Metal Ion Sensing Using ion Chemical Exchange Saturation Transfer-¹⁹F Magnetic Resonance Imaging

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Supporting Information

Experimental Section

5F-BAPTA (5,5'-difluoro BAPTA): The difluoro derivative of the tetra potassium salt of [1,2,-bis(o-aminophenoxy)ethane-N,N,-N',N', tetra-acetic acid], 5F-BAPTA, was purchased from Biotium, Inc. (Hayward, CA, USA).

Sample preparation: 5F-BAPTA was dissolved in Hepes buffer (40 mM) to a final concentration of 10 mM and the pH was adjusted to the following values: 5.6, 6.0, 6.4, 6.8, 7.0, 7.2, and 7.6. Stock 10 mM solutions of CaCl₂, MgCl₂, and ZnCl₂ were prepared in 40 mM Hepes buffer. Five μ L of the stock solution was added to 1 mL of 10 mM 5F-BAPTA resulting in a 200:1 ratio (10 mM: 50 μ M) between the free 5F-BAPTA and the M²⁺ ion following by a pH adjustment. One mL of each sample (with adjusted pH) was transferred into a 8 mm NMR tube within a 25 mm NMR tube in order to center the sample in the coil for the MRI experiments.

¹⁹*F NMR experiments*: ¹⁹F-NMR spectra were acquired with 11.7 T NMR scanner (Bruker) with a dedicated ¹⁹F coil (470 MHz). Samples contained 5F-BAPTA (5.0 mM) M^{2+} (Ca²⁺, Mg²⁺ or Zn²⁺, 0.5 mM), 5-Fluoro-Cytosine (5-FC, 0.5 mM) and D₂O (10%) that was used for signal lock. 5FC was assigned as an internal reference with a fixed frequency of -47.0 ppm. Interestingly, the data show that the frequency of free 5F-BAPTA is affected by pH, while that of bound 5F-BAPTA is not. Experiments were run at 25, 31, and 37°C to assess the effect of temperature on the frequency offsets of free 5F-BAPTA and [Ca²⁺-5F-BAPTA]. At 37°C, additional tubes containing K⁺ (120 mM) or K⁺

+ Mg²⁺ (120 mM + 1 mM, respectively) were also tested to determine the effect on $\Delta \omega$ of ions fast exchanging with 5F-BAPTA.

MRI: MRI experiments were performed on a vertical 16.4 T scanner (Bruker Avance system) at 37°C. A 25 mm birdcage radiofrequency coil was used to acquire both ¹H and ¹⁹F MR images by sweeping the coil frequency from proton (700 MHz) to fluorine (658.8 MHz) frequency.

¹*H-MRI*: A RARE sequence was used to acquire the ¹*H-MR* images with the following parameters: TR/TE=5,000/7.7 ms, RARE factor=8, 1 mm slice thickness, FOV=4×4 cm, matrix size=128×128, resolution=0.3125×0.3125 mm, and 1 average (NA=1).

¹⁹*F-iCEST experiments*: A modified RARE sequence (TR/TE=4,000/3.4 ms, RARE factor=4, 10 mm slice thickness, FOV=4×4 cm, matrix size=32×32, resolution=1.25×1.25 mm, and NA=8) including a magnetization transfer (MT) module (B₁=3.6 μ T) was used to acquire CEST-weighted images from -7.2 to +7.2 ppm around the resonance of the ¹⁹F atoms at the free 5F-BAPTA, which was assigned as 0 ppm. Saturation time (t_{sat}) was either 1500 ms to 2000 ms as indicated in the text.

¹⁹*F*-*iCEST of 500 nM of Ca*²⁺: An aqueous solution containing 0.5 mM of 5F-BAPTA and 500 nM in Hepes buffer (40 mM, pH=7.2) was transferred into a 20 mm NMR tube, which was located within a 25 mm NMR tube in order to centralize the sample in the coil for the MRI experiment. The same parameters used for ¹⁹F-iCEST experiments were used except the followings: TR=3000 ms, RARE=8, FOV=6.0 cm (resolution=1.875×1.875 mm), and 168 averages

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Magnetization Transfer Ratio (MTR) Images, iCEST images: The saturation transfer effect on the free 5F-BAPTA, i.e., the MTR or iCEST, was calculated for each voxel in the image by using a Lorentzian line shape fitting as previously described^{1,2}. The mean MTR values were calculated from a ROI for each for each testing tube after B_0 correction for each voxel. Error bars represent the inter-voxel standard deviations in each sample.

Relaxation times: The same image geometry as in the ¹⁹F-CEST experiments was used for the determination of T_1 and T_2 of the imaged samples. For T_1 measurements, a saturation recovery experiment was performed with TR values of 61, 214, 395, 615, 899, 1296 1964, and 4961 ms. A spin-echo experiment (TR=5000 ms) with multiple echoes was performed with variable echo times (TE=5, 10 15, 20, 25, 30, 35, and 40 ms) to determine the T_2 value of each sample.

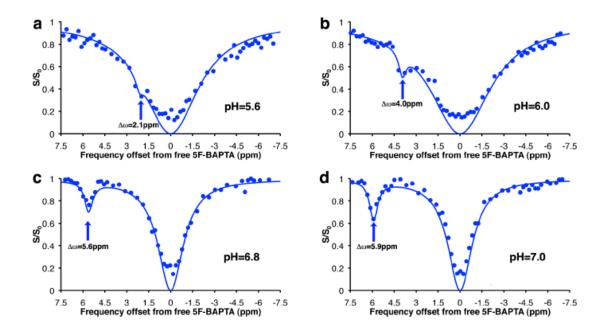
Bloch equation simulations: Numerical solutions to the six Bloch equations including direct saturation of free 5F-BAPTA were obtained as described previously³. The relaxation parameters for ¹⁹F used in the Bloch equations were $R_1 = 1/T_1$ with bound calcium $(R_{1b}) = 0.71$ s⁻¹, R_2 bound calcium $(R_{2b})= 29$ s⁻¹, whereas the R_1 and R_2 values of free 5F-BAPTA (R_{1f} and R_{2f}) were determined experimentally through inversion-recovery and saturation-recovery experiments as a function of pH on solutions containing 20 mM 5F-BAPTA as listed in Table S1.

iCEST Sensitivity: The validity of the ¹⁹F-iCEST approach is based on the detectability of the free 5F-BAPTA with ¹⁹F-MRI. As shown in Figure 4 (main text), 0.5 mM of 5F-BAPTA is detected at a lower spatial resolution compared to ¹H-MRI. iCEST images

overlaid on top of high resolution proton images will allow detailed spatial information to localize sources of iCEST signal. In addition, ¹⁹F-MRI based approaches do not demand high ¹⁹F signal-to-noise (SNR) due to the fact that any ¹⁹F signal detected originates from the fluorinated compounds, as the endogenous ¹⁹F content in biological tissues is negligible.⁴ Additionally, by acquiring imaging at $+\Delta\omega$ and $-\Delta\omega$ (but not the whole Z-spectrum as used in this study), one could increase the number of averages and improve the SNR. Each image in the data set used for obtaining the Z-spectrum at Figure 4b was obtained in ~15 min. Therefore, by acquiring a dataset that consists of $+\Delta\omega$ and $-\Delta\omega$ images, one can obtain an iCEST effect from 500 nM Ca²⁺ (10% iCEST contrast) within 30 min with the experimental setup used (i.e., 0.5 mM of free 5F-BAPTA). Published results showed no toxic effect on cells when they were loaded with more than 1 mM of 5F-BAPTA⁵. Thus, our iCEST is a valid approach, in particular when these higher concentrations of 5F-BAPTA (1 mM) can be used to reduce the experimental scan time.

As an alternative, one may consider using the iCEST approach in a localized voxel-spectroscopy setup to improve SNR and reduce the acquisition time. This will allow the detection of even lower concentrations of the free 5F-BAPTA and [Ca²⁺-5F-BAPTA, as shown before for 5-Fluoro-Uracil (5FU), which can be detected at micro-molar concentrations in vivo.⁶

In principle, the experimental acquisition parameters will determine the sensitivity of iCEST images.⁴ This includes (i) the ¹⁹F- agent T_1 and T_2 relaxation times which may be controlled by chemical probe design, (ii) improved image acquisition protocols such as SWIFT and UTE-based approaches adjusted to detect spins with short T_2 , than used previously for saturation transfer imaging^{7,8}, (iii) using MR scanners operating at high magnetic field strengths, and (iv) RF-coil design, e.g., the use of multichannel RF coils or cryogenic RF coil technologies.



Figures

Figure S1. pH dependency of ¹⁹F-CEST spectra of 5F-BAPTA: CEST-spectra of solutions containing 10 mM of 5F-BAPTA and 50 μ M of Ca²⁺ in 40 mM Hepes buffer at solutions with the pH adjusted to a) 5.6, b) 6.0, c) 6.8, and d) 7.0. Solid lines represent Bloch simulations (two pool model) and arrows point to the $\Delta\omega$ of the [Ca²⁺-5F-BAPTA] complex. Note that the resolution and number of averages for the Z-spectra (experimental points) is lower for these samples compared to those used in Figure 2 (main text). The purpose of this experiment was to demonstrate that $\Delta\omega$ could be determined from the Z-spectra. In addition, the SNR was low for this data, resulting in particularly poor quality measurements around saturation offsets of 0 ppm.

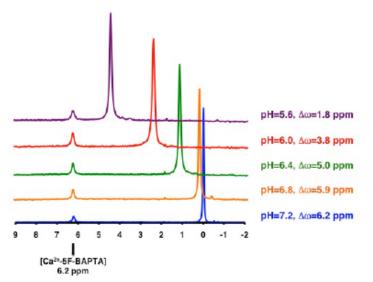


Figure S2. Chemical shift dependency of 5F-BAPTA as function of pH. ¹⁹F-NMR spectra of solutions containing Ca²⁺(0.5 mM), 5F-BAPTA (5 mM), and 5-Fluoro-Cytosine (5FC, 0.5 mM) as internal reference (set to -47 ppm) with 10% D₂O. The spectra were acquired at 470 MHz. The peak of 5-FC was calibrated at -47 ppm for convenience.

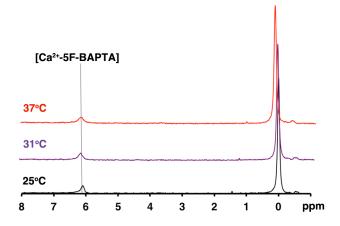


Figure S3. Effect of temperature on $\Delta \omega$ **.** ¹⁹F-NMR spectra of solutions containing Ca²⁺ (0.5 mM), 5F-BAPTA (5 mM), and 5-Fluoro-Cytosine (5FC, 0.5 mM) as internal reference (set to -47 ppm) in 10% D₂O. The spectra were acquired at 470 MHz. Increasing the temperature of the solution has no effect on the chemical shift offset of [Ca²⁺ -5F-BAPTA]. The resonance of the free 5F-BAPTA was found to be 0.0 ppm at 25°C, 0.12 ppm at 31°C, and 0.25 ppm at 37°C.

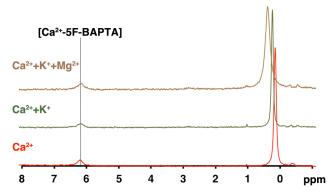


Figure S4. Effect of high concentration of fast exchanging ions (K^+ and Mg^{2+}) on $\Delta\omega$. ¹⁹F-NMR spectra of solutions containing Ca²⁺(0.5 mM), 5F-BAPTA (5 mM), and 5-Fluoro-Cytosine (5FC, 0.5 mM) as internal reference (set to -47 ppm) in 10% D₂O. The spectra were acquired at 470 MHz. Adding the fast exchanging ions K⁺ (120 mM) and K⁺+Mg²⁺(120 mM +1 mM) demonstrate a downfield shift in the resonance frequency of the free 5F-BAPTA at 0.32 ppm (for 120 mM K⁺) and 0.48 ppm (for 120 mM of K⁺ and 1 mM of Mg²⁺). The temperature was set to 37°C and the peak of 5-FC was calibrated at -47 ppm for convenience.

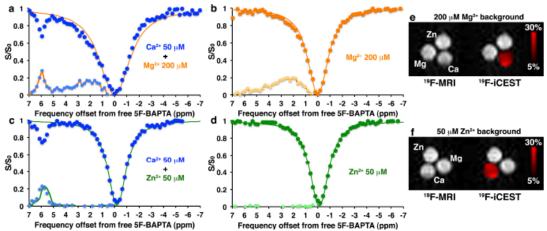


Figure S5. ¹⁹F-iCEST of Ca²⁺ in the presence of Mg²⁺ or Zn²⁺. ¹⁹F-iCEST Z-spectra of solutions containing a) Ca²⁺ (50 μ M) and Mg²⁺ (200 μ M), b) Mg²⁺ (200 μ M), c) Ca²⁺ (50 μ M), and d) Zn²⁺ (50 μ M) at pH=7.2, 37 °C, and 16.4 T. The MTR asymmetry (MTR_{asym}) plots of the background solutions are shown for 200 μ M Mg²⁺ (a, b, light blue and light orange) and for 50 μ M Zn²⁺ (c, d, light blue and light green). Solid lines represent Bloch simulations.

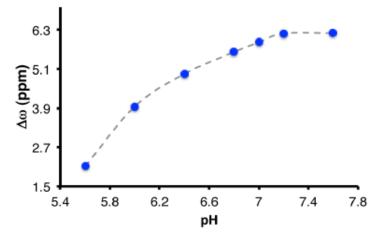


Figure S6. pH dependency of iCEST. The dependency of $\Delta\omega$ between Ca²⁺-bound and free 5F-BAPTA as obtained from iCEST Z-spectra. The phantom consisted of 8 mm NMR tubes containing 10 mM of 5F-BAPTA and 50 μ M of M²⁺ in 40 mM Hepes buffer (pH=5.6-7.6). ¹⁹F-iCEST data were acquired with B₁=3.6 μ T/2000 ms at 37 °C. The dependency of $\Delta\omega$ between Ca²⁺-bound and free 5F-BAPTA was obtained from ¹⁹F-iCEST Z-spectra

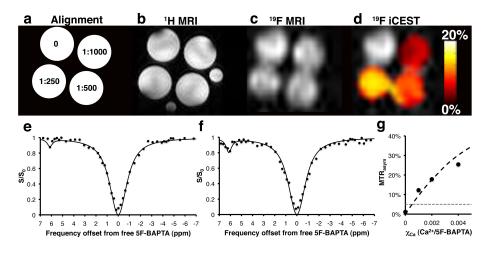


Figure S7. Sensing Ca²⁺ using iCEST. a) Alignment of four tubes containing 5 mM of 5F-BAPTA (pH=7.2) and different molar fractions (χ_{Ca} =1:250, 1:500, 1:1000, 1:0) between Ca²⁺ and 5F-BAPTA for the phantom (pH=7.2, 16.4 T, 37°C). b) ¹H-MR image. Small water tubes (shown on ¹H-MRI) were included to determine the orientation of the samples. c) ¹⁹F-MR image, and d) overlay of ¹⁹F-iCEST image ($\Delta\omega$ =6.2 ppm) on ¹⁹F-image. iCEST Z-spectra for a tube with e) χ_{Ca} =1:1000 and f) χ_{Ca} =1:500 samples. Solid lines represent Bloch simulations. g) Plot of χ_{Ca} vs. MTR_{asym} for iCEST data acquired at B₁=3.6 µT, t_{sat}=1.5 s. A 5% threshold is shown as a gray dashed line, which is reached at χ_{Ca} =1:2000.

pН	$\Delta \omega$ (ppm)	T_2 (ms)	T_1 (ms)
5.6	2.1	6 ^a	715
6.0	4.0	10	715
6.4	5.0	11	715
6.8	5.6	25	715
7.0	5.9	32	715
7.2	6.2	38	715
7.6	6.2	N.D	715

Table S1. $\Delta \omega$ and T₂ dependency on pH

^a Estimated from the Bloch equation fitting ND = Not Determined

N.D. = Not Determined

References

(1) Jones, C. K.; Polders, D.; Hua, J.; Zhu, H.; Hoogduin, H. J.; Zhou, J.; Luijten, P.; van Zijl, P. C. *Magn Reson Med* **2012**, *67*, 1579.

(2) Liu, G.; Li, Y.; Sheth, V. R.; Pagel, M. D. Mol Imag 2012, 11, 47.

(3) McMahon, M. T.; Gilad, A. A.; Zhou, J.; Sun, P. Z.; Bulte, J. W. M.; van Zijl, P. C. M. *Magn Reson in Med* **2006**, *55*, 836.

(4) Ahrens, E. T.; Zhong, J. NMR Biomed 2013. doi: 10.1002/nbm.2948

(5) Smith, G. A.; Hesketh, R. T.; Metcalfe, J. C.; Feeney, J.; Morris, P. G. *Proc Natl Acad Sci U S A* **1983**, 80, 7178.

(6) Martino, R.; Gilard, V.; Desmoulin, F.; Malet-Martino, M. J Pharm Biomed Anal 2005, 38, 871

(7) Soesbe, T. C.; Togao, O.; Takahashi, M.; Sherry, A. D. Magn Reson Med 2012, 68, 816.

(8) Du, J.; Takahashi, A. M.; Bydder, M.; Chung, C. B.; Bydder, G. M. Magn Reson Med 2009, 62, 527.