Supporting Information

Photocatalytic Oxygenation of 10-Methyl-9,10-dihydroacridine by O₂ with Manganese Porphyrins

Jieun Jung,[†] *Kei Ohkubo*,[†] *David P. Goldberg*,^{*,‡} *and Shunichi Fukuzumi*^{*,†}

[†]Department of Material and Life Science, Graduate School of Engineering, ALCA, Japan Science and Technology Agency, Osaka University, Suita, Osaka 565-0871, Japan

[‡]Department of Chemistry, The Johns Hopkins University, Baltimore, Maryland 21218, United States

E-mail: fukuzumi@chem.eng.osaka-u.ac.jp, dpg@jhu.edu



Figure S1. Time courses of formed Acr=O for the photocatalytic reaction under photoirradiation ($\lambda > 480 \text{ nm}$) of an N₂-saturated PhCN solution (2.0 mL) containing (TMP)Mn^{III}(OH) (blue, $1.0 \times 10^{-5} \text{ M}$) or (TPFPP)Mn^{III}(CH₃COO) (red, $1.0 \times 10^{-5} \text{ M}$) and AcrH₂ (0.02 M) at room temperature.



Figure S2. Time courses of formed Acr=O without photoirradiation of an O₂-saturated PhCN solution (2.0 mL) containing (TMP)Mn^{III}(OH) (blue, 1.0×10^{-5} M) or (TPFPP)Mn^{III}(CH₃COO) (red, 1.0×10^{-5} M) and AcrH₂ (0.02 M) at room temperature.



Figure S3. (a) Plots of the formation of Acr=O and (b) plot of the initial reaction rate of the photocatalytic oxygenation of AcrH₂ under irradiation ($\lambda > 480$ nm) of an O₂-saturated PhCN solution containing (TPFPP)Mn^{III}(CH₃COO) (1.0 × 10⁻⁵ M) in the presence of 5 mM (red), 10 mM (green), 20 mM (blue), or 30 mM (black) of AcrH₂ as a substrate at 298 K vs concentration of AcrH₂.



Figure S4. (a) Transient absorption spectral changes (red after 1 ps, orange after 2 ps, green after 10 ps, blue after 50 ps, and black after 2800 ps) after photoexcitation of (TPFPP)Mn^{III}(CH₃COO) in PhCN. Decay time profiles of absorbance at 588 nm due to [(TPFPP)Mn^{III}(CH₃COO)]^{*} (⁵T₁) (b) in N₂-saturetad PhCN and (c) in O₂-saturated PhCN. Time profiles of the decay of [(TPFPP)Mn^{III}(CH₃COO)]^{*} (⁷T₁) at λ = 588 nm under (d) N₂ and (e) O₂.



Figure S5. Phosphorescence spectra of singlet oxygen ($\lambda_{max} = 1270 \text{ nm}$) obtained by photoexcitation of C₆₀ (blue), (TMP)Mn^{III} (green), (TPFPP)Mn^{III} (red), and only solvent (black) as a blank in O₂-saturated C₆D₆.

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