The NADPH binding site on beef liver catalase

(catalase structure/bound NADPH/mammalian and fungal comparison)

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ABSTRACT Beef liver and human erythrocyte catalases (EC 1.11.1.6) bind NADP tenaciously [Kirkman, H. N. & Gaetani, G. F. (1984) Proc. Natl. Acad. Sci. USA 81, 4343-4348]. The position of NADP on beef liver catalase corresponds to the carboxyl-terminal polypeptide hinge in Penicillium vitale fungal catalase, which connects the common catalase structure to the additional flavodoxin-like domain. In contrast to nearly all other known structures of protein-bound NADP, NAD, and FAD, the NADP molecule of beef liver catalase is folded into a right-handed helix and bound, in part, in the vicinity of the carboxyl end of two α -helices. A water molecule (W7) occupies a pseudosubstrate site close to the C4 position of the nicotinamide and is hydrogen bonded to His-304. Although the NADP and heme groups approach each other to within 13.7 Å, there is no direct interaction. The function of the NADP remains a mystery.

Kirkman and Gaetani (1) have recently shown that tetrameric human and beef catalase (hydrogen-peroxide:hydrogenperoxide oxidoreductase, EC 1.11.1.6) binds four molecules of NADPH with a dissociation constant less than 1×10^{-8} M. While they make some suggestions as to the function of the bound dinucleotide, there is yet no true understanding of its physiological significance. Although the NADP can be removed only by a series of partially denaturing agents, it does not seem to be involved in the catalatic or peroxidatic activity.

The crystal structure of beef liver catalase (BLC) has been determined (2, 3) and more recently refined (unpublished results). The crystal structure of the fungal (*Penicillium vitale*) catalase (PVC) has also been determined (ref. 4; unpublished results) and was seen to be very similar to BLC apart from an additional carboxyl-terminal domain that has a flavodoxin-like structure. The "hinge" polypeptide, between the common catalase structure and the flavodoxin domain, binds into an external cleft of the molecule.

A significant volume of uninterpreted density occurred in the refined BLC electron density map at the equivalent position occupied by the hinge in PVC. This had tentatively been interpreted in terms of a carboxyl-terminal extension of BLC, consistent with the occasional occurrence of peptides in the digested protein (5). These peptides were homologous to a carboxyl-terminal extension of beef erythrocyte catalase (Table 1). However, when it was suggested (1) that NADP is bound to catalase, then the interpretation of the extra density as NADP was immediately obvious (Fig. 1). Its approximate position on the enzyme surface, corresponding to the hinge region leading to the flavodoxin domain of PVC, is shown in Fig. 2. The model of NADP was subsequently refined with respect to the observed structure amplitudes, giving an overall final R factor of 19.1% for 33,255 reflections extending to 2.5-Å resolution. The coordinates of the NADP

Table 1.	Erroneous carboxyl-terminal sequence extension of
BLC by a	alignment with bovine erythrocyte catalase (BEC)

Catalase	Sequence				
BLC [Schroeder et al. (5)]	500 -N-E-	505 -E-K-P-K-N(V,E,	H,G,S,H	515 I) (V-H-T-Y)	
			Th10	Thll	
BLC (x-ray) BEC [Schroeder	-N	not visible	н	– V–H–T–Y–A	
et al. (5)]	-N-E-	-E-K-P-K-N (V-E-	H-G-S-H) (V-H-T-Y,A)	
PVC [Vainshtein et al (4) x-ray]	A	495 A-A-A-A-A-A-A-A-A-A-A-A-A-A-A-A-A-A-A-	500 - A - A - A - A	505 - A-A-A-A	

The peptides Th10 and Th11 occurred only in occasional isolates of BLC. The position of the extra density of BLC had been erroneously interpreted as the peptide H-V-H-T-Y-A in a position corresponding to the hinge between the common catalase structure and the flavodoxin-like domain in PVC.

prosthetic group are given in Table 2 and are also deposited with the Brookhaven Protein Data Bank.

Conformation of NADP when bound to catalase

It has been proposed (6, 7), on the basis of NMR studies, that at neutral pH and room temperature, NAD⁺ exists in equilibrium between two folded forms and an extended form (Fig. 3). The folded forms make either left- or right-handed helices and are considered to be of lower energy than the extended form. Nevertheless, the structures of NAD [lactate dehydrogenase (9, 10), alcohol dehydrogenase (11, 12), malate dehydrogenase (13), glyceraldehyde-3-phosphate dehydrogenase (14, 15), L-3-hydroxyacyl-coenzyme A dehydrogenase (16), phosphorylase (2 sites) (17), and sorbitol dehydrogenase (18)], NADP [dihydrofolate reductase (19), glutathione reductase (20), p-hydroxybenzoate hydroxylase (by comparison with glutathione reductase) (21, 22), and 6phosphogluconate dehydrogenase (23, 24)] and FAD [glutathione reductase (25) and *p*-hydroxybenzoate hydroxylase (26)] (reduced or oxidized) have an extended conformation (base separation of at least 14 Å) when bound to enzyme in nearly all cases. The overall conformations of the bound dinucleotides are all very similar (cf. ref. 27). The only exception is that of sorbitol dehydrogenase, for which an NMR study (18) produces the paradoxical result that the base separation is merely 8.5 Å, although the enzyme is homologous to liver alcohol dehydrogenase.

Free NAD⁺ has been crystallized as a lithium salt (28) and as a free acid (29). The conformation of the dinucleotide in these two crystal structures differs greatly from the common conformation found when bound to enzymes. Indeed, the free acid NAD⁺ structure is partially folded, with the minimum distance between bases of only 9.6 Å.

When dinucleotides are bound to enzymes, they are al-

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Abbreviations: BLC, beef liver catalase; PVC, *Penicillium vitale* catalase; DHFR, dihydrofolate reductase.



FIG. 1. Two stereo views of the "extra density" in an electron density map of BLC, in which the two crystallographically independent subunits had been averaged. The structure of NADP is superimposed. The BLC map is dependent on phases calculated from the refined molecule (R = 19.1%) for which the extra density had been erroneously interpreted as a H-V-H-T-Y-A peptide (Table 1). Nevertheless, the fit to NADP is excellent and far superior to the erroneous peptide assumption.

most invariably associated with an α/β structure (a "nucleotide-binding" fold). The cofactor is then bound to the carboxyl end of strands in a parallel β -pleated sheet and to the amino end of parallel α -helices. The structural significance of this fold is probably due to the electrostatic dipole moment exerted by the α -helices on the negatively charged phosphates (30). The structure itself may be the result of divergent evolution from a primordial mononucleotide-binding fold (31, 32) or convergent evolution to a stable supersecondary structure (33). The only exception is that of 6-phospho-

Table 2. Coordinates of the NADP molecule bound to BLC

Atom	<i>P</i> , Å	<i>Q</i> , Å	<i>R</i> , Å	Atom	P, Å	Q, Å	<i>R</i> , Å	Atom	<i>P</i> , Å	Q, Å	<i>R</i> , Å
N1A	24.1	6.6	34.3	OP3	26.7	14.9	32.4	01'N	31.7	7.5	31.9
C2A	24.6	7.2	33.1	C3'A	29.0	11.9	32.8	C4'N	32.8	6.6	32.1
N3A	25.5	8.1	33.0	O3'A	29.0	12.6	31.3	C3'N	32.4	5.2	31.6
C4A	26.0	8.6	34.2	C4'A	29.6	10.5	32.6	O3'N	31.9	4.3	32.5
C5A	25.6	8.2	35.5	01'A	28.7	9.6	33.2	C2'N	31.3	5.6	30.6
C6A	24.6	7.1	35.5	C5'A	31.0	10.3	33.1	02'N	30.5	4.5	30.4
N7A	26.3	9.0	36.5	O5'A	31.7	11.6	33.0	C1'N	30.6	6.7	31.3
C8A	27.1	9.9	35.8	PA	33.2	11.6	33.3	N1N	30.1	7.7	30.6
N9A	26.9	9.6	34.5	OP1A	33.3	11.4	34.7	C2N	31.0	8.2	29.7
N6A	24.1	6.6	36.6	OP2A	33.9	12.8	32.7	C3N	30.6	9.3	28.9
C1'A	27.4	10.1	33.3	0	33.5	10.2	32.6	C4N	29.3	9.8	29.0
C2'A	27.6	11.6	33.1	PN	34.6	9.5	31.7	C5N	28.5	9.2	29.9
O2'A	26.8	12.4	32.4	OP1N	35.7	9.1	32.5	C6N	28.8	8.1	30.7
Р	26.2	13.7	33.2	OP2N	35.1	10.5	30.6	C7N	31.5	10.0	27.8
OP1	26.6	13.7	34.6	O5'N	33.7	8.4	31.0	N7N	32.8	9.5	27.7
OP2	24.7	13.6	33.2	C5'N	34.0	7.1	31.5	07N	31.1	10.9	27.2

Coordinates are given with respect to the P, Q, R molecular diad axes (3) for the "standard" subunit. In the designations of the atoms, an A as the last character refers to adenine; an N, nicotinamide.



FIG. 2. Diagrammatic view of one subunit of BLC, showing the position of the bound NADP molecule on the enzyme. A C_{α} backbone diagram that includes a view of the bound NADP and heme groups will be published elsewhere.

gluconate dehydrogenase, which is predominantly an α structure, with the possibly extended NADP coenzyme apparently binding in the proximity of the carboxyl end of helices VIb and XII (23).

NADP bound to catalase is not extended but rather is folded into a right-handed helix, although the bases are approximately perpendicular rather than parallel (Fig. 1). The nicotinamide ring is roughly parallel to the adenine ribose, with a closest approach of 3.6 Å. The binding site on the protein is toward the carboxyl ends of helices $\alpha 5$ and $\alpha 10$ (Fig. 2). The bases are close to these two helices, while the phosphates are furthest away from them.

The conformational angles [cf. ref. 9 for their definition; in addition, the dihedral angle θ'_a relates to the C2'(a)-O2'(a) bond giving the position of the O2' bound phosphate] of the coenzyme are shown in Table 3, where they are compared with the well-refined extended NADP structure found in Lactobacillus casei dihydrofolate reductase (DHFR) (19) as well as with the tentative, low-resolution, structure of NADP in 6-phosphogluconate dehydrogenase (24). The adenosine nucleoside structures of NADP bound to BLC and to DHFR are very similar, with the O2' phosphate positioned in the sterically favored trans orientation relative to the ribose ring. The puckering of the adenosine ribose ring in BLC·NADP is not clear since the C1'-C2'-C3'-C4' dihedral angle ψ is only 5°. The nicotinamide mononucleoside structures in NADP of BLC and DHFR are also similar except that BLC·NADP is syn at the glycosidic bond, whereas the DHFR NADP structure is anti. The puckering of the nicotinamide ribose ring in DHFR·NADP is given by $\psi = -36^{\circ}$ (C2' endo). The major differences in these two NADP structures are the conformational angles of the pyrophosphate

groups, producing a folded conformation in one case and extended in the other. Thus viewing down the P_a-P_n line shows the 5' oxygens nearly *cis* to each other in BLC·NADP and nearly *trans* in DHFR·NADP. The orientation of the carboxyamide group, while clearly planar with the nicotinamide ring, can be determined only tentatively by consideration of the hydrogen bonding environment. For BLC·NADP, the N7 position has been assigned *trans* to N4 of the nicotinamide. The orientation of this group in small molecule structures is *trans* for the NAD Li salt (28) and *cis* in the NAD free acid (29) and in a structure of nicotinamide (35).



FIG. 3. Folded and extended forms of NAD^+ in aqueous solution [adapted from Saenger *et al.* (8)].

Table 3.	Conformations	of	NADP a	and	NAD	molecules

		Dihedral angle, °						
Angle	Defining atoms	NADP catalase	NADP in <i>L. casei</i> DHFR*	NADP in 6PDH [†]	NAD of LDH·NAD· pyruvate in M ₄ LDH [‡]			
Xa	C4(A), N9(A), C1'(A), C2'(A)	139	105	100	89			
Ribos	e a conformation	C2' endo?	C3' endo	?	C3' endo			
θ'_{a}	C1'(A), C2'(A), O2'(A), P	125	145	-102				
ξa	C3'(A), C4'(A), C5'(A), O5'(A)	-25	-169	-169	-78			
θ_{a}	C4'(A), C5'(A), O5'(A), P(A)	-173	-163	-134	-159			
ψ_{a}	C5'(A), O5'(A), P(A), O	37	-72	110	109			
ϕ_{a}	O5'(A), P(A), O, P(N)	144	141	-156	-156			
$\phi_{\rm n}$	O5'(N), P(N), O, P(A)	-161	82	-160	49			
$\psi_{\rm n}$	C5'(N), O5'(N), P(N), O	-109	57	109	87			
 θ_{n}	C4'(N), C5'(N), O5'(N), P(N)	118	129	180	-171			
ξ'n	C3'(N), C4'(N), C5'(N), O5'(N)	-54	-172	-88	-53			
Ribos	e n conformation	C2' endo	C2' endo	?	C3' endo			
Xn	C2(N), N1(N), C1'(N), C2'(N)	-54	117	-81	106			
Short	est distance between	3.9 Å	12 Å	14 Å	14.0 Å			
ade	nine and nicotinamide	[C6(A) - N3(N)]	[C8(A)-C6(N)]		[C6(A)-C2(N)]			
α	N9(A), P(A), P(N), N1(N)		170	?	58			

The convention for the dihedral angle nomenclature is that of Arnott and Hukins (34). The angles are positive for right-handed rotations (that is, the far end of any bond is rotated clockwise relative to the near end). The zero angle of rotation is the *cis* conformation for successive bonds in the chain. *Filman *et al.* (19).

[†]Data for 6-phosphogluconate dehydrogenase from Abdallah et al. (24).

[‡]Data for lactate dehydrogenase from Chandrasekhar *et al.* (9).

Environment of NADP in BLC

A stereo view of the interacting residues and water molecules is shown in Fig. 4 and described in Table 4. The adenine moiety binds to a hydrophobic surface, as is usual in other NAD and NADP binding proteins. The O2' phosphate interacts with Arg-202, Asp-212, and Lys-236. Indeed, it had been noted by Murthy et al. (3) that Lys-236 had no apparent charge balance. This situation has now been resolved by the interpretation of the extra density as NADP. The pyrophosphate group interacts with His-304, which also interacts with the C4 position of the nicotinamide and the "substrate" water W7. There is extensive electron density between the adenine phosphate and His-304. This might represent a strong interaction (the distance between the phosphate oxygen and the imidazole nitrogen is 2.5 Å) and relate to the dissociation constant of less than 10^{-8} M (1). In contrast, the pyrophosphate moiety of many NAD-dependent dehydrogenases interacts with arginine residues (cf. ref. 14). Both OH groups of the nicotinamide ribose form hydrogen bonds with the protein. The orientation of the nicotinamide is fixed primarily by hydrogen bonding interactions with the carboxyamide group.

The interaction of Lys-236 and Arg-202 with the O2' phosphate of NADP explains why NADP binds better than NAD (1) and why the uncharged reduced cofactors bind better than the charged oxidized cofactors. The position of the dinucleotide in the cleft occupied by the hinge region in PVC also suggests why NADP does not bind as well to human erythrocyte catalase as to BLC. The former has a carboxylterminal extension (Table 1) that is likely to compete for the binding of NADP in the same cleft.

Of particular interest is the water molecule W7, which is positioned about 4 Å from the reactive C4 position of the nicotinamide relative to the A side (pro-R) of the ring. Furthermore, this water is within 3.5 Å of the N_e atom of His-304. Thus, the water is in a suitable "substrate" position for a redox reaction of a dehydrogenase. There appears to be



FIG. 4. Stereo view of NADP environment, showing hydrogen bonding and position of the W7 water molecule.

NADP mojety	BLC residue	Interaction	Distance, Å; comment
Adenine	Ser-200	0 -N6A	2.9
	Phe-197	0, 1011]
	Arg-202	Aliphatic	Hydrophobic
	Phe-445	chain	binding area
	Val-449		U U
Adenine	Arg-202	$N_{n2}-OP2$	2.7
ribose	Lys-236	$N_e - OP3$	3.5
Pyro-	His-304	N _o -OP2A	2.9
phosphate	His-304	N-OP2N	3.0
	His-304	N _o -OP2N	3.5
Nicotinamide	His-193	$N_{e} - O2'N$	2.6
ribose	Gln-441	0-03'N	2.8
	Thr-444	0 ₂ -03'N	3.6
	Phe-197	•	
	Val-301		Hydrophobic
	Phe-445		pocket
	Val-449]
Nicotinamide	Val-301	C _β -N7N	3.2
	Trp-302	O-N7N	2.9
	His-304	C _e -C4N	4.0
"Substrate"	Tyr-214	$O_{\eta}-W7$	2.8
water (W7)	Lys-236	N _e -W7	4.4
	His-304	Ν _ε -W7	3.5
	Adenine ribose	O3'A-W7	2.9
	Nicotin- amide	C4N-W7	3.8

Table 4 NADP environment in BLC

some feedback control on its binding, since the His-304 interacts both with this water substrate and the phosphates.

There is no direct interaction of the heme group and NADP. The closest approach of these two moieties is 13.7 Å between atoms C4N on NADP and the vinyl methyl group of pyrrole II of the heme group. The heme iron atom is 19.5 Å from the C4 atom of the nicotinamide ring. The β -sheet of the β -barrel domain lies between these two prosthetic groups and, in particular, the polypeptide $\beta 4$ is interposed. It is noteworthy that the essential Asn-147 is situated on β 4. The β -barrel domain of catalase consists of two halves. Each has four consecutive antiparallel β -sheets, $\beta 1$, $\beta 2$, $\beta 3$, and $\beta 4$ and $\beta 5$, $\beta 6$, $\beta 7$, and $\beta 8$. The binding of the heme group is effected primarily through residues on the $\beta 1-\beta 4$ section together with the "essential helix" $\alpha 9$ in the wrapping domain. The NADP molecule is bound primarily by residues in the $\beta 5 - \beta 8$ barrel section (including $\alpha 5$) together with helix $\alpha 10$ in the α -helical domain.

There is no evidence for bound NADP in fungal catalases as observed by binding experiments to Aspergillus niger (1) or in the electron density map of PVC. Indeed, the BLC NADP site is occupied by the hinge polypeptide in PVC (Table 1). The presence of the additional flavodoxinlike $(\alpha/\beta$ structure) domain in PVC might suggest that its function is similar to the NADP-binding function in BLC. Yet the position of the presumed FMN in PVC would be different from that of the NADP site observed in BLC (unpublished results). Furthermore, the α/β structures would select an extended FMN formation rather than the folded NADP structure found on BLC.

Conclusions

The site and conformation of NADP bound to BLC have been clearly established. Its relative closeness to the heme pocket suggests a relationship to the catalase activity, although no obvious function can be suggested. The folded conformation of the NADP is quite unlike the extended conformations found in numerous other structures. Its site closer to the carboxyl end of α -helices (with the phosphates furthest from the helix ends) is the converse to the well-characterized situation in many dehydrogenases or reductases. The position of a water molecule near the C4 carbon atom of the nicotinamide and its relationship to His-304 suggest a redox function.

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- 1. Kirkman, H. N. & Gaetani, G. F. (1984) Proc. Natl. Acad. Sci. USA 81, 4343-4348.
- Reid, T. J., III, Murthy, M. R. N., Sicignano, A., Tanaka, N., Musick, 2. W. D. L. & Rossmann, M. G. (1981) Proc. Natl. Acad. Sci. USA 78, 4767-4771.
- 3. Murthy, M. R. N., Reid, T. J., III, Sicignano, A., Tanaka, N. & Ross-Maining, M. R. (1981) J. Mol. Biol. 152, 465-499. Vainshtein, B. K., Melik-Adamyan, W. R., Barynin, V. V., Vagin,
- 4. A. A. & Grebenko, A. I. (1981) Nature (London) 293, 411-412.
- Schroeder, W. A., Shelton, J. R., Shelton, J. B., Robberson, B., Apell, 5. G., Fang, R. S. & Bonaventura, J. (1982) Arch. Biochem. Biophys. 214, 397-421
- 6. Catterall, W. A., Hollis, D. P. & Walter, C. F. (1969) Biochemistry 8, 4032-4036.
- Blumenstein, M. & Raftery, M. A. (1973) Biochemistry 12, 3585-3590. Saenger, W., Reddy, B. S., Mühlegger, K. & Weimann, G. (1977) in 8. Pyridine Nucleotide-Dependent Dehydrogenases, ed. Sund, H. (de Gruyter, Berlin), pp. 222-236.
- Chandrasekhar, K., McPherson, A., Jr., Adams, M. J. & Rossmann, 9. M. G. (1973) J. Mol. Biol. 76, 503-518.
- 10. Grau, U. M., Trommer, W. E. & Rossmann, M. G. (1981) J. Mol. Biol. 151, 289-307.
- 11. Abdallah, M. A., Biellmann, J. F., Nordström, B. & Brändén, C. I. (1975) Eur. J. Biochem. 50, 475–481.
- 12. Cedergren-Zeppezauer, E., Samama, J. P. & Eklund, H. (1982) Biochemistry 21, 4895-4908
- 13. Webb, L. E., Hill, E. J. & Banaszak, L. J. (1973) Biochemistry 12, 5101-5109.
- Moras, D., Olsen, K. W., Sabesan, M. N., Buehner, M., Ford, G. C. & 14. Rossmann, M. G. (1975) *J. Biol. Chem.* **250**, 9137–9162. Biesecker, G., Harris, J. I., Thierry, J. C., Walker, J. E. & Wonacott,
- 15. A. J. (1977) Nature (London) 266, 328-333.
- Holden, H. M. & Banaszak, L. J. (1983) J. Biol. Chem. 258, 2383-2389. 16. Stura, E. A., Zanotti, G., Babu, Y. S., Sansom, M. S. P., Stuart, D. I., 17.
- Wilson, K. S., Johnson, L. N. & Van de Werve, G. (1983) J. Mol. Biol. 170. 529-565.
- 18. Gronenborn, A. M., Clore, G. M. & Jeffery, J. (1984) J. Mol. Biol. 172, 559-572.
- Filman, D. J., Bolin, J. T., Matthews, D. A. & Kraut, J. (1982) J. Biol. 19. Chem. 257, 13663-13672.
- 20. Thieme, R., Pai, E. F., Schirmer, R. H. & Schulz, G. E. (1981) J. Mol. Biol. 152, 763-782.
- 21. Wierenga, R. K., de Jong, R. J., Kalk, K. H., Hol, W. G. J. & Drenth, J. (1979) J. Mol. Biol. 131, 55-73.
- Wierenga, R. K., Drenth, J. & Schulz, G. E. (1983) J. Mol. Biol. 167, 22 725-739.
- Adams, M. J., Archibald, I. G., Bugg, C. E., Carne, A., Gover, S., Helliwell, J. R., Pickersgill, R. W. & White, S. W. (1983) EMBO J. 2, 23. 1009-1014.
- Abdallah, M. A., Adams, M. J., Archibald, I. G., Biellmann, J. F., Helliwell, J. R. & Jenkins, S. E. (1979) Eur. J. Biochem. 98, 121-130. 24.
- Schulz, G. E., Schirmer, R. H. & Pai, E. F. (1982) J. Mol. Biol. 160, 25. 287-308
- Hofsteenge, J., Vereijken, J. M., Weijer, W. J., Beintema, J. J., Wier-26. enga, R. K. & Drenth, J. (1980) Eur. J. Biochem. 113, 141-150.
- Rossmann, M. G., Liljas, A., Brändén, C. I. & Banaszak, L. J. (1975) in 27 The Enzymes, ed. Boyer, P. D. (Academic, New York), 3rd Ed., Vol. 11. pp. 61-102.
- Reddy, B. S., Saenger, W., Mühlegger, K. & Weimann, G. (1981) J. 28. Am. Chem. Soc. 103, 907-914.
- Parthasarathy, R. & Fridey, S. M. (1984) Science 226, 969-971. 29
- Hol, W. G. J., van Duijnen, P. T. & Berendsen, H. J. C. (1978) Nature 30. (London) 273, 443-446. Rossmann, M. G., Moras, D. & Olsen, K. W. (1974) Nature (London) 31.
- 250, 194-199.
- 32. Ohlsson, I., Nordström, B. & Brändén, C. I. (1974) J. Mol. Biol. 89, 339-354
- 33. Rao, S. T. & Rossmann, M. G. (1973) J. Mol. Biol. 76, 241-256.
- Arnott, S. & Hukins, D. W. L. (1969) Nature (London) 224, 886–888. Wright, W. B. & King, G. S. D. (1954) Acta Crystallogr. 7, 283–288. 34.
- 35.