

Supporting Information for

SMARTCyp – a 2D-method for Prediction of Cytochrome P450 Mediated Drug Metabolism

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Table S1. SMARTS rules and corresponding energies.

Table S2. Structures and information on all QM calculations for all SMARTS rules.

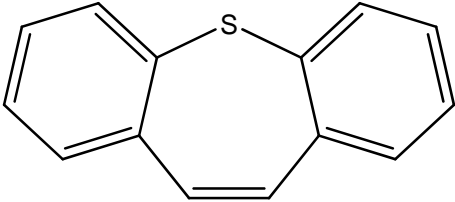
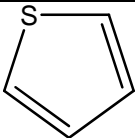
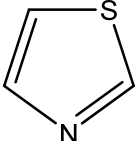
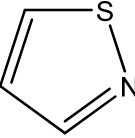
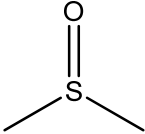
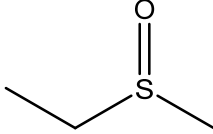
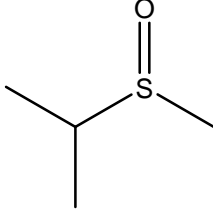
Table S1. SMARTS rules and corresponding energies (kJ/mol).

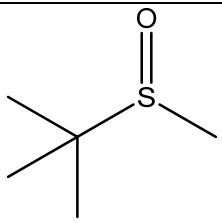
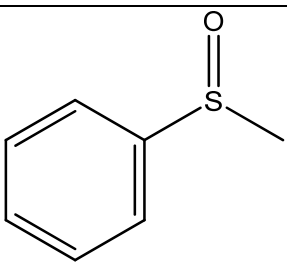
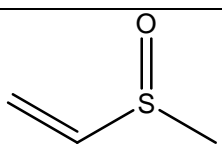
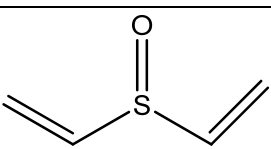
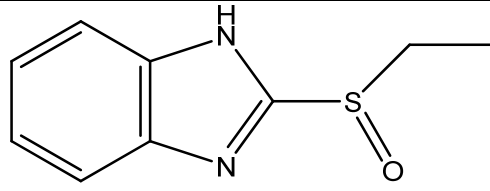
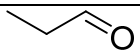
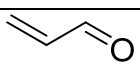
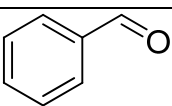
Sulphur atoms	
[SX2H1]	41.5
[\$([SX2H0]);!\$([S][*^2]);!\$([S][CX4H0])]	26.3
[\$([SX2H0][*^2]);!\$([S](~[^2])[^2]);!\$([S][CX4H0])]	34.4
[\$([S][*D4H0]);\$([SX2H0])]	44.4
[\$([SX2H0]([*^2])[*^2]);!\$([S][CX4H0])]	46.9
[sX2r5]	70.0
[\$([#16X3]=[OX1]);\$([#16]);!\$([#16X3]=[OX1][#6^2](~[#7^2]))]	30.4
[\$([#16X3]=[OX1]);\$([#16]);\$([#16X3]=[OX1][#6^2](~[#7^2]))]	46.9
Aldehyde carbon atoms	
[\$([CX3H1](=O)[#6])]	40.2
Phosphor with double bonded sulphur	
[\$([PX4]);\$([P]=[S])]	13.3
sp3 hybridized carbon atoms	
[\$([CX4]([#6^2])([#6^2])[#6^2]);!\$([CH0])]	33.1
[\$([CX4][N]);!\$([CH0]);!\$([C][N]([*^2])[*^2]);!\$([C][N]=[#6X3]);!\$([CX4][NX3][C]=[O])]	39.8
[!\$([CH0]);\$([CX4]([#6^2])[#6^2]),\$([CX4][#7]=[#6X3]),\$([CX4]([#8])[#8]);!\$([CX4]([#8])[#8][C]=[O])]	48.6
[\$([CX4][S]);!\$([CH0]);!\$([C][S]=[O])]	57.7
[\$([CX4][#6^2]~[#8]),\$([CX4][#6^2](~[#7]~[#7]),\$([CX4][#6^1]),\$([CX4][C^2]=[C^2]-[#6^2]);!\$([CH0]);!\$([CX4][C]=[O][NX3]);!\$([CX4][#6^2]=[#8]-[#8]);!\$([CX4][C^2]([C^2])=[C^2]-[#6^2])]	60.0
[\$([CX4][O]);!\$([CH0]);!\$([C][O][C]=[O]);!\$([CX4]1[O][C]1)]	62.2
[\$([CX4][#6^2]);!\$([CH0]);!\$([CX4][C]=[O][NX3])]	66.7
[\$([CX4][#7](~[*^2])~[*^2]);!\$([CH0])]	69.1
[\$([CX4][S](=[O])=[O]);!\$([CH0])]	69.5
[CX4;CH1,CH2;!\$([CX4][NX3H1][C]=[O])]	77.7
[CX4H3;!\$([CX4][NX3H1][C]=[O])]	89.6
[\$([CX4][NX3H1][C]=[O]);!\$([CH0])]	94.6
sp2 hybridized carbon atoms	
[\$([CX3H2]);\$([C]=[*^2]-[*^2])]	40.1
[\$([CX3H1]);\$([C]=[*^2]-[*^2]);!\$([C](-[*^2])=[*^2]-[*^2])]	52.4
[\$([cH1]);\$([c](:[#7]):[#7])]	55.8
[\$([ch1]);!\$([ch1]1:[c](-[N^3]-[*^2]):[c]:[c]:[c]:[c]1);!\$([c]1:[c]:[c]:[c](-[N^3]-[*^2]):[c]:[c]1);\$([ch1]1:[c](-[N^3]):[c]:[c]:[c]:[c]1),\$([c]1:[c]:[c]:[c](-[N^3]):[c]:[c]1)]	59.5
[\$([CX3]);\$([CX3]=[CX3]);!\$([CH0]);!\$([CX3](-[*^2])=[CX3]);!\$([CX3]=[CX3]-[*^2])]	65.6
[\$([cH1]);\$([c]:[#16])]	67.1
[\$([ch1]);\$([c]1:[c]:[c]:[c](~[#7X2H0]~[c^2,C^2]):[c]:[c]1),\$([c]1:[c]:[c]:[c](~[#7X3H1]~[c^2,C^2]):[c]:[c]1);!\$([c]1:[c]:[c]:[c](-[NH]-[C]=[O]):[c]:[c]1)]	69.2
[\$([ch1]);!\$([c]1:[c]:[c]:[c](-[O]-[C]=[O]):[c]:[c]1),\$([c]1:[c]:[c]:[c](-[NH]-[C]=[O]):[c]:[c]1),	75.3

$\$(\text{[c]1}:\text{[c]}:\text{[c]}:\text{[c]}(-[\text{O},\text{SX2}]):\text{[c]}:\text{[c]1})$	
$[\$(\text{[ch1]}); \$(\text{[c]1}:\text{[c]}:\text{[#16]}:\text{[c]}:\text{[c]1})]$	76.7
$[\$(\text{[ch1]}); \$(\text{[ch1]1}:\text{[c]}(\sim\text{[#7X2H0]}\sim\text{[c}^2,\text{C}^2]):\text{[c]}:\text{[c]}:\text{[c]}:\text{[c]1}), \$(\text{[ch1]1}:\text{[c]}(\sim\text{[#7X3H1]}\sim\text{[c}^2,\text{C}^2]):\text{[c]}:\text{[c]}:\text{[c]}:\text{[c]1})$ $, \$(\text{[ch1]}:\text{[c]}-[\text{O},\text{SX2}]); !\$(\text{[ch1]}:\text{[c]}-[\text{O}]-[\text{C}]=[\text{O}])]$	77.3
$[\$(\text{[CX3H1},\text{cX3H1}]); !\$(\text{[ch1]1}:\text{[c]}:\text{[c]}:\text{[#7]}:\text{[c]}:\text{[c]1}); !\$(\text{[ch1]1}:\text{[c]}:\text{[#7]}:\text{[c]}:\text{[c]}:\text{[c]1}); !\$(\text{[ch1]1}:\text{[c]}:\text{[#7]}:\text{[c]}:\text{[c]1}); !\$(\text{[ch1]}:\text{[#7]})]$	82.3
$[\$(\text{[ch1]1}:\text{[c]}:\text{[c]}:\text{[#7]}:\text{[c]}:\text{[c]1}), \$(\text{[ch1]1}:\text{[c]}:\text{[#7]}:\text{[c]}:\text{[c]}:\text{[c]1}), \$(\text{[ch1]1}:\text{[c]}:\text{[#7]}:\text{[c]}:\text{[c]1}), \$(\text{[ch1]}:\text{[#7]})]$	89.6
Nitrogen atoms	
$[\$(\text{[N}^3\text{H0}]); !\$(\text{[N}^3][^*^2])]$	41.0
$[\$(\text{[N}^3]); \$(\text{[H1},\text{H2}])]$	54.1
$[\$(\text{[N]}(\text{[#6}^2]1)[\text{#6}^2]=[\text{#6}^2][\text{#6}^3][\text{#6}^2]=1)]$	61.9
$[\$(\text{[N}^3\text{H0}]); \$(\text{[N}^3][^*^2]); !\$(\text{[N}^3][(^*^2)][^*^2])]$	63.9
$[\$(\text{[nr6]}), \$(\text{[N}^2]=[\text{C}])]$	75.6
$[\$(\text{[N]}); \$(\text{[NX3H0]}(\text{[^*^2]}[^*^2]), \$(\text{[N}^2\text{H1]}[\text{C}]=[\text{O}])]$	89.6
$[\text{nr5H0}]$	92.1

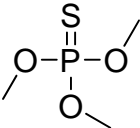
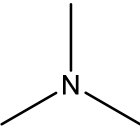
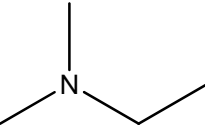
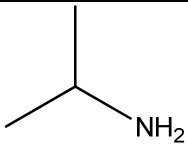
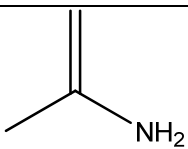
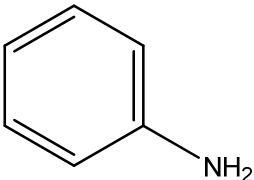
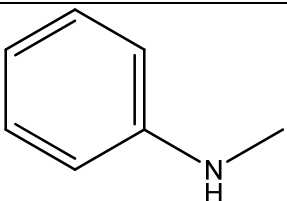
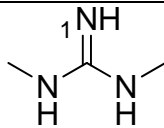
Table S2. Structures and information on all QM calculations for all SMARTS rules. Spin state in parenthesis after compound name (d is doublet, q is quartet). If there are multiple possible sites the computed site is marked with the number 1 unless otherwise written.

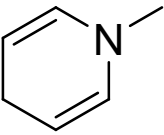
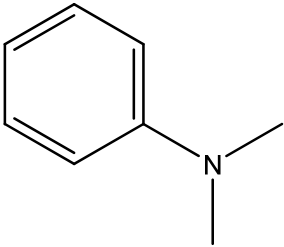
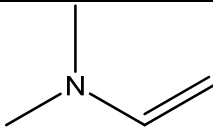
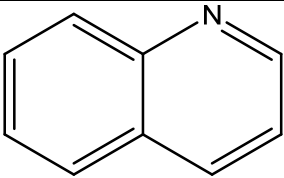
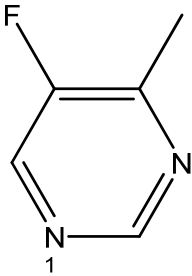
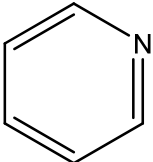
Sulphur Oxidations		
[SX2H1]		Energy (kJ/mol)
Methylsulfane (d)		41.5
[\$([SX2H0]);!\$([S][*^2]);!\$([S][CX4H0])]		Energy (kJ/mol)
Methylethylsulfane (d)		28.7 ^a
Dimethylsulfane (d)		23.8 ^b
Average:		26.3
Standard deviation:		3.5
[\$([SX2H0][*^2]);!\$([S](~[^2])[^2]);!\$([S][CX4H0])]		Energy (kJ/mol)
Phenylmethylsulfane (d)		33.8 ^a
Vinylmethylsulfane (d)		35.0 ^a
Average:		34.4
Standard deviation:		0.9
[\$([S][*D4H0]);\$([SX2H0])]		Energy (kJ/mol)
Tertbutylmethylsulfane (d)		44.4 ^a
[\$([SX2H0]([*^2])[*^2]);!\$([S][CX4H0])]		Energy (kJ/mol)
Divinylsulfane (d)		45.1 ^a

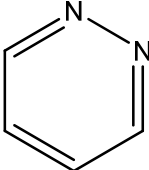
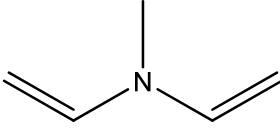
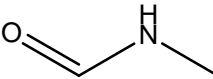
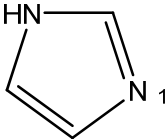
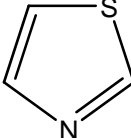
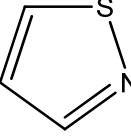
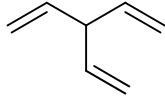
Dibenzo[b,f]thiepine (d)		48.8 ^a
Average:		46.9
Standard deviation:		2.6
[sX2r5]		Energy (kJ/mol)
Thiophene (d)		69.1
Thiazole (d)		74.0
Isothiazole (d)		66.9
Average:		70.0
Standard deviation:		3.6
[(#16X3)(=[OX1]);!(#16);!(#16X3)(=[OX1])[#6^2](~[#7^2])]		Energy (kJ/mol)
Dimethylsulfoxide (d)		27.3 ^b
Methylsulfinylethane (d)		25.3 ^a
Methylsulfinylisopropane (d)		33.6 ^a

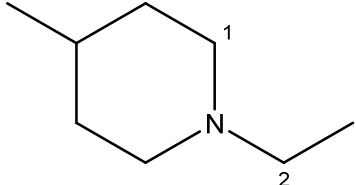
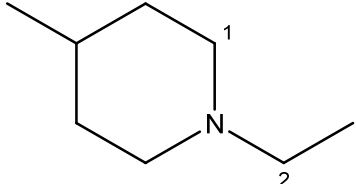
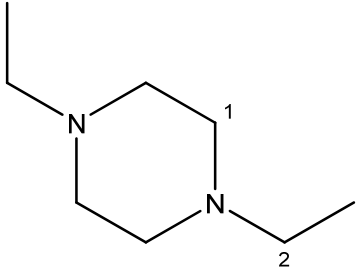
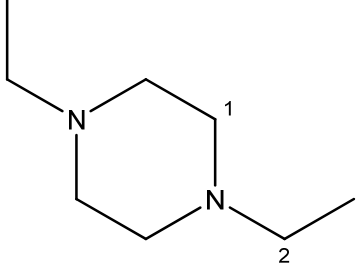

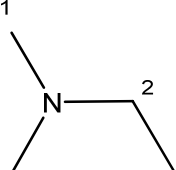
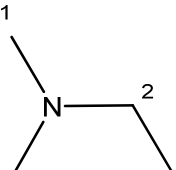
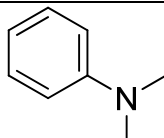
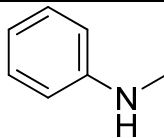
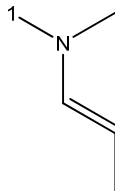
Methylsulfinyltertbutane (d)		34.8 ^a
Methylsulfinylbenzene (d)		31.3 ^a
Methylsulfinylethene (d)		27.4 ^a
Divinylsulfoxide (d)		32.8 ^a
Average:		30.4
Standard deviation:		3.7
$[\$(\text{[#16X3]}(=\text{[OX1]}));\$(\text{[#16]});\$(\text{[#16X3]}(=\text{[OX1]})[\#6^2](\sim[\#7^2]))]$		Energy (kJ/mol)
2-(ethylsulfinyl)-1H-benzo[d]imidazole (d)		46.9 ^a
Aldehyde oxidation (of C=O carbon)		
$[\$(\text{[CX3H1]}(=\text{O})[\#6])]$		Energy (kJ/mol)
Propionaldehyde (q)		37.0
Acrylaldehyde (q)		42.5
Benzaldehyde (q)		41.1
Average:		40.2
Standard deviation:		2.8

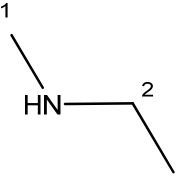
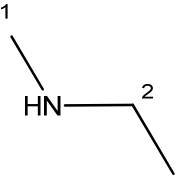
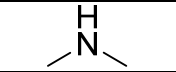
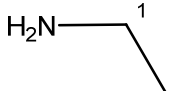
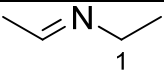
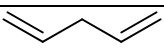
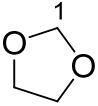

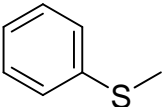
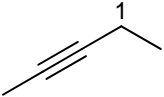
Phosphor with double bonded sulphur (desulphurization)

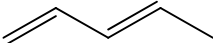
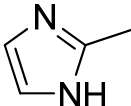
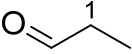
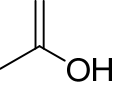


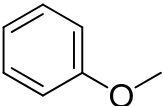


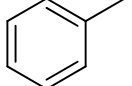
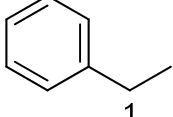
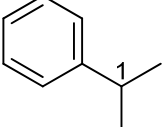
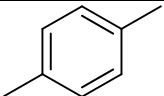
Energy (kJ/mol)		Energy (kJ/mol)
Phosphor with double bonded sulphur (desulphurization)		
[[PX4]];[[P]=[S]]		
O,O,O-trimethyl phosphorothioate (d)		13.3 ^c
Nitrogen oxidations		
[[N^3H0]];![[N^3][*^2]]		Energy (kJ/mol)
Trimethylamine (d)		39.8 ^b
N-ethyl-dimethylamine (d)		42.3
Average:		41.0
Standard deviation:		1.7
[[N^3]];[[H1,H2]]		Energy (kJ/mol)
Propane-2-amine (d)		51.6 ^a
Propene-2-amine (d)		55.9 ^a
Aniline (d)		56.3
N-methylaniline (d)		50.0
1,3-dimethylguanidine (d)		56.9 ^a

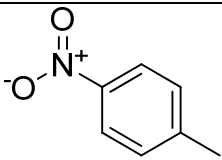
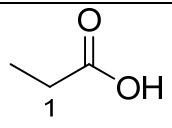
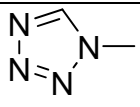
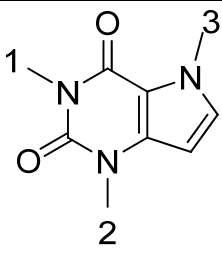
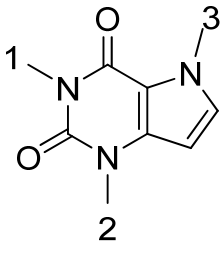
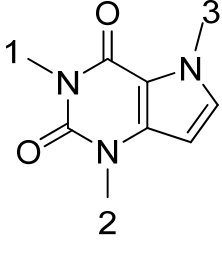
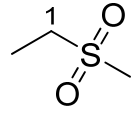
Average:		54.1
Standard deviation:		3.1
[$([N]([6^2]1)[6^2]=[6^2][6^3][6^2]=1)$]		Energy (kJ/mol)
1-methyl-4-hydropyridine (d)		61.9
[$([N^3H0]);([N^3][^2]);!([N^3]([^2])[^2])$]		Energy (kJ/mol)
<i>N,N</i> -dimethylaniline (d)		62.5
<i>N,N</i> -dimethylethenamine (d)		65.3
Average:		63.9
Standard deviation:		2.0
[$([nr6]),([N^2]=[C])$]		Energy (kJ/mol)
Quinoline (d)		72.7
5-fluoro-4-methylpyrimidine (d)		83.5
Pyridine (d)		71.0

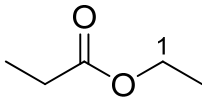
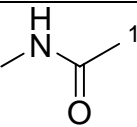
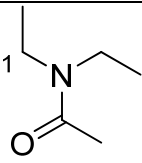


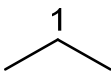
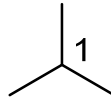
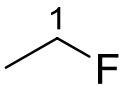
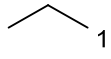
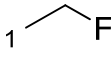
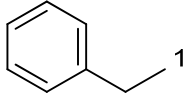
Pyridazine (d)		75.0
Average:		75.6
Standard deviation:		5.5
[$[\text{N}]$]; $[\text{NX3H0}][^*^2][^*^2]$, $[\text{N}^2\text{H1}][\text{C}=\text{O}]$		Energy (kJ/mol)
<i>N,N</i> -divinylmethaneamine (d)		93.1
<i>N</i> -methylformamide (d)		86.2
Average:		89.6
Standard deviation:		4.9
[nr5H0]		Energy (kJ/mol)
Imidazole (d)		85.8
Thiazole (d)		88.4
Isothiazole (d)		102.1
Average:		92.1
Standard deviation:		9.4
sp³ carbons		
[$[\text{CX4}][\#6^2][\#6^2][\#6^2]$];! $[\text{CH0}]$		Energy (kJ/mol)
3-vinylpenta-1,4-diene (q)		33.1
[$[\text{CX4}][\text{N}]$];! $[\text{CH0}]$;! $[\text{C}][\text{N}][^*^2][^*^2]$; ! $[\text{C}][\text{N}]=[\#6\text{X3}]$;! $[\text{CX4}][\text{NX3}][\text{C}=\text{O}]$		Energy (kJ/mol)

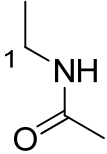
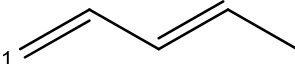
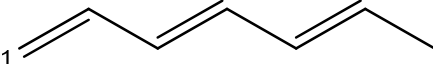
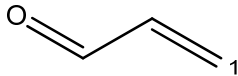
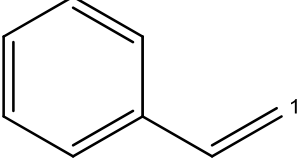
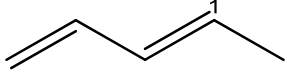
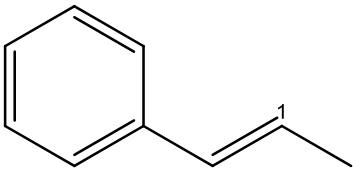
1-ethyl-4-methylpiperidine (q) (1)		35.9
1-ethyl-4-methylpiperidine (q) (2)		37.5
1,4-diethylpiperazine (q) (1)		44.5
1,4-diethylpiperazine (q) (2)		39.6
Trimethylamine (q)		38.4 ^b
N-ethyl –dimethylamine (q) (1)		38.8
N-ethyl –dimethylamine (q) (2)		37.1
<i>N,N</i> -dimethyl aniline (q)		40.7 ^d
<i>N</i> -methyl aniline (q)		44.0 ^d
Dimethylpropenamine (q)		32.9

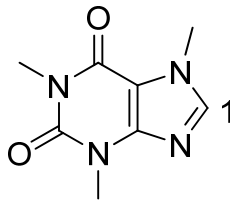
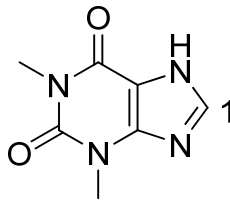
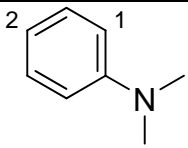
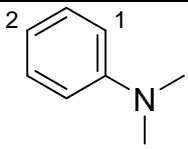

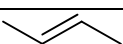
N-ethyl-methylamine (q) (1)		41.9
N-ethyl-methylamine (q) (2)		39.5
Dimethylamine (q)		42.6 ^d
Ethanamine (q) (1)		43.5
Average:		39.8
Standard deviation:		3.3
[!\$([CH0]);\$(CX4)(#6^2)(#6^2),\$(CX4)[#7]=[#6X3), \$(CX4)(#8)[#8];!\$(CX4)(#8)[#8][C]=[O]]		Energy (kJ/mol)
N-ethylideneethaneamine (q)		48.6
Penta-1,4-diene (q)		48.4
1,3-dioxolane (q)		48.7
Average:		48.6
Standard deviation:		0.1
[\$(CX4)[S];!\$(CH0);!\$(C)[S]=[O]]		Energy (kJ/mol)
Dimethylsulfane (q)		58.1 ^b
Methyl(phenyl)sulfane (q)		57.3 ^d
Average:		57.7
Standard deviation:		0.5
[\$(CX4)[#6^2]~[#8),\$(CX4)[#6^2](~[#7])~[#7]),\$(CX4)[#6^1), \$(CX4)[C^2]=[C^2]-[#6^2]);!\$(CH0);!\$(CX4)[C](=[O])[NX3]); !\$(CX4)[#6^2](=[#8])-[#8];!\$(CX4)[C^2]([C^2])=[C^2]-[#6^2]]		Energy (kJ/mol)
Pent-2-yne (q)		62.3

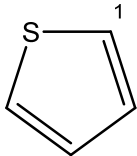
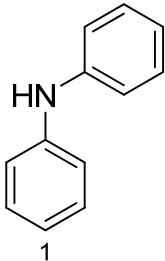
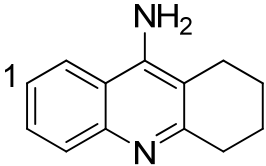
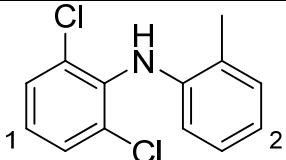
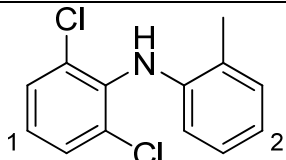
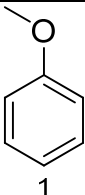
Penta-1,3-diene (q)		57.7
2-methyl-1 <i>H</i> -imidazole (q)		59.1
Propionaldehyde (q)		61.3
Prop-1-en-2-ol (q)		59.4 ^d
Average:		60.0
Standard deviation:		1.8
[\$([CX4][O]);!(CH0);!(C)[O][C]=[O];!(CX4)1[O][C]1]		Energy (kJ/mol)
Isopropanol (q)		55.7
Dimethylether (q)		63.1 ^d
Anisole (q)		68.0 ^d
Average:		62.2
Standard deviation:		6.2
[\$([CX4][#6^2]);!(CH0);!(CX4)[C](=[O])[NX3]		Energy (kJ/mol)
Propene (q)		67.0 ^d
2-fluoro-prop-1-ene (q)		69.1 ^d
Toluene (q)		67.3 ^d
Ethylbenzene (q)		64.6 ^d
1-methylethylbenzene (q)		69.7 ^d
Para-xylene (q)		65.9 ^d

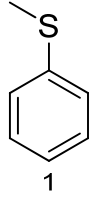
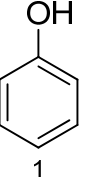
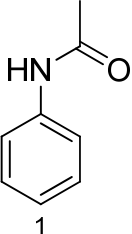
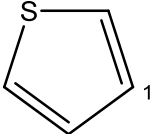
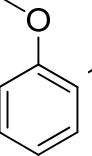
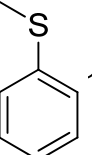
1-methyl-4-nitro-benzene (q)		62.6 ^d
Propionic acid (q)		67.2
Average:		66.7
Standard deviation:		2.3
[\$([CX4][#7](~[*^2])~[*^2]);!\$([CH0])]		Energy (kJ/mol)
1-methyl-1 <i>H</i> -tetrazole (q)		65.7
Caffeine (q) (1)		75.8 ^e
Caffeine (q) (2)		70.1 ^e
Caffeine (q) (3)		64.7 ^e
Average:		69.1
Standard deviation:		5.0
[\$([CX4][S](=[O])=[O]);!\$([CH0])]		Energy (kJ/mol)
Methylethylsulfone (q)		69.5
[CX4;CH1,CH2;!\$([CX4][NX3H1][C]=[O])]		Energy (kJ/mol)

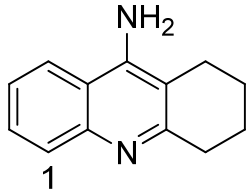
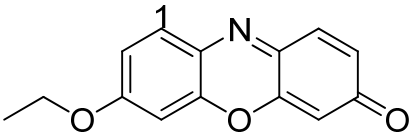
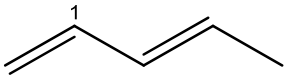
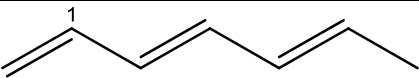
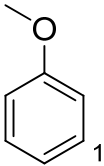
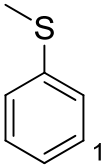
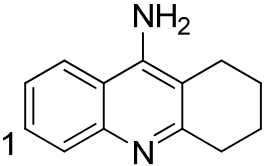

Ethyl propionate (q)		81.9
N-methylacetamide (q)		81.7
N,N-diethylacetamide (q)		79.6
2,3-diethyloxirane (q)		73.0
Dimethylsulphoxide (q)		77.0 ^b
Propane (q)		77.0 ^d
Isobutane (q)		74.7 ^d
Fluoroethane (q)		76.5 ^d
Average:		77.7
Standard deviation:		2.3
[CX4H3;!\$(CX4)[NX3H1][C]=[O]]		Energy (kJ/mol)
Propane (q)		88.4 ^d
Fluoroethane (q)		93.0 ^d
Ethylbenzene (q)		87.4 ^d
Average:		89.6
Standard deviation:		3.0
!\$(CX4)[NX3H1][C]=[O];!\$(CH0)]		Energy (kJ/mol)

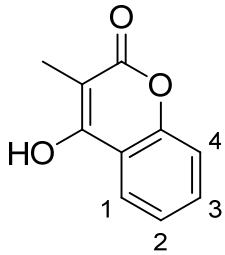
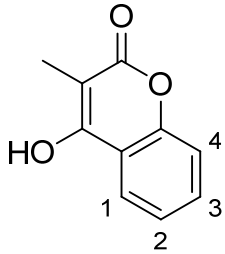
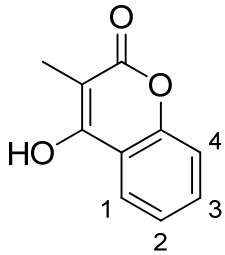
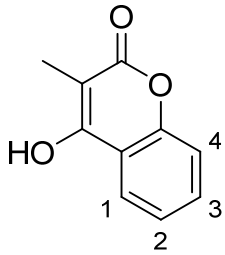
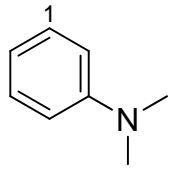
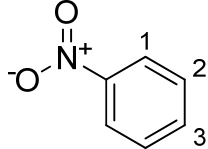
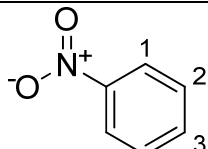
<i>N</i> -ethylacetamide (q)		94.6
sp² carbons		
[$\\$(\text{CX3H2})$;$\\$(\text{C}=[*^2]-[*^2])$]		Energy (kJ/mol)
Penta-1,3-diene (q)		44.0
Hepta-1,3,5-triene (q)		29.3
Acrylaldehyde (q)		42.2
Styrene (q)		44.9
Average:		40.1
Standard deviation:		7.3
[$\\$(\text{CX3H1})$;$\\$(\text{C}=[*^2]-[*^2])$;$\!(\text{C}(-[*^2])=[*^2]-[*^2])$]		Energy (kJ/mol)
Penta-1,3-diene (q)		52.5
Methylstyrene (q)		52.3
Average:		52.4
Standard deviation:		0.1
[$\\$(\text{cH1})$;$\\$(\text{c}(:[#7]):[#7])$]		Energy (kJ/mol)

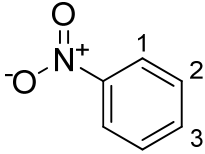
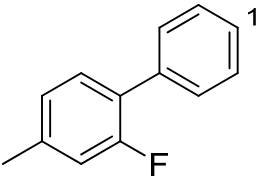
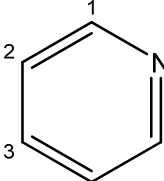
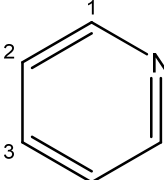
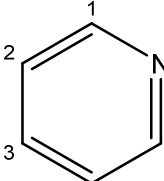
Caffeine (d)		54.4 ^e
Theophylline (d)		57.1 ^e
Average:		55.8
Standard deviation:		1.9
$[\$(\text{[ch1]});!\$(\text{[ch1]1}:[\text{c}](-[\text{N}^3]-[\text{*}^2]):[\text{c}]:[\text{c}]:[\text{c}]:[\text{c}]1);$ $!\$(\text{[c]1}:[\text{c}]:[\text{c}]:[\text{c}](-[\text{N}^3]-[\text{*}^2]):[\text{c}]:[\text{c}]1);$ $\$(\text{[ch1]1}:[\text{c}](-[\text{N}^3]):[\text{c}]:[\text{c}]:[\text{c}]:[\text{c}]1),\$(\text{[c]1}:[\text{c}]:[\text{c}]:[\text{c}](-[\text{N}^3]):[\text{c}]:[\text{c}]1)]$		Energy (kJ/mol)
<i>N,N</i> -dimethylaniline (d) (1)		58.8 ^f
<i>N,N</i> -dimethylaniline (d) (2)		60.2 ^f
Average:		59.5
Standard deviation:		0.9
$[\$(\text{[CX3]});!\$(\text{[CX3]}=[\text{CX3}]);!\$(\text{[CH0]});!\$(\text{[CX3]}(-[\text{*}^2])=[\text{CX3}]);$ $!\$(\text{[CX3]}=[\text{CX3}]-[\text{*}^2])]$		Energy (kJ/mol)
Ethene (q)		67.5 ^t
2-butene (d)		63.6 ^f
Average:		65.6
Standard deviation:		2.7
$[\$(\text{[cH1]});!\$(\text{[c]}:[\text{\#16}])]$		Energy (kJ/mol)

Thiophene (d)		67.1
$[\$(\text{ch1});\$(\text{c1}:\text{c}:\text{c}:\text{c}:\text{c}(\sim\text{X2H0})\sim\text{c}^2,\text{C}^2):\text{c}:\text{c}1),$ $\$(\text{c1}:\text{c}:\text{c}:\text{c}:\text{c}(\sim\text{X3H1})\sim\text{c}^2,\text{C}^2):\text{c}:\text{c}1);$ $!\$(\text{c1}:\text{c}:\text{c}:\text{c}:\text{c}(-\text{NH})-\text{C}=\text{O}):\text{c}:\text{c}1]$		Energy (kJ/mol)
Diphenylamine (d)		64.9
Tacrine (d)		68.6 ^e
2,6-dichloro-N-o-tolylaniline (d) (1)		69.3 ^f
2,6-dichloro-N-o-tolylaniline (d) (2)		73.9 ^f
Average:		69.2
Standard deviation:		3.7
$[\$(\text{ch1});!\$(\text{c1}:\text{c}:\text{c}:\text{c}:\text{c}(-\text{O})-\text{C}=\text{O}):\text{c}:\text{c}1);$ $\$(\text{c1}:\text{c}:\text{c}:\text{c}:\text{c}(-\text{NH})-\text{C}=\text{O}):\text{c}:\text{c}1),$ $\$(\text{c1}:\text{c}:\text{c}:\text{c}:\text{c}(-\text{O},\text{SX2}):\text{c}:\text{c}1]$		Energy (kJ/mol)
Anisole (d)		74.4

Methyl(phenyl)sulfane (d)		74.3
Phenol (d)		75.9
N-phenylacetamide (d)		76.7
Average:		75.3
Standard deviation:		1.2
$[\$(\text{[ch1]}); \$(\text{[c]1}:\text{[c]}:\text{[#16]}:\text{[c]}:\text{[c]1})]$		Energy (kJ/mol)
Thiophene (d)		76.7
$[\$(\text{[ch1]}); \$(\text{[ch1]1}:\text{[c]}(\sim\text{[#7X2H0]}\sim\text{[c]^2,C^2}):\text{[c]}:\text{[c]}:\text{[c]}:\text{[c]1}),$ $\$(\text{[ch1]1}:\text{[c]}(\sim\text{[#7X3H1]}\sim\text{[c]^2,C^2}):\text{[c]}:\text{[c]}:\text{[c]}:\text{[c]1}),$ $\$(\text{[ch1]}:\text{[c]}-\text{[O,SX2]}); !\$(\text{[ch1]}:\text{[c]}-\text{[O]}-\text{[C]}=\text{[O]})]$		Energy (kJ/mol)
Anisole (d)		78.2
Methyl(phenyl)sulfane (d)		77.2

Tacrine (d)		76.5 ^e
7-ethoxyresorufin (d)		77.3 ^e
Average:		77.3
Standard deviation:		0.7
$[\$(\text{CX3H1},\text{cX3H1})];!\$(\text{ch1})1:[\text{c}]:[\text{c}]:[\#7]:[\text{c}]:[\text{c}1];$ $!\$(\text{ch1})1:[\text{c}]:[\#7]:[\text{c}]:[\text{c}]:[\text{c}1];!\$(\text{ch1})1:[\text{c}]:[\#7]:[\text{c}]:[\text{c}1];$ $!\$(\text{ch1}:[\#7])]$		Energy (kJ/mol)
Penta-1,3-diene (d)		81.8
Hepta-1,3,5-triene (q)		77.0
Anisole (d)		85.8
Methyl(phenyl)sulfane (d)		87.1
Tacrine (d)		82.0 ^e
Benzene (d)		87.3 ^f

Warfarin (d) (1)		83.4 ^f
Warfarin (d) (2)		80.8 ^f
Warfarin (d) (3)		81.0 ^f
Warfarin (d) (4)		78.7 ^f
<i>N,N</i> -dimethylaniline (d)		86.7 ^f
Nitrobenzene (q) (1)		78.5 ^f
Nitrobenzene (d) (2)		84.5 ^f

Nitrobenzene (q) (3)		79.2 ^f
2-fluoro-4-methylbiphenyl (d)		80.7 ^f
Average:		82.3
Standard deviation:		3.4
$[\$(\text{[ch1]1:[c]:[c]:[#7]:[c]:[c]1}),\$(\text{[ch1]1:[c]:[#7]:[c]:[c]:[c]1}),\$(\text{[ch1]1:[c]:[#7]:[c]:[c]1}),\$(\text{[ch1]:[#7]})]$		Energy (kJ/mol)
Pyridine (q) (1)		90.6
Pyridine (d) (2)		85.5
Pyridine (q) (3)		92.6
Average:		89.6
Standard deviation:		3.6

^a Transition state geometry from scan crossing point.

^b Taken from the publication: P. Rydberg, U. Ryde, L. Olsen, *J. Chem. Theory Comput.* 2008, 4, 1369-1377

^c Energy taken from the calculation with the small basis set relative to reactant complex.

^d Taken from the publication: L. Olsen, P. Rydberg, TH. Rod, U. Ryde, *J. Med. Chem.* 2006, 49, 6489-6499.

^e Taken from the publication: P. Rydberg, P. Vasanthanathan, C. Oostenbrink, L. Olsen, *ChemMedChem* 2009, 4, 2070-2079.

^f Taken from the publication: P. Rydberg, U. Ryde, L. Olsen, *J. Phys. Chem. A* 2008, 112, 13058-13065.