

Pulsed EPR Spectroscopy of ^{33}S -Labeled Molybdopterin in Sulfite Oxidase

Eric L. Klein,^{1,3} Abdel Ali Belaïdi,² Arnold M. Raitsimring,¹ Amanda C. Davis,¹ Tobias Krämer,³ Andrei V. Astashkin,¹ Frank Neese,³ Günter Schwarz,² John H. Enemark^{1*}

1. Department of Chemistry and Biochemistry, The University of Arizona, 1306 E. University Blvd., Tucson, AZ 85721-0041, USA
2. Institute of Biochemistry, University of Cologne, Zülpicher Str. 47, 50674 Cologne, Germany
3. Max Planck Institute for Chemical Energy Conversion, Stiftstr. 34-36, 45470 Mülheim an der Ruhr, Germany

*jenemark@email.arizona.edu

Phone: 520 621-2245

Fax: 520 621-8065

Supporting Information

Orbital Analysis for the Energy-Minimized Mo(V) DFT Model

The frontier orbital energy diagram for the optimized structure of the SO active site model is depicted in Figure S1, showing the canonical Kohn-Sham orbitals. The orbital scheme is typical for a Mo(V) center with d^1 configuration. Spin polarization effects strongly stabilize the majority spin orbitals such that the singly-occupied MO 168α is shifted below the highest occupied orbitals of mainly ligand character. As a result of strong intermixing between metal- and ligand-based orbitals, the d_{xy} -character (in the chosen coordinate system) is spread over orbitals 168α and 169α ($\sim 26\%$ Mo character in each orbital, see Table S1). In the former, the d_{xy} orbital mixes with the out-of-plane p_z orbitals of the MPT S atoms, while the latter mixes d_{xy} orbital character with the S^V orbital of cysteine.^{1,2} Similar interactions have been found for the DFT model $[\text{MoO}(\text{SCH}_3)(\text{dithiolene})\text{OH}]^-$ (dithiolene = SCHCHS)³ for example, which features the same inner-sphere coordination environment as SO. The empty β -spin d_{xy} orbital (172β) corresponds to the lowest unoccupied MO (LUMO) and is of predominantly Mo character (77%). Above the d_{xz} orbital lie the remaining unoccupied metal d -orbitals, having d_{yz} (173β), d_{xz} (174β), $d_{x^2-y^2}$ (176β), and d_z^2 (177β) character, respectively.

Additionally, four corresponding empty α -spin d-orbitals can be readily identified within the vacant α -spin manifold.

A chemically more intuitive picture can be obtained by visualizing the electronic structure in terms of localized molecular orbitals (LMOs). The Pipek-Mezey population-localization protocol yields 10 molecular orbitals of mainly d-character localized on the Mo center, shown in Figure S2. Note that the MO diagram focuses entirely on the metal-centered orbital manifold, omitting ligand-orbitals. Furthermore, the energy splitting is represented in a schematic way since all information about orbital energies is lost in the process. All 10 orbitals of predominantly metal d-character can be identified in the diagram, and from this the d_{xy} character of the singly occupied orbital becomes much more recognizable.

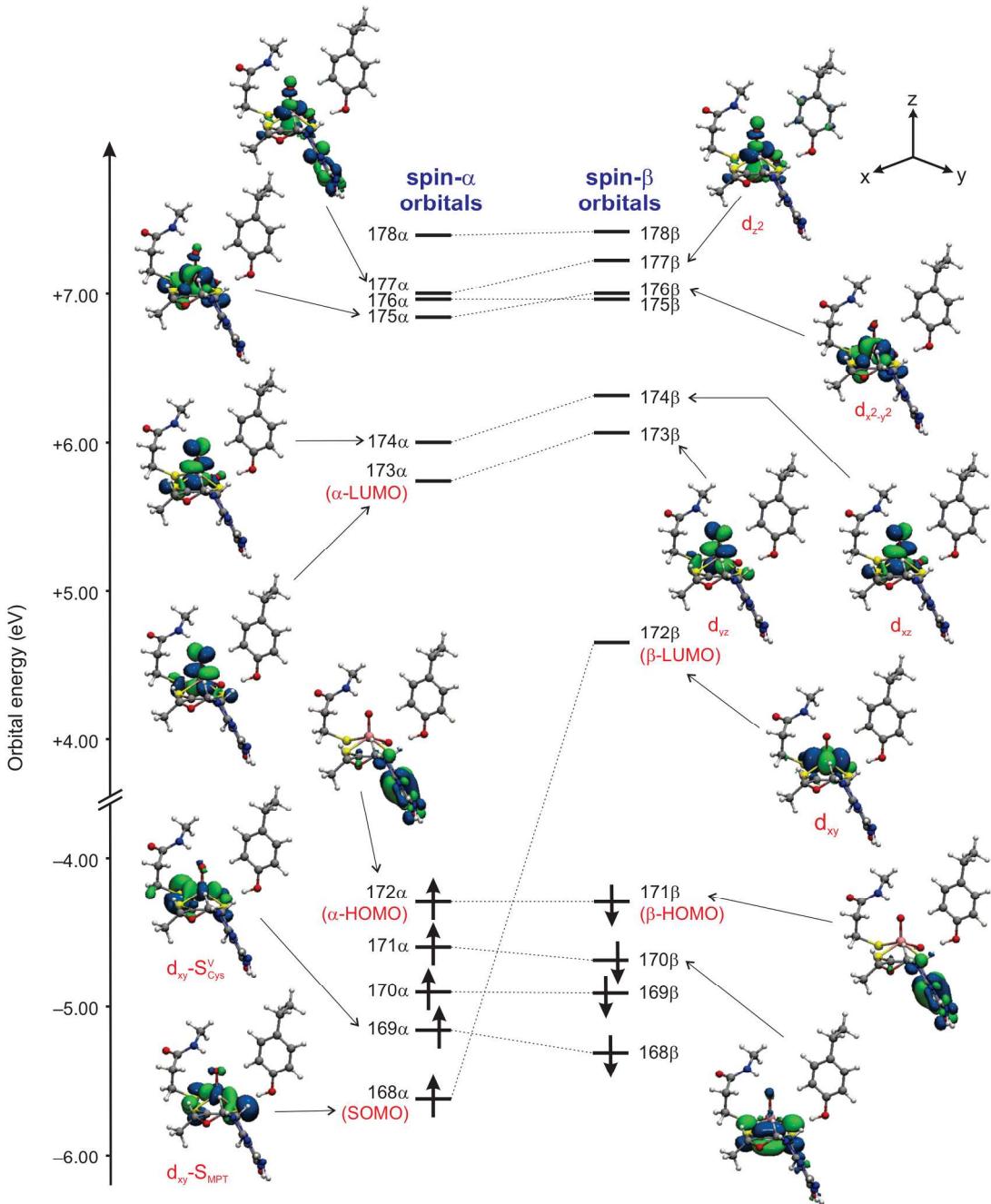


Figure S1. Canonical Kohn-Sham molecular orbital diagram (frontier orbital region) for the computational SO active site model (B3LYP/TZVP/ZORA). HOMO = Highest Molecular Orbital; LUMO = Lowest Unoccupied Molecular Orbital.

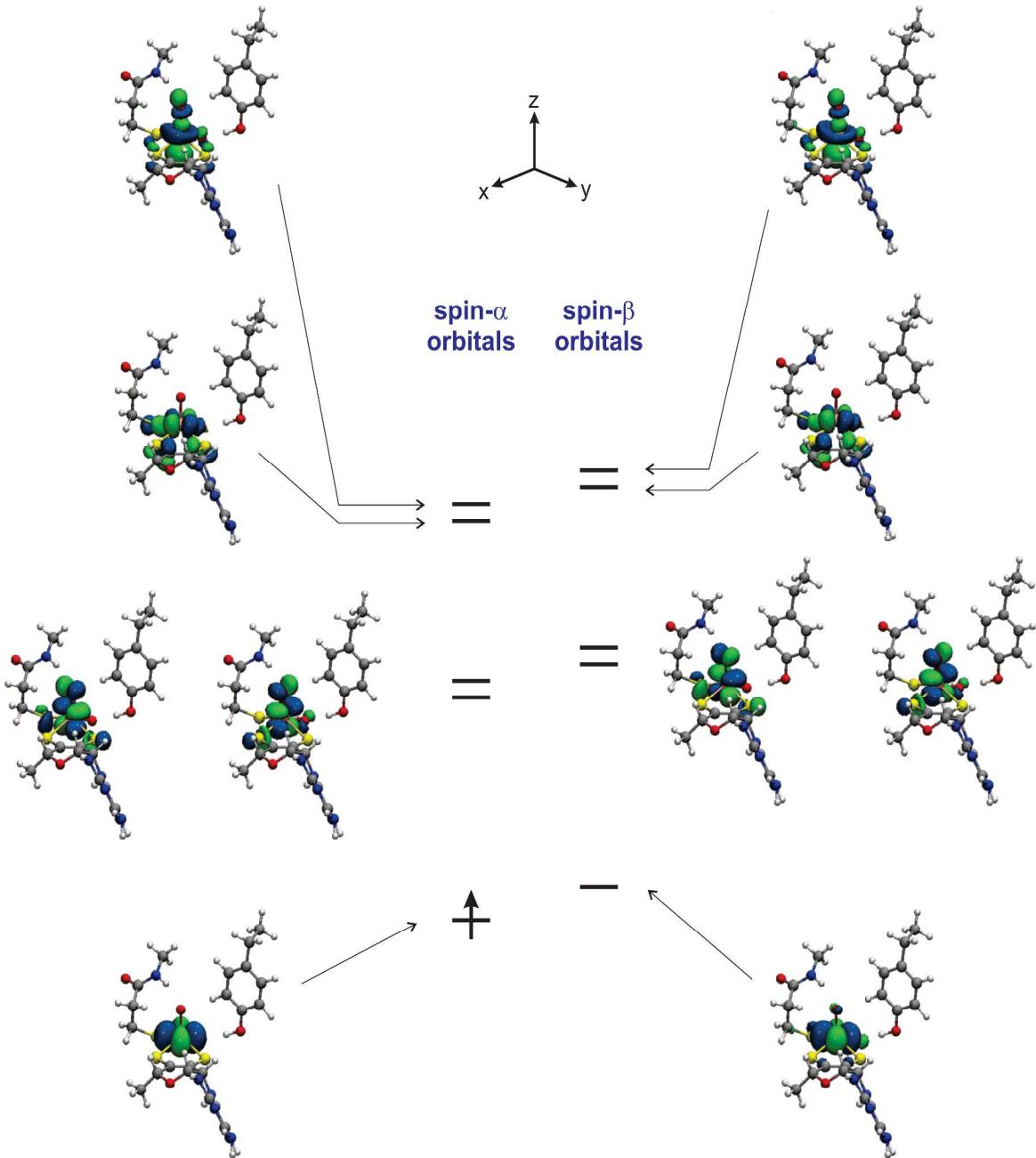


Figure S2. Pipek-Mezey localized molecular orbitals of the Mo *d*-orbital manifold of the computational SO active site model. Note that the diagram is a schematic representation of the ligand field splitting.

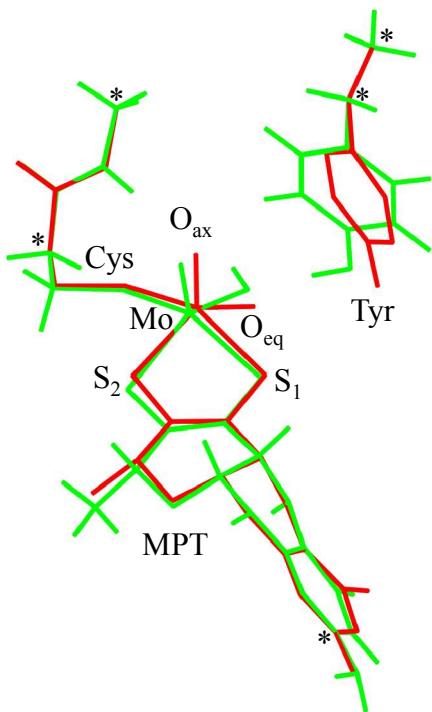


Figure S3. Overlay of the crystallographic atom coordinates (from pdb 1SOX) and the minimum-energy geometry optimized Mo(V) structure, including added H atoms (red and green structures, respectively). Atoms marked by asterisks were constrained to their relative crystallographic coordinate positions. S₁ and S₂ correspond to index numbers 30 and 32, respectively, in each of the DFT models (see below).

Table S1. Compositions of selected canonical KS orbitals in the frontier orbital region.^a

Spin- α	Center					
	Mo(36)	S(4)	S(30)	S(32)	O(16)	O(37)
168 α	26.4%	14.0%	28.0%	12.7%	4.3%	2.9%
169 α	26.8%	38.6%	10.6%	4.3%	2.2%	2.1%
173 α	59.0%	8.2%	8.6%	0.0%	0.0%	19.1%
174 α	61.5%	0.6%	0.8%	9.7%	4.5%	18.5%
175 α	41.5%	11.8%	8.6%	3.8%	4.2%	5.1%
177 α	34.1%	0.8%	31.0%	5.6%	3.5%	8.9%

Spin- β	Center					
	Mo(36)	S(4)	S(30)	S(32)	O(16)	O(37)
172 β	76.7%	6.4%	2.4%	2.6%	2.2%	0.5%
173 β	62.1%	7.4%	7.5%	0.0%	0.0%	17.5%
174 β	64.2%	0.6%	0.8%	9.0%	4.2%	16.4%
176 β	36.9%	9.9%	7.5%	4.7%	4.1%	2.0%
177 β	48.3%	1.8%	4.6%	5.9%	4.0%	13.0%

^aThe numbers in parentheses correspond to the atom coordinates of the DFT model (see below).

ORCA Input Files and Optimized Model Coordinates

Input used for geometry optimization calculations:

```
! BP86 TZVP RI TZV/J OPT GRID4 NormalPrint VDW10
%geom
Constraints
{ C 0 C }
{ C 6 C }
{ C 7 C }
{ C 8 C }
{ C 18 C }
# additional angle constraints for the models with dihedral angles defined at
# 120, 150, 180, 210, and 240 degrees. Atoms 68 and 69 are dummy atoms located
# between the dithiolene sulfurs and adjacent carbon atoms, as described within
# the Materials and Methods section.
{ A 36 68 69 C }
{ A 30 68 32 C }
{ A 29 69 31 C }
end
end
* xyz -1 2
<atom coordinates>
*
%cosmo
epsilon 4
end
```

Input used for all properties calculations:

```
! B3LYP TZVP ZORA COSMO TightSCF NPA GRID4 LargePrint VDW10
* xyz -1 2
<atom coordinates>
*
%cosmo
epsilon 4
end
%rel
    SOCType 3
    SOCFlags 1,3,3,0
    method ZORA
    modelpot 1,1,1,1
    picturechange true
    modeldens rhoZORA
end
%eprnmr
    gtensor 1
    Nuclei = all Mo {aiso, adip, aorb, rho, fgrad}
    Nuclei = 38 {aiso, adip, aorb, rho, fgrad}
    Nuclei = 17 {aiso, adip, aorb, rho, fgrad}
    Nuclei = 68 {aiso, adip, aorb, rho, fgrad}
    Nuclei = 53 {aiso, adip, aorb, rho, fgrad}
    Nuclei = 31 {aiso, adip, aorb, rho, fgrad}
    Nuclei = 33 {aiso, adip, aorb, rho, fgrad}
    Nuclei = 5 {aiso, adip, aorb, rho, fgrad}
end
```

Optimized atom coordinates:

Dihedral Angle = "relaxed"

C	-1.6330	2.8589	1.6059
C	-0.5316	3.5803	2.3827
O	-0.4775	4.8226	2.4282
C	-1.5257	3.0956	0.0964
S	-0.0422	2.3194	-0.7186
N	0.3581	2.7822	3.0276
C	1.5025	3.3233	3.7538
C	5.5613	-1.6784	5.8455
C	4.1389	-2.2351	5.5830
C	3.8534	-2.2162	4.0966
C	3.3564	-1.0577	3.4803
C	4.2123	-3.2897	3.2626
C	3.2330	-0.9571	2.0941
C	4.0886	-3.2111	1.8729
C	3.6029	-2.0363	1.2709
O	3.5176	-1.9897	-0.0886
O	1.8811	0.0000	-0.7847
N	-4.1717	-7.7399	-1.1233
C	-3.9495	-8.4585	-2.2020
N	-4.5112	-9.7163	-2.3188
N	-3.1297	-8.0373	-3.2105
C	-2.3905	-6.8255	-3.1749
O	-1.6836	-6.5077	-4.1504
N	-1.8619	-4.8941	-1.7725
C	-2.0087	-4.2296	-0.4701
C	-3.4228	-4.4100	0.1244
N	-3.7829	-5.7926	0.0959
C	-2.5858	-6.0802	-1.9688
C	-3.5109	-6.5438	-1.0346
C	-1.7091	-2.7561	-0.5839
S	0.0039	-2.3053	-0.7575
C	-2.7037	-1.8339	-0.5774
S	-2.3268	-0.0932	-0.6646
C	-4.1730	-2.2162	-0.4754
O	-4.3677	-3.6378	-0.6458
C	-5.0660	-1.5336	-1.5078
Mo	0.0000	0.0000	0.0000
O	0.0000	0.0000	1.7379
H	-2.5885	3.2917	1.9463
H	-1.6545	1.7849	1.8302
H	-2.4215	2.7142	-0.4114
H	-1.4396	4.1712	-0.1128
H	0.3276	1.7798	2.8221
H	1.1975	4.2360	4.2804
H	5.8044	-1.6767	6.9203
H	5.6463	-0.6467	5.4738
H	4.0656	-3.2618	5.9758
H	3.3971	-1.6253	6.1226
H	3.0657	-0.2051	4.1008
H	4.6004	-4.2102	3.7089
H	2.8392	-0.0480	1.6393
H	4.3600	-4.0566	1.2375
H	2.9515	-1.2034	-0.3755
H	-5.2754	-9.8740	-1.6653
H	-4.7140	-10.0448	-3.2603
H	-2.9684	-8.6205	-4.0315
H	-1.8823	-4.2570	-2.5695
H	-1.2997	-4.6876	0.2425
H	-3.4369	-4.0586	1.1720
H	-4.5968	-6.0861	0.6316

H	-4.5242	-1.9411	0.5424
H	-5.0410	-0.4435	-1.3793
H	-4.7282	-1.7761	-2.5251
H	-6.1018	-1.8790	-1.3821
H	1.8484	2.5811	4.4835
H	2.3345	3.5772	3.0769
H	6.3158	-2.2864	5.3247
H	2.2653	0.8785	-0.9614

Dihedral Angle = 120

C	-1.6921	2.9107	1.4729
C	-0.6236	3.5406	2.3737
O	-0.5315	4.7702	2.4983
C	-1.4101	3.1671	-0.0107
S	0.1136	2.3213	-0.6629
N	0.1915	2.6664	3.0308
C	1.3138	3.1384	3.8337
C	4.9720	-2.1356	5.9945
C	3.5441	-2.6083	5.6206
C	3.3589	-2.5089	4.1223
C	2.9682	-1.3000	3.5273
C	3.7334	-3.5596	3.2684
C	2.9690	-1.1284	2.1445
C	3.7277	-3.4116	1.8809
C	3.3569	-2.1854	1.2977
O	3.3982	-2.0744	-0.0522
O	1.9076	0.0000	-0.7338
N	-4.0184	-7.1352	1.9856
C	-4.3008	-8.0929	1.1401
N	-4.9899	-9.2354	1.5703
N	-3.8976	-8.0628	-0.1671
C	-3.1189	-7.0149	-0.7371
O	-2.8302	-7.0372	-1.9429
N	-1.9434	-4.9640	-0.1781
C	-1.6261	-3.8876	0.7643
C	-2.7064	-3.7802	1.8611
N	-2.9689	-5.0774	2.4234
C	-2.7574	-6.0071	0.2209
C	-3.2544	-6.0991	1.5173
C	-1.4978	-2.5620	0.0505
S	0.0046	-2.2902	-0.8693
C	-2.5052	-1.6421	0.1012
S	-2.3398	-0.0541	-0.6996
C	-3.8009	-1.8971	0.8490
O	-3.9243	-3.2604	1.3032
C	-5.0447	-1.6284	0.0050
Mo	0.0000	0.0000	0.0000
O	0.0000	0.0000	1.7335
H	-2.6428	3.4051	1.7320
H	-1.8099	1.8337	1.6547
H	-2.2666	2.8442	-0.6181
H	-1.2433	4.2417	-0.1750
H	0.1746	1.6837	2.7348
H	0.9950	4.0057	4.4265
H	5.1553	-2.1893	7.0807
H	5.1249	-1.0965	5.6680
H	3.3952	-3.6464	5.9609
H	2.7985	-1.9869	6.1430
H	2.6609	-0.4637	4.1625
H	4.0386	-4.5175	3.7015
H	2.6542	-0.1818	1.7067
H	4.0027	-4.2370	1.2222

H 2.8965 -1.2444 -0.3370
 H -5.3330 -9.0951 2.5197
 H -5.7429 -9.5326 0.9493
 H -4.0773 -8.8451 -0.7944
 H -1.9622 -4.7356 -1.1694
 H -0.6687 -4.1062 1.2697
 H -2.3394 -3.1086 2.6589
 H -3.5848 -5.0839 3.2348
 H -3.8099 -1.2277 1.7373
 H -5.0863 -0.5745 -0.2989
 H -5.0238 -2.2597 -0.8939
 H -5.9413 -1.8713 0.5922
 H 1.6407 2.3309 4.5018
 H 2.1643 3.4507 3.2041
 H 5.7233 -2.7567 5.4846
 H 2.2812 0.8876 -0.8852

Dihedral Angle = 150

C -1.5758 2.9381 1.5716
 C -0.4472 3.5790 2.3847
 O -0.3243 4.8096 2.4572
 C -1.4111 3.1734 0.0680
 S 0.0527 2.3239 -0.7017
 N 0.3801 2.7130 3.0384
 C 1.5298 3.1948 3.7964
 C 5.3233 -2.0450 5.8012
 C 3.8842 -2.5300 5.4918
 C 3.6288 -2.4462 4.0040
 C 3.1689 -1.2569 3.4203
 C 3.9937 -3.4933 3.1413
 C 3.0888 -1.1009 2.0379
 C 3.9101 -3.3608 1.7554
 C 3.4653 -2.1549 1.1823
 O 3.4239 -2.0623 -0.1696
 O 1.8892 0.0000 -0.7719
 N -4.4523 -7.5467 -1.2104
 C -4.3391 -8.1895 -2.3450
 N -5.0484 -9.3797 -2.5591
 N -3.4959 -7.7764 -3.3406
 C -2.6458 -6.6360 -3.2460
 O -1.9397 -6.3038 -4.2090
 N -1.9134 -4.9012 -1.7076
 C -1.9970 -4.2088 -0.4197
 C -3.3964 -4.3892 0.2073
 N -3.7792 -5.7750 0.1815
 C -2.7323 -5.9815 -1.9688
 C -3.6516 -6.4484 -1.0342
 C -1.6985 -2.7338 -0.5638
 S 0.0101 -2.2981 -0.7956
 C -2.6930 -1.8096 -0.5397
 S -2.3307 -0.0663 -0.6626
 C -4.1549 -2.1937 -0.3872
 O -4.3653 -3.6166 -0.5171
 C -5.0690 -1.5460 -1.4240
 Mo 0.0000 0.0000 0.0000
 O 0.0000 0.0000 1.7343
 H -2.5046 3.4370 1.8950
 H -1.6806 1.8648 1.7809
 H -2.3112 2.8405 -0.4677
 H -1.2609 4.2457 -0.1244
 H 0.3279 1.7213 2.7857
 H 1.2391 4.0918 4.3586

H 5.5521 -2.0885 6.8792
 H 5.4554 -1.0076 5.4602
 H 3.7581 -3.5659 5.8480
 H 3.1575 -1.9099 6.0415
 H 2.8679 -0.4248 4.0640
 H 4.3529 -4.4357 3.5668
 H 2.7193 -0.1703 1.6084
 H 4.1781 -4.1832 1.0900
 H 2.9087 -1.2347 -0.4330
 H -5.7177 -9.5232 -1.8039
 H -5.5063 -9.4391 -3.4690
 H -3.3748 -8.3227 -4.1919
 H -1.5680 -4.3808 -2.5102
 H -1.2672 -4.6483 0.2849
 H -3.3606 -4.0494 1.2600
 H -4.6446 -5.9975 0.6706
 H -4.4819 -1.8854 0.6307
 H -5.0489 -0.4525 -1.3296
 H -4.7330 -1.8198 -2.4334
 H -6.0980 -1.9036 -1.2780
 H 1.8564 2.4097 4.4906
 H 2.3717 3.4647 3.1367
 H 6.0571 -2.6654 5.2655
 H 2.2571 0.8768 -0.9818

Dihedral Angle = 180

C -1.6844 2.8868 1.6207
 C -0.6005 3.5537 2.4739
 O -0.5518 4.7845 2.6067
 C -1.4663 3.1430 0.1249
 S 0.0589 2.3568 -0.5859
 N 0.2860 2.7115 3.0786
 C 1.4190 3.2345 3.8361
 C 5.3443 -1.8972 5.8668
 C 3.9168 -2.4194 5.5641
 C 3.6560 -2.3685 4.0759
 C 3.1854 -1.1951 3.4694
 C 4.0135 -3.4346 3.2340
 C 3.0862 -1.0728 2.0847
 C 3.9110 -3.3359 1.8466
 C 3.4526 -2.1467 1.2494
 O 3.3924 -2.0938 -0.1040
 O 1.8957 0.0000 -0.7402
 N -3.6754 -6.7564 -4.5861
 C -3.0410 -7.0798 -5.6845
 N -3.5345 -8.0906 -6.5232
 N -1.8487 -6.5056 -6.0375
 C -1.1825 -5.5094 -5.2683
 O -0.1181 -5.0151 -5.6686
 N -1.3500 -4.2534 -3.1919
 C -1.9276 -4.0403 -1.8717
 C -3.4527 -4.2819 -1.9314
 N -3.7444 -5.5318 -2.5791
 C -1.8743 -5.1926 -4.0475
 C -3.0812 -5.8291 -3.7709
 C -1.6703 -2.6403 -1.3654
 S -0.0350 -2.3189 -0.7682
 C -2.6315 -1.6881 -1.3898
 S -2.3166 -0.0636 -0.7401
 C -4.0225 -1.9520 -1.9439
 O -4.0973 -3.2151 -2.6412
 C -4.4973 -0.9012 -2.9436

Mo	0.0000	0.0000	0.0000
O	0.0000	0.0000	1.7351
H	-2.6383	3.3540	1.9148
H	-1.7592	1.8084	1.8182
H	-2.3274	2.7790	-0.4544
H	-1.3524	4.2223	-0.0524
H	0.2876	1.7286	2.7882
H	1.0749	4.0443	4.4930
H	5.5723	-1.9188	6.9454
H	5.4540	-0.8626	5.5096
H	3.8143	-3.4516	5.9378
H	3.1761	-1.8077	6.1045
H	2.8891	-0.3488	4.0965
H	4.3817	-4.3654	3.6769
H	2.7058	-0.1537	1.6408
H	4.1750	-4.1730	1.1982
H	2.8855	-1.2711	-0.3942
H	-4.4571	-8.3818	-6.2024
H	-3.5577	-7.8389	-7.5121
H	-1.3311	-6.8291	-6.8531
H	-0.5069	-3.7569	-3.4637
H	-1.5123	-4.7612	-1.1386
H	-3.8397	-4.3225	-0.8946
H	-4.7275	-5.7977	-2.5876
H	-4.7321	-1.9872	-1.0866
H	-4.5708	0.0830	-2.4636
H	-3.7812	-0.8343	-3.7738
H	-5.4820	-1.1898	-3.3381
H	1.8494	2.4245	4.4387
H	2.1986	3.6448	3.1724
H	6.0923	-2.5092	5.3411
H	2.3046	0.8847	-0.7607

Dihedral Angle = 210

C	-1.6302	2.9796	1.4550
C	-0.5407	3.6350	2.3111
O	-0.4553	4.8666	2.4111
C	-1.3627	3.1713	-0.0436
S	0.1681	2.3238	-0.6594
N	0.3106	2.7809	2.9507
C	1.4660	3.2844	3.6868
C	5.1528	-1.9228	5.9581
C	3.7071	-2.3963	5.6616
C	3.4625	-2.3618	4.1700
C	3.0801	-1.1709	3.5354
C	3.7688	-3.4622	3.3525
C	3.0192	-1.0645	2.1475
C	3.7010	-3.3799	1.9617
C	3.3334	-2.1733	1.3361
O	3.3051	-2.1381	-0.0179
O	1.9219	0.0000	-0.6925
N	-2.0696	-5.0380	-6.7831
C	-1.0667	-5.1430	-7.6175
N	-1.2034	-5.8893	-8.7948
N	0.1629	-4.6007	-7.3641
C	0.4769	-3.8501	-6.1912
O	1.6133	-3.3808	-6.0392
N	-0.5164	-2.9776	-4.1347
C	-1.4305	-3.2979	-3.0384
C	-2.8616	-3.4196	-3.5996
N	-2.8920	-4.3301	-4.7089
C	-0.6349	-3.7389	-5.2893

C	-1.8368	-4.3623	-5.6120
C	-1.4060	-2.2251	-1.9798
S	-0.0518	-2.2984	-0.8455
C	-2.3648	-1.2626	-1.9332
S	-2.3646	-0.0355	-0.6487
C	-3.4971	-1.1903	-2.9458
O	-3.3425	-2.1397	-4.0259
C	-3.6153	0.1736	-3.6229
Mo	0.0000	0.0000	0.0000
O	0.0000	0.0000	1.7285
H	-2.5738	3.4889	1.7091
H	-1.7463	1.9118	1.6894
H	-2.2117	2.7930	-0.6315
H	-1.2227	4.2407	-0.2588
H	0.2855	1.7914	2.6859
H	1.1627	4.1522	4.2868
H	5.3783	-1.9309	7.0376
H	5.3022	-0.9006	5.5805
H	3.5625	-3.4165	6.0536
H	2.9887	-1.7446	6.1854
H	2.8276	-0.2968	4.1432
H	4.0694	-4.4063	3.8177
H	2.7095	-0.1309	1.6797
H	3.9248	-4.2421	1.3313
H	2.8504	-1.2905	-0.3324
H	-2.1788	-6.1507	-8.9307
H	-0.8218	-5.4403	-9.6274
H	0.9557	-4.7754	-7.9797
H	0.4373	-2.7408	-3.8605
H	-1.1763	-4.2704	-2.5658
H	-3.5179	-3.8133	-2.7982
H	-3.8038	-4.5320	-5.1119
H	-4.4474	-1.4190	-2.4127
H	-3.8456	0.9507	-2.8834
H	-2.6631	0.4220	-4.1105
H	-4.4117	0.1394	-4.3803
H	1.8445	2.4915	4.3448
H	2.2736	3.6059	3.0075
H	5.8777	-2.5735	5.4469
H	2.3214	0.8878	-0.7572

Dihedral Angle = 240

C	-1.6237	3.0005	1.4545
C	-0.5544	3.6764	2.3183
O	-0.4983	4.9091	2.4251
C	-1.3355	3.1805	-0.0428
S	0.2057	2.3341	-0.6311
N	0.3166	2.8389	2.9547
C	1.4643	3.3624	3.6892
C	5.3092	-1.7672	5.8730
C	3.8783	-2.2793	5.5694
C	3.6283	-2.2543	4.0792
C	3.1911	-1.0820	3.4452
C	3.9679	-3.3453	3.2621
C	3.1042	-0.9858	2.0579
C	3.8758	-3.2734	1.8723
C	3.4474	-2.0867	1.2467
O	3.3891	-2.0597	-0.1048
O	1.8932	0.0000	-0.7813
N	0.5941	-2.4700	-7.6904
C	1.8430	-2.3317	-8.0573
N	2.2229	-2.5865	-9.3819

N	2.8364	-2.0020	-7.1771	H	5.4333	-0.7410	5.4970
C	2.6238	-1.7556	-5.7868	H	3.7612	-3.3029	5.9624
O	3.5749	-1.4414	-5.0595	H	3.1412	-1.6485	6.0929
N	0.8609	-1.6614	-4.0943	H	2.9136	-0.2154	4.0530
C	-0.3783	-2.2815	-3.6361	H	4.3120	-4.2746	3.7272
C	-1.4697	-2.0592	-4.7008	H	2.7553	-0.0665	1.5886
N	-1.0099	-2.5053	-5.9845	H	4.1250	-4.1293	1.2427
C	1.2498	-1.9202	-5.3984	H	2.8819	-1.2373	-0.4082
C	0.3122	-2.2876	-6.3602	H	1.3951	-2.7170	-9.9617
C	-0.8191	-1.6807	-2.3254	H	2.8461	-1.8858	-9.7836
S	-0.0018	-2.3071	-0.8852	H	3.8137	-1.9877	-7.4645
C	-1.7720	-0.7023	-2.2959	H	1.6029	-1.5558	-3.4012
S	-2.3481	-0.0129	-0.7531	H	-0.2639	-3.3775	-3.4963
C	-2.4149	-0.1568	-3.5617	H	-2.3637	-2.6500	-4.4172
O	-1.8164	-0.6712	-4.7731	H	-1.6788	-2.4658	-6.7498
C	-2.3116	1.3639	-3.6739	H	-3.4878	-0.4558	-3.5445
Mo	0.0000	0.0000	0.0000	H	-2.8922	1.8456	-2.8775
O	0.0000	0.0000	1.7219	H	-1.2622	1.6708	-3.5706
H	-2.5770	3.5005	1.6896	H	-2.6955	1.6842	-4.6530
H	-1.7288	1.9329	1.6944	H	1.8240	2.5971	4.3894
H	-2.1749	2.7883	-0.6350	H	2.2869	3.6459	3.0111
H	-1.1991	4.2486	-0.2673	H	6.0541	-2.3976	5.3650
H	0.3111	1.8512	2.6869	H	2.2722	0.8854	-0.9393
H	1.1551	4.2580	4.2430				
H	5.5290	-1.7704	6.9536				

References

- (1) McNaughton, R. L.; Mondal, S.; Nemykin, V. N.; Basu, P.; Kirk, M. L. *Inorg. Chem.* **2005**, *44*, 8216.
- (2) McNaughton, R. L.; Tipton, A. A.; Rubie, N. D.; Conry, R. R.; Kirk, M. L. *Inorg. Chem.* **2000**, *39*, 5697.
- (3) Hernandez-Marin, E.; Seth, M.; Ziegler, T. *Inorg. Chem.* **2009**, *48*, 2880.