

## Supporting information

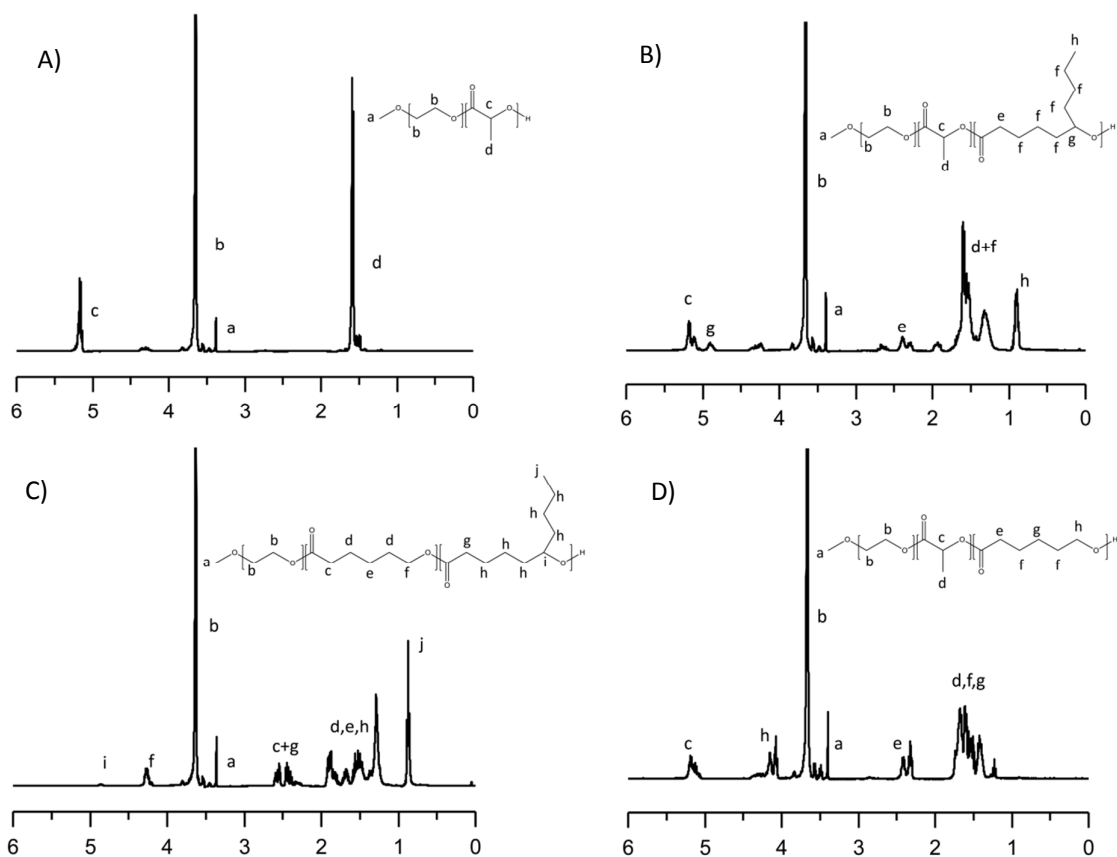
### Achieving micelle control through core crystallinity

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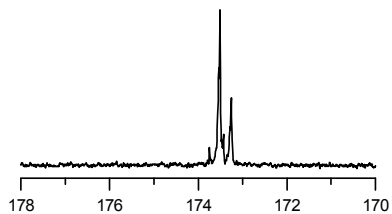
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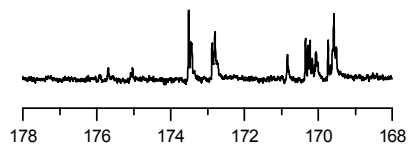
#### CHARACTERIZATION OF COPOLYMERS



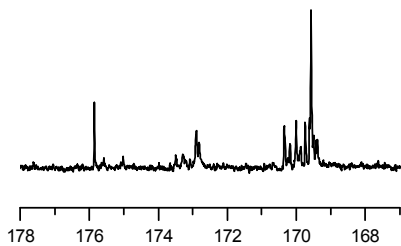
**Figure 1.** Determination of the chemical structure of A) PEG<sub>2k</sub>-PLA<sub>2k</sub>, B) PEG<sub>2k</sub>-P(LA/ $\epsilon$ DL)<sub>2k</sub>, C) PEG<sub>2k</sub>-P(CL/ $\epsilon$ DL)<sub>2k</sub> and D) PEG<sub>2k</sub>-P(CL/LA)<sub>2k</sub>. The molecular weights were calculated using  $^1\text{H-NMR}$  by integrating the methoxy peak of mPEG at 3.37 ppm and the peak for the repeating units of CL (at 4.06 ppm),  $\epsilon$ -DL (at 0.86 ppm), and LA (at 5.17 ppm).



**Figure 2.**  $^{13}\text{C}$ -NMR of  $\text{PEG}_{2k}\text{-P}(\text{CL}/\epsilon\text{DL})_{2k}$  in the carbonyl region.



**Figure 3.**  $^{13}\text{C}$ -NMR of  $\text{PEG}_{2k}\text{-P}(\text{LA}/\epsilon\text{DL})_{2k}$  in the carbonyl region.



**Figure 4.**  $^{13}\text{C}$ -NMR of  $\text{PEG}_{2k}\text{-P}(\text{CL}/\text{LA})_{2k}$  in the carbonyl region.

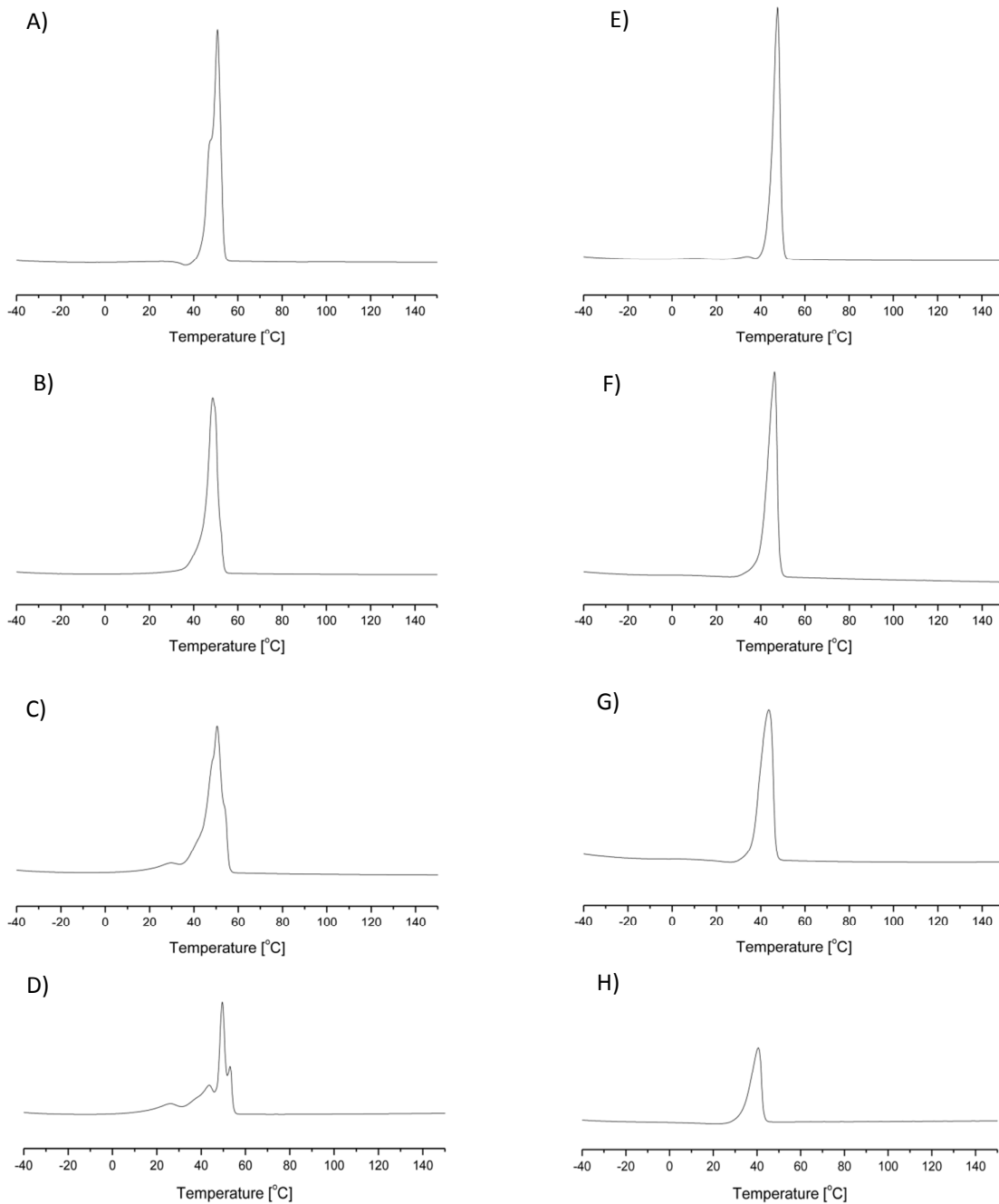
**Table 1. The mass ratio of the hydrophobic block of the copolymers with two monomers in the hydrophobic block.**  
The ratios were calculated using values obtained from  $^1\text{H}$ -NMR.

<b>Sample name</b>	<b>Mass ratio of the hydrophobic block [<math>M_1:M_2</math>]</b>
<i>PEG<sub>2k</sub>-P(CL/εDL)<sub>2k</sub></i>	40:60
<i>PEG<sub>2k</sub>-P(LA/εDL)<sub>2k</sub></i>	46:54
<i>PEG<sub>2k</sub>-P(CL/LA)<sub>2k</sub></i>	44:56

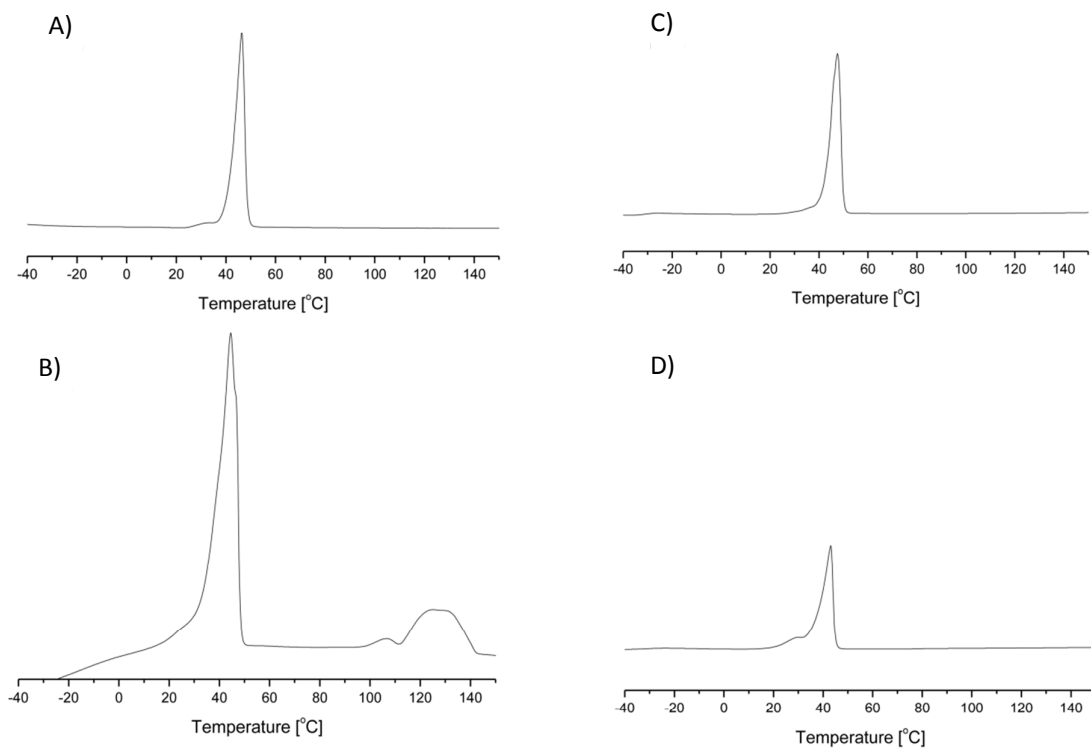
**Table 2. The melting enthalpies of the all the copolymers.**

<b>Sample name</b>	<b>Melting enthalpy [J/g]</b>
<i>PEG<sub>2k</sub>-PCL<sub>1k</sub></i>	141 <sup>a</sup>
<i>PEG<sub>2k</sub>-PCL<sub>2k</sub></i>	121 <sup>a</sup>
<i>PEG<sub>2k</sub>-PCL<sub>3k</sub></i>	97 <sup>a</sup>
<i>PEG<sub>2k</sub>-PCL<sub>4k</sub></i>	89 <sup>a</sup>
<i>PEG<sub>2k</sub>-PεDL<sub>1k</sub></i>	107
<i>PEG<sub>2k</sub>-PεDL<sub>2k</sub></i>	85
<i>PEG<sub>2k</sub>-PεDL<sub>3k</sub></i>	71
<i>PEG<sub>2k</sub>-PεDL<sub>4k</sub></i>	55
<i>PEG<sub>2k</sub>-PLA<sub>2k</sub></i>	83 <sup>a</sup>
<i>PEG<sub>2k</sub>-P(CL/εDL)<sub>2k</sub></i>	85
<i>PEG<sub>2k</sub>-P(LA/εDL)<sub>2k</sub></i>	79
<i>PEG<sub>2k</sub>-P(CL/LA)<sub>2k</sub></i>	87

<sup>a</sup> The melting enthalpies reported are combined enthalpies for both the PEG and the PCL/PLA-block.

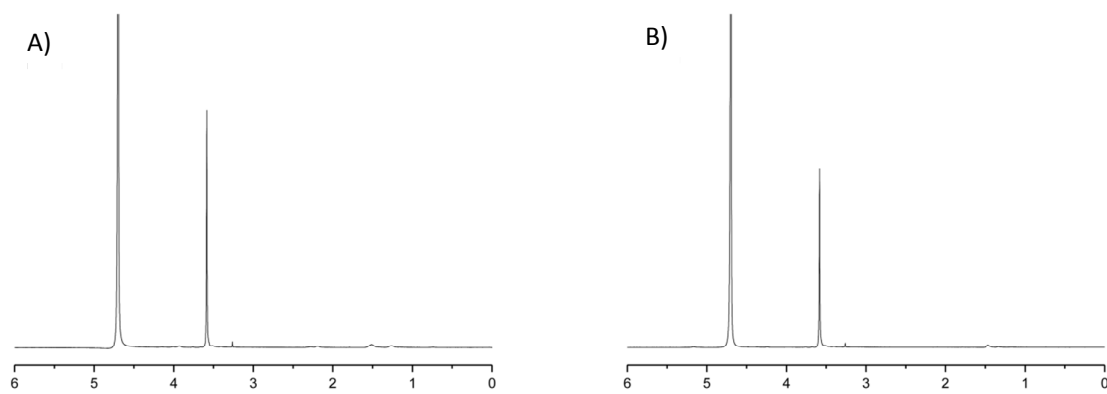


**Figure 5.** The DSC traces of A) PEG<sub>2k</sub>-PCL<sub>1k</sub>, B) PEG<sub>2k</sub>-PCL<sub>2k</sub>, C) PEG<sub>2k</sub>-PCL<sub>3k</sub>, D) PEG<sub>2k</sub>-PCL<sub>4k</sub>, E) PEG<sub>2k</sub>-PεDL<sub>1k</sub>, F) PEG<sub>2k</sub>-PεDL<sub>2k</sub>, G) PEG<sub>2k</sub>-PεDL<sub>3k</sub> and H) PEG<sub>2k</sub>-PεDL<sub>4k</sub>.



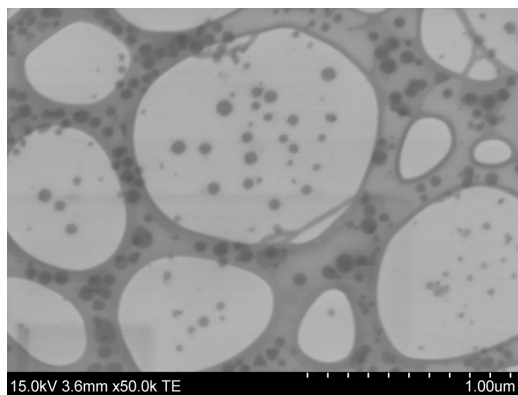
**Figure 6.** The DSC traces of A) PEG<sub>2k</sub>-P(CL/εDL)<sub>2k</sub>, B) PEG<sub>2k</sub>-PLA<sub>2k</sub>, C) PEG<sub>2k</sub>-P(LA/CL)<sub>2k</sub>, D) PEG<sub>2k</sub>-P(LA/εDL)<sub>2k</sub>.

## SELF-ASSEMBLY

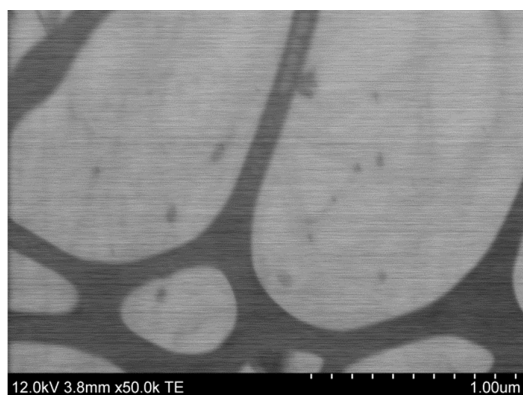


**Figure 7.** Evidence of the self-assembly of A) PEG<sub>2k</sub>-PCL<sub>2k</sub> and B) PEG<sub>2k</sub>-PLA<sub>2k</sub>. <sup>1</sup>H-NMR in D<sub>2</sub>O.

## MORPHOLOGY



**Figure 8.** Morphology of PEG<sub>2k</sub>-PεDL<sub>2k</sub> at 0.6 mg/mL.



**Figure 9.** Morphology of PEG<sub>2k</sub>-PCL<sub>2k</sub> at 0.6 mg/mL.

## MICELLE STABILITY

**Table 3.** The stability of the micelles (0.6 mg/mL) after 4 weeks measured by DLS.

Sample name	Size <sup>a</sup> [nm]	Size <sup>b</sup> [nm]
<i>PEG<sub>2k</sub>-PCL<sub>2k</sub></i>	29	42
<i>PEG<sub>2k</sub>-PLA<sub>2k</sub></i>	57	62
<i>PEG<sub>2k</sub>-PεDL<sub>2k</sub></i>	27	27
<i>PEG<sub>2k</sub>-P(CL/εDL)<sub>2k</sub></i>	41	46
<i>PEG<sub>2k</sub>-P(LA/εDL)<sub>2k</sub></i>	24	38
<i>PEG<sub>2k</sub>-P(LA/CL)<sub>2k</sub></i>	27	26

<sup>a</sup> Z-average value

<sup>b</sup> Z-average value after 4 weeks