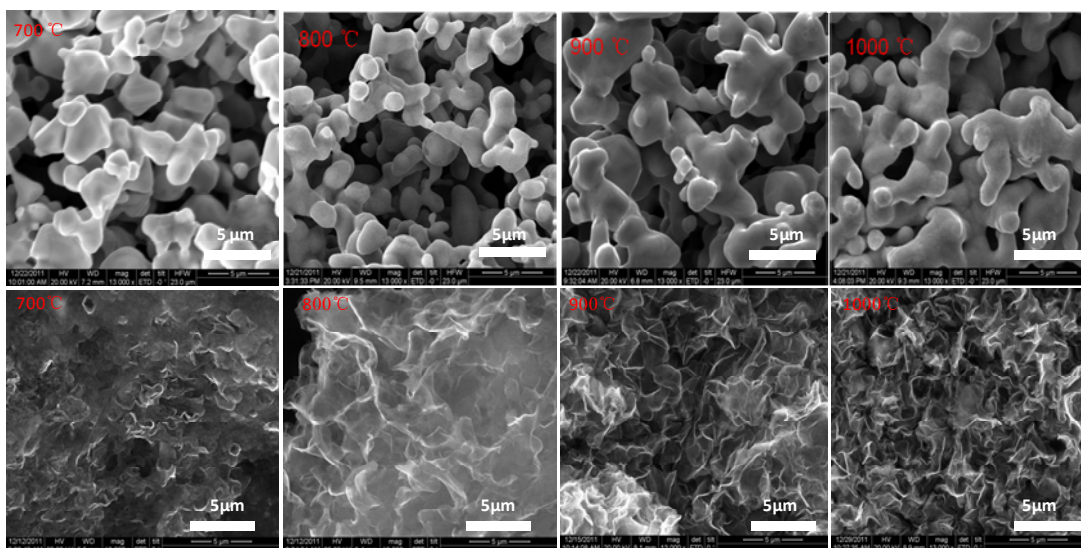
Weiwei Li^{1,2}, Song Gao¹, Liqiong Wu¹, Shengqiang Qiu¹, Yufen Guo^{1,2}, Xiumei Geng¹, Mingliang Chen¹, Shutian Liao¹, Chao Zhu¹, Youpin Gong¹, Mingsheng Long¹, Jianbao Xu^{1,2}, Xiangfei Wei¹, Mengtao Sun³ & Liwei Liu^{1*}

¹Key Laboratory of Nanodevices and Applications, Suzhou Institute of Nano-Tech and Nano-Bionics, Chinese Academy of Sciences, Suzhou, Jiangsu 215123, People's Republic of China.

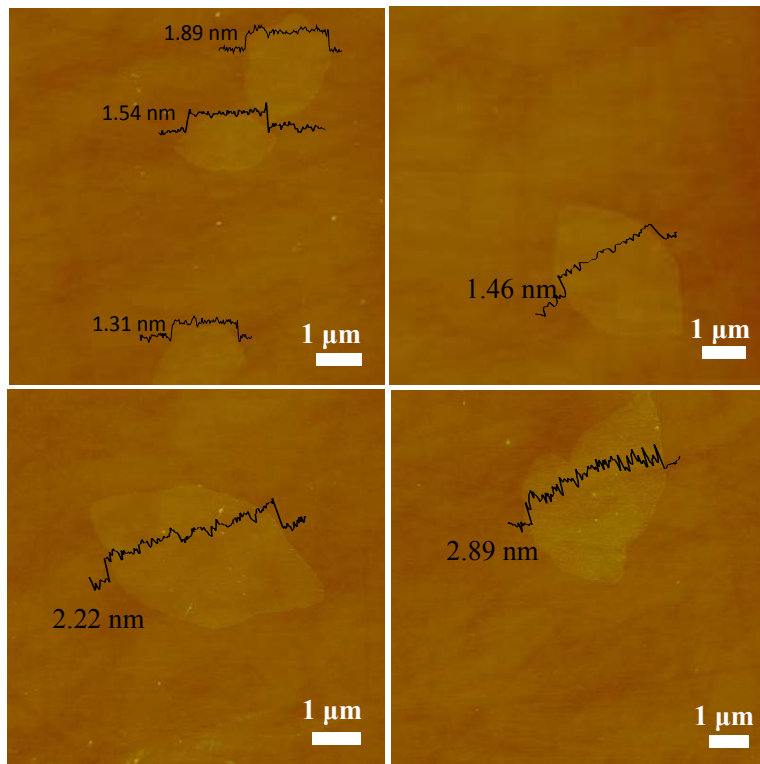
²University of Chinese Academy of Sciences, 19A Yuquan Road, Beijing, 100049, People's Republic of China.

³Institute of Physics, Chinese Academy of Sciences, Beijing 100190, People's Republic of China.

*e-mail: lwliu2007@sinano.ac.cn

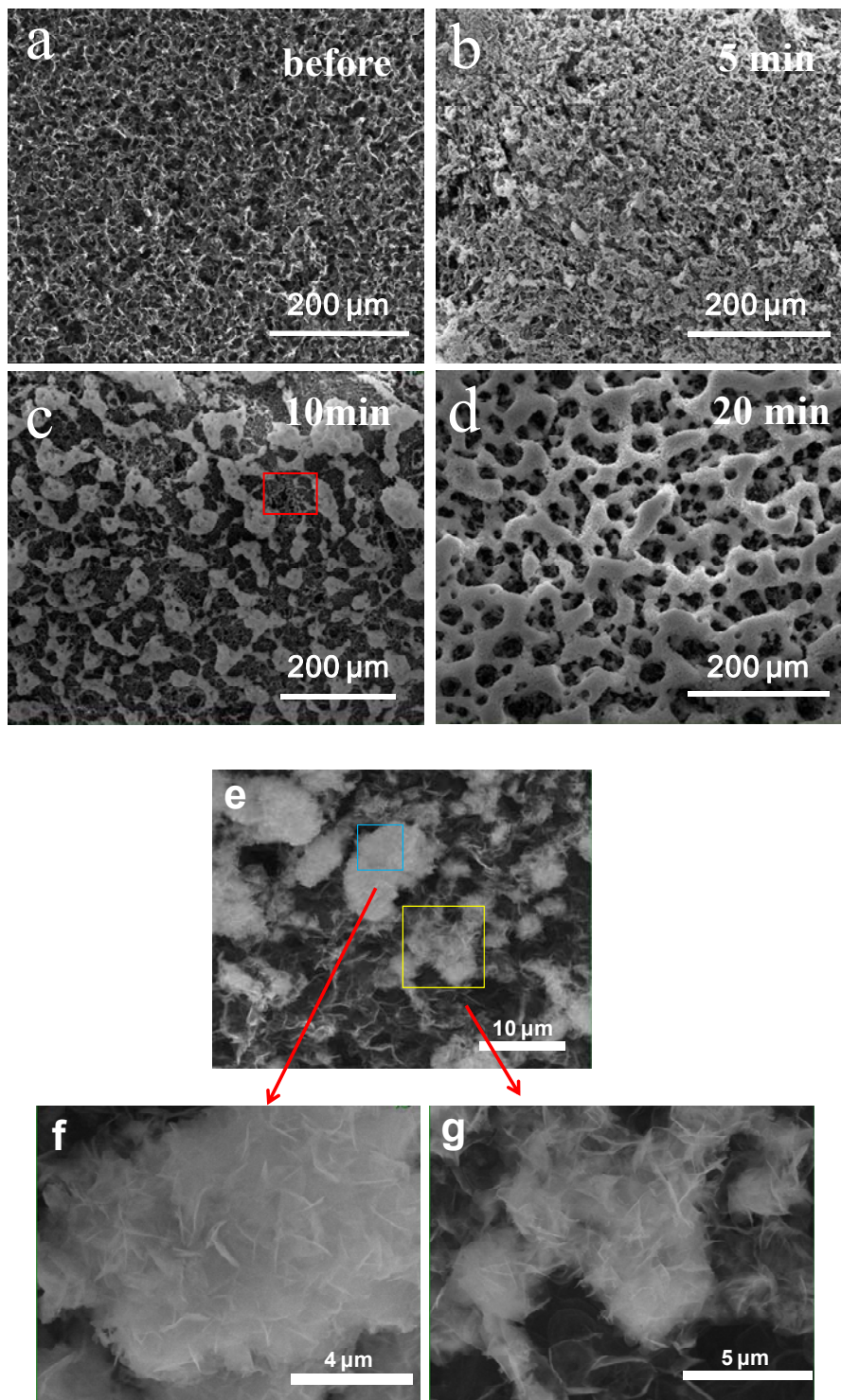


Supplementary Figure S1 | SEM images of the 3D-GMOs grown at different temperatures. Ni skeletons are removed (down) and not removed (up). The Ni particles are prone to cross-link and the quality of 3D-GMOs is gradually improved with increasing the growth temperature. When the growth temperature is lower to 700 °C, the SEM image shows that there is dominantly amorphous carbon, which is consistent with the Raman spectrum grown at 700 °C.



Supplementary Figure S2 | AFM images of the graphene layers of a 3D-GMO.

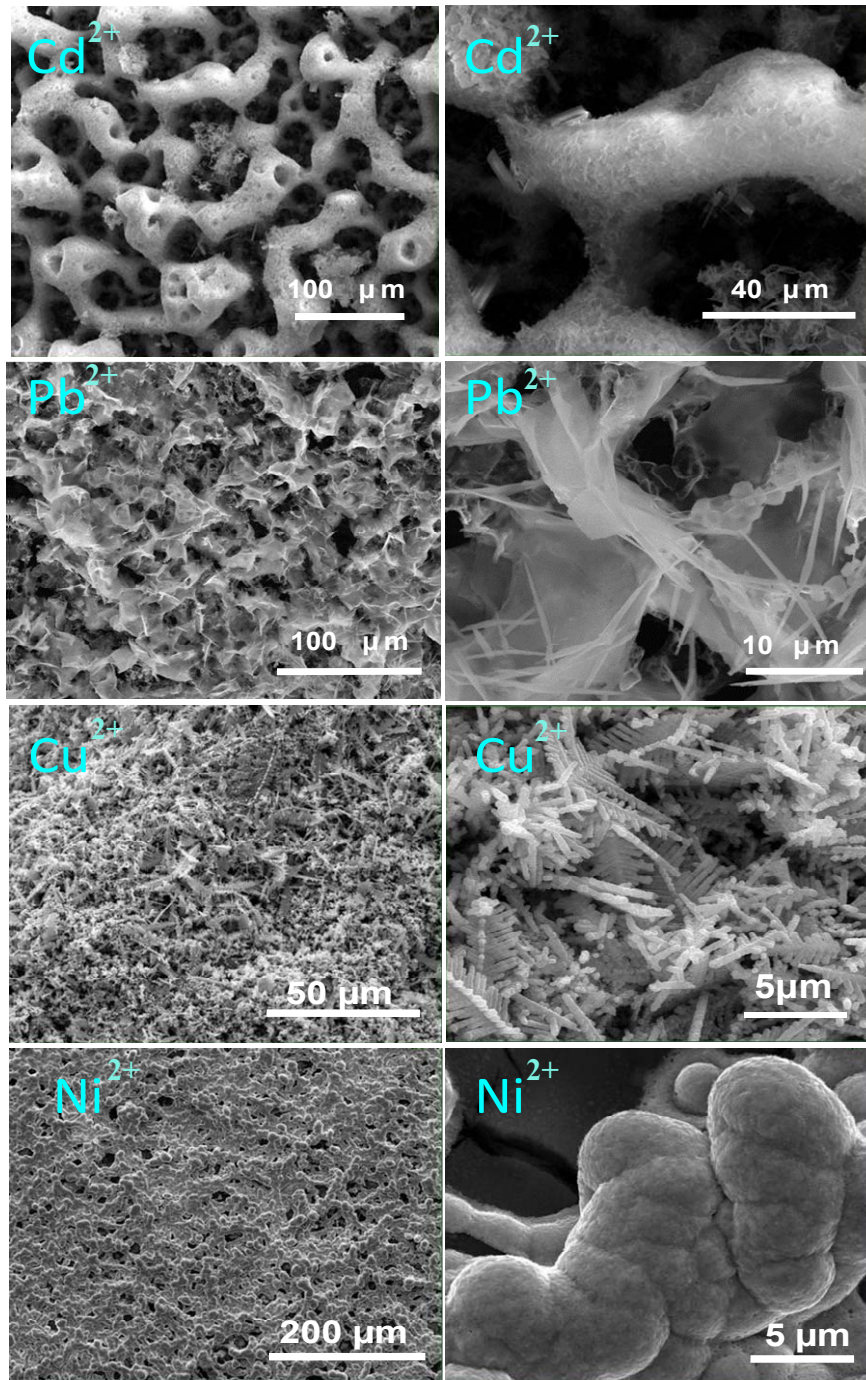
AFM images show the height profile of the surface morphology of the graphene layers, and the step height are less than 3 nm, corresponding to the dominated range of graphene layer numbers of 1-7 L.



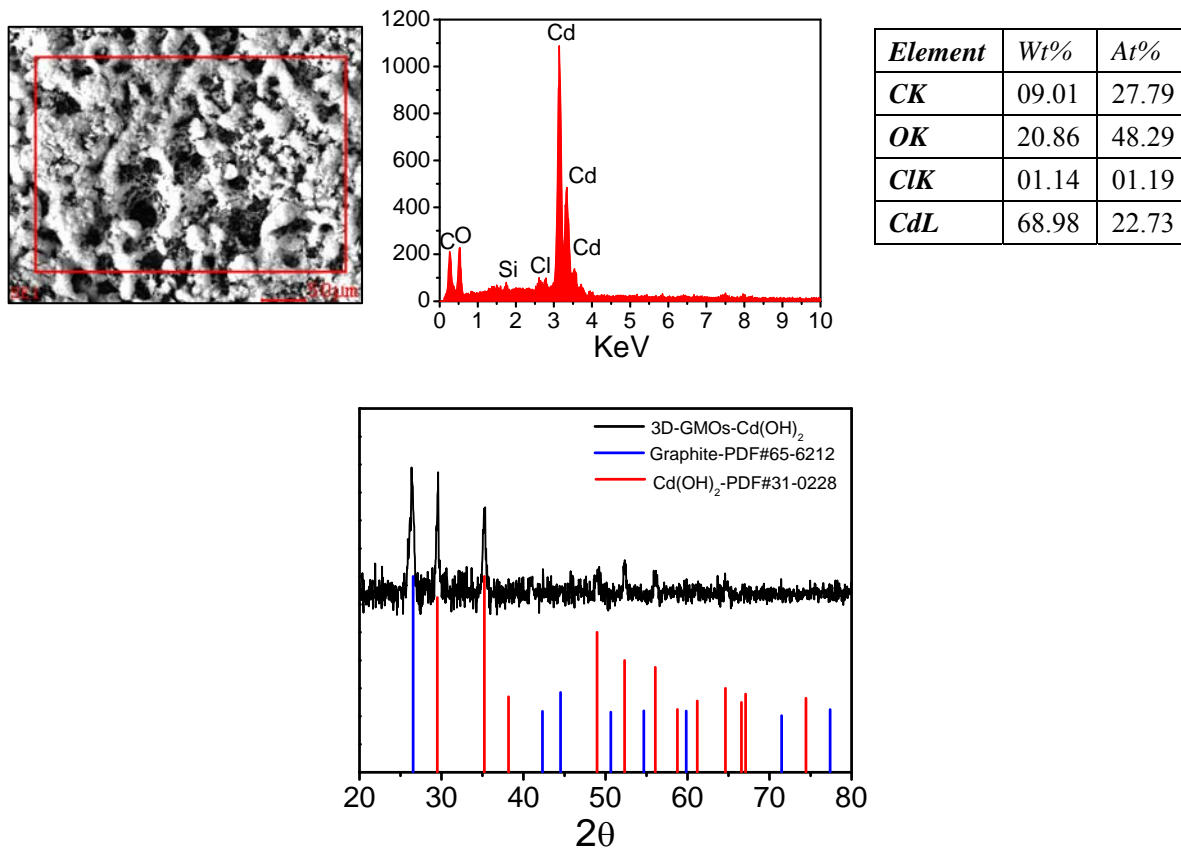
Supplementary Figure S3 | SEM images of 3D-GMOs before and after electrolytic deposition removing Cd²⁺. (a) SEM image of the as-prepared 3D-GMOs.

SEM images of 3D-GMOs after electrolytic deposition removing Cd^{2+} for (b) 5 min, (c) 10 min and (d) 20 min, respectively. (e) SEM images of 3D-GMOs after electrolytic deposition removing Cd^{2+} for 10 min in the red rectangle area of Figure c. (f, g) Magnified SEM images of the blue rectangle area and yellow rectangle area in Figure e, respectively, which show the graphene layers of 3D-GMOs covered with thin layers of deposited products.

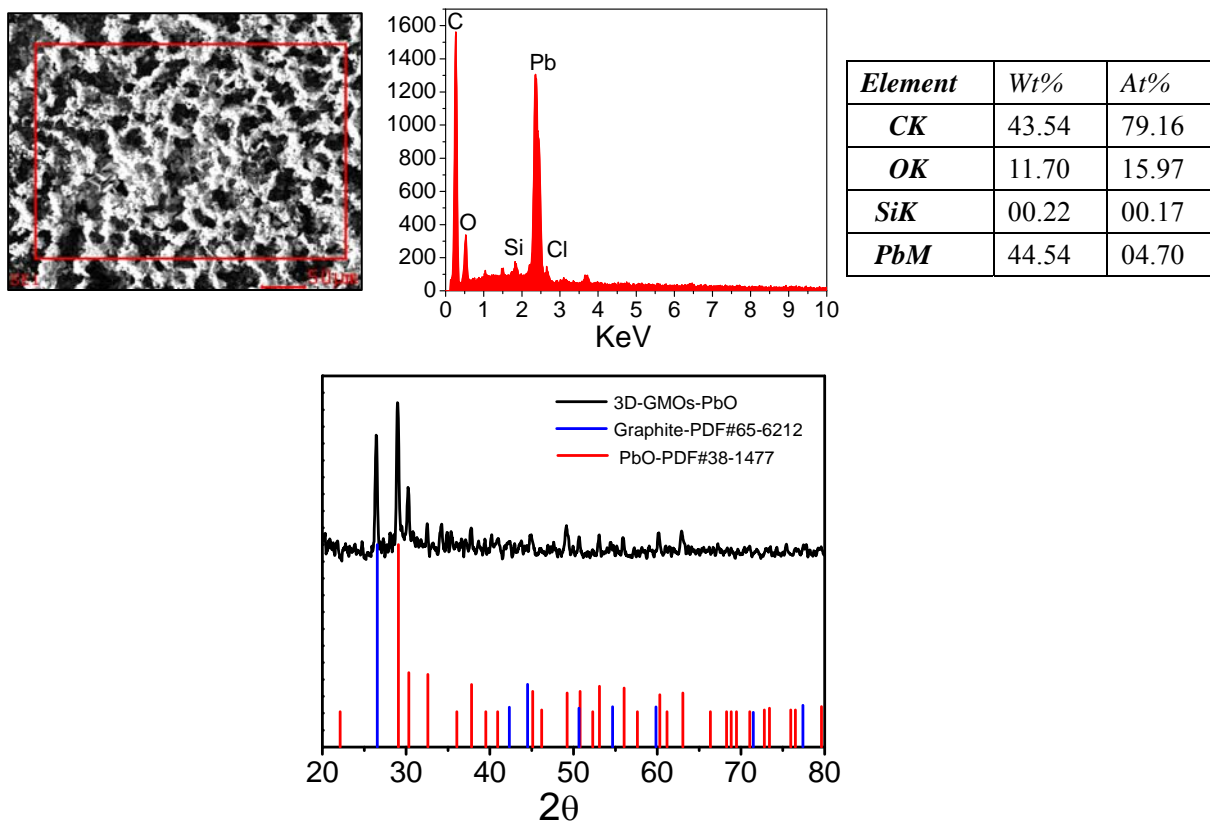
Initially, the honeycomb-like 3D-GMOs provide the large-area templates for deposition. At the beginning, the graphene layers are covered by the thin layer of deposited products. With the deposition time increasing, the covered layer of the deposited products become thicker to form 3D porous layers, which continuously offers the 3D porous template for the subsequent electrolytic deposition.



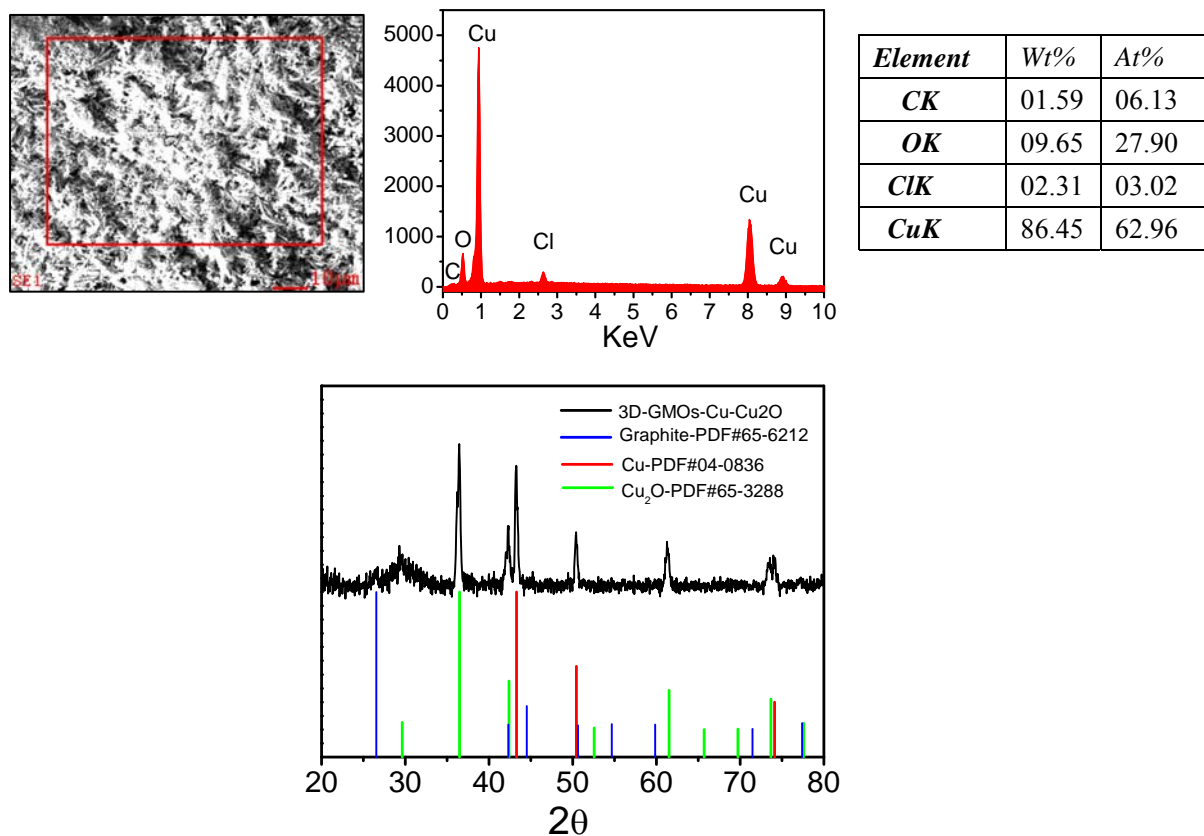
Supplementary Figure S4 | SEM images of 3D-GMOs after electrolytic deposition removing heavy metals ions. SEM images with different magnifications of the electrolytically deposited products of Cd²⁺, Pb²⁺, Cu²⁺ and Ni²⁺ (from top to down) on 3D-GMOs for 20 min, respectively. The products deposited on 3D-GMOs show 3D porous structures.



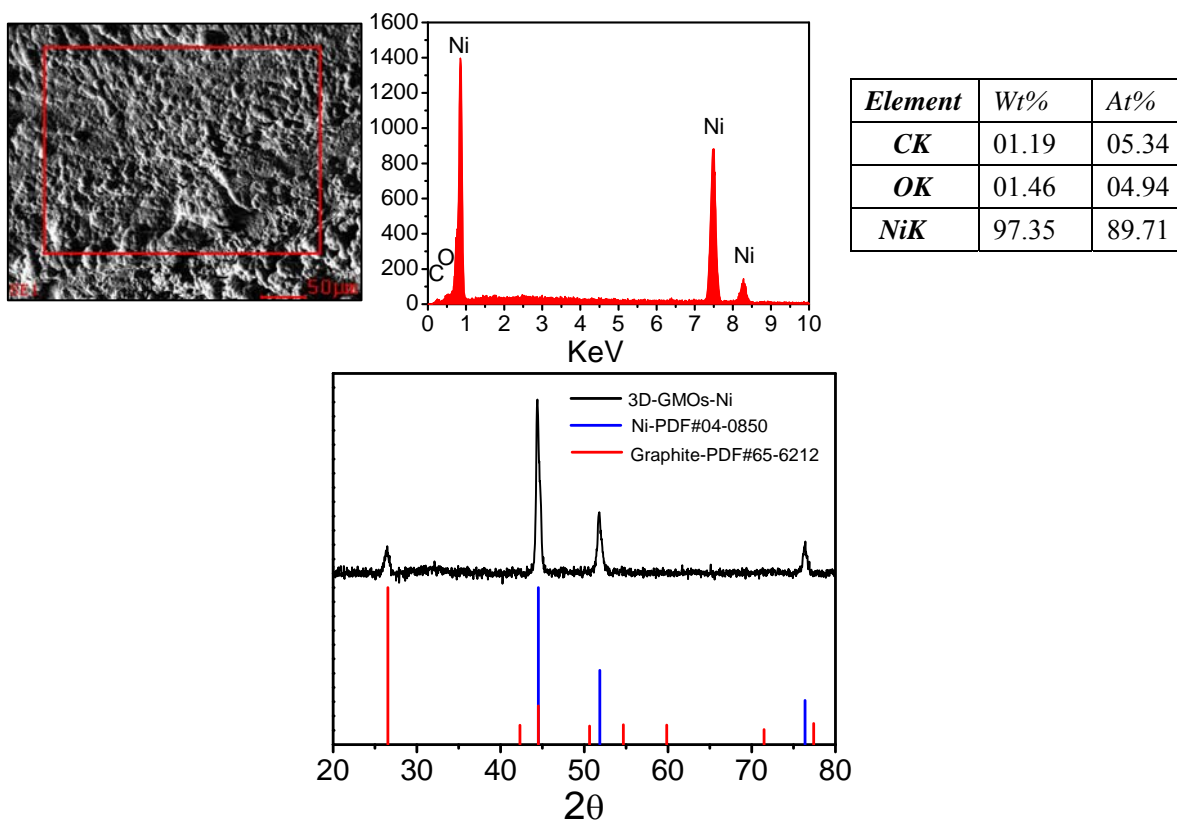
Supplementary Figure S5 | EDX spectrum (up) and XRD pattern (down) of the deposited product of Cd^{2+} for 20min. EDX shows that elements of the selected area are mainly C, O, Cd and Cl, respectively, and the proportions of the elements are list in the upper right table. XRD pattern further confirms that the main product deposited on the 3D-GMO is $\text{Cd}(\text{OH})_2$.



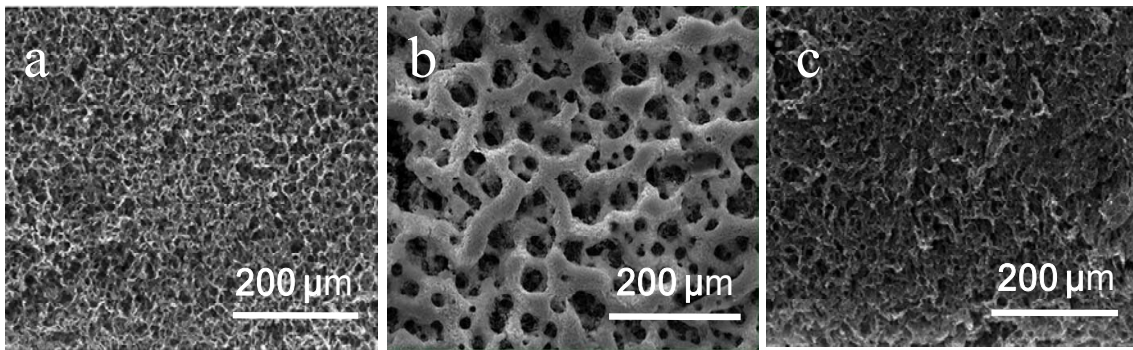
Supplementary Figure S6 | EDX spectrum (up) and XRD pattern (down) of the deposited product of Pb^{2+} for 20min. EDX shows that elements of the selected area are mainly C, O, Pb, respectively, and the proportions of the elements are list in the upper right table. XRD pattern further confirms that the main product deposited on 3D-GMO is PbO.



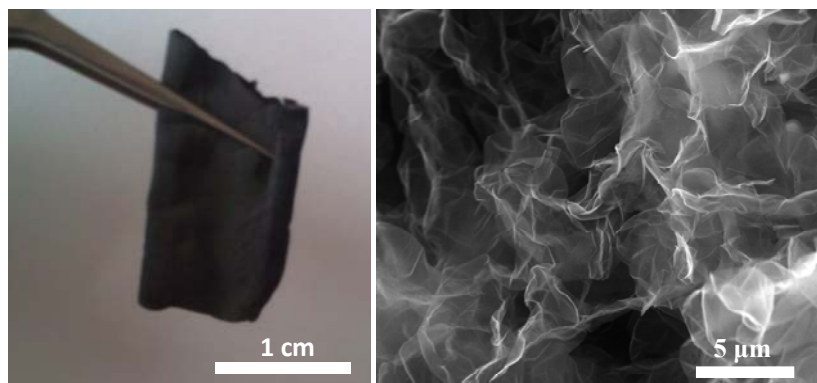
Supplementary Figure S7 | EDX spectrum (up) and XRD pattern (down) of the deposited product of Cu^{2+} for 20min. EDX shows that elements of the selected area are mainly C, O, Cu and Cl, respectively, and the proportions of the elements are list in the upper right table. XRD pattern further confirms that the main products deposited on 3D-GMO are the mixture of Cu_2O and Cu.



Supplementary Figure S8 | EDX spectrum (up) and XRD pattern (down) of the deposited product of Ni^{2+} for 20min. EDX shows that elements of the selected area are dominantly Ni and a small amount of C and O, respectively, and the proportions of the elements are list in the upper right table. XRD pattern further confirms that the product deposited on the 3D-GMO is Ni metal.



Supplementary Figure S9 | The recovery performance of the deposited products of Cd²⁺ on 3D-GMOs. SEM images of 3D-GMOs before (a) and after electrolytic deposition removing Cd²⁺ for 20 min (b) followed by the desorption process for 1 min (c). Compared with Figure b, the 3D deposited product almost disappeared in Figure c(a desorption efficiency >96%).



Supplementary Figure S10 | The morphology characterization of 3D-GMOs preparation by using the recycled NiCl_2/HCl as the catalyst precursor. Optical image (left) and SEM (right) image of 3D-GMOs after etching the recycled Ni templates by 3M HCl solution. The morphology of the 3D-GMOs grown on the recycled Ni templates is in accordance with that grown on the fresh Ni templates.

Table S1 | Maximum adsorption capacity of related heavy metal ions on graphene based absorbents.

Adsorbents	Heavy Metal ions	Adsorbtion Capacity (mg/g)	Reference
EDTA-Graphene Oxide	Pb ²⁺	479 ± 46	1
Graphene-c-MWCNT hybrid aerogel	Pb ²⁺	104.9	2
	Cu ²⁺	33.8	
Few-Layered Graphene Oxide Nanosheets	Cd ²⁺	106.3	3
Folding/aggregation of graphene oxide	Cu ²⁺	46.6	4
Ordered porous chitosan-gelatin/graphene oxide monoliths	Pb ²⁺	99	5
	Cu ²⁺	130	
Polydopamine-Functionalized Graphene Hydrogel	Pb ²⁺	336.32	6
	Cd ²⁺	145.48	
Functionalized graphene	Pb ²⁺	406.6	7
	Cd ²⁺	73.42	
3D graphene/R-FeOOH hydrogel	Pb ²⁺	373.8	8
Low-Temperature Exfoliated Graphene Nanosheets	Pb ²⁺	40	9
Few-layered graphene oxide	Pb ²⁺	842 (293K)	10
		1150 (313K)	
		1850 (333K)	
2- or 3-layered graphene		400	
Flower-like TiO ₂ -graphene oxide	Pb ²⁺	65.6± 2.7	11
	Cd ²⁺	72.8±1.6	
GO/Fe ₃ O ₄	Cu ²⁺	18.26 (293K)	12
	Pb ²⁺	882	
	Cd ²⁺	434	
	Cu ²⁺	3820	
3D-GMOs	Ni ²⁺	1683	This work
GNS/δ-MnO ₂	Ni ²⁺	46.55	13

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