

SUPPORTING INFORMATION

Antimicrobial Polymers Prepared by ROMP with Unprecedented Selectivity:

A Molecular Construction Kit Approach

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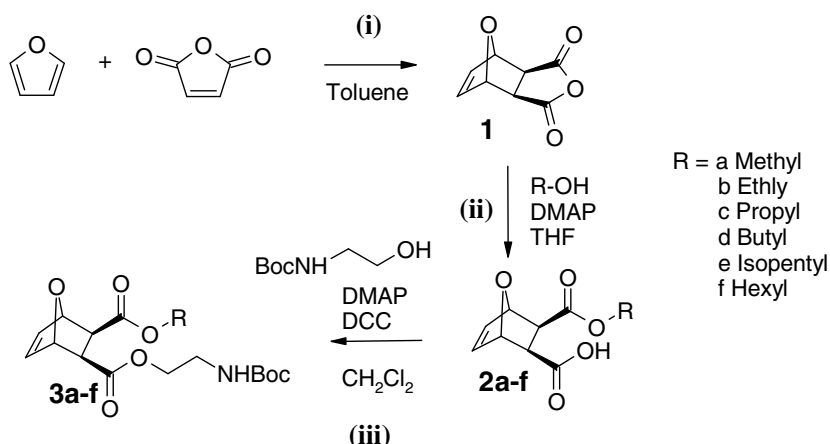
Experimental:

General: Maleic anhydride, furane, 4-dimethyl aminopyridine (DMAP), 1,3-dicyclohexylcarbodiimide (DCC), methanol, ethanol, 1-propanol, 1-butanol, isopentanol, 1-hexanol, ethylvinyl ether and trifluoroacetic acid (TFA) were obtained as reagent grade from Aldrich, Fluka or Acros and used as received.

3rd generation Grubbs catalyst (Dichloro-di(3-bromopyridino)-N,N'-Dimesitylenoimidazolino-Ru=CHPh; G3) was synthesized as described previously by Grubbs et al.¹. The HPLC grade solvents N,N-dimethylformamide (DMF), toluene, ethyl acetate and hexane were purchased from Aldrich, Fisher Scientific or Acros and used as received. THF (HPLC grade, Fisher Scientific) was distilled from sodium/benzophenone under nitrogen. Dichloromethane (HPLC grade, Fisher Scientific) was distilled from CaH₂ under nitrogen.

Gel permeation chromatography (DMF/0.01 M LiCl, calibrated with polystyrene standards, toluene as flow marker, 50°C) was measured on a PL50 GPC setup (Polymer Laboratories, Amherst, MA) with a PL Gel 5 µm pre-column and two 10 µm analytical Mixed-D columns (Polymer Laboratories, Amherst, MA). NMR spectra were recorded on a Bruker DPX300 spectrometer (Bruker, Madison, WI). High resolution mass spectra were obtained from a JEOL JMS 700 instrument (JEOL, Peabody, MA); Matrix Assisted Laser Desorption and Ionization Time of Flight Mass Spectra (MALDI-TOF MS) were measured on a Bruker Daltonics Reflex III (Bruker, Madison, WI). The biological activity of the polymer samples (the minimal concentration inhibiting 90% of bacterial growth (*MIC*₉₀) for *Escherischia coli* (D31) and *Staphylococcus aureus* (ATCC25923), and the concentration that lyses 50% of red blood cells (*HC*₅₀) were determined as reported previously.²

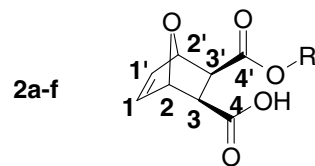
Monomer Synthesis:



(i) Using a literature procedure, maleic anhydride (100 g, 1.02 mol) was dissolved in 1 L toluene. 150 mL (140.7 g, 2.05 mmol) furan were added. The solution was stirred for 3 days. The product (**1**) was then filtered, washed with toluene and dried under vacuum. A colorless powder was obtained. Yields and spectroscopic data matched those reported previously.³

(ii) **1** (5 g, 30.0 mmol) was dissolved in THF. Two equivalents of the respective alcohol (e.g. 5.26 g (60 mmol) isopentanol for R = isopentyl) and 10 mol% DMAP were added. After stirring over night, the solvent was removed by vacuum evaporation at room temperature. The unreacted alcohol was removed by dynamic vacuum ($5 \cdot 10^{-2}$ mbar). Crystallization from dichloromethane/hexane yielded, e.g. for R = isopentyl, 4.7 g (18.5 mmol, 62 %) of the product (**2**).

2a: R = Methyl: colorless solid, ¹H-NMR (300 MHz, CDCl₃): 2.86 (m, 2H, H3 & H3'), 3.71 (s, 3H, CH₃), 5.27 & 5.32 (m, 2H, H2 & H2'), 6.47 (m, 2H, H1 & H1').



2b: R = Ethyl: colorless solid, ¹H-NMR (300 MHz, CDCl₃): 1.25 (t, $J = 7.2$ Hz, 3H, CH₂-CH₃), 2.83 (m, 2H, H3 & H3'), 4.16 (q, $J = 7.1$ Hz, 2H, CH₂-CH₃), 5.25 & 5.31 (s, 2H, H2 & H2'), 6.46 (m, 2H, H1 & H1'), 7.53-8.21 (br s, 1H, OH).

2c: R = Propyl: colorless solid, $^1\text{H-NMR}$ (300 MHz, CDCl_3): 0.92 (t, $J = 7.3$ Hz, 3H, $\text{CH}_2\text{-CH}_3$), 1.63 (m, 2H, $\beta\text{-CH}_2$), 2.84 (m, 2H, H3 & H3'), 4.05 (m, 2H, $\alpha\text{-CH}_2$), 5.23 & 5.29 (m, 2H, H2 & H2'), 6.46 (m, 2H, H1 & H1'), 9.12-9.68 (br s, 1H, OH).

2d: R = Butyl: colorless oil, $^1\text{H-NMR}$ (300 MHz, CDCl_3): 0.85 (t, $J = 7.1$ Hz, 3H, $\text{CH}_2\text{-CH}_3$), 1.30 (m, 2H, $\gamma\text{-CH}_2$), 1.54 (m, 2H, $\beta\text{-CH}_2$), 2.78 (m, 2H, H3 & H3'), 4.04 (m, 2H, $\alpha\text{-CH}_2$), 5.21 & 5.27 (m, 2H, H2 & H2'), 6.39 (m, 2H, H1 & H1').

2e: R = iso-Pentyl: colorless oil, $^1\text{H-NMR}$ (300 MHz, CDCl_3): 0.90 (d, $J = 6.6$ Hz, 6H, $\text{CH}_2\text{-CH}_3$), 1.53 (m, 1H, CH-CH_3), 1.68 (m, 2H, $\beta\text{-CH}_2$), 2.84 (m, 2H, H3 & H3'), 4.08 (m, 2H, $\alpha\text{-CH}_2$), 5.25 & 5.31 (m, 2H, H2 & H2'), 6.45 (m, 2H, H1 & H1'), 8.70-9.68 (br s, 1H, OH).

2f: R = Hexyl: colorless solid, $^1\text{H-NMR}$ (300 MHz, CDCl_3): 0.88 (t, $J = 6.4$ Hz, 3H, $\text{CH}_2\text{-CH}_3$), 1.28 (m, 6H, $\gamma\text{- to } \varepsilon\text{-CH}_2$), 1.61 (m, 2H, $\beta\text{-CH}_2$), 2.85 (m, 2H, H3 & H3'), 4.13 (m, 2H, $\alpha\text{-CH}_2$), 5.25 & 5.29 (m, 2H, H2 & H2'), 6.47 (m, 2H, H1 & H1').

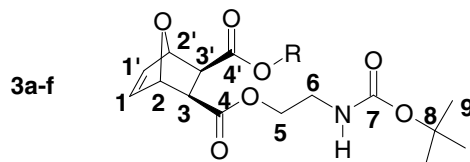
(iii) 1 eq **2a-f** (e.g. 2 g (7.88 mmol) for R = isopentyl) was dissolved in 25 mL dichloromethane. 1 eq (1.27 g, 7.88 mmol) boc-protected 2-amino ethanol and 10 mol% of DMAP were added. The solution was cooled to 0°C . 1 eq (206.3 g, 7.88 mmol) DCC was added, and the suspension was stirred over night. It was then filtered through a short alumina column (5 cm neutral Al_2O_3 /dichloromethane). The solvent was removed by vacuum evaporation and the crude product was chromatographed (15 cm silica gel, ethyl acetate:hexane 9:1 to 1:1). Evaporation of the solvent yielded the pure monomer. The yield ranged from 70-80%.

3a: R = Methyl: colorless solid, $^1\text{H-NMR}$ (300 MHz,

CDCl_3): 1.43 (s, 9H, H9), 2.82 (m, 2H, H3 & H3'), 3.40 (m,

2H, H6), 3.73 (s, 3H, CH_3), 4.16 (m, 2H, H5), 5.03 (br s,

1H, NH), 5.25 & 5.29 (m, 2H, H2 & H2'), 6.46 (m, 2H, H1 & H1'). $^{13}\text{C-NMR}$ (75 MHz, CDCl_3):



28.4 (C9), 39.5 (C6), 46.9 & 47.1 (C3, C3'), 52.4 (R, CH₃), 64.8 (C5), 79.3 (C8), 80.4 & 80.6 (C2, C2'), 136.6 & 136.7 (C1, C1'), 171.6 (C4, C4').

HR-MS (FAB): calc. 341.36, found 342.15.

3b: R = Ethyl: colorless solid, ¹H-NMR (300 MHz, CDCl₃): 1.27 (t, *J* = 7.2 Hz, 3H, CH₂-CH₃), 1.44 (s, 9H, H₉), 2.81 (m, 2H, H₃ & H₃'), 3.39 (m, 2H, H₆), 4.16 (m, 4H, CH₂-CH₃ and H₅), 5.05 (br s, 1H, NH), 5.25 & 5.31 (m, 2H, H₂ & H₂'), 6.46 (m, 2H, H₁ & H₁'). ¹³C-NMR (75 MHz, CDCl₃): 14.4 (R, CH₃), 28.4 (C9), 39.4 (C6), 46.8 & 47.1 (C3, C3'), 61.3 (R, CH₂), 64.8 (C5), 79.3 (C8), 80.4 & 80.7 (C2, C2'), 136.6 & 136.7 (C1, C1'), 155.9 (C7), 171.6 & 171.7 (C4, C4'). HR-MS (FAB): calc. 355.39, found 356.17.

3c: R = Propyl: colorless solid, ¹H-NMR (300 MHz, CDCl₃): 1.25 (m, 3H, CH₂-CH₃), 1.43 (s, 9H, H₉), 1.67 (m, 2H, β-CH₂), 2.81 (q, 2H, H₃ & H₃'), 3.40 (m, 2H, H₆), 4.12 (m, 4H, α-CH₂ and H₅), 5.07 (br s, 1H, NH), 5.24 & 5.30 (m, 2H, H₂ & H₂'), 6.46 (m, 2H, H₁ & H₁'). ¹³C-NMR (75 MHz, CDCl₃): 10.4 (R, CH₃), 21.9 (R, β-CH₂), 28.4 (C9), 39.5 (C6), 46.8 & 47.2 (C3, C3'), 64.8 (C5), 66.9 (R, α-CH₂), 79.3 (C8), 80.4 & 80.7 (C2, C2'), 136.6 & 136.7 (C1, C1'), 155.9 (C7), 171.6 & 171.8 (C4, C4'). HR-MS (FAB): calc. 369.42, found 370.19

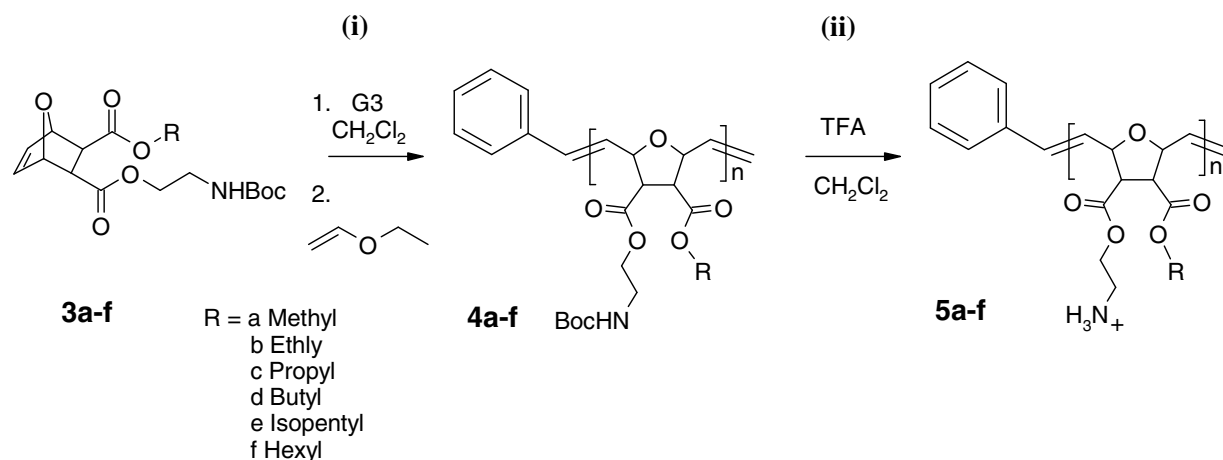
3d: R = Butyl: colorless oil, ¹H-NMR (300 MHz, CDCl₃): 0.93 (m, 3H, CH₂-CH₃), 1.37 (m, 2H, γ-CH₂), 1.44 (s, 9H, H₉), 1.62 (m, 2H, β-CH₂), 2.82 (q, 2H, H₃ & H₃'), 3.39 (m, 2H, H₆), 4.12 (m, 4H, α-CH₂ and H₅), 5.06 (br s, 1H, NH), 5.23 & 5.29 (m, 2H, H₂ & H₂'), 6.46 (m, 2H, H₁ & H₁'). ¹³C-NMR (75 MHz, CDCl₃): 13.7 (R, CH₃), 19.0 (R, γ-CH₂), 28.3 (C9), 30.4 (R, β-CH₂), 39.4 (C6), 46.7 & 47.1 (C3, C3'), 64.7 (C5), 65.1 (R, α-CH₂), 79.2 (C8), 80.4 & 80.6 (C2, C2'), 136.5 & 136.6 (C1, C1'), 155.9 (C7), 171.5 & 171.8 (C4, C4'). HR-MS (FAB): calc. 383.45, found 384.21.

3e: R = iso-Pentyl: colorless oil, ¹H-NMR (300 MHz, CDCl₃): 0.92 (d, *J* = 6.4 Hz, 6H, CH-CH₃), 1.44 (s, 9H, H₉), 1.53 (m, 1H, CH-CH₃), 1.69 (m, 2H, β-CH₂), 2.82 (q, 2H, H₃ & H₃'), 3.41 (m,

2H, H6), 4.18 (m, 4H, α -CH₂ and H5), 5.07 (br s, 1H, NH), 5.23 & 5.29 (m, 2H, H2 & H2'), 6.46 (m, 2H, H1 & H1'). ¹³C-NMR: n.d., HR-MS (FAB): n.d.

3f: R = Hexyl: colorless oil, ¹H-NMR (300 MHz, CDCl₃): 0.88 (t, *J* = 6.6 Hz, 3H, CH₂-CH₃), 1.43 (s, 9H, t-butyl), 1.37 (m, 6H, γ - ϵ -CH₂) 1.62 (m, 2H, β -CH₂), 2.82 (m, 2H, H3 & H3'), 3.42 (m, 2H, H6), 4.14 (m, 4H, α -CH₂ and H5), 5.07 (br s, 1H, NH), 5.23 & 5.29 (m, 2H, H2 & H2'), 6.46 (m, 2H, H1 & H1'). ¹³C-NMR (75 MHz, CDCl₃): 10.9 (R, CH₃), 22.5 (R, δ -CH₂), 15.4 (R, γ -CH₂), 28.3 (C9), 28.5 (R, β -CH₂), 39.3 (C6), 46.6 & 46.9 (C3, C3'), 64.4 (C5), 65.2 (R, α -CH₂), 78.9 (C8), 80.3 & 80.5 (C2, C2'), 136.4 & 136.6 (C1, C1'), 155.8 (C7), 171.5 & 171.7 (C4, C4'). HR-MS (FAB): calc. 411.50, found 412.24.

Polymer Synthesis:



(i) The monomer **3a-f** and G3-catalyst amounts (see Table S1 for details) were dissolved in 1 mL dichloromethane each and subject to three freeze-thaw cycles. The catalyst was added in one shot to the vigorously stirring monomer solution at room temperature under argon. After 30 min, the living polymer chain was end-capped with an excess of ethylvinyl ether (1 mL, 754 mg, 10.5 mmol). The solution was allowed to stir over night. After evaporation of the solvent and

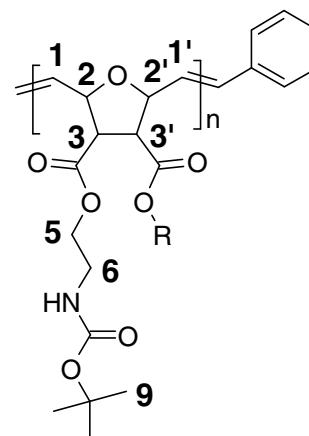
drying, an aliquot of each polymer was taken for GPC and NMR analysis. The product was a brown solid. Attempts to remove the catalyst by filtration over alumina, precipitation into ether/hexane, dialysis and ultrafiltration reduced the brown color, however the color could not be removed quantitatively.

Table S1: Experimental parameters for homopolymer synthesis

Sample	Monomer	N_{repeat} units	M_n Target $g\ mol^{-1}$	$M_{Monomer}$ $g\ mol^{-1}$	$n_{Monomer}$ mmol	$m_{Monomer}$ mg	$M_{Catalyst}$ $g\ mol^{-1}$	$n_{Catalyst}$ mmol	$m_{Catalyst}$ mg
Methyl_3k	Methyl	9.7	3000	341	0.290	100	885	0.030	29.0
Methyl_10k	Methyl	29	10000	341	0.290	100	885	0.010	8.9
Ethyl_3k	Ethyl	8.6	3000	355	0.563	200	885	0.065	57.6
Ethyl_10k	Ethyl	28	10000	355	0.563	200	885	0.020	17.7
Propyl_3k	Propyl	8.4	3000	369	0.282	104	885	0.034	29.7
Propyl_10k	Propyl	29	1000	369	0.278	103	885	0.010	8.50
Butyl_3k	Butyl	7.2	3000	383	0.525	202	885	0.073	64.5
Butyl_10k	Butyl	27	1000	383	0.565	217	885	0.021	18.6
Isopentyl_3k	Isopentyl	7.9	3000	397	0.515	205	885	0.065	57.5
Isopentyl_8k	Isopentyl	20.1	8000	397	1.926	766	885	0.096	85.0
Isopentyl_10k	Isopentyl	26	10000	397	0.508	202	885	0.020	17.5
Hexyl_3k	Hexyl	7.7	3000	412	0.516	212	885	0.067	59.0
Hexyl_10k	Hexyl	25	10000	412	0.493	203	885	0.020	17.7

4a: R = Methyl: 1H -NMR (300 MHz, $CDCl_3$): 1.43 (s, 9H, H9), 3.12 (br m, 2H, H3 & H3'), 3.34 (br m, 2H, H6), 3.71 (s, 3H, CH_3), 4.16 (br m, 2H, H5), 4.71 (m, 1H, H2 & H2' trans), 5.10 (br s, 1 H, H2 & H2' cis), 5.40 (br s, 1H, NH), 5.58 (br m, 1H, H1 & H1' cis) and 5.88 (br m, 1H, H1 & H1' trans).

4b: R = Ethyl: 1H -NMR (300 MHz, $CDCl_3$): 1.24 (s, 3H, CH_2-CH_3), 1.42 (s, 9H, H9), 3.09 (br m, 2H, H3 & H3'), 3.34 (br m, 2H, H6), 4.16 (br m, 4H, CH_2-CH_3 and H5), 4.72 (br m, 1H, H2 & H2' trans), 5.10 (br m, 1H, H2 & H2' cis), 5.30 (br s, 1H, NH), 5.58 (br m, 1H, H1 & H1' cis), 5.88 (br m, 1H, H1 & H1' trans).



4c: R = Propyl: $^1\text{H-NMR}$ (300 MHz, CDCl_3): 0.92 (m, 3H, $\text{CH}_2\text{-CH}_3$), 1.43 (s, 9H, H9), 1.62 (m, 2H, $\beta\text{-CH}_2$), 3.12 (br m, 2H, H3 & H3'), 3.34 (br m, 2H, H6), 4.10 (m, 4H, $\alpha\text{-CH}_2$ and H5), 4.69 (br m, 1H, H2 & H2' trans), 5.12 (br m, 1H, H2 cis & H2'), 5.31 (br m, 1H, H1 & H1' cis), 5.59 (br s, 1H, NH), 5.88 (br m, 1H, H1 & H1' trans).

4d: R = Butyl: $^1\text{H-NMR}$ (300 MHz, CDCl_3): 0.87 (m, 3H, $\text{CH}_2\text{-CH}_3$), 1.29 (m, 2H, $\gamma\text{-CH}_2$), 1.43 (s, 9H, H9), 1.59 (m, 2H, $\beta\text{-CH}_2$), 3.11 (br m, 2H, H3 & H3'), 3.37 (br m, 2H, H6), 4.10 (m, 4H, $\alpha\text{-CH}_2$ and H5), 4.73 (br m, 1H, H2 & H2' trans), 5.11 (br m, 1H, H2 & H2' cis), 5.35 (br s, 1H, NH), 5.59 (br m, 1H, H1 & H1' cis), 5.88 (br m, 1H, H1 & H1' trans).

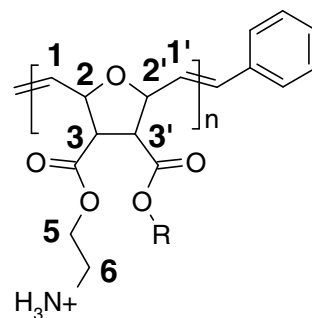
4e: R = iso-Pentyl: $^1\text{H-NMR}$ (300 MHz, CDCl_3): 0.91 (m, 6H, CH-CH_3), 1.43 (s, 9H, H9), 1.52 (m, 1H, CH-CH_3), 1.62 (m, 2H, $\beta\text{-CH}_2$), 3.10 (br m, 2H, H3 & H3'), 3.35 (br m, 2H, H6), 4.15 (m, 4H, $\alpha\text{-CH}_2$ and H5), 4.69 (br m, 1H, H2 & H2' trans), 5.11 (br m, 1H, H2 & H2' cis), 5.34 (br s, 1H, NH), 5.59 (br m, 1H, H1 & H1' cis), 5.88 (br m, 1H, H1 trans).

4f: R = Hexyl: $^1\text{H-NMR}$ (300 MHz, CDCl_3): 0.88 (m, 3H, $\text{CH}_2\text{-CH}_3$), 1.29 (m, 2H, $\gamma\text{-}\epsilon\text{-CH}_2$), 1.43 (s, 9H, H9), 1.56 (m, 2H, $\beta\text{-CH}_2$), 3.10 (br m, 2H, H3 & H3'), 3.36 (br m, 2H, H6), 4.10 (m, 4H, $\alpha\text{-CH}_2$ and H5), 4.70 (br m, 1H, H2 & H2' trans), 5.12 (br m, 1H, H2 & H2' cis), 5.38 (br s, 1H, NH), 5.58 (br m, 1H, H1 & H1' cis), 5.89 (br m, 1H, H1 & H1' trans).

(ii) The crude polymers were dissolved in 2 mL dichloromethane. An excess of TFA (2 mL, 2.97 g, 26.0 mmol) was added and the solution was stirred at room temperature over night. The excess acid was removed by azeotropic distillation with dichloromethane (2x 15 mL) and methanol (1x 15 mL) at the rotary evaporator. The samples were dried in vacuum over night and dissolved in 30 mL Milli-Q water or DMSO depending on solubility. They were dialyzed against

Milli-Q water until the conductivity of the water was 0.1 μ S after 12 h of dialysis (total dialysis time 4-7 days). The hydrolyzed polymers were then freeze dried.

5a: R = Methyl: $^1\text{H-NMR}$ (300 MHz, DMSO): 3.04 (br m, 2H, H3 & H3'), 3.30 (br m, 2H, H6), 3.59 (s, 3H, CH₃), 4.16 (br m, 2H, H5), 4.57 (m, 1H, H2 & H2' trans), 4.91 (br m, 1H, H2 & H2' cis), 5.61 (br m, 1H, H1 & H1' cis), 5.82 (br m, 1H, H1 & H1' trans).



5b: R = Ethyl: $^1\text{H-NMR}$ (300 MHz, D₂O): 1.28 (br s, 3H, CH₂-CH₃), 3.12 (br m, 2H, H3 & H3'), 3.32 (br m, 2H, H6), 3.49-3.81 (br m, 4H, CH₂-CH₃ and H5), 5.10-5.40 (br m, 2H, H2 & H2'), 5.30-5.90 (br m, 2H, H1 & H1').

5c: R = Propyl: $^1\text{H-NMR}$ (300 MHz, D₂O): 0.94 (br m, 3H, CH₂-CH₃), 1.67 (br m, 2H, β -CH₂), 3.31 (br m, 2H, H3 & H3'), 3.50 (br m, 2H, H6), 4.12 (m, 4H, α -CH₂ and H5), 4.39 (br m, 1H, H2 & H2' trans), 5.12 (br m, 1H, H2 & H2' cis), 5.31 (br m, 1H, H1 & H1' cis), 5.88 (br m, 1H, H1 & H1' trans).

5d: R = Butyl: $^1\text{H-NMR}$ (300 MHz, D₂O): 0.93 (m, 3H, CH₂-CH₃), 1.36 (m, 2H, γ -CH₂), 1.61 (m, 2H, β -CH₂), 3.10 (br m, 2H, H3 & H3'), 3.48 (br m, 2H, H6), 4.14 (m, 4H, α -CH₂ and H5), 4.38 (br m, 1H, H2 & H2' trans), 5.11 (br m, 1H, H2 & H2' cis), 5.35-6.20 (br m, 2H, H1 & H1').

5e: R= iso-Pentyl: $^1\text{H-NMR}$ (300 MHz, DMSO): 0.91 (m, 6H, CH-CH₃), 1.55 (m, 1H, CH-CH₃), 1.69 (m, 2H, β -CH₂), 3.31 (br m, 2H, H3 & H3'), 3.48 (br m, 2H, H6), 4.15 and 4.40 (m, 4H, α -CH₂ and H5), 4.69 (br m, 1H, H2 & H2' trans), 5.11 (br m, 1H, H2 & H2' cis), 5.34 (br m, 1H, H1 & H1' cis), 5.88 (br m, 1H, H1 & H1' trans).

5f: R = Hexyl: $^1\text{H-NMR}$ (300 MHz, D₂O): 0.90 (m, 3H, CH₂-CH₃), 1.31 (m, 2H, γ - ϵ -CH₂), 1.62 (m, 2H, β -CH₂), 3.30 (br m, 2H, H3 & H3'), 3.47 (br m, 2H, H6), 4.13 and 4.40 (m, 4H, α -CH₂ and H5), 4.70-5.12 (br m, 2H, H2 & H2'), 5.50-6.20 (br m, 2H, H1 & H1').

Copolymer Synthesis, Deprotection and Characterization: Copolymers were synthesized and hydrolyzed using the same methods as for the homopolymers. Each comonomer was dissolved in 0.7 mL dichloromethane separately under argon. The monomer solutions were mixed, and the respective amount of G3 catalyst in 1 mL dichloromethane was added in one shot. Details are given in Table S2. The NMR spectroscopic data for each repeat unit proton matched the assignment of the analogous protons in the homopolymers. Where possible (no peak overlap), integration revealed that the copolymer composition matched the monomer feed ratio. Deprotection was performed as described for the homopolymers.

Table S2: Experimental parameters for copolymer synthesis: a) ethyl-propyl copolymers, b) methyl-ethyl copolymers, c) methyl-propyl copolymers.

a)

Sample	Monomers	N_{repeat} units	M_n Target g mol^{-1}	n_{Ethyl} mmol	m_{Ethyl} mg	n_{Propyl} mmol	m_{Propyl} mg	$n_{Catalyst}$ mmol	$m_{Catalyst}$ mg
E1:P9	Ethyl:Propyl	10	3500	0.06	20.2	0.54	192	0.060	53.1
E1:P2	Ethyl:Propyl	10	3400	0.20	68.3	0.40	137	0.060	53.1
E1:P1	Ethyl:Propyl	9	3000	0.14	48.0	0.14	51.4	0.037	32.2
E2:P1	Ethyl:Propyl	10	3500	0.40	130	0.20	70.0	0.060	53.1
E9:P1	Ethyl:Propyl	10	3400	0.54	183	0.06	20.1	0.060	53.1

b)

Sample	Monomers	N_{repeat} units	M_n Target g mol^{-1}	n_{Methyl} mmol	m_{Methyl} mg	n_{Ethyl} mmol	m_{Ethyl} mg	$n_{Catalyst}$ mmol	$m_{Catalyst}$ mg
M1:E9	Methyl-Ethyl	10	3400	0.81	276.5	0.09	32.0	0.09	79.7
M1:E1	Methyl-Ethyl	10	3500	0.45	153.6	0.45	159.9	0.09	79.7
M9:E1	Methyl-Ethyl	10	3550	0.09	30.7	0.81	287.9	0.09	79.7

c)

Sample	Monomers	N_{repeat} units	M_n Target g mol^{-1}	n_{Methyl} mmol	m_{Methyl} mg	n_{Ethyl} mmol	m_{Ethyl} mg	$n_{Catalyst}$ mmol	$m_{Catalyst}$ mg
M1:P9	Methyl-Propyl	10	3400	0.81	276.5	0.09	33.2	0.09	79.7
M1:P1	Methyl- Propyl	10	3600	0.45	153.6	0.45	166.2	0.09	79.7
M9:P1	Methyl- Propyl	10	3700	0.09	30.7	0.81	299.2	0.09	79.7

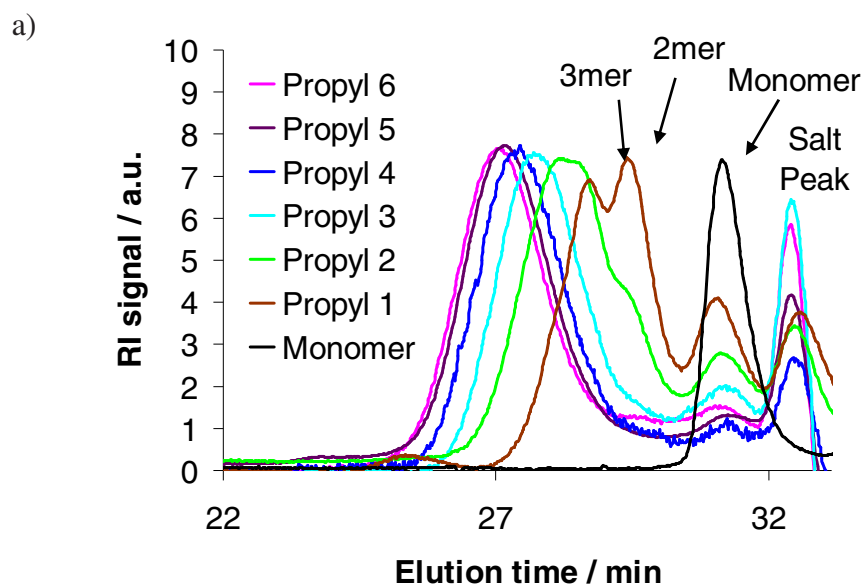
Propyl Oligomer Synthesis

Propyl Oligomers were synthesized and hydrolyzed using the same methods as for the homopolymers. Experimental data are included in Table S3. GPC traces and distribution functions obtained from the MALDI-TOF MS spectra of all oligomers are included in Fig. S1.

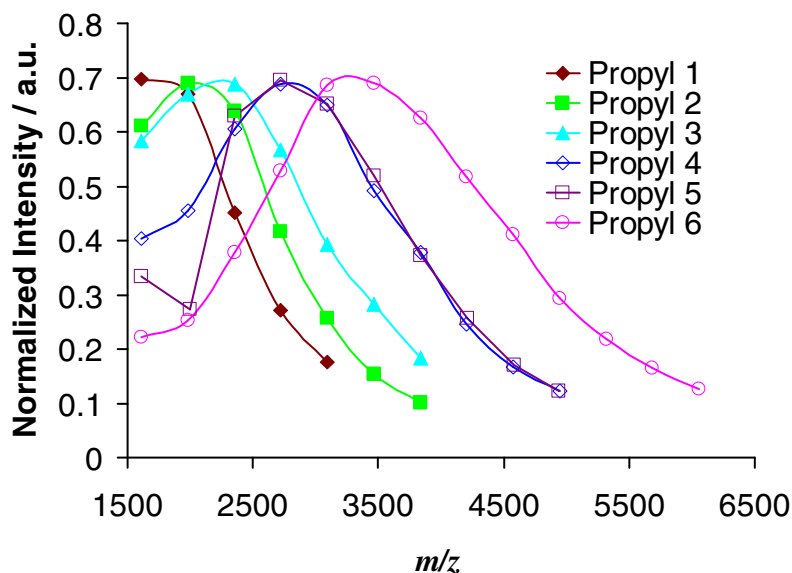
Table S3: Experimental parameters for oligomer synthesis

Sample	Monomers	N_{repeat} units	M_n Target g mol ⁻¹	n_{Propyl} mmol	m_{Propyl} mg	$n_{Catalyst}$ mmol	$m_{Catalyst}$ mg
Oligo 1	Propyl	2	740	0.18	66.5	0.09	79.7
Oligo 2	Propyl	3	1100	0.27	99.7	0.09	79.7
Oligo 3	Propyl	4	1480	0.36	133.0	0.09	79.7
Oligo 4	Propyl	5	1850	0.45	166.2	0.09	79.7
Oligo 5	Propyl	6	2220	0.54	199.5	0.09	79.7
Oligo 6	Propyl	7	2600	0.18	66.5	0.09	79.7

Fig S1: Characterization of ALL propyl oligomers. a) GPC traces, b) MALDI-TOF MS peaks and distributions



b)



Biological Data

Representative curves for the determination of the MIC₉₀ as well as the HC₅₀ values are given in the following section. The experiments were performed as described previously.² The MIC₉₀ data was obtained from the plot of % growth vs. concentration. The concentration value below 10% growth was taken as the MIC₉₀. Controls with inactive polymers and polymers with known activity were included on each plate to ensure data reproducibility. Additionally, the data was collected in quadruplicate. Representative examples are given below (Fig. S2-S3). An example of the dose-response behavior for the different bacteria is shown in Fig. S2 c) and d). The shape of the dose-response curve is largely the same for both bacteria. What is sometimes seen for hydrophobic polymers such as **Propyl_3k** is that they appear to become less toxic at higher concentrations. This is an artifact that comes from decreased solubility or aggregation at these concentrations. When the bacterial count is determined in these wells, there are zero remaining

viable bacteria, or CFUs. Examples of typical HC_{50} curves are shown in Fig S4. The HC_{50} value was taken as the average value of the data at 50 % lysis.

Fig. S2: Examples of a highly active polymer: a) **E1:M1** copolymer, MIC testing against *S. aureus*, $MIC_{90} = 12.5 \mu\text{g/mL}$, Trial 1 and 2, b) same system, Trial 3 and 4, c) **Proyl_3k**, MIC testing against *E. coli*, d) Proyl_3k, MIC testing against *S. aureus*.

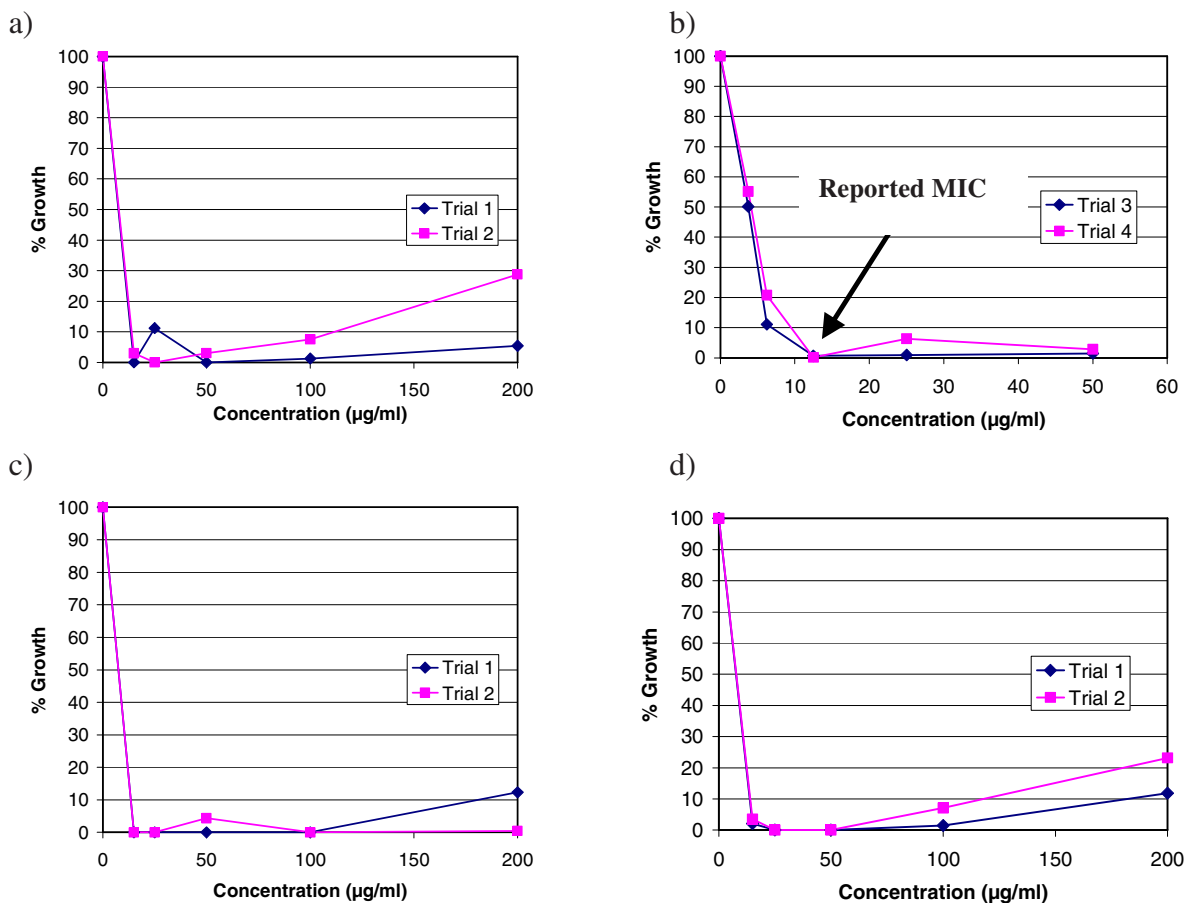


Fig. S3: Example of a) a moderately active polymer: **Propyl_10k**, MIC testing against *S. aureus*,, MIC₉₀ = 200 µg/mL, and b) an inactive polymer: **Propyl_10k**, MIC testing against *E. coli*,, MIC₉₀ >200 µg/mL

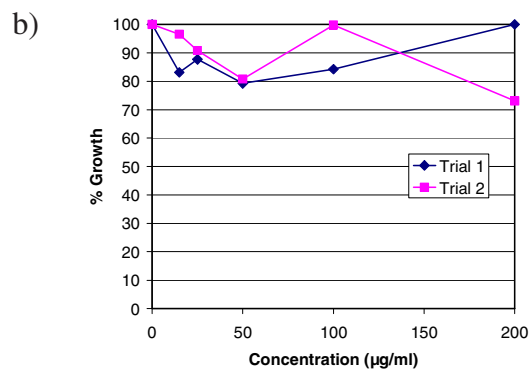
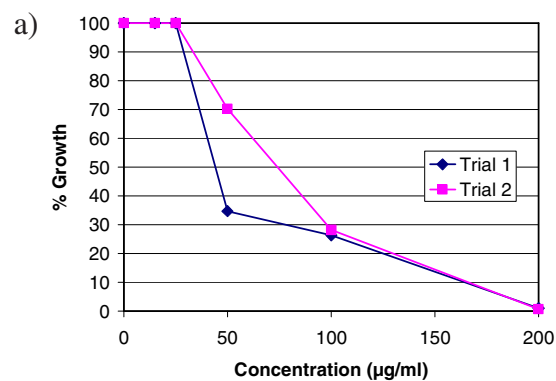
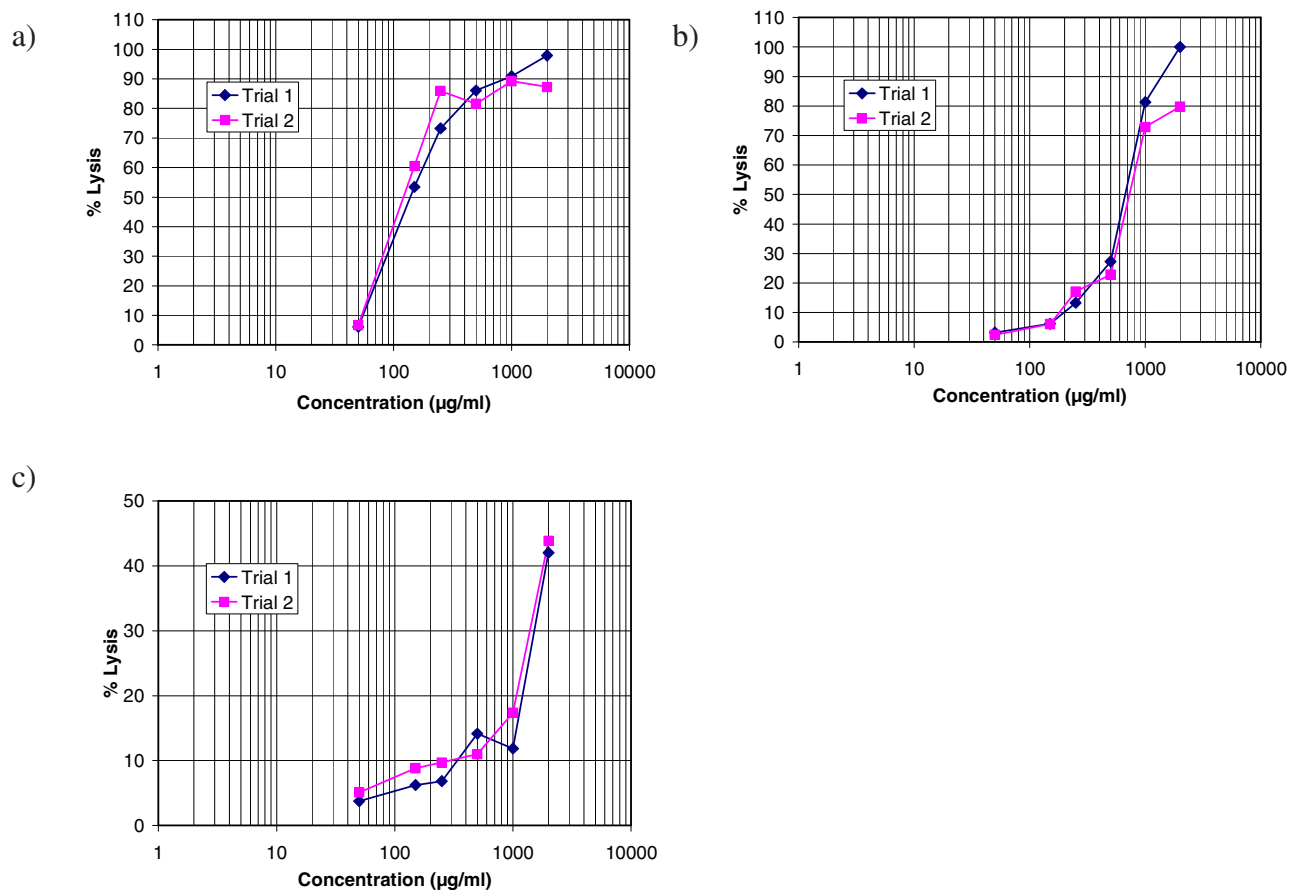


Fig. S4: HC₅₀ curve of a) a very hemolytic polymer: Oligo 6, HC₅₀ = 150 μg/mL; b) a moderately hemolytic polymer: E1:M9, HC₅₀ = 700 μg/mL; and c) a non-hemolytic polymer: M9:P1, HC₅₀ > 2000 μg/mL



To compare the differences between our reported MIC₉₀ (Tables 1b, 2b and 3b), and the value of MIC₁₀₀ which is sometime reported, Table S4 reports these MIC₁₀₀ values. When they differ from the MIC₉₀ values in the main text, they are highlighted in blue.

As can be seen when comparing the tables, the difference is usually a factor of two and for the very high selectivities polymers the difference is less since we obtained a clean 100% kill.

Table S4: Characterization of the antimicrobial homopolymers by biological assays: Inhibitory activity towards bacterial growth of *E. coli* and *S. aureus* bacteria. (MIC_{100} = minimal inhibitory concentration preventing 100% bacterial growth) and hemolytic activity towards red blood cells (HC_{50} = concentration lysing 50% of blood cells). a) Homopolymers, b) oligomers, and c) copolymers.

a)

Sample	MIC_{100} $\mu\text{g mL}^{-1}$		HC_{50} $\mu\text{g mL}^{-1}$	Selectivity	
	<i>E. coli</i>	<i>S. aureus</i>		<i>E. coli</i>	<i>S. aureus</i>
Methyl_3k	>200	>200	2000	<10	10
Ethyl_3k	50	200	1400	28	7
Propyl_3k	6.25	25	50	8.0	2.0
Butyl_3k	15	25	<50	<3.3	<2.0
Isopentyl_3k	12.5	50	<50	<4.0	<1.0
Hexyl_3k	>200	>200	<50	<0.3	<0.3
Methyl_10k	>200	>200	50	<10	<10
Ethyl_10k	>200	>200	1250	<6.3	<6.3
Propyl_10k	6.25	>200	<50	<8.0	<0.25
Butyl_10k	20	>200	<50	<2.5	<0.3
Isopentyl_8k	50	>200	<50	<0.3	<0.3
Isopentyl_10k	50	200	<50	<0.3	<0.3
Hexyl_10k	100	>200	n.d.	n.d.	n.d.

b)

Sample	MIC_{100} $\mu\text{g mL}^{-1}$		HC_{50} $\mu\text{g mL}^{-1}$	Selectivity	
	<i>E. coli</i>	<i>S. aureus</i>		<i>E. coli</i>	<i>S. aureus</i>
Monomer (3c , hydrolyzed)	>200	>200	n.d.	n.d.	n.d.
Oligo 1	>200	6.25	1050	5.25	168
Oligo 2	200	12.5	800	4.0	64
Oligo 3	200	25	1250	6.3	50
Oligo 4	>200	50	>2000	>10	40
Oligo 5	100	>200	1000	10	5.0
Oligo 6	50	50	150	3.0	3.0
Propyl_3k	6.25	25	50	8.3	2.0
Propyl_10k	6.25	>200	<50	<8.0	<0.25

c)

Sample	Monomer	MIC_{90} $\mu\text{g mL}^{-1}$		HC_{50} $\mu\text{g mL}^{-1}$	Selectivity	
		<i>E. coli</i>	<i>S. aureus</i>		<i>E. coli</i>	<i>S. aureus</i>
Propyl_3k	Propyl	6.25	25	50	8.0	2.0
E1:P9	Ethyl:Propyl	50	15	<50	<1.0	<3.3
E1:P2	Ethyl:Propyl	15	15	<50	<3.3	<3.3
E1:P1	Ethyl:Propyl	100	200	1000	10	5.0
E2:P1	Ethyl:Propyl	>200	200	>2000	10	>10
E9:P1	Ethyl:Propyl	>200	50	500	<2.5	10
Ethyl_3k	Ethyl	50	200	1400	28	7
Methyl_3k	Methyl	>200	>200	2000	<10	<10
M9:E1	Methyl-Ethyl	>200	12.5	700	<3.5	56
M1:E1	Methyl-Ethyl	>200	25	1200	<6.0	48
M1:E9	Methyl-Ethyl	>200	12.5	1500	<7	120
Ethyl_3k	Ethyl	50	200	1400	28	7
Methyl_3k	Methyl	>200	>200	2000	10	<10
M9:P1	Methyl-Propyl	>200	3.75	> 2000	10	533
M1:P1	Methyl-Propyl	>200	3.75	> 2000	10	533
M1:P9	Methyl-Propyl	>200	3.75	> 2000	10	533
Propyl_3k	Propyl	6.25	25	50	8.3	2.0

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