## Three-dimensional structures of aspartate carbamoyltransferase from *Escherichia coli* and of its complex with cytidine triphosphate

(allosteric behavior/enzyme complex/protein crystallography/x-ray diffraction/conformational changes)

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ABSTRACT X-ray diffraction studies to nominal resolutions of 3.0 Å for unliganded aspartate carbamoyltransferase (EC 2.1.3.2) (R32 crystal symmetry) and of 2.8 Å for the complex of aspartate carbamov/transferase with cytidine triphosphate (P321 crystal symmetry) have yielded traces of the polypeptide chains of the catalytic (C) and regulatory (R) chains in the hexameric C<sub>6</sub>R<sub>6</sub> molecules. The independent molecular structures of the liganded and unliganded forms of the enzyme are very nearly identical. In the regulatory chain there is a CTP-binding domain that interacts with an adjacent regulatory subunit and a zinc-binding domain that interacts with the catalytic subunit. In the catalytic chain a polar domain shows interactions be-tween adjacent pairs of C chains to form each trimer C<sub>3</sub> while an equatorial domain shows intramolecular C3-C3 interactions. The active site is at or near the interface between adjacent C chains within the trimers. Probably each active center involves amino acid residues from adjacent C chains.

Aspartate carbomyltransferase (EC 2.1.3.2) catalyzes the formation of carbamoyl-L-aspartate from L-aspartate and carbamoylphosphate (1, 2). In Escherichia coli this reaction is the first unique step in the biosynthesis of pyrimidines. The endproduct of this pathway, cytidine triphosphate (CTP) inhibits this allosteric enzyme by interacting with a regulatory chain R, which is distinct from the catalytic chain C (3, 4). Also ATP, a purine, stimulates this enzyme so that a tendency toward balance in the pyrimidine and purine syntheses is achieved. The enzyme is hexameric (5, 6), and can be dissociated by use of mercurials into two catalytic trimers C<sub>3</sub> and three regulatory dimers R<sub>2</sub> (7). Molecular weights are 34,000 for C and 17,000 for R (5, 8, 9). Each regulatory unit contains one  $Zn^{2+}$ , which has been presumed to coordinate to the four sulfhydryl groups of this chain (10, 11), and which is necessary for reconstitution of the enzyme from the subunits  $C_3$  and  $R_2$ . Other properties of this enzyme have been summarized recently (12).

Two crystallographic forms of molecular aspartate carbamoyltransferase have been studied at low resolution in this laboratory. Crystals of R32 symmetry of the unliganded form (containing neither allosteric effectors nor substrate analogues) show at 5.5-Å resolution that the molecular symmetry is  $D_{3}$ , that the two catalytic trimers are almost eclipsed when viewed along the molecular threefold axis, and that there is a large aqueous cavity in the center of the molecule (13). Crystals of symmetry P321 of the form grown in the presence of CTP showed that there are no gross structural differences when compared to the unliganded R32 form (14). The additional electron density associated with CTP was verified by a further differencedensity study in which 5-iodo-CTP was bound to the enzyme. In this brief paper we summarize x-ray diffraction studies at 3.0-Å resolution for the unliganded R32 form and at 2.8-Å resolution for the CTP-P321 form. The major features of secondary structure are presented below.

## STRUCTURE DETERMINATION

The unliganded R32 form has one RC group per asymmetric unit in a unit cell of dimensions a = 132 Å and c = 199 Å, while the CTP-P321 form has two RC units per crystallographic asymmetric unit in a unit cell of dimensions a = 122 Å and c = 142 Å. The x-ray diffraction data were collected by use of the oscillation method (15) from the native enzyme, and from heavy atom derivatives listed in Table 1. Conditions of preparation of these derivatives have been given previously (13, 14), except for the third CTP-P321 derivative, for which crystals were immersed in 5 mM KAu(CN)<sub>2</sub> and 40 mM K<sub>2</sub>HPO<sub>4</sub> at pH 5.8 for two days. Heavy atom coordinates were found in difference electron density maps, starting from the phases obtained at low resolution, and were refined by alternate cycles of refinement of phases and least squares parameters (13, 14, 16). For the unliganded R32 form we used initially only those 5452 independent reflections that were observed in the native protein and in all four derivatives, and in later refinements phased 9518 independent reflections. In the P321 form we employed those 21,983 independent reflections present in the native crystals and at least two heavy atom derivatives. In all, we phased 74% of the theoretical number of reflections to 3.0 Å for R32, and 76% to 2.8 Å for P321 crystals.

In the CTP-P321 form there is a noncrystallographic twofold axis that was obeyed to observable accuracy for the  $C_3$  trimer and the Zn domain (but not the CTP-binding domain) of the  $R_2$  dimer. The phases of this electron density map were further improved by flattening the solvent density and averaging the molecular density, except in the CTP-binding domain, about this twofold axis in direct space, back transforming to reciprocal space, and combining the calculated phases with the observed structure factors for 24,238 independent reflections (17, 18).

From these final electron density maps a model was built in a Richards optical comparator (19) on a scale of 1.25 cm/Å.

## DESCRIPTION OF THE MOLECULE

In Fig. 1 we show the gross molecular structure in  $D_3$  symmetry. Three catalytic monomers are closely associated to form each trimer, and the two nearly eclipsed  $C_3$  units are in close contact with each other in both the unliganded R32 and the CTP-P321 forms. Each catalytic monomer C is associated with a C monomer in the opposite  $C_3$  subunit by interactions through a regulatory dimer. The two C units thus related through a given  $R_2$  subunit are 120° apart about the threefold axis (Fig. 1).

In more detail, we show a  $C_3R_3$  unit in Fig. 2 as the half of Fig. 1 closer to the reader. The polar domain (containing helices H1–H4 and a five-stranded parallel  $\beta$  sheet) of the C chain is closer to the threefold axis than is the equatorial domain (containing helices H6–H8 and a six-stranded parallel  $\beta$  sheet),

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Abbreviations: C, catalytic chain of aspartate carbamoyltransferase; R, regulatory chain; MSA, 2-acetoxymercuri-4-nitrosalicylaldehyde.

	No. of		Root mean square*						
Derivative	sites		14.0–4.5 Å	4.5–3.8 Å	3.8–3.3 Å	3.3–3.0 Å	3.0–2.8 Å	All data	
MSA	3	Fc	380	279	234	186	—	301	
		Ε	165	142	122	92		140	
KAu(CN) <sub>2</sub>	3	Fc	150	117	97	78		122	
	-	E	101	107	95	93		100	
K <sub>2</sub> Pt(NO <sub>2</sub> ) <sub>4</sub>	9	Fc	280	185	142	119	·	218	
	Ū	E	223	216	187	218	_	212	
IrBr <sub>3</sub>	7	Fc	285	166	115	88		212	
	•	E	170	170	190	245	-	181	
P321									
MSA	6	Fc	157	126	112	97	86	127	
		Ε	83	76	66	. 62	58	72	
$UO_2(NO_3)_2$	12	Fc	205	153	136	117	99	159	
		E	103	105	97	79	67	96	
KAu(CN) <sub>2</sub>	5	Fc	80	64	60	55	49	66	
		Ε	67	62	57	56	<b>59</b>	61	
No. of reflections (unliganded R32)			3539	2120	2395	1464	_	9,518	
No. of reflections									
(CTP-P321)			6711	4196	4889	3526	2661	21,983	
Average figure of merit (unliganded R32)			0.738	0.647	0.604	0.528	_	0.652	
Average figure of merit (CTP-P321)			0.668	0.629	0.602	0.562	0.523	0.611	

Table 1.	Heavy atom	derivatives and	refinement	criteria
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MSA, 2-acetoxymercuri-4-nitrosalicylaldehyde.

\* rms  $Fc = |\Sigma_h f_{hj}^2/n|^{1/2}$ , in which  $f_{hj}$  is the heavy atom scattering amplitude for reflection h of derivative j and n is the number of reflections. The unit is electrons. The rms structure factor of the native protein is 1055 electrons for the unliganded R32 form and 552 electrons for the CTP-P321 form. rms  $E = |\Sigma_h e_{hj}^2/n|^{1/2}$ , in which  $e_{jh}$  is the lack of closure for reflection h of derivative j and n is the number of reflections. The unit is also electrons. The figure of merit is the mean value of the cosine of the phase angle error.

which is closer to the equatorial region containing the twofold axis. These domains are connected by helices H5 and H9, which is the COOH-terminal helix. Both the NH<sub>2</sub>-terminus and the COOH terminus of the C chain are in the polar domain. These elements of secondary structure are well defined throughout the C chain, perhaps somewhat better in the polar domain, and have allowed us to fit the partial sequence (kindly supplied by



FIG. 1. Subunit assembly of two  $C_3$  and three  $R_2$  subunits in aspartate carbamoyltransferase. Each larger sphere represents a catalytic monomer C, and each smaller sphere represents a regulatory monomer R. The view is along the threefold axis, and the twofold axes are indicated by the usual symbol. This choice of enantiomer has been made on the basis of right-handed  $\alpha$ -helices in the high-resolution structures.

W. Konigsberg) to the polar domain and to helices H5 and H9. The C–C interactions in the  $C_3$  subunit are between helix H2 of one C chain and helix H3 and strand S3 of the adjacent C chain. Interactions between the two  $C_3$  subunits occur between the loops of chain connecting S7 to H7 and S9 to S10.

In the regulatory chain, the fitting of the known sequence (20) is complete in the Zn domain (residues 102–152), but only partially complete in the CTP-binding domain (residues 1-96) because of the intrinsic disorder in the electron density maps. The CTP-binding domain includes helices H1' and H2' and a four-stranded  $\beta$  sheet. The CTP site, indicated in Fig. 2, is just to the left of this  $\beta$  sheet in the region of electron density that is present in the CTP-P321 form but absent in the unliganded R32 form. The approximately 4-Å distance of this "CTP density" from the four-stranded  $\beta$  sheet favors the previously preferred (14) assignment of the regulatory region. The part of this additional density associated with the pyrimidine ring has been tentatively assigned on the basis of the iodine position as determined at low resolution (14) from the 5-iodo-CTP derivative of this CTP-P321 crystal form. However, details of the binding of CTP are not yet clear because of some disorder in this region of the molecule. Strands S2' and S3' are connected, but we are unsure whether there are some 4 or 5 amino acids or as many as 15 amino acids in this region. The former choice would require that the extra 10 or 11 amino acids be added at the NH2-terminus in a disordered region. Hence, we have to



C<sub>3</sub>R<sub>3</sub> of the molecule of aspartate carbamoyltransferase. (Lower) Labeled RC unit. The CTP domain of the R chain is furthest to the left, while the Zn domain is adjacent and connected. The chain drawn slightly heavier is the catalytic monomer, in which the polar domain contains helices H1-H4 and the equatorial domain contains helices H6-H8. The active site is near the thiol pocket and the CTP (second site) crevice at a boundary between adjacent C monomers within a  $C_3$ trimer. Helices and strands of  $\beta$  sheet have been numbered in order from the NH<sub>2</sub>-terminus (N) of the C chain, and similarly, with primes added, for the R chain.



FIG. 3. Drawing of the R (left) and C (right) polypeptide chains in one RC unit of the hexameric  $C_6R_6$  molecule. About 306 amino acids (±1% or so) are indicated in the catalytic chain in the electron density map.

leave a decision on this part of the trace for a later study. The four-stranded  $\beta$  sheet (S1'–S4') of one regulatory chain is related to a similar sheet of the other regulatory chain of the dimer R<sub>2</sub> by a molecular twofold axis in such a way that an eight-stranded sheet is formed. Other structures that contribute to the stability of the dimer R<sub>2</sub> are helix H1' and the chain that connects it to S2', which are in contact with the chain that, in the other R unit, connects S2' to S3'. Of course, a similar reverse interaction occurs, related by the molecular twofold axis.

Also in the regulatory chain, the Zn domain is dominated by the tetrahedral bonding of Zn<sup>2+</sup> to cysteines 109, 114, 137, and 140 and the four-stranded antiparallel  $\beta$  sheet (S5'-S8'). The amino acid sequence can be reasonably well fitted to our electron density maps from Lys-56 to the COOH-terminal Asn-152. In particular, the Zn domain is a comparatively well-ordered region, and its contact with the polar domain of the C chain is well defined. Indeed, all of the substructures (e.g., loops ending at Cys residues) defined by the coordination of Zn<sup>2+</sup> to the four Cys residues of the R chain participate in the interactions between R and C chains. It is therefore clear why mercurials that react with these Cys residues of the R chain dissociate the C<sub>6</sub>R<sub>6</sub> molecule into smaller fragments. In particular, the regions of R chain from Ser-112 to Ser-121, and the Lys-138-Tyr-139 pair (in Cys... Cys loops) are in very close contact in the C chain with helix H4 and that region immediately before it in sequence. In addition, Glu-141 and Lys-142 of the R chain interact with another part of the C chain in the region before H4 in the amino acid sequence. A more detailed drawing of the R and C chains is shown in Fig. 3.

Chemical modification of Lys-84 (21, 22) destroys catalytic activity. We have placed this residue towards the NH2-terminal direction at the beginning of helix H3. Also, chemical modification of the only Cys residue of the C chain suggests that it is near the active site: modification by small uncharged groups leaves the enzyme active (23) while modification by 2-chloromercuri-4-nitrophenol inactivates the enzyme (24). Also, oxidation of this C-chain thiol to sulfonate by potassium permanganate inactivates the enzyme (25). From the mercury derivatives (Table 1) we have located this thiol in a pocket between polar  $\beta$  sheet and helix H3 of one monomer and helix H2 of the adjacent monomer. This "thiol pocket" can accommodate an organomercurial such as MSA, and it opens most directly into the central cavity of the hexameric C<sub>6</sub>R<sub>6</sub> molecule. Nearby, there is a crevice, in which a second 5-iodo-CTP molecule binds (14); this CTP crevice (Fig. 2) is between strands S6, S9, and S10 of the equatorial domain and helices H2, H5, and H9, all within the same C chain. The CTP crevice also opens into the central cavity. Helix H2 is at the interface between these two cavities, forming one boundary of the CTP crevice within the same C chain, and also of the thiol pocket of the adjacent C chain.

The portion of regulatory chain that is closest to these pockets is the part of the large Cys-114–Cys-137 loop that is nearer Cys-114. This region of R is in contact with helix H4 and in close proximity with helix H3. In the thiol pocket, adjacent to helix H3, reaction of this C-chain thiol with MSA also involves modification of Lys-84 (26). The nearby CTP crevice (Fig. 2) may also be a candidate for a binding region for substrates or inactivating reagents. If so, the active site appears to be shared between adjacent C chains within a given catalytic trimer  $C_3$ , perhaps in a manner resembling that suggested (27) for glutathione reductase.

Our description here indicates that there is very little difference between the electron density of the unliganded molecule (R32) and the CTP-liganded molecule (P321), in spite of the different buffers (Tris/imidazole and citrate), their concentrations (0.05/0.05 and 0.04 M), and their pH values (6.35and 5.8), respectively. In fact, the major difference is density ascribed to CTP in the P321 crystals as compared with no electron density in the corresponding regions of the R32 crystals. It is likely that this conformation is therefore of biochemical significance. However, conclusions with respect to activity are to be made very cautiously, because this is not the active conformation of aspartate carbamoyltransferase.

We have grown two crystal forms of the complex of aspartate carbamoyltransferase with its putative transition state analogue N-phosphonacetyl-L-aspartate. The crystal symmetry of one of them is R32 with unit cell dimensions a = 123 Å and c = 709Å. The other crystalline form of this derivative of the enzyme has the space group P321 and unit cell dimensions of a = 122Å and c = 157 Å. One can argue from these data that the symmetry of this enzyme when bound to molecules of this substrate analogue is  $D_3$  from the former and at least  $C_3$  from the latter crystals. If, as sedimentation ultracentrifuge experiments suggest (28), there is a large conformational change, it probably is at least an elongation of the molecule along the threefold axis, a change that is consistent with these unit cell measurements but that is not proved by them.

With this paper, we recognize Prof. E. Havinga on the occasion of his 70th birthday anniversary.

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- Jones, M. E., Spector, L. & Lipman, F. (1955) J. Am. Chem. Soc. 77, 819–820.
- 2. Reichard, P. & Hanshoff, G. (1956) Acta Chem. Scand. 10, 548-566.

- Yates, R. A. & Pardee, A. B. (1956) J. Biol. Chem. 221, 757– 770.
- Gerhart, J. C. & Pardee, A. B. (1962) J. Biol. Chem. 237, 891– 896.
- 5. Weber, K. (1968) Nature (London) 218, 1116-1119.
- 6. Wiley, D. C. & Lipscomb, W. N. (1968) Nature (London) 218, 1119-1121.
- 7. Gerhart, J. C. & Schachman, H. K. (1965) Biochemistry 4, 1054-1062.
- Meighen, E. A., Pigiet, V. & Schachman, H. K. (1970) Proc. Natl. Acad. Sci. USA 65, 234-241.
- Rosenbusch, J. P. & Weber, K. (1971) J. Biol. Chem. 246, 1644–1657.
- Rosenbusch, J. P. & Weber, K. (1971) Proc. Natl. Acad. Sci. USA 68, 1019–1023.
- Nellbach, M. E., Pigiet, V. P., Gerhart, J. C. & Schachman, H. K. (1972) *Biochemistry* 11, 315–327.
- Jacobson, G. R. & Stark, G. R. (1973) in *The Enzymes*, ed. Boyer, P. D. (Academic, New York), Vol. 9, pp. 225–308.
- Warren, S. G., Edwards, B. F. P., Evans, D. R., Wiley, D. C. & Lipscomb, W. N. (1973) Proc. Natl. Acad. Sci. USA 70, 1117– 1121.
- Edwards, B. F. P., Evans, D. R., Warren, S. G., Monaco, H. L., Landfear, S. M., Eisele, G., Crawford, J. L., Wiley, D. C. & Lipscomb, W. N. (1974) Proc. Natl. Acad. Sci. USA 71, 4437– 4441.
- 15. Crawford, J. L. (1977) Dissertation (Harvard University, Cambridge, MA).
- Dickerson, R. E., Kopka, M. L., Varnum, J. C. & Weinzierl, J. E. (1967) Acta Crystallogr. 23, 511–522.
- 17. Fletterich, R. J. & Steitz, T. A. (1976) Acta Crystallogr. A32, 125-131.
- 18. Bricogne, G. (1976) Acta Crystallogr. A32, 832-847.
- 19. Richards, F. M. (1968) J. Mol. Biol. 37, 225-230.
- 20. Weber, K. (1968) Nature (London) 218, 1116-1119.
- 21. Greenwell, P., Jewett, S. L. & Stark, G. R. (1973) J. Biol. Chem. 248, 5994-6001.
- Kempe, T. D. & Stark, G. R. (1975) J. Biol. Chem. 250, 6861– 6869.
- Vanaman, T. C. & Stark, G. R. (1970) J. Biol. Chem. 245, 3565–3573.
- Evans, D. R., McMurray, C. H. & Lipscomb, W. N. (1972) Proc. Natl. Acad. Sci. USA 69, 3638–3642.
- 25. Benisek, W. F. (1971) J. Biol. Chem. 246, 3151-3959.
- 26. Landfear, S. M. (1978) Dissertation (Harvard University, Cambridge, MA).
- 27. Schulz, G. E., Schirmer, R. H., Sachsenheimer, W. & Pai, E. F. (1978) Nature (London) 273, 120-124.
- Gerhart, J. C. & Schachman, H. K. (1968) Biochemistry 7, 538–552.
- Britton, E. G., Lipscomb, J. S. & Pique, M. E. (1978) Computer Graphics 12 (3), 222–227.