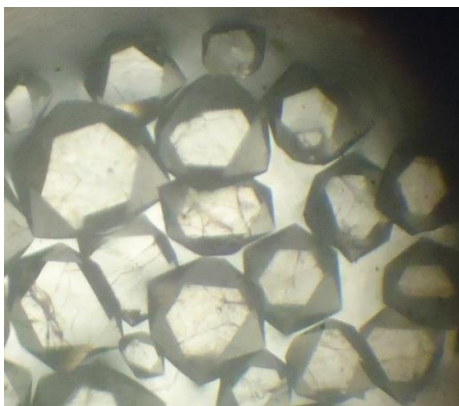
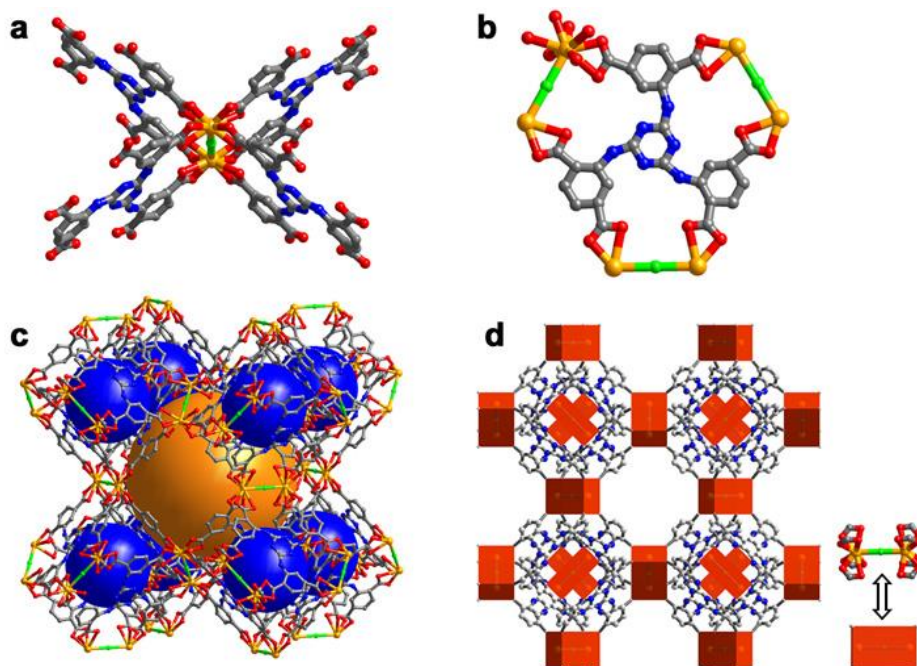


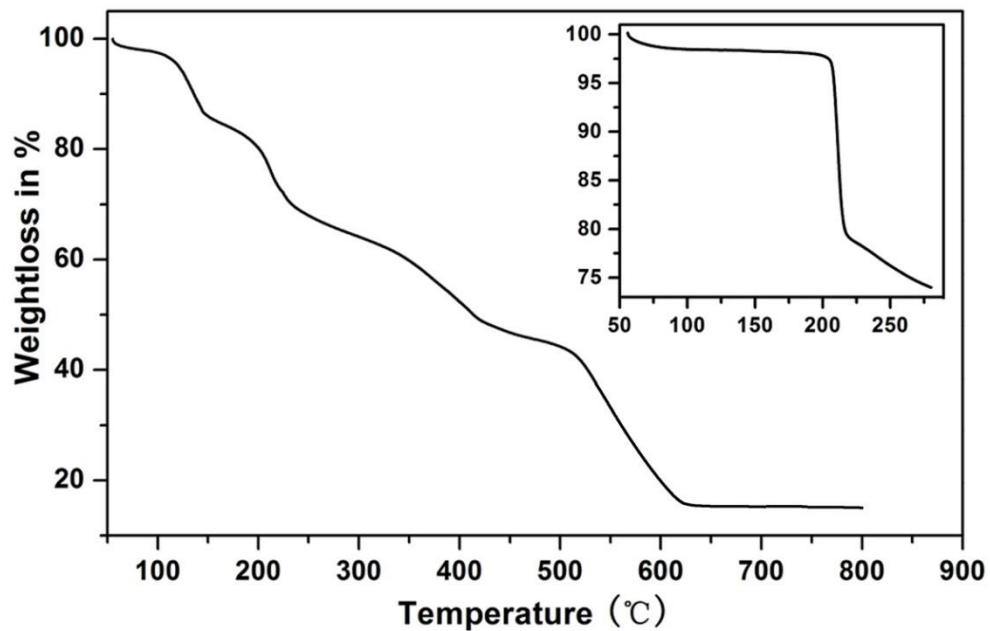
Supplementary Figure S1 Scheme of synthesis of the H₆TATPT ligand.



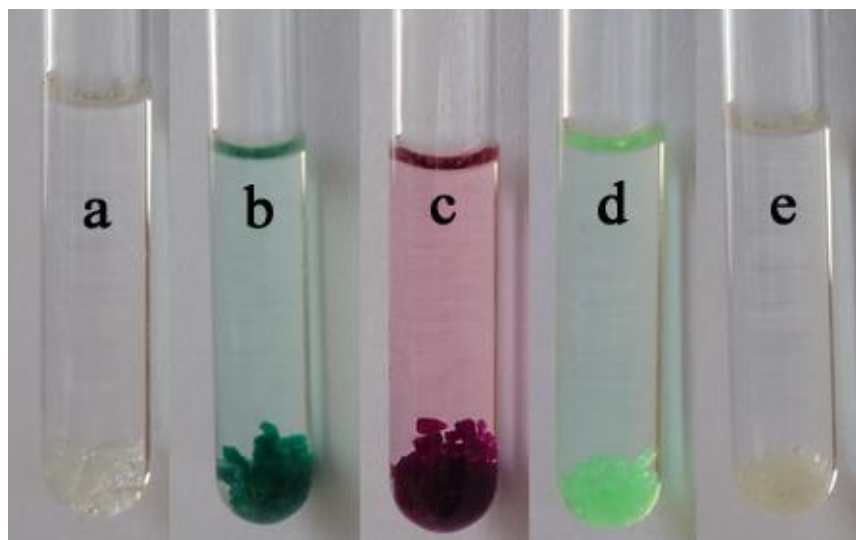
Supplementary Figure S2 Photograph of crystals of compound 1.



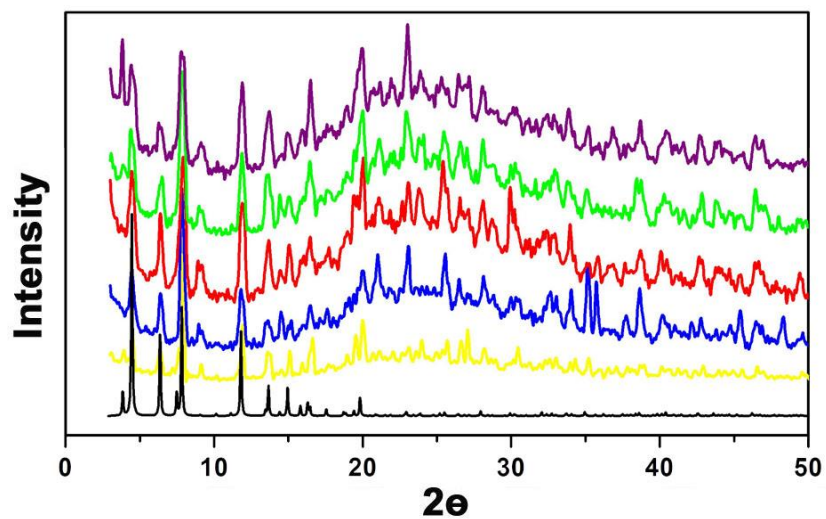
Supplementary Figure S3 Crystal structures of compound 1. (a) The coordination environment of Cd²⁺ ion and its connectivity with TATPT ligands, C gray, O red, N blue, Cl green and Cd orange. (b) Coordination mode of TATPT ligand. (c) Ball and stick representations show the 3D structure in **1** viewed along the *a*-axis. The orange and blue spheres represent the void inside the cages. (d) Polyhedral representations of the crystal structure viewed along a axis in which the cuboid represented the Cd₂(μ₂-Cl)(CO₂)₈ secondary building unit (SBU).



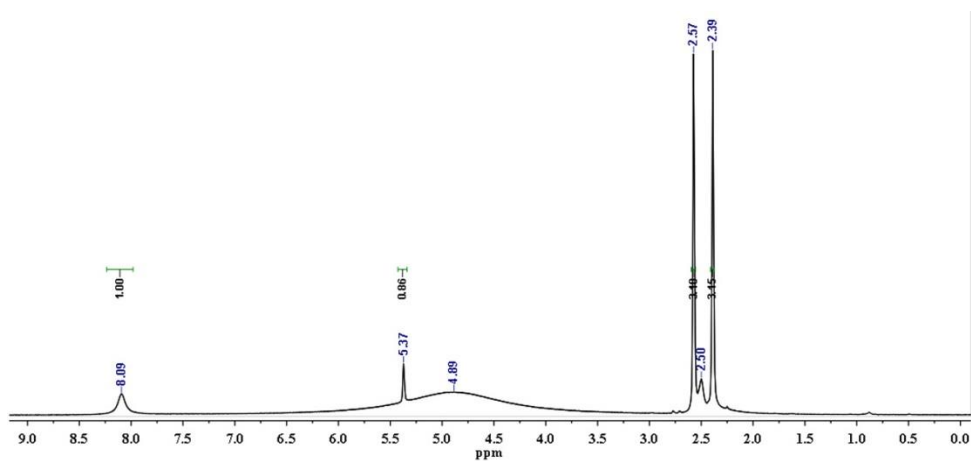
Supplementary Figure S4 TG profile. TG profile of compound **1** under nitrogen gas atmosphere with a heating rate of $10\text{ }^{\circ}\text{C}\cdot\text{min}^{-1}$. The inset is the TG profile of desolvated **1** from 50 to 280 $^{\circ}\text{C}$ under nitrogen gas with a heating rate of $2\text{ }^{\circ}\text{C}\cdot\text{min}^{-1}$.



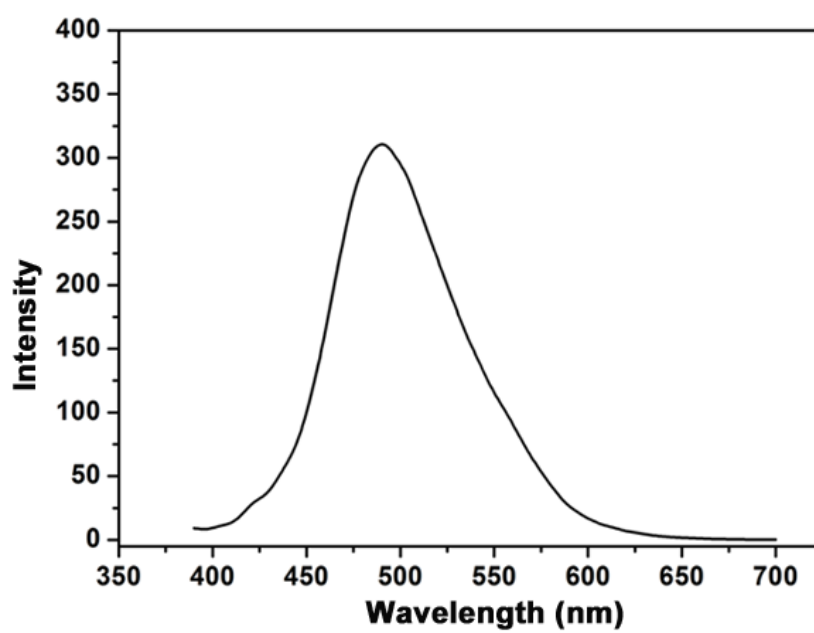
Supplementary Figure S5 Photographs of 1 and samples after ion exchange with transition metals. (a) As-synthesized 1. (b) Cu²⁺⊃1. (c) Co²⁺⊃1. (d) Ni²⁺⊃1. (e) Zn²⁺⊃1.



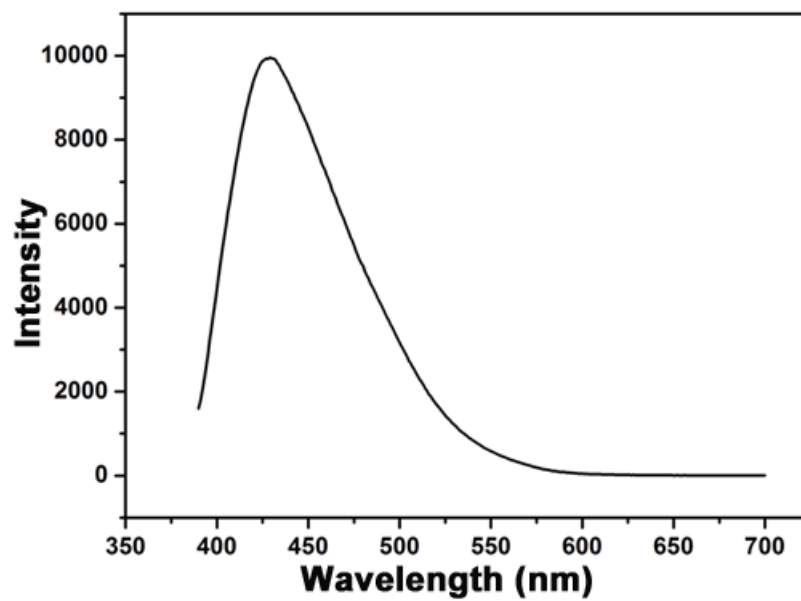
Supplementary Figure S6 PXRD patterns. PXRD patterns for **1** and transition metal ion exchanged **1**. simulated, black; as-synthesized **1**, yellow; Co²⁺⊃**1**, blue; Ni²⁺⊃**1**, red; Zn²⁺⊃**1**, green; Cu²⁺⊃**1**, purple.



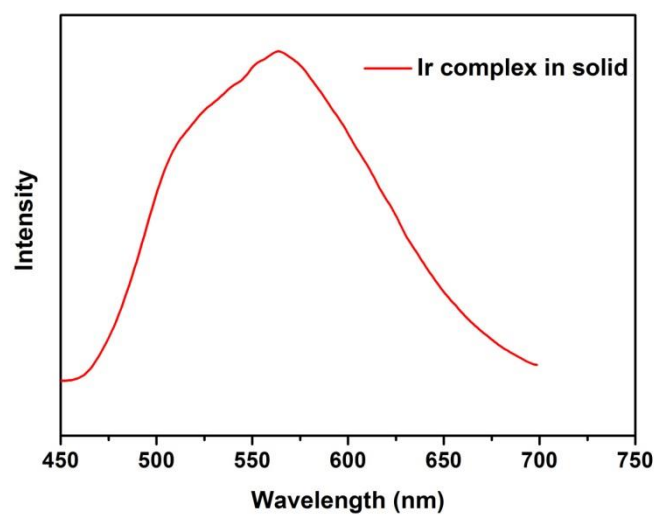
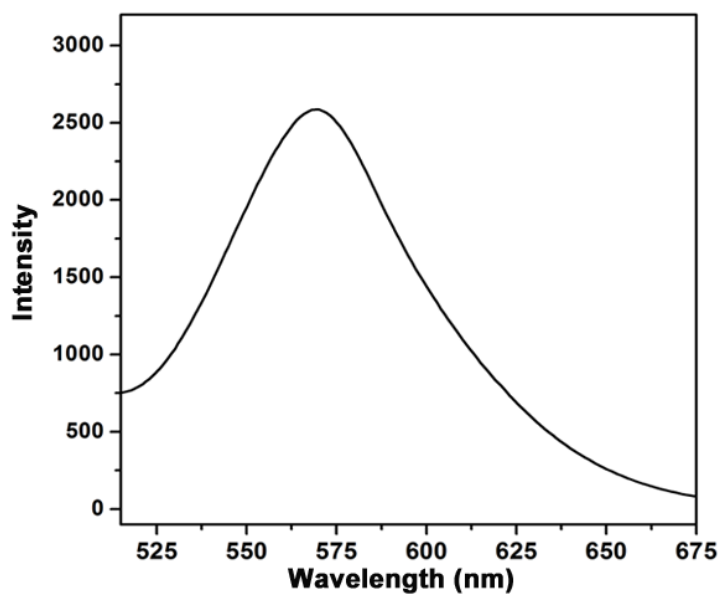
Supplementary Figure S7 ^1H NMR spectrum. The ^1H NMR spectrum in dms0-d_6 solution taken after ion exchange experiment.



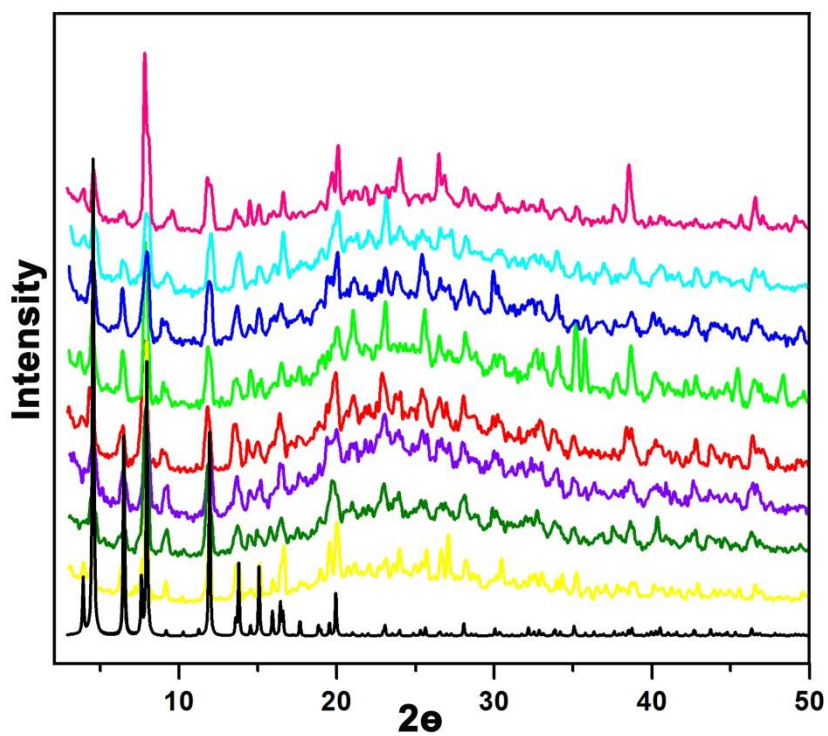
Supplementary Figure S8 photoluminescence spectrum of ligand. Solid photoluminescence spectrum of H₆TATPT ligand at room temperature ($\lambda_{\text{ex}} = 370$ nm).



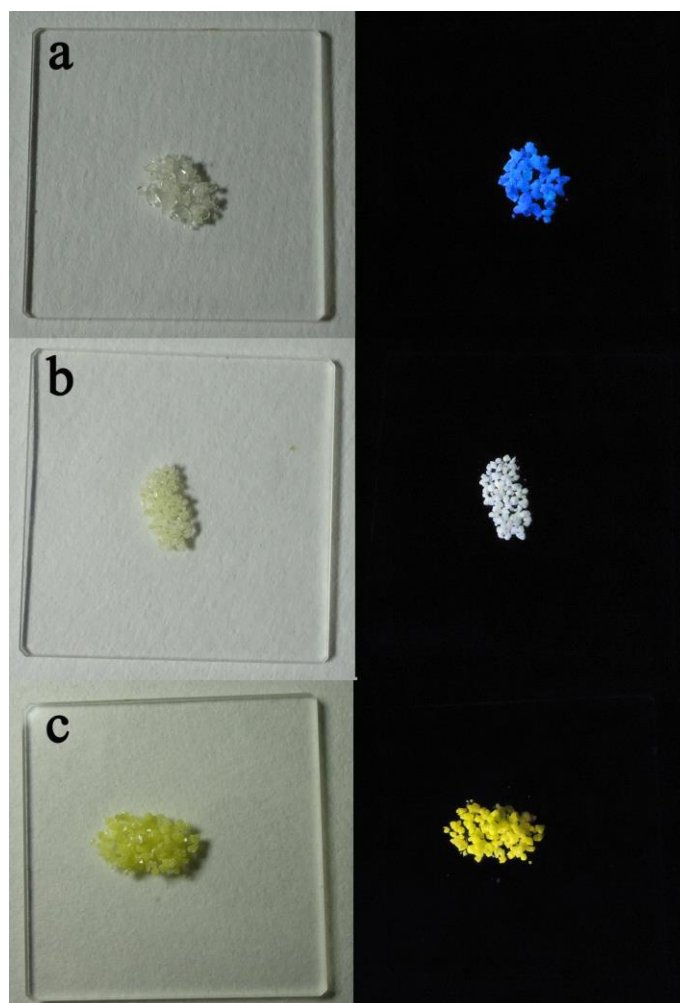
Supplementary Figure S9 photoluminescence spectrum of **1**. Solid photoluminescence spectrum of compound **1** at room temperature upon excitation at 370 nm.



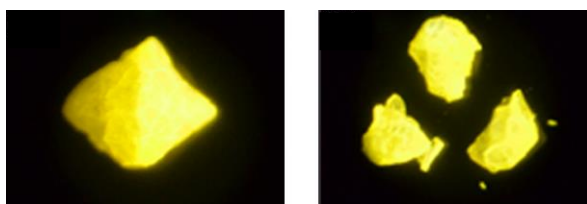
Supplementary Figure S10 photoluminescence spectra of [Ir(ppy)₂(bpy)][PF₆]. Room temperature photoluminescence spectra of [Ir(ppy)₂(bpy)][PF₆] in DMF solvent (top) and in solid state (bottom) upon



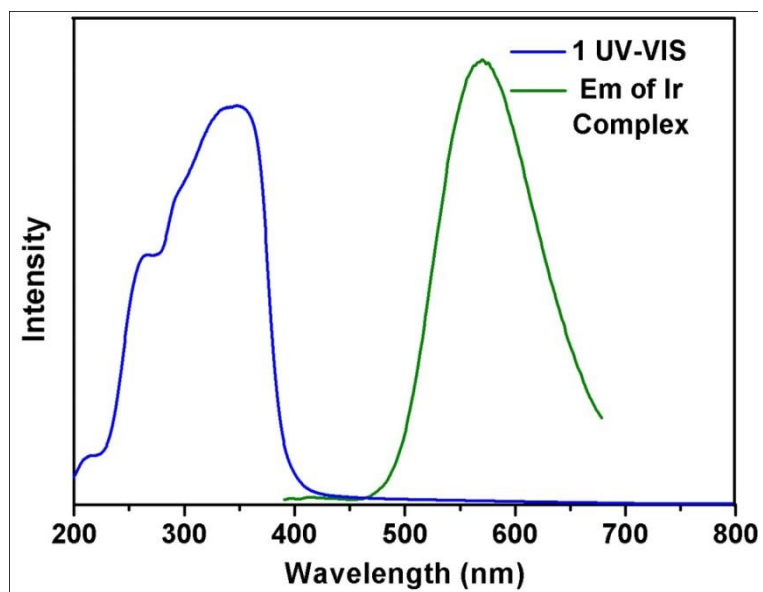
Supplementary Figure S11 XRD patterns. XRD patterns for **1** (simulated, black; as-synthesized**1**, yellow) and $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+$ encapsulated **1** with different concentrations of $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+$ (0.52wt%, olive; 1.04wt%, violet; 3.5wt%, red; 3.7wt%, green; 4.5wt%, blue; 7.5wt%, cyan; 8.8wt% pink).



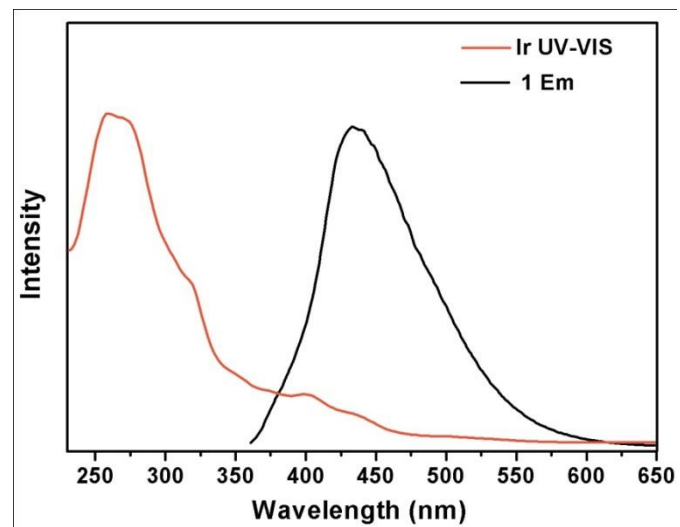
Supplementary Figure S12 Photographs of 1 and $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+@1$ (under natural light (left) and laboratory UV light (365 nm, right)). (a) As-synthesized 1. (b) 3.5wt% contained $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+@1$. (c) 8.8wt% contained $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+@1$.



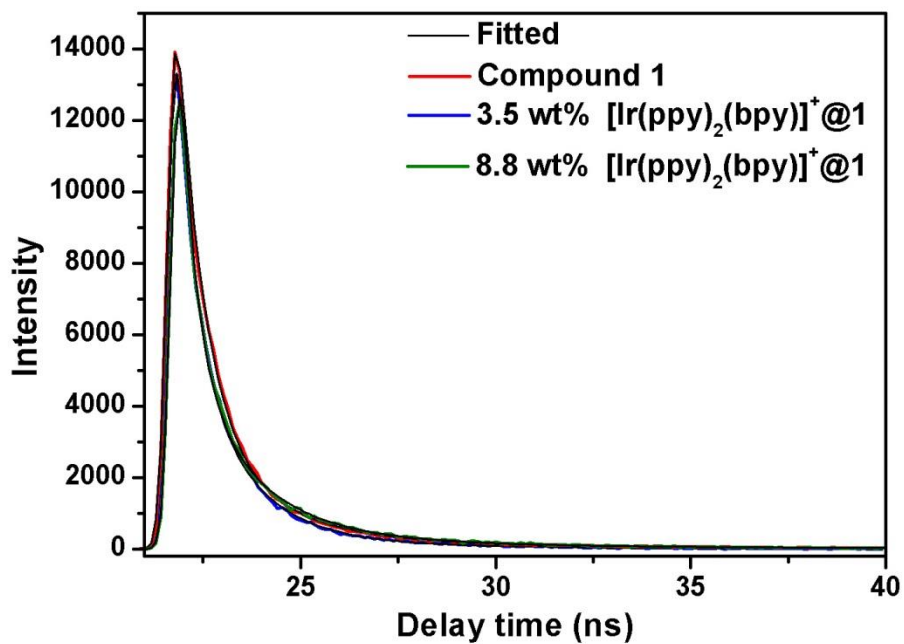
Supplementary Figure S13 Photographs of crystals. Photographs of a single crystal of a freshly grown $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+@1$ (left) and crushed fragments of the same crystal (right).



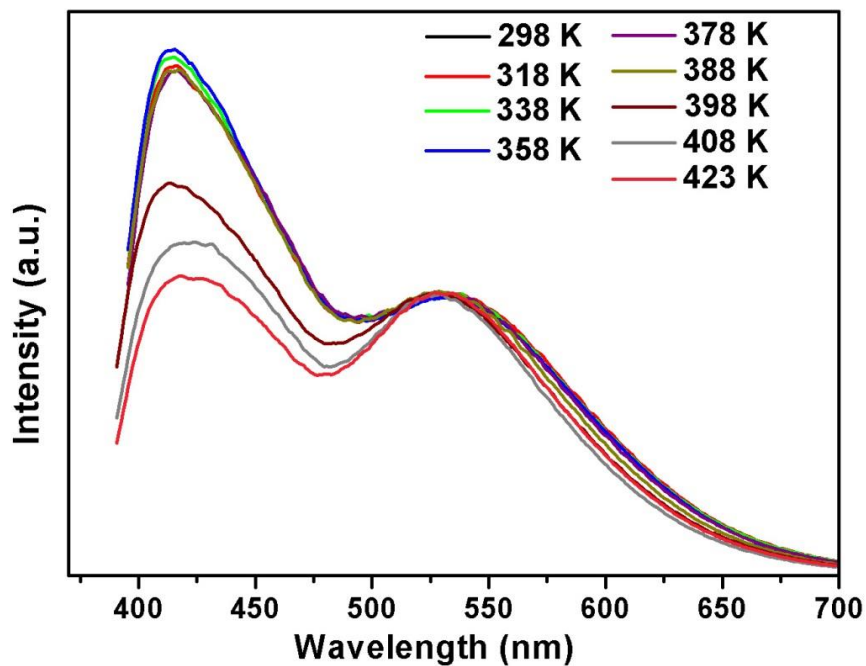
Supplementary Figure S14 Optical spectra. The absorption spectrum of **1** (blue), and emission spectrum of the $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+$ complex excited at 370 nm (green).



Supplementary Figure S15 Optical spectra. The absorption spectrum of the $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+$ complex (red), and emission spectrum of **1** excited at 340 nm (black).

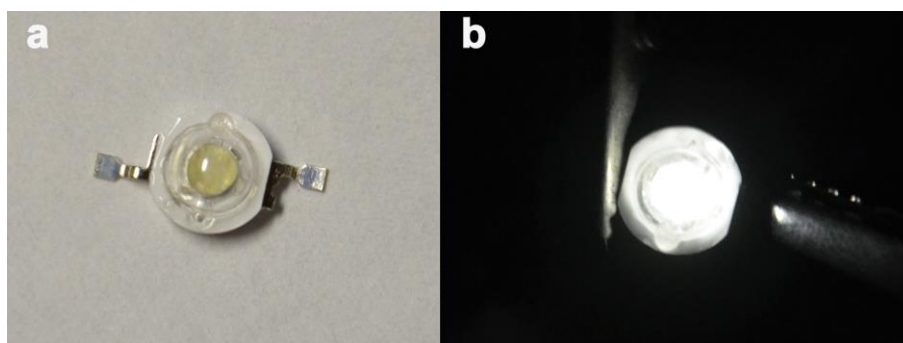


Supplementary Figure S16 Fluorescence decay profiles. Fluorescence decay profiles of **1** and [Ir(ppy)₂(bpy)]⁺@**1** at $\lambda_{\text{ex}} = 370$ nm and $\lambda_{\text{em}} = 425$ nm: black, after fitted; red, compound **1**; blue, 3.5% [Ir(ppy)₂(bpy)]⁺@**1**; green, 8.8% [Ir(ppy)₂(bpy)]⁺@**1**.

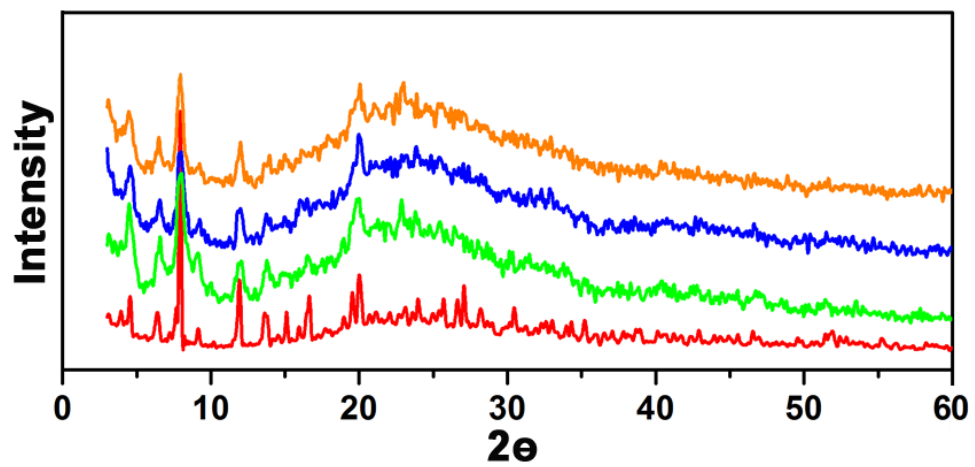


Supplementary Figure S17 Temperature-dependent photoluminescence spectra.

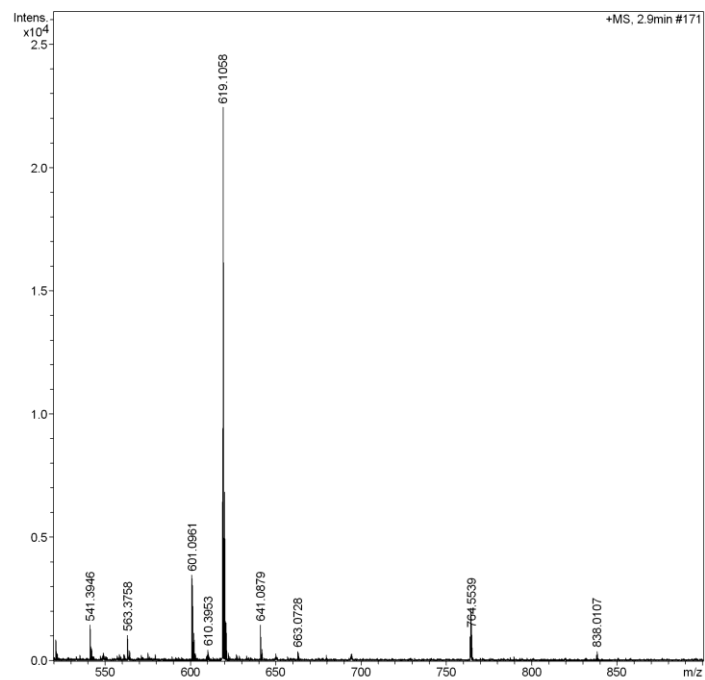
Photoluminescence spectra of white-emitting [Ir(ppy)₂(bpy)]⁺@1 at different temperatures: black, 298K; red, 318K; green, 338K; blue, 358K; violet, 378K; dark yellow, 388K; wine, 398K; gray, 408K; pink, 423K. All measurements were performed on solid samples at an excitation wavelength λ_{ex} of 370 nm.



Supplementary Figure S18 Photographs of the white LED fabricated using a UV InGaAsN LED chip and white phosphor of 3.8 wt% $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+@1$. (a) Without current. (b) After applying a forward current of 150 mA.



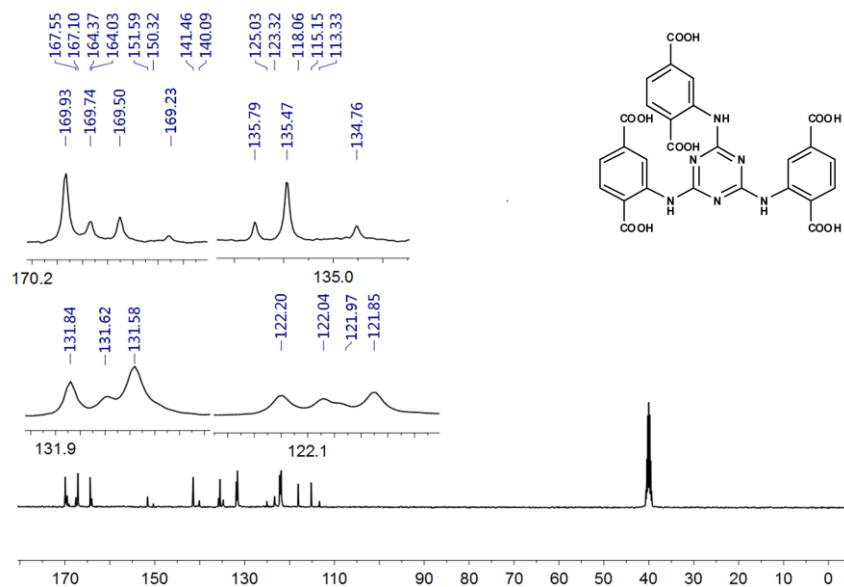
Supplementary Figure S19 PXRD patterns. PXRD patterns for **1** and transition metal exchanged **1**: as-synthesized **1**, red; Eu^{3+} @**1**, green; **Tb**@**1**, blue; **Eu/Tb**@**1**, orange.



Supplementary Figure S20 HRMS spectrum for the ligand (H₆TATPT). High

resolution mass spectra (HRMS) were recorded on Bruker microTof using ESI method.

(ESI) m/z calcd for C₂₇H₁₈N₆O₁₂ [M+H]⁺: 619.1016 found: 619.1058.



Supplementary Figure S21 ¹³C-NMR spectrum for the ligand (H₆TATPT, C₂₇H₁₈N₆O₁₂). Data were recorded on a Bruker AV400 NMR (400 MHz) at 25 °C.

Supplementary Table S1 Crystal data and structural refinement of compound **1**; for atomic coordinates, equivalent isotropic displacement parameters, bond lengths, angles, and anisotropic displacement parameters please see the CIF (CCDC code 916964).

Empirical formula	$C_{174}H_{288}N_{51}O_{78}Cl_3Cd_6$
Formula weight	5147.52
Temperature (K)	296
Crystal system	cubic
Space group	<i>Fm-3m</i>
Unit cell dimensions	
a (Å)	38.527(2)
b (Å)	38.527(2)
c (Å)	38.527(2)
α (°)	90
β (°)	90
γ (°)	90
Volume (Å ³)	57186.8 (6)
Z	8
Calculated density (mg/m ³)	1.196
F(000)	12600
Crystal size (mm ³)	0.86 × 0.44 × 0.24
R_{int}	0.0697
Goodness-of-fit on F^2	1.052
Final R indices [$I > 2\sigma(I)$]	$R_1 = 0.0896$, $wR_2 = 0.2749$

Supplementary Table S2 The concentration of Cu^{2+} , Co^{2+} , Ni^{2+} , and Zn^{2+} in compound **1**.

Transition metal	The concentration of transition	The concentration of transition
	metal (wt%)	metal per formula
Co	5.268	0.37
Cu	7.021	0.46
Ni	3.615	0.26
Zn	4.307	0.28

Supplementary Table S3 The CIE coordinates of $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+@1$ at various concentration of $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+$.

Sample	CIE coordinate		The concentration of $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+$ (respect to Cd wt%)
	X	Y	
1	0.175	0.145	0
2	0.231	0.206	0.526
3	0.254	0.239	1.04
4	0.311	0.330	3.52
5	0.322	0.352	3.73
6	0.350	0.429	4.52
7	0.371	0.468	7.50
8	0.400	0.501	8.83

Supplementary Table S4 Summary of the quantum yield of the reported white-emitting MOFs.

Compound	Quantum yield	Reference number
PbL2 ^a	2–3%	26
Eu-SMOF-1 ^b	4.3%	25
ZJU-1:1.5% Tb ³⁺ , 2.0% Eu ³⁺ ^c	6.8%	12
[AgL] _n nH ₂ O ^d	10.86%	23
Eu/Tb@1 ^e	11.3%	this work
[Ir(ppy) ₂ (bpy)] ⁺ @1 ^e	20.4%	this work

^aL2 = 2,5-bis(((S)-2-hydroxypropyl)thio)terephthalic acid

^bSMOF-1 = In(BTB)_{2/3}(OA)(DEF)_{3/2} (BTB = 1,3,5-Tris(4-carboxyphenyl)benzene, OA = oxalic acid,

DEF = N,N'-diethylformamide)

^cZJU-1 = Na₃[La(PDA)₃](H₂O)₁₂ (PDA = pyridine-2,6-dicarboxylate)

^dL = 4-cyanobenzoate

^e1 = [(CH₃)₂NH₂]_{1.25}[(Cd_{0.5}Cl_{0.25})(TATPT)_{1/3}]·DMF·1.5H₂O, (TATPT =

2,4,6-tris(2,5-dicarboxylphenylamino)-1,3,5-triazine, DMF = N,N-Dimethylformamide)

Supplementary Table S5 Quantum yield of white-emitting $[\text{Ir}(\text{ppy})_2(\text{bpy})]^+@1$ at different temperatures.

Temperature (K)	298	378	388	398	408	423
Quantum yield (%)	20.7	21.1	20.1	19.7	18.5	17.9

Supplementary Table S6 Correlated color temperature (CCT) and color rendering index (CRI) for white LED.

CCT (K)	CRI	CIE	
		x	y
5409	84.5	0.30	0.35