

Check neglect of convection of enthalpy terms

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In[549]:= Remove["Global`*"]

In[550]:= Θvar = 298.15; (* Θair *)
ψvar = 0; (* ψr *)

In[552]:= R = 8.3145; (*gas constant Joules per mole per Kelvin*)

In[553]:= Patm = 1.013 × 10^5; (*atm pressure in Pa*)

In[554]:= Al = .7(* .1263 oakair fraction, say 0.5 just mesophyll*)
Av = 1 - Al; (*Area fraction tissue in leaf from 2011 data 12.63%*)

Out[554]= 0.7

In[556]:= Cpv = 28.03; (* Heat Capacity water vapor J/mol at 100 c wiki *)
Cpl = 74.53; (* Heat Capacity water liquid J/mol at 100 or 25 c wiki*)

In[558]:= c[ψ_, θ_] := 1.28  $\frac{R \cdot 298.15}{Patm} \text{Exp}\left[-\frac{44000}{R} \left(\frac{1}{θ} - \frac{1}{298.15}\right)\right]$ 
 $\text{Exp}\left[\frac{\psi \cdot 18.07}{R \cdot \theta}\right]$  (*mol fraction for psi in MPa*)

In[559]:= Cvθ = θ_ c[ψvar, θ] /. θ → Θvar (* linearization of dx/dT 1/k *)
Cvψ = ψ_ c[ψ, Θvar] /. ψ → ψvar (* linearization of dx/dpsi 1/mpa *)
Out[559]= 0.00186474

Out[560]= 0.000228328

In[561]:= kvh = .026;(* heat cond air J m-2 s-1 K-1*)

In[562]:= klh = .28614;(* heat cond tissue*)

In[563]:= λ = 44000 - 43 (Θvar - 298.15) (* Joules per Mol*)

Out[563]= 44000.

In[564]:= L = .72 × 131 × 10^(-6) (*-20 10^(-6)*)
Out[564]= 0.00009432

In[565]:= SRsun = 400
Out[565]= 400

In[566]:= Q = 0.2 * SRsun / L (* w m-3 volumetric heat source,
.2 is fraction of total abs solar rad abs in spongy *)
Out[566]= 848 176.

In[567]:= Dv =  $\frac{Patm}{R \Theta var} 2.13 * \left(\frac{\Theta var}{273.15}\right)^{1.8} *$ 
 $10^{-5}$  (* cDv for mol frac driving force out of leaf *)
Out[567]= 0.00101901
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Groups

$$\Piψ = 1 + \frac{Al \cdot klh + Av \cdot kvh}{λ \cdot Av \cdot Dv \cdot Cvθ} + \frac{Al \cdot klh + Av \cdot kvh}{λ \cdot Al \cdot kl} \frac{Cvψ}{Cvθ}$$

$$\Piθ = 1 + \frac{λ \cdot Al \cdot kl}{Al \cdot klh + Av \cdot kvh} \frac{Av \cdot Dv \cdot Cvθ}{Al \cdot kl + Av \cdot Dv \cdot Cvψ}$$

$$\frac{1}{Piψ} + \frac{1}{Piθ}$$

$$\Deltaθ = - \frac{Q \cdot L^2}{Piθ (Al \cdot klh + Av \cdot kvh)}$$

(* Here neg as char gradients must have opp sign by mol con *)

$$\Deltaψ = - \frac{Q \cdot L^2}{Piψ λ \cdot Al \cdot kl}$$

Terms

$$\text{CondTerm} = \frac{Al \cdot klh + Av \cdot kvh}{λ \cdot Al \cdot kl} \left(\frac{Δθ}{Δψ} \right)$$

$$\text{MolTerm} = -1$$

$$\text{MixTerm} = \left(\frac{Cpl \cdot Δθ}{λ} + \frac{Cpv \cdot Av \cdot Dv \cdot Cvψ \cdot Δθ}{λ \cdot Al \cdot kl} \right)$$

$$\text{SqTerm} = \frac{Cpv \cdot Av \cdot Dv \cdot Cvθ \cdot Δθ^2}{λ \cdot Al \cdot kl} \frac{Δψ}{Δψ}$$

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In[579]:= SourceTerm =  $\frac{\Omega L^2}{\lambda A l k l \Delta \psi}$ 
Out[579]= 10.4707
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Noting that temp and psi char grads opposite signs, treat temp as neg rel to source

```
In[580]:= Econ = CondTerm + MolTerm - MixTerm + SqTerm + SourceTerm
Out[580]= 0.00008236

In[581]:= MixTerm - SqTerm
Out[581]= -0.00008236

In[582]:= CondTerm + MolTerm + SourceTerm
Out[582]= 1.77636 × 10-15
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