

Molecular recognition of fluorine impacts substrate selectivity in the fluoroacetyl-CoA thioesterase FIK

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Supplementary Figures

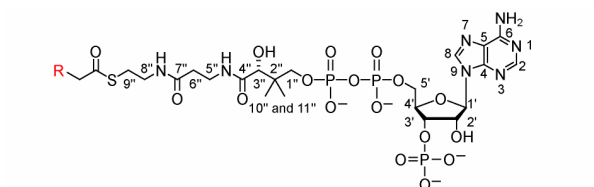
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Table S1. Acyl-CoA numbering scheme for 1D NMRs and $^1\text{H}/^{13}\text{C}$ HMBC crosspeaks.



<i>Crosspeak numbering for HMBC spectra</i>							
1	$\text{H}_{1''}\text{C}_{2''}$	10	$\text{H}_{10''}\text{C}_{11''}$	19	$\text{H}_{5''}\text{C}_{7''}$	28	H_2C_4
2	$\text{H}_{1''}\text{C}_{3''}$	11	$\text{H}_{10''}\text{C}_{2''}$	20	$\text{H}_{5''}\text{C}_{6''}$	29	H_2C_6
3	$\text{H}_{1''}\text{C}_{11''}$	12	$\text{H}_{10''}\text{C}_{1''}$	21	$\text{H}_{6''}\text{C}_{7''}$	30	H_8C_4
4	$\text{H}_{1''}\text{C}_{10''}$	13	$\text{H}_{10''}\text{C}_{3''}$	22	$\text{H}_{6''}\text{C}_{5''}$	31	H_8C_5
5	$\text{H}_3''\text{C}_{11''}$	14	$\text{H}_{11''}\text{C}_{10''}$	23	$\text{H}_{6''}\text{C}_{7''}$	32	H_1C_2
6	$\text{H}_3''\text{C}_{10''}$	15	$\text{H}_{11''}\text{C}_{2''}$	24	$\text{H}_{8''}\text{C}_{9''}$	33	H_1C_4
7	$\text{H}_3''\text{C}_{2''}$	16	$\text{H}_{11''}\text{C}_{1''}$	25	$\text{H}_{9''}\text{C}_{1\text{RAC}}$	34	$\text{H}_1\text{C}_{4'}$
8	$\text{H}_3''\text{C}_{1''}$	17	$\text{H}_{11''}\text{C}_{3''}$	26	$\text{H}_{9''}\text{C}_{8''}$	35	$\text{H}_2\text{C}_{1'}$
9	$\text{H}_3''\text{C}_{4''}$	18	$\text{H}_5''\text{C}_{4''}$	27	$\text{H}_{2\text{RAC}}\text{C}_{1\text{RAC}}$	36	$\text{H}_3\text{C}_{4'}$

Figure S1. $^1\text{H}/^{13}\text{C}$ HMBC spectrum of (S)-2-fluoropropionyl-CoA in D_2O .

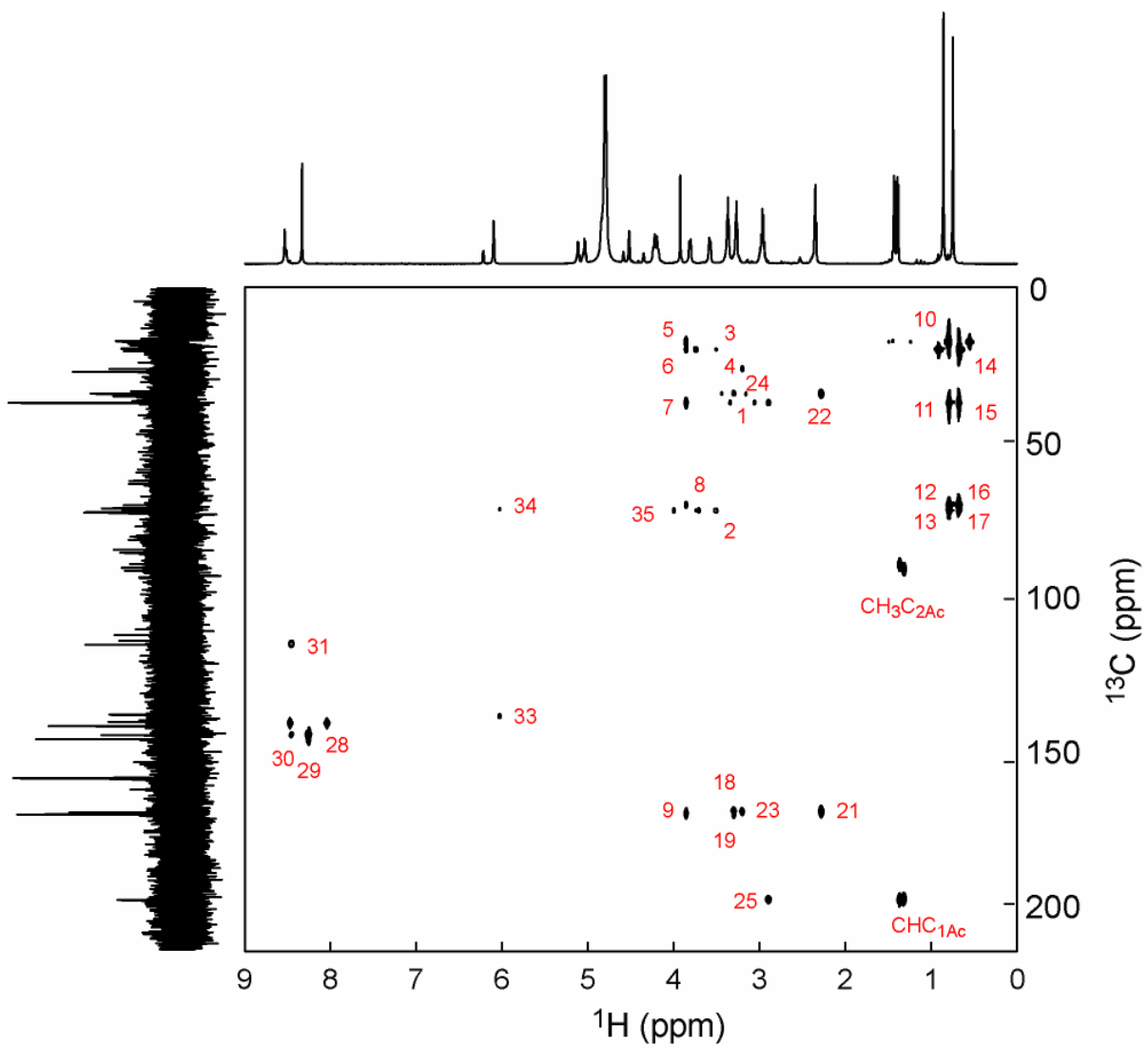


Figure S2. $^1\text{H}/^{13}\text{C}$ HMBC spectrum of (*R*)-2-fluoropropionyl-CoA in D_2O .

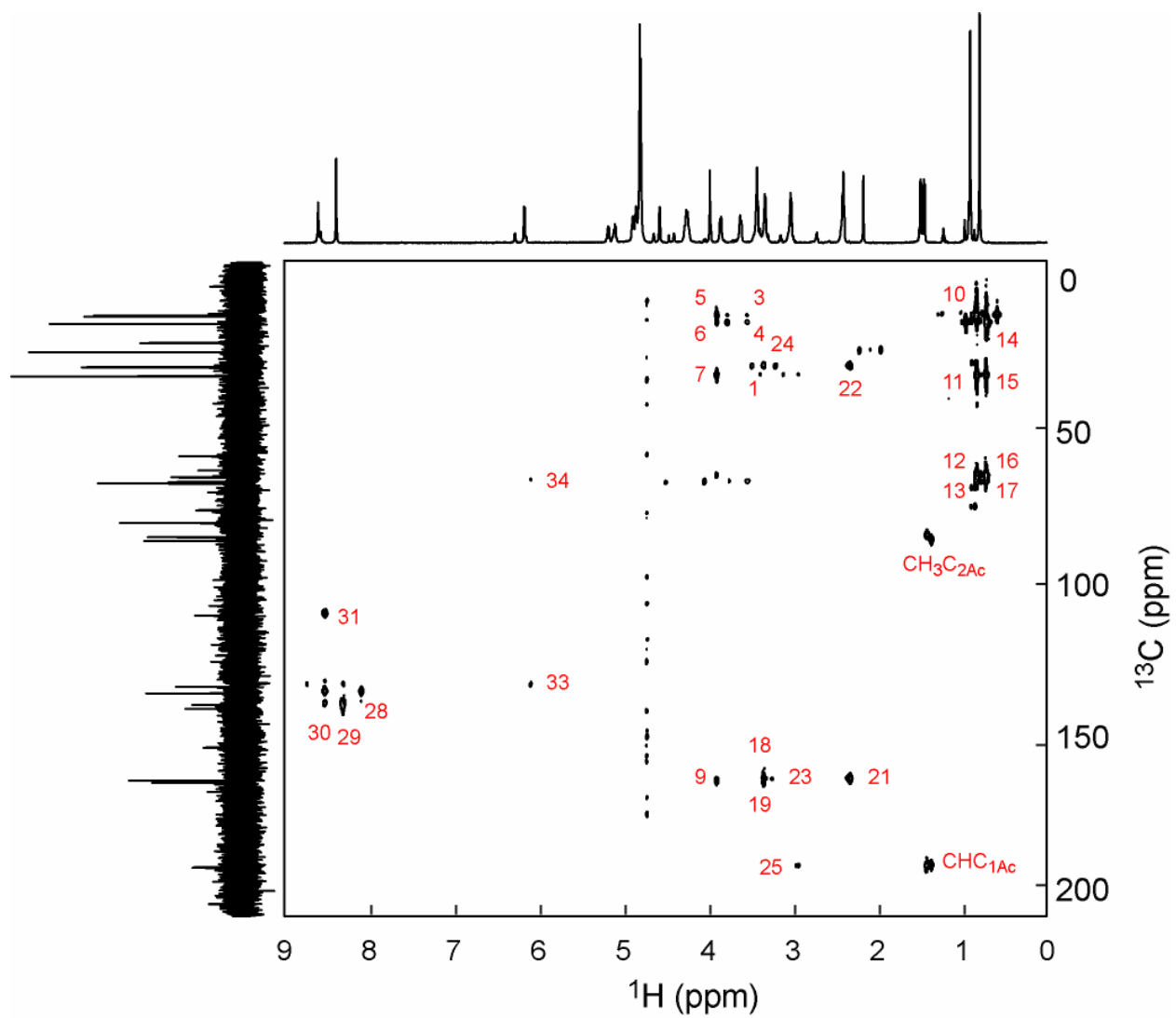


Figure S3. The catalytic triad of FIK-T42S. In the FIK-T42S mutant (PDB ID: 3KVU), both Ser 42 and His 76 populate two different rotatmers, and Glu 50 is rotated relative to its position in the wild-type enzyme (*I*). Chain A, grey; chain B, slate; carbon, grey; nitrogen, blue; oxygen red.

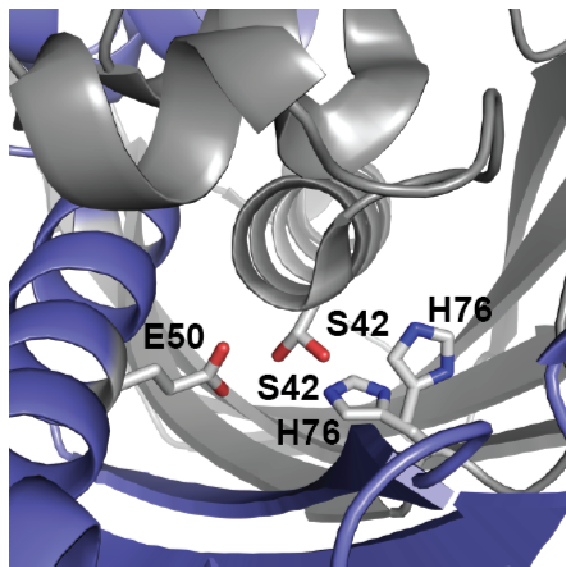


Figure S4. Simulated steady-state kinetic data for FIK-T42S-catalyzed acetyl-CoA hydrolysis. Data were simulated using the same K_D for both wild-type and mutant and using the acylation and deacylation rate constants measured using pre-steady-state kinetic analysis.

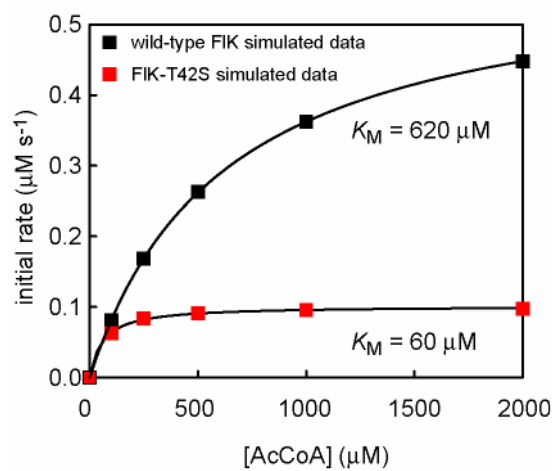


Figure S5. Time courses for hydrolysis of (*R*,*S*)-2-fluoropropionyl-CoA. (A) 50 μ M. (B) 100 μ M. (C) 200 μ M. (D) 250 μ M.

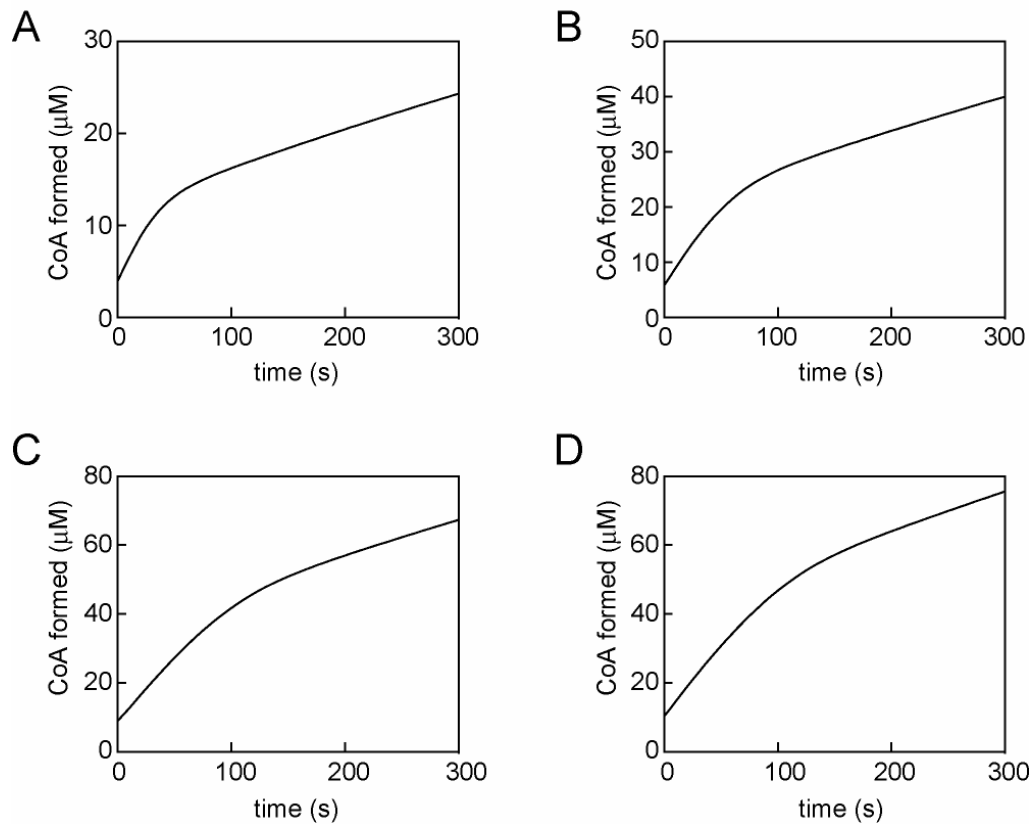


Figure S6. Taft plot for k_{cat} of various acyl-CoA FIK substrates. 2-fluoropropionyl-CoA substrates are shown in red. Data shown in black are from (2). σ^* values for acyl-CoAs with a single α -substituent are from (3). The σ^* value for 2-fluoropropionyl-CoA was calculated by adding the values for the F and Me substituents as described in (4). Et, butyryl-CoA; Me, propionyl-CoA; H, acetyl-CoA; Br, bromoacetyl-CoA; Cl, chloroacetyl-CoA; F, fluoroacetyl-CoA; (S)-F, Me, (S)-2-fluoropropionyl-CoA; (R)-F, Me, (R)-2-fluoropropionyl-CoA; CN, cyanoacetyl-CoA.

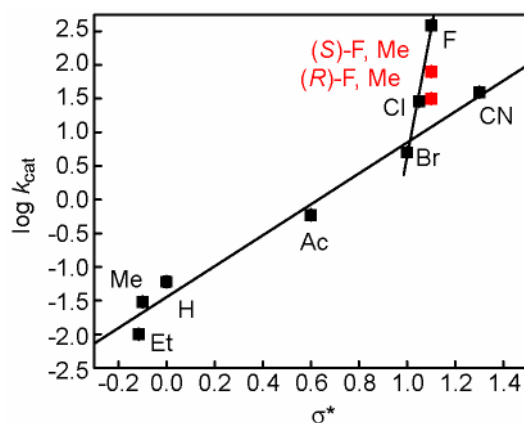
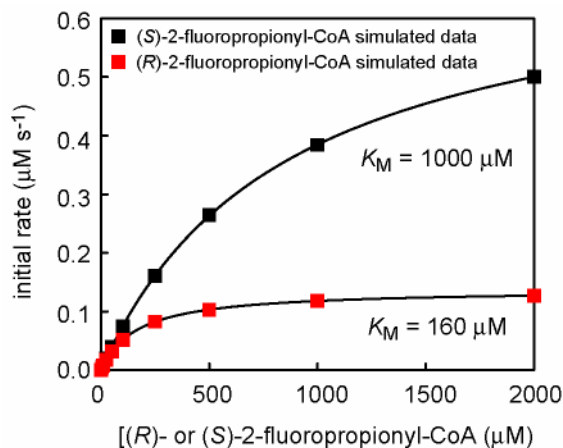


Figure S7. Simulated steady-state kinetic data for FIK-catalyzed hydrolysis of (S)- and (R)-2-fluoropropionyl-CoA. At constant K_D , the changes in the measured kinetic constants are sufficient to explain the difference in K_M between the two substrates.



Literature cited

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