# Exploiting Pseudo C2-Symmetry for an Efficient Synthesis of the F-Ring of the Spongistatins

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## Supporting Information

**General Information.** All reactions were carried out under an atmosphere of nitrogen in flame-dried glassware with magnetic stirring unless otherwise indicated. Degassed solvents were purified by passage through an activated alumina column. Buffered silica gel (pH 7) was prepared by adding 10% (by weight) pH 7 aqueous phosphate buffer solution to silica gel and mixing until homogeneous. <sup>1</sup>H NMR spectra were recorded on a Bruker DPX-300 (300 MHz), Bruker DPX-400 (400 MHz), or Bruker Avance III 500 (500 MHz) spectrometer and are reported relative to protiated solvent signals (CDCl<sub>3</sub> = 7.27 ppm; C<sub>6</sub>D<sub>6</sub> = 7.16 ppm; DMSO-d<sub>6</sub> = 2.50). Data are based on apparent multiplicities and are reported as follows: (bs= broad singlet, s = singlet, d = doublet, t = triplet, q = quartet, quin = quintet, sep = septet, m = multiplet, dd = doublet of doublets, ddd = doublet of doublet of doublets; coupling constant(s) in Hz). Proton decoupled <sup>13</sup>C NMR spectra were recorded on a Bruker DPX-300 (75 MHz), Bruker DPX-400 (100 MHz), or Bruker Avance III 500 (126 MHz) spectrometer and are reported in ppm from CDCl<sub>3</sub> internal standard (77.23 ppm), C<sub>6</sub>D<sub>6</sub> (128.39 ppm) or DMSO-d<sub>6</sub> (39.52 ppm). Infrared spectra were recorded on a Perkin Elmer Paragon 1000 FT-IR spectrometer. Optical rotations were recorded on a Jasco DIP-1000 digital polarimeter. Melting points were recorded on a SRS DigiMelt MPA160 melting point apparatus and are uncorrected.



**Procedure 1:** To a solution of 3-methyl-1,4-pentadiene 7 (12 mL, 98 mmol) and *tert*-butyl acrylate (36 mL, 246 mmol, 2.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (480 mL) was added Hoveyda-Grubbs 2<sup>nd</sup>

Generation catalyst (308 mg, 0.492 mmol, 0.5 mol %). After 16h, an additional portion of *tert*-butyl acrylate (18 mL, 123 mmol, 1.25 equiv) was added followed by an additional portion of the HG-II catalyst (154 mg, 0.25 mmol, 0.25 mol %), and the reaction mixture was heated to 30 °C. After 30h, an additional portion of *tert*-butyl acrylate (18 mL, 123 mmol, 1.25 equiv) was added followed by an additional portion of the HG-II catalyst (154 mg, 0.25 mmol, 0.25 mol, 0.25 mol), and the temperature was raised to 45 °C. Each day for the next two days, an additional portion of *tert*-butyl acrylate (18 mL, 123 mmol, 1.25 mol), and the temperature was raised to 45 °C. Each day for the next two days, an additional portion of *tert*-butyl acrylate (18 mL, 123 mmol, 1.25 equiv) was added followed by an additional portion of the HG-II catalyst (154 mg, 0.25 mmol, 0.25 mol), and the temperature was raised to 45 °C. Each day for the followed by an additional portion of *tert*-butyl acrylate (18 mL, 123 mmol, 1.25 equiv) was added followed by an additional portion of the HG-II catalyst (154 mg, 0.25 mmol, 0.25 mol), and the temperature was coled to ambient temperature and concentrated. The residue was treated with hexanes and the mixture was concentrated. The residue was purified by silica gel flash column chromatography (5% EtOAc/Hex) to yield diene **10b** (22 g, 78 mmol, 80%) as a pale yellow oil.

**Procedure 2:** To a solution of 3-methyl-1,4-pentadiene **7** (5.62 mL, 45 mmol, 1 equiv) and 4methoxyphenol (25 mg, 0.199 mmol, 0.47 mol %) (**Note**: 4-methoxyphenol is a stabilizer for the *t*butyl acrylate, and we have found that we get slightly better results by adding additional stabilizer in this fashion) in *t*-butyl acrylate (135 ml, 0.9 mol, 20 equiv) was added Hoveyda-Grubbs  $2^{nd}$  Generation catalyst (65.6 mg, 0.105 mmol, 0.23 mol %). The flask was fitted with a reflux condenser and the reaction mixture was heated to 50 °C (oil bath, external temperature). After 1.5 h, an additional portion of the HG-II catalyst (65.6 mg, 0.105 mmol, 0.23 mol %) was added. After 1.5 h, an additional portion of the HG-II catalyst (65.6 mg, 0.105 mmol, 0.23 mol %) was added. After 2 h, the reaction mixture was cooled to ambient temperature and concentrated. The residue was treated with toluene and the mixture was concentrated. The residue was purified by silica gel flash column chromatography (4% EtOAc/Hex) to yield diene **10b** (7.55 g, 26.7 mmol, 58%) as a pale yellow oil.

Data for **10b**: TLC  $R_f = 0.45$  (10% EtOAc/Hex); IR (thin film) 2977, 2933, 1713, 1614, 1392, 1318, 1153 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  6.76 (dd, J = 15.7, 7.0 Hz, 2H), 5.73 (dd, J = 15.7, 1.3 Hz, 2H), 3.17-3.08 (m, 1H), 1.47 (s, 18H), 1.20 (d, J = 6.9 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  165.9, 148.4, 123.2, 80.6, 38.5, 28.3, 18.6; HRMS for  $C_{16}H_{27}O_4$  (FAB+): calcd 283.1904 ([M+H]<sup>+</sup>), found 283.1910 ([M+H]<sup>+</sup>).



To a mechanically-stirred solution of K<sub>3</sub>Fe(CN)<sub>6</sub> (148 g, 450 mmol, 6 equiv), K<sub>2</sub>CO<sub>3</sub> (62.1 g, 450 mmol, 6 equiv), NaHCO<sub>3</sub> (37.8 g, 450 mmol, 6 equiv) and CH<sub>3</sub>SO<sub>2</sub>NH<sub>2</sub> (14.3 g, 150 mmol, 2 equiv) in H<sub>2</sub>O (750 mL) was added a solution of (DHQD)<sub>2</sub>PHAL (3.00 g, 3.85 mmol, 0.05 equiv) in t-BuOH (500 mL). The mixture was cooled to 0 °C and K<sub>2</sub>OsO<sub>4</sub>•2H<sub>2</sub>O (1.10 g, 3.00 mmol, 0.04 equiv) was added, followed 10 minutes later by a solution of diene 10b (21.2 g, 74.9 mmol) in t-BuOH (250 mL). The bright orange suspension was stirred vigorously at 0 °C for 18 h. Solid Na<sub>2</sub>SO<sub>3</sub> (94 g, 749 mmol, 10 equiv) was added and the reaction mixture was stirred for a further 45 min. The layers of the cold reaction mixture were separated, and the aqueous layer was extracted with EtOAc (3 x 150 mL). The combined organic layers were stirred with solid NaCl, the resulting brine layer was removed. The organic phase was washed with additional brine (350 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. The residue was purified by silica gel flash column chromatography (50% EtOAc/Hex) to yield tetraol 11 (16.4 g, 46.7 mmol, 62%) as the major product of a 4.5:1 mixture of diastereomers (as judged by <sup>1</sup>H NMR spectroscopy – see below) as a colorless gum that solidified to a colorless solid on standing. This mixture was used as is in the next step, but for the purposes of characterization, an analytically pure sample was obtained by careful flash chromatography: TLC  $R_f = 0.3$  (60% EtOAc/Hex); mp 87-89 °C;  $[\alpha]^{21}_{D} = -12.7^{\circ}$  (c = 0.56, CHCl<sub>3</sub>); IR (thin film) 3473, 2978, 2934, 1716, 1394, 1288 cm<sup>-1</sup>; <sup>1</sup>H NMR  $(400 \text{ MHz}, \text{CDCl}_3) \delta 4.15 \text{ (dd}, J = 5.7, 3.0 \text{ Hz}, 1\text{H}), 4.12 \text{ (dd}, J = 5.7, 1.6 \text{ Hz}, 1\text{H}), 4.06 \text{ (dt}, J = 8.4, 3.3 \text{ Hz})$ Hz, 1H), 3.99 (td, J = 8.9, 1.3 Hz, 1H), 3.55 (d, J = 8.5 Hz, 1H), 3.52 (d, J = 5.7 Hz, 1H), 3.34 (d, J = 5.7 Hz, 1H), 3.22 (d, J = 9.1 Hz, 1H), 2.26-2.14 (m, 1H), 1.50 (s, 18H), 1.07 (d, J = 7.0 Hz, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ 173.0, 83.5, 83.3, 75.0, 74.2, 72.2, 71.9, 40.0, 28.2, 13.1; HRMS for  $C_{16}H_{31}O_8$  (FAB+): calcd 351.2013 ([M+H]<sup>+</sup>), found 351.2018 ([M+H]<sup>+</sup>).



To a solution of tetraol **11** (12.7 g, 36.2 mmol) and trimethyl orthoacetate (8.90 mL, 54.3 mmol, 1.5 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (630 mL) was added pyridinium *p*-toluenesulfonate (PPTS) (9.10 g, 36.2 mmol, 1 equiv). After 16 h, the reaction mixture was poured into saturated aqueous NaHCO<sub>3</sub> (400 mL) and the layers were separated. The aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 x 75 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. The residue was purified by silica gel flash column chromatography (20% EtOAc/Hex) to yield orthoacetate **16** as a pale yellow gum (9.77 g, 26.1 mmol, 72%). TLC R<sub>f</sub> = 0.65 (50% EtOAc/hex);  $[\alpha]^{22}_{D} = +18.0^{\circ}$  (*c* = 1.53, CHCl<sub>3</sub>);

IR (thin film) 3512, 3004, 1752, 1728, 1403, 1293, 1131 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  4.49 (s, 1H), 4.47 (d, *J* = 3.5 Hz, 1H), 4.07 (dd, *J* = 8.3, 1.7 Hz, 1H), 3.86 (dd, *J* = 10.4, 1.7 Hz, 1H), 2.92 (d, *J* = 8.3 Hz, 1H), 2.51 (dqd, *J* = 10.4, 6.9, 3.5 Hz, 1H), 1.64 (s, 3H), 1.48 (s, 9H), 1.48 (s, 9H), 0.97 (d, *J* = 6.9 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  171.7, 169.7, 120.7, 83.1, 82.5, 81.7, 76.7, 74.1, 69.6, 30.7, 28.18, 28.15, 21.6, 12.1; HRMS for C<sub>18</sub>H<sub>31</sub>O<sub>8</sub> (FAB+): calcd 375.2013 ([M+H]<sup>+</sup>), found 375.2014 ([M+H]<sup>+</sup>).



To a cooled (-78 °C) solution of alcohol **16** (5.54 g, 14.8 mmol) in THF (150 mL) was added a solution of NaHMDS (2.66 g, 14.5 mmol, 0.98 equiv) in THF (15 mL) by syringe pump over 30 min. Five minutes following complete addition, benzyl bromide (3.52 mL, 29.6 mmol, 2 equiv) was added in one portion and the reaction mixture was warmed to 0 °C. After 3.5 h, the reaction was quenched by the addition of saturated aqueous NH<sub>4</sub>Cl (150 mL). The layers were separated, and the aqueous layer was extracted with EtOAc (3 x 50 mL). The combined organic layers were washed with brine (200 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. The residue was purified by silica gel flash column chromatography (10% EtOAc/Hex) to yield orthoacetate **18** (5.40 g, 11.6 mmol, 78%) as a colorless gum. TLC R<sub>f</sub> = 0.35 (20% EtOAc/hex);  $[\alpha]^{20}_{D}$  = -42.3° (*c* = 0.64, CHCl<sub>3</sub>); IR (thin film) 2976, 1751, 1723, 1403, 1368, 1151, 1135 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.38-7.28 (m, 5H), 4.96 (d, *J* = 12.3 Hz, 1H), 4.40 (s, 2H), 4.39 (d, *J* = 15.0 Hz, 1H), 3.91 (dd, *J* = 10.5, 2.6 Hz, 1H), 3.83 (d, *J* = 2.7 Hz, 1H), 2.44 (dqd, *J* = 10.3, 6.9, 3.5 Hz, 1H), 1.68 (s, 3H), 1.50 (s, 9H), 1.46 (s, 9H), 0.57 (d, J = 6.9 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  169.8, 169.3, 137.4, 128.9, 128.6, 128.3, 120.9, 82.4, 82.2, 81.8, 76.4, 75.2, 74.1, 72.7, 30.8, 28.3, 28.2, 21.7, 11.7; HRMS for C<sub>25</sub>H<sub>37</sub>O<sub>8</sub> (FAB+): calcd 465.2483 ([M+H]<sup>+</sup>), found 465.2501 ([M+H]<sup>+</sup>).



To a cooled (-78 °C) solution of NaHMDS (7.78 g, 42.4 mmol, 4.5 equiv) in THF (200 mL) was added a solution of MeOAc (3.75 mL, 47.1 mmol, 5 equiv) in THF (16.5 mL) by syringe pump over 40 min. Upon complete addition, the reaction mixture was stirred for a further 2 h, and then a

solution of orthoacetate **18** (4.38 g, 9.43 mmol) in THF (25 mL) was added by syringe pump over 15 min. After 5 h, the still cold reaction mixture was poured into saturated aqueous NH<sub>4</sub>Cl (400 mL). The layers were separated and the aqueous layer was extracted with Et<sub>2</sub>O (3 x 100 mL). The combined organic layers were washed with brine (300 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. The residue was purified by silica gel flash column chromatography (20% EtOAc/Hex) to yield  $\beta$ -ketoester **19** (3.50 g, 7.53 mmol, 80%) as a pale yellow oil. Analysis by <sup>1</sup>H NMR spectroscopy revealed a complex mixture of tautomers, and the material was taken into the next step without further characterization.

To a solution of β-ketoester **19** (5.20 g, 11.2 mmol) in MeOH (85 mL) was added H<sub>2</sub>O (30 mL) and pyridinium *p*-toluenesulfonate (2.81 g, 11.2 mmol, 1 equiv). After 39 h, solid NaHCO<sub>3</sub> (excess) was added and the reaction mixture was stirred until bubbling ceased. The reaction mixture was diluted with brine (300 mL) and extracted with EtOAc (4 x 50 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated. The residue was purified by silica gel flash column chromatography (40% EtOAc/Hex) to yield hemiketal **20** (5.32 g, 11.0 mmol, 98%) as a pale yellow oil that solidified to a colorless solid on standing. TLC R<sub>f</sub> = 0.25 (50% EtOAc/Hex); mp 144-146 °C;  $[\alpha]^{22}_{D} = -48.5^{\circ}$  (*c* = 0.45, CHCl<sub>3</sub>); IR (thin film) 3435, 2977, 2934, 1743, 1497, 1369, 1236, 1158, 1053 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.39-7.29 (m, 5H), 5.99 (s, 1H), 4.96 (t, *J* = 10.3 Hz, 1H), 4.95 (d, *J* = 12.9 Hz, 1H), 4.39 (d, *J* = 12.0 Hz, 1H), 4.08 (d, *J* = 10.2 Hz, 1H), 3.89 (s, 1H), 3.68 (s, 3H), 3.21 (t, *J* = 10.4 Hz, 1H), 3.01 (d, *J* = 16.4 Hz, 1H), 2.68 (d, *J* = 16.4 Hz, 1H), 2.13 (s, 3H), 2.17-2.07 (m, 1H), 2.04 (d, *J* = 12.0 Hz, 1H), 1.50 (s, 9H), 0.54 (d, *J* = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 173.9, 172.0, 169.0, 137.2, 129.1, 128.7, 128.4, 98.1, 82.1, 75.9, 75.9, 75.2, 74.4, 73.1, 52.2, 39.0, 36.1, 28.3, 21.3, 12.1; HRMS for C<sub>24</sub>H<sub>35</sub>O<sub>10</sub> (FAB+): calcd 483.2225 ([M+H]<sup>+</sup>), found 483.2238 ([M+H]<sup>+</sup>).



To a cooled (-78 °C) solution of hemiketal **20** (0.90 g, 1.9 mmol) and triethylsilane (3.0 mL, 19 mmol, 10 equiv) in  $CH_2Cl_2$  (27 mL) was added triethylsilyl trifluoromethanesulfonate (1.7 mL, 7.5 mmol, 4 equiv) slowly down the side of the flask. After 10 min, the reaction mixture was warmed to 0 °C. After 1.3 h, triethylamine (2.6 mL, 19 mmol, 10 equiv) was added slowly down the side of the

flask. After 1 h, saturated aqueous NaHCO<sub>3</sub> (30 mL) was added and the resulting mixture was vigorously stirred. The layers were separated, and the aqueous layer was acidified with AcOH before being extracted with EtOAc (3 x 15 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. Excess AcOH was removed by azeotroping with hexanes, and the remaining volatiles were removed *in vacuo* overnight. The residue was purified by silica gel flash column chromatography (gradient 20 to 40 to 100% EtOAc/Hex) to yield carboxylic acid **21** (0.71 g, 1.4 mmol, 72%) as a colorless solid. TLC R<sub>f</sub> = 0.07 (EtOAc); mp 104-110 °C;  $[\alpha]^{20}_{D} = -13.3^{\circ}$  (*c* = 0.51, CHCl<sub>3</sub>); IR (thin film) 3378, 2954, 2878, 1740, 1615, 1497, 1231, 1131, 1091 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.32-7.28 (m, 5H), 4.99 (d, *J* = 12.2 Hz, 1H), 4.74 (dd, *J* = 9.8, 8.9 Hz, 1H), 4.32 (d, *J* = 12.3 Hz, 1H), 3.86 (s, 1H), 3.73 (s, 3H), 3.62 (d, *J* = 10.2 Hz, 1H), 3.56-3.44 (m, 2H), 2.84 (d, *J* = 14.1 Hz, 1H), 2.64 (dd, *J* = 14.3, 6.5 Hz, 1H), 2.14 (s, 3H), 2.04-1.97 (m, 1H), 0.95 (t, *J* = 7.9 Hz, 9H), 0.58 (q, *J* = 7.9 Hz, 6H), 0.44 (d, *J* = 5.8 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  175.8, 173.5, 170.6, 137.6, 129.1, 128.6, 128.2, 82.4, 79.4, 77.6, 77.4, 73.2, 72.2, 52.6, 37.2, 36.8, 21.6, 12.2, 7.0, 5.4; HRMS for C<sub>26</sub>H<sub>40</sub>NaO<sub>9</sub>Si (FAB+): calcd 547.2334 ([M+Na]<sup>+</sup>), found 547.2354 ([M+Na]<sup>+</sup>).



To a cooled (0 °C) solution of 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (EDCI) (543 mg, 2.83 mmol, 1.1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was added HOBt•H<sub>2</sub>O (394 mg, 2.57 mmol, 1 equiv) and *N*-methylmorpholine (0.62 mL, 5.66 mmol, 2.2 equiv), followed by a solution of carboxylic acid **21** (1.35 g, 2.57 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (15 mL) slowly. After 30 min, the reaction mixture was cooled to 0 °C and Me(OMe)NH•HCl (276 mg, 2.83 mmol, 1.1 equiv) was added. After 5 min, the cooling bath was removed and the reaction mixture was allowed to warm to ambient termperature. After 16 h, 40 h, and 48 h, additional portions of *N*-methylmorpholine (62 µL, 0.22 equiv) and Me(OMe)NH•HCl (50 mg, 0.2 equiv) were added. After 64 h total, the reaction mixture was poured into H<sub>2</sub>O (50 mL), and the layers were separated. The aqueous layer was extracted with EtOAc (4 x 10 mL). The combined organic layers were washed with saturated aqueous NH<sub>4</sub>Cl (30 mL), and the aqueous layer was back-extracted with EtOAc (20 mL). The combined organic layers were washed with saturated aqueous NH<sub>4</sub>Cl (30 mL), and the fibrine (80 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. The residue was purified by silica gel flash column chromatography (50% EtOAc/Hex) to yield Weinreb amide **22** (1.20 g, 2.11 mmol, 82%)

as a colorless gum that slowly crystallized to colorless crystals. TLC  $R_f = 0.6$  (EtOAc); mp 108-110 °C;  $[\alpha]_{D}^{20} = -18.4^{\circ}$  (c = 0.52, CHCl<sub>3</sub>); IR (thin film) 2954, 2878, 1741, 1686, 1619, 1232, 1130 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.36-7.28 (m, 5H), 4.87 (d, J = 11.9 Hz, 1H), 4.77 (dd, J = 10.1, 9.3 Hz, 1H), 4.29 (d, J = 12.0 Hz, 1H), 4.24 (s, 1H), 3.65 (s, 3H), 3.65-3.60 (m, 2H), 3.59 (s, 3H), 3.47 (t, J =9.0 Hz, 1H), 3.24 (s, 3H), 2.69 (d, J = 15.1 Hz, 1H), 2.49 (dd, J = 15.0, 10.9 Hz, 1H), 2.19-2.10 (m, 1H), 2.13 (s, 3H), 0.92 (t, J = 7.9 Hz, 9H), 0.60-0.52 (m, 9H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  171.7, 170.8, 169.5, 137.1, 129.1, 128.5, 128.3, 79.9, 79.6, 78.0, 75.0, 73.4, 72.4, 61.3, 51.7, 37.2, 37.1, 32.9, 21.6, 12.3, 7.0, 5.4; HRMS for C<sub>28</sub>H<sub>46</sub>NO<sub>9</sub>Si (FAB+): calcd 568.2936 ([M+H]<sup>+</sup>), found 568.2936 ([M+H]<sup>+</sup>).



To a solution of Weinreb amide **22** (1.20 g, 2.11 mmol) in THF (10.6 mL) was added tetrabutylammonium fluoride (10.6 mL of a 1M solution in THF, 10.6 mmol, 5 equiv). After 16 h, the reaction mixture was poured into saturated aqueous NH<sub>4</sub>Cl (40 mL). The layers were separated, and the aqueous layer was extracted with EtOAc (4 x 20 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. The residue was purified by silica gel flash column chromatography (EtOAc) to yield diol **23** (0.760 g, 1.86 mmol, 88%) as a colorless gum. TLC R<sub>f</sub> = 0.1 (EtOAc);  $[\alpha]^{20}_{D} = -35.5^{\circ}$  (c = 0.50, CHCl<sub>3</sub>); IR (thin film) 3444, 2936, 2876, 1736, 1660, 1438, 1164, 1066 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>)  $\delta$  7.33-7.27 (m, 2H), 7.15-7.03 (m, 3H), 4.75 (d, J = 11.6 Hz, 1H), 4.35 (d, J = 2.9 Hz, 1H), 4.21 (d, J = 11.6 Hz, 1H), 3.90-3.76 (m, 2H), 3.68 (dd, J = 10.2, 2.9 Hz, 1H), 3.47 (s, 3H), 3.46 (s, 1H), 3.34 (td, J = 8.9, 3.9 Hz, 1H), 3.24 (s, 3H), 3.29-3.15 (m, 1H), 3.11 (s, 3H), 3.02 (dd, J = 15.6, 2.6 Hz, 1H), 2.64 (dd, J = 15.6, 9.8 Hz, 1H), 2.36-2.17 (m, 1H), 0.98 (d, J = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>)  $\delta$  172.8, 170.5, 138.1, 129.5, 129.0, 128.7, 81.3, 79.2, 78.0, 77.3, 75.6, 72.9, 61.2, 51.7, 38.6, 38.3, 33.8, 13.1; HRMS for C<sub>20</sub>H<sub>30</sub>NO<sub>8</sub> (FAB+): calcd 412.1966 ([M+H]<sup>+</sup>), found 412.1978 ([M+H]<sup>+</sup>).



To a solution of diol **23** (0.67 g, 1.6 mmol) and 2,2-dimethoxypropane (8 mL) in CH<sub>2</sub>Cl<sub>2</sub> (8 mL) was added (±)-camphorsulfonic acid (16 mg, 0.065 mmol, 4 mol %). After 3 d, saturated aqueous NaHCO<sub>3</sub> (10 mL) was added, and the layers were separated. The aqueous layer was extracted with EtOAc (4 x 4 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. The residue was purified by silica gel flash column chromatography (gradient 50 to 100% EtOAc/Hex) to yield acetonide **24** (0.51 g, 1.13 mmol, 69%) as a colorless oil, along with recovered diol **23** (0.14 g, 0.34 mmol, 21%). TLC R<sub>f</sub> = 0.4 (EtOAc);  $[\alpha]^{20}_{D}$  = -32.6° (*c* = 0.61, CHCl<sub>3</sub>); IR (thin film) 2983, 2936, 1739, 1678, 1371, 1171, 1087, 1065, 1051 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.34-7.24 (m, 5H), 4.84 (d, *J* = 11.9 Hz, 1H), 4.29 (d, *J* = 12.1 Hz, 1H), 4.24 (s, 1H), 3.90-3.81 (m, 1H), 3.62 (s, 3H), 3.57 (s, 3H), 3.52 (d, *J* = 9.4 Hz, 1H), 3.22 (s, 3H), 3.20-3.13 (m, 2H), 2.63 (d, *J* = 15.1 Hz, 1H), 2.57 (dd, *J* = 15.3, 10.1 Hz, 1H), 2.35-2.22 (m, 1H), 1.39 (s, 3H), 1.37 (s, 3H), 0.74 (d, *J* = 6.3 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  171.2, 169.6, 137.1, 128.9, 128.5, 128.3, 110.6, 83.9, 81.2, 78.3, 76.0, 74.8, 72.4, 61.1, 51.7, 37.6, 36.6, 33.3, 27.0, 26.7, 12.7; HRMS for C<sub>23</sub>H<sub>34</sub>NO<sub>8</sub> (FAB+): calcd 452.2279 ([M+H]<sup>+</sup>), found 452.2294 ([M+H]<sup>+</sup>).



To a cooled (0 °C) flask containing dry CeCl<sub>3</sub> (0.87 g, 3.5 mmol, 8 equiv) was added THF (4.5 mL) slowly with vigorous stirring, and the resulting suspension was sonicated for 15 min. The ice water bath was removed and the resulting suspension was stirred overnight. The mixture was sonicated again for 20 min and then cooled to -78 °C. TMSCH<sub>2</sub>MgCl (1.87 mL, 2.44 mmol, 5.5 equiv) was then added dropwise over 5 min. After 3 h, a solution of acetonide **24** (197 mg, 0.436 mmol) in THF (4 mL) was added dropwise. After 2 h, the mixture was allowed to warm to ambient temperature. After 6 h, saturated aqueous NH<sub>4</sub>Cl (15 mL) was added, and the biphasic mixture was poured into H<sub>2</sub>O (15 mL) and EtOAc (15 mL). The layers were separated, and the aqueous layer was extracted with EtOAc (3 x 15 mL). The combined organic layers were washed with brine (50 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (20 mL), silica gel (4 g) was added, and the resulting suspension was stirred for 14 h. The reaction mixture was filtered, and the silica gel was washed with EtOAc. The filtrate was concentrated and the residue was purified by silica gel flash column chromatography (30% EtOAc/Hex) to yield allylsilane **25** (131 mg, 0.259 mmol, 59%) as a

colorless oil. TLC  $R_f = 0.7$  (60% EtOAc/Hex);  $[\alpha]^{20}_{D} = -20.6^{\circ}$  (c = 0.63, CHCl<sub>3</sub>); IR (thin film) 2983, 2938, 2877, 1741, 1685, 1370, 1231, 1156, 1087, 1043 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.37-7.28 (m, 5H), 4.87 (d, J = 11.5 Hz, 1H), 4.67 (br s, 1H), 4.56 (br s, 1H), 4.33 (d, J = 12.1 Hz, 1H), 4.26 (s, 1H), 3.59 (s, 3H), 3.51 (d, J = 8.8 Hz, 2H), 3.25 (s, 3H), 3.21-3.13 (m, 2H), 2.30 (d, J = 14.9 Hz, 2H), 2.21 (dd, J = 15.2, 9.4 Hz, 1H), 1.56 (d, J = 13.5 Hz, 1H), 1.50 (d, J = 13.5 Hz, 1H), 1.42 (s, 3H), 1.40 (s, 3H), 0.77 (d, J = 6.2 Hz, 3H), 0.01 (s, 9H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  169.6, 144.1, 137.2, 129.0, 128.6, 128.3, 110.2, 109.1, 84.1, 81.4, 79.2, 79.0, 75.3, 72.5, 61.1, 40.6, 36.8, 33.3, 27.8, 27.1, 26.9, 13.0, -1.1; HRMS for C<sub>27</sub>H<sub>44</sub>NO<sub>6</sub>Si (FAB+): calcd 506.2932 ([M+H]<sup>+</sup>), found 506.2929 ([M+H]<sup>+</sup>).



To a cooled (-78 °C) solution of allylsilane 25 (28 mg, 0.056 mmol) in THF (0.65 mL) was added MeMgBr (79 µL, 0.11 mmol, 2 equiv) slowly down the side of the flask. After 5 min, the reaction mixture was warmed to 0 °C. After 45 min, the reaction mixture was recooled to -78 °C and an additional portion of MeMgBr (79 µL, 0.11 mmol, 2 equiv) was added slowly down the side of the flask. After 5 min, the reaction mixture was warmed to 0 °C. After 15 min, saturated aqueous NH<sub>4</sub>Cl (2 mL) was added, and the layers were separated. The aqueous layer was extracted with EtOAc (3 x 0.75 mL). The combined organic layers were washed with brine (2 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated to yield methyl ketone 12 (26 mg, 0.056 mmol, 99%) as a colorless oil. TLC  $R_f = 0.7$  $(30\% \text{ EtOAc/Hex}); [\alpha]^{22}_{D} = -32.6^{\circ} (c = 0.86, \text{CHCl}_3); \text{ IR (thin film) } 2954, 2876, 1713, 1635, 1371,$ 1230, 1087, 1043 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.41-7.29 (m, 5H), 4.82 (d, J = 11.9 Hz, 1H), 4.61 (br s, 1H), 4.57 (br s, 1H), 4.32 (d, J = 11.9 Hz, 1H), 3.78 (d, J = 1.7 Hz, 1H), 3.53 (t, J = 8.4 Hz, 1H), 3.27 (dd, J = 9.7, 1.6 Hz, 1H), 3.16 (p, J = 8.7 Hz, 2H), 2.34 (d, J = 15.4 Hz, 1H), 2.26 (s, 3H), 2.25-2.17 (m, 2H), 1.54 (d, J = 14.2 Hz, 1H), 1.46 (d, J = 13.7 Hz, 1H), 1.43 (s, 3H), 1.41 (s, 3H), 0.68 (d, J = 6.4 Hz, 3H), 0.01 (s, 9H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  212.8, 143.5, 136.9, 129.0, 128.9, 128.7, 110.4, 109.5, 83.7, 83.7, 82.5, 79.1, 78.4, 73.7, 41.0, 37.0, 28.0, 27.2, 27.1, 26.9, 13.0, -1.1; HRMS for  $C_{26}H_{41}O_5Si$  (FAB+): calcd 461.2718 ([M+H]<sup>+</sup>), found 461.2737 ([M+H]<sup>+</sup>).

### X-ray Structure of 22:

X-ray quality crystals of **22** were obtained as described above, and its structure was solved by Dr. Aaron Sattler (Parkin Group). This structure confirms the stereostructure of **22** is as shown (Fig. S1).



Figure S1. X-ray structure of 22.

### NMR Structure of 15:

The structure of compound **15** was assigned from a series of NMR experiments, including <sup>1</sup>H and <sup>13</sup>C spectra, as well as COSY, HSQC, and NOESY spectra. The key nOe interactions are shown in Fig. S2, and the spectra are reproduced below.



Figure S2. NOESY enhancements observed for 15 establish its stereostructure.







S - 12





S - 14



S - 15





S - 17



S - 18



S - 19









S - 23



270 260 250 240 230 220 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 f1 (ppm)



